A Publication of Strem Chemicals, Inc. THE STREM CHEMIKER VOL. XXX No. 1 February, 2018

High Surface Area Silica Nanoparticles by Jean-Marie Basset and Pradeep Doggali

Table of Contents

Biographical Sketches:		of contents	
Prof. Jean-Marie Basset, Ph.I Dr. Pradeep Kumar Doggali	D		1 2
Article 1: High Surface Area Fibro	ous Silic	a Nanoparticles (KCC-1)	3-23
High Surface Area Silica Nanor	oarticle	s Available from Strem	24-25
New Products Introduced Since	e Chem	niker XXIX (July 2016)	26-99
New Kits Introduced Since Che	miker 2	XXIX (July 2016)	101-110
AntPhos and WingPhos Kit BI-DIME Ligand Kit Buchwald Biaryl Phosphine L Bond Formation, Suzuki C Buchwald Palladacycle Preca biphenyl-2-yl-Palladacycle CalB immo KIT™ - Immobilize Graphene Quantum Dots (GC Graphene Quantum Dots (GC Graphene Quantum Dots (GC Lipase immo Kit - Immobilize Solvias Josiphos Nickel Catal	igand M oupling italyst K s Gen. 3 ed enzy RKIT1. QDs) Ma QDs) Mi QDs) Mi d enzym yst Kit	lini Kit 3 for Aromatic Carbon-Heteroatom and Negishi Cross-coupling it 2b (Methanesulfonato-2'-amino-1,1'- 3) me aster Kit ni Kit (Powders) ni Kit (Liquids) nes	101 102 103-104 105-106 107 107 108 108 109 109 110
Available Booklets			112
American Chemical Society Award for Distinguished Service in the Advancement of Inorganic Chemistry	2018 2017 2016 2015	Prof. Thomas B. Rauchfuss, University of Illinois Prof. William B. Tolman, University of Minnesota Prof. Vincent L. Pecoraro, University of Michigan Prof. Kim R. Dunbar, Texas A&M University	
Canadian Society for Chemistry Award for Pure or Applied Inorganic Chemistry	2018 2017 2016 2015	Assoc. Prof. Eric Rivard, University of Alberta Prof. Dwight Seferos, University of Toronto Prof. Curtis P. Berlinguette, University of British Colu Prof. Muralee Murugesu, University of Ottawa	ımbia
Headquarters			
Strem Chemicals, Inc. 7 Mulliken Way Newburyport, MA 01950 USA		Tel: (978) 499-1600 Fax: (978) 465-3104 Toll free numbers below US & Canada or Tel: (800) 647-8736 Fax: (800) 517-8736 Email: info@strem.com	nly
European Offices			
15, rue de l'Atome Zone Industrielle F-67800 BISCHHEIM, France		Tel: +33 (0) 3 88 62 52 60 Fax: +33 (0) 3 88 62 26 81 Email: info.europe@strem.com	
Postfach 1215 D-77672 KEHL, Germany		Tel: +49 (0) 7851 75879 Fax: +33 (0) 3 88 62 26 81 Email: info.europe@strem.com	

Strem UK An Independent Distributor of Strem Chemicals Products Newton Hall Town Street, Newton Cambridge, CB22 7ZE UK

©Copyright 2018 by



The Strem Chemiker Vol. XXX No.1 February, 2018

www.strem.com

Tel: 0845 643 7263

Fax: 0845 643 7362

Email: enquiries@strem.co.uk



Cover design and art by Renegade Studios

Glossary of Terms

[α] _p	 Specific rotation
AAS	 Atomic Absorption Standard
ACS	 Conforms to American Chemical Society specifications
air sensitive	 Product may chemically react with atmospheric oxygen or carbon dioxide
	at ambient conditions. Handle and store under an inert atmosphere of
	nitrogen or argon.
amp	 Ampouled
b.p.	 Boiling point in °C at 760mm, unless otherwise noted
d.	 Density
dec.	 Decomposes
elec. gr.	 Electronic Grade, suitable for electronic applications
f.p.	 Flash point in °F
gran.	 Granular
heat sensitive	 Product may chemically degrade if stored for prolonged periods of time at
	ambient temperatures or higher. Store at 5°C or lower.
hydrate	Unspecified water content which may vary slightly from lot to lot
hydroscopic	 Product may absorb water if exposed to the atmosphere for prolonged
	 periods of time (dependent on humidity and temperature). Handle and
	store under an inert atmosphere of nitrogen or argon
light sensitive	Product may chemically degrade if exposed to light
lia	 Liquid
mn	 Melting point in °C
moisturo sonsitivo	 Product may chemically react with water. Handle and store under an inert
moisture sensitive	 atmosphere of nitrogen or argon
NMP grado	Suitable as a Nuclear Magnetic Deconance reference standard
	For optical applications
optical grade	 Pol optical applications
pwur.	 Lload to propage reference standards and standardize valumetric solutions
	 Dised to prepare reference standards and standardize volumetric solutions
FURAIREN	 A grade higher than technical, often used where there are no official
purmed	 A grade higher than technical, often used where there are no official
D Vel	Standards
P. VOI.	 Pore volume
pyropnoric	 Product may spontaneously ignite if exposed to air at ambient conditions
reagent	 High purity material, generally used in the laboratory for detecting,
	measuring, examining or analyzing other substances
REU	 Rare Earth Oxides. Purity of a specific rare-earth metal expressed as a
	percentage of total rare-earths oxides.
SA	 Surface area
store cold	 Product should be stored at -18°C or 4°C, unless otherwise noted (see
	product details)
subl.	 Sublimes
superconductor grade	 A high purity, analyzed grade, suitable for preparing superconductors
tech. gr.	 Technical grade for general industrial use
TLC	 Suitable for Thin Layer Chromotography
v.p.	 Vapor pressure mm of Hg
xtl.	 Crystalline

About Purity

Chemical purity Metals purity	 is reported after the chemical name, e.g. Ruthenium carbonyl, 99% is reported in parentheses with the respective element, e.g. Gallium (III) bromide, anhydrous, granular (99.999%-Ga) PURATREM where 100% minus the metal purity is equal to the maximum allowable percentage of trace metal impurity
	percentage of trace metal impunty

Biographical Sketches



Prof. Jean-Marie Basset, Ph.D.

Prof. Jean-Marie Basset is a Distinguished Professor in the Physical Sciences and Engineering Division at King Abdullah University of Science & Technology (KAUST). From his start at KAUST in 2009 until October 2017 he served as Director of the KAUST Catalysis Center (KCC) of which he was also a founding member. He continues his work today at the KCC in Chemical Science.

Catalysis is a strategic domain for the world and particularly Saudi Arabia. It is a critical enabling science for the Kingdom's energy future. Improvement in catalytic processes across the chemical and petroleum industries will increase resources, energy utilization efficiencies and reduce waste. The need for breakthroughs within

the crucial domains of catalysis are imperative to meet the energy challenges "beyond petroleum" such as the use of CO_2 Natural Gas, CO utilization, and efficient photocatalysts for water splitting. Prof. Basset has been working to meet some of these challenges and help to catalysis field to advance from catalyst discovery to catalysis by design. He has recently developed a theory to predict any catalytic reaction on the basis of the catalysis by design using single site catalysis.

Before his move to KAUST, Prof. Basset served as the Research Director at the Centre National de la Recherche Scientifique (CNRS) since 1987. He first came to CNRS in 1971 and occupied several positions in his tenure including vice-director of the Institute of Catalysis (Lyon). He founded the consortium "Actane" on alkane activation with 11 university labs and 5 different companies. He founded his Laboratory of Surface Organometallic Chemistry in 1994. During this time he was also acting Scientific Director of the School of Chemistry, Physics and Electronics (CPE) at the University of Lyon in France (1992-1994). Prof. Basset's Lyon lab was home to over 50 scientists, including Nobel Laureate Yves Chauvin. This lab has trained over 450 chemists through three-year scholarship programs.

He has written more than 600 publications in international journals and holds 70 patents most of which are protected internationally.

Prof. Basset's achievements include:

- 2011 Member of the European Academy of Sciences and Arts (AUS)
- 2008 Doctor Honoris Causa, University of Xiamen (CN)
- 2008 Doctor Honoris Causa, TUM (DE)
- 2006 Augustine Award of the ORCS (USA)
- 2005 Distinguished Achievements Award of IMPI (USA)
- 2003 Chevalier dans l'Ordre National du Mérite (FR)
- 2003 Member of the European Academy of Sciences (EU)
- 2002 "Visiting Professor University of Hokkaido" (JAP)
- 2001 Member of the French Academy of Technologies (FR)
- 1999 "August-Wilhem-Von-Hofman-Vorselung" Lecturer (DE)
- 1998 Academy of Sciences Award, Prix de l'Institut Français du Pétrole (FR)
- 1998 Seaborg Lecturer in Inorganic Chemistry (Université de Berkeley) (USA)
- 1997 Procope Award for French –German collaboration (FR)
- 1997 Société Française de Chimie (Grand Prix) Prix Süe 1997 (FR)
- 1993 Corresponding member of the de French Academy of Sciences (FR)
- 1992 Grammaticakis Neuman Award of the French Academy of Sciences (FR)
- 1991 Max Plank Award with W. Herrmann (DE)
- 1987 Japan Society for the Promotion of Sciences Award (JAP)
- 1987 Alexander Von Humboldt Award (DE)
- 1984 Pacific Coast Lecturer West Coast (USA)

Biographical Sketches



Dr. Pradeep Kumar Doggali

Dr. Pradeep Kumar Doggali is a Postdoctoral Fellow at King Abdullah University of Science and Technology (KAUST), Saudi Arabia. He completed his Master of Science in Chemistry (Specialization: Industrial Chemistry) in the year 2007 from Jawaharlal Nehru Technological University (JNTU), Hyderabad, India. That same year, he joined the National Environmental Engineering Research Institute (NEERI), a laboratory of Council of Scientific and Industrial Research (CSIR) in India to pursue his doctoral studies. His research work focused on developing Perovskite and Mixed Oxide type catalytic materials for coal combustion and post combustion control of carbon monoxide (CO) and particulate matter (PM) emissions. During his Ph.D. studies, he was

on deputation to Kyushu University, Japan as a visiting research student (2009) at the Laboratory of Prof. Y. Teraoka. There he worked on transition metal-based perovskite type catalysts for the control of diesel soot emissions from vehicles and stationary sources like generator sets. In 2012 he joined the R&D unit of Larson and Toubro (L&T) Power Limited (Vadodara, India) as a Research Executive. At L&T, Dr. Doggali contributed in the development of suitable biomass additives for the clean combustion of coal for gasification based systems.

As a Research Consultant in KAUST (2013), he worked on developing shape, size and morphology controlled nano structured metal oxides and mixed oxides using template assisted routes. Dr. Doggali's postdoctoral work (2015) is primarily in the area of developing improved methods to synthesize Nano Fibrous Silica (KCC-1) to make it suitable for a variety of industrial applications. He is also developing various nanoparticle supported KCC-1 based catalysts for different reactions of significance for use in energy and environmental fields. He was conferred with DST (Department of Science and Technology), India, Young Scientist award in the year 2015.

Awards:

- GLOBAL-COE Internship Award at Kyushu University, Japan
- Best business plan presentation award at Technology Led Entrepreneurship Program of Indian Institute of Chemical Technology (CSIR-IICT) by the faculty of Indian Institute of Management (IIM)
- Best Oral presentation award on "Enviro 2025-Challenges, Interventions & Mitigations" held at NEERI, Nagpur, India.
- Awarded Senior Research Fellowship (SRF) by CSIR (Council of Scientific and Industrial Research, Government of India).

High Surface Area Fibrous Silica Nanoparticles (KCC-1)

Jean-Marie Basset and Pradeep Doggali KAUST Catalysis Center 4700 King Abdullah University of Science & Technology (KAUST), Thuwal 23955-6900, Saudi Arabia

Introduction

Given the growing demand for silica materials for various applications including catalysis, drug delivery, bio sensing, liquid chromatography, coating & paints, optically active materials, as well as in cosmetics, much research has been carried out to control the shape, structure, and pore-size leading to various types of mesoporous silica materials. However, the performance, and therefore the applicability, of mesoporous silica has been hindered by the limitation in accessibility to the active sites within the porous structure for many reasons, such as clogging of the pores by the sintering of active metals and also by accumulation of the carbonaceous by products inside the pores or at the external surface of porous materials. The morphology of the mesoporous silica could also influence the way drugs gets encapsulated and their subsequent release. Therefore, development and fabrication of novel silica materials with unique morphology that leads to high performances, which results in super efficiency is imperative to meet the emerging applications.

KAUST's novel fibrous silica nanospheres (KCC-1)^[1] offers a unique alternative shape, other than porosity, that has never before been seen in silica materials: a fibrous surface morphology arranged in three dimensional structure to form spheres. Unlike traditional pore-based silica, these nanospheres possess a fibrous structure that increases accessibility to most of the available surface area; this, for example, increases catalytic activity significantly. KCC-1 has been attracting the attention of researchers all over the world owing to its unique morphology and high surface area and therefore, making it ideal for use in a wide variety of catalytic reactions, sorbent and biological applications such as immobilization of enzymes and delivering DNAs.

A range of Nano catalysts synthesized using KCC-1 as a support have been showing excellent catalytic activity for various reactions of industrial importance. As a catalyst support/sorbent/carrier, KCC-1 has been demonstrating superior activity as compared to commercially available mesoporous silica materials such as MCM-41 and SBA-15. Some of these KCC-1 based materials and their corresponding reactions are included in Table.1.

S.N	KCC-1 based material/catalyst/sorbent	Catalytic reaction/sorbent application/ carrier for DNAs and genes	Reference
1.	Tantalum hydride catalyst supported on KCC-1	Hydrometathesis of olefins	2
2.	Ruthenium (Ru) nanoparticles supported on KCC-1	Hydrogenolysis of Propane	3
3.	KCC-1 based silver nanocatalyst	Hydrogenation of dimethyl oxalate (DMO)	4
4.	Amine-functionalized KCC-1	CO ₂ Capture	5
5.	Nitridated KCC-1	Knoevenagel condensation and transesterification	6
6.	KCC-1 nanospheres	CO ₂ methanation	7
7.	KCC-1 supported Ag catalysts	CO oxidation	8
8.	Palladium nanoparticles supported on KCC-1	hydrogenation of alkenes and α , β -unsaturated carbonyl compounds	9
9.	KCC-1 nanoparticles	Synthesis of triazolo[1,2-a] indazole-triones	10
10.	Manganese (Mn) incorporated KCC-1	catalytic ozonation of oxalic acid	11
11.	Amine-grafted KCC-1	As a carrier to transport and deliver DNAs & genes	12
12.	Atomic layer deposited (ALD) TiO_2 on KCC-1	Photocatalytic dye degradation	13

Our group at catalysis research center (KCC) of KAUST has been working on catalytic applications using KCC-1 for different reactions. Successful attempts have been made in studying KCC-1 based catalysts for olefin and alkane metathesis, C-C Coupling, H₂ production from natural gas, oxidation reactions and photocatalytic water splitting. Our research work on above mentioned application using KCC-1 is briefly discussed.

I) Hydro-metathesis of olefins

Hydro-metathesis is a catalytic reaction of transition metal hydride, which in the presence of hydrogen catalyzes the direct conversion of olefins into alkanes having higher and lower numbers of carbon atoms. In this work, Tantalum hydride grafted on KCC-1 transforms an olefin in the presence of hydrogen at moderate temperatures into the expected corresponding alkane, and also transforms the same olefin into alkanes having a higher and lower number of carbon atoms. The catalytic hydro-metathesis of propylene (1:1 mixture of olefin and hydrogen) was performed and compared with 1-butene hydrometathesis on a micropilot equipped with a stainless steel reactor at atmospheric pressure at 150°C. The reactions proceeded efficiently with 37–40% conversion of propylene and 48-51% of 1-butene. In the case of propylene, in addition to the expected hydrogenation product propane, butane (35–40%) and ethane (40%) were the major products formed, with methane, iso-butane, and pentanes as the minor products. In case of 1-butene, major products formed include butane, propane (47–61%) and hexanes (13–25%) whereas ethane, propylene, pentanes, and heptanes formed as minor products. Remarkably, in both cases catalyst possessed high stability and regeneration ability^[2]. Graphical representation of the process is presented below.



II) Hydrogenation of alkenes

Hydrogenation of alkenes and α , β -unsaturated carbonyl compounds have been studied using an efficient KCC-1 based catalyst (palladium nanoparticles supported on KCC-1). This resulted excellent yields of the corresponding products with very high chemo-selectivity in the case of α , β -unsaturated carbonyl compounds. We have observed the superior performance of KCC-1 in comparison with commercial mesoporous silica supports like MCM-41 and SBA-15 under identical conditions. These results are attributed to superior accessibility of catalytically active sites along with high loading and excellent dispersion of palladium in case of KCC-1^[10].



1,3-diphenyl-2-propene-1-one

1,3-diphenyl-2-propane-1-one Butane-1,3-diyldibenzene

III) Influence of metal particle (Au nanoparticles) size on catalytic CO oxidation

In order to study the possible advantage of using the fibrous nature of KCC-1, we synthesized gold nanoparticles (Au NPs) with different size and supporting them on KCC-1. The activity of the resulting catalysts was and tested for CO oxidation reaction. It is evidenced by the HR-TEM studies that the size and the location of the Au NPs on the support observed to be dependent on the method of synthesis. The catalytic activity of the Au NPs proved to be size dependent. We have also observed that larger Au NPs of Au/KCC-1- NH₂ were not able to penetrate the pores of KCC-1 and showed only limited catalytic activity. In this study, it was possible to place highly dispersed Au NPs with uniform size inside the fibers of KCC-1-NH₂ by grafting HAuCl₄ on Au/KCC-1-NH₂ followed by reduction^[14].



IV) Coupling of methane to ethane and hydrogen

We have developed a method to selectively produce hydrogen and ethane from methane by thermocatalytic decomposition of methane using Ru nanoparticles supported on Nano fibrous Silica (KCC-1) catalyst. These catalysts have been synthesized by post modification of KCC-1 with amine groups using APTS (3-aminopropyl triethoxysilane) followed by supporting nanoparticles on KCC-1. In this work, suitable temperatures and pressures have been found to produce a product having either hydrogen with solid carbon or ethane with solid carbon. The novelty of this approach also associated with the catalyst system that is efficient to produce the hydrogen and ethane as mentioned above with feed gas comprising less than 1000 ppm water. Also, the formation of hydrogen and ethane has not produced carbon dioxide (CO₂) but a just carbon. Based on the structure of the carbon (formed in this reaction), it can have applications as carbon graphite/ carbon fiber/ carbon nanotube^[15].



V) Photocatalytic water splitting

We have also been successful in utilizing fibrous nature of KCC-1 for photocatalytic reactions. Triazine-based carbon nitride (CN) synthesis in the presence of KCC-1, resulted in favorable effects to advance photocatalytic stability. The photocatalyst synthesized using this approach yielded a HER photocatalyst with a significant AQE of $22.1 \pm 3\%$ at 400 nm for solar to HER. The improved lifetime for the ground-state bleaching of the excited charge carriers was evidenced by time-resolved transient absorption spectroscopy. The synthesized CN/KCC-1 photocatalyst has been consistent with the higher photocatalytic performance than that of CN without KCC-1^[16].



Experiments are also in progress to test KCC-1 for different industrially important reactions considering its superior performance as compared to conventional silica supports.

Several researchers all over the globe have successfully utilized KCC-1 for various applications after our discovery of KCC-1 in 2010. Therefore, work carried out by research groups from other institutions has been discussed to sum-up the highlights of some potential applications of KCC-1.

VI) Hydrogenation of dimethyl oxalate (DMO) to Methylene Glycol (MG)

Mengyao Ouyang et al. studied Ag/KCC-1 nanocatalyst for hydrogenation of dimethyl oxalate (DMO) to Methyl Glycolate (MG) and they compared the activity of Ag/KCC-1 with Ag/SBA-15 and Ag/MCM- 41. The results showed that dispersion of metals was completely dependent on the nature silica support used, which results in different dispersion and accessibility of active sites. Authors concluded that KCC-1 with hierarchical pore channels has higher accessible internal surface area and therefore displays high catalytic activity^[4].



VII) Hydrogenation of functionalized aromatic compounds

M. Dhiman et al. synthesized ultrasmall platinum (Pt) nanoparticles on PEI (polyethylenimine) functionalized KCC-1 (KCC-1-PEI/Pt) and studied their activity for hydrogenation of 3-nitrostyrene. Authors observed an extraordinary selectivity for the said reaction and Pt (1%) supported on PEI functionalized KCC-1 was found to be selective towards reduction to 3-vinylaniline. On the other hand, 10% and 5% Pt loaded on KCC-1-PEI demonstrated high activity and selectivity towards 3-ethylaniline. It was observed that among studied catalysts KCC-1-PEI/Pt (1%) was found to be an efficient catalyst for the hydrogenation of a variety of substituted aromatic compounds. From obtained results it was described, how selectivity can be tuned just by changing the particle size of Pt nanoparticles. In this study, most active sites were observed to be subnanometer particles and pseudo-single atoms of Pt ^[17].



3-nitroaniline

3-vinylaniline

X. Le et al. investigated palladium supported KCC-1 catalyst for reduction of nitrophenols (4-NP) and hydrodeclorination of chlorophenols (4-CP). In order to synthesize catalysts, Pd precursor was added the functionalized KCC-1 followed by reduction with NaBH4. The Pd nanoparticles were well dispersed on fibers of KCC-1 and showed good catalytic activity for both the reactions. The higher catalytic activities was explained on the basis of morphological characteristics of the support which favors uniform dispersion and accessibility to Pd nanoparticles^[18].



In another one of their studies, X. Le et al. have explored magnetically retrievable core-shell catalyst Pd/Fe₃O₄@SiO₂@KCC-1. The catalytic performance of the synthesized catalyst was determined by reduction reaction of 4-nitrophenol (4-NP) to 4-aminophenol (4-AP). The reduction rate was more for Pd/Fe₃O₄@SiO₂@KCC-1 as compared to Pd on other supports. The catalyst was also tested for Suzuki coupling reaction and found to have good stability and reusability (up to five cycles)^[19]. M. Dhiman et al.

reported a facile way for synthesis of metal nanoparticles (Rh, Ru, Pd) supported on KCC-1, wherein they used PEI as a pseudochelator. Their studies revealed that the use of PEI had an edge over complex dendrimer, making the process relatively inexpensive. The catalytic activities of synthesized catalysts (KCC-1-PEI/Rh, KCC-1-PEI/Ru, and KCC-1-PEI/Pd) were tested for hydrogenation of phenylacetylene and styrene. The activity of KCC-1-PEI/Rh was more for both hydrogenation of phenylacetylene and styrene, as compared to that of KCC-1-PEI/Ru, and KCC-1-PEI/Pd. The difference in the activity of these catalysts was attributed to varying adsorption coefficients for the different metal atoms. The higher activities exhibited by studied catalysts was explained on the basis of easy accessibility to metal atoms with KCC-1 despite PEI infiltration. KCC-1-PEI/Rh and KCC-1-PEI/Ru also showed excellent stability, which was maintained almost up to five cycles without any deterioration^[20].



X. Le et al. also Studied Ni@Au core-shell nanoparticles supported KCC-1 for the reduction of 4-nitrophenol to 4-aminophenol in the presence of NaBH4. The examined catalyst showed superior catalytic activity in comparison with to Ni@Au NPs. This was explained by the high accessibility provided by KCC-1 and also due to the less aggregation of Ni@Au NPs on the KCC-1 nano-silica support. The Ni@Au/KCC-1 was also found to be an excellent catalysts to reduce 2-nitroaniline to o-Phenylenediamine. In this study, Ni@Au NPs as the active sites reduced the use of Au content not effecting the catalytic activity. Also possess paramagnetic nature, which makes the catalyst easy to recover for reuse. Authors have also shown that Ni@Au/KCC-1 can be reused at least ten times without significant reduction in catalytic activity. The sustainable activity was attributed to the inhibition of leaching of Ni@Au NPs from the mercaptopropyl groups functionalized KCC-1^[21].



VIII) C-C bond forming reactions

Several research group have demonstrated use of KCC-1 for various catalytic reactions including, Suzuki coupling^[22], Suzuki–Miyaura cross-coupling^[23], Knoevenagel Condensation^[6] and also synthesizing various industrially important organic compounds.

Fihri et al. investigated Pd nanoparticles supported KCC-1 for Suzuki coupling of aromatic halides. In order to prepare the catalysts, KCC-1 was first functionalized with amine groups using 3-aminopropyltriethoxysilane to obtain KCC-1-NH₂. Later, the functionalized surface was treated with PdCl₂ to prepare KCC-1-NH₂/Pd. Authors found that the synthesized catalyst was active for different aryl bromides and iodides with aryl boronic acids and demonstrates good to excellent yields. It was possible to easily recover the catalyst from the reaction mixture and researchers reused it for a number of cycles with no significant change in the catalyst activity. Thus authors confirms the heterogeneity and good stability of this catalyst system. TEM investigations carried out in this study indicates no change in the morphology of fresh and used catalyst, which indicated no Ostwald ripening^[22].



Aryl bromides

P. Gautam et al. synthesized palladium nanoparticles (Pd NPs) supported on KCC-1 and developed an economically viable and sustainable protocol for the carbonylative Suzuki–Miyaura cross coupling reaction. Pd nano particles supported on KCC-1 demonstrates the turnover number (TON) 28- times and a turnover frequency (TOF) 51-times greater than the best Palladium supported catalyst reported for the carbonylative cross-coupling between 4-iodoanisole and phenylboronic acid. Catalysts tested in this study could be recycled up to ten times with negligible loss in activity till eighth cycle^[23].



IX) Knoevenagel condensation

M.Bouhrara et al. synthesized Nitridated KCC-1 (solid base) by ammonolysis under a flow of ammonia (NH₃). It was found that the synthesized catalyst is highly active for Knoevenagel condensation reactions of different aldehydes. Tested catalysts in this research work were active for the transesterification of esters with a wide range of alcohols compared to the reported catalysts. Authors explained the improvement in activity by the presence of both basic (amines) and acidic (silanols) sites and more importantly excellent accessibility of these sites in case of KCC-1. The unique structure of KCC-1 assists the substrates to easily penetrate and interact with basic amine and helps in improving the catalytic activity^[6].



Benzaldehyde derivatives Active methylene compounds



X) Synthesizing different organic compounds

S.M. Sadeghzadeh explored KCC-1 based catalysts for construction of different organic compounds. Author reported a green and efficient method for synthesis of various triazolo [1,2-a]indazole-triones using KCC-1 catalyst at room temperature under solvent free conditions by visible light. Nano-SiO₂, MCM-41, SBA-15 were also tested for synthesis of triazolo [1,2-a]indazole-triones using the same conditions. It was observed that the yields of the desired product was fair to good with Nano-SiO₂, MCM-41, SBA-15. However, KCC-1 shows excellent yields for the same reaction. Non-negligible activity observed using KCC-1 was explained based on its shape, composition and morphology^[10].



triazolo[1,2-a]indazole-triones

In a different study, S.M. Sadeghzadeh synthesized ionic liquid functionalized KCC-1 and their activity was tested for the synthesis of quinazoline-2,4 (1H, 3H)-diones from CO₂ and 2-aminobenzonitriles under mild conditions. Author compared the catalytic performance of ionic liquid functionalized KCC-1 and literature reported catalysts. Results in this study clearly indicate that the synthesized catalysts show good to excellent yields at lower temperature with small amount of catalyst, lower pressure of carbon dioxide and shorter reaction time. The synthesis was shown to be green, low-cost and was effective for the development of other materials^[24].



quinazoline-2,4(1H,3H)-dione

The author also investigated $\text{Fe}_{3}O_4$ /KCC-1/BPAT MNPs core–shell nanocomposite for the synthesis of N-substituted 1,4-dihydropyridines with excellent yield under mild reaction condition. It was found that the studied catalyst was non-toxic, green and economically viable and recyclable for the synthesis of N-substituted 1,4-dihydropyridines^[25].



P. K. Kundu et al. studied KCC-1-PEI/Pd catalysts for decarbonylation of aldehydes, wherein they developed an efficient heterogeneous catalytic protocol for the decarbonylation of a variety of aldehydes with complete conversion to products. By doing this, authors have found an easy way for product purification (without chromatography) using KCC-1-PEI/Pd catalyst. The developed method in this study not only limited to aromatic aldehydes, but the decarbonylation of hetero aromatic, alkane, and alkenylaldehyde derivatives. Authors found that the studied catalyst have excellent recyclability even after eightcyles, which proves the sustainability of the studied catalyst system^[26].



