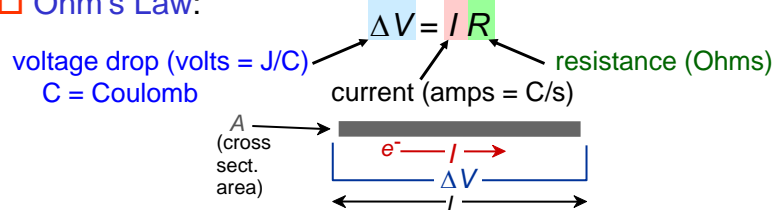


## Chapter 18 Electrical properties

- Introduction
- Ohm's law
- Electrical conductivity
- Electronic and ionic conduction
- Energy band structures in solids
- Conduction in terms of band and atomic bonding atoms
- Electron mobility
- Electrical resistivity of metals
- Electrical characteristics of commercial alloys
- Intrinsic semiconductors
- Extrinsic semiconductors

## Electrical conduction

### □ Ohm's Law:



### □ Resistivity, $\rho$ and conductivity, $\sigma$ :

- geometry-independent forms of Ohm's Law
- Resistivity is a material property & is independent of sample

$$\frac{\Delta V}{L} = \frac{I}{A} \rho$$

$E$ : electric field intensity

resistivity (Ohm-m)

$J$ : current density

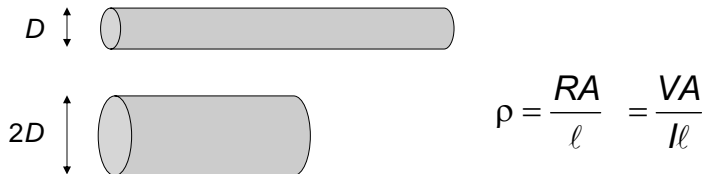
### □ Resistance:

$$R = \frac{\rho L}{A} = \frac{L}{A\sigma}$$

conductivity  $\rightarrow \sigma = \frac{1}{\rho}$

## Electrical properties

- Which will conduct more electricity?



- Analogous to flow of water in a pipe
- So resistance depends on sample geometry, etc.

## Definitions

- Further definitions

- $J = \sigma \varepsilon$  =< another way to state Ohm's law

- $J \equiv$  current density =  $\frac{\text{current}}{\text{surface area}} = \frac{I}{A}$  like a flux

- $\varepsilon \equiv$  electric field potential =  $V/\ell$  or  $(\Delta V/\Delta \ell)$

$$J = \sigma (\Delta V/\Delta \ell)$$

↑                      ↑                      ↑  
Electron flux    conductivity    voltage gradient

### Current carriers

- electrons in most solids
- ions can also carry (particularly in liquid solutions)

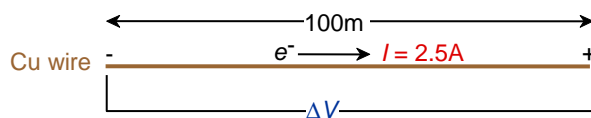
## Conductivity: comparison

□ Room  $T$  values  $(\text{Ohm}\cdot\text{m})^{-1} = (\Omega \cdot \text{m})^{-1}$

<b>METALS</b>	<b>conductors</b>	<b>CERAMICS</b>	
Silver	$6.8 \times 10^7$	Soda-lime glass	$10^{-10}$ - $10^{-11}$
Copper	$6.0 \times 10^7$	Concrete	$10^{-9}$
Iron	$1.0 \times 10^7$	Aluminum oxide	$<10^{-13}$
<b>SEMICONDUCTORS</b>		<b>POLYMERS</b>	
Silicon	$4 \times 10^{-4}$	Polystyrene	$<10^{-14}$
Germanium	$2 \times 10^0$	Polyethylene	$10^{-15}$ - $10^{-17}$
GaAs	$10^{-6}$		
	<b>semiconductors</b>		<b>insulators</b>

## Example: Conductivity problem

What is the minimum diameter ( $D$ ) of the wire so that  $\Delta V < 1.5 \text{ V}$ ?

