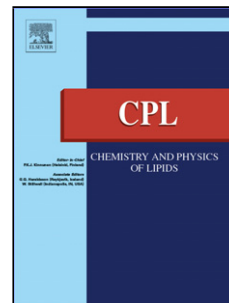


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4 **Antiproliferative effect of alkylglycerols as vehicles of butyric acid on colon cancer**
5 **cells**

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Abbreviations: AKG, 1-O-octadecyl glycerol; DMEM, Dulbecco's modified Eagle's medium; DMSO, dimethyl sulfoxide; D-SCAKG, 1-O-octadecyl-2,3-dibutyroilglycerol; FBS, fetal bovine serum; GC, gas chromatography; M-SCAKG, 1-O-octadecyl-2-butyroilglycerol; MTT, thiazolyl blue tetrazolium bromide; PAF, platelet-activating factor

16 **Abstract**

17 The anticarcinogenic activity of synthetic 1-O-octadecyl-2,3-dibutyroilglycerol (D-SCAKG)
18 in tumor-cell line of colonocytes (SW620) was performed. The effect of the previously
19 digested D-SCAKG under *in vitro* intestinal conditions was compared to the bioactivity of
20 non-digested D-SCAKG. Antiproliferative activity of each individual product from digestion
21 (1-O-octadecyl-2-butyroilglycerol; 1-O-octadecyl glycerol; butyric acid) was also performed.

22 The impact of solubilization of lipid products within micellar structures was also tested.

23 The 1-O-octadecyl glycerol was the most active compound, followed by 1-O-octadecyl-2-
24 butyroilglycerol, D-SCAKG and butyric acid. The 1-O-octadecyl glycerol and butyric acid
25 were the only molecules that showed antiproliferative effect in absence of micelles. Digested
26 D-SCAKG was 4-fold more effective than non-digested D-SCAKG. A synergism between 1-
27 O-octadecyl-2-butyroilglycerol and 1-O-octadecyl glycerol was evidenced.

28 As summary, the synthetic D-SCAKG seems to be an interesting antitumoral lipid against
29 colonocytes, especially after previous intestinal digestion, and mainly due to the synergism of
30 the major products, namely 1-O-octadecyl-2-butyroilglycerol and 1-O-octadecyl glycerol. At
31 the same time, 1-O-octadecyl-2-butyroilglycerol would constitute a stable esterified form of
32 butyric acid for its vehiculization.

33

34 **Keywords:** alkylglycerols, butyric acid, lipid digestion, lipid-delivery pro-drugs,
35 antiproliferative compounds

36

36 **1. Introduction**

37 Alkylglycerols, alkylglycerophospholipids and their derivatives, namely ether lipids, have
38 been the subject of much attention last years because of their special physiological functions
39 (Magnusson and Haraldsson, 2011). These glycerides contain O-alkyl or O-alk-1-enyl groups
40 at sn-1 position. Alkylmonoacylglycerols or alkyl-diacylglycerols can be also found. Ether
41 lipids are especially abundant as major components of total lipids in liver oil of marine
42 invertebrates and vertebrates, especially shark liver oil. In the human body, they can be found
43 in the cells of the immune system and in human breast milk as membrane components and
44 cellular signaling molecules (Magnusson and Haraldsson, 2011; Torres et al., 2011). As anti-
45 tumour agents, they have been related to inhibitory growth, antimetastatic activity, anti-
46 neoangiogenic action, and induce differentiation and apoptosis in cancer cells (Brohult et al.,
47 1978; Deniau et al., 2011; Krotkiewski et al., 2003; Pedrono et al., 2004; Pedrono et al.,
48 2007). Immune stimulation properties have also been attributed to these substances
49 (Braverman and Moser, 2012; Kantah et al., 2012; Mitre et al., 2005).

50 Within ether lipids, one of the most popular is the platelet-activating factor (PAF), a 1-O-
51 alkyl-2-acetyl-sn-glycero-3-phosphocholine, with diverse cell mediator effects in a variety of
52 tissues and systems, including circulation, inflammation, development and reproduction
53 (Hanahan, 1986; Prescott et al., 2000). Other important ether lipids are 1-O-alkyl-sn-
54 glycerols, which have been also claimed to display various beneficial effects on human health
55 and seems to amplify the production of PAF by incorporation within the PAF precursor,
56 namely the 1-O-alkyl-2-acyl-sn-glycero-3-phosphocholine (Himachi et al., 1997; Kantah et
57 al., 2012). On the other hand, 1-O-alkyl-2-acylglycerols are considered of interest as analogs
58 of diacylglycerols, which would inhibit protein kinase C. The 1-O-alkyl-2-acylglycerols are
59 produced following cell stimulation after alkylglycerol incorporation in membrane
60 phospholipids. These products bind but do not activates protein kinase C- ϵ , and compete with

61 diacylglycerol, which could result in cell growth arrest (Deniau et al., 2011; Heymans et al.,
62 1987; Houk et al., 2008; Marigny et al., 2002).

63 Taking into account the attractive properties of these lipids, synthetic ether lipids that do not
64 occur in nature have been prepared to find therapeutic agents (Hartvigsen et al., 2006), such
65 as PAF-like lipids, which retain a short-chain residue at the sn-2 position other than an acetyl
66 group (Tokumura, 1995). In this sense, it has been shown that the shorter the sn-2 chain
67 residue the more active the PAF-like lipid (Kern et al., 1998; Tanaka et al., 1995).

