

Assessing microvascular changes

- The most common MR techniques for assessing microvascular changes:
 - Dynamic contrast agent enhanced MRI (DCE-MRI)
 - Dynamic susceptibility change MRI (DSC-MRI)
- Both require rapid temporal sampling, with preferred sampling rates on the order of
 - 5 10 sec per image set for DCE-MRI
 - 1 2 sec per image set for DSC-MRI
- Both require the infusion of exogenous contrast agents.

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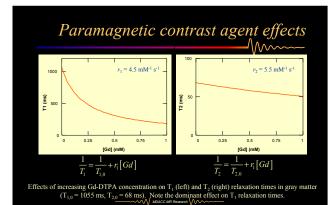
Common MRI contrast agents

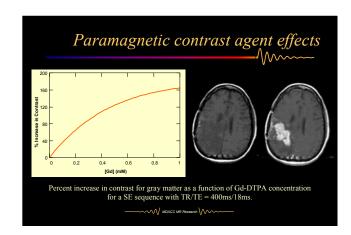
Paramagnetic Contrast Agents

- Gadolinium is the most common paramagnetic atom used in MR agents
- Gd is toxic must be tightly chelated
- Three common Gd agents:
 - Magnevist (gadopentetate dimeglumine)
 - Omniscan (gadodiamide)

non-ionic

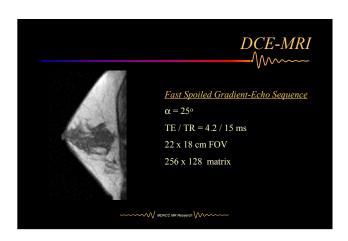
- Prohance (gadoteridol) non-ionic
 Osmotic loads of all three are significantly less than iodinated agents
- Affect both T₁ and T₂ relaxation times, with the dominant effect being shortening of the T₁ relaxation time (at routine clinical doses).