S. Sadeghzadeh synthesized palladium nanoparticles (Pd NPs) dispersed on KCC-1 functionalized with different groups such as APTPOSS@KCC-1, hyperbranched polyglycerol@KCC-1 (HPG@KCC-1), and ionic liquid@KCC-1 (IL@KCC-1). These catalysts are evaluated for synthesis of benzo[c] pyrazolo[1,2-a]cinnolin-1-one. It was observed that APTPOSS@KCC-1 displayed good catalytic

reaction and also exhibited excellent reusability in the catalytic reaction. The high activity of APTPOSS functionalized KCC-1 was attributed to its high accessibility and minimum aggregation. Also its tolerance to leaching of the nanoparticles^[27].



benzo[c]pyrazolo[1,2-a]cinnolin-1-one

XI) Adsorbent for CO₂ capture

U. Patil et al. studied the use of silicon oxynitrides as adsorbents for CO₂ capture. In this investigation authors synthesized three series of functionalized materials based on KCC-1, SBA-15 and MCM-41 with Si–NH₂ groups in a single step process via thermal ammonolysis. Authors demonstrates that studied materials overcome several limitations associated with conventional aminegrafted mesoporous silica. Adsorbents prepared by ammonalysis shows good CO₂ capture capacity, faster adsorption–desorption kinetics as well as effective regeneration and reuse. Also the studied materials demonstrate excellent thermal and mechanical stability^[28]. Graphical representation of the process is presented below.



B. Singh et al. synthesized KCC-1 based hybrid materials functionalized by different amine molecules using physisorption and covalent attachment. Synthesized materials were tested for their CO₂ capture capacity. Among the studied materials, the tetraethylene pentamine (TEPA) based KCC-1 (KCC-1-TEPAads) showed superior capacity in comparison with MCM-41 in terms of textural stability, CO₂ capture capacity, rate of adsorption as well as thermal stability. KCC-1-TEPAads demonstrates the adsorption capacity of 91.5 mmol g-1, whereas MCM-41-TEPAads CO₂ capture capacity was observed to be only 73.1 mmol g-1. These results clearly indicate the superior stability of the KCC-1-based sorbents as compared to the conventional mesoporous materials^[5]. Graphical representation of the above mentioned process is shown below.





Zafar Ali et al. also studied the immobilization of lipase on KCC-1. Author's method of synthesis, activation, and optimization of immobilization conditions, produced better resistance to temperature and pH inactivation as compared to the free lipase. Therefore, it widened the reaction pH and temperature regions, with the optimum pH and temperature of 7.5 and 40°C, respectively. The immobilized Lipase i.e Candida Ragusa (ICRL) showed above 81% of the initial activity even after 28 days. Also 80% activity after 8 repeated cycles. Authors also demonstrated the ICRL improved storage stability, reusability as well as 700 U/g of protein as immobilization efficiency^[29]. Graphical representation of Immobilization of enzyme on KCC-1 is shown below.



Xiaoxi Huang et al. performed DNA adsorption studies on a series of KCC-1- and MCM-41-based nanomaterials that are functionalized with organoamine groups. Authors observed higher adsorption capacities for KCC-1 based materials as compared to MCM-41 based nanomaterials. It was explained that DNA adsorption reaches saturation once the outer surfaces of the MCM-41 nanoparticles are occupied by the DNA molecules, leaving their inner channel spaces to play negligible role in the adsorption of DNA molecules. However, due to the unique fibrous morphology, KCC-1-based nanomaterials absorb DNAs efficiently both on their external and internal surfaces. Cellular toxicity tests performed in this study demonstrates the biocompatibility of KCC-1^[12]. Graphical representation of DNA adsorbed KCC-1 is shown below.



XIII) Photo catalysis

Bayal et al. also reported a new method to fabricate TiO₂ on the fibers of KCC-1. The synthesized KCC-1/TiO₂ photocatalyst showed a very high yield of H₂ (26.4 mmolh-1g-1 TiO₂) in presence of UV light. The yield reported in this research study was one of the best reported so far for photocatalytic H₂ generation. In addition, authors do not observe decrease in photocatalytic activity of KCC-1/TiO₂, unlike conventional mesoporous silica materials and this effect was attributed to open morphology of KCC-1^[30].



R. Singh et al. designed and synthesized KCC-1 based photo catalysts by coating TiO_2 on fibers of KCC-1 using atomic layer deposition (ALD). Catalysts developed by authors showed superior catalytic activity for photocatalytic dye degradation in comparison to MCM-41 and SBA- 15 based TiO_2 catalysts synthesized with aid of ALD as well as literature reported silica-supported TiO_2 catalysts. Authors have described the advantages of KCC-1/TiO_2 catalysts over MCM-41 and SBA-15 based TiO_2 . Some of these include uniform and conformal coating, less reduction in surface, enhanced light harvesting properties and better accessibility of active sites^[13].

$$H_{2}O \xrightarrow[KCC-1 catalyst]{} H_{2} + 1/2 O_{2}$$

XIV) Optical sensing properties

KCC-1 has been also explored as sensor with optical sensing properties and is used for detection of Hg^{2+} detection with very good selectivity, and sensitivity. The sensor was developed by immobilizing the rhodamine-based receptor (RB-Si) within the fibers of KCC-1. Authors observed that spirolactam ring opening of the rhodamine groups are the main reason for the fluorescence enhancement responses of RB-KCC-1 and obtained a detection limit of 9.05×10^{-7} . Based on the results obtained authors conclude that the synthesized materials are potential candidates for the detection and effective removal of Hg^{2+} in biological, environmental, and industrial fields^[31]. The below figure indicates the Graphical representation of RB-Si Immobilized KCC-1 acted as a sensor for the detection of Hg^{+2} ions.



XV) Effect of pore structure of silica on CO oxidation activity

J. Xu et al. studied the CO oxidation reaction using different Ag supported on mesoporous supports such as MCM-41, SBA-15 and KCC-1. Authors observed that the morphology of the supports influence the dispersion of Ag nanoparticles, consequently shows difference in catalytic activity. It was mentioned that when KCC-1 was used, the migration and agglomeration of Ag particles can be effectively hindered by the open access mesopores. Authors therefore conclude that in comparison with SBA-15 and MCM-41, KCC-1 shows very high catalytic activity^[8].

$$2CO + O_2 \xrightarrow{Ag/KCC-1 \\ Catalyst} 2CO_2$$

XVI) Oxidation of organosilanes to silanols

M. Dhiman et al. reported ultrasmall and pseudo single atoms of gold supported on KCC-1 for the oxidation of organosilanes to silanols with high Turnover Number (TON), which is nearly half a million (591,000) for dimethylphenyl silane as a model substrate). The Extraordinary catalytic activity and stability in this study was attributed to dispersion of ultrasmall nanoparticles and pseudo-single atoms of Au as well as Au δ + species on fibers of KCC-1. This oxidation was carried out only at 45°C using water as an oxidant and hydrogen was observed to be the by-product in this study. Authors also mention that, it was possible to easily separate Au based KCC-1 catalyst and they are stable for several cycles. These results clearly indicates the reusability of catalyst^[32].



XVII) CO, methanation

Hamid et al. studied the catalytic activity of KCC-1 for CO₂ methanation and found that KCC-1 is a probable candidate for the examined reaction as compared to MCM-41. The high activity of KCC-1 was attributed to the presence of abundant oxygen vacancy facilitating the CO₂ adsorption/dissociation. With the help of IR studies, it was also explained that KCC-1 has significantly higher basicity in addition to oxygen vacancy than that of MCM-41. These two properties of the catalysts were explained as reasons for improved catalytic performance of the catalyst. KCC-1 in this research work show good stability throughout 90 h of reaction due to the resistance towards Ostwald ripening^[7]. Graphical representation of CO₂ methanation using KCC-1 is presented below.



XVIII) n-Heptane isomerization

n-Heptane -

Fatah et al. synthesized Molybdenum oxide (MoO₃) supported KCC-1 and phosphorus (P) loaded on MoO₃/KCC-1 (P/MoO₃/KCC-1) and studied its activity for n-Heptane isomerization. Synthesis was carried out by mixing the MoO₃ and KCC-1 powder in a ball mill equipment. For doing this, an agate container having 300 cm³ volume and six-piece of 20 mm diameter agate ball (weight of approximately 11 g each) were used. Authors found a slight change in the morphology after the addition of MoO₃ and modification with H₃PO₄. It was also observed MoO₃/KCC-1 shows excellent catalytic activity for n-heptane isomerization. This was attributed to the high amount and high accessibility of the active sites as well as the high dispersion of MoO₃^[33]. Graphical representation of the process is presented below.



Isomerized products of n-Heptane

XIX) Benzene hydroxylation and 1-hexene epoxidation reactions

Yang et al. prepared nanosized monodispersed amphiphilic titanosilicate composite (TS-1@KCC-1) materials, which exhibited superior thermal stability as well as catalytic activity. These surface active titanosilicates were evaluated for their activity in benzene hydroxylation and 1-hexene epoxidation reactions. They observed that Rh(OH)₃ species supported TS-1@KCC-1 exhibited higher catalytic activity in benzene hydroxylation with hydrogen peroxide under Pickering Interface Catalysis (PIC) in comparison with phase-boundary catalysis (PBC). The PIC reaction system in this study demonstrates excellent chemical stability and reasonable reusability.^[34]



XX) Synthesis of cycliccarbonate from carbon dioxide and epoxides

S.M. Sadeghzadeh also studied the KCC-1 supported heteropolyacid-based ionic liquid (KCC-1/IL/ HPW) for the synthesis of cycliccarbonate from carbon dioxide and epoxides under mild conditions. Authors observed increase in catalytic efficiency of IL/HPW after immobilization onto KCC-1 NPs and KCC-1/IL/HPW NPs are found to be recovered from the reaction mixture during the work-up procedure. Also the recycled catalyst was reused for ten consecutive cycles and observed to be active without any significant loss in catalytic activity. The developed method of synthesis in this study indicate a green and low-cost protocol for making heteropolyacid-based catalysts and is also promising for developing similar type of materials^[35].



XXI) Synthesis of tetrahydrodipyrazolo pyridines

Using the developed heteropolyacid-based ionic liquid (KCC-1/IL/HPW) catalyst S.M. Sadeghzadeh also investigated the synthesis of tetrahydrodipyrazolo pyridines. Catalyst was found to be active for the studied reaction and also it was possible to recover and reuse it several times without any significant decrease in activity and selectivity. The high activity of the catalyst was attributed to high accessibility of active sites available with KCC-1 based catalyst^[36].



XXII) Synthesis of high surface area Silicon carbides (SIC)

Eunjin Jung et al. made an attempt to make high-surface area SiC spheres by the nano casting technology using a KCC-1 as a template. Polycarbosilane and poly(phenyl carbosilane) were used as the precursor for silicon carbide and it was possible to obtain hollow spheres by the inversion of a template structure. In this work, poly (phenyl carbosilane) worked as a binding agent due to its higher molecular weight. Fibrous strcture of KCC-1 was obtained for hollow spheres of silicon carbides. The resulting fibrous hollow spheres possessed SiC nano crystals and also shows 407 m²/g BET surface area^[37]. Graphical representation of the process that describes the synthesis of high surface area Silicon carbides (SIC) is shown below.



XXIII) Synthesis of β-enaminones

Z. N. Siddigui et al. developed an efficient protocol for the preparation of β-enaminones using KCC-1 sulphuric acid catalyst. Synthesis of KCC-1 sulfuric acid (KCC-1-SA) was carried out by dispersing KCC-1 (1.0 g) in CH₂Cl₂ (10 mL) in a flask. This was achieved by first dissolving chlorosulfonic acid (8 mmol) in CH₂Cl₂ (10 mL) and the resulting solution was added to the KCC-1 suspension through a constant-pressure dropping funnel under stirring for around 30 min at room temperature. After this addition, the resulting mixture was stirred for another 30 min at room temperature. The obtained brown solid was collected by filtration and it was washed with ether (50 mL). Procedure was concluded with a drying step at room temperature. Authors carried out the reactions under thermal solvent-free conditions and after a few minutes they obtained excellent yields. It was concluded that the amount of catalyst used is very low and the followed methodology that is environmentally benign in comparison with H₂SO₄, in terms of the amount, hazardous nature and reaction conditions. It was possible to recover the catalyst and is recyclable up to seven cycles without much loss in activity^[38].



R - H, p-OMe, m-NO₂, p-Cl, p-OH, p-CH₃



R - H, p-OMe, m-NO₂, m-NO₂, p-Cl, p-OH, p-CH₃

XXIV) Hydrogenolysis of Alkanes

Fihri et al. studied Ru nanoparticle supported on KCC-1 for the hydrogenolysis of propane and ethane. It was observed from the TEM images that Ru nanoparticles are fully loaded on fibers of KCC-1. Authors proved that KCC-1 showed better catalytic performance as compared to conventional mesoporous silica materials such as MCM-41 and SBA-15. This examination was carried out by loading comparable Ru content on aforementioned supports and testing under identical conditions. It was also explained that the superior catalytic activity of KCC-1 based catalyst is attributed to high accessibility of Ru supported on KCC-1. In addition, high activity of Ru/KCC-1 catalyst was explained through HR-TEM studies. Authors mention that in case of Ru/KCC-1 hexagonal shaped nanoparticles were present having several corners and sharp edges and these possess reactive atoms with lowest coordination numbers. Catalyst based on KCC-1 was observed to be stable with excellent lifetime. Deactivation was not observed even after 8 days^[3].





M. L. Firmansyah et al. prepared ZSM-5 zeoilte with KCC-1 by combining a microemulsion technique with zeolite seed-assisted crystallization. The resulting material (FZSM-5) possessed very good physicochemical properties such as high surface area, fibrous morphology and strong acid sites. Authors explained that the cumene hydrocracking activity of the material could be attributed to the large number of active sites as well as their high accessibility. High dispersion of the metal particle was also observed, which was mainly due to the large surface area of the FZSM-5 type catalyst. The catalytic activity for cumene hydrocracking and EB dehydrogenation was enhanced by the protonation and introduction of Pt into the FZSM-5 type catalyst^[39].



XXVI) Dehydrogenation of aqueous solutions of HCOOH/HCOONa

S.M. Sadeghzadeh synthesized a novel PbS (lead sulphide) containing Ionic liquid IL-based KCC-1 catalysts (KCC-1/IL/PbS) and studied their catalytic applications in dehydrogenation of aqueous solutions of HCOOH/HCOONa to H₂ and CO₂ gas. It was found that the activity was due to dendritic fibrous morphology of the KCC-1 as well as the synergistic effect between KCC-1/IL and the small PbS NPs. Catalysts investigated in this study could be recovered and reused at least ten times with no decrease in activity^[40].

Formic acid

Carbon dioxide Hydrogen

XXVII) Spidery catalyst for the synthesis of quinazoline-2,4(1H,3H)-diones

S.M. Sadeghzadeh synthesized multi-carboxylic hyperbranched polyglycerol groups (HPG) functionalized KCC-1 and nanocatalysts based on noble metal (Au, Pd, and Cu) were developed with the functionalized KCC-1. These resulting catalysts observed to be having ultrasmall size and monodisperse particles with uniform distribution, and high loading capacity. With the help of synthesized catalysts, an economically viable and green protocol was developed for reaction of CO₂ with 2-aminobenzonitrile to obtain quinazoline-2,4(1H, 3H)-diones. Authors examined the impact of HPG by synthesizing catalyst with HPG functionalization and without HPG functionalization i.e KCC-1/HPG/X (X = Au, Pd, Cu) and KCC-1/X (X = Au, Pd, Cu). It was observed that the amount of motionless nanoparticles in KCC-1/HPG/X were about twice that in KCC-1/X. From the results, the effectiveness of HPG functionalization on KCC-1 was demonstrated^[41].



XXVIII) Production of 5 hydroxymethylfurfural (HMF)

N. Chermahini et al. functionalized KCC-1 with propylsulfonic acid groups (Pr-SO₃H) and investigated resulting catalysts efficiency for the production of 5 hydroxymethylfurfural (HMF) by the dehydration of fructose. Authors achieved 67.71% yield and 68.32% selectivity for the HMF at a 99.11% fructose conversion. Although the acid loading on the catalyst surface was low, the results observed in this study are better as compared to the reported studies. It was explained that the catalyst was reusable and can be synthesized by an environmentally friendly process for the heterogeneous catalytic manufacture of 5-HMF^[42].



XXIX) Ozonation of oxalic acid

S. Afzala et al. investigated the Manganese incorporated fibrous silica nanosphere (MnOx-0.013/ KCC-1) for ozonation of oxalic acid. Synthesized catalysts show excellent activity and stability with minimum Mn leaching in comparison to MnOx loaded on MCM-41. The high catalytic activity of studied catalysts was attributed to the generation of hydroxyl radical. Authors mentioned that surfacehydroxyl groups observed with the help of phosphates and ATR-FTIR were found to be active sites^[43]. Graphical representation of Mn loaded KCC-1 used for ozonation of oxalic acid is shown below.



S.M. Sadeghzadeh studied gold (III) phosphorus complex containing HPG-based KCC-1 and tested its catalytic activity for the cyclization of propargylic amines with CO₂ to provide 2-oxazolidinones. It was observed that hot filtration tests and selective catalyst poison examinations showed the presence of soluble Au species during the reaction process. However, recovery studies demonstrated that no significant decrease has occurred in the activity and metal content of recovered KCC-1/IL/Au. On the other hand catalysts were able to be recovered and reused at least ten times with no loss in activity and selectivity. Through the obtained results it was concluded that superior effectiveness of HPG@ KCC-1/PPh₂/Au nanocatalyst was attributed to isolated HPG units incorporated in the fibers of KCC-1, which could control the mechanistic aspects through preventing the formation of agglomerated gold (III) phosphorus complex and stabilization of active catalytic gold species^[44].



2-oxazolidinone

XXX) One-pot synthesis of benzamide

Hong-gen Peng et al. Synthesized a core–shell material consisting of KCC-1 and encapsulated TS-1 zeolite (Si/Ti=40) in a microemulsion system. Rh(OH)₃ species were supported on KCC-1 and the resulting bifunctional Rh(OH)₃/TS-1@KCC-1 catalyst was used for obtaining primary amides from aldehyde, NH₃ and H₂O₂ through one-pot ammoximation and rearrangement(benzamide from benzaldehyde). Authors explained that Titanosilicates are potential candidates for catalyzing the oxidation of NH₃ with H₂O₂ to yield the hydroxylamine intermediate. The formed intermediate again reacts with aldehydes or ketones via non-catalytic oximation to produce oximes. In this study, the Rh(OH)₃ species are considered to be the catalytic sites for the rearrangement of aldehyde oximes to amides. The synthesized TS-1@KCC-1 catalyst was also found to have superior hydrothermal and mechanical stability and is a robust support for dispersing and stabilizing the Rh(OH)₃ species^[45].



Catalog Products of KCC-1

Different grades of KCC-1 products have been included in the catalogue and these products based on their varied properties can have different applications in catalysis, paints& coatings, cosmetics and drug delivery. Table 1 indicates the particle size, specific surface area and pore volume of different grades of KCC-1. Figure (1-6) show the Transition Electron Microscopy (TEM) images of KCC-1 product portfolio.

Category	Grade	Particle size (nm)	Spec.Surface Area (m²/g)	Pore Volume (cm³/g)
	KCC-1 L1	~900-1000	~ 700	~1.40
Large	KCC-1 L2	~900-1000	~ 600	~1.20
	KCC-1 L3	~900-1000	~ 550	~0.80
Modium	KCC-1 M1	~400-450	~ 400	~0.50
Medium	KCC-1 M2	~300-350	~ 600	~0.60
Small	KCC-1 S1	~ 130-190	~380	~ 0.80

Table.2: Summary of properties of KCC-1 product portfolio



Figure.1: Transition Electron Microscopy images of KCC-1 (L1 grade)



Figure.2: Transition Electron Microscopy images of KCC-1 (L2 grade)



Figure 3: Transition Electron Microscopy images of KCC-1 (L3 grade)



Figure 4: Transition Electron Microscopy images of KCC-1 (M1 grade)



Figure 5: Transition Electron Microscopy images of KCC-1 (M2 grade)



Figure 6: Transition Electron Microscopy images of KCC-1 (S1 grade)

References:

- 1. High-Surface-Area Silica Nanospheres (KCC-1) with a Fibrous Morphology. Vivek Polshettiwar, Dongkyu Cha, Xixiang Zhang, Jean-Marie Basset. Angewandte Chemie International Edition 2010 (49) 9652 –9656
- Hydro-metathesis" of Olefins: A Catalytic Reaction Using a Bifunctional Single-Site Tantalum Hydride Catalyst Supported on Fibrous Silica (KCC-1) Nanospheres. Vivek Polshettiwar, Jean Thivolle-Cazat, Mostafa Taoufik, Francois Stoffelbach, Sebastien Norsic, and Jean-Marie Basset. Angewandte Chemie International Edition 2011 (50) 2747 –2751
- Fibrous Nano-Silica (KCC-1)-Supported Palladium Catalyst: Suzuki Coupling Reactions Under Sustainable Conditions. Aziz Fihri, Dongkyu Cha, Mohamed Bouhrara, Noor Almana, and Vivek Polshettiwar. ChemSusChem 2012 (5) 85–89
- Ouyang, M.; Wang, Y.; Zhang, J.; Zhao, Y.; Wang, S.; Ma, X. Three Dimensional Ag/KCC-1 Catalyst with A Hierarchical Fibrous Framework for The Hydrogenation of Dimethyl Oxalate. RSC Adv. 2016, 6, 12788-12791.
- Design of CO2 sorbents using functionalized fibrous nanosilica (KCC-1): insights into the effect of the silica morphology (KCC-1 vs. MCM-41), Baljeet Singh and Vivek Polshettiwar, Journal of Materials Chemistry A, 2016, 4, 7005–7019-7005
- Nitridated Fibrous Silica (KCC-1) as a Sustainable Solid Base Nanocatalyst. Mohamed Bouhrara, Chanakya Ranga, Aziz Fihri, Rafik R. Shaikh, Pradip Sarawade, Abdul-Hamid Emwas, Mohamed N. Hedhili, and Vivek Polshettiwar. ACS Sustainable Chemical Engineering. 2013 (1)1192–1199
- Oxygen vacancy-rich mesoporous silica KCC-1 for CO2 methanation. M.Y.S. Hamid, M.L. Firmansyah, S. Triwahyono, A.A. Jalil , R.R. Mukti, E. Febriyanti, V. Suendod, H.D. Setiabudi, M. Mohamed, W. Nabgan. Applied Catalysis A: General 532 (2017) 86–94
- Ag supported on meso-structured SiO2 with different morphologies for CO oxidation: On the inherent factors influencing the activity of Ag catalysts. Juan Xu, Jingyan Zhang, Honggen Peng, Xianglan Xu, Wenming Liu, Zheng Wang, Ning Zhang, Xiang Wang, Microporous and Mesoporous Materials 2017 (242) 90-98
- Palladium Nanoparticles Supported on Fibrous-Structured Silica Nanospheres (KCC-1): An Efficient and Selective Catalyst for the Transfer Hydrogenation of Alkenes, Ziyauddin S. Qureshi, Pradip B. Sarawade, Matthias Albert, Valerio D'Elia, Mohamed N. Hedhili, Klaus Kçhler, and Jean-Marie Basset, ChemCatChem 2015, 7, 635 – 642
- A multicomponent reaction on a 'free' KCC-1 catalyst at room temperature under solvent free conditions by visible light. Seyed Mohsen Sadeghzadeh. RSC Adv., 2016, 6, 54236
- 11. Synthesis of manganese incorporated hierarchical mesoporous silicananosphere with fibrous morphology by facile one-pot approach for efficient catalytic ozonation. Shahzad Afzal, Xie Quan, Shuo Chen, Jing Wang, Dost Muhammad. Journal of Hazardous Materials 318 (2016) 308–318

References (continued):

- 12. Dendritic Silica Nanomaterials (KCC-1) with Fibrous Pore Structure Possess High DNA Adsorption Capacity and Effectively Deliver Genes In Vitro. Xiaoxi Huang, Zhimin Tao, John C. Praskavich, Jr. Anandarup Goswami, Jafar F. Al-Sharab, Tamara Minko, Vivek Polshettiwar, Tewodros Asefa. Langmuir 2014, 30, 10886–10898
- Atomic Layer Deposited (ALD) TiO2 on Fibrous Nano-Silica (KCC-1) for Photocatalysis: Nanoparticle Formation and Size Quantization Effect, Rustam Singh, Rudheer Bapat, Lijun Qin, Hao Feng, Vivek Polshettiwar, ACS Catalysis. 2016, 6, 2770–2784
- Gold Nanoparticles Supported on Fibrous Silica Nanospheres (KCC-1) as Efficient Heterogeneous Catalysts for CO Oxidation, Ziyauddin S. Qureshi et al., ChemCatChem 2016, 8, 1671 – 1678
- Hydrogen generation. J. M. Basset, Vivek Polshettiwar, M. Bouhrara, Y. Saih, US Patent application 2012/ 6158968
- Dendritic Tip-on Polytriazine-Based Carbon Nitride Photocatalyst with High Hydrogen Evolution Activity. Manas K. Bhunia, Sigismund Melissen, Manas R. Parida, Pradip Sarawade,[⊥] Jean-Marie Basset, Dalaver H. Anjum, Omar F. Mohammed, Philippe Sautet, Tangui Le Bahers, and Kazuhiro Takanabe. Chemistry of Materials 2015 (27) 8237–8247
- Ultrasmall nanoparticles and pseudo-single atoms of platinum supported on fibrous nanosilica (KCC-1/Pt): engineering selectivity of hydrogenation reactions. Mahak Dhiman and Vivek Polshettiwar, Journal of Materials Chemistry A, 2016 (4)12416
- Fibrous nano-silica supported palladium nanoparticles: An efficient catalyst for the reduction of 4-nitrophenol and hydrodechlorination of 4-chlorophenol under mild conditions. Xuanduong Le, Zhengping Dong, Xinlin Li, Wei Zhang, Minhdong Le, Jiantai Ma, Catalysis Communications 59 (2015) 21–25
- Palladium nanoparticles immobilized on core-shell magnetic fibers as a highly efficient and recyclable heterogeneous catalyst for the reduction of 4-nitrophenol and Suzuki coupling reactions. Xuanduong Le,a Zhengping Dong, Yansheng Liu, Zhicheng Jin,aThanh-Do Huy, Minhdong Le and Jiantai Ma. Journal of Material Chemistry A 2014(2) 19696–19706
- Efficient Synthesis of Monodisperse Metal (Rh, Ru, Pd) Nanoparticles Supported on Fibrous Nanosilica (KCC-1) for Catalysis. Mahak Dhiman, Bhagyashree Chalke, and Vivek Polshettiwar. ACS Sustainable Chemical Engineering. 2015 (3)3224–3230
- Fibrous nano-silica containing immobilized Ni@Au core-shellnanoparticles: A highly active and reusable catalyst for thereduction of 4-nitrophenol and 2-nitroaniline. Xuanduong Le, Zhengping Dong, Wei Zhang, Xinlin Li, Jiantai Ma. Journal of Molecular Catalysis A: Chemical 395 (2014) 58–65
- Fibrous Nano-Silica (KCC-1)-Supported Palladium Catalyst: Suzuki Coupling Reactions Under Sustainable Conditions. Aziz Fihri, Dongkyu Cha, Mohamed Bouhrara, Noor Almana, and Vivek Polshettiwar. ChemSusChem 2012 (5) 85–89
- KCC-1 supported palladium nanoparticles as an efficient and sustainable nano catalyst for carbonylative Suzuki–Miyaura cross-coupling, Prashant Gautam, Mahak Dhiman, Vivek Polshettiwar and Bhalchandra M. Bhanage, Green Chem., 2016 (18)5890
- Ionic liquid immobilized onto fibrous nano-silica: A highly active and reusable catalyst for the synthesis of quinazoline-2,4(1 H,3 H)-diones, Seyed Mohsen Sadeghzadeh, Catalysis Communications 72 (2015) 91–96
- Bis(4-pyridylamino)triazine-stabilized magnetite KCC-1: a chemoselective, efficient, green and reusable nanocatalyst for the synthesis of N-substituted 1,4-dihydropyridines. Seyed Mohsen Sadeghzadeh. RSC Advances 2016 (6) 99586
- Palladium Nanoparticles Supported on Fibrous Silica (KCC-1-PEI/Pd): A Sustainable Nanocatalyst for Decarbonylation Reactions. Pintu K. Kundu, Mahak Dhiman, Atanu Modak, Arindam Chowdhury, Vivek Polshettiwar, and Debabrata Maiti. ChemPlusChem 2016(81)1142 – 1146
- Pd/APTPOSS@KCC-1 as a new and efficient support catalyst for C–H activation. Seyed Mohsen Sadeghzadeh, Rahele Zhiani and Shokufe Emrani. RSC Adv., 2017, 7, 24885
- 28. Silicon oxynitrides of KCC-1, SBA-15 and MCM-41 for CO2 capture with excellent stability and regenerability. Umesh Patil, Aziz Fihri, Abdul-Hamid Emwas and Vivek Polshettiwar. Chemical Science 2012 (3) 2224
- Immobilization of lipase on mesoporous silica nanoparticles with hierarchical fibrous pore. Zafar Ali, Lei Tian, Panpan Zhao, Baoliang Zhang, Nisar Ali, Muhammad Khan, Qiuyu Zhang. Journal of Molecular Catalysis B: Enzymatic 134 (2016) 129–135
- Nanostructured Silica-Titania Hybrid using Fibrous Nanosilica as Photocatalysts. Nisha Bayal, Rustam Singh, and Vivek Polshettiwar. DOI: 10.1002/cssc.201700135
- Multifunctional fibrous silica composite with high optical sensing performance and effective removal ability toward Hg2+ ions. Zebin Sun, Dan Guo, Li Zhang, Haizhen Li, Bo Yang and Shiqiang Yan, Journal of Materials Chemistry B, 2015 (3)3201-3210
- Organosilane Oxidation with a Half Million Turnover Number using Fibrous Nanosilica Supported Ultrasmall Nanoparticles and Pseudo-Single Atoms of Gold. Mahak Dhiman, Bhagyashree Chalke, and Vivek Polshettiwar. J. Mater. Chem. A, 2017, DOI: 10.1039/C6TA09434A
- n-Heptane isomerization over molybdenum supported on bicontinuous concentric lamellar silica KCC1: Influence of phosphorus and optimization using response surface methodology (RSM). N.A.A. Fataha S. Triwahyonob, A.A. Jalila , N. Salamun, C.R. Mamat, Z.A. Majid. http://dx.doi.org/10.1016/j.cej.2016.12.028
- Amphiphilic Titano silicates as Pickering Interfacial Catalysts for Liquid-Phase Oxidation Reactions. Yulin Yang, Wen-Juan Zhou, Armin Liebens, Jean-Marc Clacens, Marc Pera-Titus, and Peng Wu. Journal of physical chemistry 2015 (119) 25377–25384