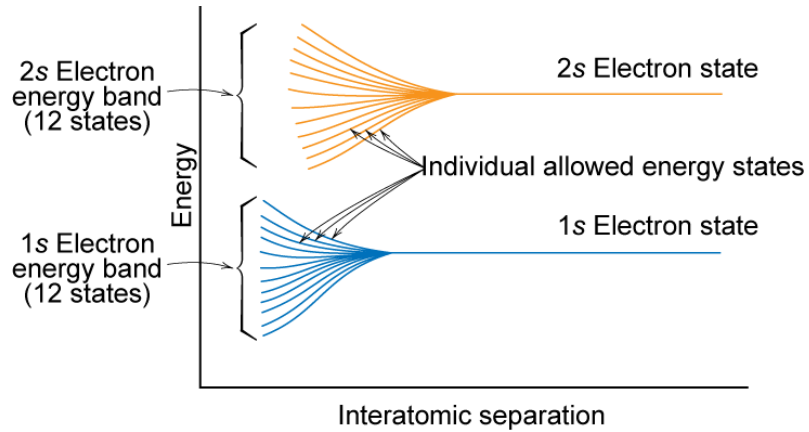


$$R = \frac{L}{A\sigma} = \frac{\Delta V}{I}$$

$\frac{\pi D^2}{4}$

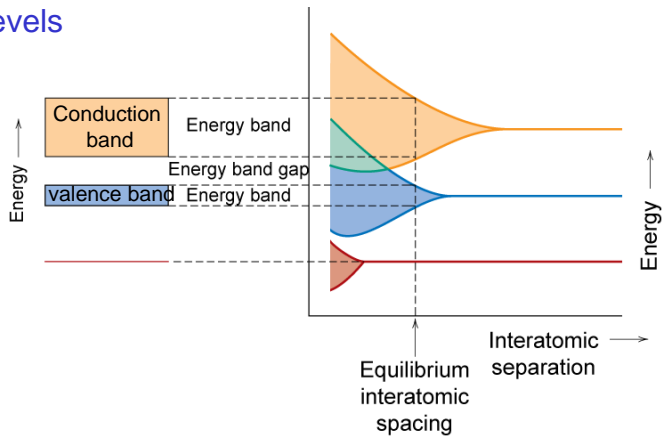
Solve to get  $D > 1.87 \text{ mm}$

## Electronic band structures



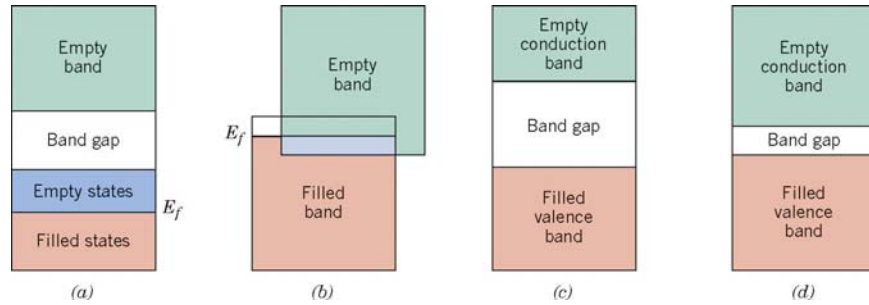
## Band structure

- Valence band – filled – highest occupied energy levels
- Conduction band – empty – lowest unoccupied energy levels



## Various possible electron band structures

□ Fermi energy  $E_f$ : the energy corresponding to the highest filled state at 0 K



