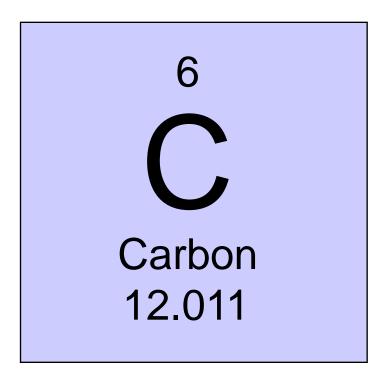
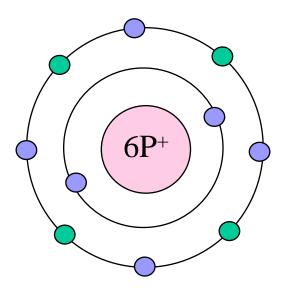
Chapter 2-3 Carbon Compounds

A. Organic compounds- originally thought to be compounds produced by living organism, now it refers to compounds containing carbon.

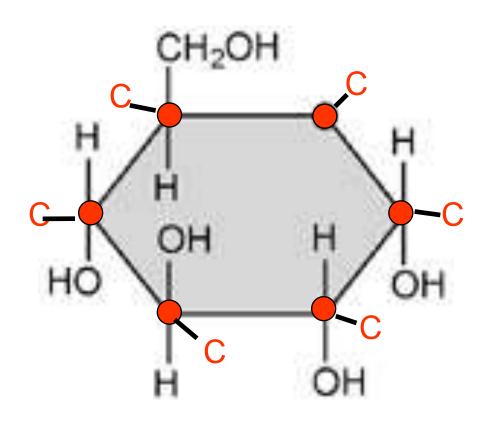


- 1. The carbon atom is unique and carbon compounds are found in living organisms.
 - a. Carbon atoms have 4 valence electrons

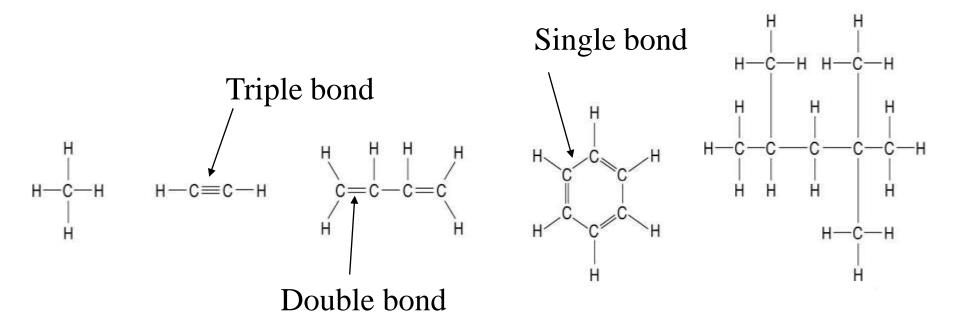


Carbon atoms need to share with 4 additional electrons to achieve an octet—this opens up all sorts of bonding possibilities.

b. Carbon atoms can bond to other carbon atoms so that carbon compounds can form chains and rings.

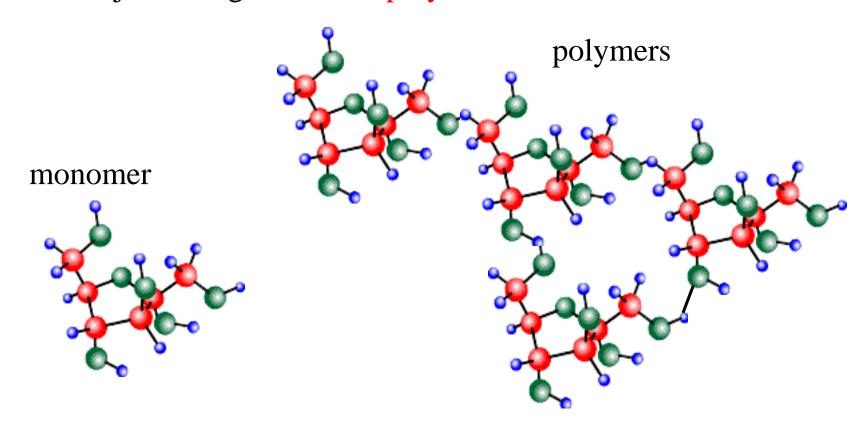


c. Carbon can form millions of different large and complex structures. It is a great building material for living organisms.