References (continued):

- 35. A heteropolyacid-based ionic liquid immobilized onto fibrous nano-silica as an efficient catalyst for the synthesis of cyclic carbonate from carbon dioxide and epoxides, Seyed Mohsen Sadeghzadeh, Green Chemistry, 2015(17) 3059–3066
- 36. A heteropolyacid-based ionic liquid immobilized onto magnetic fibrous nano-silica as robust and recyclable heterogeneous catalysts for the synthesis of tetrahydrodipyrazolopyridines in water. Seyed Mohsen Sadeghzadeh, RSC Advances 2016 (6)75973
- The Synthesis of High-Surface Area Silicon Carbide by Conversion Method Using Carbosilane Polymer. Eunjin Jung, Yoon Joo Lee, Woo Teck Kwon, Younghee Kim, Dong Geun Shin, and Soo Ryong Kim. Defect and Diffusion Forum 353 (2014) 244-247
- Nano Fibrous Silica Sulphuric Acid as an Efficient Catalyst for the Synthesis of β-Enaminone. Zeba N. Siddiqui, Kulsum Khan, Nayeem Ahmed. Catalysis Letters 2014 (144)623–632
- Synthesis and characterization of fibrous silica ZSM-5 for cumene hydrocracking, M. L. Firmansyah, A. A. Jalil, S. Triwahyono, H. Hamdan, M. M. Salleh, W. F. W. Ahmadd, G. T. M. Kadj, Catalysis Science & Technology, 2016, 6, 5178–5182
- 40. PbS based ionic liquid immobilized onto fibrous nano-silica as robust and recyclable heterogeneous catalysts for the hydrogen production by dehydrogenation of formic acid, Microporous and Mesoporous Materials 234 (2016) 310-316
- Spidery catalyst for the synthesis of quinazoline- 2,4IJ1H,3H)-diones, Seyed Mohsen Sadeghzadeh, Catalysis Science & Technology, 2016, 6, 1435–1441-1435
- 42. Production of 5-hydroxymethylfurfural from fructose using a spherically fibrous KCC-1 silica Catalyst, Alireza Najafi Chermahini, Fereshte Shahangi, Hossein A. Dabbagh Mohammad Saraji, RSC Advances, 2016, 6, 33804
- 43. Synthesis of manganese incorporated hierarchical mesoporous silicananosphere with fibrous morphology by facile one-pot approach forefficient catalytic ozonation. Shahzad Afzal, Xie Quana, Shuo Chen, Jing Wang, Dost Muhammad, Journal of Hazardous Materials 318 (2016) 308–318
- 44. A green approach for the synthesis of 2- oxazolidinones using gold(I) complex immobilized on KCC-1 as nanocatalyst at room temperature. Seyed Mohsen Sadeghzadeh. Applied Organo metalic Chemistry 2016(30) 835–842
- 45. One-pot synthesis of benzamide over a robust tandem catalyst based on center radially fibrous silica encapsulated TS-1. Hong-gen Peng, Le Xu, Haihong Wu, Kun Zhang and Peng Wu. Chemical Communications., 2013 (49) 2709-2711

SILICON (Compounds)

14-6100 NEW

High Surface area Silica nanoparticles, large, particle size ~900-1000 nm, surface area ~700 m²/g, (KCC-1 L1) (*112945-52-5*) SiO₂; FW: 60.09; white pwdr.; SA: ~700 m²/g; P.Vol. ~1.4 cm³/g Note: Diameter: ~900-1000nm; This product is under license of patented Technology from King Abdullah University of Science and Technology – KAUST. Patent PCT/IB2010/002421.

Technical Note:

1. Novel fibrous shaped silica nanospheres, denoted as KCC-1 (KAUST Catalysis Center)^[1], have unique physical properties which have never before been reported in silica materials. These nanomaterials have been developed by Prof. J. M. Basset of King Abdullah University of Science and Technology (KAUST). A fibrous surface morphology arranged in three-dimensional structure forms the spheres (Fig. 1). Unlike traditional pore-based silica, these nanospheres possess a fibrous structure that increases accessibility to the available surface area; this in turn, significantly increases the catalytic activity.

These materials exhibit excellent physical properties, including a high surface area, a fibrous surface morphology, good thermal and hydrothermal stabilities and high mechanical stability (Table 1). The fibrous morphology of KCC-1 remains unaffected even after mechanical compression up to 216 MPa pressure. This is superior to the conventional MCM-41 type of silica, which is affected at pressure 86 MPa.^[1]

A range of heterogeneous catalysts, prepared using KCC-1 as a supporting material, have been showing excellent catalytic activity for various transformations of research and industrial importance. As a catalyst support, sorbent or carrier, KCC-1 is able to demonstrate superior activity as compared to regular mesoporous silica materials in energy related processes^[2,3], a variety of organic reactions^[4-7], biomedical applications and drug delivery systems^[8], optoelectronic devices^[9] and many others.





Product #	Category	Grade	Particle Size (nm)	Surface Area (m²/g)	Pore Volume (cm ³ /g)
14-6100	Large	(KCC-1 L1)	~900-1000	~700	~1.4
14-6110	Large	(KCC-1 L2)	~900-1000	~600	~1.2
14-6120	Large	(KCC-1 L3)	~900-1000	~550	~0.9
14-6200	Medium	(KCC-1 M1)	~400-450	~400	~0.7
14-6210	Medium	(KCC-1 M2)	~300-350	~600	~0.6
14-6300	Small	(KCC-1 S1)	~130-190	~380	~0.8

References:

- 1. Angew. Chem. Int. Ed. 2010, 49, 9652
- 2. Chem. Mater. 2015, 27, 8237
- 3. ACS Catalysis. 2016, 6, 2770
- 4. ChemSusChem **2012**, 5, 85
- 5. Green Chem., 2016, 18, 5890
- 6. Angew. Chem. Int. Ed. 2011, 50, 2747
- 7. RSC Adv., 2017, 7, 24885
- 8. Langmuir 2014, 30, 10886
- 9. J. Mater. Chem. B, 2015, 3, 3201

1g

5g

	•	
SILICON	(Compounds)	
14-6110	High Surface area Silica nanoparticles, large, particle size	1a
NEW	~900-1000 nm, surface area ~600 m²/g, (KCC-1 L2) (112945-52-5)	5g
NEW	SiO ₂ ; FW: 60.09; white pwdr.; SA: ~600 m²/g; P.Vol. ~1.20 cm³/g	
	Note: Diameter: ~900-1000nm; This product is under license of patented	
	Technology from King Abdullah University of Science and Technology –	
Technical Note:	KAUST. Paleni PCT/162010/002421.	
1. See 14-	6100 (page 24)	
14-6120	High Surface area Silica nanoparticles, large, particle size	1g
NEW	~900-1000 nm, surface area ~550 m²/g (KCC-1 L3) (112945-52-5)	5g
	SiO ₂ ; FW: 60.09; white pwdr.; SA: ~550 m²/g; P.Vol. ~0.97 cm³/g	
	Technology from King Abdullah University of Science and Technology	
	KAUST. Patent PCT/IB2010/002421.	
Technical Note:		
1. See 14-	6100 (page 24)	
14-6200	High Surface area Silica nanoparticles, medium, particle size	1g
NEW	~400-450 nm, surface area ~400 m²/g, (KCC-1 M1) (112945-52-5)	5g
_	SIO ₂ ; FW: 60.09; White pwdr.; SA: ~400 m²/g; P.Vol. ~0.7 cm³/g	
	Technology from King Abdullah University of Science and Technology	
	KAUST. Patent PCT/IB2010/002421.	
Technical Note:		
1. See 14-	6100 (page 24)	
14-6210	High Surface area Silica nanoparticles, medium, particle size	1g
NEW	~300-350 nm, surface area ~600 m²/g, (KCC-1 M2) (112945-52-5)	5g
_	SIO ₂ ; FW: 60.09; White pwdr.; SA: ~600 m²/g; P.Vol. ~0.6 cm³/g	
	Technology from King Abdullah University of Science and Technology –	
	KAUST. Patent PCT/IB2010/002421.	
Technical Note:		
1. See 14-	6100 (page 24)	
14-6300	High Surface area Silica nanoparticles, small, particle size ~130-190	1g
NEW	nm, surface area ~380 m ² /g, (KCC-1 S1) $(112945-52-5)$	5g
	SIO ₂ , FW. 60.09, While to beige pwol., SA. ~360 m²/g, F.Vol. ~0.6 cm²/g	
	Technology from King Abdullah University of Science and Technology –	
	KAUST. Patent PCT/IB2010/002421.	
Technical Note:		
1. See 14-	6100 (page 24)	
14-6310	High Surface area Silica nanoparticles, small, particle size ~40-50	1g
NEW	nm, surface area ~520 m²/g, (KCC-1 S2) (112945-52-5)	5g
	SiO ₂ ; FW: 60.09; white to beige pwdr.; SA: ~520 m²/g; P.Vol. ~1.3 cm³/g	
	Note: Diameter: ~40-50nm; This product is under license of patented	
	KAUST. Patent PCT/IB2010/002421.	

ALUMINUM (Compounds)

 13-0300
 Aluminum hydroxide isophthalate MOF (CAU-10, Isophthalate:Al=0.9-1.0) (1416330-84-1)

 Al(OH)(C₈H₄O₄)_x, X = 0.9-1.0; white solid; SA: 620-640 m²/g; P.Vol. 0.23-0.27 cm³/g Note: Particle size: 0.4-0.7 micron, Thermal stability: 400°C, Activation temperature: 150°C Sold under license from Inven2 AS for research purposes only. PCT/GB2009/001087.



500mg 2g

Technical Note:

 MOF exhibits water adsorption characteristics which make it a promising adsorbent for application in heat-exchange processes¹

References:

- Water adsorption behaviour of CAU-10-H: a thorough investigation of its structure-property relationships, J. Mater. Chem. A, 2016, 4, 11859.
- Structures, Sorption Characteristics, and Nonlinear Optical Properties of a New Series of Highly Stable Aluminum MOFs., Chem. Mater. 2013, 25, 17–26.

BIOCATALYSTS (Compounds) CalB immo 1090™ - Immobilized enzyme 07-3155 10q 50g white to slightly yellow spherical beads, dry NEW (store cold) 250g Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 1 year; Particle Size: 300-710 micron; CalB immo 1090 is an adsorbed preparation and is suitable for applications in solvent-free systems like oils, as well as organic solvents and it and can be used for (regio- and stereoselective) esterifications and transesterifications. CalB Immo 1090 has many advantages including high activity and the possibility to use in oils, organic solvent and bi-phasic systems. Sold in collaboration with Purolite for research purposes only. 07-3152 CalB immo 5587™ - Immobilized enzyme 10g white to slightly yellow spherical beads, dry 50g NEW (store cold) 250g Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 1 year; Particle Size: 300-710 micron; CalB immo 5587 is an adsorbed preparation and is particularly suitable for applications where cost is an essential parameter, like biodiesel or industrial oil manufacture. CalB Immo 5587 has many advantages including cost-effectiveness in processes like biodiesel manufacture. It is also a highly robust carrier, particularly suitable for column configurations. Sold in collaboration with Purolite for research purposes only. CalB immo 5872[™] - Immobilized enzyme 10g 07-3159 50g white to slightly yellow spherical beads, dry NEW (store cold) 250g Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 1 year; Particle Size: 300-1500 micron;CalB immo 5872 is an adsorbed preparation and is suitable for applications in solvent-free systems like oils, as well as organic solvents and it and can be used for (regio- and stereoselective) esterifications and transesterifications. CalB Immo 5872 has many advantages including cost-effectiveness and the possibility to use in oils, organic solvent and bi-phasic systems. Sold in collaboration with Purolite for research purposes only.

BIOCATAL	YSTS (Compounds)	
07-3142 New	CalB immo 8285 [™] - Immobilized enzyme white to slightly yellow spherical beads, dry (store cold) Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 1 year; Particle Size: 100-710 micron; CalB immo 8285 is covalently immobilized and is suitable for applications in water, organic solvents as well as solvent- free systems and can be used for (regio- and stereoselective) hydrolysis, esterifications and transesterifications. The lipase is immobilized by covalent immobilization onto Purolites highly hydrophobic carrier Purolite ECR8285 (an epoxy/butyl methacrylate co-polymer). Sold in collaboration with Purolite for research purposes only.	10g 50g 250g
07-3148 (NEW)	CalB immo 8806 [™] - Immobilized enzyme white to slightly yellow spherical beads, dry (<i>store cold</i>) Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 1 year; Particle Size: 300-710 micron; CalB immo 8806 is an adsorbed preparation and is suitable for applications in solvent-free systems like oils, as well as organic solvents and it and can be used for (regio- and stereoselective) esterifications and transesterifications. CalB Immo 8806 has many advantages including high activity and the possibility to use in oils, organic solvent and bi-phasic systems. Sold in collaboration with Purolite for research purposes only.	10g 50g 250g
96-4050	CalB immo KIT™ - Immobilized enzyme See page 107	
07-3130 NEW	CalB immo Plus [™] - Immobilized enzyme white to off white spherical beads, dry (store cold) Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 1 year; Particle Size: 300-710 micron; CalB immo Plus is suitable for applications in organic solvents as well as solvent-free systems and can be used for (regio- and stereoselective) esterifications and transesterifications. CalB Immo Plus has many advantages including high activity and high mechanical stability. Sold in collaboration with Purolite for research purposes only.	10g 50g 250g
07-3133 New	CalB immo Plus Food Grade [™] - Immobilized enzyme white to slightly yellow spherical beads, dry (store cold) Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 1 year; Particle Size: 300-710 micron; CalBimmo Plus Food Grade is supplied in food-grade quality and conforms to the General Specifications and Considerations for Enzyme Preparations Used in Food Processing of the Joint FAO/WHO Expert Committee on Food Additives (JECFA). CalB immo Plus is suitable for applications in organic solvents as well as a solvent-free systems and can be used for (regio- and stereoselective) esterifications and transesterifications. CalB Immo Plus has many advantages including high activity and high mechanical stability. Sold in collaboration with Purolite for research purposes only.	10g 50g 250g
06-0925 (NEW)	Enzyme carrier Lifetech [™] ECR1030M White to off white spherical beads (wet); SA: 80 - 120 m²/g (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-710 micron; Pore Diameter: 220-340 Å; Lifetech ECR1030M is a copolymer of divinylbenzene (DVB) and methacrylate with no functional groups. It is used for enzyme immobilization by adsorption (hydrophobic interaction) and it is particularly suitable for lipase immobilization such as CALB. Lifetech ECR1030M main features are high mechanical stability compared to other existing resins, low surface area that grants high enzyme activity at low protein loading. Sold in collaboration with Purolite for research purposes only.	50g 5x50g

BIOCATAL	YSTS (Compounds)	
07-2215	Enzyme carrier Lifetech™ ECR1504	50a
NEW	White to off white spherical beads (wet)	5x50g
	(store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-1200 micron; Lifetech ECR1504 is a copolymer of divinylbenzene (DVB) and styrene functionalized with tertiary amines. It is used for enzyme immobilization by ionic interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the tertiary amines on the polymer. It is particularly suitable for immobilization of enzymes with iP in the range 3 - 5 like many glycosidaseses. Lifetech ECR1504 main features are possibility to regenerate the resin, pH adjustment before immobilization and large particle size for column applications. Sold in collaboration with Purolite for research purpose	
07-2220	Enzyme carrier Lifetech M ECB1508	500
NEW	White to off white spherical beads (wet) (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-1200 micron; Lifetech ECR1508 is copolymer of divinylbenzene (DVB) and styrene functionalized with tertiary amines. It is used for enzyme immobilization by ionic interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the tertiary amines on the polymer. It is particularly suitable for immobilization of enzymes with iP in the range 3 - 5 like many glycosidaseses. Lifetech ECR1508 main features are possibility to regenerate the resin, pH adjustment before immobilization and large particle size for column applications. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
07-2224	Enzyme carrier Lifetech™ ECR1604	50g
NEW	White to off white spherical beads (wet) (<i>store cold</i>) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-1200 micron; Lifetech ECR1604 is a copolymer of divinylbenzene (DVB) and styrene functionalized with quaternary amines. It is used for enzyme immobilization by ionic interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the tertiary amines on the polymer. It is particularly suitable for immobilization of enzymes with iP in the range 3 - 5 like many glycosidaseses. Lifetech ECR1604 main features are possibility to regenerate the resin, pH adjustment before immobilization and large particle size for column applications. Sold in collaboration with Purolite for research purposes only.	5x50g
07-2230	Enzyme carrier Lifetech™ ECR1640	50g
NEW	white spherical beads (wet) (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-1200 micron; Lifetech ECR1640 is a copolymer of divinylbenzene (DVB) and styrene functionalized with quaternary amines. It is used for enzyme immobilization by ionic interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the tertiary amines on the polymer. It is particularly suitable for immobilization of enzymes with iP in the range 3 - 5 like many glycosidaseses. Lifetech ECR1640 main features are possibility to regenerate the resin, pH adjustment before immobilization and large particle size for column applications. Sold in collaboration with Purolite for research purposes only.	5x50g
06-0928	Enzyme carrier Lifetech™ ECR1061M	50g
NEW	White to off white spherical beads (wet); SA: 400 - 510 m²/g (<i>store cold</i>) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-710 micron; Pore Diameter: 600-750 Å; Lifetech ECR1061M is a copolymer of divinylbenzene (DVB) with methacrylate with no functional groups. It is used for enzyme immobilization by adsorption (hydrophobic interaction) and it is particularly suitable for lipase immobilization as CALB. Lifetech ECR1061M main features are higher porosity compared to Lifetech ECR1030M to allow better diffusion for bulky substrates or better application in viscous systems. Sold in collaboration with Purolite for research purposes only.	5x50g

BIOCATA	LYSTS (Compounds)	
06-0905 New	Enzyme carrier Lifetech [™] ECR1090F White to off white spherical beads (wet); SA: 750 - 850 m²/g (<i>store cold</i>) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 150-300 micron; Pore Diameter: 900-1100 Å; Lifetech ECR1090F is a copolymer of divinylbenzene (DVB) and styrene with high porosity and no functional groups. It is used for enzyme immobilization by adsorption (hydrophobic interaction) and it is particularly suitable for lipase immobilization. Lifetech ECR1090F main features are high porosity, high mechanical stability and high surface area. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
06-0913 New	Enzyme carrier Lifetech [™] ECR1090M White to off white spherical beads (wet); SA: 750 - 850 m²/g (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-710 micron; Pore Diameter: 900-1100 Å; Lifetech ECR1090M is a copolymer of divinylbenzene (DVB) and styrene with high porosity and no functional groups. It is used for enzyme immobilization by adsorption (hydrophobic interaction) and it is particularly suitable for lipase immobilization. Lifetech ECR1090M main features are high porosity, high mechanical stability and high surface area. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
06-0922 New	Enzyme carrier Lifetech [™] ECR1091M White to off white spherical beads (wet); SA: > 450 m ² /g (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-710 micron; Pore Diameter: 950-1200 Å; Lifetech ECR1091M is a copolymer of divinylbenzene (DVB) and styrene with very high porosity and no functional groups. It is used for enzyme immobilization by adsorption (hydrophobic interaction) and it is particularly suitable for lipase immobilization. Lifetech ECR1091M main features are high porosity, high mechanical stability and high surface area. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
06-0810 NEW	Enzyme carrier Lifetech [™] ECR8204F White to off white spherical beads (wet) (<i>store cold</i>) Note: Store in dry conditions (2-8°C). Do not freeze. Shelf life: 6 months; Particle Size: 150-300 micron; Pore Diameter: 300-600 Å; Lifetech ECR8204F is a methacrylate polymer functionalized with epoxy groups, used for covalent enzyme immobilization. Epoxides form very stable covalent linkages with different protein surface groups as ε-NH2 in Lys or nucleophiles (amino, thiol, phenolic). Immobilization is performed under very mild experimental conditions of pH and temperature, at high ionic buffer strength. Lifetech ECR8204F low porosity, the hydrophilicity, high mechanical strengh and it is optimal for use in batch reactors. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
06-0813 New	Enzyme carrier Lifetech [™] ECR8204M White to off white spherical beads (wet) (store cold) Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 6 months; Particle size: 300-710 micron; Pore Diameter: 300-600 Å; Lifetech ECR8204M is a methacrylate polymer functionalized with epoxy groups, used for covalent enzyme immobilization. Epoxides form very stable covalent linkages with different protein surface groups as ε-NH ₂ in Lys or nucleophiles (amino, thiol, phenolic). Immobilization is performed under very mild experimental conditions of pH and temperature, at high ionic buffer strength. Lifetech ECR8204M main features are the low porosity, the hydrophilicity, high mechanical strengh and it is optimal for use in batch reactors and columns.	50g 5x50g

BIOCATAL	YSTS (Compounds)	
06-0817 New	Enzyme carrier Lifetech™ ECR8209F white to off white spherical beads (wet) (<i>store cold</i>) Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 6 months; Particle Size: 150-300 micron; Pore Diameter: 300-600 Å; Lifetech ECR8209F is a methacrylate polymer functionalized with epoxy groups, used for covalent enzyme immobilization. Epoxy groups form very stable covalent linkages with different protein surface groups as ε-NH ₂ in Lys or nucleophiles (amino, thiol, phenolic). Immobilization is performed under very mild experimental conditions of pH and temperature, at high ionic buffer strength. Lifetech ECR8209F main features are the high porosity, the hydrophilicity and it is optimal for use in batch reactors.	50g 5x50g
07-1512 New	Enzyme carrier Lifetech [™] ECR8309F white to off white spherical beads (wet); SA: 70 min. m²/g (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 150-300 micron; Pore Diameter: 600-1200 Å; Lifetech ECR8309F is a methacrylate polymer functionalized with amino groups on a short spacer (C2). It is used for covalent enzyme immobilization by pre-activation of the resin with glutaraldehyde and to subsequently form very stable covalent linkages with different protein groups (amino, thiol, phenolic) under very mild experimental conditions of pH and temperature, at low ionic buffer strength. It can also be used for enzyme immobilization by ionic interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the charged amines on the polymer. Lifetech ECR8309F main features are the medium porosity, the hydrophilicity and its optimal use in batch reactors. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
07-1515 (New)	Enzyme carrier Lifetech [™] ECR8309M White to off white spherical beads (wet); SA: 70 min. m²/g (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-710 micron; Pore Diameter: 600-1200 Å; Lifetech ECR8309M is a methacrylate polymer functionalized with amino groups on a short spacer (C2). It is used for covalent enzyme immobilization by pre-activation of the resin with glutaraldehyde and to subsequently form very stable covalent linkages with different protein groups (amino, thiol, phenolic) under very mild experimental conditions of pH and temperature, at low ionic buffer strength. It can also be used for enzyme immobilization by ionic interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the charged amines on the polymer. Lifetech ECR8309M main features are the medium porosity, the hydrophilicity and its optimal use in batch reactors and columns. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
07-1518 New	Enzyme carrier Lifetech [™] ECR8315F white to off white spherical beads (wet); SA: 60 min. m²/g (<i>store cold</i>) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 150-300 micron; Pore Diameter: 1200-1800 Å; Lifetech ECR8315F is a methacrylate polymer functionalized with amino groups on a short spacer (C2). It is used for covalent enzyme immobilization by pre-activation of the resin with glutaraldehyde and to subsequently form very stable covalent linkages with different protein groups (amino, thiol, phenolic) under very mild experimental conditions of pH and temperature, at low ionic buffer strength. Lifetech ECR8315F main features are the high porosity, the hydrophilicity and its optimal use in batch reactors. Sold in collaboration with Purolite for research purposes only.	50g 250g

BIOCATAL	_YSTS (Compounds)	
07-1520 New	Enzyme carrier Lifetech [™] ECR8315M white to off white spherical beads (wet); SA: 60 min. m ² /g (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-710 micron; Pore Diameter: 1200-1800 Å; Lifetech ECR8315M is a methacrylate polymer functionalized with amino groups on a short spacer (C2). It is used for covalent enzyme immobilization by pre-activation of the resin with glutaraldehyde and to subsequently form very stable covalent linkages with different	50g 250g
	protein groups (amino, thiol, phenolic) under very mild experimental conditions of pH and temperature, at low ionic buffer strength. It can also be used for enzyme immobilization by ionic interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the charged amines on the polymer. Lifetech ECR8315M main features are the medium porosity, the hydrophilicity and its optimal use in batch reactors and columns. Sold in collaboration with Purolite for research purposes only.	
07-1523 New	Enzyme carrier Lifetech™ ECR8409F white to off white spherical beads (wet); SA: 70 min. m²/g (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 150-300 micron; Pore Diameter: 600-1200 Å; Lifetech ECR8409F is a methacrylate polymer functionalized with amino groups on a long spacer (C6). It is used for covalent enzyme immobilization by pre-activation of the resin with glutaraldehyde and to subsequently form very stable covalent linkages with different protein groups (amino, thiol, phenolic) under very mild experimental conditions of pH and temperature, at low ionic buffer strength. It can also be used for enzyme immobilization by ionic interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the charged amines on the polymer. Lifetech ECR8409F main features are the medium porosity, the hydrophilicity and its optimal use in batch reactors. Sold in collaboration with Purolite for research purposes only.	50g 250g
07-1525 New	Enzyme carrier Lifetech ™ ECR8409M white to off white spherical beads (wet); SA: 70 min. m ² /g (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-710 micron; Pore Diameter: 600-1200 Å; Lifetech ECR8409M is a methacrylate polymer functionalized with amino groups on a long spacer (C6). It is used for covalent enzyme immobilization by pre-activation of the resin with glutaraldehyde and to subsequently form very stable covalent linkages with different protein groups (amino, thiol, phenolic) under very mild experimental conditions of pH and temperature, at low ionic buffer strength. It can also be used for enzyme immobilization by ionic interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the charged amines on the polymer. Lifetech ECR8409M main features are the medium porosity, the hydrophilicity and its optimal use in batch reactors and columns. Sold in collaboration with Purolite for research purposes only.	50g 250g
07-1528 New	Enzyme carrier Lifetech™ ECR8415F white to off white spherical beads (wet); SA: 60 min. m²/g (<i>store cold</i>) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 150-300 micron; Pore Diameter: 1200-1800 Å; Lifetech ECR8415F is a methacrylate polymer functionalized with amino groups on a long spacer (C6). It is used for covalent enzyme immobilization by pre-activation of the resin with glutaraldehyde and to subsequently form very stable covalent linkages with different protein groups (amino, thiol, phenolic) under very mild experimental conditions of pH and temperature, at low ionic buffer strength. It can also be used for enzyme immobilization by ionic interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the charged amines on the polymer. Lifetech ECR8415F main features are the high porosity, the hydrophilicity and its optimal use in batch reactors. Sold in collaboration with Purolite for research purposes only.	50g 250g

BIOCATALYSTS (Compounds)			
07-1530 NEW	Enzyme carrier Lifetech™ ECR8415M White to off white spherical beads (wet); SA: 60 min. m²/g (store cold)	50g 5x50g	
	Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-710 micron; Pore Diameter: 1200-1800 Å; Lifetech		
	ECR8415M is a methacrylate polymer functionalized with amino groups on a long spacer (C6). It is used for covalent enzyme immobilization by pre-		
	activation of the resin with glutaraldehyde and to subsequently form very		
	stable covalent linkages with different protein groups (amino, thiol, phenolic) under very mild experimental conditions of pH and temperature, at low		
	ionic buffer strength. It can also be used for enzyme immobilization by ionic		
	interaction of the ionizable surface aminoacids (Lys, Arg, His, Asp, Glu) with the charged amines on the polymer. Lifetech ECR8415M main features are		
	the high porosity, the hydrophilicity and its optimal use in batch reactors. Sold		
07 4522	in collaboration with Purolite for research purposes only.	50%	
NEW	White to off white spherical beads (wet); SA: 70 min. m ² /g (store cold)	50g 5x50g	
	Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 150-300 micron; Pore Diameter: 350-600 Å; Lifetech		
	ECR8806F is a methacrylic polymer functionalized with octadecyl groups. It		
	and it is particularly suitable for lipase and transaminases immobilization.		
	Lifetech ECR8806F main features are very enzyme activity achieved upon		
	Purolite for research purposes only.		
07-1535	Enzyme carrier Lifetech™ ECR8806M	50g	
NEW	(store cold)	5x50g	
	Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years;		
	is a methacrylic polymer functionalized with octadecyl groups. It is used		
	for enzyme immobilization by adsorption (hydrophobic interaction) and it is		
	ECR8806M main features are very high enzyme activity achieved upon		
	immobilization compared to other existing resins. Optimal for column packed		
06-0820	Enzyme carrier Lifetech™ ECR8209M	50g	
NEW	white to off white spherical beads (wet); SA: 70 min. m ² /g	5x50g	
	Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 6 months; Particle		
	Size: 300-710 micron; Pore Diameter: 600-1200 Å; Lifetech ECR8209M is a methacrylate polymer functionalized with enoxy groups, used for covalent enzyme		
	immobilization. Epoxy groups form very stable covalent linkages with different		
	protein surface groups as ε-NH2 in Lys or nucleophiles (amino, thiol, phenolic).		
	temperature, at high ionic buffer strength. Lifetech ECR8209M main features are the		
06 0922	high porosity, the hydrophilicity and it is optimal use in batch reactors and columns.	500	
06-0825	white to off white spherical beads (wet); SA: 60 min. m ² /g	5x50g	
NEW	(store cold) Note: Store in dry conditions (2.9°C). Do not freeze. Shelf Life: 6 months: Particle		
	Size: 150-300 micron; Pore Diameter: 1200-1800 Å; Lifetech ECR8215F is a		
	methacrylate polymer functionalized with epoxy groups, used for covalent enzyme		
	protein surface groups as ε -NH ₂ in Lys or nucleophiles (amino, thiol, phenolic).		
	Immobilization is performed under very mild experimental conditions of pH and		
	the very high porosity, the hydrophilicity and it is optimal for use in batch reactors.		