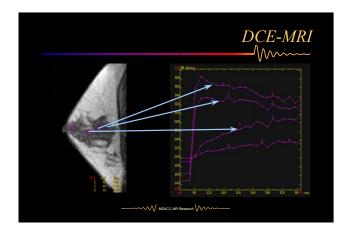


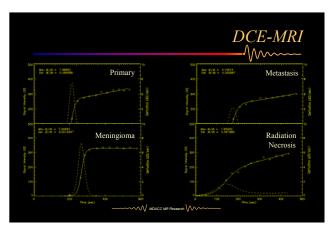


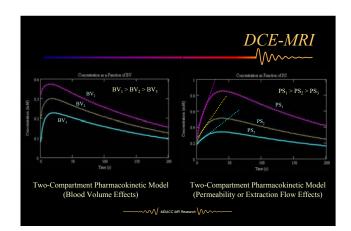
DCE-MRI

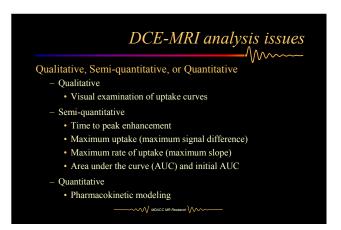
DC

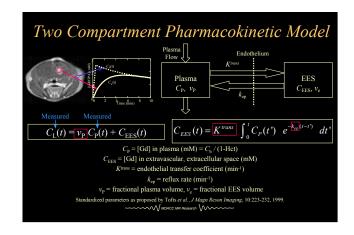


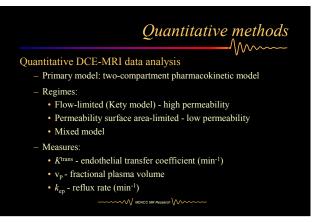


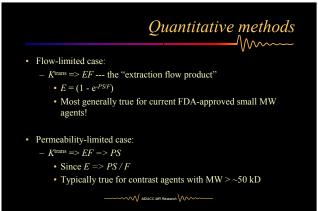


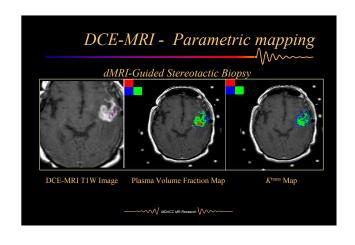


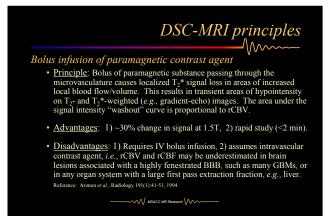


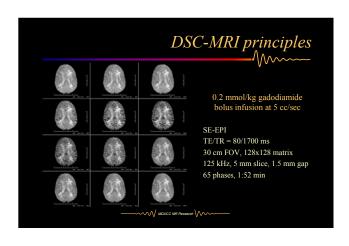


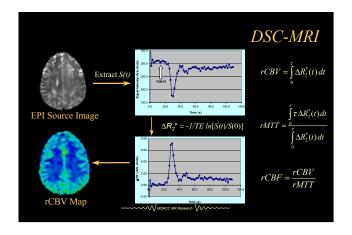


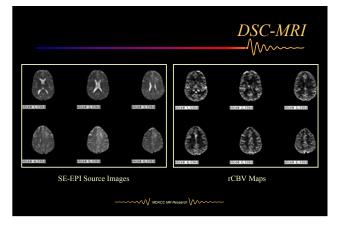


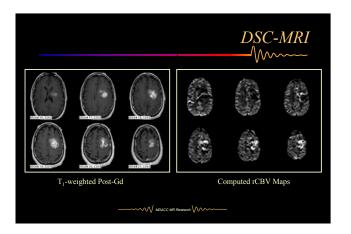


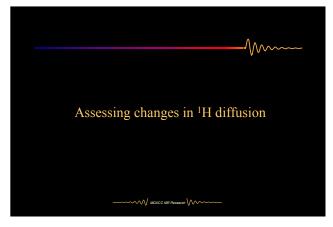










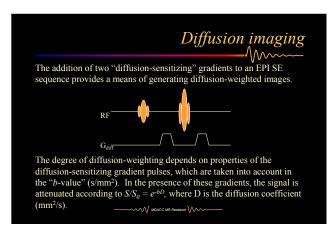


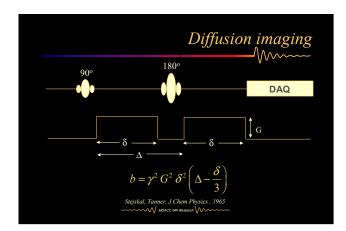
Diffusion imaging

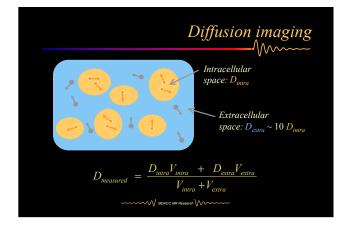
With the appropriate modification of a SE sequence, the image contrast can be made to depend on the rate of random, Brownian, motion of the water protons, *i.e.*, diffusion-weighted image contrast is obtained.

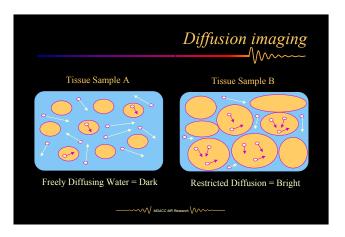
As the range of motion in diffusion processes is quite small compared to physiologic and gross patient motion, image acquisition rates need to be quite rapid. Therefore, successful diffusion imaging typically requires EPI class scanners capable of acquiring images in 50 - 100 ms per image.

