



CATALYTIC OXIDATION TECHNOLOGY TRANSFER PROGRAM

 Guild
Associates, Inc.

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CATOX

CBR Filtration Team

Technology Transfer Program (TTP)

Report Documentation Page

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Overall Objective

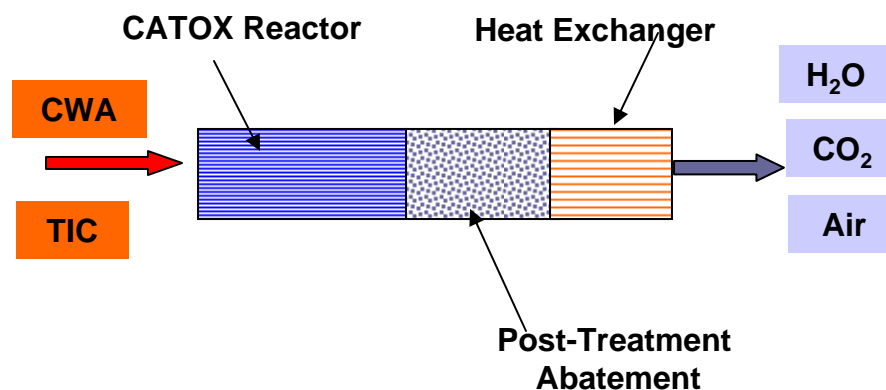


- Design, construct and evaluate a catalytic air purification system for collective protection applications
 - Demonstrate Broader Protection of Catalytic versus Single Pass Filtration Technology
 - Optimize Power, Weight and Size of Catalytic Process



Approach

- Incorporate commercial/newly developed catalysts for chemical, biological and TICs destruction (leverage w/ Advanced Adsorbents Program)
- Establish design relationships for predicting system size and energy requirements for potential applications



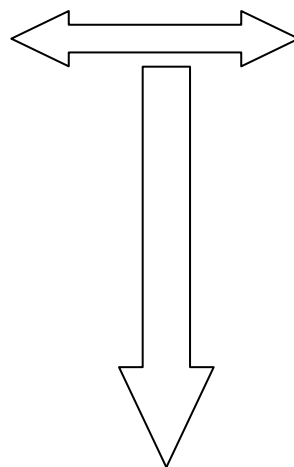


Program Leverage

DARPA Investment

TECHNOLOGY READINESS LEVELS

- identification and evaluation of acid gas abatement strategies



ECBC/INDUSTRY

COLLABORATION

LAB-SCALE TESTING (COMPONENT)

- evaluation of commercial catalysts

SYSTEM-SCALE TESTING

- evaluation and optimization of catalyst system components
 - integration of system components
- determine size and energy requirements of scalable AP system



DARPA Investment



– Objectives

- **Technology maturity level/Technology readiness level**
- **Applicability/relevance to DoD CB Defense Acquisition Program**
- **Potential to mature for evaluation in field environment**
- **Availability of other funding to leverage Technology Transition investment**
- **Availability to government of data bases, methodology details, and design concepts**



Chemical delivery sensitivities

- Single pass filters have their capacity defined in terms of CT (conc x time).
- This is a misleading indicator for CATOX.
- CATOX challenges must be as close to the application's threat to account for performance sensitivities to:
 - # of attacks
 - Dosage per attack
 - Peak concentration
 - Frequency of attacks
- CATOX systems can succeed or fail for a given CT if any of these variables are changed.



Benefits

Current filter technology

- limited capacity for agents that are removed by chemical reaction and/or weakly adsorbed
- minimal protection versus several of the toxic industrial chemicals (TICs)
- prolonged environmental exposure has been shown to reduce the capacity of these filters for agents that are removed by chemical reaction

Catalytic oxidation

An alternative air purification technology

- (1) broad and universal protection against the chem-bio threat,
- (2) reduced logistics due to long operational life,
- (3) greatly increased capacity for CB agents and TICs compared to current NBC collective protection technologies and
- (4) lower energy costs relative to other regenerative filtration technologies.
- (5) Catalytic oxidation is a destructive technology, converting CB agents and TICs to CO₂ (catalysts exist that are capable of decomposing nitrogen-containing compounds with minimum NO_x formation), H₂O and haloacids (should halogens be associated with the parent compound).



Challenges

- **Issues**
 - **Toxic By-products**
 - **Catalyst Stability**
 - **Energy**
 - **Post treatment**

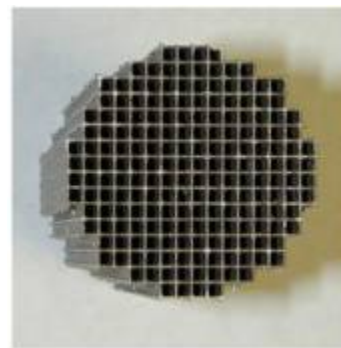
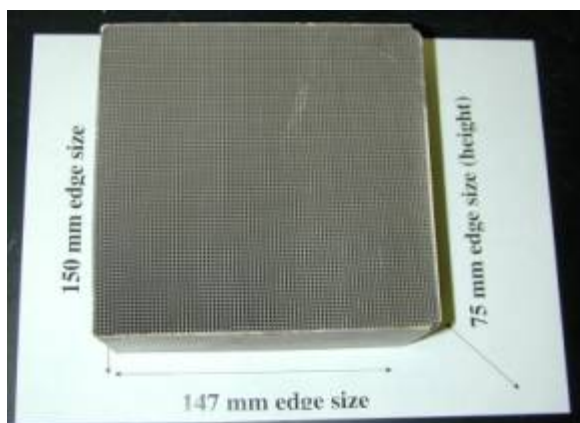
- **State of the Art**
 - **High activity**
 - **Stable catalysts**
 - **NO_x, acid retention**
 - **Improved acid abatement technologies**

- **Mitigation of reaction product emissions**
- **Maximize heat recovery to minimize energy utilization**



Material Development

-Incorporate commercial/newly developed catalysts for chemical, biological and TICs destruction



Downselect (light-off curves and selectivity)

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Monolithic Bed Catalyst Suppliers



Manufacturer	Catalyst Designation	Cell Density
Engelhard Corporation	Engelhard Catalyst	200 Cells/in ²
United Emissions Catalyst	NB001-73-01	200 Cells/in ²
United Emissions Catalyst	NB001-73-02	200 Cells/in ²
Sud Chemie Prototech	LS02-03145	400 Cells/in ²
Sud Chemie Prototech	Misc-03144	400 Cells/in ²
Guild Associates	No-NO _x	400 Cells/in ²
Guild Associates	3X	400 Cells/in ²
Johnson Matthey	CatalyK6 Sample	400 Cells/in ²

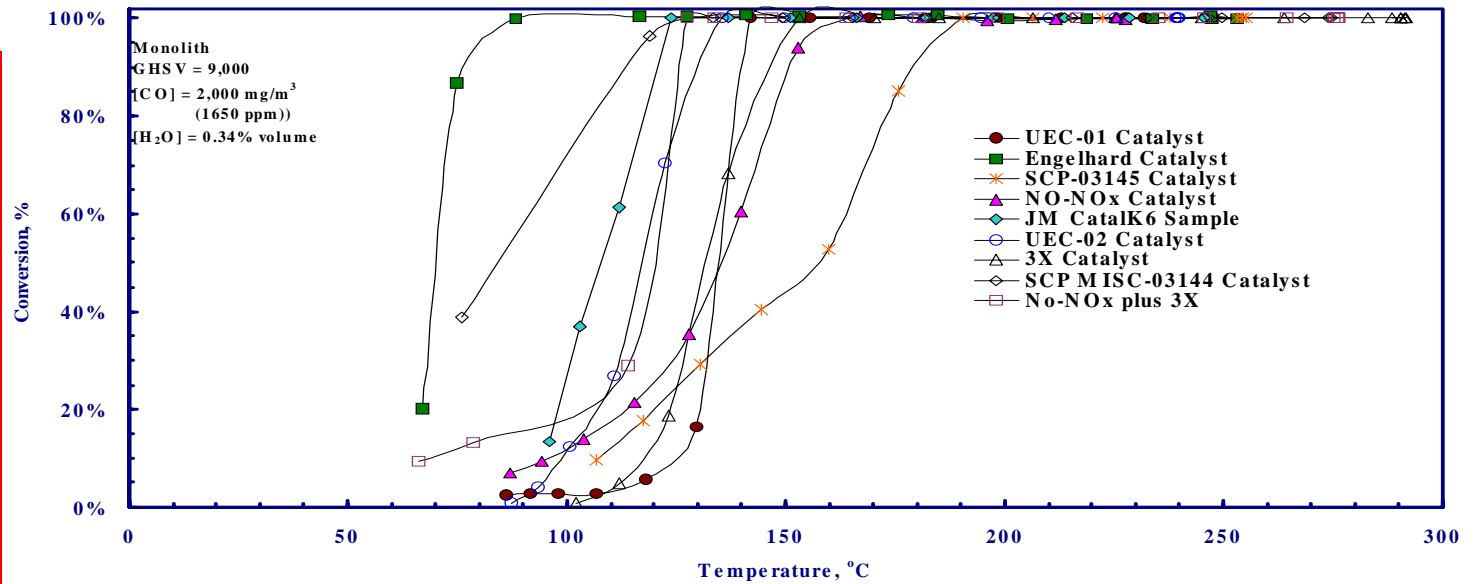
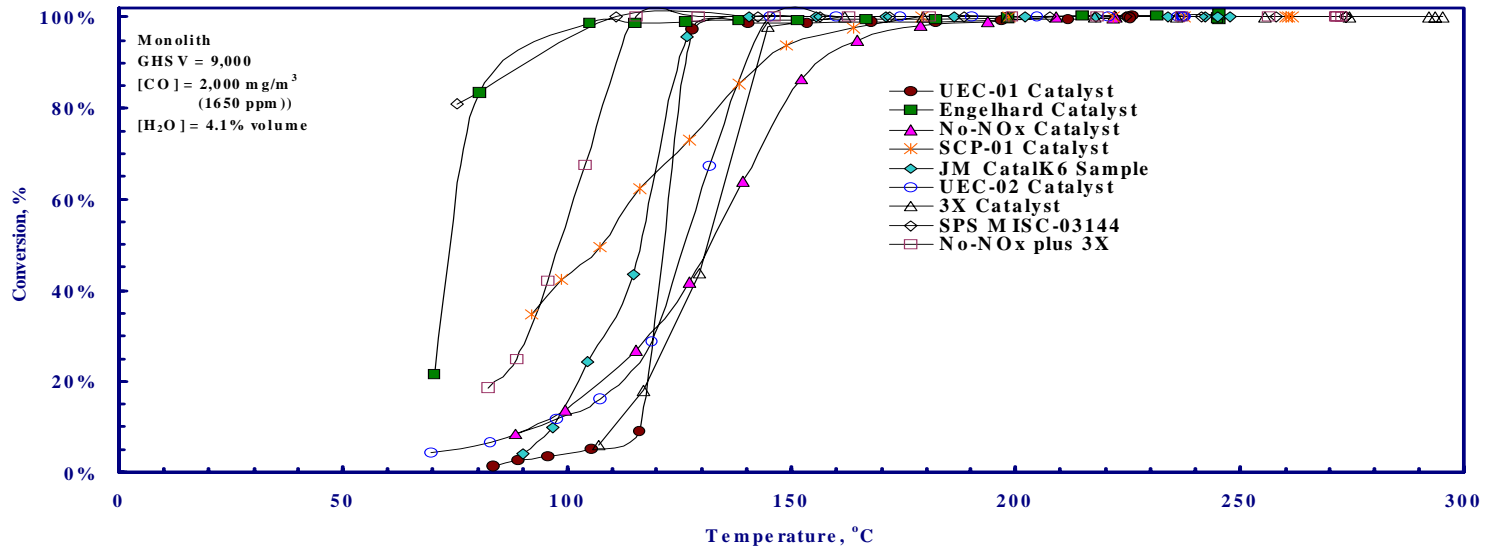


Lab-scale results

Catalyst oxidation of CO by catalytic materials under low and high humidity conditions (materials improvement)

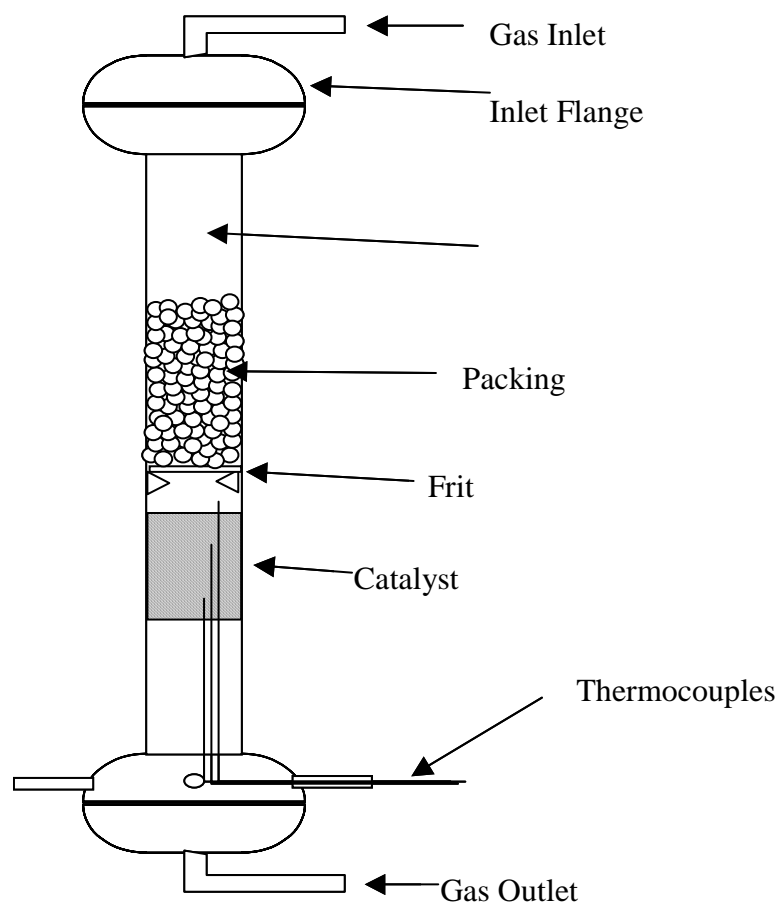


• good indicator of catalyst activity (very clean parent and byproduct)
• poor effective filtration by current NBC systems



