

Coulometric Karl Fischer titration: diaphragm-free cell, cell design and applications

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Outline

- Coulometric Karl Fischer titration
- Diaphragm-free cell
- Applications, Instrumentation, Examples

Volumetric and coulometric Karl Fischer titration for water determination

The Karl Fischer reaction: Iodine reacts stoichiometrically with water



Volumetric Karl Fischer Titration:

Iodine is added with burette during titration.

Water as a major component: 100 ppm - 100 %

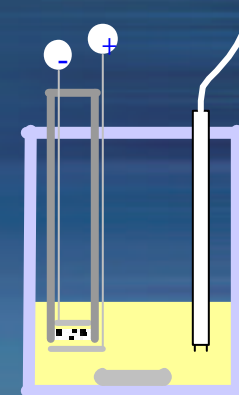


Coulometric Karl Fischer Titration:

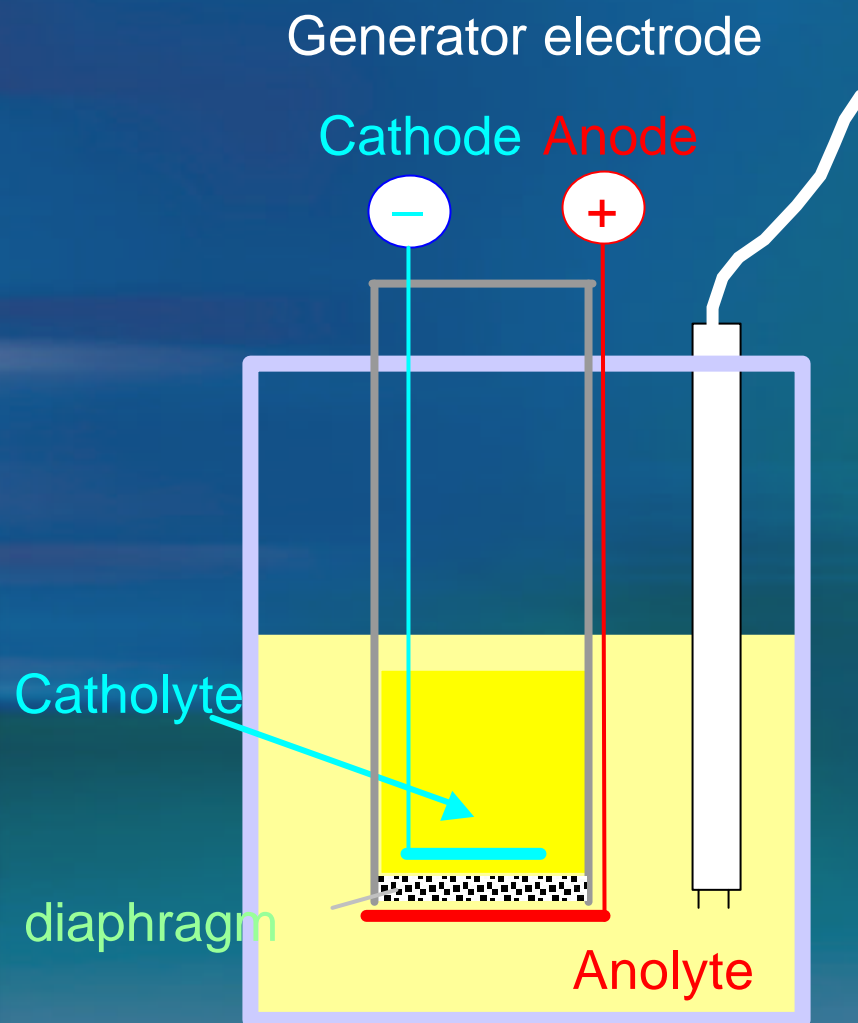
Iodine is generated electrochemically during titration



Water in trace amounts: 1 ppm - 5 %



Coulometric Karl Fischer titration cell



Indicator electrode:
Double platinum pin electrode

Anolyte:

Sulfur dioxide, Imidazole, Iodide,
various solvents for various
applications:

Methanol or Ethanol with
Chloroform, Octanol, Ethylene
glycol

Catholyte:

Same or modified solution

Control of the Karl Fischer titration: indication

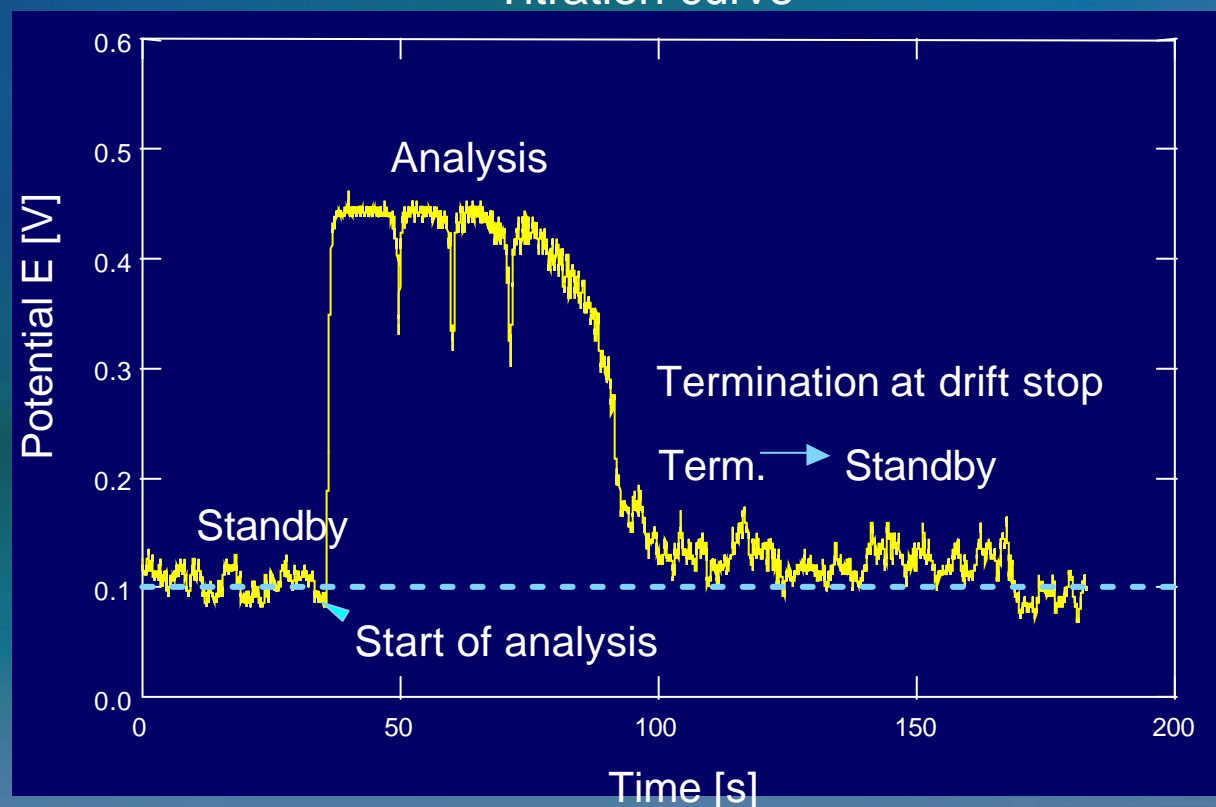
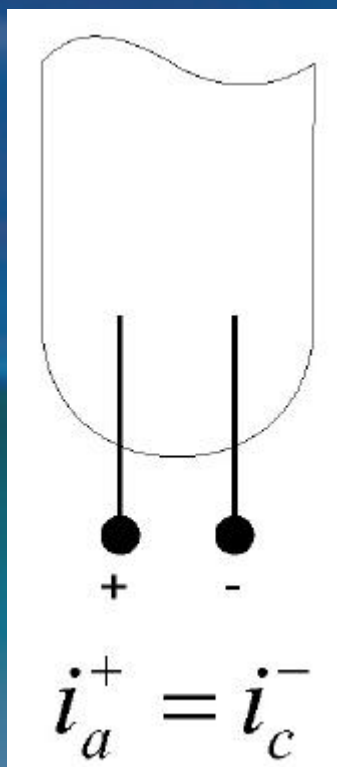
METTLER TOLEDO DL32/DL39:

DC polarization at double platinum pin electrode (two-electrode potentiometry)