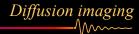
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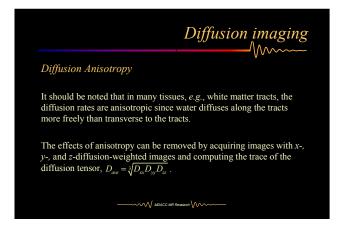


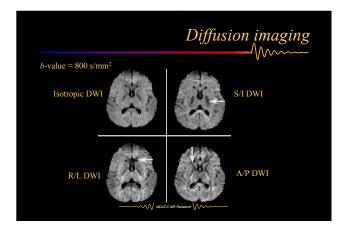


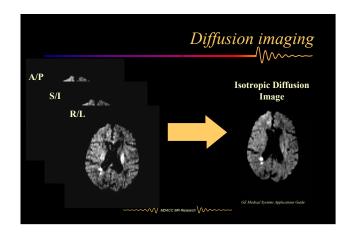


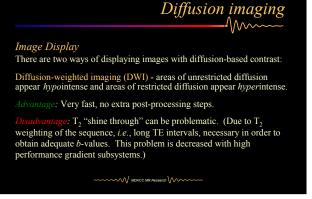
- Diffusion is described by a tensor.
- In materials with isotropic diffusion, the off-diagonal elements of the tensor are zero.
- In non-isotropic diffusing materials, the off-diagonal elements are non-zero, but (ideally) are symmetric.
- By applying the diffusion sensitizing gradient directions appropriately, the tensor elements can be completely defined.

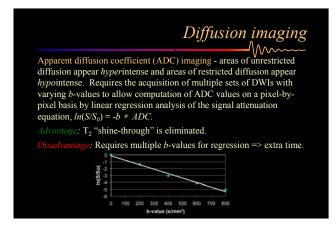
$$\overline{\overline{D}} = \begin{bmatrix} D_{xx} & D_{xy} & D_{xz} \\ D_{xy} & D_{yy} & D_{yz} \\ D_{xz} & D_{yz} & D_{zz} \end{bmatrix}$$

