#### BACHELOR OF SCIENCE IN INDUSTRIAL CHEMISTRY (Qualification type: Professional Bachelor's Degree) Qualification code: BPIY20 - NQF Level 8 (480 credits) SAQA ID: 111429. CHE NUMBER: H/H16/E114CAN

Campus where offered:

Arcadia Campus

#### REMARKS

- a. Admission requirement(s) and selection criteria:
- APPLICANTS WHO OBTAINED A SENIOR CERTIFICATE BEFORE 2008:

#### Admission requirement(s):

A Senior Certificate with a matriculation endorsement or equivalent qualification, with a C symbol at Higher Grade or a B symbol at Standard Grade for English, Mathematics, and Physical Science.

APPLICANTS WHO OBTAINED A NATIONAL SENIOR CERTIFICATE IN OR AFTER 2008:

#### Admission requirement(s):

A National Senior Certificate or an equivalent qualification, with a bachelor's degree endorsement, or an equivalent qualification, with an achievement level of at least 4 for English (home language or first additional language), 5 for Mathematics and 5 for Physical Sciences.

#### Selection criteria:

To be considered for this qualification, candidates must have an Admission Point Score (APS) of at least **24** (excluding Life Orientation).

b. Assessment procedure(s):

No further assessment will be done. Applicants who achieve the minimum APS will be considered until the programme complement is full. All completed applications received within the published due dates will be ranked. After consideration of the Departmental Student Enrolment Plan, only the top ranking applicants will be selected. Once a programme is full, a waiting list will be in place to provide an opportunity for applicants to fill places of those who did not register on time. Applicants will be informed of their status per official letter from the Office of the Registrar, alternatively, they can check their application status on the TUT website, www.tut.ac.za.

- c. Minimum duration: Four years.
- d. Presentation: Day classes.
- e. Intake for the qualification: January only.
- f. Recognition of Prior Learning (RPL), equivalence and status: See Chapter 30 of Students' Rules and Regulations.
- g. Exclusion and readmission: See Chapter 2 of Students' Rules and Regulations.



## CURRICULUM

FIRST YEAR							
CODE	MODULE	NQF-L	CREDIT	PREREQUISITE MODULE(S)			
CHM105X COE105X	Chemistry I Communication for Academic	(5) (5)	(24) (10)				
CPL105X MAT105X PHI105X	Computer Literacy Mathematics I Physics I	(5) (5) (5)	(10) (24) (24)				
FIRST SEMESTER							
ALI125X LFS125X	Academic Literacy (block module) Life Skills (block module)	(5) (5)	(2) (2)				
TOTAL CRE	EDITS FOR THE FIRST YEAR:		96				
SECOND YEAR							
CODE	MODULE	NQF-L	CREDIT	PREREQUISITE MODULE(S)			
MAT206P	Mathematics II	(6)	(24)	Mathematics I			
FIRST SEMESTER							
CAL216P INA216X OCA216X PCA216X	Chemical Process Industries II Inorganic Chemistry IIA Organic Chemistry IIA Physical Chemistry IIA	(6) (6) (6) (6)	(18) (12) (12) (12)	Chemistry I Chemistry I Chemistry I Chemistry I Mathematics I			
SECOND SEMESTER							
INB216X OCB216X PCB216X	Inorganic Chemistry IIB Organic Chemistry IIB Physical Chemistry IIB	(6) (6) (6)	(18) (18) (18)	Inorganic Chemistry IIA Organic Chemistry IIA Physical Chemistry IIA			
TOTAL CRE	EDITS FOR THE SECOND YEAR:		132				
THIRD YEAR							
CODE	MODULE	NQF-L	CREDIT	PREREQUISITE MODULE(S)			
ESY307P	Environmental Science and Technology III	(7)	(24)	Inorganic Chemistry IIA Organic Chemistry IIA Physical Chemistry IIA			
FIRST SEMESTER							
ISY317P	Industrial Synthesis III	(7)	(18)	Chemical Process Industries II Inorganic Chemistry IIA Mathematics II Organic Chemistry IIA Physical Chemistry IIA			
MSA317P	Material Science IIIA	(7)	(12)	Inorganic Chemistry IIA Mathematics II Organic Chemistry IIA Physical Chemistry IIA			



