Chapter 5 Cellular Respiration

Releasing Stored Energy

- 1. Aerobic Cellular Respiration- Carried out by organisms in oxic (oxygen containing) environment.
- 2. Anaerobic Cellular Respiration- Anoxic (nooxygen containing) environment.
- 3. Fermentation- Modified process- anaerobic respiration.

5.3 Cellular Respiration

- A series of chemical reactions that <u>break down</u> <u>glucose</u> to release energy
- The energy is then stored in the form of ATP
- Formula is:



Cellular respiration banks energy as ATP

- <u>1 molecule of glucose</u> has a LOT of energy (too much at once!)
- Cells <u>change it</u> into a more <u>useable</u> form = <u>ATP</u>
- It's like changing a \$100 bill for 100 loonies (easier to spend loonies)
- <u>64%</u> of energy is released as <u>HEAT-</u> (keeps you <u>warm</u>!)
 (Exam Question)











Cellular Respiration Takes Place in Four Stages

- Glycolysis Anaerobic
 Cytoplasm
- **2.** Pyruvate Oxidation- AerobicMitochondria
- 3. Kreb's cycle- Aerobic
 - Matrix of mitochondria
- 4. Electron Transport Chain and Chemiosmosis- Aerobic
 Inner membrane of mitochondria

































Questions

Section 7.1 Questions

- Briefly describe one cellular process that involves the use of active transport. How is ATP involved in this process?
 Why is cellular respiration necessary?
 What are the four stages of aerobic respiration?
- What are the characteristics of glucose that make it well suited as an energy supply molecule within our bodies?
 3. Briefly active ta

 The conversion of glucose energy to ADP energy is less than 50 % efficient. In what way do birds and mammals take advantage of this inefficiency?
 4. Wity is 5. What at take advantage of this inefficiency?

 1
 Glucose has a birdy naparruy content is relative to advantage of this inefficiency?
- 1. Glucose has a <u>high energy</u> content, is relatively small, and is highly soluble.
- 2. Both animals use the waste energy (heat) to keep their bodies warm. Both organisms are warm-blooded.
- 3. Two processes that require the use of ATP are the <u>movement of</u> <u>material</u> up the concentration gradient (low to high concentration) and the movement of specific molecules <u>through proteins</u> in the cell membrane. Both of these processes require ATP to change the nature of the cell membrane or activate the membrane proteins.
- 4. Cellular respiration is required to convert stored food energy into the <u>usable form of ATP</u>.
- 5. Pyruvate Oxidation, Krebs cycle, and Oxidative Phosphorylation.



3. Kreb's Cycle

- Takes place in the matrix of the mitochondria
- Starts with <u>Acetyl-CoA</u>
- Acetyl-CoA is <u>oxidized</u>, NAD+ and FAD are reduced

3. Kreb's Cycle

- 2 carbons enter as acetyl CoA
- Binding to <u>Oxaloacetate (4 carbon)</u>
- Forming **6** carbon molecule
- 2 Carbons leave as CO2
- <u>2 NAD</u>⁺ are reduced to form 2 <u>NADH</u>
- ATP is formed
- FAD is reduced to form FADH₂
- Additional <u>NAD</u>⁺ are reduced to form <u>NADH</u>
 - Note: 2 pyruvate enter cycle- per glucose molecule









The diagram is a simplified view of stage 2 and stage 3 of aerobic respiration.

Circle the part of the diagram that shows pyruvate oxidation.
 What happens to the carbon atoms in acetyl CoA?
 The carbon atoms in acetyl CoA are used in the formation of CO2.

3. Identify the product(s) of the Krebs cycle that provides energy for cell processes. How many molecules of this(these) are produced for every turn of the Krebs cycle?

The product of the Krebs cycle that provides energy for cell processes is ATP. Each turn of the Krebs cycle produced 2 ATP, 6 NADH, and 2 FADH- per glucose molecule.

4. Identify the product(s) that provides reducing energy for cell processes. How many molecules of this(these) are produced for every turn of the Krebs cycle?

The products of the Krebs cycle that provide reducing energy for cell processes are NADH and FADH2. Each turn of the Krebs cycle produces 6 NADH molecules and 2 FADH2 molecules







As the electrons move ...

 $\ensuremath{\textcircled{\bullet}}$ A small amount of energy is released as they are passed from protein to protein

This energy is used to move <u>H+ into the intermembrane space</u>

OXYGEN

● Is the final electron acceptor ● It is reduced to form water ● $2 H^+ + \frac{1}{2} O_2 + 2e^- \longrightarrow H_2 O_2$



What happens to the ETC if there is no O₂?

A lack of oxygen causes the system to back up all the way to glycolysis because the NADH and FADH₂ can't be recycled. Hence why if we don't take in O₂, no ATP = cells die = you die!