68 Recently, Torres et al. (2009 a) synthesized structured alkyldiacylglycerols containing
69 residual short-chain fatty acids at sn-2 and sn-3 locations by enzymatic transesterification of
70 1-O-octadecyl glycerol and ethyl butyrate (1-O-octadecyl-2,3-dibutyroilglycerol; D-SCAKG).

71 Ethyl butyrate was not only chosen by the nature of “short chain residue”, but also on the
72 bioactive importance of butyrate by itself. Besides having physiological role as the main
73 metabolic fuel for colonocytes and control of colonic inflammation, butyrate seems to
74 interfere with the pathogenesis of diverse cancers by inhibiting cell proliferation or inducing
75 apoptosis, such as colorectal cancer, hepatocarcinoma, leukemia, breast and prostate cancer
76 (Avivi-Green et al., 2001; Coradini et al., 1999; Kuefer et al., 2004; Kuroiwa-Trzmielina et
77 al., 2009; Ooi et al., 2009). However, the potential application of butyrate as antitumor agent
78 is limited by the problem to reach enough plasma concentrations required to exert its
79 antiproliferative/differentiating actions. Moreover, it is rapidly metabolized, showing a short
80 half-life (Kuroiwa-Trzmielina et al., 2009; Li et al., 2009). Therefore, there is a current
81 interest on overcoming these drawbacks in order to allow its application as therapeutic agent.

82 Pro-drugs, such tributyrin, have been proposed as alternative of natural butyrate (Clarke et al.,
83 2008; Heidor et al., 2012; Kuroiwa-Trzmielina et al., 2009). Torres et al. (2009 a) suggested
84 that the vehiculization of butyric acid by alkylglycerol backbone might be also proposed as
85 interesting butyric pro-drug, due to the double advantage of alkylglycerols as potential lipid

86 vehicle of butyrate and the own bioactivity of each individual compound, alkylglycerol and
87 butyrate, within the same molecule.

88 Preliminary studies have been performed in order to validate the synthesized D-SCAKG
89 molecule. As first approach, the potential bioaccessibility was assayed by simulating the
90 intestinal digestion of the molecule by pancreatic enzymes under *in vitro* conditions, since it
91 would be the preliminary step in case of oral intake of this pro-drug (Martin et al., 2011). It
92 was demonstrated that the major hydrolysis product after intestinal digestion of D-SCAKG
93 was the molecule 1-O-octadecyl-2-butyroilglycerol, together with free butyric acid, 1-O-
94 octadecyl glycerol and minor non-digested D-SCAKG. Therefore, regardless of the short-
95 chain as butyrate, the major digestion product would be the potential bioactive structure under
96 the form of 1-O-alkyl-2-acylglycerol. On the other hand, regardless of the alkylglycerol
97 backbone, the major digestion product 1-O-alkyl-2-butyroilglycerol was a stable esterified
98 form of butyrate. This was considered a positive result since it has been stated that efficient
99 pro-drugs of butyrate should have a sufficiently stable bond between the carrier and butyrate
100 residue to increase its *in vivo* half-life (Coradini et al., 1999). In fact, in the performed *in vitro*
101 intestinal digestion study, it was shown that the traditional tributyrin proposed as pro-drug of
102 butyrate was completely hydrolyzed to free butyric acid, which would not be the desired
103 situation in case of butyrate pro-drugs. The potential of D-SCAKG as carrier of butyrate was
104 concluded. Furthermore, the additional bioactive ether lipid, namely 1-O-octadecyl glycerol,
105 commonly known as batyl alcohol, was also an interesting product of the *in vitro* intestinal
106 digestion of the molecule D-SCAKG. Batyl alcohol has been previously tested as anti-tumour
107 lipid (Ando et al., 1972) or antiinflammatory agent (Burford and Gowdey, 1968). Therefore, all
108 the digestion products of the synthetic D-SCAKG after *in vitro* intestinal lipolysis might be
109 considered attractive products for bioactive properties.

110 Taking into account all these preliminary data, the aim of the current study was to study a
111 potential anticarcinogenic activity of the synthetic D-SCAKG in tumor-cell line of
112 colonocytes. To perform this study, the digested D-SCAKG under *in vitro* intestinal
113 conditions was tested, since this last would be the more probable situation in case of oral
114 intake of this molecule, where the hydrolysis products would be the available compounds for
115 intestinal cells. The bioactivities of hydrolysis products of digestion were also tested
116 individually in order to explain the obtained results properly. Furthermore, the impact of
117 solubilization of lipid products within micellar structures of bile salts and phospholipids on
118 bioactivities was also tested, as it would be the most probable situation after *in vitro* intestinal
119 digestion into the lumen gut environment (Martin et al., 2011).