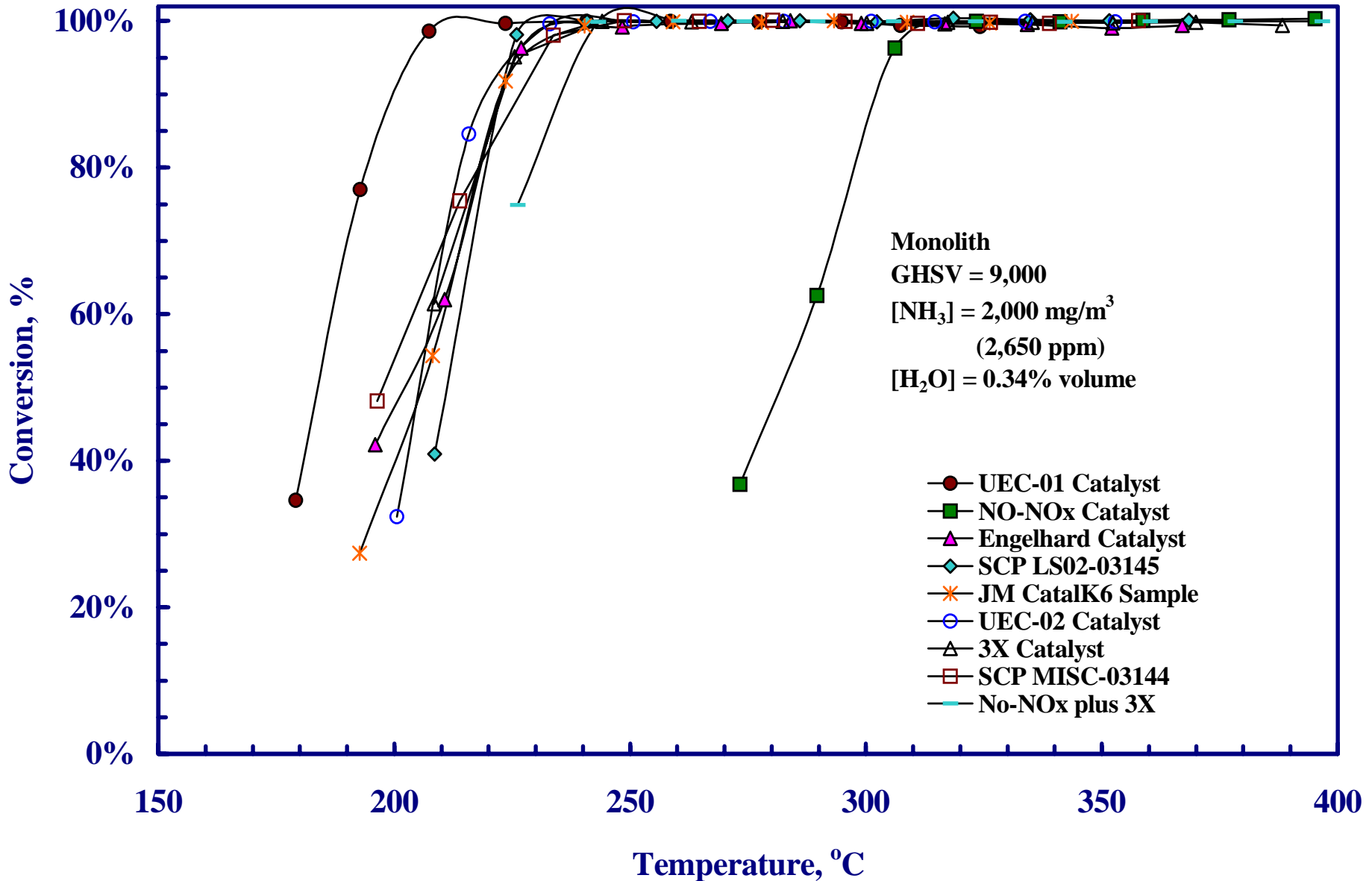


Schematic representation of catalytic reactor



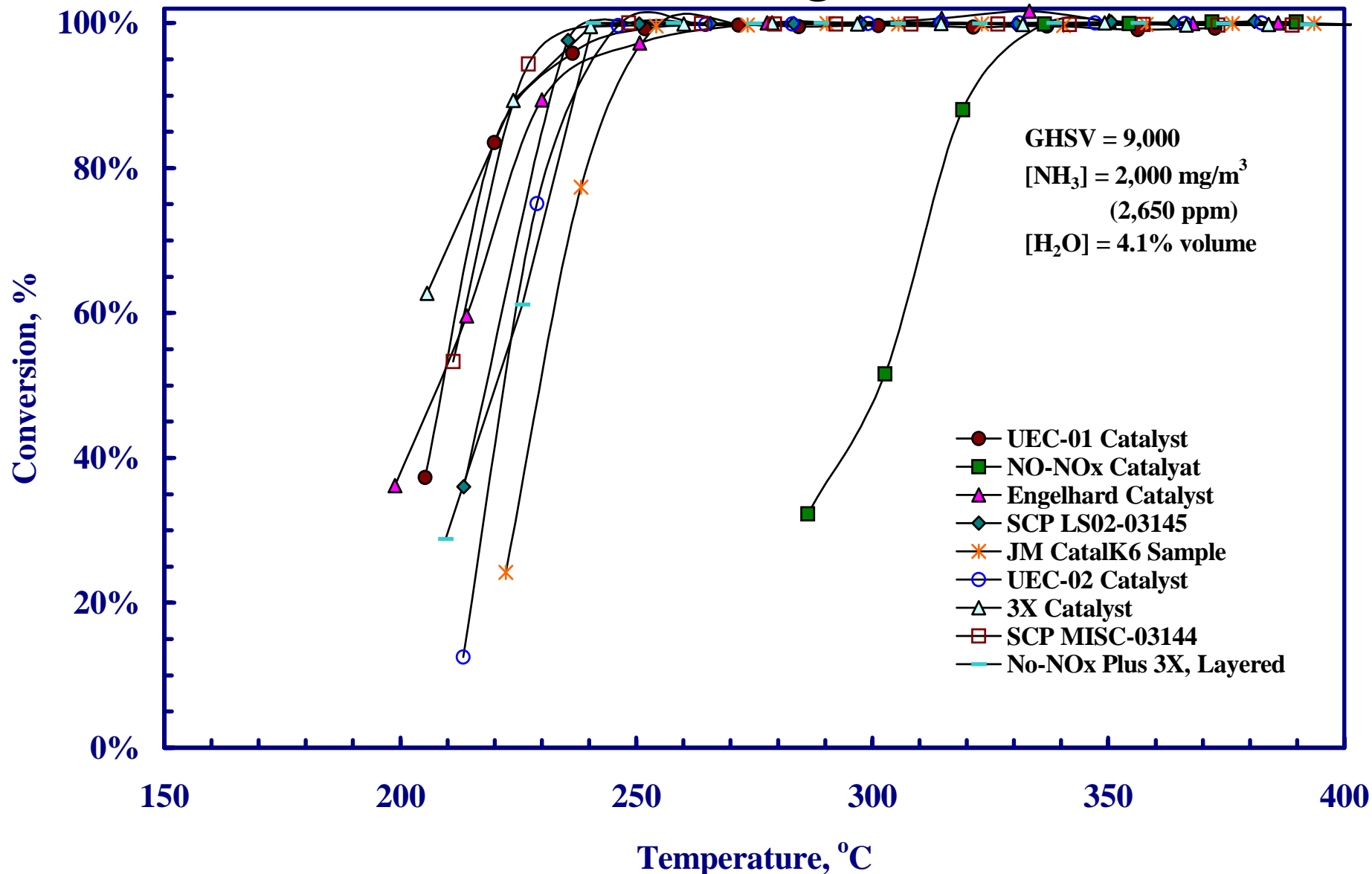


Dry NH₃ tests



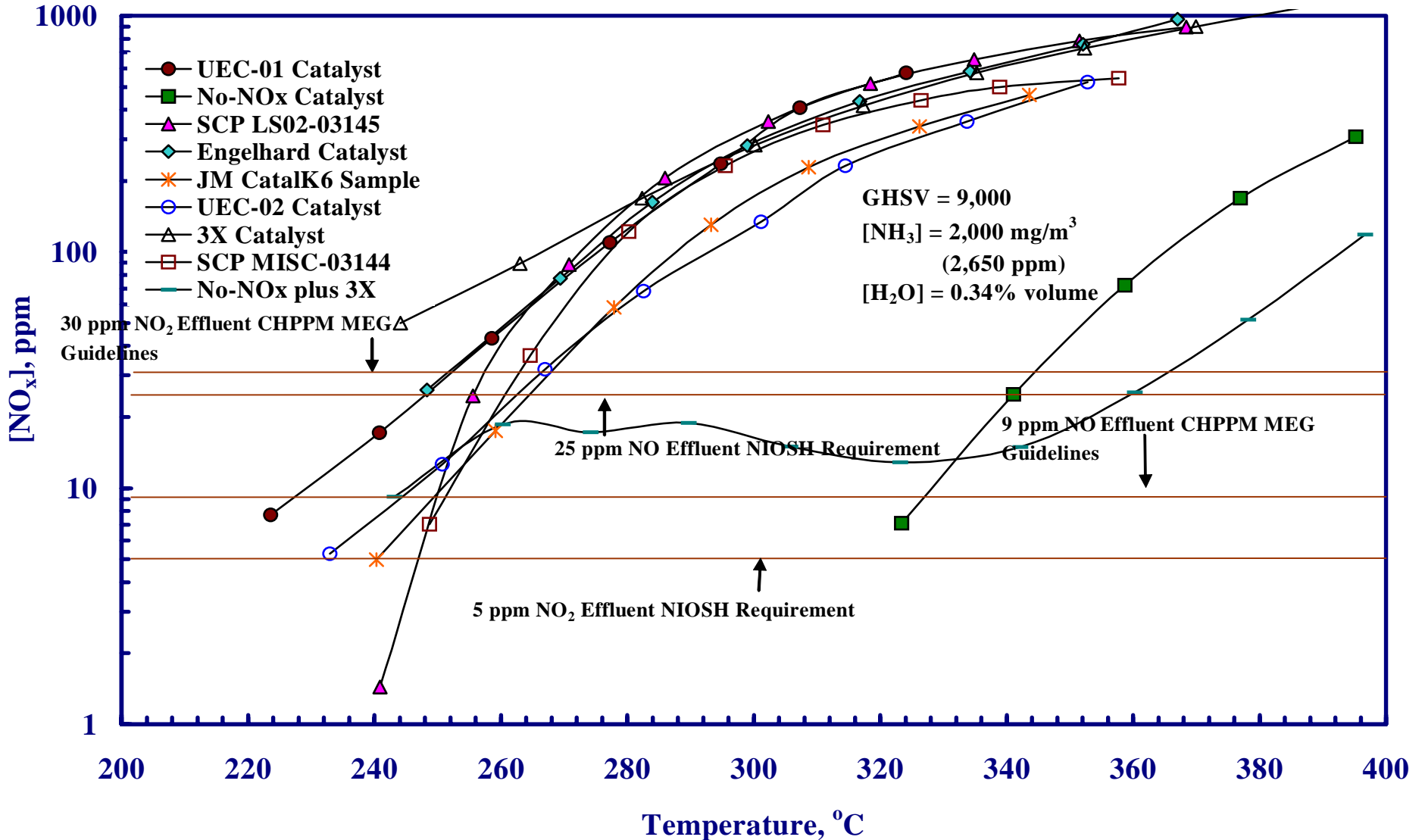


Humid NH₃ tests



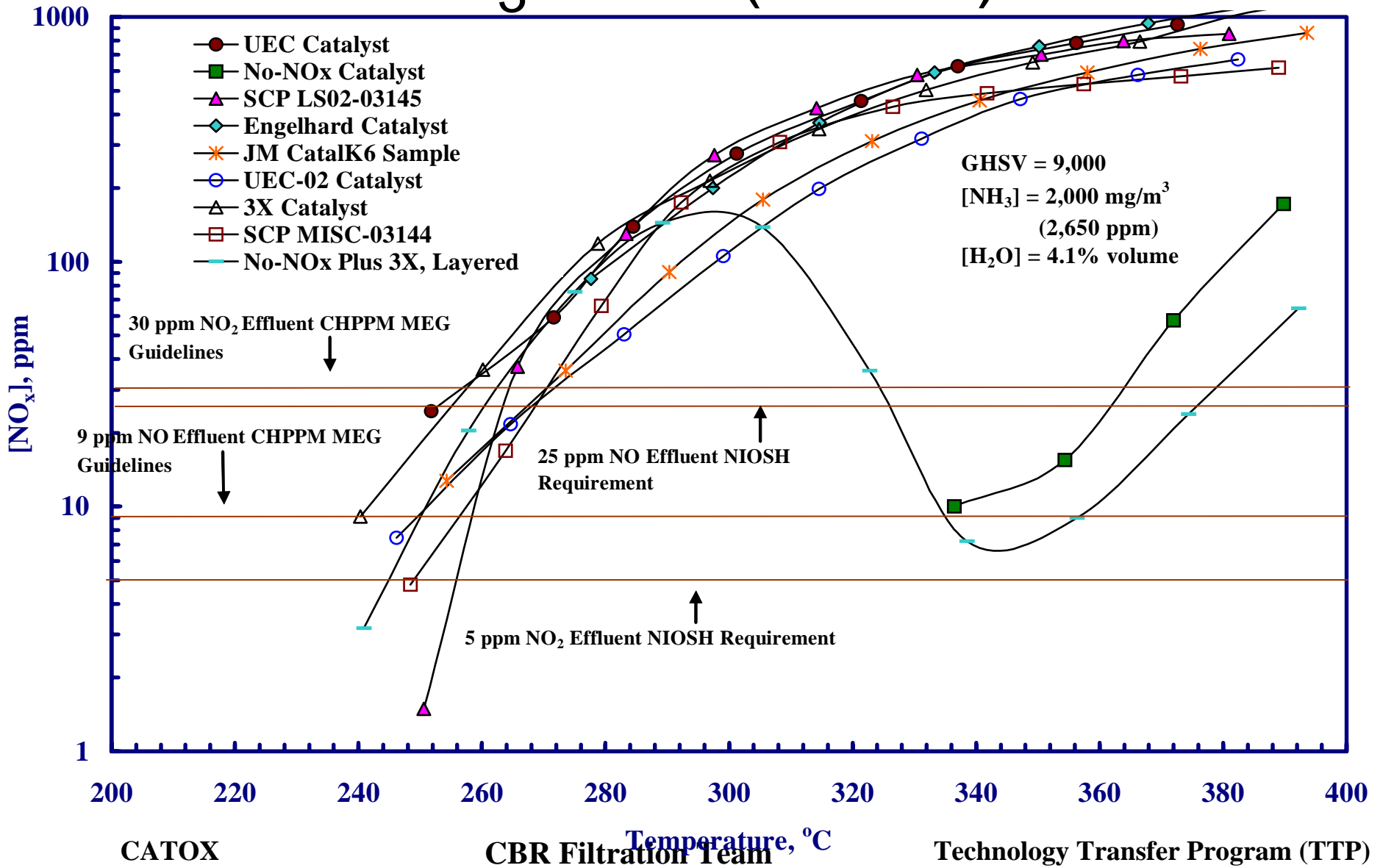


Selectivity (NO_x): NH_3 tests (dry)