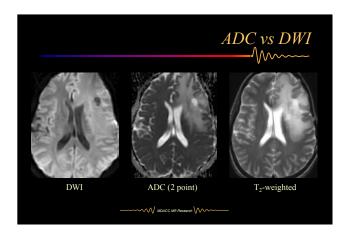


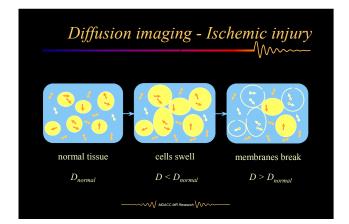


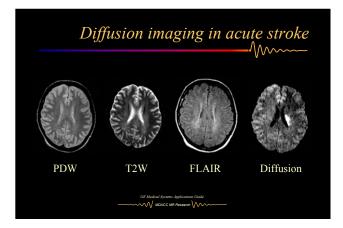


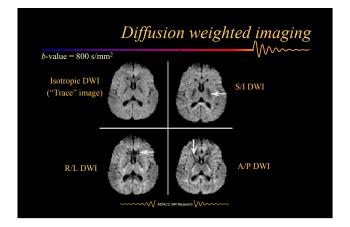


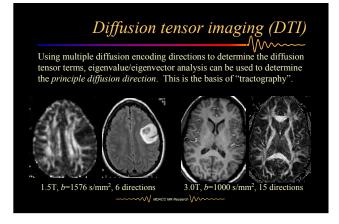


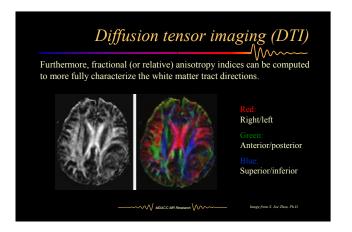


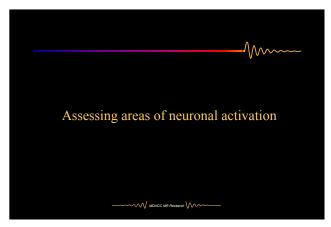




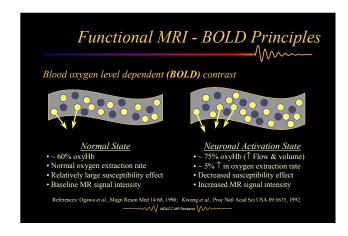


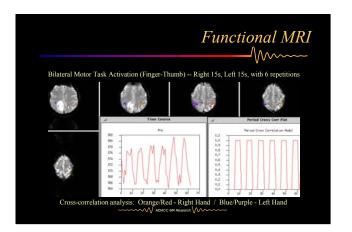


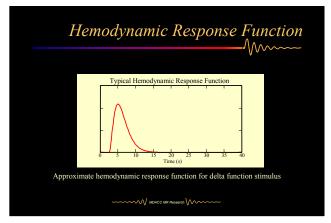


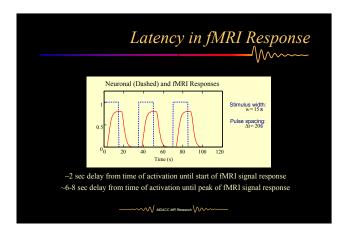


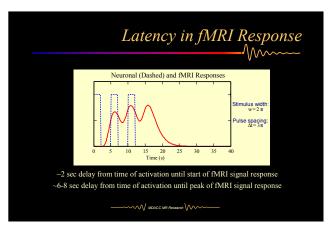
Blood oxygen level dependent (BOLD) contrast • Principle: Uses the difference in the magnetic state of oxyhemoglobin (diamagnetic) vs deoxyhemoglobin (paramagnetic) to provide image contrast. • Advantage: Totally noninvasive. Requires no infusion. • Disadvantage: Much smaller change in signal intensity compared to bolus injection technique (~1-5% changes at 1.5T). References: Ogawa et al., Magn Resson Med 14:68, 1990; Kwong et al., Proc Natl Acad Sci USA 89:5675, 1992.