120

121 **2. Materials and methods**

122 *2.1. Reagents and Materials*

123 The D-SCAKG was synthesized according to a methodology previously described by our
124 group (Torres et al., 2009 a). Trizma, maleic acid, pancreatin, bile salts, phosphatidyl choline
125 from egg yolk, butyric acid and 1-O-octadecyl glycerol were purchased from Sigma-Aldrich
126 Chemie GmbH (Steinheim, Germany). Hydrochloric acid, sodium sulphate anhydrous,
127 sodium chloride, calcium chloride were from Panreac (Barcelona, Spain). N-dodecane for
128 synthesis was purchased from Merck (Darmstadt, Germany). All solvents used were of HPLC
129 grade from Lab-Scan (Dublin, Ireland). Human colon cancer cells (SW620) were obtained
130 from ATCC (Rockville, MD, USA). Dulbecco's modified Eagle's medium (DMEM), fetal
131 bovine serum (FBS), glutamine and antibiotic-antimycotic solution were from Gibco BRL
132 (Grand Island, NY, USA).

133

134 *2.2. In vitro lipid digestion of D-SCAKG*

135 The *in vitro* lipid digestion model was performed according to Martin et al. (2011). Sample (1
136 g) of D-SCAKG was mixed with 0.5 g of bile salts, 0.2 g of lecithin, 5mM CaCl₂, 150mM
137 NaCl and 54 mL of 1 M Trizma-maleate buffer pH 7.5. The mixture was homogenized (Ultra-
138 Turrax IKA T18) for 20 min at 7000 rpm. The homogenized was placed in a thermostatically
139 controlled vessel (37°C) under continuous stirring by magnetic stir bar at 1000 rpm.
140 Simulation of intestinal digestion was started by addition of fresh pancreatin extract (1 g of
141 pancreatin in 6 mL of Trizma-maleate buffer pH 7.5 stirred during 10 min and centrifuged at
142 1600 x g during 15 min) and continued up to 60 min. *In vitro* digestion of each sample was
143 performed in triplicate.

144 The total lipids from samples were extracted by hexane/methyl-tert-butyl ether (50:50, v/v).
145 Furthermore, this medium was acidified by hydrochloric acid (150 mM) in order to stop the
146 enzymatic reaction and to enhance the recovery of butyric acid. n-Dodecane (10 mg) was
147 added as internal standard. The mixture was vortexed for 1 min and centrifuged for 10 min at
148 15,000 rpm. Organic phase containing separated lipids was collected and anhydrous sodium
149 sulfate was added before further analysis.

150 Hydrolysis products were determined according to Torres et al. (2009 b) by gas
151 chromatography (GC) (Hewlett-Packard 5890 series II) with on-column injection using a 7 m
152 5% phenyl methyl silicone capillary column (Quadrex Corporation, New Haven, CT), 0.25
153 µm i.d. A deactivated column of 12 cm 530 µm i.d. was used as pre-column. Injector and
154 detector temperature was 43°C and 360°C, respectively. The temperature program was as
155 follows: starting at 40°C and then heating to 250°C at 42°C min⁻¹ with 10 min hold, followed
156 by heating from 250°C to 325°C at 7.5°C min⁻¹ with 30 min hold. Helium was used as carrier
157 gas at a pressure of 5.2 psi. The peaks were computed using GC chemstation software
158 (Agilent Technologies, Santa Clara, CA) and quantified according to the internal standard n-
159 dodecane.

160 The lipid composition of the digested D-SCAKG is shown in Table 1. This composition was
161 subsequently simulated by mixing the individual compounds at the proper proportions
162 according to Table 1 in order to perform the *in vitro* cell assays. Previously, the individual
163 hydrolysis product 1-O-octadecyl-2-butyroilglycerol (M-SCAKG) from *in vitro* digestion of
164 D-SCAKG was purified by semi-preparative HPLC, since it cannot be found as commercial
165 standard. The rest of hydrolysis products, namely butyric acid and 1-O-octadecyl glycerol
166 (AKG) were commercial standards.

167

168 2.3. *In vitro* colonocytes assay

169 2.3.1. Cell culture

170 Human colon cancer cells SW620 were cultured in DMEM supplemented with 10% FBS,
171 2mM glutamine and 1% of antibiotic-antimycotic solution (containing 10,000 units/mL of
172 penicillin base, 10,000 µg/mL of streptomycin base and 25,000 ng/mL of amphotericin B).
173 They were maintained under standard conditions of temperature (37°C), humidity (95%) and
174 carbon dioxide (5%).

175

176 2.3.2. Cell viability assay

177 The antiproliferative activity of D-SCAKG and its derivatives was measured by MTT (thiazolyl
178 blue tetrazolium bromide) assay. Cells in exponential growth phase were placed in 96-well
179 plates, using 6000 cells per well in a final volume of 200 µL. After 24h incubation, cells were
180 treated by replacing medium with new culture medium (blank wells), medium containing
181 different concentrations of the tested products, or micelles suspensions. Stocks of all
182 treatment compounds, both for mixing directly with DMEM as well as for preparing micelles,
183 were prepared in Ethanol.

184 Cell viability was determined at the same time of treatment and after 48h. To determine the
185 number of viable cells, 20 μ L of MTT solution (5 mg/mL in PBS) was added to each well and
186 incubated during 3 h, subsequently the medium was removed and 200 μ L of DMSO (dimethyl
187 sulfoxide) was added in order to lyse the cells and resuspend the formazan (MTT metabolic
188 product). Quantities of formazan product, which is directly related to the number of viable
189 cells, were measured at 560 nm using a scanning spectrophotometer microplate reader (UVM
190 340 Biochrom, Cambridge). Concentration values corresponding to the parameters IC50
191 (50% cell viability inhibition), GI50 (50% growth inhibition), TGI (total growth inhibition)
192 and LC50 (50% cell death) were calculated according to the NIH definitions using a logistic
193 regression.