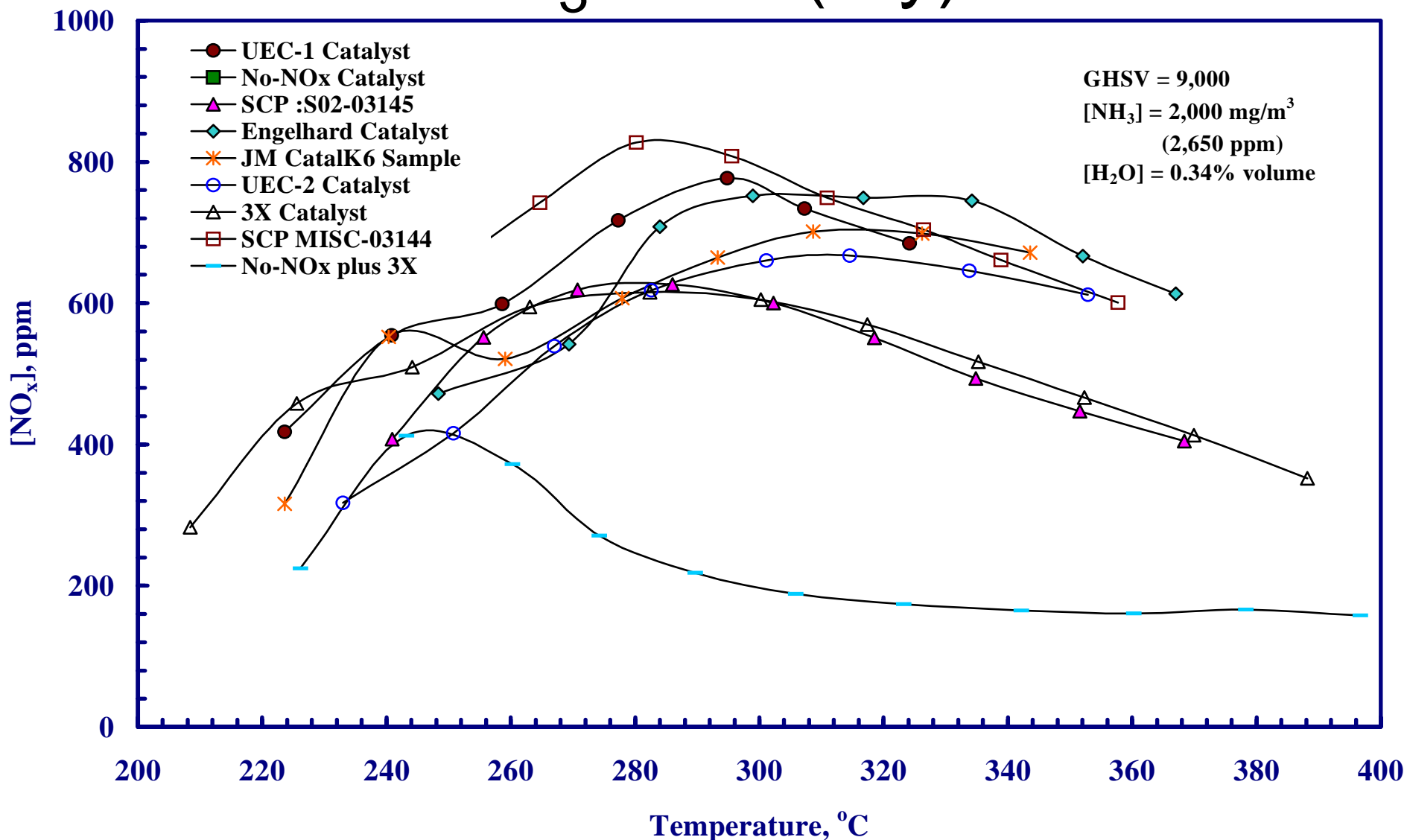


Selectivity (NO_x): NH_3 tests (humid)



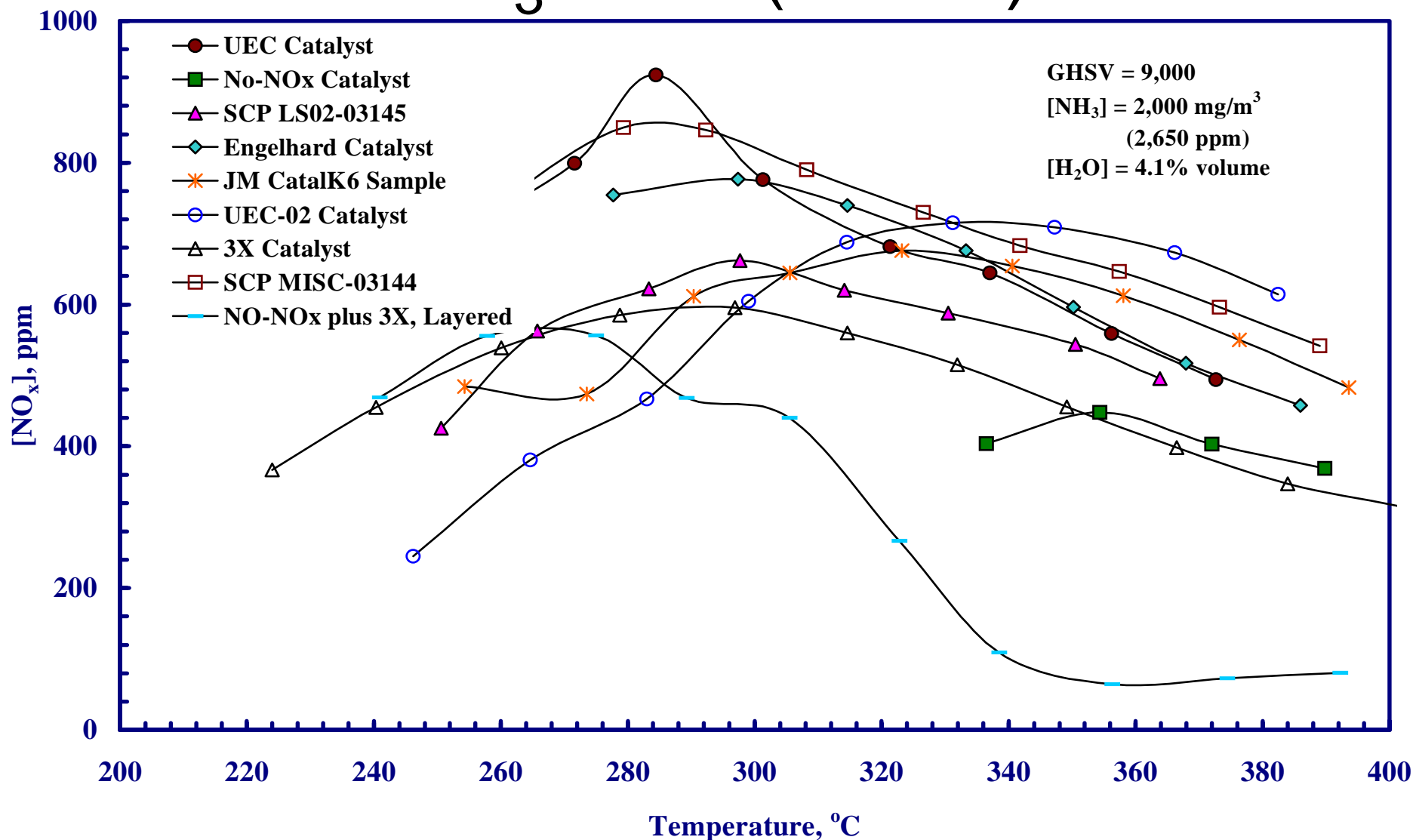


Selectivity (N_2O): NH_3 tests (dry)



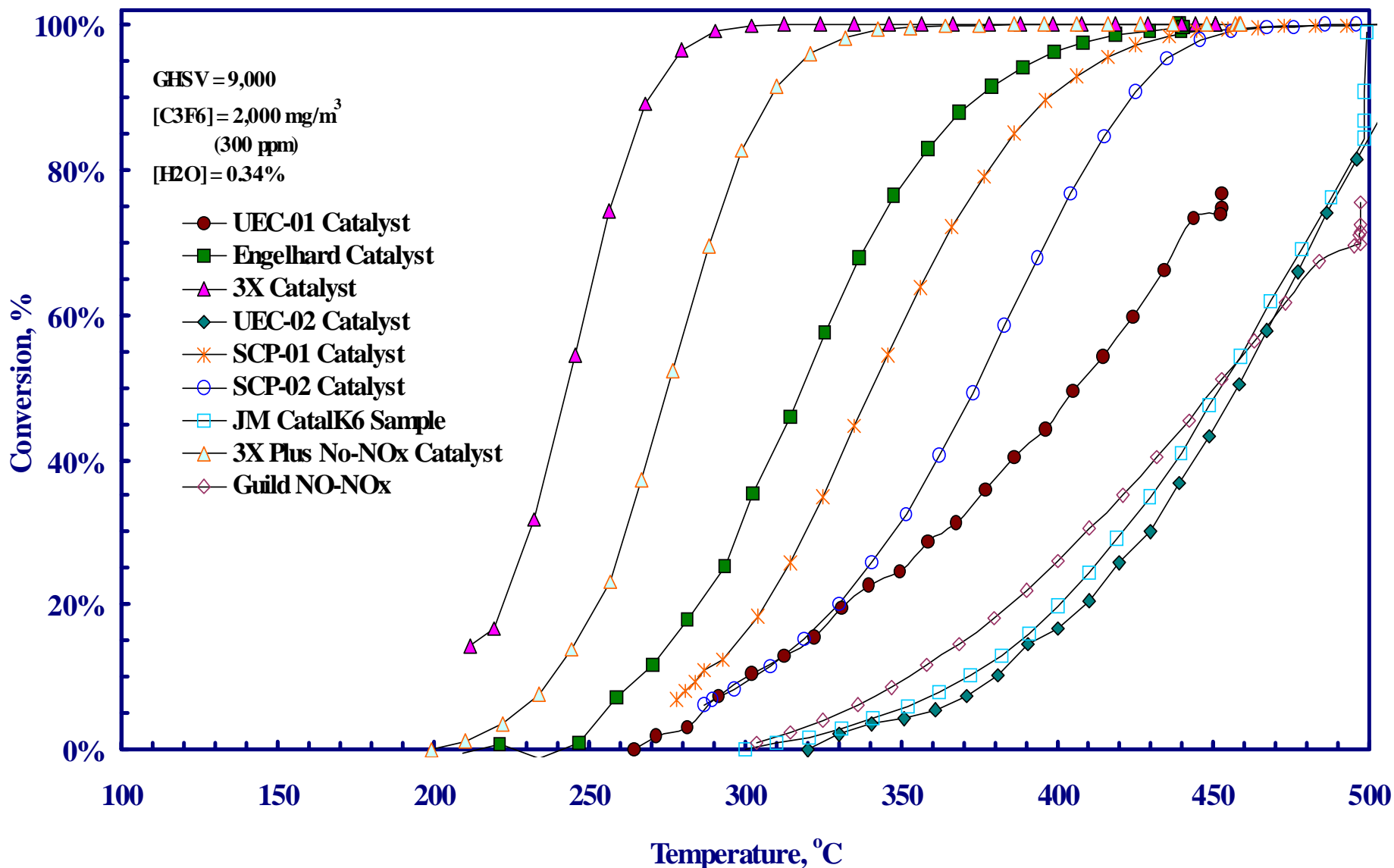


Selectivity (N_2O): NH_3 tests (humid)