BIOCATAL	YSTS (Compounds)	
06-0826	Enzyme carrier Lifetech™ ECR8215M	50g
NEW	white to off white spherical beads (wet); SA: 60 min. m²/g	5x50g
_	(store cold) Note: Store in dry conditions (2-8°C). Do not freeze, Shelf Life: 6 months: Particle	
	Size: 300-710 micron; Pore Diameter: 1200-1800 Å; Lifetech ECR8215M is a	
	methacrylate polymer functionalized with epoxy groups, used for covalent enzyme immobilization. Epoxy groups form very stable covalent linkages with different protein	
	surface groups as ϵ -NH2 in Lys or nucleophiles (amino, thiol, phenolic). Immobilization	
	is performed under very mild experimental conditions of pH and temperature, at high	
	hydrophilicity and its optimal use in batch reactors and columns.	
06-0828	Enzyme carrier Lifetech™ ECR8285	50g
NEW	white to off white spherical beads (wet); SA: 100-200 m²/g	5x50g
_	(store cold) Note: Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 6 months:	
	Particle Size: 250-1000 micron; Pore Diameter: 400-600 Å; Lifetech ECR8285	
	is a methacrylate polymer functionalized with both butyl and epoxy groups. This	
	optimal for immobilization of hydrophobic enzymes like lipases and transaminases.	
	Epoxides form very stable covalent linkages with different protein groups (amino,	
	thiol, phenolic) under very mild experimental conditions of pH and temperature.	
	hydrophobic property combined with epoxy groups allowing the use in bi-phasic	
	systems. Sold in collaboration with Purolite for research purposes only.	
96-0255	Enzyme carrier Lifetech™ ECRKIT1 See page 107	
96-4065	Lipase immo Kit - Immobilized enzymes	
CALCIUM	(Compounds)	
20-8200	Bis(N,N'-diisopropylformamidinato)calcium(II) dimer, (99.99 %-Ca) PURATREM (1959584-78-1) IPr i Pr	1g 5g
NEW	$C_{28}H_{60}Ca_2N_8$; FW: 588.99; tan to light-brown pwdr.	
	air sensitive, moisture sensitive IPr, Ca, IPr Note: Product sold under use subject to terms and N.	
	conditions of label license at www.strem.com/harvard2. $H \rightarrow H$	
1 Calcium	amidinate precursor for the atomic layer deposition (ALD) of iPr	
calcium c	ontaining thin films. iPr^{N}	
References:	Kim, Chuanyi Yang, Tamara Powers, Luke M. Davis	
Xiabing L	ou, and Roy G. Gordon, Angew. Chem. Int. Ed., 2016 , 55, 10228 –10233.	
	Elemental Forms)	
06-2530	Granhene oxide (4mg/ml water dispersion) - low Mn (1034343-98-0)	100ml
NEW	C; brown liq.	500ml
	Note: Diameter: 5-30 micron flakes;	
06-0330	Graphene Quantum Dots (GQDs), Aqua-Green Luminescent (1034343-98-0) See page 48	
06-0332	Graphene Quantum Dots (GQDs) in water, Aqua-Green Luminescent (1034343-98-0) See page 48	
06-0334	Graphene Quantum Dots (GQDs), Blue Luminescent (1034343-98-0) See page 49	
06-0336	Graphene Quantum Dots (GQDs) in water, Blue Luminescent (1034343-98-0) See page 49	
06-0340	Graphene Quantum Dots (GQDs) in water, Cyan Luminescent (1034343-98-0) See page 49	
CARBON	(Elemental Forms)	
------------------------------	---	-------------------
06-0365 New	High Strength Metallurgical Graphene on GLASS HSMG® (10x10mm) (1034343-98-0) C; FW: 12.011 Note: HSMG® Sold under license for research purposes only. U.S. Patent no. 9,284,640 B2.	1рс
06-0345 New	High Strength Metallurgical Graphene on PMMA HSMG® (10x10mm) (1034343-98-0) C; FW: 12.011 Note: HSMG® Sold under license for research purposes only. U.S. Patent no. 9,284,640 B2.	1рс
06-0355 New	High Strength Metallurgical Graphene on PMMA HSMG® (25x25mm) (1034343-98-0) C; FW: 12.011 Note: HSMG® Sold under license for research purposes only. U.S. Patent no. 9,284,640 B2.	1рс
06-0360 New	High Strength Metallurgical Graphene on PMMA HSMG® (50x50mm) (1034343-98-0) C; FW: 12.011 Note: HSMG® Sold under license for research purposes only. U.S. Patent no. 9,284,640 B2.	1рс
CHROMIU	M (Compounds)	
24-0000 New	Chromium(III) acetate, 97% (1066-30-4) Cr(CH ₃ CO ₂) ₃ ; FW: 229.13; green pwdr.	100g 500g
24-2427 NEW HAZ	Chromium (III) naphthenate, 30-40% in heavy naphtha (2-2.5% Cr) (61788-69-0) green liq.	100g 500g
COBALT	Compounds)	
27-1025 NEW	Bis(1,4-di-t-butyl-1,3-diazabutadienyl)cobalt(II) Co(DAD) ₂ , min. 98% (99.999%-Co) PURATREM (177099-51-3) C ₂₀ H ₄₀ CoN ₄ ; FW: 395.49; dark green-blue xtl. <i>air sensitive</i> Note: U.S. Patent Application No. 13/818,154. Product sold under, use subject to, terms and conditions of label license at www.strem.com/waynestate1	250mg 1g 5g
27-0486 NEW amp HAZ	Bis(N,N'-di-i-propylacetamidinato)cobalt(II), min. 98% (99.99%-Co) PURATREM (Co(iPr- MeAMD) ₂ (635680-58-9) C ₁₀ H ₃₄ CoN ₄ ; FW: 341.40; green xtl. <i>air sensitive, moisture sensitive</i> Note: Product sold under, use subject to, terms and conditions of label license at www.strem.com/harvard2.	250mg 1g 5g
27-4010 NEW	lambda-Tris[(15,25)-1,2-diphenyl-1,2-ethane- diamine]cobalt(III) chloride tetrakis[3,5-bis(tri- fluoromethyl)phenyl]borate dihydrate SKJ-1 ($1542135-29-4$) $C_{74}H_{60}BCI_2CoF_{24}N_6$; FW: 1629.92(1665.92); orange pwdr. Note: U.S. Patent 14/417655 Note: U.S. Patent 14/417655 H_{1} , H_{2} , H	50mg 250mg

COBALT (Compounds)

27-4011 NEW

delta-Tris[(1S,2S)-1,2-diphenyl-1,2-ethanediamine]cobalt(III) chloride tetrakis(2,3,4,5, 6-pentafluorophenyl)borate trihydrate SKJ-3 (1867120-15-7) $C_{\rm e6}H_{\rm 48}BCl_2CoF_{20}N_6$; FW: 1445.74 (1501.80); orange solid Note: U.S. Patent 14/417655



SCF₃

50mg 250mg

COPPER (Compounds)

29-0225 New	Copper(I) bromide, dimethyl sulfide complex, 99% (54678-23-8) BrC ₂ CuH ₆ S; FW: 205.59; off-white to light-green pwdr. <i>moisture sensitive</i>	5g 25g
29-5001 NEW	Copper(II) trifluoromethanesulfonate, 99% (99.9%-Cu) (Copper triflate) (34946-82-2) Cr(25-CO) > F O O F F	5g 25g
HAZ	$\begin{array}{c} Cu(Cr_3SO_3)_2, \ \text{Fight green solid} \\ air sensitive \\ \end{array} \qquad \qquad$	TUUg
29-5515	Trifluoromethylthiolato(2,2-bipyridine)copper(I), 97% (1413732-47-4) C ₁₁ H ₈ CuF ₃ N ₂ S; FW: 320.80; red xtl. air sensitive, moisture sensitive	250mg 1g

Technical Notes:

- 1. Catalyst for nucleophilic trifluoromethylthiolation of aryl halides.
- 2. Catalyst for synthesis of α -trifluoromethylthio- and seleno- α , β -unsaturated carbonyl compounds.
- 3. Catalyst for trifluoromethylthiolation of vinyl bromides.
- 4. Catalyst for trifluoromethylthiolation of bromopyridines.



Tech. Note (1) Ref. (1,2)

X = Br, I

R = H, Me, Ph, Cl, CO₂Me, CN, NO₂, OMe



Tech. Note (3) Ref. (4)

Tech. Note (4) Ref. (4)

Tech. Note (2) Ref. (3)

R = OMe, NMe, NO₂, CF₃, F, Cl, et al.



References:

1. Angew. Chem. Int. Ed. 2013, 52, 1548.

(bpy)Cu(SCF₃)

KF, Diglyme

- 2. Tetrahedron, 2013, 69, 6046.
- 3. Tetrahedron, 2014, 70, 672.
- 4. J. Org. Chem. **2015**, 80, 2912.
- 5. Adv. Synth. Catal. **2016**, 358, 386.

_

ELECTROPOLISHED STAINLESS STEEL BUBBLERS (Vertical)

95-4002 New	Stainless steel bubbler, 1200ml, vertical, electropolished with fill-port, high temp valves (315°C), DOT 4B, UN stamped Note: See the Technical Note tab at strem.com for drawings.	1cyl
95-3000 New	Stainless steel bubbler, 150ml, vertical, electropolished with fill-port, replaceable-seat valves with rotated handles, DOT 4B, UN stamped Note: See the Technical Note tab at strem.com for drawings.	1cyl

GERMAN	IUM (Compounds)	
32-3215	Germanium(II) chloride dioxane adduct (28595-67-7)	2g
NEW	C₄H ₈ Cl₂GeO₂; FW: 231.65 air sensitive, moisture sensitive	10g
GOLD (EI	emental Forms)	
79-0921	Gold nanoparticles, 1% on carbon black (surfactant and reactant-free)	5g
NEW	(7440-57-5) Au: EW: 106-70: block colid	25g
_	(store cold)	
	Note: Manufactured by laser ablation. Sold in collaboration with Particular®	
	for research purposes only.	
79-0916	Gold nanoparticles, 1% on Titania (anatase) (surfactant and reactant-free)	5g
NEW	(7440-57-5)	25g
	Au; FW: 196.70; dark purple pwdr.	
	Note: Manufactured by laser ablation. Sold in collaboration with Particular® for	
	research purposes only.	
79-0905	Gold nanoparticles, 1% on Titania (rutile) (surfactant and reactant-free)	5g
NEW	(7440-57-5)	25g
	Au; FW: 196.70; purple solid (store cold)	
	Note: Manufactured by laser ablation. Sold in collaboration with Particular® for	
	research purposes only.	
79-0926	Gold nanoparticles, 5% on carbon black (surfactant and reactant-free)	5g
NEW	(7440-57-5)	25g
	AU; FW: 196.70; black solid (store cold)	
	Note: Manufactured by laser ablation. Sold in collaboration with Particular® for	
	research purposes only.	
79-0935	Gold nanoparticles, 10% on Titania (anatase) (surfactant and reactant-free)	1g
NEW	(7440-57-5)	5g
_	Au; FW: 196.70; purple solid	
	Note: Manufactured by laser ablation. Sold in collaboration with Particular® for	
	research purposes only.	
79-0930	Gold nanoparticles, 10% on Titania (rutile) (surfactant and reactant-free)	1g
NEW	(7440-57-5)	5g
_	Au; F vv: 196.70; dark purple solid (store cold)	
	Note: Manufactured by laser ablation. Sold in collaboration with Particular® for	
	research purposes only.	

GOLD (Compounds)

79-1230 NEW

Chloro{1,3-bis[2,6-bis(1-methylethyl) phenyl]-1,3-dihydro-4,5-dimethyl-2H-imidazol-2-ylidene}gold(I), 98% IPrMeAuCI (1192141-66-4) $C_{29}H_{40}AuCIN_2$; FW: 649.0; white pwdr. *air sensitive*



100mg 500mg

5g

25g

INDIUM (Compounds)

49-4901 Indium(III) acetylacetonate (99.99+%-In) PURATREM (14405-45-9) In(CH₃COCHCOCH₃)₃; FW: 412.15; off-white pwdr.; m.p. 180-185; b.p. 260-280 subl.; d. 1.41



 PF_6

F₃C

F₂C

50mg 250mg

IRIDIUM (Compounds)

77-0220 (2,2'-Bipyridine)bis[3,5-difluoro-2-[5-(trifluoro-methyl)-2-pyridinyl-kN][phenyl-kC]iridium(III) hexafluorophosphate, 95% (1092775-62-6) C₃₄H₁₀F₁₆IrN₄P; FW: 1009.70; yellow pwdr. air sensitive Note: Photocatalyst

Technical Notes:

- 1. Photocatalyst used for the chemo-, regio, and stereoselective trifluoromethylation of styrene.
- Photoredox catalyst used in cross-coupling: Ir/Ni dual catalysts for the synthesis of benzylic ethers.
- 3. Iridium complex used for catalytic olefin hydroamidation enabled by proton-coupled electron transfer.
- Catalyst used for visible light photoredox cross-coupling of acyl chlorides with potassium alkoxymethyltrifluoroborates.
- Iridium catalyst used in the photoredox/nickel dual catalytic cross-coupling of secondary alkyl β-trifluoroboratoketones and –esters with aryl bromides.
- 6. Photocatalyst used in the cross-coupling of trifluoroalkylboranes.







1. See 77-0220 (page 37)







IRIDIUM (Compounds) 77-6580 Tris[(2-(2-pyridinyl-kN)-5-(trifluoromethyl) 50mg F₃C phenyl-kC]iridium(III), 95% (500295-52-3) 250mg NEW C₃₆H₂₁F₉IrN₃; FW: 858.78; yellow solid air sensitive Ν Note: Photocatalyst CF₃ CF₃ **IRON** (Compounds) 26-0873 Bromocarbonyl[(1S,2S)-2,3-100mg Ph Ph diphenylethylenediamine-N,N'-bis(2-500mg NEW diphenylphosphinoethyllidene)]iron(II) tetraphenylborate, FeATHer-II Catalyst Br (1257252-03-1) BPh₄ C₆₇H₆₅BBrFeN₂OP₂; FW: 1122.75; yellow pwdr. Fe air sensitive, moisture sensitive Note: Sold in collaboration with GreenCentre C Ш for research purposes only. Patents: PCT/ Ph Ph 0 CA2013/050405, PCT 2013/010275. Technical Note: Catalyst used in the transfer hydrogenation of ketones and imines. 1 0.016 to 0.05 mol% Fe catalyst HO R² 0.033 - 0.40 mol% KOtBu Tech. Note (1) mm Ref. (1) ⁱPrOH. 28 °C R¹ R² R¹ 10s to 1 hour PPh₂ 1.0 mol% Fe cat. PPh₂ 8 mol% KO^tBu Tech. Note (1) HN Ref. (1) ⁱPrOH. 28 °C Me 10 s to 3 min R¹ R¹ Me References: Science, 2013, 342, 1080. 1. 26-1240 1-Diphenylphosphino-1'-(di-t-butylphosphino) 250mg PPh₂ ferrocene, 97% (95408-38-1) 1g NEW C₃₀H₃₆FeP₂; FW: 514.40; yellow to orange pwdr.; Fe m.p. 75-79° (tBu)₂P⁻ air sensitive

ANTHANUM (Compounds) 57-0505 Lanthanum(III) bromide, anhydrous (99.99%-La) (REO) PURATREM 2g (13536-79-3)10g NEW LaBr₃; FW: 378.62; white xtl. air sensitive, moisture sensitive 57-1250 250mg Tris[N,N,N,N-tetramethylguanidinium] [tris(1S)-(1.1'-binaphalene)-2.2'-diolato] 1g NEW lanthanate La-HTMG-B (1611526-71-6) Me₂N NMe₂ C₇₅H₇₈N₉O₆La; FW: 1340.38; tan pwdr. Me₂N N-H-Note: U.S. Patent 14/898,925. òo -H-N NMe₂ Technical Notes: 1. Catalyst used for an asymmetric Michael addition. Ċ 2. Catalyst used for an asymmetric Michael addition of C 1,3-dicarbonyls to enones. 3 Catalyst used for the asymmetric aza-Michael addition of methylhydroxylamine to chalcone derivatives. Me₂N NMe₂ 4. Catalyst used for the asymmetric direct aldol reaction of acetophenone and pivaldehyde. Lan. catalyst (10 mole%) Na-X (30 mole%) H₂O (X moles) THF, 25°C, 12 h O O ↓ ↓ OBn -Tech. Note (1) OBn Ref. (1) [`]O BnO $\hat{}$ Lan. catalyst (10 mole%) Nal (3x mole%) H₂O (3x mole%) Tech. Note (2) OR^1 Ref. (1) THF. 0°C. 12 h R¹O Lan. catalyst (x mole%) Lil (3x mole%) Tech. Note (3) NHOCH₃ Drierite 1 equiv Ref. (1) $_{\rm A}$ + CH₃ONH₂ THF, -20°C, 48 h Lan. catalyst (8 mole%) Li-I (24 mol%) Tech. Note (4) KO(tBu) (8 mol%) H₂O (16 mol%) THE, -20°C. 20 h Ref. (1) THE -20°C 20 h References: 1. J. Am. Chem. Soc., 2014, 136, 8034.

MANGAN	ESE (Compounds)	
25-1345	Manganese(II) chloride anhydrous (99.995%-Mn) PURATREM	5g
NIEW	(7773-01-5)	25g
INE W	MnCl ₂ ; FW: 125.84; pink solid; m.p. 650°; b.p. 1190°; d. 2.98	100g
	hygroscopic	

METALS SCAVENGING AGENTS (Compounds) 07-2203 1,4-Bis(2-isocyanopropyl)piperazine (SnatchCat 1g NC Metal Scavenger) (51641-96-4) 5q NEW C12H20N4; FW: 220.31; white xtl. (store cold) Note: Sold in collaboration with Apeiron Synthesis, Inc. U.S. Patent 14/443,048; PCT/IB2014/062564. CN Technical Note: SnatchCat represents a universal tool for metal removal with a proven efficiency for Ru and Pd metals. The 1 product is an immediate metathesis reaction guencher, allowing reduced metal content to 10ppm after 20 min scavenging time and a simple workup by silica gel filtration. Compatible with a broad range of functional groups and solvents, the product is stable, non-toxic, non-volatile and odor-free. 1. Nitro-Grela, 1 mol% 70°C, 60 min COOF 2. SnatchCat OOF 70°C, 30 min 3. Filtration through silica gel Tech. Note (1) Ref. (1) PhB(OH)₂ (PPh₃)₂PdCl₂ (5 mol% 2. SnatchCat, 30 min 3. Filtration through silica gel References: 1. ChemSus Chem., 2015, 8, 4139. 06-1522 Chelating/scavenger resin with aminophosphonic - S940 50g spherical beads (wet) 5x50q NEW (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 425-850 micron; Purolite S940 is a chelating resin of macroporous structure, with a polystyrene matrix crosslinked with divinylbenzene (DVB) substituted with weakly acidic aminophosphonic active groups. This chemical structure facilitates the formation of complexes with metallic ions. The aminophosphonic chelating resins have a greater affinity for certain cations, and form more stable complexes with cations of low atomic mass metals than their iminodiacetic resin counterparts. Hence Purolite S940 is capable of fixing one or more specific cations from a larger range even from solutions which are highly concentrated. Sold in collaboration with Purolite for research purposes only. 06-1525 Chelating/scavenger resin with aminophosphonic - S950 50g spherical beads (wet) 5x50a NEW (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-1000 micron; Purolite S950 is a macroporous aminophosphonic acid chelating resin, designed for the removal of cations of toxic metals such as lead, copper and zinc from industrial effluents at low pH. At somewhat higher pH values, calcium, magnesium and barium, as well as the toxic metals cadmium, nickel, and cobalt are strongly complexed and may be separated from guite high concentrations of univalent cations. Purolite S950 is highly selective (under the appropriate conditions) for a range of both heavy metal and common divalent ions. Hence its use may be recommended where it is necessary to remove calcium or magnesium in order to avoid possible precipitation, or where its selectivity for a particular range of metals offers advantages. Sold in collaboration with Purolite for research purposes only.

METALS S	SCAVENGING AGENTS (Compounds)	
06-1508 (New)	Chelating/scavenger resin with aminoxime - S910 spherical beads (wet) (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-1200 micron; Purolite® S910 is an amidoxime chelating resin, designed for the removal of cations of metals such as copper and iron from water or other solvent even at relatively low pH. It can also be used for the recovery of traces of precious metals from dilute solutions. It cannot be used for removal of alkaline earth metals. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
06-1530 New	Chelating/scavenger resin with bispicolylamine - S960 spherical beads (wet) (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 425-1000 micron; Purolite® S960 is a macroporous poly- styrene divinylbenzene copolymer within which weakly basic bis-picolylamine chelating functional groups are covalently bonded. This resin has high selectivity for metals such as nickel, copper, and cobalt as compared to iron, aluminum, calcium, and magnesium. Purolite® S960 is active within the pH range of 0 8 and may function outside that range depending on the metals involved. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
06-1520 New	Chelating/scavenger resin with iminodiacetic - S930Plus spherical beads (wet) (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 425-1000 micron; Purolite S930Plus is a macroporous polystyrene; based chelating resin, with iminodiacetic groups designed for the removal of heavy metals from industrial effluents. Purolite S930Plus finds use in processes for extraction and recovery of metals from ores, galvanic plating solutions, pickling baths and effluents. Further uses include the decalcification of brine for chloralkali processes, where Purolite S930Plus shows advantages under certain operating conditions over the typically used aminophosphonic type resins such as Purolite S940. Purolite S930Plus has high selectivity and capacity for hardness and strontium and has excellent osmotic stability. Purolite S930Plus is susceptible to oxidation. Hence direct treatment of brine solutions containing free chlorine should be avoided, for instance by preliminary reaction with sulphur dioxide, sulphide or, by use of a treatment with activated carbon. Brine solutions can often contain significant concentrations of chlorates. In this case it is necessary to ensure that the displacement rinse prior to the acid regeneration is efficient, so as to avoid the formation of free chlorine from contact of chlorates in the brine solution with the regeneration acid. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
06-1514 NEW	Chelating/scavenger resin with isothiouronium - S920Plus spherical beads (wet) (<i>store cold</i>) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-1200 micron; Purolite® S920Plus is a macroporous polystyrenic based chelating resin, with thiouronium groups designed for the selective removal of mercury and for the recovery of precious metals from industrial effluents. Mercury is strongly bound to the functional groups to form highly stable complexes, with high selective affinity compared with those of other heavy metals. These properties are largely unaffected by high chloride (or sulfate) content of the effluent. Effluent solutions which may typically contain 2 - 20 ppm of mercury can be treated to reduce the concentration in solution to less than 0.005 ppm. Purolite S920Plus can load up to 200g of mercury, or gold, or approximately 60g of platinum or palladium for each liter of resin, equivalent to 12.5, and 3.75 lb/f3 respectively. Purolite S920Plus is designed for the removal of low concentrations of soluble mercury salts from waste streams and for the recovery of precious metals from rinse waters in the galvanic and electronic industries. It is also used in hydrometallurgy for the separation of precious metals from acid liquors. Mercury and precious metals are so strongly held, and run lengths are so long (thousands of hours) that it is not normally considered economic to regenerate the resin for reuse. Purolite S920Plus is more resistant to oxidation than many thiol based resins and contact with the atmosphere is not detrimental, however free chlorine and other strong oxidizing agents may damage the resin and their removal from solution by filtering through activated carbon is recommended. Sold in collaboration with Purolite for research purposes only.	50g 5x50g

METALS	SCAVENGING AGENTS (Compounds)	
06-1501 (NEW)	Chelating/scavenger resin with N-methylglucamine - S108 spherical beads (wet) (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 425-630 micron; Purolite® S108 is a macroporous polystyrenic based resin with excellent kinetics and functional groups specially designed for the selective removal of salts of boron from aqueous solutions. It is effective for such solutions over a wide range of pH values, and over a wide range of boron concentrations. The presence of boron ions in water for potable and agriculture/horticulture use, even in relatively small (ppm) concentrations can give rise to major problems. Even where concentrations of other ions are reasonably high, Purolite® S108 will reduce boron concentrations by an order of magnitude. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
06-1528 (NEW)	Chelating/scavenger resin with phosphonic and sulfonic acid - S957 spherical beads (wet) (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 425-1000 micron; Purolite S957 is a specially developed Monophos chelating resin, which incorporates phosphonic, and sulfonic functional groups on a mechanically and osmotically resistant matrix. These combined properties give it high selectivity for iron and other transitional metals, even in acidic solutions. Purolite S957 has been especially designed for the selective removal of ferric iron from acidic solutions, such as copper electrolyte or from nickel, cobalt and zinc processing solutions. Purolite S957 can also be used in potable water applications for the selective removal of trace levels of selected metals from neutral pH waters. Its selectivity for uranium and other lanthanide elements should give good opportunities for its successful employment in other areas. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
06-1532 New	Chelating/scavenger resin with polyamine - S985 spherical beads (wet) (store cold) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-1200 micron; Purolite® S985 is a high capacity, macroporous, weak base anion exchange resin with a polyacrylic matrix supporting functional groups of the polyamine type. The carefully formulated, macroporous acrylic matrix ensures excellent exchange kinetics for the removal of trace heavy metals from waste water streams and the special polyamine functionality produces very interesting operating capacities and makes the uptake of metallic cations possible even when they are present in the waste stream as organic anionic complexes. Its tough and resilient macroporous structure also affords excellent mechanical, strength and resistance to osmotic shock. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
06-1518 NEW	Chelating/scavenger resin with thiol - S924 spherical beads (wet) (<i>store cold</i>) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-1000 micron; Purolite S924 is a chelating resin, polystyrene based and designed for the selective removal of mercury. The mercury is strongly bound to the functional groups to form highly stable complexes with high selective affinity compared with those of other heavy metals. Even so the high selective affinity such as silver, copper, lead, cadmium, nickel and cobalt, makes this resin useful in waste treatment and hydrometallurgical processes. The high selectivity for mercury is largely unaffected by high chloride or sulphate content of the effluent. Effluent solutions that may typically contain 0.01-25ppm of mercury can be treated to reduce the concentration to significantly less than 5ppb residual mercury. Purolite S924 can load up to 150 g (16 lb/cu.ft) of mercury per litre of resin. Purolite S924 is designed for the removal of moderately low concentrations of soluble mercury sells from brine streams used to produce caustic soda and chlorine where mercury sells are used, and may be regenerated with concentrated hydrochloric acid solutions. In the process for the manufacture of caustic soda and chlorine from brine, where all or part of the production uses mercury cells, the mercury rich regenerant acid may be neutralized with the sodium hydroxide to produce a recovered brine solution that may be recycled to the mercury cell process. Mercury may be present at very low concentrations and consequently run lengths are often long (thousands of hours). It is sometimes not economic to regenerate the resin for re-use. In such cases Purolite S920 may be preferred because of its higher capacity. Purolite S924 is prone to oxidation and long-term contact with the atmosphere is detrimental. It is recommended that this resin is shipped and stored under water. Also, free chlorine and other strong oxidizing agents may damage the resin. Their removal	50g 5x50g

METALS SCAVENGING AGENTS (Compounds)

06-1512 New	Chelating/scavenger resin with thiourea - S914 spherical beads (wet) (<i>store cold</i>) Note: Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 years; Particle Size: 300-1200 micron; Purolite S914 is a macroporous type chelating resin with thiourea functionality. It has very high selectivity for mercury and platinum group metals such as platinum, gold and silver. Purolite S914 is stable over the whole pH range. The applications include mercury removal from brine and effluent in chloralkali process, mercury removal from flue gas scrubber effluent, recovery of platinum group metals from effluents amongst others. Sold in collaboration with Purolite for research purposes only.	50g 5x50g
MOFS ANI	D LIGANDS FOR MOF SYNTHESIS (Compounds)	
22-1070	Hexakis[µ-(2-amino-1,4-benzenedicarboxylato)][tetra-µ-hydroxyoc- ta-µ-oxooctatitanium], NH2-MIL-125(Ti), CONEKTIC™ T125 (1309760-94-8) See page 96	
40-1109	Zirconium aminobenzenedicarboxylate MOF (UiO-66-BDC-NH2, BDC- NH2:Zr=0.9-1.0) (1260119-00-3) See page 97	
40-1108	Zirconium benzenedicarboylate MOF (UiO-66-BDC, BDC:Zr=0.66-0.98) See page 98	
40-1112	Zirconium biphenyldicarboxylate MOF (UiO-66-BPDC/UiO-67, BPDC:Zr=0.9-1.0) See page 98	
40-1114	Zirconium Fumarate MOF (UiO-66-FA, FA:Zr=0.66-0.98) See page 99	
40-1106	Zirconium trans-1, 2-ethylenedicarboxylic acid MOF (UiO-66-FA, FA:Zr=1) See page 99	
40-1111	Zirconium trimellitate MOF (UiO-66-BDC-COOH, BDC-COOH:Zr=0.9-1.0) See page 100	
NANOMAT	[ERIALS (Elemental Forms)	
79-0921	Gold nanoparticles, 1% on carbon black (surfactant and reactant-free) (7440-57-5) See page 36	
79-0916	Gold nanoparticles, 1% on Titania (anatase) (surfactant and reac- tant-free) (7440-57-5) See page 36	
79-0905	Gold nanoparticles, 1% on Titania (rutile) (surfactant and reactant-free) (7440-57-5) See page 36	
79-0926	Gold nanoparticles, 5% on carbon black (surfactant and reactant-free) (7440-57-5) See page 36	
79-0935	Gold nanoparticles, 10% on Titania (anatase) (surfactant and reac- tant-free) (7440-57-5) See page 36	
79-0930	Gold nanoparticles, 10% on Titania (rutile) (surfactant and reactant-free) (7440-57-5) See page 36	
06-2530	Graphene oxide (4mg/ml water dispersion) - low Mn. (1034343-98-0) See page 33	

NANOMATERIALS (Elemental Forms)

06-0330 Graphene Quantum Dots (GQDs), Aqua-Green Luminescent (1034343-98-0) C; dark red-brown pwdr. light sensitive, (store cold) C; dark red-brown pwdr. Note: Particle diameter: <5 nm. Sold in collaboration with Dotz Nano Ltd. for research purposes only. Suggested use within 6 months of purchase. Do not freeze. Store in DARK.