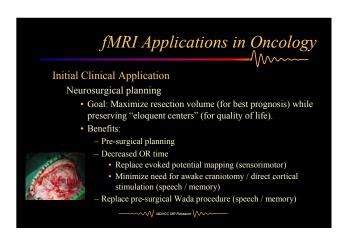


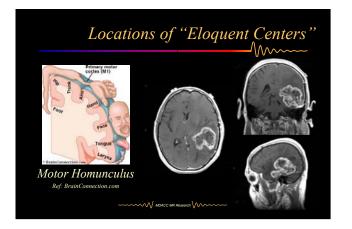


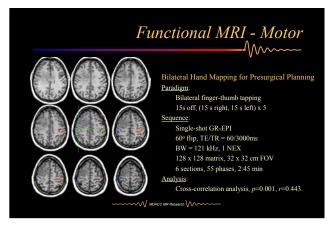


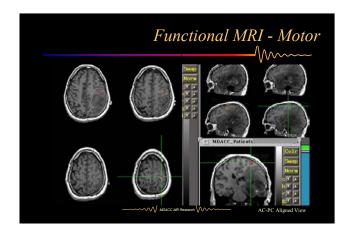


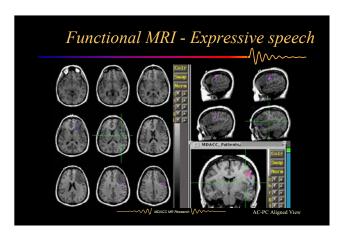


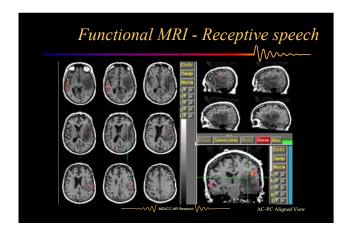


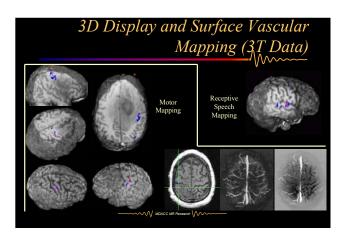


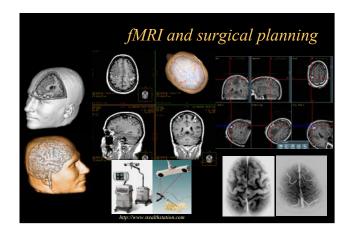


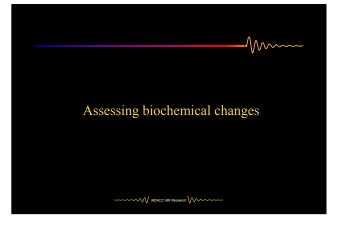












Introduction to spectroscopy

 Recall the Larmor relation that relates resonant frequency to magnetic field strength:

$$v = \gamma B_{nucleus}$$

where ν is the resonant frequency (MHz), γ is the gyromagnetic ratio ($\gamma'(2\pi)$ =42.57 MHz/T for protons), and $B_{nucleus}$ is the applied magnetic field strength (T) at a given nucleus.

 However, the value of B_{nucleus} depends on the local electronic environment, i.e., it is the value of the applied field, B_o, modified by the magnetic field due to the chemical environment.

Introduction to spectroscopy

 Therefore, nuclei in different chemical (electronic) environments will have slightly different resonance frequencies depending on the amount of local nuclear shielding, σ:

$$v = \gamma B_o (1 - \sigma)$$

 It is this local shielding effect that results in spectra with multiple peaks for a given nuclear species, where the peak positions depend on the local chemical environment.

Introduction to spectroscopy

 The position of a given spectral peak is usually given in terms of chemical shift with respect to some reference,

$$\delta_x = \left[\left(\nu_x - \nu_{ref} \right) / \nu_{ref} \right] x \ 10^6$$

- It is given in *parts per million* (ppm) to make the separations between the peaks independent of applied field strength.
- Note, however, that the separation of the peaks (in Hz) does depend on field strength. Therefore, the spectral resolution improves as field strength increases. (So does the SNR of the spectral peaks.)
- · For ¹H MRS applications, the reference is usually water.

H₂O H₂O H₂O H₃O Chemical Shift, δ (ppm)

Requirements

-∕\∧~~~

The success of an MRS examination depends upon the following:

- <u>High quality localization</u>

The volume from which the chemical information (spectrum) is obtained must be accurately known.

- Highly homogeneous magnetic field

Linewidths of peaks are inversely proportional to T₂*, so improved homogeneity results in narrower peaks (improved spectral resolution).

- Efficient water suppression (1H MRS)

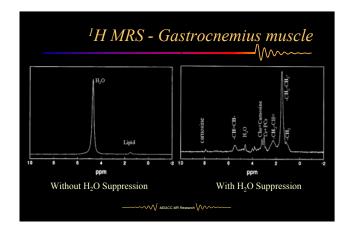
This is aided by improved homogeneity as well.

Spectral quantitation

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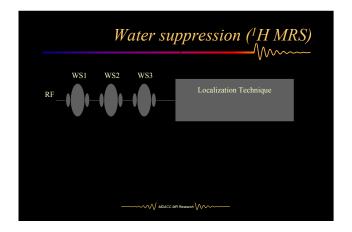
Water suppression (¹H MRS)

- In ¹H MRS studies, water is seldom the molecule of interest.
 While the fact that we're about 80% water is good news for MRI, it is very bad news for MRS.
- The metabolites of interest are usually about a factor of 8,000 less in concentration than water. Therefore, we must have a very efficient means of suppressing the water resonance in order to readily detect the metabolite resonances.



Water suppression (¹H MRS)

- The most commonly utilized method for water suppression is based on the same principle as "fat sat". (For MRS sequences, the suppression pulses are commonly referred to as CHESS pulses - chemically selective saturation.)
- Typically, multiple (often 3), narrow bandwidth (~50 Hz) pulses are applied at the water resonance frequency preceding the localization sequence.
- Multiple pulses are used to improve the degree of water suppression.