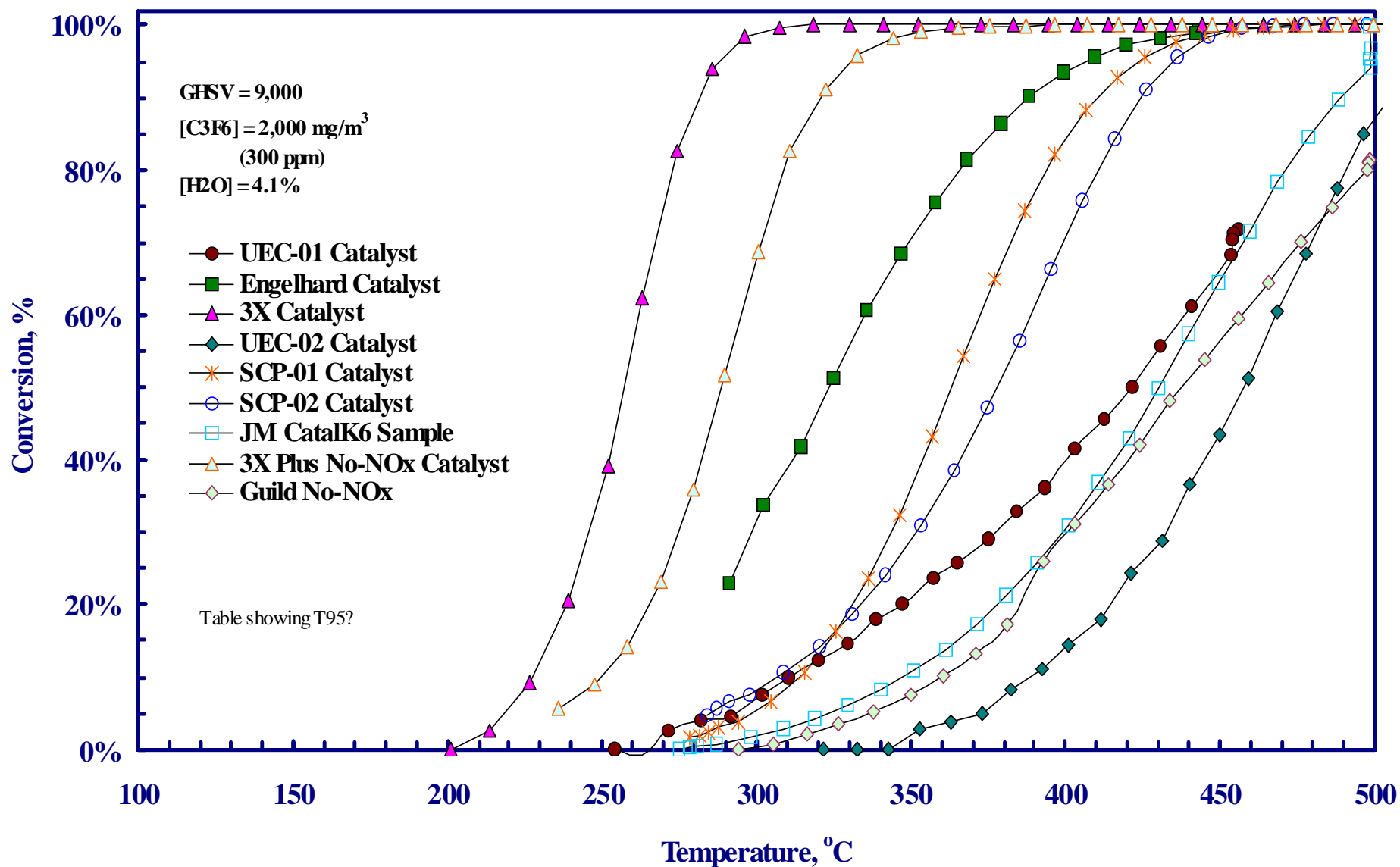


Dry C₃F₆ tests





Humid C_3F_6 tests





Catalyst Performance

Catalyst performance summary information

Catalyst	Design Lim. Chemical ¹	Temperature ²	Ammonia [NO _x] ³	Ammonia [N ₂ O] ³
Guild No-NO _x	C ₃ F ₆	T > 500°C	> 500 ppm	> 300 ppm
Guild No-NO _x Plus 3X	C ₃ F ₆	350°C	15 ppm	160 ppm
Guild 3X	C ₃ F ₆	310°C	400 ppm	550 ppm
Engelhard #164217005	C ₃ F ₆	440°C	> 1,000 ppm	~ 300 ppm
UEC NB001-73-001	C ₃ F ₆	T > 500°C	> 1,000 ppm	~ 300 ppm
UEC NB001-73-002	C ₃ F ₆	T > 500°C	> 1,000 ppm	~ 300 ppm
SCP LS02-03145	C ₃ F ₆	450°C	> 1,000 ppm	~ 300 ppm
SCP MISC-03144	C ₃ F ₆	450°C	> 1,000 ppm	~ 300 ppm
JM CatalyK6 Sample	C ₃ F ₆	T > 500°C	> 1,000 ppm	~ 300 ppm

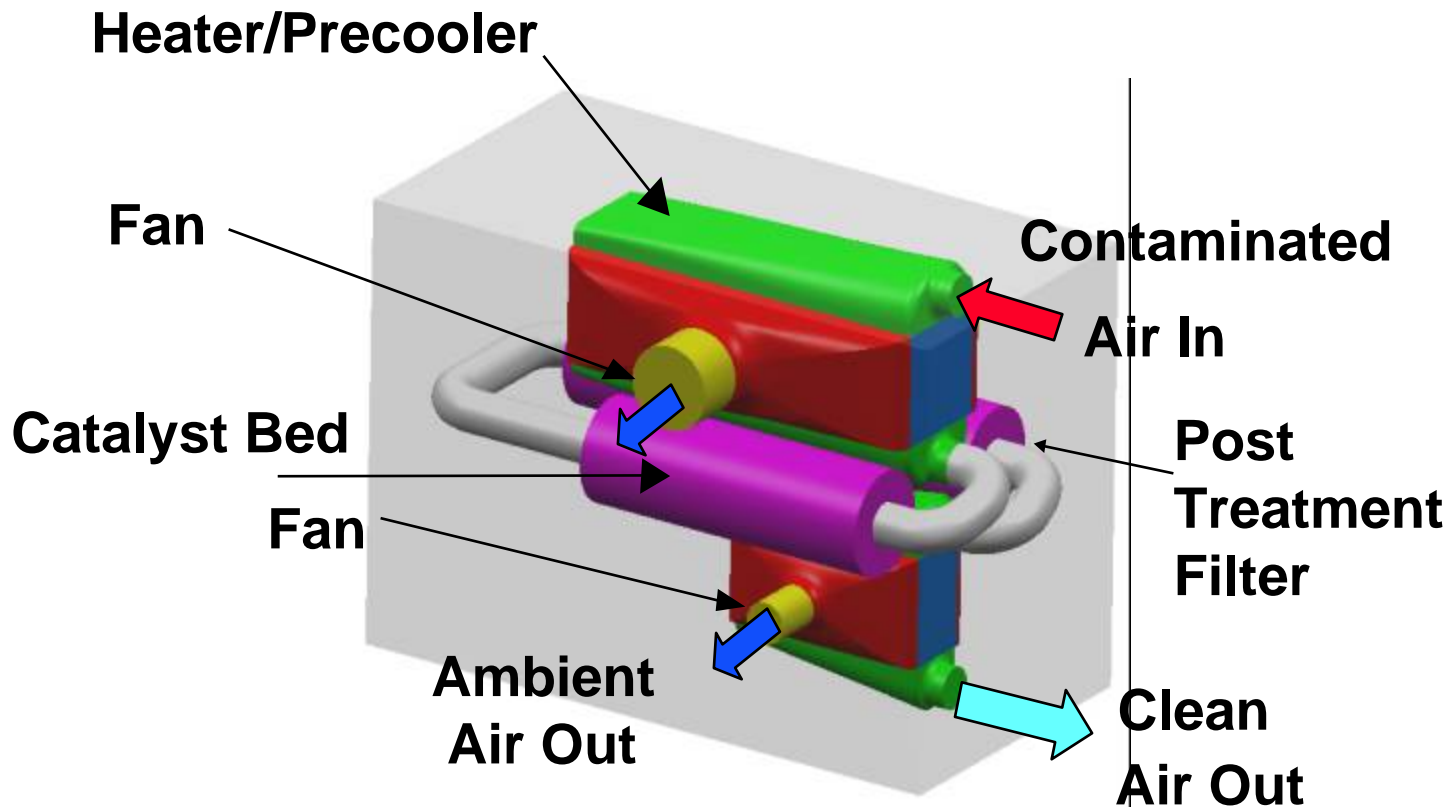
¹Chemical requiring greatest temperature to achieve 99% destruction

²Temperature required to achieve 99% destruction of design limiting chemical

³NO_x or N₂O concentration formed during destruction of NH₃ at temperature



NBC Subsystem Summary



Targeted for:

FCS Application

Scalable for building protection

Transportable shelters (JECP)

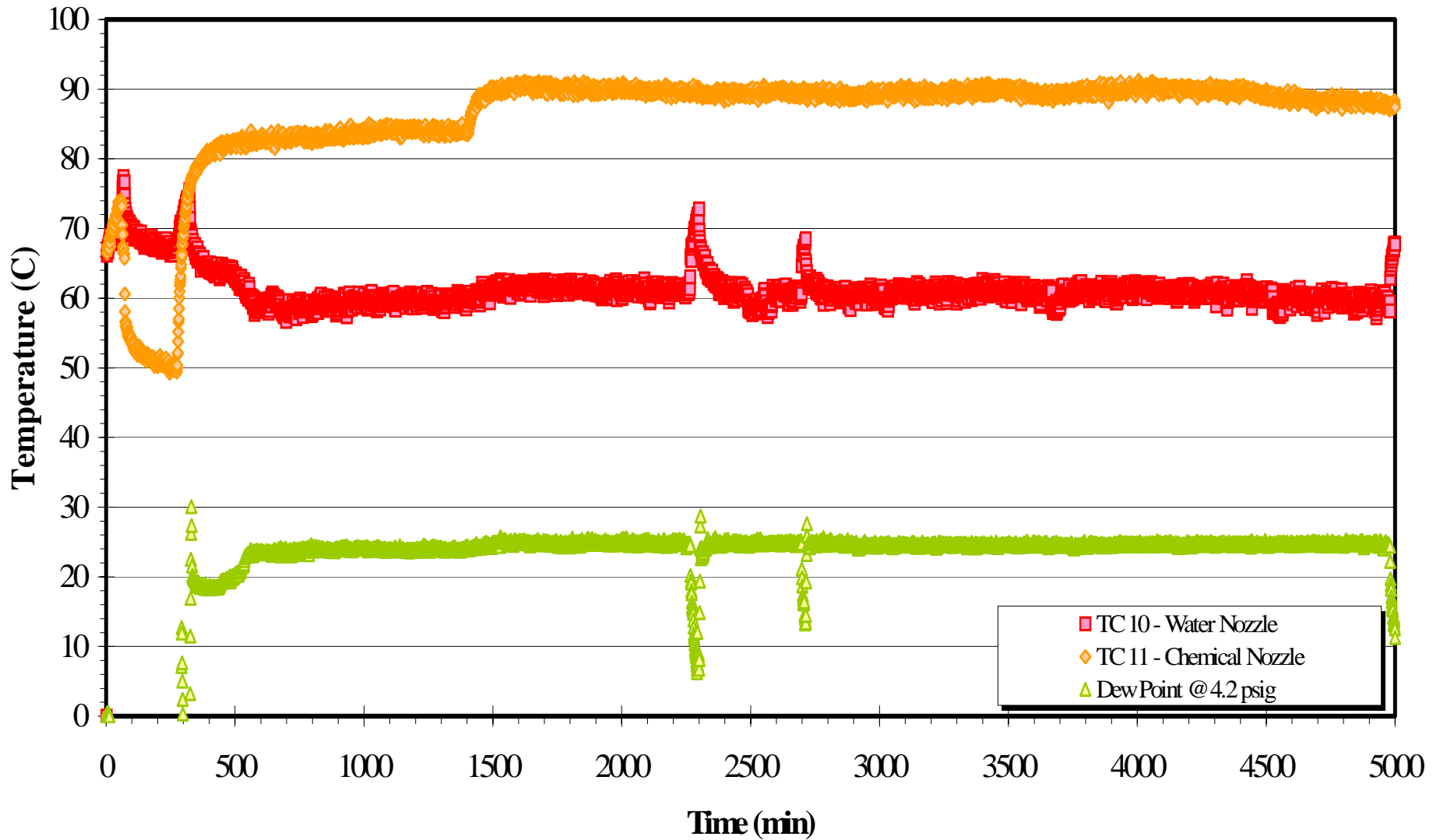


Ammonia (HC): Feed Temperature

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Feed Conditions



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ATD Experimental Design

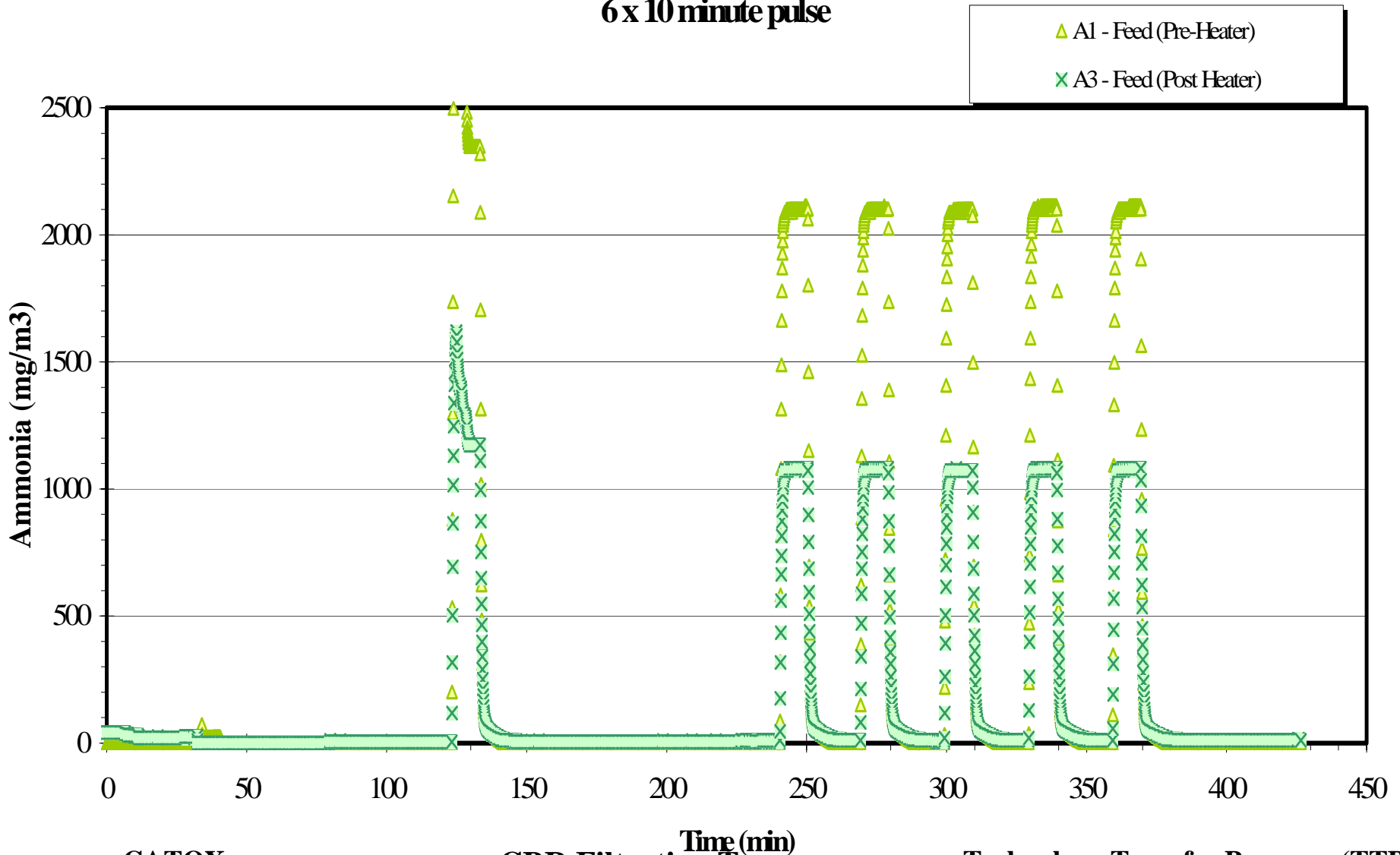
- 6 x 2000 mg/m³ x 10 minutes
(C_T = 120,000 mg-min/m³)
 - 6 x 200 mg/m³ x 100 minutes
(C_T = 120,000 mg-min/m³)
-