Suggested Applications:

Graphene quantum dots (GQDs), sheets of few-layered graphene and lateral dimensions smaller than 100nm possess strong quantum confinement and edge effects. Thus, they possess unique physical properties such as strong photoluminescence, which can be tailored for specific applications by controlling their size, shape, defects and functionality.

In contrast to classic QDs, such as metal or silicon quantum dots, GQDs are biocompatible, photostable and inherit superior thermal, electrical and mechanical properties from the graphene. These features can greatly contribute to various state-of-the-art applications: optical brighteners, taggants for security applications¹, bioimaging markers², fluorescent polymers³, antibacterial⁴, antibiofouling⁵, and disinfection systems⁶, heavy metals⁷, humidity and pressure⁸ sensors, batteries⁹, flash memory devices¹⁰, photovoltaic devices¹¹ and light-emitting diodes¹².

Itom #	Photoluminescence			
nem #	QY* *	λ _{max} *	Max emission	FWHM *
06-0330 / 06-0332	>17%	485 nm	525 nm	70 nm
06-0334 / 06-0336	>65%	350 nm	445 nm	65 nm
06-0340	>25%	420 nm	490 nm	80 nm
Abbreviations		-		
QY+	Quantum Yield			
λ _{max} Maximum excitation		wavelength		
FWHM	Full width at half max	ximum		



References:

- 1. http://onlinelibrary.wiley.com/doi/10.1002/anie.201206791/abstract
- 2. http://onlinelibrary.wiley.com/doi/10.1002/ppsc.201400219/abstract
- 3. http://pubs.acs.org/doi/abs/10.1021/acsami.5b06057
- 4. http://pubs.acs.org/doi/abs/10.1021/acsami.6b01765
- 5. http://www.nature.com/articles/srep20142
- 6. http://pubs.acs.org/doi/abs/10.1021/nn501640q
- 7. http://www.sciencedirect.com/science/article/pii/S0013468615000468
- 8. http://pubs.acs.org/doi/abs/10.1021/nl4003443
- 9. http://pubs.acs.org/doi/abs/10.1021/nl504038s
- 10. http://iopscience.iop.org/article/10.1088/0957-4484/25/25/255203/meta
- 11. http://onlinelibrary.wiley.com/doi/10.1002/anie.200906291/abstract
- 12. http://link.springer.com/article/10.1007/s10853-012-7016-8

06-0332	Graphene Quantum Dots (GQDs) in water, Aqua-Green Luminescent	100ml
	(1034343-98-0)	
NEW	C; cloudy orange lig.	
	light sensitive, (store cold)	
	Note: Particle diameter: <5 nm. Concentration: 1 mg/ml. Sold in	
	collaboration with Dotz Nano Ltd. for research purposes only. Suggested use	
	within 6 months of purchase. Do not freeze. Store in DARK.	
Technical Note:	· · · · · · · · · · · · · · · · · · ·	
1. See 06-0)330 (page 48)	

100mg

NANOMA	TERIALS (Elemental Forms)	
06-0334 NEW	Graphene Quantum Dots (GQDs), Blue Luminescent (1034343-98-0) C; dark brown pwdr. light sensitive, (store cold) Note: Particle diameter: <5 nm. Sold in collaboration with Dotz Nano Ltd. for research purposes only. Suggested use within 6 months of purchase. Do not freeze. Store in DARK.	100mg
Technical Note: 1. See 06-	0330 (page 48)	
06-0336 New	Graphene Quantum Dots (GQDs) in water, Blue Luminescent (1034343-98-0) C; cloudy colorless liq. <i>light sensitive, (store cold)</i> Note: Particle diameter: <5 nm. Concentration: 1 mg/ml. Sold in collaboration with Dotz Nano Ltd. for research purposes only. Suggested use within 6 months of purchase. Do not freeze. Store in DARK	100ml
Technical Note: 1. See 06-	0330 (page 48)	
06-0340 NEW	Graphene Quantum Dots (GQDs) in water, Cyan Luminescent (1034343-98-0) C; cloudy brown liq. <i>light sensitive, (store cold)</i> Note: Particle diameter: <5 nm. Concentration: 1 mg/ml. Sold in collaboration with Dotz Nano Ltd. for research purposes only. Suggested use within 6 months of purchase. Do not freeze. Store in DARK.	100ml
Technical Note: 1. See 06-	0330 (page 48)	
96-7410	Graphene Quantum Dots (GQDs) Master Kit (1034343-98-0) See page 108	
96-7425	Graphene Quantum Dots (GQDs) Mini Kit (Powders) (1034343-98-0) See page 108	
96-7420	Graphene Quantum Dots in water (GQDs) Mini Kit (Liquids) (1034343-98-0) See page 109	
06-0365	High Strength Metallurgical Graphene on GLASS HSMG® (10x10mm) (1034343-9 See page 34	8-0)
06-0345	High Strength Metallurgical Graphene on PMMA HSMG® (10x10 mm) (1034343-9 See page 34	8-0)
06-0355	High Strength Metallurgical Graphene on PMMA HSMG® (25x25mm) (1034343-98 See page 34	8-0)
06-0360	High Strength Metallurgical Graphene on PMMA HSMG® (50x50mm) (1034343-98 See page 34	8-0)
28-0015	Nickel/palladium alloy nanoparticle on graphene (G-Ni ₃₃ Pd ₆₇) See page 62	
78-3015	Platinum nanoparticles, 1% on carbon black (surfactant and reactant-free) (7440- See page 80	-06-4)
78-3020	Platinum nanoparticles, 5% on carbon black (surfactant and reactant-free) (7440- See page 80	-06-4)
78-3030	Platinum nanoparticles, 10% on carbon black (surfactant and reactant-free) (744) See page 80	0-06-4)
78-3032	Platinum nanoparticles, 20% on carbon black (surfactant and reactant-free) (744) See page 80	0-06-4)
78-3035	Platinum nanoparticles, 30% on carbon black (surfactant and reactant-free) (744) See page 80	0-06-4)
78-3012	Platinum nanoparticles, 1% on Titania (anatase) (surfactant and reactant-free) (7 See page 81	440-06-4)
78-3005	Platinum nanoparticles, 1% on Titania (rutile) (surfactant and reactant-free) (7440 See page 81	0-06-4)
78-3026	Platinum nanoparticles, 10% on Titania (anatase) (surfactant and reactant-free) (See page 81	7440-06-4)
78-3023	Platinum nanoparticles, 10% on Titania (rutile) (surfactant and reactant-free) (744 See page 81	40-06-4)

NANOMA	TERIALS (Compounds)	
14-6100	High Surface area Silica nanoparticles, large, particle size ~900-1000 nm, surface area ~700 m²/g, (KCC-1 L1) <i>(112945-52-5)</i> See page 92	
14-6110	High Surface area Silica nanoparticles, large, particle size ~900-1000 nm, surface area ~600 m²/g, (KCC-1 L2) (112945-52-5) See page 93	
14-6120	High Surface area Silica nanoparticles, large, particle size ~900-1000 nm, surface area ~550 m²/g (KCC-1 L3) (112945-52-5) See page 93	
14-6200	High Surface area Silica nanoparticles, medium, particle size ~400-450 nm, surface area ~400 m²/g, (KCC-1 M1) (112945-52-5) See page 93	
14-6210	High Surface area Silica nanoparticles, medium, particle size ~300-350 nm, surface area ~600 m²/g, (KCC-1 M2) (112945-52-5) See page 94	
14-6300	High Surface area Silica nanoparticles, small, particle size ~130-190 nm, surface area ~380 m²/g, (KCC-1 S1) <i>(112945-52-5)</i> See page 94	
14-6310	High Surface area Silica nanoparticles, small, particle size ~40-50 nm, surface area ~520 m²/g, (KCC-1 S2) (112945-52-5) See page 94	
	Compounds)	
28-1210	(Acetonitrile)dichloronickel(II), 99% (18897-44-4)	1g
NEW	C ₂ H ₃ Cl ₂ NNi; FW: 170.65; yellow pwdr. air sensitive, hygroscopic	5g
28-0009 New HAZ	Allyl(cyclopentadienyl)nickel(II), min. 97% (12107-46-9) C ₈ H ₁₀ Ni; FW: 164.86; dark purple liq.; d. 1.31 air sensitive, moisture sensitive, pyrophoric	1g 5g
28-0225 New	Bis(1,4-di-t-butyl-1,3-diazabutadienyl)nickel(II) Ni(DAD)2, min. 98% (99.999%-Ni) PURATREM C ₂₀ H ₄₀ N ₄ Ni; FW: 395.25; Bis(1,4-di-t-butyl-1,3-diazabutadienyl)nickel(II) Ni(DAD)2, min. 98% (99.999%-Ni) PURATREM <i>air sensitive</i> Note: U.S. Patent Application No. 13/818,154. Product sold under, use subject to, terms and conditions of label license at www.strem.com/waynestate1	250mg 1g 5g
28-0178 New	$\begin{array}{c} \mbox{Chloro}(4\mbox{-cyanophenyl})\{(R)\mbox{-1-[(S)-2-(bis(4\mbox{-fluo-rophenyl})phosphinoferrocenyl]ethyl(di-t-butyl-phosphine)}nickel(II) (2049086\mbox{-37-3}) \\ C_{39}H_{42}\mbox{Cl}_{2}\mbox{Fe}NNip_2; FW: 774.69; red-violet solid Note: Sold in collaboration with Solvias for research purposes only.} \\ \end{array}$	100mg 500mg
Technical Note: 1. See 28-	0170 (page 51)	
28-0172 NEW	$\begin{array}{c} \label{eq:constraint} \begin{tabular}{lllllllllllllllllllllllllllllllllll$	100mg 500mg
 See 28- 	0170 (page 51)	

NICKEL (Compounds)

28	-0	1	7	1

'5	Chloro(4-cyanophenyl){(R)-1-[(S)-2-(dicyclo-
	hexylphosphino)ferrocenyl]ethyl (diphenyl-
	phosphine)}nickel(II) (2049086-36-2)
	C ₄₃ H ₄₈ ClFeNNiP ₂ ; FW: 790.79; orange solid
	Note: Sold in collaboration with Solvias for
	research purposes only.



100mg 500mg

Technical Note: See 28-0170 (page 51) 1.

28-0170 Chloro(4-cyanophenyl){(R)-1-[(S)-2-(diphen-100mg Me ylphosphino)ferrocenyl]ethyl(di-t-butyl) 500mg NEW P(tBu)₂ phosphine} nickel(II) (2049086-34-0) Ph C₃₀H₄₄ClFeNNiP₂; FW: 738.71; red-violet solid Note: Sold in collaboration with Solvias for research purposes only. Technical Notes: Versatile, air-stable, low cost nickel catalyst alternative to palladium for carbon-carbon and 1 carbonheteroatom cross-coupling reactions. 2. Used to react substituted aryl and heteroaryl halides and tosylates with ammonia to produce diverse aryl and

- heteroaryl amines1. 3. Used in monoarylation experiments using commercially available ammonia gas, ammonium salts or ammonia stock solutions^{2,3}.
- 4. Catalyzes the coupling of aryl chlorides with gaseous amines in the form of their hydrochloride salts³. References:
 - Org.Process Res. Dev., 2015, 19, 1936. 1.
 - Angew. Chem, Int. Ed., 2015, 54, 3773. 2.
 - 3. Angew. Chem, Int. Ed., 2015, 54, 3768.



Technical Note:

Air-stable nickel precatalyst for the amination of aryl chlorides, sulfamates, mesylates and triflates. 1.

2.5-1 Lie	0 mol% Ni catalyst Ot-Bu or K ₃ PO ₄			
(Het)ArX + R ₂ NH	>	(Het)Ar-NR ₂	Tech. Note (1)	
X= Cl, OTf, OSO ₂ NMe ₂ , OMs		60-98% yield	Ker. (1)	
References: 1. Org. Lett., 2014 , 16,	220.			
96-3660 Solvias J See page	osiphos Nickel Catal 110	lyst Kit		

NITROGEN (Compounds) 07-2061 (2S)-2-[[[[(1R,2R)-2-Aminocyclohexyl] 50mg amino]thioxomethyl]amino]-N-(diphenyl-NEW methyl)-N,3,3-trimethylbutanamide, 98%, NH₂ (99% ee) (1421052-39-2) N C₂₇H₃₈N₄OS; FW: 466.70; white to yellow pwdr. Me tBu S Note: Sold in collaboration with Daicel for research purposes only. (2S)-2-[[[[(1S,2S)-2-Aminocyclohexyl] 07-2060 50mg amino]thioxomethyl]amino]-N-(diphenyl-NEW methyl)-N,3,3-trimethylbutanamide, 98%, ′NH₂ н (99% ee) (1421697-46-2) Ph C₂₇H₃₈N₄OS; FW: 466.70; white to yellow pwdr. Ьe tBu S Note: Sold in collaboration with Daicel for research purposes only. 07-0955 (2R)-2-[[[(1S,2S)-2-Aminocyclohexyl] 50ma tBu Me amino]thioxomethyl]amino]-N,3,3-NEW trimethyl-N-(phenylmethyl)butanamide, н 98%, (99% ee) н ÑH₂ C₂₁H₃₄N₄OS; FW: 390.60; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 07-0956 (2S)-2-[[[[(1R,2R)-2-Aminocyclohexyl]amino]thioxomethyl]amino]-50mg N,3,3-trimethyl-N-(phenylmethyl)butanamide, 98%, (99% ee) NEW (479423-21-7)C₂₁H₃₄N₄OS; FW: 390.60; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 07-0988 N-I(1R.2R)-2-Aminocyclohex-100ma н Н yl]-N'-[3,5-bis(trifluoromethyl)phenyl] NEW CF₃ thiourea, 98%, (99% ee) (860994-58-7) C₁₅H₁₇F₆N₃S; FW: 385.40; white to yellow pwdr. 'NH₂ Note: Sold in collaboration with Daicel for research purposes only. CFa 07-7018 (2S)-2-Amino-3,3-dimethyl-N-(phenylmethyl)butanamide, 98%, (99% 100ma ee) (207121-91-3) NEW C₁₃H₂₀N₂O; FW: 220.30; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 07-7019 (2R)-2-Amino-3,3-dimethyl-N-(phenylmethyl) 500ma \cap butanamide, 98%, (99% ee) (268556-62-3) NH₂ NEW C₁₃H₂₀N₂O; FW: 220.30; white to yellow pwdr. Note: Sold in collaboration with Daicel for ī-Bu research purposes only. 07-2081 N-[(1S,2S)-2-Amino-1,2-diphenylethyl]-1,1,1-100mg trifluoromethanesulfonamide, 95%, (99% ee) (167316-28-1) NEW C₁₅H₁₅F₃N₂O₂S; FW: 344.40; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 07-2080 N-[(1R,2R)-2-Amino-1,2-diphenylethyl]-1,1,1-tri-100mg Ph Ph fluoromethanesulfonamide, 98%, (99% ee) NEW (852212-89-6) $C_{15}H_{15}F_3N_2O_2S$; FW: 344.40; white to yellow pwdr. H₂N CF₃ Note: Sold in collaboration with Daicel for research purposes only. 07-8033 (2S)-2-Amino-N,3,3-trimethyl-N-(phenylmethyl) 100mg 0 butanamide, 98%, (99% ee) (947383-62-2) NEW tBu C₁₄H₂₂N₂O; FW: 234.30; white to yellow pwdr. Ph N Note: Sold in collaboration with Daicel for research $\bar{N}H_2$ Me purposes only.

NITROGEN (Compounds) 07-8034 (2R)-2-Amino-N.3.3-trimethyl-N-(phenylmethyl) 100mg 0 ĭ butanamide, 98%, (99% ee) t-Bu NEW C₁₄H₂₂N₂O; FW: 234.30; white to yellow pwdr. Note: Sold in collaboration with Daicel for research NH₂ Me purposes only. 07-8410 N.N'-(S)-[1.1'-Binaphthalene]-2.2'-divlbis[N'-100ma CF₃ [3.5-bis(trifluoromethyl)phenyl]thiourea]. NEW 95%, (99% ee) (914497-25-9) C₃₈H₂₂F₁₂N₄S₂; FW: 826.70; white to yellow pwdr. CF₃ Note: Sold in collaboration with Daicel for CF₃ research purposes only. ĊF₃ 07-0750 2.2'-Bipvrazine, 95% (10199-00-5) 250mg C₈H₆N₄; FW: 158.16; light-brown solid 1a NEW air sensitive Note: Ligand for photocatalyst synthesis. Technical Notes: Ligand for the ruthenium- promoted catalytic water oxidation reaction. 1. 2 Ligand for the ruthenium promoted photocatalytic Diels-Alder cycloaddition. 3. Ligand for the ruthenium photocatalyzed intermolecular [3+2] cycloaddition of cyclopropylamines with olefins. 4. Ligand for the ruthenium mediated photocatalytic reaction for the preparation of N-arylindoles. 5. Endoperoxide synthesis by photocatalytic aerobic [2+2+2] cycloadditions. 6. [Ru(bpz)₃](PF₆)₂ catalyzed anti-Markovnikov hydrothiolation of olefins with a variety of thiols. 7. [Ru(bpz)₃](PF₆)₂ catalyzed [3+2] photooxygenation of aryl cyclopropanes. 8. $[Ru(bpz)_3](PF_6)_2$ catalyzed intermolecular [3 + 2] annulation of cyclopropylanilines with alkynes. Tech. Note (1) [Ru(tpy)(bpz)(OH₂)](PF₆)₂ O₂ + 4e⁻ = 4H⁺ 2H₂O Ref. (1) MeC MeC [Ru(bpz)3]2-Tech. Note (2) Ref. (2) Me Me Me Me R³ Tech. Note (3) [Ru(bpz)3](PF6)2 Ref. (3) Ŕ2 [Ru(bpz)₃](PF₆)₂ Tech. Note (4) NH Ref. (4) ÓB

53

NITROGEN (Compounds)



NITROGE	N (Compounds)	
07-6321 New	N-[(1S,2S)-2-[Bis(phenylmethyl)amino]cyclohexyl]-N'-[3,5- bis(trifluoromethyl)phenyl]thiourea, 98%, (99% ee) (1233369-39-5) $C_{29}H_{29}F_6N_3S$; FW: 565.60; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	50mg
07-7051 New	2,6-Bis[(4R)-4-tert-butyloxazolin-2-yl]pyridine, 98%, (99% ee) (185346-17-2) $C_{19}H_{2r}N_3O_2$; FW: 329.40; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. t-Bu t-Bu	100mg
07-7050 NEW	2,6-Bis[(4S)-4-tert-butyloxazolin-2-yl]pyridine, 98%, (99% ee) (118949-63-6) $C_{19}H_{zr}N_3O_2$; FW: 329.40; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	100mg
07-7130 New	$ \begin{array}{c} (\textbf{R})\textbf{-1-[3,5-Bis(trifluoromethyl)phenyl]-3-[1-(dimethylamino)-3-methylbutan-2-yl]thiourea, \\ \textbf{98\%}, (\textbf{99\%} ee) (1048692-61-0) \\ \textbf{C}_{16}H_{21}F_6N_3S; FW: 401.40; white to yellow pwdr. \\ Note: Sold in collaboration with Daicel for research purposes only. } \\ \end{array} $	50mg
07-7029 New	3-[[3,5-Bis(trifluoromethyl)phenyl]amino]-4- [[(8α ,9S)-10,11-dihydro-6'-methoxycinchon- an-9-yl]amino]-3-cyclobutene-1,2-dione, 98%, (99% ee) (1352957-59-5) C ₃₂ H ₃₀ F ₆ N ₄ O ₃ , FW: 632.60; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. CF ₃ H H H H CF ₃ H H H CF ₃ H H H CF ₃ H H H CF ₃ H H H H H H H H H H H H H	50mg
07-7028 New	3-[[3,5-Bis(trifluoromethyl)phenyl]amino]-4- [[(9R)-10,11-dihydro-6'-methoxycinchonan- 9-yl]amino]-3-cyclobutene-1,2-dione, 98%, (99% ee) (1407166-63-5) $C_{32}H_{30}F_{6}N_{4}O_{3}$; FW: 632.60; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. CF ₃ H H MeO N	50mg
07-7128 New	$\begin{array}{c} \textbf{3-[[3,5-Bis(trifluoromethyl)phenyl]ami-}\\ \textbf{no]-4-[[(1S,2S)-2-(dimethylamino)cyclohexyl]}\\ amino]-3-cyclohutene-1,2-dione, 95\%,\\ (\textbf{99\% ee)} (1263205-96-4)\\ C_{20}H_{21}F_8N_3O_2; FW: 449.40; white to yellow pwdr.\\ Note: Sold in collaboration with Daicel for research purposes only.\\ \end{array}$	100mg
07-7129 NEW	3-[[3,5-Bis(trifluoromethyl)phenyl]amino]-4-[[(1R,2R)-2-(dimethylamino) cyclohexyl]amino]-3-cyclobutene-1,2-dione, 98%, (99% ee) (1211565-07-9) C ₂₀ H ₂₁ F ₆ N ₃ O ₂ ; FW: 449.40; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	100mg
07-1025 New	3-[[3,5-Bis(trifluoromethyl)phenyl]ami - no]-4-[[(1R,2R)-2-(dimethylamino)-1,2-di- phenylethyl]amino]-3-cyclobutene-1,2-dione, 98%, (99% ee) ($1223105-89-2$) $C_{28}H_{23}F_6N_3O_2$; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	50mg

NITROGEN (Compounds) 07-8510 3-[[3,5-Bis(trifluoromethyl)phenyl]amino]-4-50mg CF₃ [[(1R,2R)-2-(dipentylamino)cyclohexyl]amino]-NEW 3-cyclobutene-1,2-dione, 98%, (99% ee) (1411983-40-8) C₂₈H₃₇F₆N₃O₂; FW: 561.60; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 07-8511 3-[[3,5-Bis(trifluoromethyl)phenyl]amino]-4-[[(1S,2S)-2-(dipentylamino) 50ma cyclohexyl]amino]-3-cyclobutene-1,2-dione, 98%, NEW (99% ee) (1411983-41-9) C₂₈H₃₇F₆N₃O₂; FW: 561.60; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 07-8436 3-[[3.5-Bis(trifluoromethyl)phenyl]amino]-4-50mg CF₃ [[(8a,9S)-6'-methoxycinchonan-9-yl]amino]-NEW 3-cyclobutene-1,2-dione, 95%, (99% ee) н ^{нм} CF₃ (1256245-84-7) N $C_{32}H_{28}F_{6}N_{4}O_{3}$; FW: 630.60; white to yellow pwdr. Note: Sold in collaboration with Daicel for research MaC purposes only. 07-8435 3-[[3,5-Bis(trifluoromethyl)phenyl]amino]-50mg ÇF₃ 4-[[(9R)-6'-methoxycinchonan-9-yl]amino]-NEW 3-cyclobutene-1,2-dione, 98%, (99% ee) HN (1256245-79-0) CE_{2} C₃₂H₂₈F₆N₄O₃; FW: 630.60; white to yellow pwdr. Note: Sold in collaboration with Daicel for research MeC purposes only. 07-5610 3-[[3,5-Bis(trifluoromethyl)phenyl]amino]-4-50mg CF₃ [[(1R,2R)-2-(1-piperidinyl)cyclohexyl]amino]-3-NEW cyclobutene-1,2-dione, 98% (1211565-11-5) C23H25F6N3O2; FW: 489.50; orange pwdr. CF₃ Note: Sold in collaboration with Daicel for research purposes only. Н 07-5611 3-[[3,5-Bis(trifluoromethyl)phenyl]amino]-4-[[(1S,2S)-2-(1-pyrrolidinyl) 50mg cyclohexyl]amino]-3-cyclobutene-1,2-dione, 98%, (99% ee) (1346683-42-8) NEW C₂₂H₂₃F₆N₃O₂; FW: 475.40; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 07-0630 (2R)-2-[[[[3,5-Bis(trifluoromethyl)phenyl] 50mg CF₃ amino]thioxomethyl]amino]-3,3-dimethyl-N-NEW (phenylmethyl)butanamide, 98%, (99% ee) C₂₂H₂₃F₆N₃OS; FW: 491.50; white to yellow pwdr. ī-Bu S Note: Sold in collaboration with Daicel for CF₃ research purposes only. 07-0631 (2S)-2-[[[[3,5-Bis(trifluoromethyl)phenyl]amino]thioxomethyl] 50mg amino]-3,3-dimethyl-N-(phenylmethyl)butanamide, 98%, (99% ee) NEW (1490388-03-8) C₂₂H₂₃F₆N₃OS; FW: 491.50; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 07-6300 N-[3,5-Bis(trifluoromethyl)phenyl]-N'-[(1S,2R)-100mg 2,3-dihydro-2-hydroxy-1H-inden-1-yl]thiourea, F₃C NEW 95%, (99% ee) (949480-57-3) C₁₈H₁₄F₆N₂OS; FW: 420.4; white to yellow pwdr. Note: Sold in collaboration with Daicel for research ĊF₃ purposes only.