PCH317P	Physical Chemistry III	(7)	(12)	Mathematics II Physical Chemistry IIB				
SRS317P	Separation Science III	(7)	(18)	Inorganic Chemistry IIB Mathematics II Organic Chemistry IIB Physical Chemistry IIB				
SECOND SEMESTER								
MSB317P MSP317P	Material Science IIIB Molecular Spectroscopy III	(7) (7)	(12) (18)	Material Science IIIA Inorganic Chemistry IIB Mathematics II Organic Chemistry IIB Bweical Chemistry IIB				
PCL317P	Process Control III	(7)	(18)	Inorganic Chemistry IIB Mathematics II Organic Chemistry IIB Physical Chemistry IIB				
TOTAL CREDITS FOR THE THIRD YEAR: 132								
FOURTH YEAR								
CODE	MODULE	NQF-L	CREDIT	PREREQUISITE MODULE(S)				
FIRST SEMESTER								
ISY418P MSP418P PCH418P RIY418P	Industrial Synthesis IV Molecular Spectroscopy IV Physical Chemistry IV Research Methodology IV (after completion of all first-, second and third-year modules)	(8) (8) (8) (8)	(18) (18) (12) (12)	Industrial Synthesis III Molecular Spectroscopy III Physical Chemistry III				
TOTAL CR	EDITS FOR THE SEMESTER							
SECOND SEMESTER								
SECOND S	EMESTER		60					
SECOND S	SEMESTER Industrial Chemistry Practice IV (after completion of all first-, occord and third wars mediate)	(8)	60 (30)					
ICC418P	EDITOT OF THE CLINEOFLIK EMESTER Industrial Chemistry Practice IV (after completion of all first-, second and third-year modules) Industrial Chemistry Project IV	(8) (8)	60 (30) (30)	Industrial Synthesis IV Material Science IIIB Molecular Spectroscopy IV Physical Chemistry IV				
ICC418P ICC418P ICR418P ICR418R	SEMESTER Industrial Chemistry Practice IV (after completion of all first-, second and third-year modules) Industrial Chemistry Project IV (re-registration) (first semester module)	(8) (8) (8)	60 (30) (30) (0)	Industrial Synthesis IV Material Science IIIB Molecular Spectroscopy IV Physical Chemistry IV Research Methodology IV				
ICC418P ICC418P ICR418P ICR418R TOTAL CR	SEMESTER Industrial Chemistry Practice IV (after completion of all first-, second and third-year modules) Industrial Chemistry Project IV (re-registration) (first semester module) EDITS FOR THE SEMESTER:	(8) (8) (8)	60 (30) (30) (0) 60	Industrial Synthesis IV Material Science IIIB Molecular Spectroscopy IV Physical Chemistry IV Research Methodology IV				
ICC418P ICR418P ICR418P ICR418R TOTAL CR	SEMESTER Industrial Chemistry Practice IV (after completion of all first-, second and third-year modules) Industrial Chemistry Project IV (re-registration) (first semester module) EDITS FOR THE SEMESTER: EDITS FOR THE FOURTH YEAR:	<ul><li>(8)</li><li>(8)</li><li>(8)</li></ul>	60 (30) (30) (0) 60 <b>120</b>	Industrial Synthesis IV Material Science IIIB Molecular Spectroscopy IV Physical Chemistry IV Research Methodology IV				



#### 4

#### MODULE INFORMATION (OVERVIEW OF SYLLABUS)

Please take note that for the 2020 academic year certain modules will be assessed as continuous assessment. Please contact the Academic Department for further information.

The syllabus content is subject to change to accommodate industry changes. Please note that a more detailed syllabus is available at the Department or in the study guide that is applicable to a particular module. On 29 November 2019, the syllabus content was defined as follows:

#### Α

#### ACADEMIC LITERACY (ALI125X)

(Module custodian: Directorate of Library and Information Services)

Introduction of information literacy. Development of a search strategy and application of a search string to search engines and academic databases. Evaluation of information sources. Ethical and legal use of information. (Total tuition time:  $\pm$  20 hours)

С

#### CHEMICAL PROCESS INDUSTRIES II (CAL216P)

(Module custodian: Department of Chemical, Metallurgical and Materials Engineering) Introduction to chemical process industries; Petroleum refining; Pulp and paper manufacturing; Mineral Processing; Bulk manufacturing of ammonia and hydrogen; Bulk manufacturing of nitric acid, carbon dioxide and sulphuric acid; Air separation technologies: oxygen and nitrogen; Coal and coal processing; and Renewable energy technologies. (Total tuition time: ± 120 hours)