194

195 2.3.3. Preparation of Mixed Micellar and Control Solution

196 Micelles were prepared with egg yolk lecithin and sodium taurocholate at final concentrations
197 of 150 μ M and 0.5mM respectively. Micelle solutions were prepared by first adding
198 appropriate amounts of each treatment compounds to a glass vial containing 0.5ml of hexane.
199 For control solutions, a vial with 0.5ml of hexane was prepared. Vials content were dried
200 using a nitrogen evaporator (N-Evap 111, Organomation Associates). Then 0.23 mg of
201 lecithin dissolved in 0.575 ml of hexane was added to each vial and dried again in the same
202 conditions; this was followed by the addition of 500ul of 2mM taurocholate prepared in
203 ethanol and another evaporation step. Finally, dried content was resuspended in 1ml of
204 DMEM without supplements and mixed with 1ml of complete DMEM, and 200ul of this
205 mixture was added to each well.

206

207 **3. Results and discussion**

208 In order to analyze the potential anticarcinogenic activity of the synthetic D-SCAKG in
209 tumoral colonocytes, the bioactivity of digested D-SCAKG under *in vitro* intestinal conditions
210 was evaluated. As a first approach, the antiproliferative activity of each individual lipid
211 product (D-SCAKG, M-SCAKG, AKG and butyric acid) was determined as reference. In
212 addition, since the formation of micellar structures might constitute the most usual situation
213 under intestinal digestion into the lumen gut environment (Martin et al., 2011), the impact of
214 solubilization of lipid products within micellar structures of bile salts and phospholipids on
215 the observed bioactivity was also tested, showing a noticeable difference when determining the
216 effect of the lipid product under free form (without micellization) or previously solubilized
217 within mixed micelles.

218 As it can be observed in Figure 1, when compounds were added without previous inclusion in
219 micellar structures of bile salts and phospholipids, the AKG and butyric acid were the only
220 molecules that showed a bioactive effect at the assayed concentrations in tumor colonocytes,
221 whereas lack of activity was detected for D-SCAKG and M-SCAKG. Under the free forms,
222 the greatest activity was found for AKG (μM dose range) compared to butyric acid (mM dose
223 range), accordingly with previous studies concerning other alkylglycerols or butyric prodrugs
224 (Reynolds et al., 2000; Wächtershäuser and Stein, 2000). Concerning the specific case of
225 AKG, the obtained bioactivity of this compound was not in agreement with the previous study
226 of Deniau et al. (2011), because these authors reported a lack of antitumour effect of the same
227 AKG (1-O-octadecyl glycerol) compared to other individual alkylglycerols. Nevertheless,
228 such study was performed under *in vivo* conditions in solid grafted tumours in mice after oral
229 administration, and studying the effect on tumour volume, spleen weight, or lung metastases,
230 so the conditions would not be comparable to the present study. On the contrary, Ando et al.
231 (1972) showed a positive anti-tumor activity of batyl alcohol (1-O-octadecyl glycerol) in
232 Ehrlich carcinoma in mice, but such activity was worse than that for alkylglycerols with

233 shorter alkyl chains. The divergences between our and previous studies show the interest on
234 go in depth on the bioactivities of individual alkylglycerols as potential antitumor agents,
235 because such information is scarce.

236 When lipid compounds were added under a previous micellization with bile salts and
237 phospholipids, bioactivities were also found for D-SCAKG and M-SCAKG. Again, the best
238 antiproliferative effect was found for AKG, and the bioactivities for D-SCAKG and M-
239 SCAKG were found at higher concentrations, being similar between them (Figure 1). Control
240 assays of micelles without inclusion of the experimental compounds showed a lack of effect
241 of such vehicles.

242 With the aim of further evaluating the bioactivity of the different conditions, growth
243 inhibition and promotion of cell death were analyzed by IC50, GI50, TGI and LC50
244 calculation in those conditions in which a bioactive effect was initially observed. As it is
245 shown in Table 2, AKG displayed greater bioactivity than D- and M-SCAKG even without
246 including in micelles. Under micellization, AKG resulted 5-fold more active than the other
247 two lipid compounds (48.3 in cell sensitivity to AKG with respect to 282.02 and 237.04 for
248 D- and M-SCAKG respectively, and 65.38 in citostaticity with respect to 385.28 and 395.09).
249 Regarding cell death, AKG was the only compound displaying an efficient cytotoxic effect at
250 reached concentrations (85.9uM for LC50).