57

NITROGEN (Compounds)

MINCOL		
07-0912 New	$\begin{array}{c} \textbf{N-[3,5-Bis(trifluoromethyl)}\\ \textbf{phenyl]-N'-[(2S)-2-pyrrolidinylmethyl]}\\ \textbf{thiourea, 98\% (904928-30-9)}\\ C_{r_4}H_{r_5}F_6N_3S; FW: 371.30; white to yellow pwdr.\\ Note: Sold in collaboration with Daicel for research purposes only.\\ \hline CF_3 \end{array}$	100mg
07-7173 New	2-[(4R)-4-tert-Butyl-4,5-dihydro-2-oxazolyl]- 5-(trifluoromethyl)pyridine, 98%, (99% ee) (1428537-19-2) $C_{13}H_{15}F_{3}N_{2}O;$ FW: 272.30; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. F _3 C	50mg
07-7124 New	$\begin{array}{l} \textbf{2-[(4S)-4-tert-Butyl-4,5-dihydro-2-oxazolyl]-5-(trifluoromethyl)pyridine, 98\%\\ \textbf{(99\% ee)} \ (1416819-91-4)\\ C_{13}H_{15}F_{3}N_{2}O; \ FW: 272.30; \ \text{white to yellow pwdr.}\\ \text{Note: Sold in collaboration with Daicel for research purposes only.} \end{array}$, 100mg
07-6312 New	N,N'-(1R,2R)-1,2-Cyclohexanediyl- bis[N'-[3,5-bis(trifluoromethyl)phenyl] thiourea], 98%, (99% ee) (743458-79-9) $C_{24}H_{20}F_{12}N_4S_{21}$; FW: 656.6; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. NH CF ₃ NH CF ₃ CF ₃	100mg
07-1813 NEW	2,6-Dichloro-1-fluoropyridinium triflate, 95% (<i>130433-68-0</i>) C ₈ H ₃ Cl ₂ F ₄ NO ₃ S; FW: 316.06; white to light-yellow pwdr. <i>air sensitive, moisture sensitive, (store cold)</i>	500mg 2g
07-1280 New	2-(2,4-Difluorophenyl)-5-methylpyridine, 95% (583052-21-5) C ₁₂ H ₉ F ₂ N; FW: 205.20; white solid <i>air sensitive</i> Note: Ligand for photocatalyst synthesis.	500mg 2g
07-1923 NEW HAZ Technical Notes: 1. Palladium 2. Copper-c 3. Synthesis 4. Iridium-ca 5. Palladium 6. A Copper- pyrrolo[2,	4,7-Dimethoxy-1,10-phenanthroline, 98% ($92149-07-0$) $C_{14}H_{10}N_2O_2$; FW: 238.24; white to off-white pwdr.; m.p. 210-212°; d. 1.25 <i>air sensitive</i> Note: Ligand for photocatalyst synthesis. catalyzed synthesis of benzofurans and coumarins from phenols and olefins. atalyzed benzylic C(sp3)-H alkoxylation of heterocyclic compounds. of amides via copper-catalyzed amidation of aryl halides using isocyanides. talyzed silylation of aryl C-H bonds. catalyzed intramolecular cyclization of nitroalkenes: synthesis of thienopyrroles. -catalyzed N-alkynylation route to 2-substitued N-alkynyl pyrroles and their cyclization-1-ones	250mg 1g ation into
R'+ OH	Pd(OAc) ₂ , 1-10 phenanthroline Cu(OAc) ₂ . NaOAc air, CICH ₂ CH ₂ CI, 110°C	Tech. Note (1) Ref. (1)

+ R1€

0

NITROGEN (Compounds)



NITROGEN	I (Compounds)	
07-2063 New	1-[(1R,2R)-2-(Dimethylamino)cyclohexyl]-3- (perfluorophenyl)thiourea, 98%, (99% ee) $C_{15}H_{18}F_{5}N_{3}S$; FW: 367.40; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	100mg
07-2064 NEW	1-[(1S,2S)-2-(Dimethylamino)cyclohexyl]-3-(perfluorophenyl) thiourea, 98%, (99% ee) $C_{15}H_{18}F_5N_3S$; FW: 367.40; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	100mg
07-8020 NEW	N,N'-[(1S,2S)-1,2-Diphenyl-1,2-ethanediyl]bis[N'-[3,5- bis(trifluoromethyl)phenyl]thiourea], 95%, (99% ee) (1416334-72-9) $C_{32}H_{22}F_{12}N_4S_2$; FW: 754.70; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	100mg
07-0821 New	N,N'-[(1R,2R)-1,2-Diphenyl-1,2-ethanediyl] bis[N'-[3,5-bis(trifluoromethyl)phenyl] thiourea], 98%, (99% ee) (1012051-90-9) $C_{32}H_{22}F_{12}N_4S_2$; FW: 754.70; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. Ph Ph Ph F ₃ C S N N N S F ₃ C CF ₃	100mg
07-1828 NEW	1-Fluoropyridinium triflate, 95% (107263-95-6) C ₆ H ₅ F ₄ NO ₃ S; FW: 247.17; white to pale-yellow pwdr. <i>air sensitive, moisture sensitive, (store cold)</i>	500mg 2g
07-1835 NEW	1-Fluoro-2,4,6-trimethylpyridinium triflate (107264-00-6) C ₉ H ₁₁ F ₄ NO ₃ S; FW: 289.25; white to pale-yellow pwdr. <i>air sensitive, moisture sensitive, (store cold)</i>	500mg 2g
07-0984 New	(2S)-N-[(1R,2S)-2-Hydroxy-1,2-diphenyleth- yl]-2-pyrrolidinecarboxamide, 98%, (99% ee) (529486-23-5) $C_{19}H_{22}N_2O_2$; FW: 310.40; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	100mg
07-0983 New	$\begin{array}{c} \textbf{(2S)-N-[(1S,2R)-2-Hydroxy-1,2-diphenyleth-}\\ \textbf{y]]-2-pyrrolidinecarboxamide, 98%, (99% ee)}\\ (529486-25-7)\\ C_{19}H_{22}N_2O_2, \ FW: \ 310.40; \ whie \ to \ yellow \ pwdr.\\ Note: \ Sold \ in \ collaboration \ with \ Daicel \ for \ research \\ purposes \ only. \end{array} \qquad \begin{array}{c} H \\ N \\ H \\ O \\ H \\ O \\ H \\ O \\ H \end{array}$	100mg
07-1040 New	$\begin{array}{c} \textbf{2-[(4R)-4-lsobutyl-4,5-dihydro-2-oxaz-oyl]-5-(trifluoromethyl)pyridine, 98%, (99% ee) \\ C_{1_3}H_{1_5}F_{3}N_2O; FW: 272.30; white to yellow pwdr. \\ Note: Sold in collaboration with Daicel for research purposes only. \\ \end{array} \qquad \begin{array}{c} F_3C \\ N \\ I \\ I$	50mg
07-7058 New	$\begin{array}{c} \textbf{2-[(4R)-4-IsopropyI-4,5-dihydro-2-oxazolyI]-5-(tri-fluoromethyI)pyridine, 98%, (99% ee) \\ C_{12}H_{13}F_{3}N_{2}O; FW: 258.20; white to yellow pwdr. \\ Note: Sold in collaboration with Daicel for research purposes only. \\ \end{array} \qquad \begin{array}{c} F_{3}C \\ N \\ I \\ N \\ I \\ I \\ I \\ I \\ I \\ I \\ I$	100mg
07-1035 NEW	(3aR,3'aR,8aS,8'aS)-2,2'-(1-Methylethylidene) bis[3a,8a-dihydro-8H-indeno[1,2-d]oxazole], 95%, (99% ee) ($189623-45-8$) C ₂₃ H ₂₂ N ₂₀ O ₂ ; FW: 358.40; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	100mg

NITROGEN	N (Compounds)		
07-1034 New	(3aS,3'aS,8aR,8'aR)-2,2'-(1-Methylethylidene) bis[3a,8a-dihydro-8H-indeno[1,2-d]oxazole], 98%, (99% ee) (175166-51-5) C ₂₃ H ₂₂ N ₂ O ₂ ; FW: 358.40; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	Me Me	100mg
07-7105 (NEW)	(4S,4'S)-2,2'-[2-Phenyl-1-(phenylmethyl) ethylidene]bis[4-(1-methylethyl)-4,5- dihydrooxazole], 95%, (99% ee) ($444575-98-8$) $C_{27}H_{34}N_2O_2$; FW: 418.60; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	Ph Ph O ///O ///O iPr iPr	50mg
07-1780 New	2-Phenylpyridine , 95% (1008-89-5) $C_{11}H_{9}N$; FW: 155.20; amber liquid; b.p. 268-270°; f.p. 230°; d. 1.086 <i>air sensitive</i> Note: Ligand for photocatalyst synthesis.	N Ph	1g
07-0905 New	3-[[(1\$,2\$)-2-(1-Piperidinyl)cyclohexyl] amino]-4-[[4-(trifluoromethyl)phenyl]amino]- 3-cyclobutene-1,2-dione, 95%, (99% ee) (1312991-08-4) C ₂₂ H ₂₈ F ₃ N ₃ O ₂ ; FW: 421.50; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.		50mg
07-2625 New	2-[4-(Trifluoromethyl)phenyl]pyridine , 95% (203065-88-7) $C_{12}H_8F_3N$; FW: 223.19; white to yellow solid <i>air sensitive</i> Note: Ligand for photocatalyst synthesis.	F ₃ C	1g
07-1840 New	$\begin{array}{l} \textbf{2-[(Trifluoromethyl)thio]-1,1-dioxide-1,2-benzisothiazol-3(2H)-one, 97% (1647073-46-8)\\ C_8H_4F_3NO_3S_2; FW: 283.25; white solid air sensitive, moisture sensitive \end{array}$	O, O N-SCF ₃ O	250mg 1g



PALLADIUM (Elemental Forms)

46-1660

NEW

Palladium on carbon - 1 wt % loading, activated synthetic carbon pellet (7440-05-3) Pd; FW: 106.42; black pellet; SA: Pd 130 m²/g; P.Vol. 1.2 cm³/g

Technical Notes:

The enhanced dispersion of these Pd particles on the high 1 purity carbon support (in both powder and pellet forms) enables the catalysts' operation under mild conditions (from RT to 60 °C) with higher selectivity and minimum unwanted side reactions.

One of the key differences of the Pd/C catalyst in pellet form



(46-1660) is its more accessible surface area. This product has macroporosity (10 micron) and mesoporosity (6 nm) which makes carbon considerably lower in density (0.27 g/mL versus 0.5 -0.8 g/mL) and lighter than other carbon supports. This macroporosity allows palladium to be distributed throughout the carbon support instead of just on the outside. As a result, the metal surface area is much greater (130 m²/g vs 20 m²/g) and has a much smaller particle size (3.5 nm vs 40 nm).

46-1610	Palladium on carbon - 1 wt % loading, activated synthetic carbon powder	10g
NEW	(7440-05-3) Pd: EW/: 106.42: black pwdr : SA: Pd 150 m2/a: PV/ol. 0.35 cm3/a	50g
	1 d, 1 W. 100.42, black pwdi., 3A. 1 d 130 m/g, 1. Vol. 0.33 cm/g	
46-1630	Palladium on carbon - 5 wt % loading, activated synthetic carbon powder	5g
	(7440-05-3)	25g
NEW	Pd; black pwdr.; SA: Pd 70 m ² /g; P.Vol. 0.35 cm ³ /g	-
DALLADIUM (Compoundo)		

compounds)

46-0700 NEW	Bis(acetonitrile)palladium(II) p-tolu C ₁₈ H ₂₀ N ₂ O ₆ S ₂ Pd; FW: 530.91; yellow <i>air sensitive, (store cold)</i>	enesulfonate, 98% (114 pwdr.	757-66-3)	50mg 250mg
46-0241 NEW	Bis{[2-(Diadamantylphosphi- no)-3-methoxy-2',4',6'-tri-i-pro- pyl-3'-(2,3,5,6-tetrafluoro-4-bu- tylphenyl)-1,1'-biphenyl] palladium(0)}1,5-cyclooctadiene, [AIPhos Palladium complex] ($1805783-51-0$) C ₁₁₂ H ₁₄₆ F ₈ O ₂ P ₂ Pd ₂ ; FW: 1951.13; yellow-green solid Note: Patents: US 6,395,916, US 6,307,087	iPr Ad ₂ P. iPr F F Bu	nBu F F F iPr PAd2 iPr MeO	50mg 250mg

Technical Notes:

Ligand for the Palladium-Catalyzed Fluorination of Five-Membered Heteroaryl Bromides. 1.

2. Ligand for the Palladium-Catalyzed Fluorination of Aryl Triflates and Bromides.



10q 50q

PALLADIUM (Compounds) 46-0308 BisI(trimethylsilyl)methyl](1,5-cy-250mg SiMe₃ clooctadiene)palladium(II), 98% 1g NEW (225931-80-6) C₁₆H₃₄PdSi₂; FW: 389.03; gray pwdr. SiMe₃ air sensitive, (store cold) 96-5506 Buchwald Palladacvcle Precatalvst Kit 2b (Methanesulfonato-2'-amino-1.1'-biphenvl-2-vl- Palladacvcles Gen. 3) See page 105 46-0347 Methanesulfonato(2-bis(3,5-di(tri-100mg Me₂N OMe fluoromethyl)phenylphosphi-500mg NEW no)-3,6-dimethoxy-2',6'-bis(di-CF₃ methylamino)-1,1'-biphenyl) (2'-methylamino-1,1'-biphenyl-2-yl) PR₂ Me₂N OMe R palladium(II) [Palladacycle Gen. 4] (H)Me-N-Pd--OMs (1810068-35-9) CFa C48H44F12N3O5PPdS; FW: 1140.32; yellow to orange solid Note: Patents: PCT/US2013/030779, US Serial No. 13/799620 46-0935 Methanesulfonato{N-[2-(di-1-250mg adamantylphosphino)phenyl]morpholine} 1g NEW (2'-amino-1,1'-biphenyl-2-yl)palladium(II) dichloromethane adduct, min. 98% [Mor-PAd₂ Dalphos Palladacycle Gen. 3] C43H55N2O4PPdS; FW: 833.37; Beige to brown solid H₂N-Rd-OMs Note: Patents: PCT/US2013/030779, US Serial No. 13/799620 Technical Note: Palladium catalyst for ammonia arylation 1. 2. Palladium catalyst for multicomponent one-pot synthesis of indoles. 3. Palladium catalyst for primary aliphatic amination of aryl mesylates. 4. Palladium catalyst for ketone mono- α -arylation of aryl mesylates. NH₂ Tech. Note (1) Ref. (1) X = CI or OTsR= electron-donating or withdrawing group MorDalPhos/Pd (cat.) H₂NR¹ Tech. Note (2) CS₂CO₃ Ref. (2) R1 toluene, 90° OMs MorDalPhos/Pd (cat.) Tech. Note (3) tBuOH/1.4-dioxane Ref. (3)

K₃PO₄ 110°. 16-20 h

PALLADIUM (Compounds)



PHOSPHORUS (Compounds)



4-(Anthracen-9-yl)-3-(t-butyl-2,3-dihydrobenzo[d][1,3]oxaphosphole, 98+% rac-AntPhos (1268693-24-8) C₂₅H₂₃O₃P; FW: 370.42; pale yellow pwdr. *air sensitive, (store cold)* Note: Sold in collaboration with Zejun for research purposes only. Patents ZL201310020371.1, CN 201610056390.



25mg 100mg 500mg

Technical Notes:

- 1. Ligand/palladium catalyst for general Miyaura borylation reactions.
- 2. Ligand/palladium catalyst for general and sterically demanding Suzuki-Miyaura cross-coupling reactions.
- 3. Ligand/palladium catalyst for aryl-alkyl Suzuki-Miyaura cross-coupling reactions.
- 4. Ligand/nickel catalyst for intramolecular reductive cyclization.
- 5. Ligand/palladium catalyst for Dearomative cyclization.



PHOSPHC	RUS (Compounds)	
15-1967 NEW	(S)-4-(Anthracen-9-yl)-3-(t-butyl-2,3-dihydrobenzo[d][1,3] oxaphosphole,99+% (>99% ee) [(S)-AntPhos] (1807740-34-6) C ₂₅ H ₂₃ O ₃ P; FW: 370.42; light yellow xtl. <i>air sensitive,</i> (store cold) Note: Sold in collaboration with Zejun for research purposes only. Patents ZL201310020371.1, CN 201610056390.	25mg 100mg 500mg
1. Ligand fo	or the enantioselective nickel-catalyzed intramolecular reductive cyclization of alky	nones.
	R'	
O O O N R''	$\begin{array}{c c} Ni, P^* \\ \hline Et_3SiH \end{array} & O \\ \hline n \\ R'' \end{array} OSiEt_3 \\ \hline R'' \end{array}$	Tech. Note (1) Ref. (1)
References:	Chem Int Ed. 2015 54 2520	
96-0650	BI-DIME Ligand Kit See page 102	
15-0293 New	(11aR)-3,7-Bis([1,1'-biphenyl]-4-yl)-10,11,12,13- tetrahydro-5-hydroxy-diindeno[7,1-de:1',7'- fg][1,3,2]dioxaphosphocin, 98%, (99% ee) (1297613-77-4) C ₄₁ H ₃₁ O ₄ P; FW: 618.70; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	25mg
15-0294 New	(11aS)-3,7-Bis([1,1'-biphenyl]-4-yl)-10,11,12,13-tetrahydro-5-hydroxy- diindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosphocin, 98%, (99% ee) (1297613-77-4) $C_{41}H_{31}O_4P$; FW: 618.67; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	25mg
15-1366	(11bR)-2,6-Bis[3,5-bis(trifluoromethyl)phenyl]- CF3	100mg
NEW	4-hydroxy-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2] dioxaphosphepin, 98%, (99% ee) (791616-62-1) $C_{36}H_{17}F_{12}O_4P$; FW: 772.5; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	
15-1367 New	(11bS)-2,6-Bis[3,5-bis(trifluoromethyl)phenyl]-4-hydroxy-4-oxide- dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin, 98%, (99% ee) (878111-17-2)	100mg

 $C_{36}H_{17}F_{12}O_4P$; FW: 772.5; White to light-yellow pwdr.

Note: Sold in collaboration with Daicel for research purposes only.

PHOSPHO	DRUS (Compounds)	
15-1376 (NEW)	(11bR)-2,6-Bis[3,5-bis(trifluoromethyl)phenyl]- 8,9,10,11,12,13,14,15-octahydro-4-hydroxy-4- oxide-dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphos- phepin, 98%, (99% ee) (1011465-24-9) $C_{36}H_{25}F_{12}O_4P$; FW: 780.5; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. C_{F_3}	25mg 100mg
15-1377 New	(11bS)-2,6-Bis[3,5-bis(trifluoromethyl)phenyl]-8,9,10,11,12,13,14,15- octahydro-4-hydroxy-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2]dioxa- phosphepin, 98%, (99% ee) $C_{36}H_{25}F_{12}O_4P$; FW: 780.5; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	25mg 100mg
15-0276 New	(11aR)-3,7-Bis(4-chlorophenyl)-10,11,12,13-tetra- hydro-5-hydroxy-diindeno[7,1-de:1',7'-fg][1,3,2] dioxaphosphocin, 98%, (99% ee) $C_{29}H_{21}Cl_2O_4P$; FW: 535.36; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	50mg
15-0275 New	(11aS)-3,7-Bis(4-chlorophenyl)-10,11,12,13-tetrahydro-5-hydroxy- diindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosphocin, 98%, (99% ee) (1258327-05-7) C ₂₉ H ₂₁ Cl ₂ O ₄ P; FW: 535.36; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	50mg
15-3280 (NEW)	(11bR)-2,6-Bis[4-(1,1-dimethylethyl)phenyl]-4- hydroxy-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2] dioxaphosphepin, 98%, (99% ee) (861909-30-0) C ₄₀ H ₃₇ O ₄ P; FW: 612.71; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	50mg
15-0339 NEW	(11aR)-3,7-Bis((4-(1,1-dimethylethyl)phenyl)- 10,11,12,13-tetrahydro-5-hydroxy-diindeno[7,1- de:1',7'-fg][1,3,2]dioxaphosphocin, 98%, (99% ee) $C_{37}H_{39}O_4P$; FW: 578.7; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	25mg
15-0338 NEW	(11aS)-3,7-Bis((4-(1,1-dimethylethyl)phenyl)-10,11,12,13-tetrahydro-5-hy- droxy-diindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosphocin, 98%, (99% ee) C ₃₇ H ₃₉ O ₄ P; FW: 578.70; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	25mg

PHOSPHORUS (Compounds) 15-1368 (11bR)-2,6-Bis(3,5-dimethylphenyl)-4-hydroxy-4-100mg CH₃ oxide-dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphos-NEW phepin, 98%, (99% ee) (861909-53-7) $C_{36}H_{20}O_4P$; FW: 556.6; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research P^{≈0} 0. purposes only. 0 ÒН CH₃ ĊH₃ 15-1369 (11bS)-2,6-Bis(3,5-dimethylphenyl)-4-hydroxy-4-oxide-di-100mg naphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin, 98%, (99% ee) NEW (1170736-59-0) C₃₆H₂₀O₄P; FW: 556.6; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 15-1373 (11bR)-2,6-Bis(3,5-dimethylphenyl)-25mg CH₃ 8,9,10,11,12,13,14,15-octahydro-4-hydroxy-4-ox-100mg NEW ide-dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin, 98%, (99% ee) (1065214-95-0) CH_3 C₃₆H₃₇O₄P; FW: 564.7; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research n O, ОH purposes only. CH₃ ĊH₃ 15-1374 (11bS)-2,6-Bis(3,5-dimethylphenyl)-8,9,10,11,12,13,14,15-octahydro-25mg 4-hydroxy-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin, 100mg NEW 98%, (99% ee) C₃₆H₃₇O₄P; FW: 564.7; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 15-0374 Bis(pentafluorophenyl)phenylphosphine, 97% 250ma (5074 - 71 - 5)1q NEW C₁₈H₅F₁₀P; FW: 442.20; white pwdr.; m.p. 59-61°; b.p. 105°C/0.3mm; f.p. >110°C air sensitive 15-0277 (11aR)-3,7-Bis(1-pyrenyl)-10,11,12,13-tetrahydro-5-hy-25mg droxy-diindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosphocin, NEW 98%, (99% ee) $C_{49}H_{31}O_4P$; FW: 714.7; white to yellow pwdr. Note: Sold in collaboration with Daicel for research 0. 0 purposes only. OH 15-0278 (11aS)-3,7-Bis(1-pyrenyl)-10,11,12,13-tetrahydro-5-hydroxy-di-25ma indeno[7,1-de:1',7'-fg][1,3,2]dioxaphosphocin, 98%, (99% ee) NEW $C_{49}H_{31}O_4P$; FW: 714.7; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.
PHOSPHO	RUS (Compounds)	
15-0638 New	$\begin{array}{c} \textbf{Bis(p-sulfonatophenyl)phenylphos-}\\ \textbf{phine dihydrate dipotassium salt, 97\%}\\ (308103-66-4)\\ \textbf{C}_{18}\textbf{H}_{13}\textbf{K}_{2}\textbf{O}_{6}\textbf{PS}_{2}; FW: 498.60(534.63); \text{ white}\\ \textbf{pwdr.; m.p. 98-102^{\circ}}\\ \textbf{air sensitive} \end{array} \qquad $	100mg 500mg
15-1395 New	(R)-3,3'-Bis2,4,6-triisopropyl- phenyl)-5,5',6,6',7,7',8,8'-octahydro-1,1'-binaph- thyl-2,2'-diyl Hydrogenphosphate, 98%, (99% ee) (929294-27-9) $C_{so}H_{es}O_{4}P$; FW: 761; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	25mg 100mg
15-1394 New	(S)-3,3'-Bis2,4,6-triisopropylphenyl)-5,5',6,6',7,7',8,8'-octahydro- 1,1'-binaphthyl-2,2'-diyl Hydrogenphosphate, 98%, (99% ee) (878111-20-7) C ₅₀ H ₆₅ O ₄ P; FW: 761; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	25mg 100mg
96-5495	Buchwald Biaryl Phosphine Ligand Mini Kit 3 for Aromatic Carbon-Heteroatom Bond Formation, Suzuki Coupling and Negishi Cross-coupling See page 103	
15-6205 NEW	3-(t-Butyl)-4-(2,6-dimethoxyphenyl)-2,3- dihydrobenzo[d][1,3]oxaphosphole, min. 97% rac-BI-DIME (<i>1246888-90-3</i>) C ₁₉ H ₂₃ O ₃ P; FW: 330.36; light-yellow xtl. Note: Sold in collaboration with Zejun for research purposes only. Patents: ZL2013105048267, CN104558038	25mg 100mg 500mg
Technical Matte		

1. Ligand/palladium catalyst for general and sterically demanding Suzuki-Miyaura cross-coupling reactions

- 2. Ligand/palladium catalyst for sterically demanding Buchwald-Hartwig amination
- 3. Ligand/palladium catalyst for asymmetric Suzuki-Miyaura cross-coupling reactions.
- 4. Ligand/nickel catalyst for asymmetric intramolecular reductive cyclization
- 5. Ligand/rhodium catalyst for asymmetric hydroboration



PHOSPHORUS (Compounds)







PHOSPHO	RUS (Compounds)	
15-0870 New	$(11bS)-N-((R)-2,3-dihydro-1H-inden-1-yl)-N-(Di-n-butyl methyl)dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin-4-amine(2019254-27-2)C_{38}H_{40}NO_2P; FW: 573.7Note: Patent: UK1710941.4P-N$	
15-1720 New	$\begin{array}{c} \textbf{1-(2-Di-t-butylphosphinophenyl)-3,5-diphe-}\\ \textbf{nyl-1H-pyrazole, 98\% (628333-86-8)}\\ C_{29}H_{33}N_2P; FW: 440.56; white xtl.; m.p. 138-142^{\circ}\\ air sensitive \\ \end{array} \qquad \qquad$	250mg 1g
15-0837 (New)	(11bS)-2,6-Di-9-phenanthrenyl-4-hydroxy-4- oxide-dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphos- phepin, 98% (1043567-32-3) $C_{a8}H_{29}O_{4}P$; FW: 700.71; red brown pwdr. Note: Sold in collaboration with Daicel for research purposes only.	50mg
15-1380 NEW	$\begin{array}{c} \textbf{DiphenyI[4-(N,N-dimethylamino)phenyI]}\\ \textbf{phosphine, min. 95\%} (739-58-2)\\ C_{20}H_{20}NP; FW: 305.35; white solid; m.p. 151-154^\circ\\ air sensitive \end{array} \begin{array}{c} Ph \\ P \\ Ph \end{array} \begin{array}{c} \textbf{Me} \\ \textbf{Ph} \\ \textbf{Me} \end{array}$	1g 5g
15-0860 New	$(11bS)-N-(Diphenylmethyl)-N-isopropyldinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin-4-amine (1637749-69-9)C_{38}H_{30}NO_2P; FW: 539.6Note: Patents: US20160074852, EP2986375$	
15-0364 New	(11bR)-2,6-Di-1-pyrenyl-4-hydroxy-4-oxide- dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin, 98%, (99% ee) C _{s2} H ₃₁ O ₄ P; FW: 748.80; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	50mg
15-0365 NEW Technical Note: 1. See 15-0	(11bS)-2,6-Di-1-pyrenyl-4-hydroxy-4-oxide-dinaphtho[2,1-d:1',2'-f] [1,3,2]dioxaphosphepin, 98%, (99% ee) C ₅₂ H ₃₁ O ₄ P; FW: 748.8; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. I364 (page 74)	50mg





PHOSPHORUS (Compounds) 15-0820 (11bR)-8,9,10,11,12,13,14,15-Octahydro-4-hydroxy-50mg SiPh₃ 2,6-bis(triphenylsilyl)-4-oxide-dinaphtho[2,1-NEW d:1',2'-f][1,3,2]dioxaphosphepin, 98%, (99% ee) °. P^{__}€ 0. (957790-94-2) ЮH C₅₆H₄₉O₄PSi₂; FW: 873.13; white to yellow pwdr.; m.p. >264° Note: Sold in collaboration with Daicel for research SiPh₃ purposes only. 50mg 15-0821 (11bS)-8,9,10,11,12,13,14,15-Octahydro-4-hydroxy-2,6bis(triphenylsilyl)-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2] NEW dioxaphosphepin, 98%, (99% ee) (1157989-25-7) C₅₆H₄₉O₄PSi₂; FW: 873.13; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 15-1383 (11bR)-8,9,10,11,12,13,14,15-Octahydro-4-hydroxy-25ma 2,6-di-1-naphthalenyl-4-oxide-dinaphtho[2,1-100ma NEW d:1',2'-f][1,3,2]dioxaphosphepin, 98%, (99% ee) (1242066-20-1) C₄₀H₂₂O₄P; FW: 608.7; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research , ₽.́ purposes only. 0´ ЮH 15-1384 (11bS)-8,9,10,11,12,13,14,15-Octahydro-4-hydroxy-2,6-25mg di-1-naphthalenyl-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2] 100mg NEW dioxaphosphepin, 98%, (99% ee) C₄₀H₃₃O₄P; FW: 608.7; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 15-1378 (11bR)-8.9.10.11.12.13.14.15-Octahvdro-4-hvdroxy-25ma 2.6-di-2-naphthalenyl-4-oxide-dinaphtho[2,1-100mg NEW d:1',2'-f][1,3,2]dioxaphosphepin, 98%, (99% ee) (922711-75-9) o C₄₀H₃₃O₄P; FW: 608.7; White to light-yellow pwdr. \cap ЮΗ Note: Sold in collaboration with Daicel for research purposes only. 15-1379 (11bS)-8,9,10,11,12,13,14,15-Octahydro-4-hydroxy-2,6-25mg di-2-naphthalenyl-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2] 100ma NEW dioxaphosphepin, 98%, (99% ee) C₄₀H₃₃O₄P; FW: 608.7; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. 15-1396 (11bR)-8,9,10,11,12,13,14,15-Octahydro-4-100mg Ph hydroxy-2,6-diphenyl-4-oxide-dinaphtho[2,1-NEW d:1',2'-f][1,3,2]dioxaphosphepin, 98%, (99% ee) 0 0 (791616-65-4) OH C₃₂H₂₉O₄P; FW: 508.5; White to light-vellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. Ph (11bS)-8,9,10,11,12,13,14,15-Octahydro-4-hydroxy-2,6-diphenyl-4-15-1397 100ma oxide-dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin, 98%, (99% ee) NEW (945852-48-2) $C_{32}H_{29}O_4P$; FW: 508.5; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.