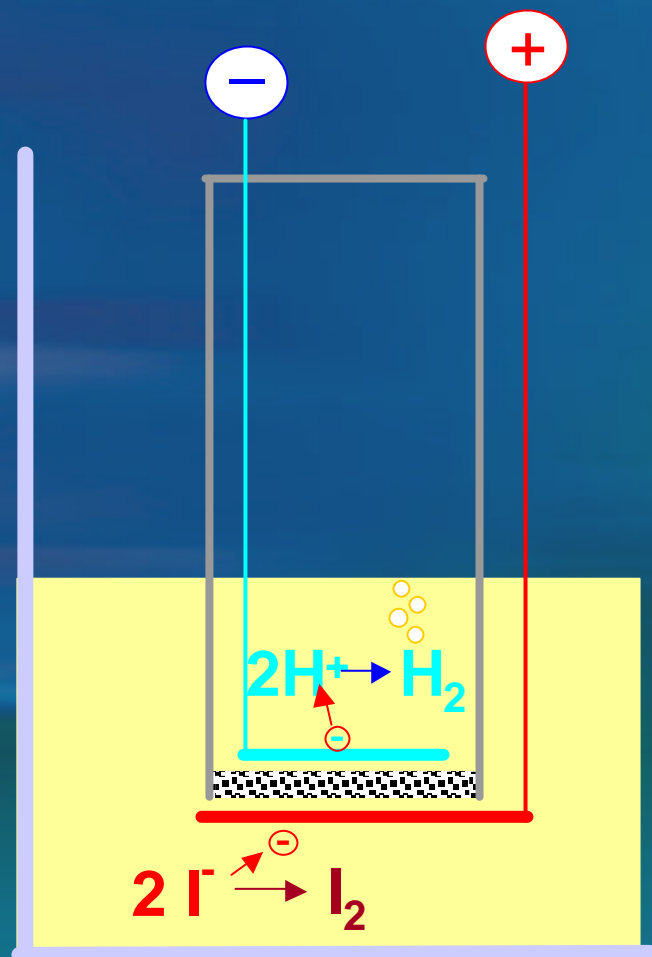
Polarization current: 1, 2 or 5 μA

Sample: 0.1 mL methanol, 1500 ppm of water

Titration curve



Coulometric Karl Fischer titration: iodine generation



Anodic reaction:

Iodine production by oxidation of iodide



Cathodic reaction:

Hydrogen production through reduction of H^+ ions



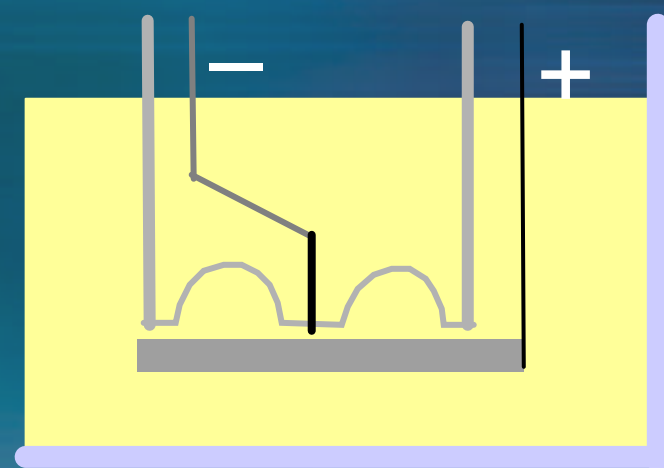
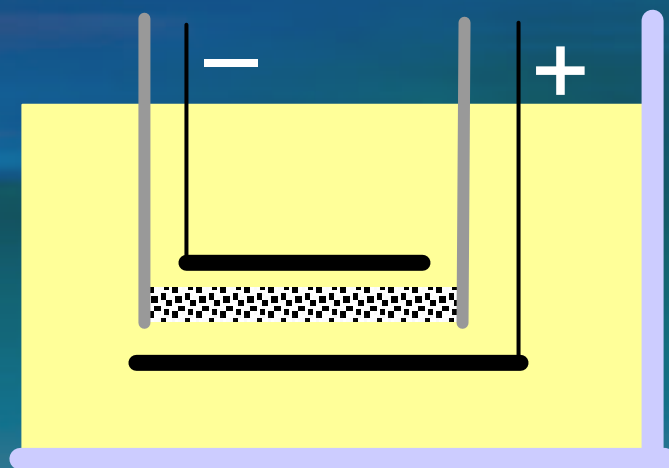
Coulometric Karl Fischer titration: cell with diaphragm and diaphragm-free cell