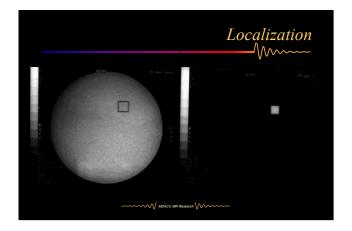


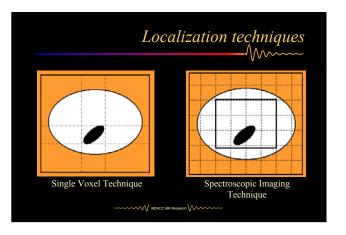
Localization

For a spectrum to have any significance, the region from which it is obtained must be accurately known. The most commonly used localization techniques are:

- Single voxel volume localization: The VOI is the intersection of three slice selective gradient/RF pulses. Each slice thickness can be individually varied to define VOI.
- Spectroscopic imaging: Uses phase-encoding for localization.
- Hybrids: Usually a combination of SVL and SI techniques.

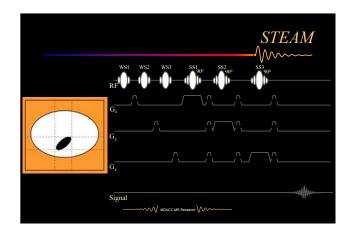
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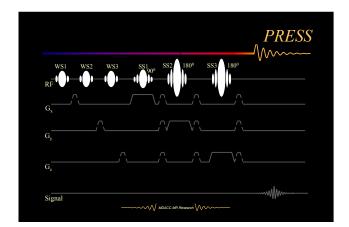


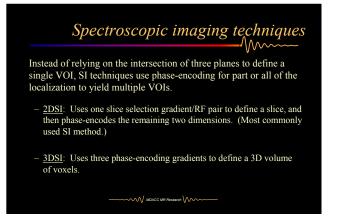


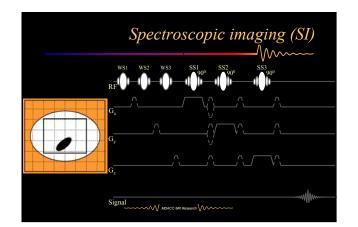
• The most common single volume localization techniques are those based on the *st*imulated *e*cho *a*cquisition *m*ode (STEAM) and *p*oint *res*olved spectroscopy (PRESS) sequences. - STEAM: 90°-90°-acquire - PRESS: 90°-180°-180°-acquire • Advantage of STEAM: shorter minimum echo times • Advantage of PRESS: 2x SNR incrase compared to STEAM

(for peaks with no *J*-coupling)





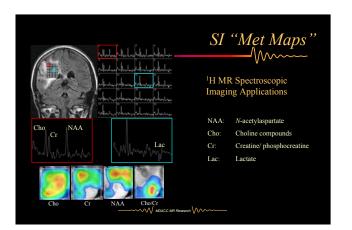




Spectroscopic imaging techniques

While SVL techniques are faster for obtaining a single localized spectrum, SI techniques have the following advantages:

- Spectra from multiple VOIs can be obtained for comparison.
 Useful for comparing suspected pathological tissue with normal-appearing contralateral region, or for better assessment of lesion heterogeneity.
- Spectra from smaller VOIs can be obtained as compared to SVL techniques.
 - Less partial volume averaging, better assessment of heterogeneity.
- "Metabolite maps", in which pixel intensity is proportional to chemical concentration, can be generated.



Spectroscopic imaging techniques

Disadvantages of SI techniques include:

- rather long acquisition times:

2DSI:
$$T_{scan} = N_{x_phase} \times N_{y_phase} \times N_{averages}$$

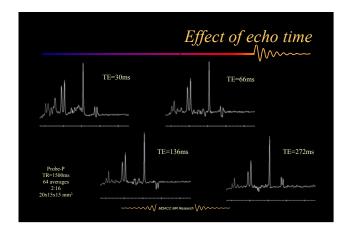
- spatially-dependent water suppression efficiency & spectral quality much larger volume over which field homogeneity must be optimized -- more difficult to accomplish than with SVL.
- "spectral-bleed" from one voxel to another is possible due to phase-encoding point spread function. (Can be minimized by increasing N_{phasess}, but this costs time.)