- The ATP synthase channel is the only place permeable to H⁺
- •As hydrogen flow back into the matrix from the intermembrane space energy is released
- This energy is used to make ATP32 per glucose





Products- Cellular Respiration

 <u>http://www.warrencountyschools.org/userfile</u> s/2623/Classes/10683/Cell%20Respiration%2 <u>0Chart.pdf</u>



Crash course Biology Review





Label Diagram 9: Chemiosmosis



Fill in the missing labels in this outline of chemiosmosis.

One molecule of ATP is synthesized from <u>ADP</u> and <u>P</u> as an H⁺ ion passes through <u>the ATP Synthase Channel</u> (complex) into the mitochondrial matrix from the H+ reservoir in the intermembrane space.

Total ATP produced:

2 ATP from glycolysis
2 ATP from Kreb's
32 ATP from the electron transport and chemiosmosis
Total = 36 ATP per glucose



Section 7.3 Questions (Page 220) #1-10 Key

- Arrange the following types of cells in order of increasing number of mitochondria in the cytoplasm: nerve cell, skin cell, fat cell, heart muscle cell. Provide a rationale for your sequence.
- 2. (a) In eukaryotic cells, where does glycolysis occur?(b) What two products of glycolysis may be transported into mitochondria for further processing?
- Heart muscle cells have the most mitochondria as they require the most energy to contract (approximately 70 times per minute). Nerve cells have the second most mitochondria as they need to maintain the membrane potential necessary to conduct a nerve impulse. Skin cells are next as their functions require little energy, followed by fat cells, which provide insulation and nutrient storage for the body and therefore contain few mitochondria.
- 2. (a) Glycolysis occurs in the cytoplasm of eukaryotic organisms.
 (b) The two products of glycolysis that enter the mitochondria are pyruvate and NADH.

Page 220 #3

- Describe two functions that mitochondrial membranes serve in energy metabolism.
- 3. Mitochondrial membranes perform several vital roles in energy metabolism. The outer membrane of the mitochondria acts as a cell membrane and houses transport proteins that allow substances in and out of the mitochondria. For instance, the outer membrane houses transport proteins, which move the two pyruvate molecules formed during glycolysis from the cytoplasm into the mitochondria, where they undergo pyruvate oxidation before entering the Krebs cycle. The inner membrane of mitochondria servers several functions. It divides the mitochondrion into two compartments: the matrix and the intermembrane space. Both of these areas play important roles in energy metabolism. For instance, the matrix is where most of the Krebs cycle reactions take place and the intermembrane space is where protons are used to create the electrochemical gradient that stores free energy, which is necessary to create ATP. The inner membrane of mitochondria also houses the numerous proteins and cofactors required ultimately to generate ATP. NADH hydrogenase, cytochrome b-c1 complex, cytochome indica membrane.

4. Why is aerobic cellular respiration a more efficient energyextracting process than glycolysis alone?

4. Glycolysis is not considered a highly effective energy-harnessing mechanism because it transfers only about 2.1 % of the free energy available in one molecule of glucose into ATP. Most of the energy is still trapped in two pyruvate molecules and 2 NADH molecules. Aerobic respiration further processes the pyruvate and NADH during pyruvate oxidation, the Krebs cycle, chemiosmosis, and electron transport. During pyruvate oxidation, the pyruvate and NADH are transformed into two molecules each of acetyl-CoA, hydrogen, carbon dioxide, and NADH. Acetyl-CoA enters the Krebs cycle and increases ATP production. By the end of the Krebs cycle, the entire original glucose molecules is consumed. It has been transformed into 6 CO2 molecules, which are released as waste, and energy, which is stored as 4 ATP molecules and 12 reduced coenzymes (NADH and FADH2). Most of the free energy stored in NADH and FADH2 will be transformed to ATP in the final stage of aerobic respiration, chemiosmosis, and electron transport. By the end of aerobic respiration, all the energy available in glucose has been harnessed.

- (a) What part of a glucose molecule provides electrons in cellular respiration?
- (b) Describe how electron transport complexes set up a proton gradient in response to electron flow.
- (c) How is the energy used to drive the synthesis of ATP?
- (d) What is the name of this process?
- (e) Who discovered this mechanism?

5. (a) The hydrogens

- (b) The electron transport chain creates a proton gradient by pumping electrons from the matrix to the inner membrane. The membrane becomes impermeable to protons due to their high concentration and forces the protons to move through proton channels to form ATP.
- (c) The potential energy stored in this gradient causes the protons to move through special protein channels, and the energy released is used to form ATP.
- (d) This process is termed chemiosmosis or oxidative phosphorylation.
- (e) Chemiosmosis was discovered by Peter Mitchell in 1961.

- (a) Distinguish between an electron carrier and a terminal electron acceptor.
 - (b) What is the final electron acceptor in aerobic respiration?
- **7.** Explain how the following overall equation for cellular respiration is misleading:

$$C_6H_{12}O_6 + 6O_2 \longrightarrow 6CO_2 + 6H_2O_2$$

- (a) An electron carrier is a molecule that can accept and donate electrons from and to various enzymes. A terminal electron acceptor is the final substance that receives electrons in an oxidation-reduction reaction.
 (b) The final electron acceptor in aerobic respiration is oxygen.
- 7. The equation does not show the formation of energy. It also does not
- include the fact that water is required on the reactant side of the equation.