251 According to these results, it seems that the potential bioactivity of ether lipids in tumor
252 colonocytes would be different depending on the esterification of the alkylglycerol backbone.
253 Thus, either alkylmonoacylglycerols or alkyldiacylglycerols would be worse antiproliferative
254 agents than non-esterified alkylglycerols. Nevertheless, it should be remarked that
255 alkylmonoacylglycerols would be slightly better than alkyldiacylglycerols, since a value for
256 LC50 was found for the former (M-SCAKG), whereas lack of LC50 dose was detected for
257 alkyldiacylglycerols (D-SCAKG) when they were included under the form of mixed micellar

258 structures. The observed differences could be related to the own molecular size of the
259 different structures of alkylglycerols. This reason could be also linked to the different
260 solubility and/or polarity of the molecules. Thus, the presence of two hydroxyl groups in the
261 case of AKG compared to the one or two acyl chains of butyrate in the case of M-SCAKG or
262 D-SCAKG, respectively, might lead to better solubility of the AKG molecule in the aqueous
263 media of the cell culture, allowing an easier access to colonocytes.

264 Regardless on the different bioactivities between compounds, the vehiculization of ether
265 lipids within micellar structures enhanced their antiproliferative effects. Thus, whereas lack of
266 activity was observed for D-SCAKG and M-SCAKG after direct addition to the media, a
267 significant antiproliferative effect was observed under micellar solubilization conditions.
268 Accordingly, AKG-mediated inhibition of tumoral cell growth and promotion of cell death
269 were enhanced by micellization. These results would be in agreement with the expected
270 situation at intestinal lumen, where most lipid compounds need the support of mixed micelles
271 to be dispersed in the aqueous media to cross the unstirred water layer close to enterocytes
272 previous to absorption (Christensen et al., 1995; Ramirez et al., 2001). At this respect, the
273 non-esterified form of AKG might allow an easier formation of mixed micelles compared to
274 D-SCAKG and M-SCAKG due to the availability of the two hydroxyl groups of the
275 molecule, which confers emulgent-like properties, allowing its dispersion into mixed micelles
276 and in turn the accessibility to colonocytes.

277 The obtained results contributed to the knowledge of the bioactivity of ether lipids, because to
278 the best of our knowledge, previous studies showing differences between non-esterified and
279 esterified alkylglycerols as antiproliferative agents by *in vitro* tumor-cell models have not been
280 found. Furthermore, the effect of micellization on the efficacy of these compounds on
281 intestinal cells has not been previously reported.

282 Nevertheless, it is important to remark that the results found for esterified forms of AKG
283 might be considered an artifact in the particular case of an oral intake of D-SCAKG. This is
284 because, as most esterified lipids, these are partially hydrolyzed at intestinal tract by digestive
285 lipases. Subsequently, the hydrolysis products are those major included in mixed micelles of
286 bile salts and phospholipids to allow their access to intestinal cells. In fact, according to Table
287 1 and in agreement to Martin et al (2011), the simulation of intestinal digestion of D-SCAKG
288 produced a stable form of esterified butyric acid as 1-O-octadecyl-2-butyroilglycerol (M-
289 SCAKG) as major hydrolysis product (> 50%). Free AKG (> 20%) and free butyric acid
290 (20%) were also released as bioactive compounds of interest, together with minor residual D-
291 SCAKG. Due to that, it was approached that the bioactivity of this mixture of compounds
292 solubilized in micellar structures, namely M-SCAKG, AKG, butyric acid and residual D-
293 SCAKG, should be tested for validate the bioactivity of the synthetic D-SCAKG in tumor
294 colonocytes, in order to simulate the potential situation after oral intake of this lipid.

295 The bioactivity of digested D-SCAKG compared to that of non-digested D-SCAKG in shown
296 in Figure 2. Furthermore, both compounds were also compared in terms of growth inhibition,
297 citostaticity and citotoxicity in Table 3.a. In addition, data for each individual component of
298 the mixture according to the proportions of Table 1 was estimated (Table 3.b). As it is shown
299 in Figure 2 and Table 3.a, the digested D-SCAKG was around 4-fold more effective than the
300 former non-digested D-SCAKG. Therefore, this result suggests that the digestion of D-
301 SCAKG would improve efficiency on its antiproliferative activity of this synthetic lipid after
302 oral intake. Of note, taking into account the dose of each individual compound in the mixture
303 of hydrolysis products (Table 3.b), much lower effective concentrations were found for all
304 compounds when they were mixed (Table 3.b) than when they were tested individually (Table
305 2). Especially attractive was the finding that the bioactivity of AKG, the most effective
306 compound individually, was greater when this compound was added together with D-

307 SCAKG, M-SCAKG and butyric acid. Thus, 86 μM of AKG was necessary to reach the
308 LC50 value when it was added individually to cells (Table 2), whereas such value was
309 reduced by three times (26 μM LC50) when AKG was included in the micellized media
310 together with the rest of hydrolysis compounds (Table 3.b). Similarly, IC50, GI50 and TGI
311 reduced by twice. Taking into account that AKG was only present in the digested mixture at
312 around 20% of total lipids (Table 1), whereas it was 100% when added individually, these
313 results suggest a potential synergism between the hydrolysis products, promoting the efficacy
314 of the individual compounds and in turn the total efficacy of the mixture.