#### **CHEMISTRY I (CHM105X)**

#### (Module custodian: Department of Chemistry)

The role and importance of chemistry in everyday life. Classification and properties of matter. Units of measurement. Atoms, molecules and ions. The modern view of atomic structure and the use of electron configurations in chemical bonding. The periodic table of elements. The use of IUPAC rules for naming inorganic compounds. Application of the mole concept in stoichiometric calculations. Reactions in aqueous solutions. Chemical equilibrium. Fundamental concepts in electrochemistry. Organic nomenclature. (Total tuition time: ± 240 hours)

#### COMMUNICATION FOR ACADEMIC PURPOSE (COE105X)

#### (Module custodian: Department of Applied Languages)

A workable knowledge of English is an essential skill for any graduate who is required to conduct themselves successfully in a professional working environment. This module will equip students with the competencies required to compose a selection of written texts related to communicating both internally and externally within a professional environment. In addition, the module includes strategies that are essential for the effective communication in various situations, including small groups to avoid unproductive conflict, a multicultural context, etc. (Total tuition time: ± 100 hours)

#### COMPUTER LITERACY (CPL105X)

#### (Module custodian: End User Computing Unit)

Introduction of information literacy. Development of a search strategy and application of a search string to search engines and academic databases. Evaluation of information sources. Ethical and legal use of information. (Total tuition time: ± 100 hours)

#### Е

# ENVIRONMENTAL SCIENCE AND TECHNOLOGY III (ESY307P) (Module custodian: Department of Chemistry)

Introduction to environmental science; Aquatic Chemistry; Geochemistry; Atmospheric Chemistry; Water Treatment; Toxicological Chemistry; and Industrial Ecology. (Total tuition time: ± 240 hours)

#### **1 X 3-HOUR PAPER**

# 1 X 3-HOUR PAPER

# 1 X 3-HOUR PAPER

**1 X 3-HOUR PAPER** 

# CONTINUOUS ASSESSMENT

CONTINUOUS ASSESSMENT

#### INDUSTRIAL CHEMISTRY PRACTICE IV (ICC418P) (Module custodian: Department of Chemistry)

The module is designed to prepare students to function effectively in a working environment, which may include the following industries: petroleum and petrochemicals, pharmaceutical, pulp and paper, food processing, sugar and starch, dyestuffs, cement, paint and coating, fermentation, agrichemicals, mining and ore processing, water and sewage treatment. (Total tuition time: ± 300 hours)

#### INDUSTRIAL CHEMISTRY PROJECT IV (ICR418P/R)

(Module custodian: Department of Chemistry)

The purpose of this module is to apply the knowledge of industrial synthesis, physical chemistry, organic chemistry, inorganic chemistry, separation science, molecular spectroscopy, process control, catalysis, material science and environmental science and technology in developing suitable methods and procedures for synthetic, production, characterisation, structural determination, transformation using both qualitative and quantitative techniques to propose and design innovative solutions to industrial process chemistry and laboratory problems. (Total tuition time: ± 300 hours)

#### INDUSTRIAL SYNTHESIS III (ISY317P)

(Module custodian: Department of Chemistry)

Electron absorption spectroscopy of various transition metal complexes; Industrial extraction, purification, properties and the chemistry of the platinum group metals; Biological and chemical catalysts; Biocatalysis; Preparation, treatment, characterisation and applications of heterogeneous catalysts. (Total tuition time: ± 180 hours)

#### INDUSTRIAL SYNTHESIS IV (ISY418P)

#### (Module custodian: Department of Chemistry)

Properties and reactions of chiral organic synthesis; Metals and metalloids-mediated organic synthesis; Biosynthetic pathways of natural products; Synthetic methods of medicinal chemistry and drug design; Preparation, treatment, characterisation and applications of homogeneous catalysts. (Total tuition time: ± 180 hours)

#### **INORGANIC CHEMISTRY IIA (INA216X)**

#### (Module custodian: Department of Chemistry)

Covalent, ionic and metallic bonding; Inorganic thermodynamics; Solvent systems and acid-base behaviour and redox chemistry; Hydrogen and some of its compounds; Alkali metals and their compounds; Alkaline earth metals and their compounds; Boron and aluminium and their compounds; Carbon and silicon and their compounds; Nitrogen, phosphorous and their compounds; Oxygen, sulfur and their compounds; The Halogens and their compounds. (Total tuition time: ± 120 hours)