Current
generator
„electrode“ for
cell with
diaphragm

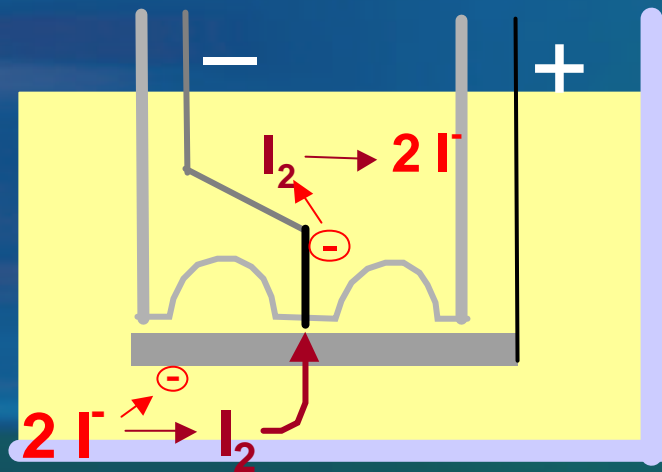


Current
generator
„electrode“
“ for
diaphragm-free cell



Coulometric Karl Fischer titration: cell without diaphragm

Efficiency loss at non-optimized diaphragm-free coulometric KF titration



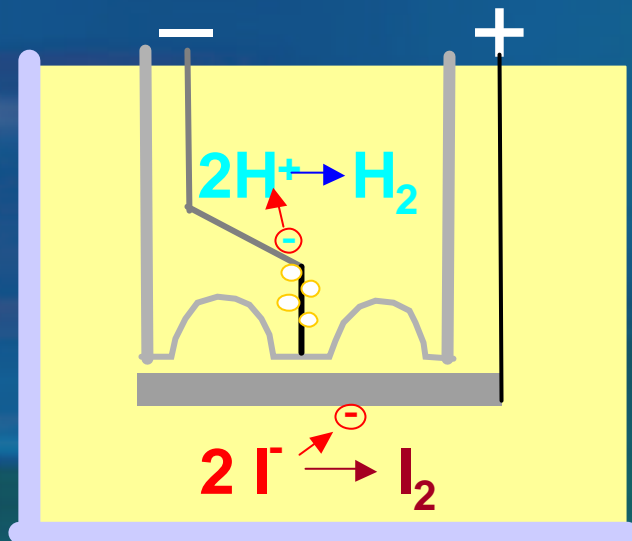
Problem:

Efficiency loss through
chemical or electrochemical
iodine reduction with
oxidizable reduction products
or at the counter-cathode
→ too high water recovery

→ Engineering of an optimized diaphragm-free current generator is required

Coulometric Karl Fischer titration: cell without diaphragm

Optimization of the diaphragm-free coulometric KF titration:
Geometry optimization



- **Anode:**
homogeneous current distribution
- **Cathode:**
small cathode surface area →
protection through hydrogen bubbling
- Sufficient stirring
- Geometrical separation of the electrodes

Coulometric Karl Fischer titration: cell without diaphragm

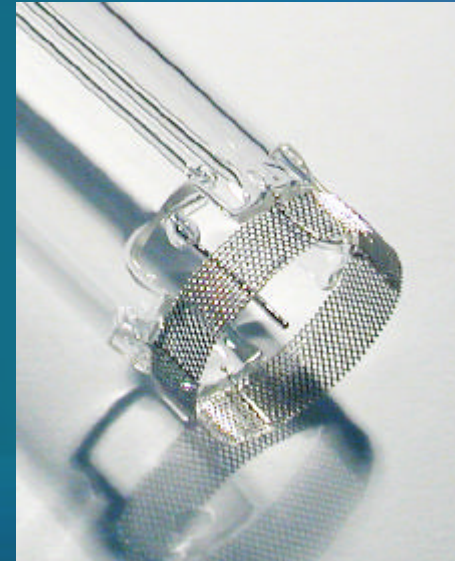
Optimization of the diaphragm-free coulometric KF titration:
Electrochemical current generation

- **Generated current pulses** of 400 mA, 200 mA, 100 mA, 60 mA are feasible. Large pulses are favourable, since the production of oxidizable reduction products is diminished.
- **Voltage** at the generator cell must be high enough for low-conducting electrolytes.
Example: 28 V available at DL32/DL39 for current generation, which is sufficient for low-conducting electrolytes to approx. 5 mS/cm.