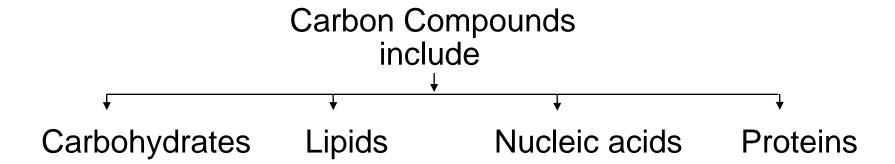


1. Macromolecules—giant molecules made from smaller molecules.

- a) Formed from polymerization, the joining of smaller molecules into large molecules.
- b) Smaller molecules are called monomers and when joined together form polymers.



2. Organic (C) compounds are classified into 4 groups.



a. Carbohydrates—

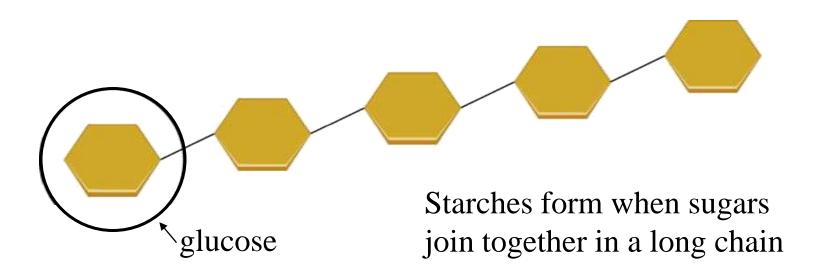
- 1. Compounds made of carbon, hydrogen and oxygen
- 2. glucose is an example: $C_6 H_{12} O_6$

3. The ratio between elements is usually 1:2:1

4. Functions are a source of quick energy and structural building.

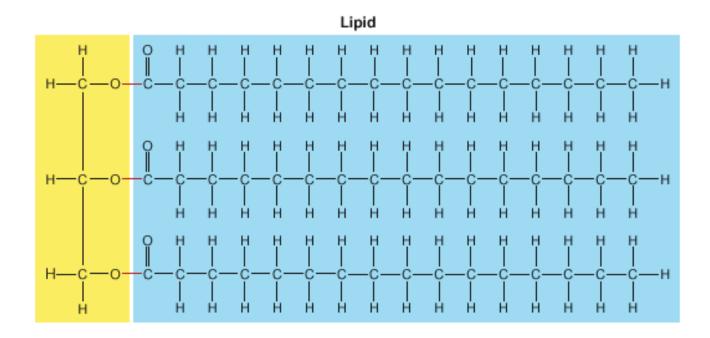
5. Sugars and starches are carbohydrates

- 6. Simple sugars are known as monosaccharides, examples would include glucose, galactose and fructose.
- 7. Polysaccharides are complex sugars made from monosaccharides, examples would be glycogen (a starch), and cellulose.