DRUS (Compounds)		
(11bR)-8,9,10,11,12,13,14,15-Octahydro-4- hydroxy-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2] dioxaphosphepin, 98%, (99% ee) (297752-25-1) $C_{20}H_{21}O_4P$; FW: 356.4; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	O POH	100mg
(11bS)-8,9,10,11,12,13,14,15-Octahydro-4-hydroxy dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin, 9 (1193697-61-8) C ₂₀ H ₂₁ O ₄ P; FW: 356.4; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research p	y-4-oxide- 8%, (99% ee) purposes only.	100mg
(11aR)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-di- 1-naphthalenyl-5-oxide-diindeno[7,1-de:1',7'- fg][1,3,2]dioxaphosphocin, 98%, (99% ee) (1297613-73-0) $C_{37}H_{27}O_4P$; FW: 566.59; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	O O O POH	50mg
(11aS)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-di-1- oxide-diindeno[7,1-de:1',7'-fg] [1,3,2]dioxaphosp (1258327-08-0) C ₃₇ H ₂₇ O ₄ P; FW: 566.59; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research p	-naphthalenyl-5- hocin 98% (99% ee) purposes only.	50mg
(11aR)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-di- 2-naphthalenyl-5-oxide-diindeno[7,1-de:1',7'- fg] [1,3,2]dioxaphosphocin, 98% (99% ee) (1297613-74-1) $C_{37}H_{27}O_4P$; FW: 566.59; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	O POH	50mg
(11aS)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-di-2- oxide-diindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosph $C_{37}H_{27}O_4P$; FW: 566.59; white to yellow pwdr. Note: Sold in collaboration with Daicel for research p	naphthalenyl-5- nocin, 98%, (99% ee) purposes only.	50mg
(11aR)-10,11,12,13-Tetrahydro-5-hydroxy-3,7- diphenyl-diindeno[7,1-de:1',7'-fg] [1,3,2]dioxa- phosphocin, 98% (99% ee) ($1297613-72-9$) $C_{29}H_{23}O_4P$; FW: 466.46; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	о-Р-он	50mg
(11aS)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-diph de:1',7'-fg] [1,3,2]dioxaphosphocin, 98% (99% ee $C_{29}H_{23}O_4P$; FW: 446.46; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research p	tenyl-diindeno[7,1-) (1258327-04-6) purposes only.	50mg
	(11bR)-8,9,10,11,12,13,14,15-Octahydro-4- hydroxy-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2] dioxaphosphepin, 98%, (99% ee) (297752-25-1) $C_{20}H_{21}O_4P$; FW: 356.4; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. (11bS)-8,9,10,11,12,13,14,15-Octahydro-4-hydroxy dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin, 9 (1193697-61-8) $C_{20}H_{21}O_4P$; FW: 356.4; White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research p (11aR)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-di-1 -naphthalenyl-5-oxide-diindeno[7,1-de:1',7'- fg][1,3,2]dioxaphosphocin, 98%, (99% ee) (1297613-73-0) $C_{37}H_{27}O_4P$; FW: 566.59; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. (11aS)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-di-1 oxide-diindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosp (1258327-08-0) $C_{37}H_{27}O_4P$; FW: 566.59; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research p (11aR)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-di- 2-naphthalenyl-5-oxide-diindeno[7,1-de:1',7'- fg][1,3,2]dioxaphosphocin, 98% (99% ee) (1297613-74-1) $C_{37}H_{27}O_4P$; FW: 566.59; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. (11aS)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-di-2 oxide-diindeno[7,1-de:1',7'-fg][1,3,2]dioxaphospf $C_{37}H_{27}O_4P$; FW: 566.59; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. (11aS)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-di- diphenyl-diindeno[7,1-de:1',7'-fg][1,3,2]dioxa- phosphocin, 98% (99% ee) (1297613-72-9) $C_{29}H_{29}O_4P$; FW: 566.59; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. (11aS)-10,11,12,13-Tetrahydro-5-hydroxy-3,7-diph de:1',7'-fg][1,3,2]dioxaphosphocin, 98% (99% ee $C_{29}H_{29}O_4P$; FW: 466.46; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only.	$\begin{aligned} & (118R) = 0, (11,12,13,14,15-Octahydro-4-hydroxy-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2]\\ & \mathsf{(dxaphosphepin, 99%, (99% ee) (297752-25-1)\\ & (Jachs,0,P, FW: 356,4: White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. \\ \\ & (\mathsf{(118R) = 0, (11,12,13,14,15-Octahydro-4-hydroxy-4-oxide-dinaphtho[2,1-d:1',2'-f][1,3,2]dioxaphosphepin, 98%, (99% ee) (1793697-61-8)\\ & \mathsf{(C_{a, H_{a}, 0, P, FW: 356,4: White to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. \\ \\ & ((118R) = 0, (11,12,13) = Tetrahydro-5-hydroxy-3,7-di-1-naphthalenyl-5-oxide-dindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosphocin, 98%, (99% ee) (1297613-73-0) \\ & \mathsf{(C_{a, H_{a}, 0, P, FW: 566.59; white to yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. \\ \\ & ((118R) = 0, (11,12,13) = Tetrahydro-5-hydroxy-3,7-di-1-naphthalenyl-5-oxide-dindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosphocin, 98%, (99% ee) (1256327-08-0) \\ & \mathsf{(C_{a, H_{a}, 0, P, FW: 566.59; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. \\ \\ & ((118R) = 0, (11,12,13) = Tetrahydro-5-hydroxy-3,7-di-2-naphthalenyl-5-oxide-dindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosphocin, 98%, (99% ee) (1257613-74-1) \\ & \mathsf{(C_{a, H_{a}, 0, P, FW: 566.59; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. \\ \\ & ((118R) = 0, (11,12,13) = Tetrahydro-5-hydroxy-3,7-di-2-naphthalenyl-5-oxide-dindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosphocin, 98%, (99% ee) (1257613-74-1) \\ & \mathsf{(C_{a, H_{a}, 0, P, FW: 566.59; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research purposes only. \\ \\ & ((118R) = 0, (11,12,13) = Tetrahydro-5-hydroxy-3,7-di-2-naphthalenyl-5-oxide-dindeno[7,1-de:1',7'-fg][1,3,2]dioxaphosphocin, 98%, (99% ee) (257613-72-9) \\ & \mathsf{C_{a, H_{a}, 0, P, FW: 566.59; white to light-yellow pwdr. Note: Sold in collaboration with Daicel for research pur$



PHOSPHORUS (Compounds)



Tris(1-adamantyl)phosphine, 97% (897665-73-5) C₃₀CH₄₅P; FW: 436.65; white to off-white pwdr. air sensitive Note: Sold in collaboration with GreenCentre for research purposes only. Patents: 62248056.

250mg

1g

5g

Tech. Note (1) Ref. (1,2)

Technical Note:

- A stable phosphine ligand which can bind to metals such as palladium 1.
 - to be used in Suzuki-Miyaura couplings in making drug intermediates. Dramatic effects in catalysis are also accessible now by using PAd₃ as a ligand during Suzuki-Miyaura cross-coupling of chloro(hetero)arenes.^{1,2}



Ar-Y + (HO)₂B-R

K₃PO₄ or K₂CO₃ (Y = CI, Br)THF, tol, or BuOH

References:

1. Synlett., 2017, 28(3), 280-288.

2. J. AIII. C	<i>Inem. Soc.</i> , 2016 , <i>130</i> , 0392.	
15-7725 NEW	Tris(2,6-dimethoxyphenyl)phosphine, min. 97% (85417-41-0) $C_{24}H_{27}O_6P$; white to off-white pwdr.; m.p. 145-147° <i>air sensitive</i>	5g
PLATINU	M (Elemental Forms)	
78-3015 New	Platinum nanoparticles, 1% on carbon black (surfactant and reactant-free) (7440-06-4) Pt; FW: 195.10; black solid (store cold) Note: Manufactured by laser ablation. Sold in collaboration with Particular® for research purposes only.	5g 25g
78-3020 New	Platinum nanoparticles, 5% on carbon black (surfactant and reactant-free) (7440-06-4) Pt; FVV: 195.10; black solid (store cold) Note: Manufactured by laser ablation. Sold in collaboration with Particular® for research purposes only.	5g 25g
78-3030 New	Platinum nanoparticles, 10% on carbon black (surfactant and reactant-free) (7440-06-4) Pt; FW: 195.10; black solid (store cold) Note: Manufactured by laser ablation. Sold in collaboration with Particular® for research purposes only.	1g 5g
78-3032 New	Platinum nanoparticles, 20% on carbon black (surfactant and reactant-free) (7440-06-4) Pt; FW: 195.10; black solid (store cold) Note: Manufactured by laser ablation. Sold in collaboration with Particular® for research purposes only.	1g 5g
78-3035 New	Platinum nanoparticles, 30% on carbon black (surfactant and reactant-free) (7440-06-4) Pt; FW: 195.10; black solid (store cold) Note: Manufactured by laser ablation. Sold in collaboration with Particular® for research purposes only.	1g 5g

PLATINU	M (Elemental Forms)	
78-3012 New	Platinum nanoparticles, 1% on Titania (anatase) (surfactant and reactant-free) (7440-06-4) Pt; FW: 195.10; gray solid (store cold) Note: Manufactured by laser ablation. Sold in collaboration with Particular® for research purposes only.	5g 25g
78-3005 New	Platinum nanoparticles, 1% on Titania (rutile) (surfactant and reactant-free) (7440-06-4) Pt; FW: 195.10; light gray solid (store cold) Note: Manufactured by laser ablation. Sold in collaboration with Particular® for research purposes only.	5g 25g
78-3026 New	Platinum nanoparticles, 10% on Titania (anatase) (surfactant and reactant-free) (7440-06-4) Pt; FW: 1955.10; dark gray solid (store cold) Note: Manufactured by laser ablation. Sold in collaboration with Particular® for research purposes only.	1g 5g
78-3023 New	Platinum nanoparticles, 10% on Titania (rutile) (surfactant and reactant-free) (7440-06-4) Pt; FW: 195.10; dark gray solid (store cold) Note: Manufactured by laser ablation. Sold in collaboration with Particular® for research purposes only.	1g 5g
PLATINUI	M (Compounds)	
78-0455 NEW	cis-Dichlorodiammineplatinum (II), CISPLATIN (USP) (15663-27-1) cis-Pt(NH ₃) ₂ Cl ₂ ; FW: 300.06; yellow to orange xtl.; m.p. 230°	1g 5g
98-1350 New	(Trimethyl)methylcyclopentadienylplatinum(IV), 99% (99.999%-Pt) PURATREM (94442-22-5) (CH ₃) ₃ (CH ₃ C ₅ H ₄)Pt; FW: 319.32; off-white pwdr.; m.p. 30-31°; b.p. (subl. 23°C/0.053mm); d. 1.88 air sensitive, (store cold)	500mg 2g 10g
PRASEO	DYMIUM (Compounds) Tris[N,N,N,N-tetramethylguanidinium][- tris(1S)-(1,1'-binaphalene)-2,2'-diolato] praseodymate Pr-HTMG-B C ₇₅ H ₇₈ N ₉ O ₆ Pr; FW: 1342.38; off-white pwdr.	250mg 1g

Technical Note:

1. See 57-1250 (page 43)

Note: U.S.Patent 14/898,925.



RUTHENIUM (Compounds)



Allylbis(2-aminoethyldiphenylphosphino) ruthenium(II) tetrafluoroborate, 98% (1352633-94-3) $C_{31}H_{37}BF_4N_2P_2Ru;$ FW: 687.46; off-white to pale yellow solid Note: Sold in collaboration with GreenCentre for research purposes only. Patents: PCT/2013/010275.



100mg 500mg

Technical Notes:

- 1. A highly active catalyst for the hydrogenation of amides to alcohols and amines.
- 2. Catalyst used for the hydrogenation of functionalized amides under basic and neutral conditions.



RUTHENIUM (Compounds) 44-0748 [1,3-Bis(2,6-di-i-propylphenyl)imidazolidin-100mg iPr 2-ylidene][(2-((1-methoxy(methyl) 500mg NEW amino)-1-oxopropan-2-yl)oxy] benzylidenediiodoruthenium(II) GreenCat-I2 iPr iP C₃₀H₅₃I₂N₃O₃Ru; FW: 966.74; green solid Ы Ru air sensitive, (store cold) Note: Sold in collaboration with Apeiron Synthesis, Inc. U.S. Patent 61/666.009 Me PCT/EP2013/062435. Store at 2-8°C under inert O atmosphere. Catalyst may be weighed in air Me Me 44-0770 1,3-Bis(2,6-di-i-propylphenyl)imidazoliiPr 100ma iPr din-2-ylidene)(2-i-propoxy-5-nitrobenzylidene) 500mg NEW ruthenium(II) dichloride Nitro-Grela SiPr iP iP (928795-51-1) °CI C₃₇H₄₉Cl₂N₃O₃Ru; FW: 755.78; green pwdr. Ru= (store cold) CI NO₂ Note: Sold in collaboration with Apeiron Synthesis, Inc. U.S. Patent 6/867,303 PCT/EP2003/01122. Store at 2-8°C under inert atmosphere. Catalyst may be weighed in air 44-0782 [1,3-Bis(2,6-di-i-propylphenyl)imidazoli-100ma iPr din-2-ylidene)(2-i-propoxy-5-nitrobenzylidene) 500mg NEW ruthenium(II) diiodide nitro-Grela I2 SIPr iPr iP (1874265-00-5) Ы C₃₇H₄₀I₂N₃O₃Ru; FW: 938.68; olive brown pwdr. air sensitive, (store cold) Note: Sold in collaboration with Apeiron Synthesis, NO₂ Inc. U.S. Patent 6/867,303 PCT/EP2003/01122. Store at 2-8°C under inert atmosphere. Catalyst may be weighed in air 44-0793 100mg [1,3-Bis(2,6-di-i-propylphenyl)imidaziPr iPr olidin-2-ylidene)(tricyclohexylphos-500mg NEW phine)-(2-oxo-5-nitrobenzylidene)ruthenium(II) iPr iP chloride LatMet SIPr (1544328-59-7) .CI C₅₂H₇₇CIN₂OPRu; FW: 913.68; dark green xtl. Ru air sensitive Cy₃P⁴ Note: Sold in collaboration with Apeiron Synthesis, Inc. U.S. Patent 9,328,132, PCT/EP2013/065839. 44-0056 Bis(N,N'-di-tert-butylacetamidinato)ruthenium(II) 1g tBu dicarbonyl, 98% (99.99%-Ru) PURATREM 5g tBu NEW (949113-49-9) \mathbf{O} amp N C₂₂H₄₂N₄O₂Ru; FW: 495.67; Beige to yellow solid; Me Ru Me m.p. 204 ĊO air sensitive, moisture sensitive tBu Note: Product sold under, use subject to, terms and İRu conditions of label license at www.strem.com/harvard2

RUTHENIUM (Compounds)





Catalyst for generation of hydrogen generation from glycerol and selective synthesis of lactic acid.

Catalyst N-formylation of morpholine with hydrogen and carbon dioxide.

$$\begin{array}{c} 0 \\ 0 \\ 0 \\ 0 \end{array} + H_2 \\ 50 \text{ atm} \end{array} \xrightarrow{\text{Catalyst, KOfBu}} H0 \\ \hline \text{THF, 140°C} \end{array} + H0 \\ \hline \text{OH} + \text{MeOH} + \text{HCOOMe} \\ \hline \text{Ref. (1)} \\ \hline \text{Ref. (1)} \\ \hline \text{Ref. (1)} \\ \hline \text{Ref. (2)} \end{array}$$



References:

1. ChemSus Chem., 2015, 8, 4139.

RUTHENIUM (Compounds)



RUTHENIUM (Compounds) 44-0460 Dihydrotetrakis(triphenylphosphine)ruthenium(II), 250mg 95% (19529-00-1) 1a NEW C₇₂H₆₂P₄Ru; FW: 1152.23; yellow to green pwdr.; Phm.p. 181-183° air sensitive Technical Notes: Catalyst for transfer hydrogenation of ketones with alcohols involving carbon-carbon bond formation. 1. 2 Catalyst for oxidative cyclization of 2-aminobenzyl alcohol with ketones. 3 Amidation catalyst of alcohols or aldehydes with amines. 4. Stereoselective semireduction of internal alkynes to Z-olefins under transfer hydrogenation conditions. 5. Catalyst for synthesis of cyclic imides from nitriles and diols via hydrogen transfer mechanism. 6. Catalyst for direct hydrogenation of carboxylic acids using triphos-type ligands. OH Tech. Note (1) Ru Catalvst OH Ref. (1) R' OH Ru Catalyst Tech. Note (2) Ref. (2) NH2 Ru Catalyst Tech. Note (3) Ref. (3) Ru Catalvst Tech. Note (4) Ref. (4) HΩ Ru Catalyst Tech. Note (5) R¹-CN $2H_2$ Ref. (5) R3

Tech	Note	(6)
Ref. (6)	

References:

ΩН

J. Org. Chem., 2001, 66, 9020. 1.

Ru Catalyst/N-(PR)₃

70 bar H₂

R

OH

- 2. Chem. Commun., 2001, 2576.
- 3. J. Org. Chem., 2010, 75, 3002.
- Chem. Eur. J., 2010, 16, 12214. 4.
- Org. Lett., 2014, 16, 4404. 5.
- 6. ChemSusChem., 2016, 9, 177.



RUTHENIUM (Compounds)



RUTHENIUM (Compounds)





SAMARIUM (Elemental Forms) 62-6235 Samarium powder (99.9% REO) 325 mesh (7440-19-9) 5g Sm; FW: 140.4; 325 mesh; m.p. 1072°; b.p. 1778°; d. 7.40 25g NEW air sensitive, moisture sensitive HAZ SCANDIUM (Compounds) 21-1200 Tris(N,N'-di-i-propylformamidinato)scandium(III), 1g iPr (99.9%-Sc) 5g NEW iPr C₂₁H₄₅ScN₆; FW: 426.58; white to off-white pwdr. air sensitive, moisture sensitive Note: Product sold under, use subject to, terms iPr and conditions of label license at www.strem.com/ harvard2 iPr SILICON (Compounds) Bis(t-butylamino)silane, BTBAS (99.999%-Si) 14-1072 1g ^tBu ^tBu PURATREM (186598-40-3) 5g NEW C₈H₂₂N₂Si; FW: 174.36; colorless lig.; b.p. 167°C; 25q f.p. 30°C; d. 0.816 HN ŇΗ moisture sensitive Si н н 14-1530 Bis(dimethylamino)dimethylsilane, 99+% 5g Me Me BDMADMS (3768-58-9) 25a NEW [N(CH₃)₂]₂(CH₃)₂Si; FW: 146.31; colorless liq.; Me Me b.p. 128-129°; f.p. -3°C; d. 0.81 moisture sensitive Me Me

Technical Notes:

- Used in the chemical vapor deposition of silicon nitride films, and also the atomic layer deposition of SiNxCy dielectric sealing layers using plasma-enhanced atomic layer deposition (PE-ALD).^{1,2,3,4}
- 2. Used as a reagent for silvlation.^{5,6}

References:

- 1. AIP Conference Proceedings, 2015, 1649(1, Irago Conference 2014), 41-46.
- 2. Materials Science in Semiconductor Processing, 2015, 29, 139-142.
- 3. Applied Surface Science, 2010, 257(4), 1196-1203.
- 4. Surface and Coatings Technology, 2008, 202(9), 1606-1614.
- 5. Microelectronic Engineering, 1991, 13(1-4), 47-50.
- 6. Journal of Vacuum Science & Technology, B: Microelectronics and Nanometer Structures, 1990, 8(6), 1481-7.

14-6100	High Surface area Silica nanoparticles, large, particle size ~900-1000 nm,	1g
	surface area ~700 m²/g, (KCC-1 L1) (112945-52-5)	5g
NEW	SiO ₂ ; FW: 60.09; white pwdr.; SA: ~700 m²/g; P.Vol. ~1.4 cm³/g	
	Note: Diameter: ~900-1000nm; This product is under license of patented	
	Technology from King Abdullah University of Science and Technology – KAUST.	
	Patent PCT/IB2010/002421.	
chnical Note		

Technical Note:

 Novel fibrous shaped silica nanospheres, denoted as KCC-1 (KAUST Catalysis Center)⁽¹⁾, have unique physical properties which have never before been reported in silica materials. These nanomaterials have been developed by Prof. J. M. Basset of King Abdullah University of Science and Technology (KAUST). A fibrous surface morphology arranged in three-dimensional structure forms the spheres (Fig. 1). Unlike traditional pore-based silica, these nanospheres possess a fibrous structure that increases accessibility to the available surface area; this in turn, significantly increases the catalytic activity.

These materials exhibit excellent physical properties, including a high surface area, a fibrous surface morphology, good thermal and hydrothermal stabilities and high mechanical stability (Table 1). The fibrous morphology of KCC-1 remains unaffected even after mechanical compression up to 216 MPa pressure. This is superior to the conventional MCM-41 type of silica, which is affected at pressure 86 MPa.⁽¹⁾

SILICON (Compounds)

14-6100 High Surface area Silica nanoparticles, large, particle size ~900-1000 nm, (cont.) surface area ~700 m²/g, (KCC-1 L1) (112945-52-5)

A range of heterogeneous catalysts, prepared using KCC-1 as a supporting material, have been showing excellent catalytic activity for various transformations of research and industrial importance. As a catalyst support, sorbent or carrier, KCC-1 is able to demonstrate superior activity as compared to regular mesoporous silica materials in energy related processes^[2-3], a variety of organic reactions^[4-7], biomedical applications and drug delivery systems^[8], optoelectronic devices^[9] and many others.



Product #	Category	Grade	Particle Size (nm)	Surface Area (m ² /g)	Pore Volume (cm ³ /g)
14-6100	Large	(KCC-1 L1)	~900-1000	~700	~1.4
14-6110	Large	(KCC-1 L2)	~900-1000	~600	~1.2
14-6120	Large	(KCC-1 L3)	~900-1000	~550	~0.9
14-6200	Medium	(KCC-1 M1)	~400-450	~400	~0.7
14-6210	Medium	(KCC-1 M2)	~300-350	~600	~0.6
14-6300	Small	(KCC-1 S1)	~130-190	~380	~0.8

References:

- 2. Chem. Mater. 2015, 27, 8237
- ACS Catalysis. 2016, 6, 2770 3.
- ChemSusChem 2012, 5, 85 4
- 5. Green Chem., 2016, 18, 5890
- 6. Angew. Chem. Int. Ed. 2011, 50, 2747
- 7. RSC Adv., 2017, 7, 24885
- 8. Langmuir 2014, 30, 10886
- 9. J. Mater. Chem. B, 2015, 3, 3201

14-6110	High Surface area Silica nanoparticles, large, particle size ~900-1000 nm,	1g
NIEW	surface area ~600 m²/g, (KCC-1 L2) (112945-52-5)	5g
INE W	SiO ₂ ; FW: 60.09; white pwdr.; SA: ~600 m²/g; P.Vol. ~1.20 cm³/g	
	Note: Diameter: ~900-1000nm; This product is under license of patented	
	Technology from King Abdullah University of Science and Technology –	
	KAUST. Patent PCT/IB2010/002421.	
chnical Note:		
1. See 14-	6100 (page 92)	

Te

Technical Note:	Surface area Silica nanoparticles, large, particle size ~900-1000 nm, surface area ~550 m²/g (KCC-1 L3) (112945-52-5) SiO ₂ ; FW: 60.09; white pwdr.; SA: ~550 m²/g; P.Vol. ~0.97 cm³/g Note: Diameter: ~900-1000nm; This product is under license of patented Technology from King Abdullah University of Science and Technology – KAUST. Patent PCT/IB2010/002421.	1g 5g
1. See 14-	6100 (page 92)	
14-6200	High Surface area Silica nanoparticles, medium, particle size ~400-450 nm, surface area ~400 m²/g, (KCC-1 M1) <i>(112945-52-5)</i>	1g 5g

	info@strem.com . order@strem.com
See	14-6100 (page 92)

Patent PCT/IB2010/002421.

Technical Note: 1.

SiO2; FW: 60.09; white pwdr.; SA: ~400 m2/g; P.Vol. ~0.7 cm3/g Note: Diameter: ~400-450nm; This product is under license of patented Technology from King Abdullah University of Science and Technology – KAUST.

^{1.} Angew. Chem. Int. Ed. 2010, 49, 9652

item i reducte i		
SILICON (Compounds)	
14-6210 NEW	High Surface area Silica nanoparticles, medium, particle size ~300-350 nm, surface area ~600 m²/g, (KCC-1 M2) (112945-52-5) SiO ₂ ; FW: 60.09; white pwdr.; SA: ~600 m²/g; P.Vol. ~0.6 cm³/g Note: Diameter: ~300-350nm; This product is under license of patented Technology from King Abdullah University of Science and Technology – KAUST. Patent PCT/IB2010/002421.	1g 5g
Technical Note: 1. See 14-6	\$100 (page 92)	
14-6300 NEW	High Surface area Silica nanoparticles, small, particle size ~130-190 nm, surface area ~380 m²/g, (KCC-1 S1) (112945-52-5) SiO ₂ ; FW: 60.09; white to beige pwdr.; SA: ~380 m²/g; P.Vol. ~0.8 cm³/g Note: Diameter: ~130-190nm; This product is under license of patented Technology from King Abdullah University of Science and Technology – KAUST. Patent PCT/IB2010/002421.	1g 5g
1. See 14-6	6100 (page 92)	
14-6310 New	High Surface area Silica nanoparticles, small, particle size ~40-50 nm, surface area ~520 m²/g, (KCC-1 S2) (<i>112945-52-5</i>) SiO ₂ ; FW: 60.09; white to beige pwdr.; SA: ~520 m²/g; P.Vol. ~1.3 cm³/g Note: Diameter: ~40-50nm; This product is under license of patented Technology from King Abdullah University of Science and Technology – KAUST. Patent PCT/IB2010/002421.	1g 5g
14-1445 New HAZ	Silicon(IV) bromide, (99.99% Si) PURATREM (7789-66-4) SiBr ₄ ; FW: 347.72; colorless liq.; m.p. 5.4°; b.p. 154°; d. 2.772 <i>moisture sensitive</i>	5g 25g
14-1943 New	2,2,4,4-Tetramethyl-3,8,11,14,17-pentaoxa-2,4-disilaoctadecane, 99+% Electrolyte solvent ANL-2SM3 ($855996-83-7$) C ₁₅ H ₃₆ O ₅ Si ₂ ; FW: 352.61; colorless liq. Note: Use for batteries for medical devices expressly excluded. U.S. Patent: 8,076,031 B1; 14/266,052.	5g 25g
Organosilicon E 1. Silicon base oligomers ir stability of li 2. Increases b 3. Are less flar batteries. 4. Improves co	Electrolytes for Lithium Ion Batteries ed electrolytes with polyethylene glycol mprove thermal and electrochemical thium-ion batteries. attery long-term stability. mmable than conventional organic carbonate-based solvents and maintain the safe ponductivity and kinetics of the lithium salts.	Me Si SiMe ₃ Me operation of
Electrochemica	I and Physical Properties:	

- Disiloxane liquid electrolyte ANL-2SM3 exhibits electrochemical stability, high thermal stability, and low viscosity.- Viscosity 3.8 cP at 25°C; The conductivity and viscosity of ANL-2SM3-based electrolyte are 3.65x10⁻⁴ Scm⁻¹ and 18 cP at 25°C [1,2] Charged cathode material is more thermally stable in the siloxane-based electrolyte than in the carbonate-based electrolyte [1]. Boiling point 269-271°C; Glass transition temperature -103.0°C
- Soluble electrolytic lithium salts: LiBOB, LiPF₆, ANL2SM3, and LiTFSIANL-2SM3 is compatible with nanostructured material-based electrodes [3].
- Disiloxane/LiBOB or Disiloxane /LiPF₆ electrolytes show conductivities up to 6.2x10⁴ Scm⁻¹ at room temperature. Disiloxane electrolytes doped with 0.8MLiBOB are stable to 4.7 V. The LiBOB/disiloxane combinations were found to be good electrolytes for lithium-ion cells [4].

- 1. J. Power Sources, 2006, 160, 645
- 2. J. Power Sources, 2006, 160, 1355
- 3. Chem. Mater., 2007, 19, 5734
- 4. J. Power Sources, 2010, 195, 6062

SILVER (Compounds)





iPr Aα CF₂H 100mg 500mg

Technical Notes:

- 1. Catalyst for Pd(dba)2/DPPF assisted direct difluoromethylation of aryl bromides and iodides.
- Pre-catalyst for copper assisted difluoromethylthiolation of aryl 2. and heteroaryl diazonium salts.
- 3. Pre-catalyst for palladium assisted difluoromethylthiolation of pyridyl, guinolinyl, benzothiazolyl, thiophenyl, carbazolyl and pyazolyl heteroaryl bromides, iodides, triflates and aryl iodides.