315 In order to study the interaction of compounds further additional experiments were performed
316 by the combination of the lipid products at the same proportion that obtained after *in vitro*
317 digestion, except that butyric acid was not added to the media. This lead to a lipid mixture
318 mainly formed by the major hydrolysis products M-SCAKG and AKG (68 and 29 g/100 g
319 total lipids, respectively) . As shown in Figure 3, the combination of M-SCAKG and AKG
320 showed almost the same results as found for the combination of all compounds, suggesting
321 that the active compounds of the digested D-SCAKG were its major hydrolysis products,
322 namely M-SCAKG and AKG. Furthermore, this result confirmed a synergism between M-
323 SCAKG and AKG, since the antiproliferative dose of both compounds in the mixture was
324 lower than the dose needed for each individual compounds separately (Table 3.b and Table 2).
325 These results suggest two promising activities for the major hydrolysis product of D-SCAKG
326 after *in vitro* intestinal digestion, namely M-SCAKG. On one hand, it would contribute
327 together with AKG to a potential antitumoral activity against tumor colonocytes, and, on the
328 other hand, M-SCAKG would constitute a stable esterified form of butyric acid for its
329 vehiculization. This is because efficient pro-drugs of butyrate should have a sufficiently stable
330 bond between the carrier and butyrate residue to increase its *in vivo* half-life (Coradini et al.,
331 1999).

332 Additional studies might be conducted to further explain the mechanism of synergism
333 between M-SCAKG and AKG. At this respect, concerning shark liver oil as the main natural
334 source of alkylglycerols, it has been suggested that the bioactivity of such oil might be likely
335 mediated by the presence of alkylglycerols (Pedrono et al., 2004), but the interaction with
336 other lipid species in the oil (free alkylglycerols, mono- and di-esterified alkylglycerols with
337 different fatty acids, and acylglycerols) could not be rejected. One explanation of the observed
338 results of synergism might be related to the formation of micellar structures, both in the
339 number or size of micelles. This is because in general, it has been shown that higher surface,
340 extension and swelling of micelles with hydrolysis products enhance the inclusion of the rest
341 of lipophilic compounds, and in turns their bioaccessibility (Porter and Charman, 2001).
342 Similarly, it might be possible that the combination of M-SCAKG and AKG led to the
343 formation of higher number and/or surface of micellar structures, compared to the sole
344 presence of each individual component, allowing the access to colonocytes. According to this
345 approach, further studies were performed by counting the number of micellar structures by
346 light microscope from micelles preparations of AKG, M-SCAKG or their combination.
347 However, no conclusive results were found (data not shown).

348 On the other hand, it would be also interesting to elucidate whether the observed responses
349 might be also related to a synergistic action of M-SCAKG and AKG on the own colonocytes
350 once the compounds were absorbed. In this regard is important to remark that the M-SCAKG
351 molecule showed the typical structure of those analogues of diacylglycerols, namely 1-O-
352 alkyl-2-acylglycerols, which have been pointed out as inhibitors of protein kinase C (Deniau
353 et al., 2011; Marigny et al., 2002; Heymans et al., 1987). Additionally, once inside the
354 enterocyte the own forms of alkylglycerols would be derived to other diverse bioactive forms
355 under the fast dynamic of lipid remodeling, leading to alternate fatty acids at sn-2 location

356 different to butyric acid, via the deacylation-reacylation pathway of the Lands's cycle (Perez
357 et al., 2006).

358 In summary, the synthetic D-SCAKG seems to be an interesting antitumoral lipid against
359 colonocytes, especially after previous intestinal digestion, after solubilization in mixed
360 micelles, and mainly due to the synergism of the major products, namely 1-O-octadecyl-2-
361 butyroylglycerol and 1-O-octadecyl glycerol. At the same time, 1-O-octadecyl-2-
362 butyroylglycerol would constitute a stable esterified form of butyric acid for its vehiculization.
363 The current study also showed the importance of considering the digestive hydrolysis process
364 of lipid compounds on the evaluation of bioactivities at intestinal level, as well as the
365 importance of vehiculize the lipid products as it would be at intestinal lumen, namely under
366 the form of mixed micelles. Nevertheless, further studies would be necessary in order to
367 explain the observed effects related to the intracellular metabolism once the lipid compounds
368 were absorbed.

369

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376

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- 479

479 **Figure captions**

480

481 **Figure 1. Antiproliferative effect of free and micellized form of the different lipid**
482 **compounds.** Bioactivity was determined on sw620 human colon cancer cells by the MTT
483 (thiazolyl blue tetrazolium bromide) assay. Butyric acid (d) was not included in the study of
484 micellization since it is solubilized in the aqueous media. Data represent the mean +/- s.e.m. of
485 at least 2 independent experiments, each performed in quadruplicate. AKG, 1-O-octadecyl
486 glycerol; D-SCAKG, 1-octadecyl-2,3-dibutyroilglycerol; M-SCAKG, 1-O-octadecyl-2-
487 butyroilglycerol.

488

489 **Figure 2. Antiproliferative effect of digested D-SCAKG on sw620 human colon cancer**
490 **cells.** To test this compound, D-SCAKG, M-SCAKG, AKG and butyric acid were combined
491 at the same proportion that obtained after in vitro digestion. AKG and their esterified forms
492 were included in micelles, and butyric acid was added after micelle resuspension in culture
493 medium. Data represents the mean +/- s.e.m. of at least 2 independent experiments each
494 performed in quadruplicate. For comparative reasons, data are expressed at molar
495 concentration of D-SCAKG equivalent. In the case of digested D-SCAKG equivalents, they
496 were estimated from the weight of lipid mixture of hydrolysis products, which corresponded
497 to a weight of non-digested D-SCAKG and, in turn to a specific molar concentration of D-
498 SCAKG.