#### INORGANIC CHEMISTRY IIB (INB216X) (Module custodian: Department of Chemistry)

Bonding and the structure of molecules; Coordination Chemistry; Crystal field theory; Descriptive chemistry of d-block elements; Descriptive chemistry of the first transition metal series; Descriptive chemistry of Cu, Ag, Au and some of their compounds; Descriptive chemistry of Zn, Cd, Hg and some of their compounds. (Total tuition time: ± 180 hours)

#### L

#### LIFE SKILLS (LFS125X)

#### (Module custodian: Directorate of Student Development and Support)

Academic, personal and socio-emotional skills development for students in higher education. Personal and social dimensions address: effective planning and self-management (goal setting and time management); Adjusting to university life (student life, diversity and change); Intra- and interpersonal skills development (conflict management, self-esteem, relationship management); Effective living (healthy living, HIV education, substance abuse); Academic dimension addresses: academic skills for university (e.g. critical thinking, creativity, managing assignments and assessments). (Total tuition time: ± 20 hours)

# : ± 180 hours)

**1 X 3-HOUR PAPER** 

**1 X 3-HOUR PAPER** 

**1 X 3-HOUR PAPER** 

WORK-INTEGRATED LEARNING

PROJECT ASSESSMENT

#### CONTINUOUS ASSESSMENT



### MATERIAL SCIENCE IIIA (MSA317P)

#### (Module custodian: Department of Chemistry)

Fundamental concepts and structure of materials; Structure of crystalline and amorphous solids; Imperfections, failure, dislocations and strengthening mechanisms in materials. (Total tuition time: ± 120 hours)

#### MATERIAL SCIENCE IIIB (MSB317P)

#### (Module custodian: Department of Chemistry)

Phase diagrams, phase transformation in metals and thermal processing of metal alloys; Ceramics: structure, properties, processing and applications; Polymer structures and composites. (Total tuition time: ± 120 hours)

#### MATHEMATICS I (MAT105X)

#### (Module custodian: Department of Mathematics and Statistics)

Consolidation of existing knowledge of and introduction to new types of functions; solution of systems of equations using matrices; introduction to vector algebra and complex numbers; application of differentiation and integration to solve well-defined problems. (Total tuition time: ± 240 hours)

#### MATHEMATICS II (MAT206P)

#### (Module custodian: Department of Mathematics and Statistics)

Series expansions: Applications of differentiation: Partial differentiation: Integration techniques: Applications of integration; and first-order ordinary differential equations (ODE's). (Total tuition time: ± 240 hours)

#### MOLECULAR SPECTROSCOPY III (MSP317P)

#### (Module custodian: Department of Chemistry)

Introduction to Molecular Spectroscopy; Ultra-violet/Visible Spectroscopy; Infrared Spectroscopy; Mass spectroscopy; Nuclear Magnetic Resonance Spectroscopy. (Total tuition time: ± 180 hours)

### MOLECULAR SPECTROSCOPY IV (MSP418P)

(Module custodian: Department of Chemistry)

Principles of Molecular Spectroscopy and Powder Diffraction; Interpretation of Spectra; Spectral Analysis of Organic and Polymer Molecules; Spectral Analysis of Inorganic Compounds and Minerals; and Industrial and Environmental Applications. (Total tuition time: ± 180 hours)

#### 0

#### **ORGANIC CHEMISTRY IIA (OCA216X)**

#### (Module custodian: Department of Chemistry)

Nature of organic compounds; Structure and reactivity of: alkanes and cycloalkanes; alkenes and alkynes; Aromatic Chemistry; Stereochemistry; Properties and reactions of: alkyl halides; alcohols; ethers and epoxides; ketones and aldehydes; carboxylic acids; carboxylic acid derivatives; amines. (Total tuition time: ± 120 hours)

#### **ORGANIC CHEMISTRY IIB (OCB216X)**

#### (Module custodian: Department of Chemistry)

Electrophilic and nucleophilic aromatic substitution reactions for the syntheses of benzene; Properties and reactions of carbonyl compounds for the syntheses of simple and complex carbonyl derivatives; Stereoisomerism in organic chemistry, Structure and properties of biomolecules; Carbohydrates, proteins, lipids and terpenes. (Total tuition time: ± 180 hours)