Ar = pyridyl, quinolinyl, benzothiazolyl, thiophenyl, carbazolyl, pyazolyl

- 1 Nat. Commun. 2014, 5, 5405.
- 2. Angew. Chem. Int. Ed. 2015, 54, 7648.
- Chem. Sci., 2016, 7, 3757. 3

SODIUM (Compounds)						
11-1146 New	Sodium hexafluorophosphate 99% (99.99%-Na) PURATRE (21324-39-0) F ₆ NaP; FW: 167.95; white pwdr.; d. 2.37 <i>air sensitive, moisture sensitive, hygroscopic</i>	M Na ⁺ PF ₆ -	2g 10g			
11-1147 New HAZ	Sodium hexafluorophosphate 99% (99.99%-Na) PURATREM (<10ppm H2O) (21324-39-0) NaPF ₆ ; FW: 167.95; white pwdr. <i>air sensitive, moisture sensitive, hygroscopic</i> Note: Suitable for battery applications		1g 5g			
11-1015 New	Sodium tetrakis(4-fluorophenyl)borate dihydrate, 97% (207683-22-5) $C_{24}H_{20}BF_4NaO_2$; FW: 414.18(450.21); white xtl.	F	250mg 1g 5g			

TITANIUM (Compounds)



YTTRIUM (Compounds)

39-5850

NEW

 $\label{eq:transform} \begin{array}{l} \textbf{Tris[N,N,N,N-tetramethylguanidinium]} \\ \textbf{[tris(1S)-(1,1'-binaphalene)-2,2'-diolato]} \\ \textbf{yttrate Y-HTMG-B} \ (1611526-73-8) \\ \textbf{C}_{75}\textbf{H}_{78}\textbf{N}_{9}\textbf{O}_{6}\textbf{Y}; \ \textbf{FW}: \ 1290.39; \ \text{orange pwdr.} \\ \textbf{Note: U.S. Patent 14/898,925.} \end{array}$

Technical Note:

1. See 57-1250 (page 43)



250mg

1g

ZIRCONIUM (Compounds)

98	8 -4042 New Haz	Tetrakis(dimethylamino)zirconium(IV 40-4115, contained in 50ml Swagelok (19756-04-8) Zr[N(CH ₃) ₂] ₄ ; FW: 267.53; light yellow xi moisture sensitive, (store cold) Note: Product sold under, use subject to license at www.strem.com/harvard1), 99% (99.99%-Zr) TDMAZ, ® cylinder (96-1070) for CVD/ALD tl. o, terms and conditions of label	25g
40	0-1109	Zirconium aminobenzenedicarbox-		500mg
	NEW	ylate MOF (UiO-66-BDC-NH2, BDC-NH2:Zr=0.9-1.0) ($1260119-00-3$) Zr ₆ O ₄ (OH) ₄ (C ₆ H ₅ NO ₄) _x , X = 5.4-6.0; yellow solid; SA: 800-1075 m²/g; P.Vol. 0.31-0.41 cm³/g Note: Particle size: 0.1-0.5 micron, Thermal stability: 300°C, Activation temperature: 150°C Sold under license from Inven2 AS for research purposes only. PCT/GB2009/001087.		2g
Technical Notes:				
1. 2. Refere	Useful Mo Catalyst I agents to nces:	DF for adsorption of CO ₂ applications ¹ . MOF used in the conversion of toxic non-toxic products ² .	H ₂ N COOH	

- 1. Carbon Dioxide Adsorption in Amine-Functionalized Mixed-Ligand Metal-Organic Frameworks of UiO-66 Topology., Chem.Sus.Chem. **2014**.
- Tailoring the Pore Size and Functionality of UiO-Type Metal-Organic Frameworks for Optimal Nerve Agent Destruction, Inorg. Chem. 2015.
- Towards Metal-Organic Framework based Field Effect Chemical Sensors: UiO-66-NH2 for Nerve Agent Detection, Chem. Sci., 2016, 7, 5827.

ZIRCONIUM (Compounds)



References:

- 1. Tuned to Perfection: Ironing Out the Defects in Metal–Organic Framework UiO-66 Chem. Mater. **2014**, 26, 4068–4071.
- H₂ storage in isostructural UiO-67 and UiO-66 MOFs Phys. Chem. Chem. Phys., **2012**, *14*, 1614–1626.
- A New Zirconium Inorganic Building Brick Forming Metal Organic Frameworks with Exceptional Stability. J. Am. Chem. Soc. 2008, 130, 13850–13851.



- UiO-67-type Metal-Organic Frameworks with Enhanced Water Stability and Methane Adsorption Capacity, Inorg. Chem. 2016, 55, 1986–1991.
- 2. H₂ storage in isostructural UiO-67 and UiO-66 MOFs, Phys. Chem. Chem. Phys., **2012**, 14, 1614–1626.
- A New Zirconium Inorganic Building Brick Forming Metal Organic Frameworks with Exceptional Stability. J. Am. Chem. Soc. 2008, 130, 13850–13851, Jasmina Hafizovic Cavka, Søren Jakobsen, Unni Olsbye, Nathalie Guillou, Carlo Lamberti, Silvia Bordiga, and Karl Petter Lillerud.

ZIRCONIUM (Compounds)



- - studies for the storage of hydrogen or methane2
 - 2. Water adsorption in MOF's for many applications such as dehumidification, thermal batteries, and delivery of drinking water in remote areas3

References:

- Water harvesting from air with metal-organic frameworks powered by natural sunlight. Science, 2017, 356, 1. 430-434
- 2. A Facile "Green" Route for Scalable Batch Production and Continuous Synthesis of Zirconium MOFs. Eur. J. Inorg. Chem. 2016, 4490-4498.
- 3. Water Adsorption in Porous Metal-Organic Frameworks and Related Materials, J. Am. Chem. Soc., 2014, 136, 4369-4381.
- 4 A water-born Zr-based porous coordination polymer: Modulated synthesis of Zr-fumarate MOF. Microporous and mesoporous materials , 2015, 203, 186-194.



Technical Notes:

- 1. Metalorganic framework used in a large number of studies for the storage of hydrogen or methane²
- 2. Water adsorption in MOF's for many applications such as dehumidification, thermal batteries, and delivery of drinking water in remote areas3

- 1 Water harvesting from air with metal-organic frameworks powered by natural sunlight. Science, 2017, 356, 430-434.
- 2. A Facile "Green" Route for Scalable Batch Production and Continuous Synthesis of Zirconium MOFs. Eur. J. Inorg. Chem. 2016, 4490-4498.
- 3. Water Adsorption in Porous Metal-Organic Frameworks and Related Materials, J. Am. Chem. Soc., 2014, 136.4369-4381.
- 4. A water-born Zr-based porous coordination polymer: Modulated synthesis of Zr-fumarate MOF. Microporous and mesoporous materials , 2015, 203, 186-194.

ZIRCONIUM (Compounds)



 Functionalized forms show the highest selectivity, good working capacity and medium ranged CO₂ adsorption enthalpy that make these materials very promising for physi-sorption-based processes²

- 1. Engineering Copper Carboxylate Functionalities on Water Stable Metal–Organic Frameworks for Enhancement of Ammonia Removal Capacities. J. Phys. Chem. C, **2017**, *121*, 3310–3319.
- Functionalizing porous zirconium terephthalate UiO-66(Zr) for natural gas upgrading: a computational exploration., Chem. Commun., 2011, 47, 9603–9605.

KITS - AntPhos and WingPhos Kit



15-1967

15-1970

15-1975

AntPhos and WingPhos Kit Sold in collaboration with Zejun for research purposes only. Patents ZL20131002037.1, CN 201610056390. Components also available for individual sale. Contains the following:



25mg

25mg

25mg

See page 67

See page 73

See page 73

(S)-4-(Anthracen-9-yl)-3-(t-butyl-2,3-dihydrobenzo[d][1,3]

ee), [(2R,2'R,3R,3'R)-WingPhos] (1884680-45-8)

ee), [(2S,2'S,3S,3'S)-WingPhos] (1435940-19-4)

oxaphosphole,99+% (>99% ee) [(S)-AntPhos] (1807740-34-6)

(2R,2'R,3R,3'R)-4,4'-Di(anthracen-9-yl)-3,3'-di-t-butyl-2,2',3,3'-

tetrahydro-2,2'-bibenzo[d][1,3]oxaphosphole, min 98% (>90%

(2S,2'S,3S,3'S)-4,4'-Di(anthracen-9-yl)-3,3'-di-t-butyl-2,2',3,3'-

tetrahydro-2,2'-bibenzo[d][1,3]oxaphosphole, min 98%, (>99%

KITS - BI-DIME Ligand Kit



BI-DIME Ligand Kit Sold in collaboration with Zejun for research purposes only. Patents: ZL2013105048267, CN104558038. Components also available for individual sale. Contains the following:



15-6205	3-(t-Butyl)-4-(2,6-dimethoxyphenyl)-2,3-dihydrobenzo[d] [1,3]oxaphosphole, min. 97% rac-BI-DIME <i>(1246888-90-3)</i>	25mg	See page 70
15-6210	(S)-3-(t-Butyl)-4-(2,6-dimethoxyphenyl)-2,3-dihydrobenzo[d] [1,3]oxaphosphole, min. 97% (S)-BI-DIME (1373432-09-7)	25mg	See page 72
15-6211	(R)-3-(t-Butyl)-4-(2,6-dimethoxyphenyl)-2,3-dihydrobenzo[d] [1,3]oxaphosphole, min. 97% (R)-BI-DIME (1373432-03-7)	25mg	See page 71
15-6220	(2S,3S)-3-(t-Butyl)-4-(2,6-dimethoxyphenyl)-2-methyl-2,3-di- hydrobenzo[d][1,3]oxaphosphole, min. 97% (S,S)-Me-BI-DIME (1373432-11-1)	25mg	See page 72
15-6225	(2R,3R)-3-(t-Butyl)-4-(2,6-dimethoxyphenyl)-2-methyl-2,3-di- hydrobenzo[d][1,3]oxaphosphole, min. 97% (R,R)-Me-BI-DIME (1477517-18-2)	25mg	See page 72
15-6230	(2S,3S)-3-(t-Butyl)-4-(2,6-dimethoxyphenyl)-2-i-propyl-2,3-di- hydrobenzo[d][1,3]oxaphosphole, min. 97% (S,S)-iPr-BI-DIME (1477517-21-7)	25mg	See page 72
15-6235	(2R,3R)-3-(t-Butyl)-4-(2,6-dimethoxyphenyl)-2-i-propyl-2,3-di- hydrobenzo[d][1,3]oxaphosphole, min. 97% (R,R)-iPr-BI-DIME (1477517-19-3)	25mg	See page 72

KITS - Buchwald Biaryl Phosphine Ligand Mini Kit 3



Buchwald Biaryl Phosphine Ligand Mini Kit 3 for Aromatic Carbon-Heteroatom Bond Formation, Suzuki Coupling and Negishi Cross-coupling

For aromatic carbon-heteroatom bond formation and Suzuki Coupling.







KITS - Buchwald Biaryl Phosphine Ligand Mini Kit 3

96-5495 Buchwald Biaryl Phosphine Ligand Mini Kit 3 for Aromatic Carbon-Heteroatom Bond (cont.) Formation, Suzuki Coupling and Negishi Cross-coupling

For aromatic carbon-heteroatom bond formation and Suzuki Coupling.

15-1063	2-Di-t-butylphosphino-4-methoxy-3,5,6-trimethyl- 2',4',6'-tri-i-propyl)-1,1'-biphenyl, min. 98% [~1:1 mixture with regioisomer, 2-Di-t-butylphosphino-5-me- thoxy-3,4,6-trimethyl-2',4',6'-tri-i-propylbiphenyl] (1359986-21-2)	250mg	Visit strem.com
15-1105	2-Dicyclohexylphosphino-2'-methoxy-4',6'-di-t-butyl-1,1'-bi- phenyl, min. 98% VPhos (1848244-75-6)	250mg	Visit strem.com
15-1125	2-Diphenylphosphino-2',6'-bis(dimethylamino)-1,1'-biphe- nyl, min. 98% PhCPhos (1447963-71-4)	100mg	Visit strem.com
15-1135	2'-Dicyclohexylphosphino-2,6-di-i-propyl-4-sulfona- to-1,1'-biphenyl hydrate sodium salt (XPhos-SO ₃ Na) (870245-84-4)	100mg	Visit strem.com
15-1142	2'-Dicyclohexylphosphino-2,6-dimethoxy-3-sulfona- to-1,1'-biphenyl hydrate sodium salt (water soluble SPhos), min. 98% (1049726-96-6)	500mg	Visit strem.com
15-1147	2-Dicyclohexylphosphino-2',6'-bis(dimethylamino)-1,1'-bi- phenyl, min. 98% CPhos (1160556-64-8)	250mg	Visit strem.com
15-1151	2-Diethylphosphino-2',6'-bis(dimethylamino)-1,1'-biphenyl di(hydrogen tetrafluoroborate) salt, min. 98% EtCPhos	100mg	Visit strem.com
15-1154	2-Dicyclohexylphosphino-4'-(N,N-dimethylamino)-1,1'-bi- phenyl, 98% (1185899-00-6)	250mg	Visit strem.com
15-1168	2-(Di-t-butylphosphino)-3-methoxy-6-methyl-2',4',6'-tri-i- propyl-1,1'-biphenyl, min. 98% RockPhos (1262046-34-3)	100mg	Visit strem.com
15-2065	2-(Diadamantylphosphino)-3-methoxy-2',4',6'-tri-i-pro- pyl-3'-(2,3,5,6-tetrafluoro-4-butylphenyl)-1,1'-biphenyl, AIPhos (1805783-60-1)	100mg	Visit strem.com
15-3010	2-(t-Butylphenylphosphino)-2',6'-dimethylamino-1,1'- biphenyl, 98% (t-Bu)PhCPhos (1660153-91-2)	250mg	Visit strem.com
15-3015	2-[Bis(3,5-trifluoromethylphenylphosphino)-3,6-dime- thoxy]-2',6'-dimethylamino-1,1'-biphenyl, 98% (1810068-30-4)	100mg	Visit strem.com
15-3020	2-[Bis(3,5-trifluoromethylphenylphosphino)-3,6-dime- thoxy]-2',6'-di-i-propoxy-1,1'-biphenyl, 98% (1810068-31-5)	250mg	Visit strem.com

KITS - Buchwald Palladacycle Precatalyst Kit 2b



Buchwald Palladacycle Precatalyst Kit 2b (Methanesulfonato-2'-amino-1,1'-biphenyl-2-yl- Palladacycles Gen. 3)

Patents: US 6,395,916, US 6,307,087. Components also available for individual sale.

Contains the following:


KITS - Buo	chwald Palladacycle Precatalyst Kit 2b		
96-5506 (cont.)	Buchwald Palladacycle Precatalyst Kit 2b (Methanesulfonato-2'-amino-1,1'-biphenyl-2-yl- Palladacycle	s Gen. 3)	
46-0348	Methanesulfonato[2-diethylphosphino-2',6'-bis(dimethyl- amino)-1,1-biphenyl](2'-amino-1,1'-biphenyl-2-yl)palladi- um(II), min. 98% [EtCPhos Palladacycle Gen. 3]	50mg	Visit strem.com
46-0357	Methanesulfonato(2-di-t-butylphosphino-1,1'-binaphthyl) (2'-amino-1,1'-biphenyl-2-yl)palladium(II), min. 95% [TrixiePhos Palladacycle Gen. 3]	100mg	Visit strem.com
46-0387	Methanesulfonato(tri-t-butylphosphino)(2'-amino-1,1'-biphe- nyl-2-yl)palladium(II), 98% [P(t-Bu)3 Palladacycle Gen. 3] (1445086-17-8)	250mg	Visit strem.com
46-0392	Methanesulfonato(1,3,5,7-tetramethyl-8-phenyl-2,4,6-tri- oxa-8-phosphaadamantane)(2'-amino-1,1'-biphenyl- 2-yl)palladium(II) dichloromethane adduct, min. 98% [MeCgPPh Palladacycle Gen. 3]	500mg	Visit strem.com
46-0487	Methanesulfonato(2-dicyclohexylphosphino-2',6'-bis(dimethy- lamino)-1,1'-biphenyl)(2'-amino-1,1'-biphenyl-2-yl)palladium(II), 98% [CPhos Palladacycle Gen. 3] (1447963-73-6)	100mg	Visit strem.com
46-0935	Methanesulfonato{N-[2-(di-1-adamantylphosphino)phenyl] morpholine}(2'-amino-1,1'-biphenyl-2-yl)palladium(II) dichloro- methane adduct, min. 98% [Mor-Dalphos Palladacycle Gen. 3]	250mg	See page 64
46-0959	Methanesulfonato[4,6-bis(diphenylphosphino)phenoxazine] (2'-amino-1,1'-biphenyl-2-yl)palladium(II), 98% [NiXantphos Palladacycle Gen. 3] (1602922-03-1)	100mg	Visit strem.com
46-2128	Methanesulfonato[1,1'-bis(diphenylphosphino)ferrocene)] (2'-amino-1,1'-biphenyl-2-yl)palladium(II), min. 98% [DPPF Palladacycle Gen. 3] (1445086-28-1)	250mg	Visit strem.com
46-2135	Methanesulfonato[2-(di-t-butylphosphino)-2'-(N,N-dimeth- ylamino)-1,1'-biphenyl](2'-amino-1,1'-biphenyl-2-yl)palladi- um(II) dichloromethane adduct, min. 98% [t-BuDavePhos Palladacycle Gen. 3] (1445085-92-6)	250mg	Visit strem.com
46-2153	Methanesulfonato[2,2'-bis(diphenylphosphino)-1,1'-binaph- thyl](2'-amino-1,1'-biphenyl-2-yl)palladium(II), min. 98% [BINAP Palladacycle Gen. 3]	250mg	Visit strem.com
46-2158	Methanesulfonato(1,1'-bis(di-t-butylphosphino)ferrocene) (2'-amino-1,1'-biphenyl-2-yl)palladium(II), min. 98% [DTBPF Palladacycle Gen. 3]	250mg	Visit strem.com
46-2163	Methanesulfonato(2-di-t-butylphosphino-3,4,5,6-tetrameth- yl-2',4',6'-tri-i-propylbiphenyl)(2'-amino-1,1'-biphenyl-2-yl) palladium(II), min. 95% [Me4 t-ButylXPhos Palladacycle Gen. 3] (1507403-85-1)	100mg	Visit strem.com

KITS - CalB immo KIT™ - Immobilized enzyme

96-4050 New	CalB immo KIT [™] - Immobilized enzyme Store in dry conditions (2-8°C). Do not freeze. Shelf Life: 1 year; Sold in collaboration with Purolite for research purposes only. Components also available for individual sale. Contains the following:		
07-3130	CalB immo Plus™ - Immobilized enzyme	10g	See page 27
07-3142	CalB immo 8285™ - Immobilized enzyme	10g	See page 27
07-3148	CalB immo 8806™ - Immobilized enzyme	10g	See page 27
07-3152	CalB immo 5587™ - Immobilized enzyme	10g	See page 26
07-3155	CalB immo 1090™ - Immobilized enzyme	10g	See page 26
07-3159	CalB immo 5872™ - Immobilized enzyme	10g	See page 26

Item #	Immobilized on	Immobilization	Enzyme activity (PLU/g dry)
07-3130	DVB/methacrylate	Adsorption	>9,000
07-3142	Epoxy/butyl methacrylate	Covelent	>10,000
07-3148	Octadecyl methacrylate	Adsorption	>10,000
07-3152	Styrene/DVB copolymer	Adsorption	>4,000
07-3155	Macroporous styrene/DVB	Adsorption	>8,000
07-3159	Styrene/DVB	Adsorption	>3,500

Appearance: White to slightly yellow spherical beads, free from foreign matter

Pricipal Applications: Screening of immobilized lipases for process development, Esterifications (regio- and stereo-selective), Transesterfication, Amidation, Fats and oils modification

Advantages: Fast screening in process development, Wide selection of enzyme carriers for different applications

KITS - Enz	yme Carrier Lifetech™ ECRKIT1		
96-0255	Enzyme carrier Lifetech™ ECRKIT1		
NEW	Store in dry conditions (2-20°C). Do not freeze. Shelf Life: 5 year rapid screening of different methods of enzyme immobilization. S for research purposes only. Components also available for individual sale. Contains the following:	s; This enzym old in collabo	e carrier kit allows ration with Purolite
06-0810	Enzyme carrier Lifetech™ ECR8204F	50g	See page 29
06-0828	Enzyme carrier Lifetech™ ECR8285	50g	See page 33
06-0913	Enzyme carrier Lifetech™ ECR1090M	50g	See page 29
06-0925	Enzyme carrier Lifetech™ ECR1030M	50g	See page 27
07-1512	Enzyme carrier Lifetech™ ECR8309F	50g	See page 30
07-1532	Enzyme carrier Lifetech™ ECR8806F	50g	See page 32

KITS - Graphene Quantum Dots (GQDs) Master Kit

96-7410 New	Graphene Quantum Dots (GQDs) Master Kit Sold in collaboration with Dotz Nano Ltd. for research purposes only. Suggested use within 6 months of purchase. Do not freeze. Store in DARK. Components also available for individual sale. Contains the following:			
06-0330	Graphene Quantum Dots (GQDs), Aqua-Green Luminescent (1034343-98-0)	100mg	See page 48	
06-0332	Graphene Quantum Dots (GQDs) in water, Aqua-Green Luminescent (1034343-98-0)	100ml	See page 48	
06-0334	Graphene Quantum Dots (GQDs), Blue Luminescent (1034343-98-0)	100mg	See page 49	
06-0336	Graphene Quantum Dots (GQDs) in water, Blue Luminescent (1034343-98-0)	100ml	See page 49	
06-0340	Graphene Quantum Dots (GQDs) in water, Cyan Luminescent (1034343-98-0)	100ml	See page 49	

	Photoluminescence					
ltem #	QY+ *	λ max *	Max emission	FWHM *		
06-0330 / 06-0332	>17%	485 nm	525 nm	70 nm		
06-0334 / 06-0336	>65%	350 nm	445 nm	65 nm		
06-0340	>25%	420 nm	490 nm	80 nm		
Particle diameter: <5 nm Topographic height: 1.0 - 2.0 nm Concentration: 1mg/ml (for liquid items)						
Abbreviations: QY* = Quantum Yield; λ max = Maximum excitation wavelength; FWHM = Full width at half maximum						

KITS - Graphene Quantum Dots (GQDs) Mini Kit (Powders)

96-7425 New	Graphene Quantum Dots (GQDs) Mini Kit (Powders) Sold in collaboration with Dotz Nano Ltd. for research purposes only. Suggested use within 6 months of purchase. Do not freeze. Store in DARK. Components also available for individual sale. Contains the following:		
06-0330	Graphene Quantum Dots (GQDs), Aqua-Green Luminescent (1034343-98-0)	100mg	See page 48
06-0334	Graphene Quantum Dots (GQDs), Blue Luminescent (1034343-98-0)	100mg	See page 49

ltem #	Color & Form	Photoluminescence				
		QY+ *	λmax *	Max emission	FWHM *	
06-0330	dark red-brown pwdr.	>17%	485 nm	525 nm	70 nm	
06-0334	dark brown pwdr.	>65%	350 nm	445 nm	65 nm	
Particle diameter: <5 nm Topographic height: 1.0 - 2.0 nm						
Abbreviations: $QY^* = Quantum Yield$; $\lambda max = Maximum excitation wavelength$; FWHM = Full width at half maximum						

KITS - Graphene Quantum Dots in water (GQDs) Mini Kit (Liquids)

96-7420 NEW	O Graphene Quantum Dots in water (GQDs) Mini Kit (Liquids) Sold in collaboration with Dotz Nano Ltd. for research purposes only. Suggested use within 6 months of purchase. Do not freeze. Store in DARK. Components also available for individual sale. Contains the following:				
06-0332	Graphene Quantum Dots (GQDs) in water, Aqua-Green Luminescent (1034343-98-0)	100ml	See page 48		
06-0336	Graphene Quantum Dots (GQDs) in water, Blue Luminescent (1034343-98-0)	100ml	See page 49		
06-0340	Graphene Quantum Dots (GQDs) in water, Cyan Luminescent (1034343-98-0)	100ml	See page 49		

Item #	Color & Form	Photoluminescence			
		QY+ *	λ max *	Max emission	FWHM *
06-0332	cloudy orange liq.	>17%	485 nm	525 nm	70 nm
06-0336	cloudy colorless liq.	>65%	350 nm	445 nm	65 nm
06-0340	cloudy brown liq.	>25%	420 nm	490 nm	80 nm
Particle diameter: <5 nm Topographic height: 1.0 - 2.0 nm Concentration: 1mg/ml					

Abbreviations: $QY^* = Quantum Yield; \lambda \max = Maximum excitation wavelength; FWHM = Full width at half maximum$

KITS - Lipase immo Kit - Immobilized enzymes

96-4065 NEW

Lipase immo Kit - Immobilized enzymes

- The KIT contains 10g of each of the following:
- CalB immo Plus™(Lipase from Candida antarctica B)*07-3130*
- CalA immo (Lipase from Candida antarctica A)
- TL immo (Lipase from Thermomyces lanuginosa)
- RM immo (Lipase from Rhizomucor miehei)
- CR immo (Lipase from Candida rugosa)
- PS immo (Lipase from Pseudomonas cepacia)

A selection of immobilized lipases on different Lifetech[™] ECR enzyme carrier resins for screening purposes. Store in dry conditions (2-8°C). Do not freeze. Sold in collaboration with Purolite for research purposes only.

Components also available for individual sale. Contains the following:

Pricipal Applications: Esterifications (regio - and stereo-selective), Transesterfication, Kinetic resolution of racemic alcohols, amines, esters and triacylclycerides, Fats and oils modification, Hydrolysis of esters

Advantages: Fast screening in process development, Wide selection of immobilized lipases, Optimal for all applictions in organic solvents.

CN

100mg

KITS - Solvias Josiphos Nickel Catalyst Kit

96-3660 NEW

Ρh

Fe

28-0170

Solvias Josiphos Nickel Catalyst Kit Sold in collaboration with Solvias for research purposes only. Components also available for individual sale.

Су

28-0172

Contains the following: Мe Me Me T P(tBu)₂ PCv₂ PPh₂ Ph Су ,Cy Fe Fe



28-0170	Chloro(4-cyanophenyl){(R)-1-[(S)-2-(diphenylphosphino) ferrocenyl]ethyl(di-t-butyl)phosphine} nickel(II) (2049086-34-0)	100mg	See page 51
28-0172	Chloro(4-cyanophenyl){(R)-1-[(S)-2-(dicyclohexylphosphino) ferrocenyl]ethyl (dicyclohexylphosphine)}nickel(II) (2049086-35-1)	100mg	See page 50
28-0175	Chloro(4-cyanophenyl){(R)-1-[(S)-2-(dicyclohexylphosphino) ferrocenyl]ethyl (diphenylphosphine)}nickel(II) (2049086-36-2)	100mg	See page 51
28-0178	Chloro(4-cyanophenyl){(R)-1-[(S)-2-(bis(4-fluorophenyl) phosphinoferrocenyl]ethyl(di-t-butylphosphine)}nickel(II) (2049086-37-3)	100mg	See page 50

CN

100mg

Ċy

28-0175

CN

100mg

Visit strem.com for the following:

- New Product Announcements
- Searchable Catalog
- Technical Notes
- Safety Data Sheets (SDS)
- Certificates of Analysis (CofA)
- Product Pricing and Availability

If you are unable to find a product you need please contact us. Custom synthesis services are available.

Available Booklets























THE STREM CHEMIKER

Strem Chemicals, Inc.

7 Mulliken Way Newburyport, MA 01950-4098 U.S.A. Tel.: (978) 499-1600 Fax: (978) 465-3104 (Toll-free numbers below US & Canada only) Tel.: (800) 647-8736 Fax: (800) 517-8736

DUR LINE OF RESEARCH CHEMICALS

Metal Alkoxides & beta-Diketonates Metal Catalysts & Chiral Catalysts Biocatalysts & Organocatalysts Metal Carbonyls & Derivatives Metal Foils, Wires, Powders & **Wetal Acetates & Carbonates** Electronic Grade Chemicals Vietal Alkyls & Alkylamides Metal Halides. Hvdrides & Ligands & Chiral Ligands High Purity Inorganics & Metal Oxides, Nitrates, Chalcogenides Alkali Metals Deuterides Ionic Liquids Elements Fullerenes

Metal Scavengers Metallocenes Nanomaterials Organofiluorines Organometallics Porphines & Arsines Porphines & Prithalocyanines Precious Metal & Rare Earth Chemicals Volatile Precursors for MOCVD, CVD & ALD

Bulk Manufacturing Custom Synthesis Custom Synthesis Custom Facilities FDA Inspected Drug Master Files Complete Documentation

Check out our new website search capabilities. Follow us on Twitter and LinkedIn. View our blogs on our home page.

www.strem.com