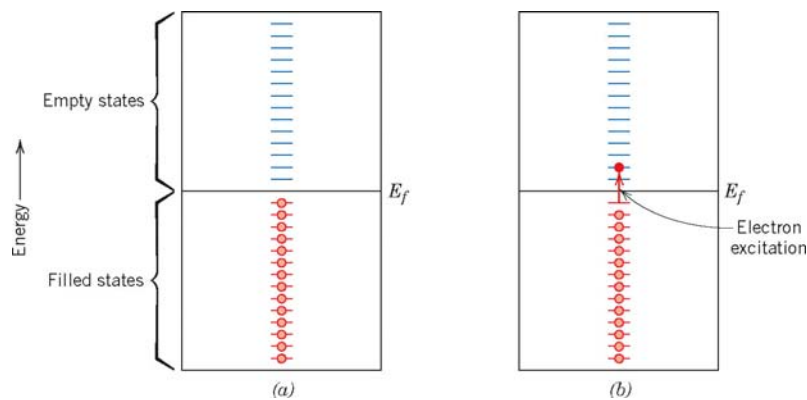
**Metal (Cu)**

**Metal (Mg)**

**Insulator**

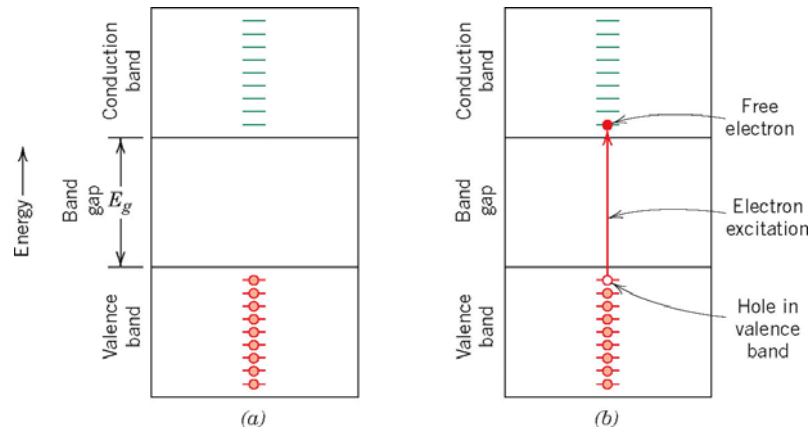
**Semiconductor**

## Conduction in terms of band and atomic bonding models



**For a metal, occupancy of electron states (a) before and (b) after an electron excitation**

## Conduction in terms of band and atomic bonding models



**For an insulator or semiconductor, occupancy of electron states (a) before and (b) after an electron excitation**

## Electron mobility

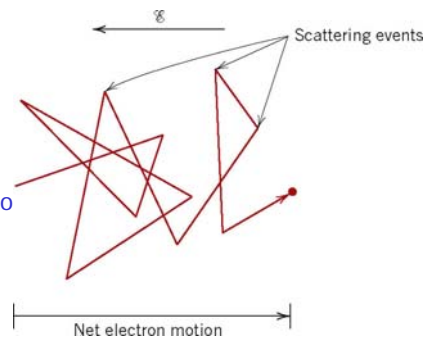
- Drifting velocity: average electron velocity in the direction of the force imposed by the applied field

$$v_d = \mu_e \mathcal{E}$$

- Electron mobility  $\mu_e$  : an indication of the frequency of scattering events ( $\text{m}^2/\text{V}\cdot\text{s}$ )

- The conductivity

$$\sigma = n |e| \mu_e$$



## Electrical resistivity of metals

□ Mathiessen's rule

$$\rho_{total} = \rho_t + \rho_i + \rho_d$$

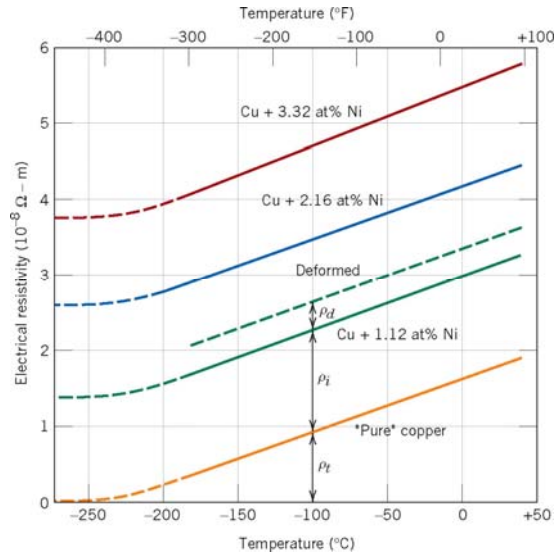
- Influence of T

$$\rho_t = \rho_0 + aT$$

- Influence of impurity

$$\rho_i = Ac_i(1-c_i)$$

- Influence of plastic deformation: increased numbers of electron-scattering dislocations

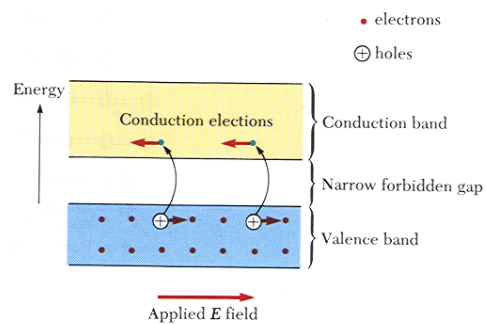


## Intrinsic semiconductor

- Intrinsic semiconductor: its electrical behavior is based on the pure material

- Intrinsic semiconductor band structure

- A current arises from the flow of electrons and holes



## Intrinsic semiconductor

### □ Elemental semiconductor

- Group IVA, Si (E<sub>gap</sub> 1.1 eV), Ge (E<sub>gap</sub> 0.7 eV)

### □ Compound semiconductor

- Group IIIA-VA, GaAs, InSb
- Group IIB-VIA, CdS and ZnTe

### □ Electric conductivity

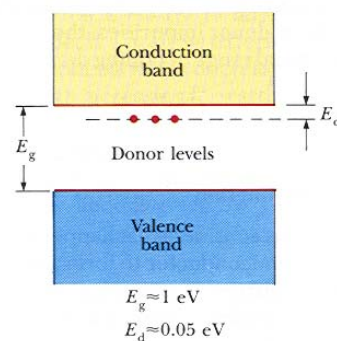
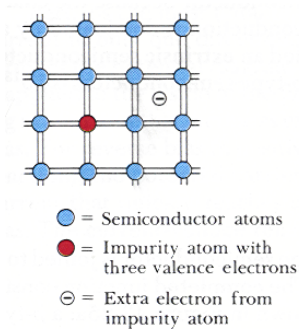
$$\sigma = n|e|\mu_e + p|e|\mu_h$$

## Extrinsic semiconductor

### □ The electrical behavior is determined by impurities.

### □ N-type semiconductor

- Majority carrier: electrons
- Minority charge carriers: holes

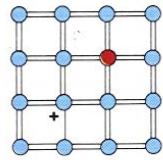




## Extrinsic semiconductors(cont.)

□ P-type semiconductor: trivalent substitutional impurities  
aluminum, boron, and gallium

- Majority carrier: holes
- Minority charge carriers: electrons



- = Semiconductor atoms
- = Impurity atom with three valence electrons
- ⊕ = Hole, or electron deficiency in a bond