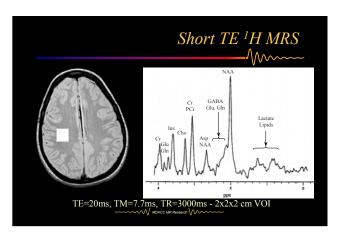
What can be seen? In ¹H MRS of the brain, the primary peaks are: - NAA: N-acetylaspartate (viable neurons only) – NAA: – Cr: 2.0 ppm total creatine (creatine + phosphocreatine) choline (phosphotidylcholine, etc.) 3.0 ppm 3.2 ppm 1.4 ppm - Cho: Lac: lactate (All of the above can be detected at short *and* relatively long TE acquisitions.) GABA: γ-aminobutyric acid Simple amino acids, e.g., alanine, glutamate, glutamine NAAG: N-acetylaspartylglutamate - Asp: aspartate (The above are in the 2.2-2.6 ppm range and typically require short TE acquisitions.) Lipids: range of chemical shifts, but dominant is methyl at 1.3 ppm myo-inositol 3.6 ppm - Glucose (at ~ 3.5 ppm) (The above require relatively short TE acquisitions.)

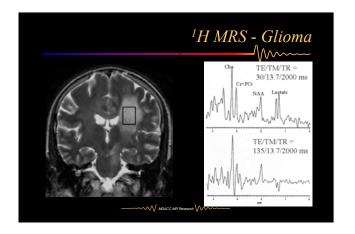
Choice of echo times

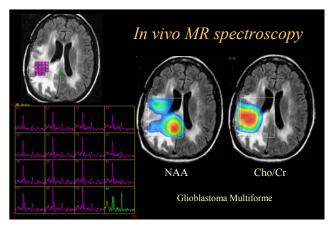
- Note that the choice of echo times in MRS exams is critical.
 As you increase TE, of course, the signal from all metabolites decreases due to spin dephasing.
- Short TE: more spectral peaks means improved chances for lesion characterization or evaluation of therapy. However, the examinations are more difficult to obtain reproducibly, mainly due to decreased water suppression efficiency.

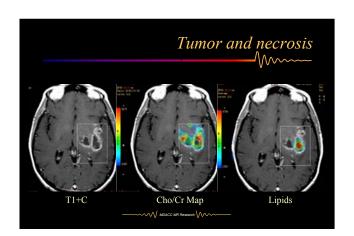
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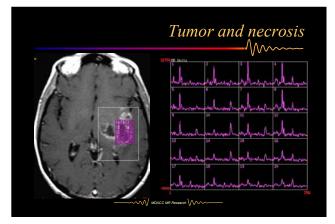












Spectral quantitation

Quantitative analysis comes in two forms:

- Relative concentrations
 - Most commonly involves taking the ratio of peak areas, *e.g.*, NAA/Cr and Cho/Cr in brain.
 - Problem: Changes in ratios can be due to changes in, for example, NAA *or* Cr.
- Absolute concentrations
 - · Much more difficult and requires some form of "standard".
 - External standard: small container of known concentration of reference sample from which reference spectrum is obtained.
 - Internal standard: most commonly taken as water.

Very important

- When comparing MRS data, be sure you take the TE and TR times into account.
- Each metabolite has its own T₁ and T₂ relaxation times.
 Therefore, as you change TE and/or TR, the relative peak areas and heights change.
- If you must compare data acquired at different TE and TR values, you can approximately normalize the data using the equations for T₂-decay and T₁-recovery appropriate for the particular localization sequence, and reported *in vivo* T₁ and T₂ relaxation times for each metabolite of interest.