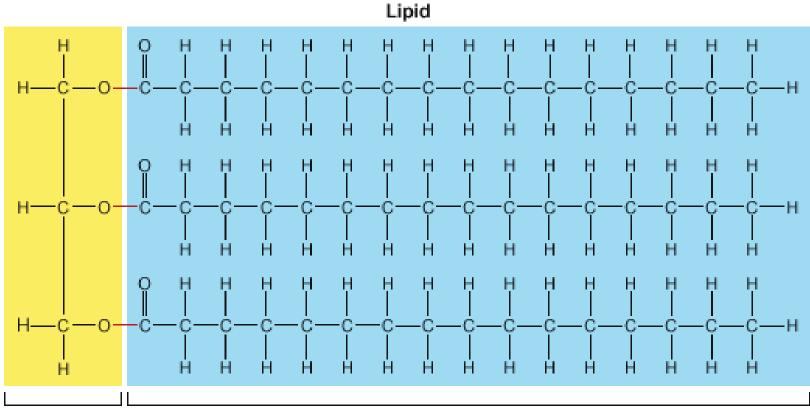


b. Lipids—

- 1. Compounds made of carbon, hydrogen and oxygen
- 2. $C_{54}H_{98}O_{6}$



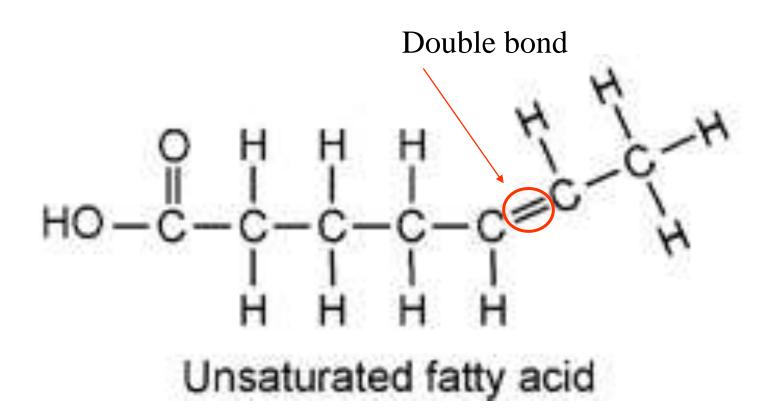
- 3. Fats, oils, waxes
- 4. 3 Fatty acids and 1 glycerol are the smaller molecules that compose lipids



Glycerol Fatty acids

5. Saturated- this term refers to a lipid compound that has C atoms joined to other C atoms by single bonds. This allows the maximum number of H atoms to bond to the C atoms.

6. Unsaturated- this refers to a lipid compound that has at least 1 double bond between C atoms. This means that the maximum number of H atoms are not bonded to the C atoms.

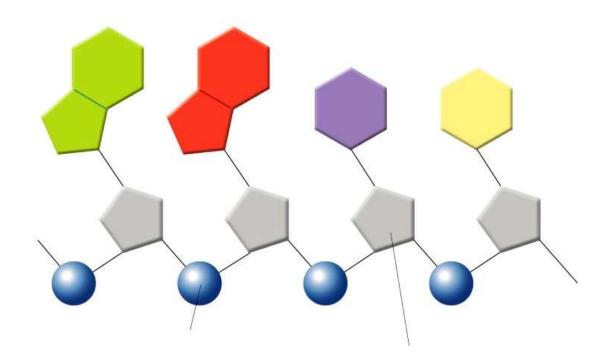


7. Polyunsaturated- lipid compounds that have more than 1 double bond between C atoms.

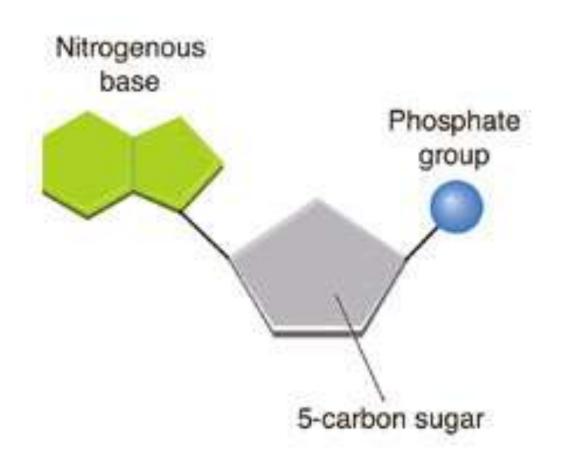
Unsaturated fatty acid

c. Nucleic Acids—

- 1. Macromolecules containing carbon, hydrogen, oxygen, nitrogen and phosphorus.
- 2. Composed of smaller molecules (monomers) called nucleotides.