C_T = 240,000 mg-min/m³
(under 2.5% water (volume) and 50 cfm)



Ammonia (HC): Feed Concentrations

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NH₃-HW-BHc Ammonia Feed Chart 5-6-04
6 x 10 minute pulse



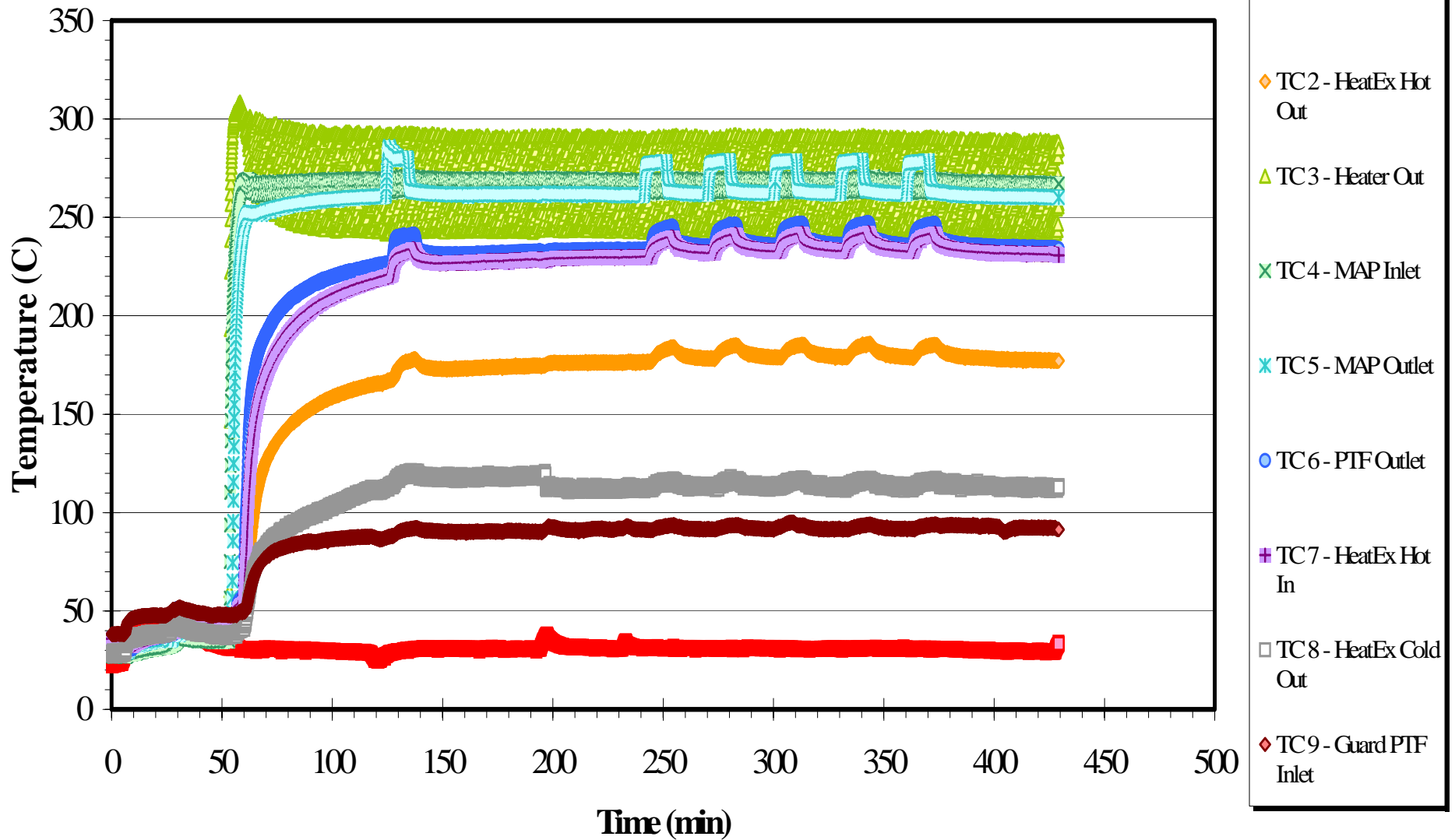


Ammonia (HC): Temperature

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Temperature Plot

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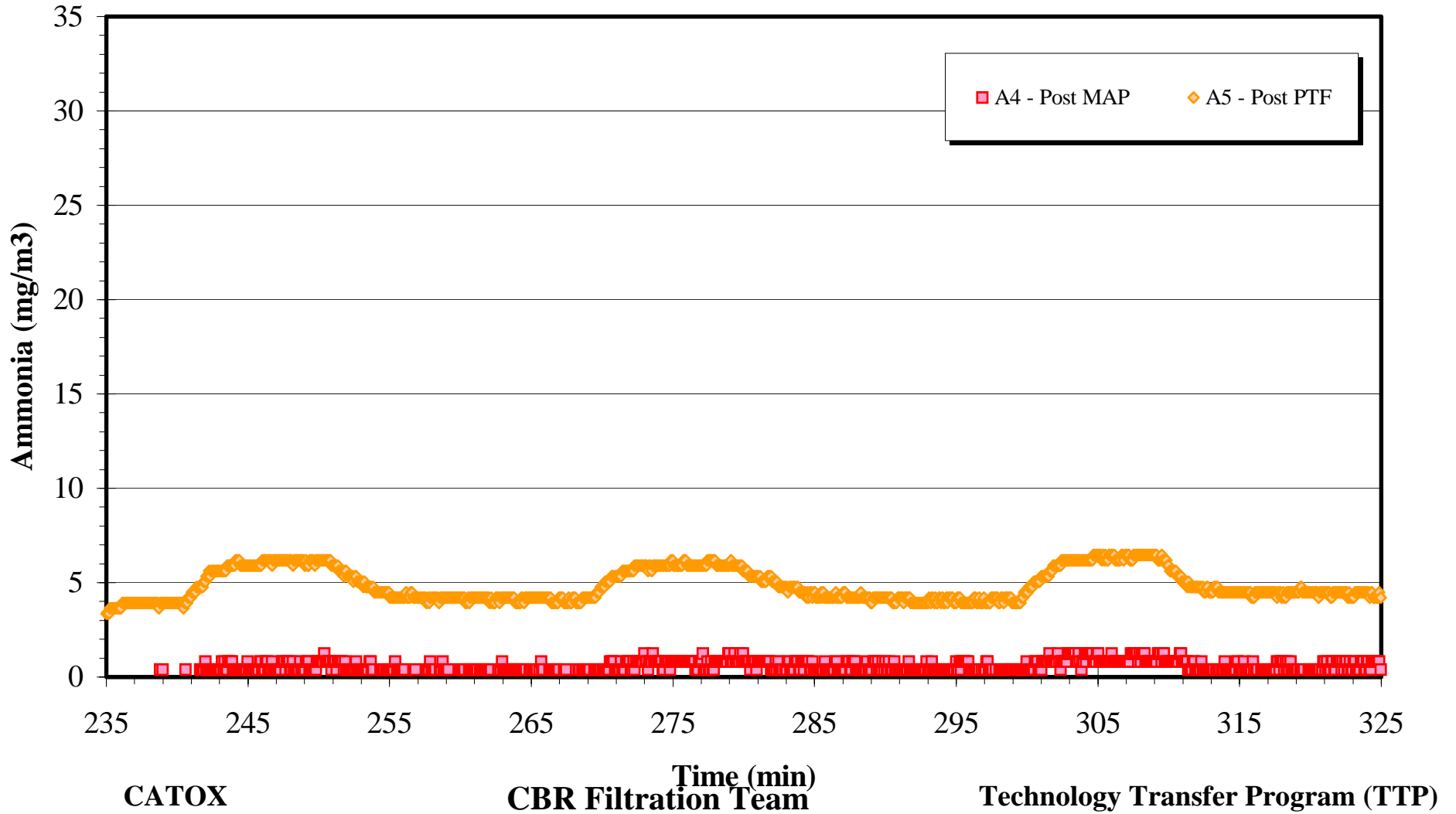
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Ammonia (HC); Parent compound

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ECBC 50 SCFM CATOX UNIT Analog Signal Chart