- Explain why CO₂ does not serve as a source of free energy in living systems.
- (a) Explain the role of FADH₂ in the electron transport chain.
 - (b) Explain why FADH₂ does not generate as many ATP molecules as NADH does.
- Aerobic cellular respiration stops if no oxygen is present. Explain why.
- 8. CO2 cannot serve as a source of free energy because the carbon atoms are fully oxidized; there are no H atoms bonded to any of the C valence electron positions. Thus, its chemical potential energy is 0 kl/mole. The bond in ATP is easy to break and releases abundant energy.
- 9. (a) FADH2 is an electron carrier. In the electron transport chain, FADH2 donates electrons.
- (b) FADH2 does not generate as many ATP molecules as NADH because it has a lower energy content and it enters the electron transport chain later in the process.
- Oxygen is necessary for aerobic cellular respiration because it is the final acceptor of electrons in the electron transport chain.

ANAEROBIC RESPIRATION

Anaerobic Respiration

- Refers to respiration without oxygen
- Without oxygen NADH and FADH₂ cannot get rid of their electrons
- This means there is no NAD⁺ for glycolysis or Kreb's cycle

When oxygen levels decrease...

- NADH and FADH₂ give their electrons to another acceptor instead of oxygen
- This allows NAD⁺ and FAD to be available for glycolysis (which produces a small amount of ATP





- Value of exercise intensity at which blood lactic acid concentration begins to increase sharply
 - OExercising beyond threshold may limit duration of exercise

Due to pain, muscle stiffness, and fatigue

 Athletic training improves blood circulation and efficiency of O₂ delivery to body cells

OResult:

- Decrease in lactic acid production
- ▼Increase in lactic acid threshold

Supplements and toxins

• Creatine phosphate

- May serve as an E source by donating its phosphate to ADP
- Occurs naturally in body and many foods
- Athletes consume compound to produce more ATP in muscles
- Compound may also buffer muscle cells and delay onset of lactic acid fermentation
- Potential harmful side effects are possible

Metabolic Poisons

- · Some poisons interfere with the electron transport chain
- Causes death quickly because electron flow stops, which stops all stages of cellular respiration
- Examples:
 - Cyanide
 - Hydrogen sulfide



Section 7.4 Review Questions - Key

- 1. List two differences between aerobic respiration and fermentation.
- 2. A student regularly runs 3 km each afternoon at a slow, leisurely pace. One day, she runs 1 km as fast as she can. Afterward, she is winded and feels pain in her chest and leg muscles. What is responsible for her symptoms?
- 1. Two differences in aerobic respiration and fermentation are as follows:
- Aerobic respiration yields 36 ATP molecules per glucose molecule, whereas fermentation yields 2 ATP molecules per glucose molecule.
- Aerobic respiration produces water and carbon dioxide, whereas fermentation produces ethanol or lactic acid.
- The student will feel soreness in her chest and legs due to the buildup of lactic acid in her muscle tissues.

What role does alcohol fermentation play in the food industry?

- Compare and contrast the use of anaerobic and aerobic microbes in waste treatment.
- Define maximum oxygen consumption, VO₂max.
- 3. Fermentation is used to produce alcoholic beverages, bread, and certain types of cheese.
- 4. Both types of respiration can be used in waste treatment. Aerobic respiration is quick but expensive. Anaerobic respiration takes longer but is inexpensive and capable of removing toxic properties from the waste that aerobic respiration cannot.
- 5. Maximum oxygen consumption, VO2 max, is a measure of a body's capacity to generate the energy required for physical activity. It is the maximum volume of oxygen that the cells of the body can remove from the bloodstream in 1 min per kilogram of body mass while the body experiences maximal exertion.

- (a) Determine the value of the lactic acid threshold from Figure 10.
 (b) What does this value mean?
- Lactic acid fermentation is used in the food industry. Use Internet and print sources to answer the following questions.
 (a) What foods depend on lactic acid fermentation?
- (b) What microbes are used in each food in (a).



- 6. (a) The lactate threshold is 3.0 mmol/L.
- (b) This value refers to a threshold. Below this level, an individual can sustain exercise for long periods; above the threshold, an individual cannot sustain exercise for long periods. Once the lactate threshold is passed in the body, the lactate concentration in the blood increases sharply, which causes pain, muscle stiffness, and fatigue.
- 9. (a) The production of wine, cheese, bread, yogurt, pickles, and some meats depends on lactic acid fermentation.
- (b) Different species of bacteria are used to produce yogurt, fungus produces cheeses, yeasts and moulds produce meats such as salami and chorizo, bacteria make pickles and wine, and bacteria and yeasts produce breads.