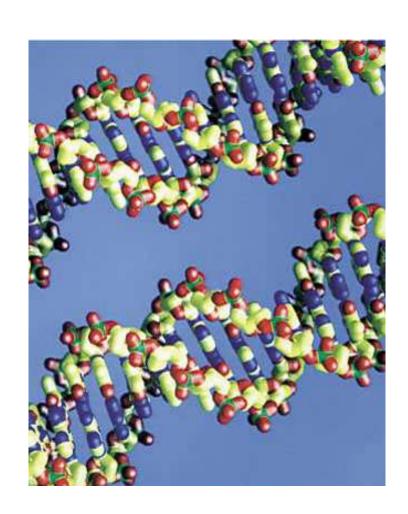
a) Nucleotides are composed of 3 parts: 5 C sugar, a phosphate group and a nitrogenous base.



3. Stores and transmits genetic information.

4. 2 types:

- a) RNA- ribonucleic acid: ribose sugar, translates the genetic information
- b) DNA- deoxyribonucleic acid: deoxyribose sugar, stores the genetic information

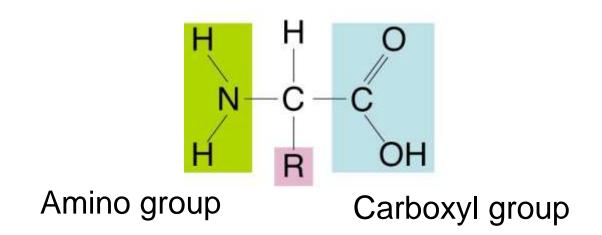


d. Proteins

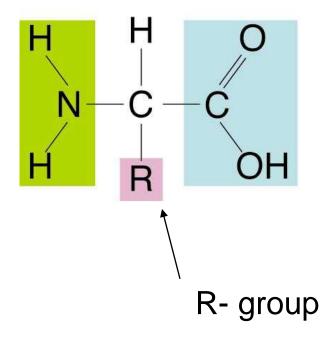
1. Macromolecules that contain Carbon, Hydrogen, Oxygen and Nitrogen.

Protein macromolecule Composed of amino acids. Amino acids

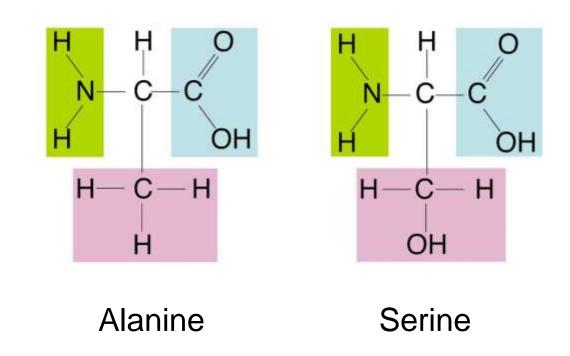
a) Amino acids are smaller molecules with an amino group (-- NH₂) on one end and a carboxyl group (-- COOH) on the other.



b) The section of the amino acids that is different is called an R-group.

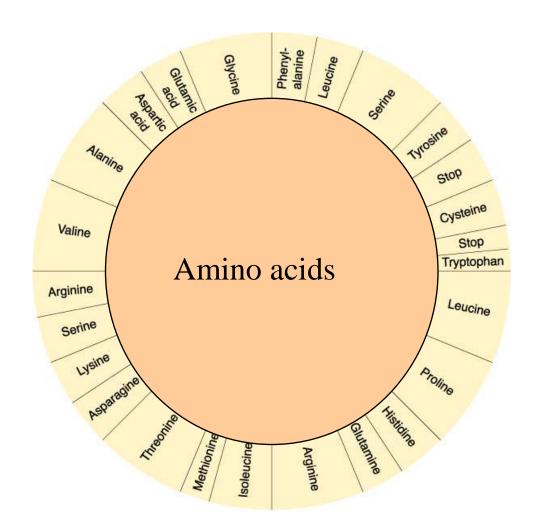


R-groups have a variety of characteristics which we use to classify amino acids into different levels.



- c) There are 4 levels based on their structure.
 - 1) The sequence of amino acids in a protein chain
 - 2) Amino acids within a chain can be twisted or folded
 - 3) The chain itself is folded and
 - 4) If multiple chains, the specific arrangement in space.

3. There are more than 20 amino acids which can be joined together to form a huge variety of proteins.



4. These proteins are extremely large molecules and are

used for:

a) building structures,

b) hormones or

c) enzymes.