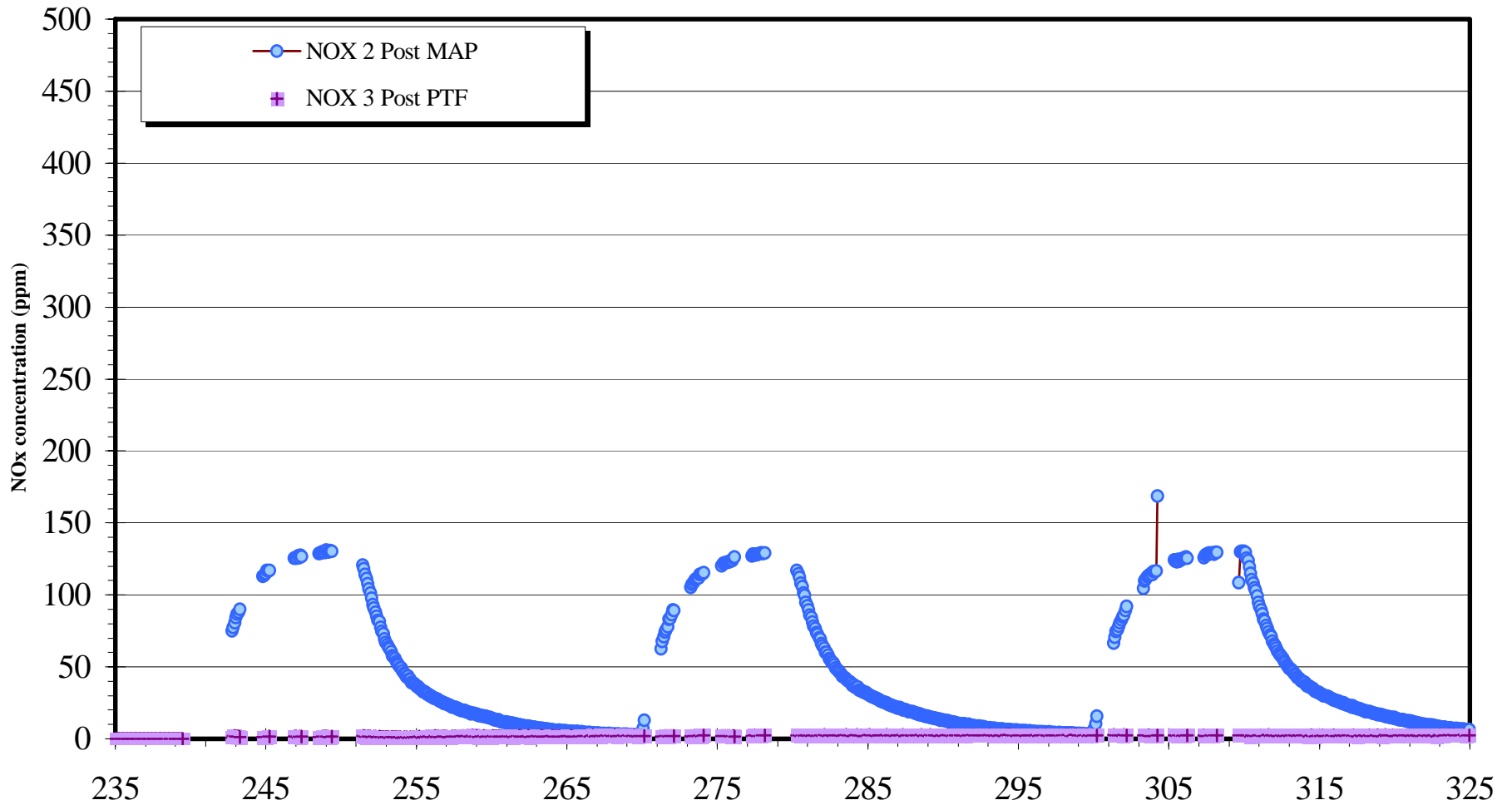




Ammonia (HC): By-products

ECBC 50 SCFM CATOX UNIT

Analog Signal Chart



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Time (min)
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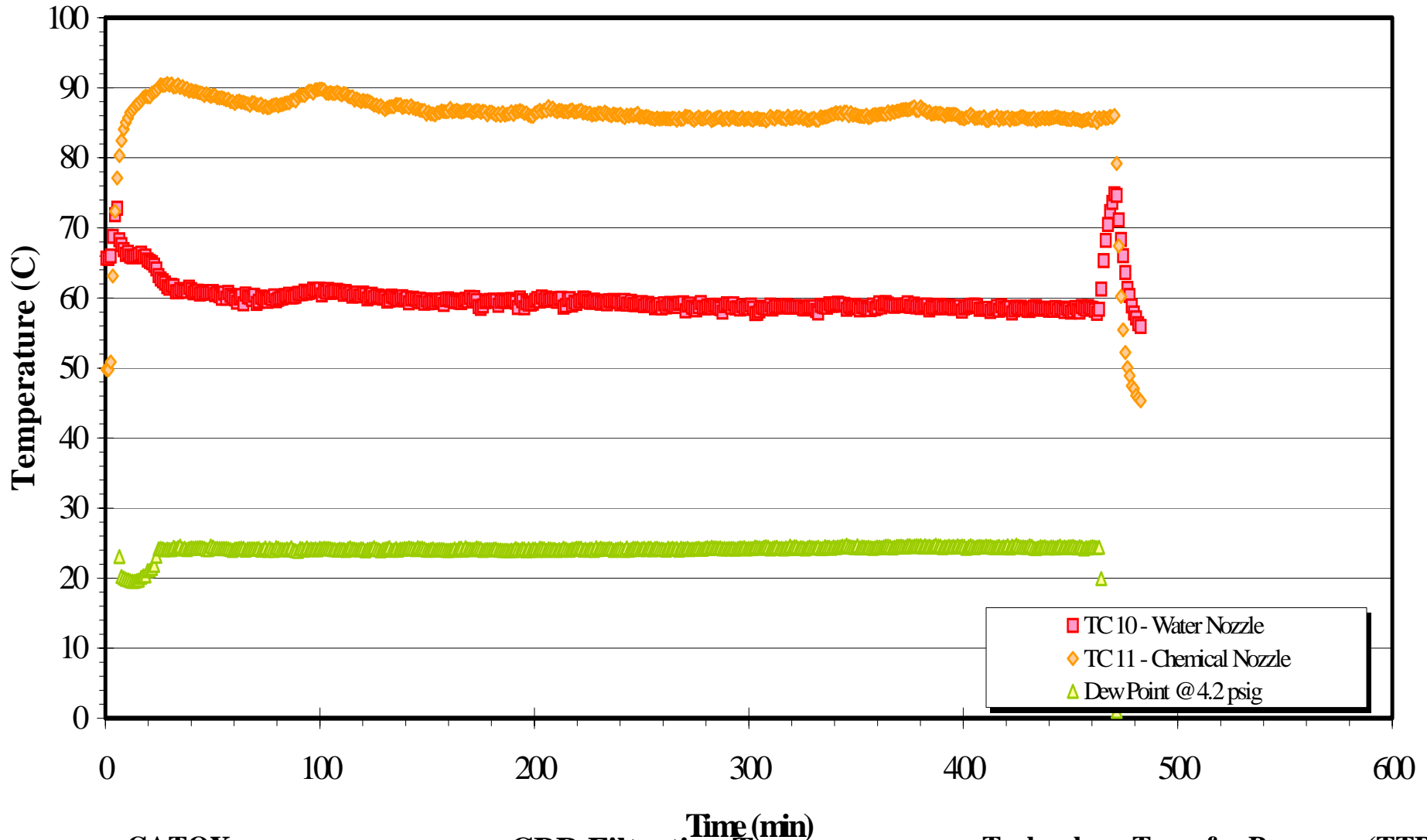


Ammonia (LC): Feed Temperature

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ECBC 50 SCFM CATOX UNIT

Feed Conditions



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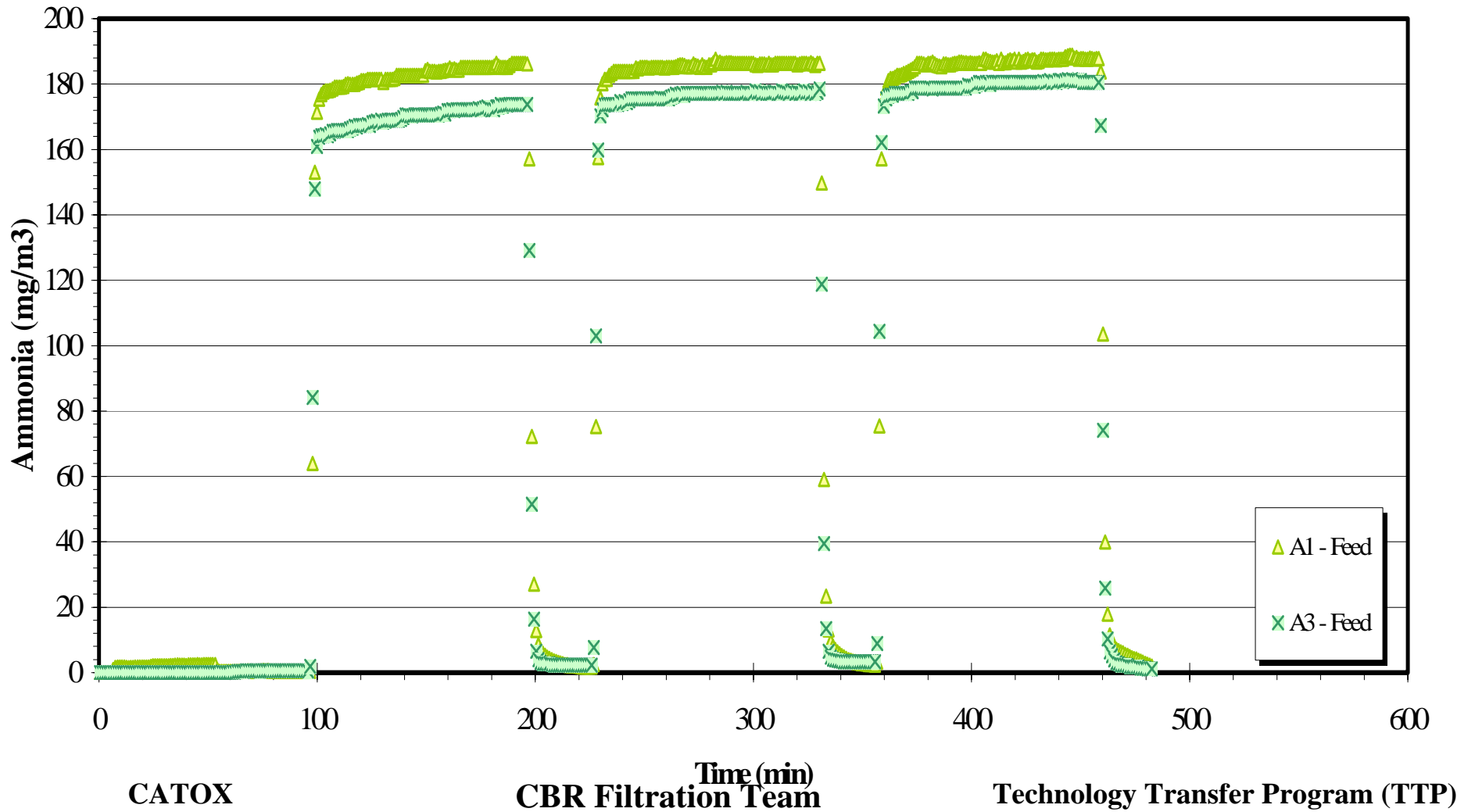
Ammonia (LC): Feed Concentrations