499 AKG, 1-O-octadecyl glycerol; D-SCAKG, 1-octadecyl-2,3-dibutyroilglycerol; M-SCAKG, 1-
500 O-octadecyl-2-butyroilglycerol

501

502 **Figure 3. Comparison of antiproliferative effect of digested D-SCAKG with a**
503 **preparation in which butyric is excluded.** M-SCAKG and AKG constitute the 97% of total
504 composition. All hydrolysis products were combined at the same proportion that obtained after
505 in vitro digestion. AKG and their esterified forms were included in micelles, while butyric
506 acid (when used) was added after micelle resuspension in culture medium. Data represent the
507 mean +/- s.e.m. of at least 2 independent experiments each performed in quadruplicate. For
508 comparative reasons, data are expressed at molar concentration of D-SCAKG equivalent, as
509 explained in figure 2.

510 AKG, 1-O-octadecyl glycerol; D-SCAKG, 1-octadecyl-2,3-dibutyroilglycerol; M-SCAKG, 1-
511 O-octadecyl-2-butyroilglycerol

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Table 1. Lipid composition of digested D-SCAKG by *in vitro* intestinal model

	g/100 g total lipids
D-SCAKG ^a	2.2
M-SCAKG ^b	54.4
AKG ^c	23.4
Butyric acid	20.0

^a D-SCAKG (1-octadecyl-2,3-dibutyroilglycerol); ^b M-SCAKG (1-O-octadecyl-2-butyroilglycerol); ^c AKG (1-O-octadecyl glycerol)

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514

514 **Table 2.** IC50, GI50, TGI and LC50 parameters of the esterified forms of AKG, as well as non-
 515 esterified AKG and butyric acid. These parameters were calculated from the results obtained in
 516 viability assays according to the NIH definitions using a logistic regression. Values are represented as
 517 the mean \pm s.e.m. of at least 2 independent experiments each performed in quadruplicate. NLO: No
 518 Lethality Observed.
 519

	D-SCAKG		M-SCAKG		AKG				BUTYRIC	
	micelles		micelles		medium		micelles		medium	
	Conc. (μ M)	s.e.m.	Conc. (μ M)	s.e.m.	Conc. (μ M)	s.e.m.	Conc. (μ M)	s.e.m.	Conc. (μ M)	s.e.m.
IC50	282,02	\pm 65,26	237,04	\pm 44,46	101,87	\pm 7,80	48,73	\pm 7,16	3791,95	\pm 1,13
GI50	150,52	\pm 28,89	178,79	\pm 31,58	64,13	\pm 1,84	26,94	\pm 6,94	1527,46	\pm 0,02
TGI	385,28	\pm 31,15	395,09	\pm 34,06	116,04	\pm 20,30	65,38	\pm 14,74	4551,65	\pm 1,33
LC50	NLO		732,76	\pm 20,93	130,56	\pm 25,57	85,90	\pm 23,14	11949,70	\pm 2,55

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Table 3. IC50, GI50, TGI and LC50 parameters of digested AKG on sw620 human colon cancer cells. (a) Digested media versus non-digested media. Values were calculated as mean values +/- s.e.m. of at least 2 independent experiments each performed in quadruplicate. For comparative reasons, data are expressed at molar concentration of D-SCAKG equivalent. In the case of digested D-SCAKG equivalents, they were estimated from the weight of lipid mixture of hydrolysis products, which corresponded to a weight of non-digested D-SCAKG and, in turn to a specific molar concentration of D-SCAKG. (b) Calculated IC50, GI50, TGI and LC50 for each product of digestion taking into account their respective proportion in the mixture of digested D-SCAKG.

a)

	Digested D-SAKG (in micelles)			D-SCAKG (in micelles)		
	(μM D-AKG equivalent)		s.e.m.	Conc. (μM)		s.e.m.
IC50	74,0	±	13,7	282,02	±	65,26
GI50	50,8	±	12,0	150,52	±	28,89
TGI	79,8	±	15,4	385,28	±	31,15
LC50	85,2	±	9,7	no data	±	

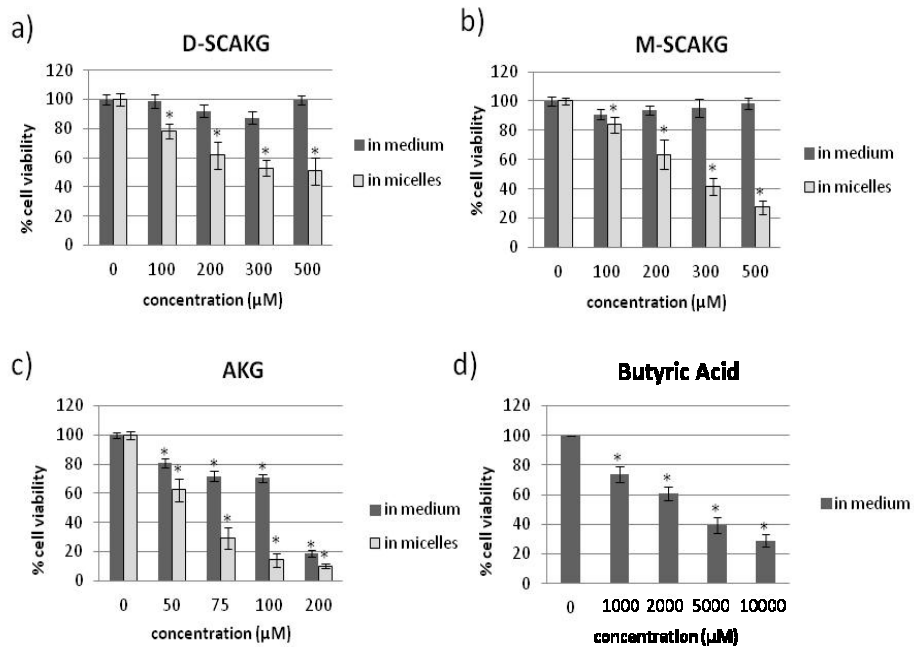
b)