#### Ρ

### PHYSICAL CHEMISTRY IIA (PCA216X)

#### (Module custodian: Department of Chemistry)

Properties of gases, liquids, solids and solutions (such as pressure, temperature, composition, density, boiling points, viscosity, surface tension, concentration, vapour pressure and colligative properties). Basic thermodynamics properties (such as internal energy, enthalpy, heat capacity, entropy and Gibbs free energy), and equilibrium process (reaction and phase equilibrium). Fundamental electrochemical properties and processes such as electrode potentials, emf, electrolysis and corrosion). (Total tuition time: ± 120 hours)

# **1 X 3-HOUR PAPER**

**1 X 3-HOUR PAPER** 

**1 X 3-HOUR PAPER** 

**1 X 3-HOUR PAPER** 

#### **1 X 3-HOUR PAPER**

## **1 X 3-HOUR PAPER**

# **1 X 3-HOUR PAPER**



#### **1 X 3-HOUR PAPER**

**1 X 3-HOUR PAPER** 

#### PHYSICAL CHEMISTRY IIB (PCB216X)

(Module custodian: Department of Chemistry)

Chemical Thermodynamics; Phase Changes; Electrochemistry; Chemical Kinetics; Quantum Chemistry; and Surface Chemistry. (Total tuition time: ± 180 hours)

#### PHYSICAL CHEMISTRY III (PCH317P)

#### (Module custodian: Department of Chemistry)

Surface Chemistry: physical and chemical behaviour of surfaces including interfaces such the solid-gas interface, liquid interface; Electrochemistry; electrical properties of interfaces, liquid interface, polarised and non-polarised electrode surfaces and their applications. (Total tuition time: ± 120 hours)

#### PHYSICAL CHEMISTRY IV (PCH418P)

#### (Module custodian: Department of Chemistry)

Statistical Thermodynamics: thermodynamic properties of macroscopic systems from microscopic constituents; energy, entropy and temperature per average atom; statistical postulates of guantum mechanics; molecular interpretation to describe heat capacity, entropy, internal energy, Gibbs free energy; Kinetics of complex reactions: reactions approaching equilibrium: the Arrhenius equation: Lindemann-Hinshelwood mechanism: the RRK (Rice-Ramsperger-Kassel) model; the Rice-Herzfeld mechanism; catalysis. (Total tuition time: ± 120 hours)

#### PHYSICS I (PHI105X)

#### (Module custodian: Department of Physics)

Basic Mathematical Concepts for Physics and measurements. Kinematics in one dimension. Kinematics in a plane (projectile motion). Forces and Newton's Laws of Motion. Momentum and Impulse. Work, Energy and Power. Rotational Kinematics. Rotational Dynamics. Elasticity. Static and dynamic fluids. Temperature and 0th Law of Thermodynamics. Thermal Expansion and Thermal Stress. First Law of Thermodynamics. Heat transfer. Gas Laws. General properties of waves. Reflection. Refraction. Interference and Diffraction of waves. Electrostatics. Electric Potential Energy and Potential Difference. Electric Circuits. (Total tuition time: ± 240 hours)

#### PROCESS CONTROL III (PCL317P)

#### (Module custodian: Department of Chemical, Metallurgical and Materials Engineering)

Introduction to process control; Control incentives and strategies; Controller principles and Control modes; Process instrumentation; and application and analysis of process instrumentation. (Total tuition time: ± 240 hours)

#### R

#### **RESEARCH METHODOLOGY IV (RIY418P)**

#### (Module custodian: Department of Chemistry)

Purpose, nature and meaning of research; Structure of a research proposal: identify a research problem; literature review; research aims, objectives and hypotheses; research design types; sampling procedures; validating results; research budget; referencing styles; Types of quantitative data; Basic principles of non-parametric tests: introduction to descriptive statistics: Probability (p-values) and hypothesis testing, introduction to inferential statistics (student's t-test, ANOVA and correlations); Interpretation of graphs and tables; Basic principles of research ethics; Dissemination of research findings. (Total tuition time: ± 120 hours)

#### s

#### SEPARATION SCIENCE III (SRS317P)

#### (Module custodian: Department of Chemistry)

Fundamentals of separation processes; Separation by phase addition or creation; Separation by barriers and solid agents; Separations that involve solid phase; and Extractions. (Total tuition time: ± 180 hours)

# **1 X 3-HOUR PAPER**

CONTINUOUS ASSESSMENT

# **1 X 3-HOUR PAPER**

**1 X 3-HOUR PAPER** 

**1 X 3-HOUR PAPER** 

**1 X 3-HOUR PAPER** 

#### **1 X 3-HOUR PAPER**