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ECBC 50 SCFM CATOX UNIT

NH₃-HW-BLc P3-5

Feed Chart 5-10-04



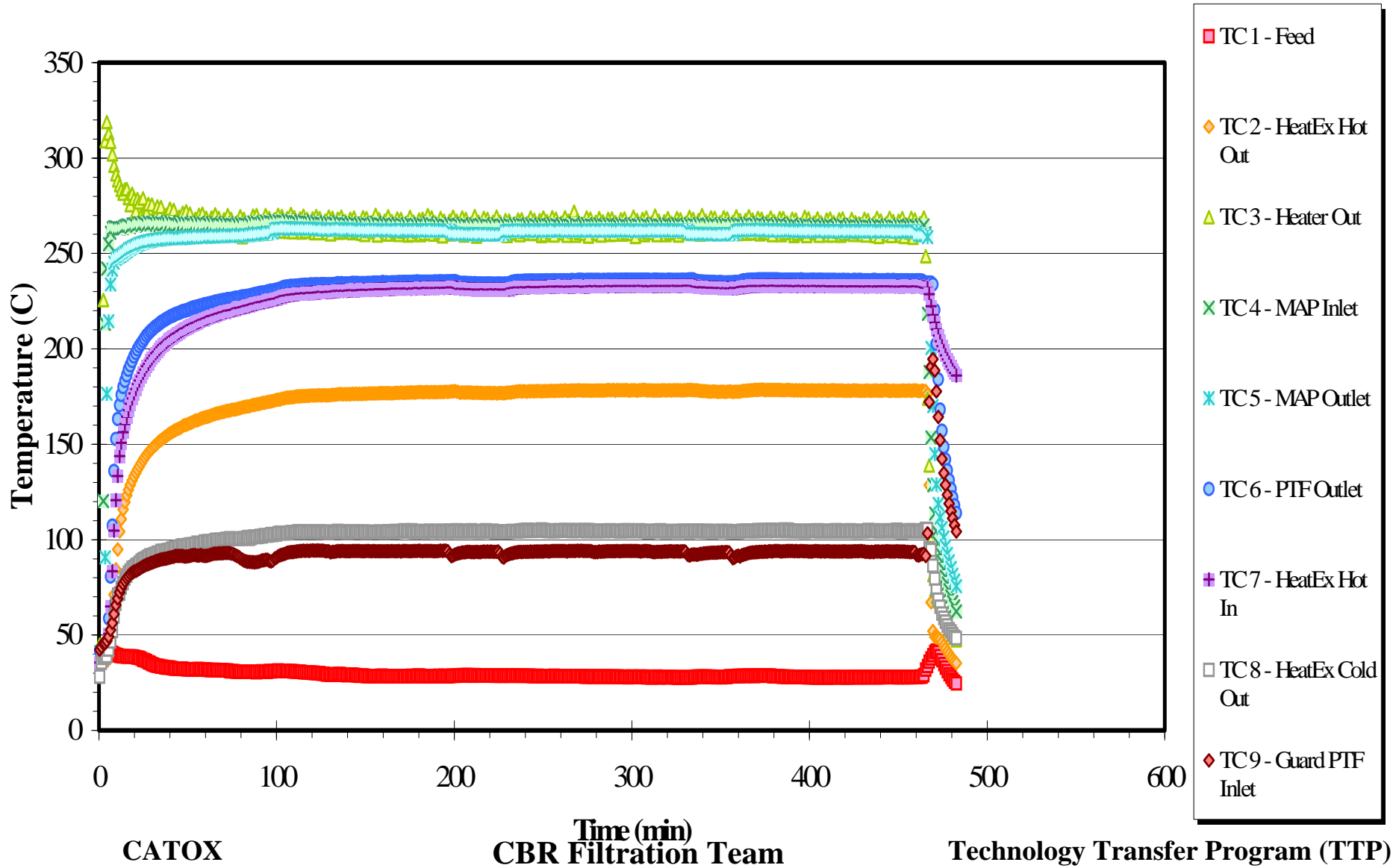


Ammonia (LC): Temperature

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ECBC 50 SCFMCATOX UNIT

Temperature Plot





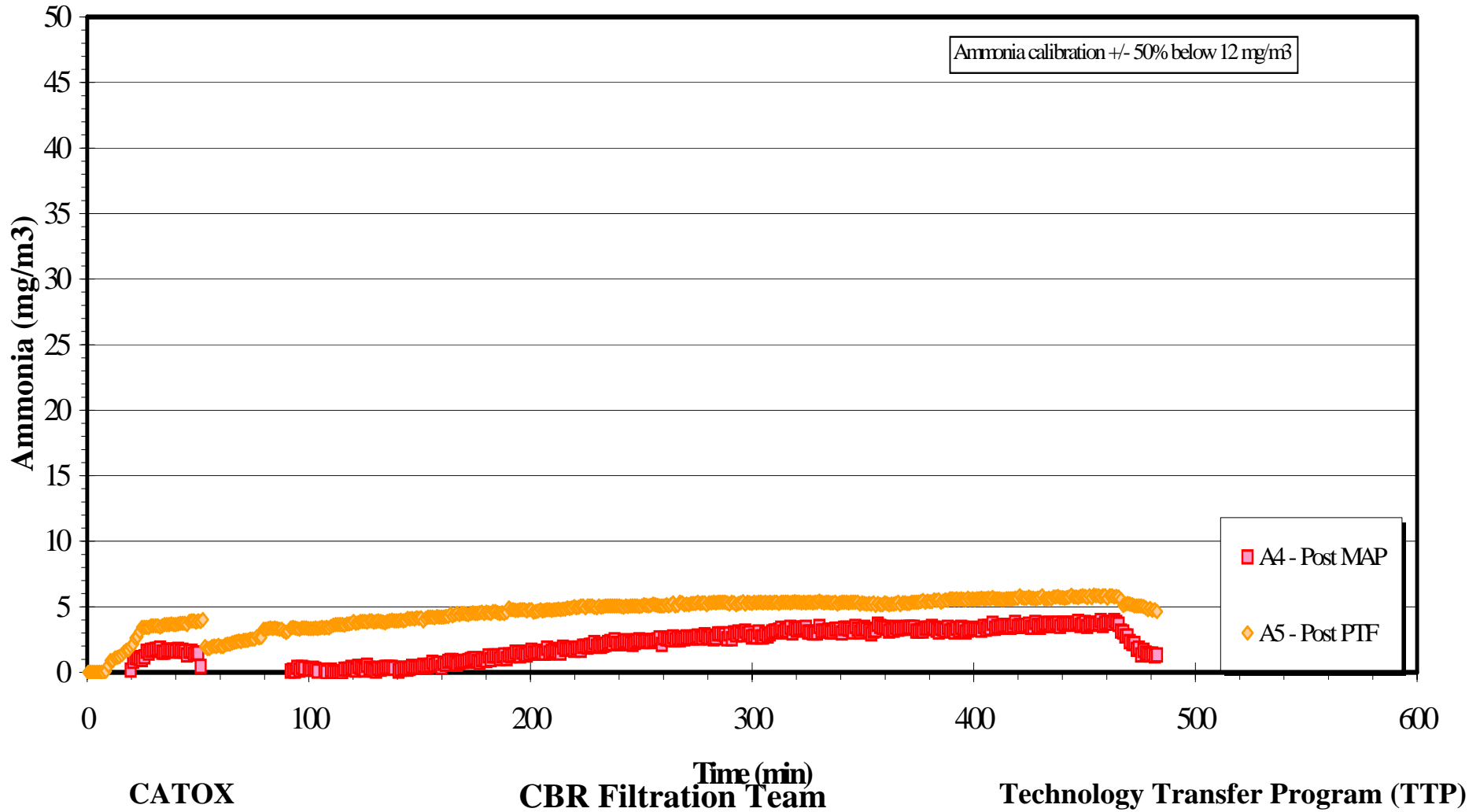
Ammonia (LC): Parent

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ECBC 50 SCFM CATOX UNIT

NH3-HW-BLc P3-5

Effluent Chart (NH3) 5-10-04



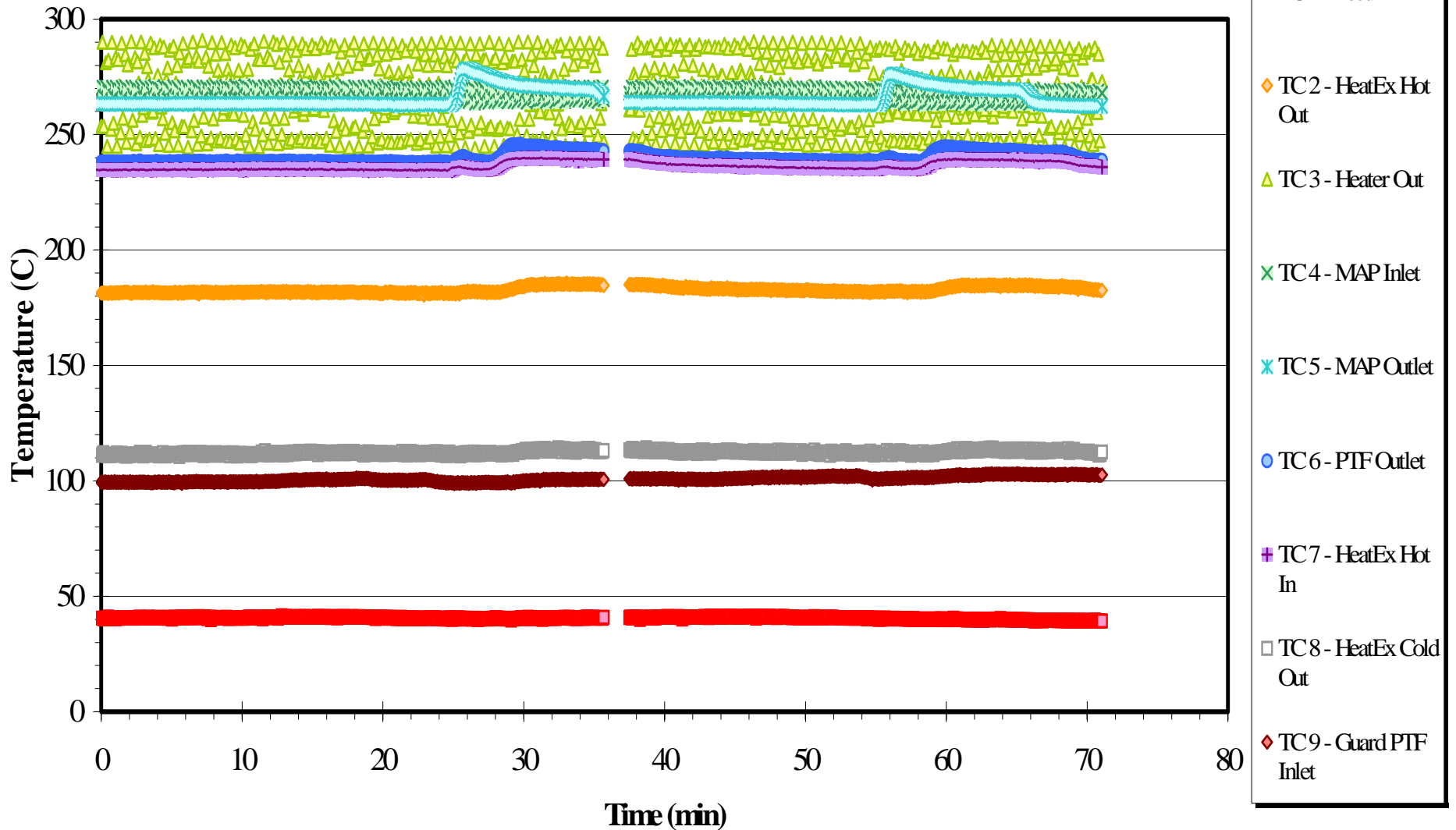


Ethylene Oxide (HC): Temperature

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Temperature Plot
EO-HW-BHc 6-18-04



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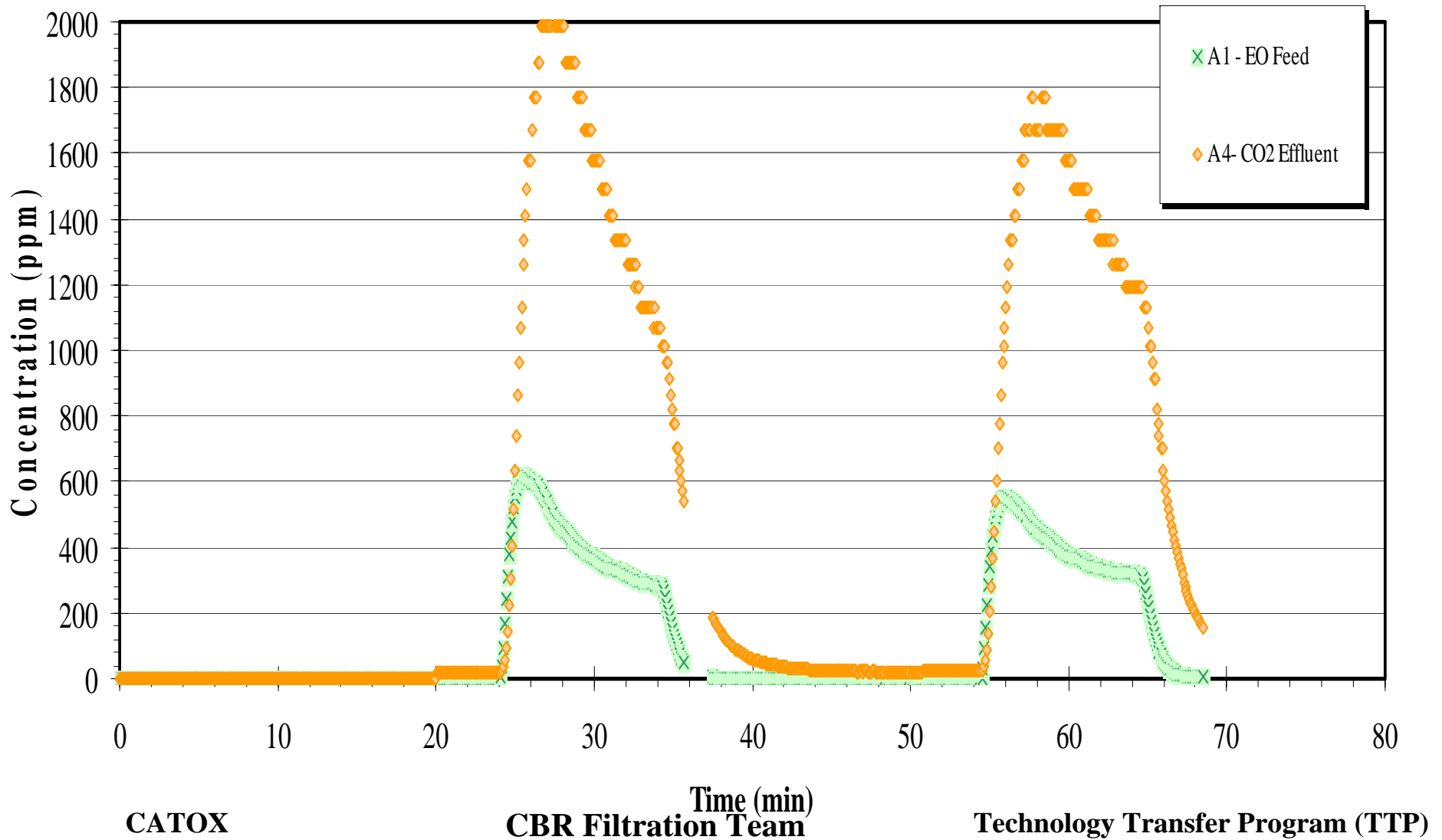
Ethylene Oxide (HC): Feed Concentrations