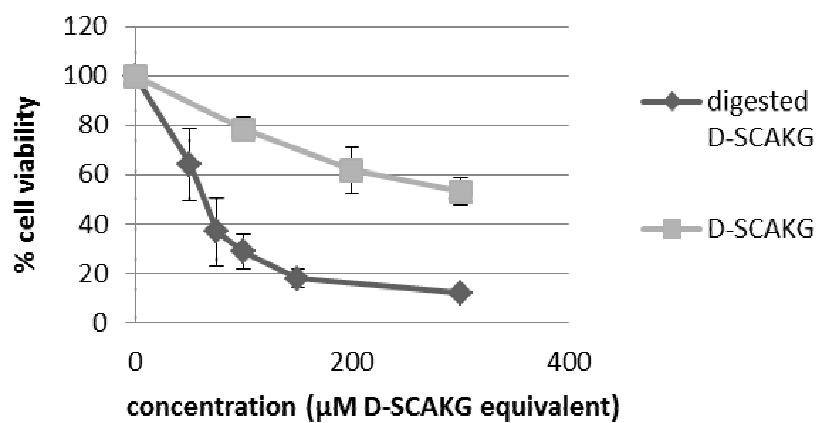
Digested D-SAKG	D-SAKG	M-SAKG	AKG	BUTIRIC
	μM	μM	μM	μM
74 μM (IC50)	1,3	44,3	22,7	78,32
50,8 μM (GI50)	0,89	30,41	15,58	53,77
79,8 μM (TGI)	1,4	47,77	24,48	84,46
85,2 μM (LC50)	1,5	51	26,14	90,17

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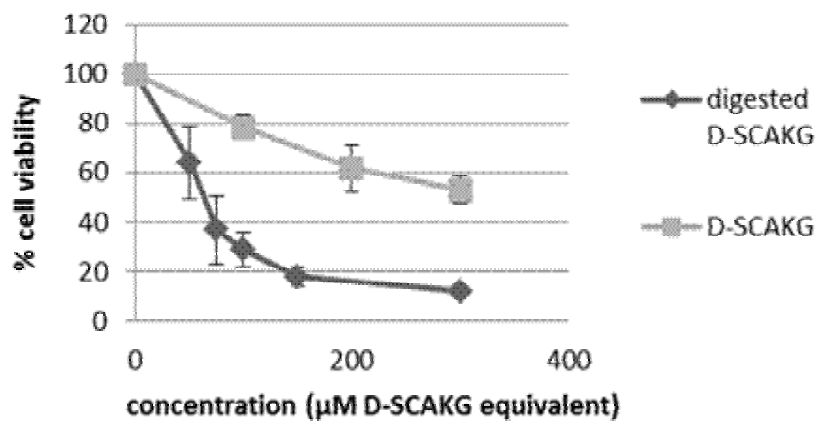
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Figure 1



555 **Figure 2**

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561 **Figure 3**

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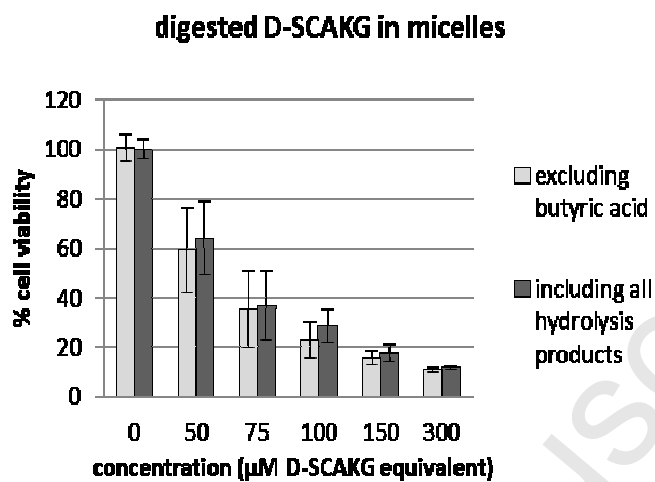
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571 **Highlights**

572 Intestinal digestion of alkylglycerols diesterified with butyric acid was simulated.

573 The antitumor effect of digested alkylglycerol was performed in tumor colonocytes.

574 Digested alkylglycerol was 4-fold more effective than the non-digested one.

575 Synergism between major digestion products was found.

576 Diesterified alkylglycerol would be a stable lipid for vehiculization of butyric acid.

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