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EO-HW-BHc 6-18-04

EO Feed - CO₂ Effluent



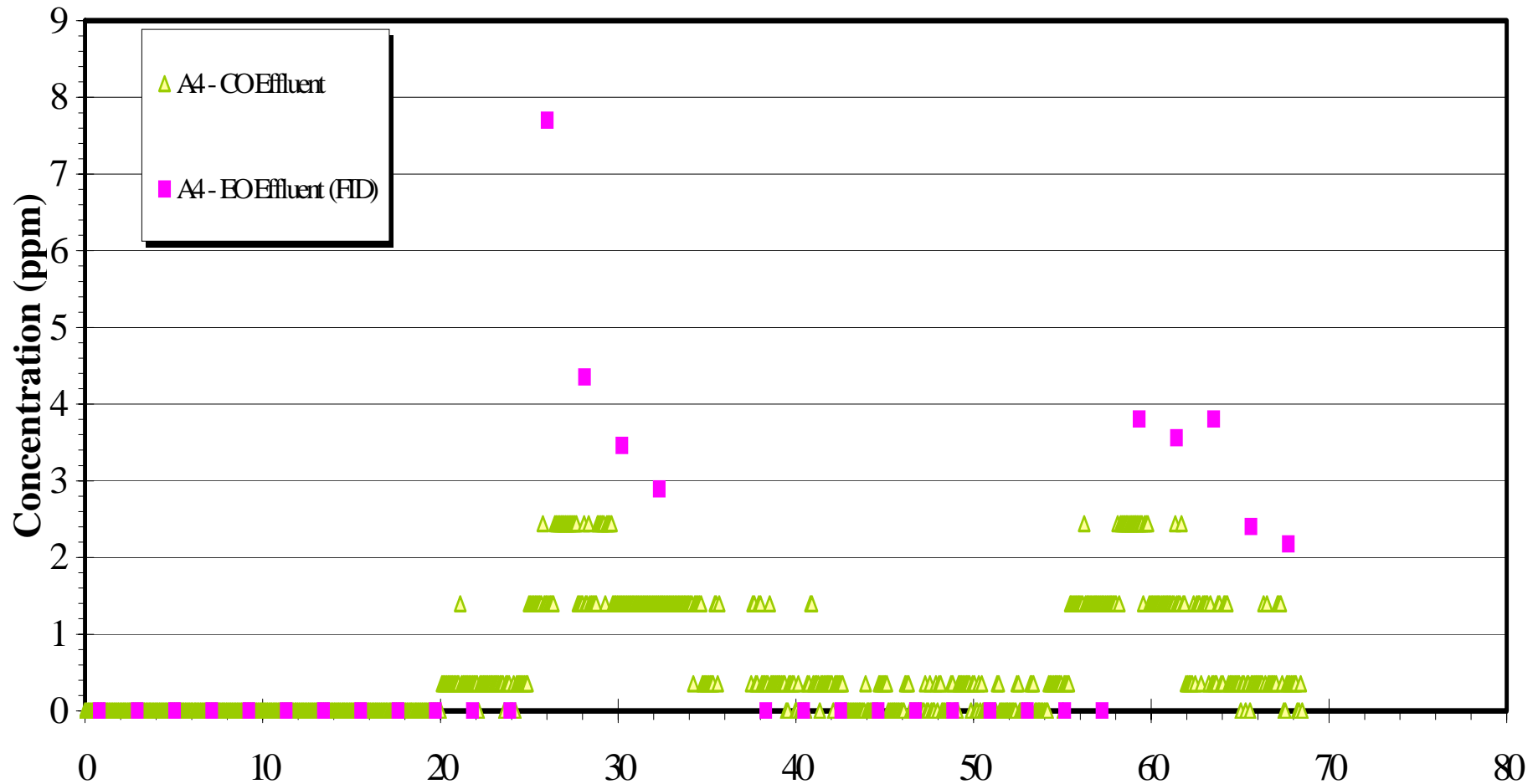


Ethylene Oxide (HC): By-products

ECBC50SCFMCATOXUNIT

EO-HW-BHc 6-18-04

EO-COEffluent



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Time (min)
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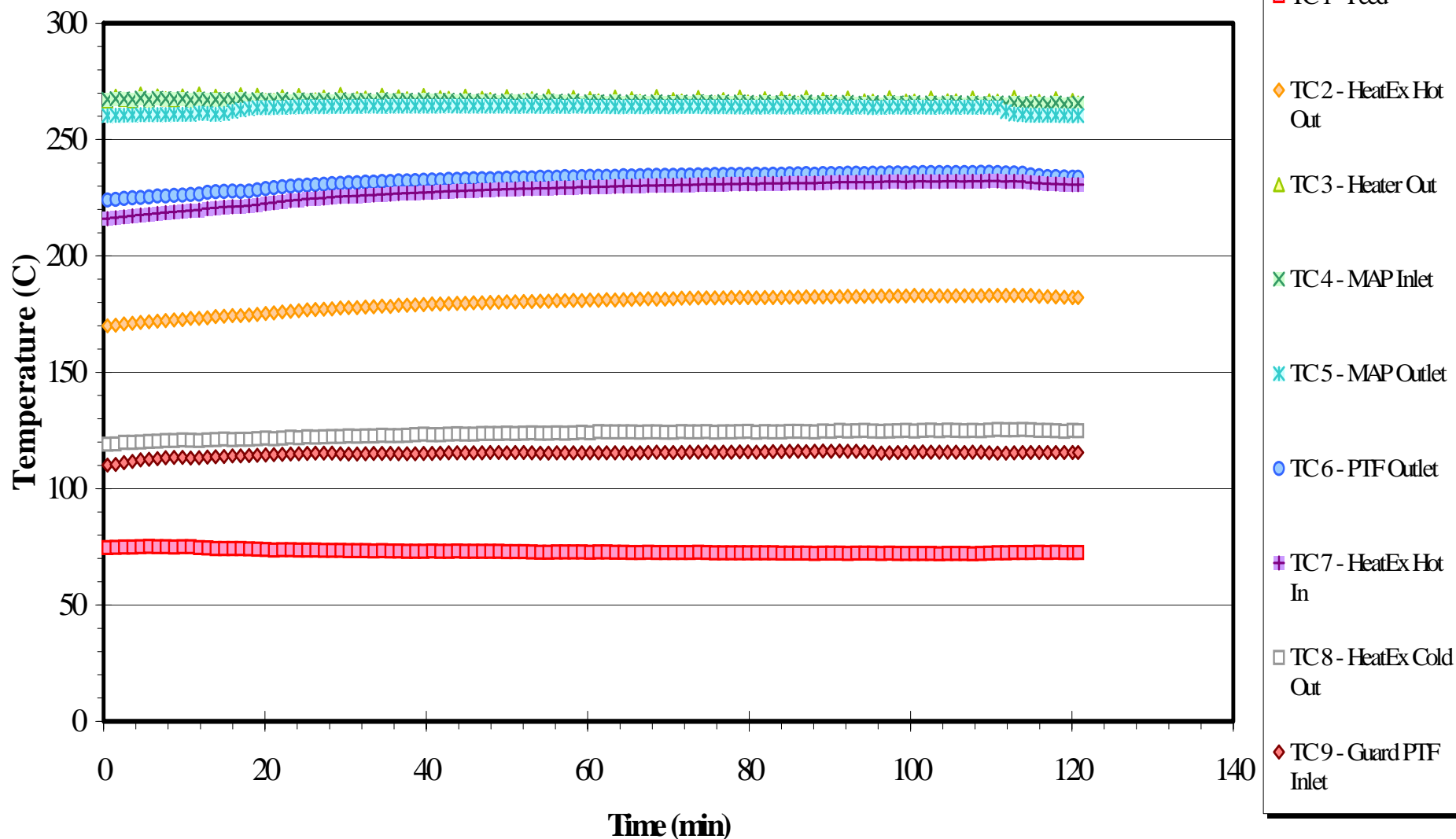


Formalin (LC)

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Temperature Plot

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CATOX

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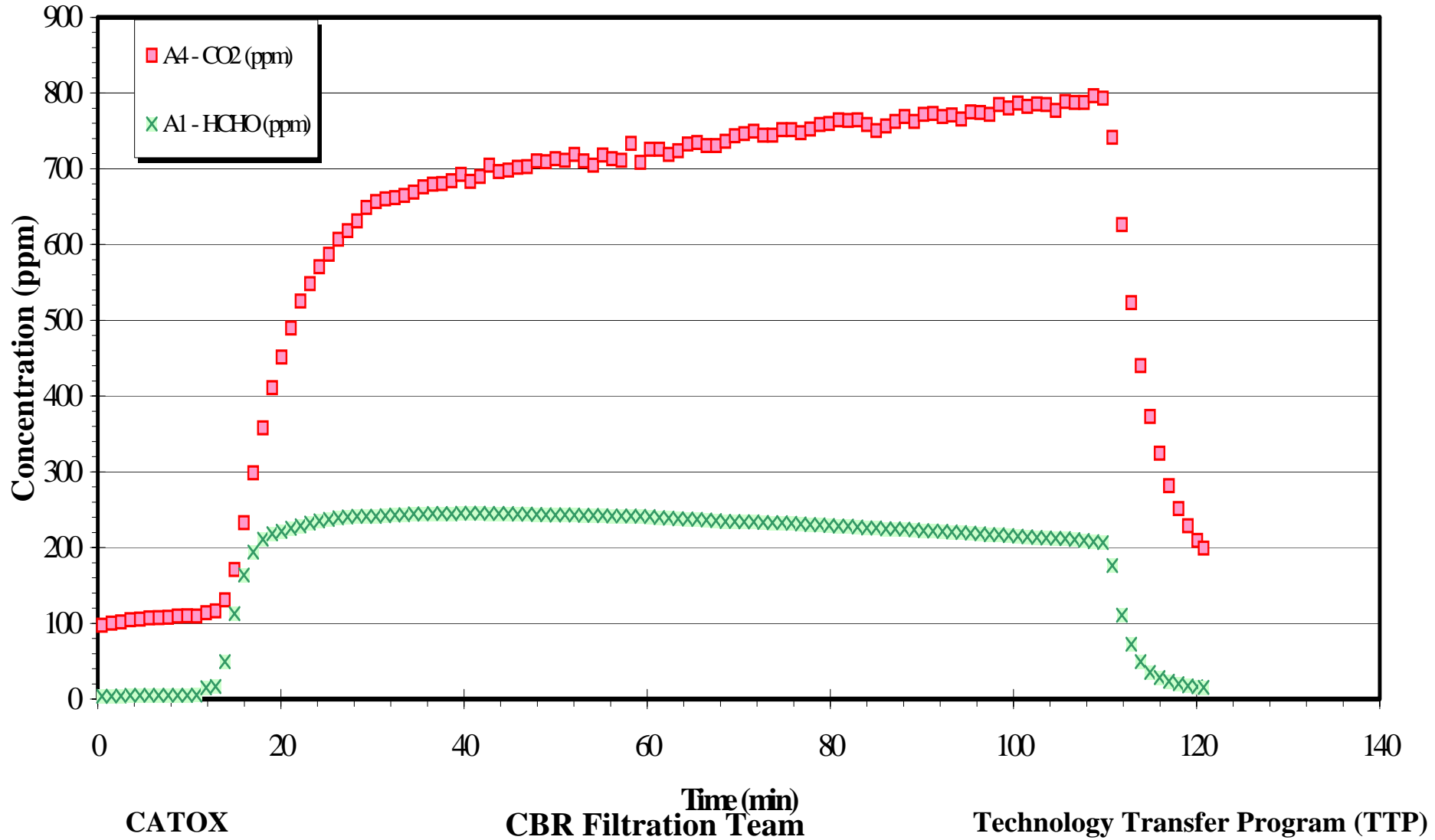


Formalin (LC)

ECBC 50 SCFM CATOX UNIT

HCHO-HW-DLc 7-14-04

A1 - HCHO A4 - CO₂



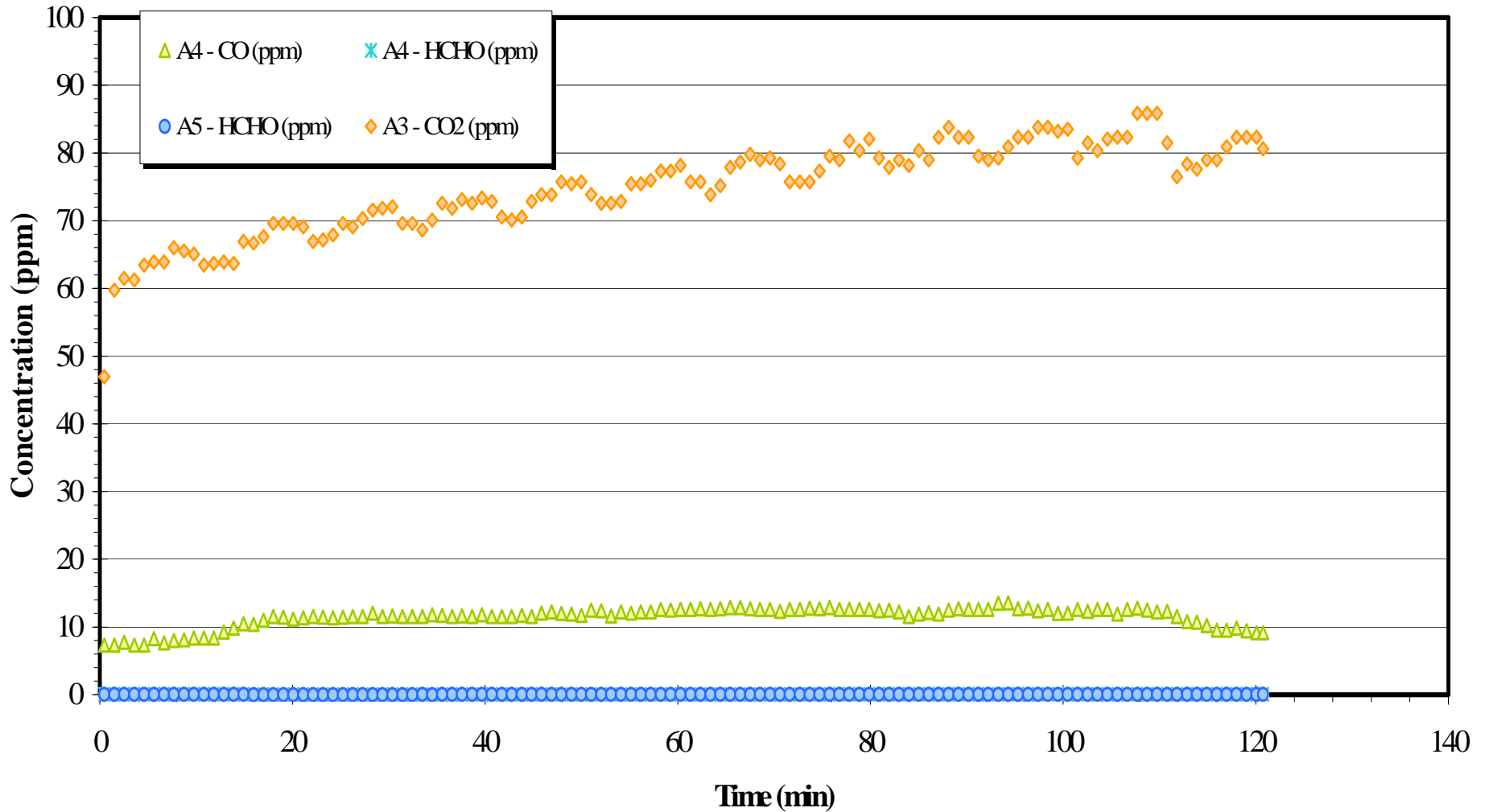


Formalin (LC)

ECBC 50 SCFMCATOX UNIT

HCHO-HW-DLc 7-14-04

A3 - CO₂ A4 - CO A4 - HCHO A5 - HCHO



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Chemicals tested:

Carbon Monoxide

Ammonia

Ethylene Oxide

Formalin

Chemical underway:

Acetonitrile

Chemicals left:

Chloroform

CK

CS₂

Nitric Acid

CEES

HF

HFP

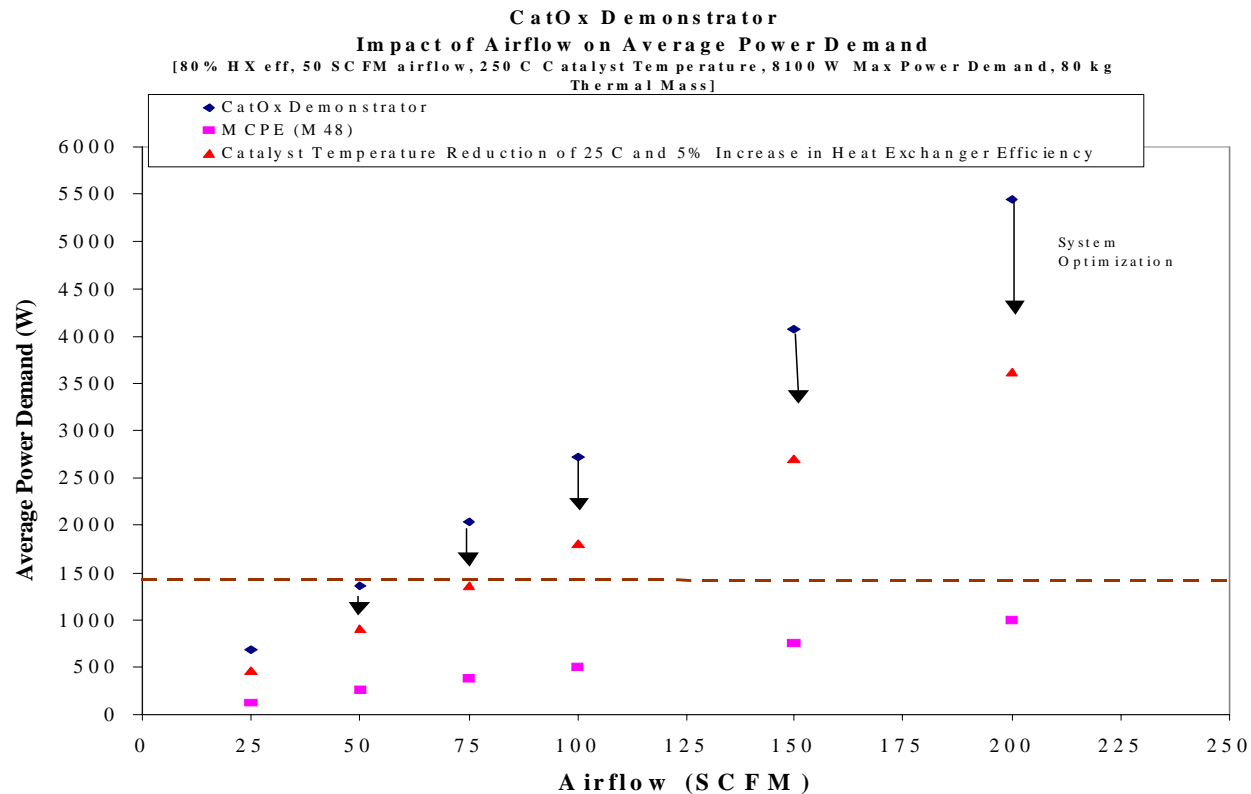
DMMP



Field conditions impacting CATOX

In a fielded system, one catalyst bed operating at one flow rate and one operating temperature will be employed.

Power and weight optimization





Improvement of Subcomponents

Next Generation PTS

Catalyst Improvements

Heat Exchanger (greater 90% heat recovery efficiency)

Modular System

Lighter, Smaller Overall Footprint



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