Coulometric Karl Fischer titration: cell without diaphragm

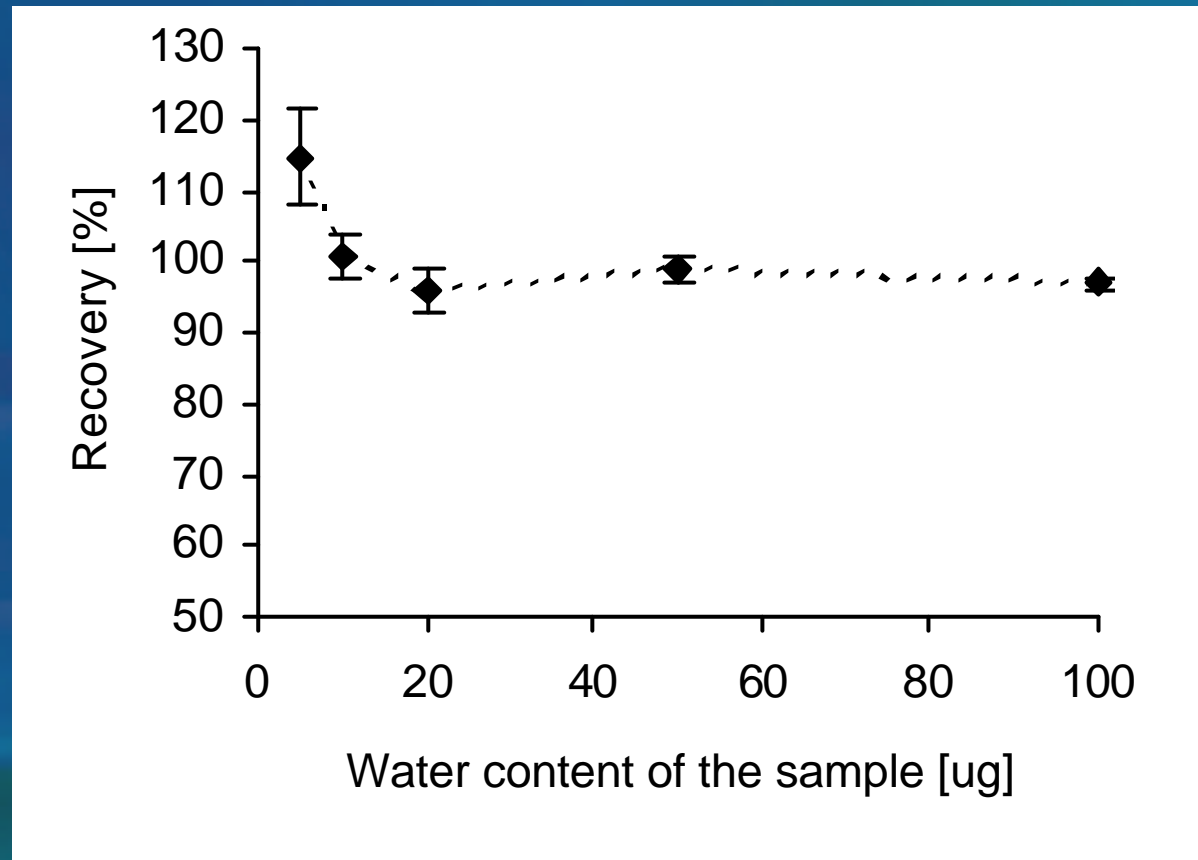
Advantages of diaphragm-free KF
coulometry: **Simple ease-of-use**

- Titration cell is easier to clean and refill
- No clogged-up diaphragm
- No contaminants from diaphragm
- Lower drift



- Optimized diaphragm-free coulometric KF titration allows precise trace water determinations **down to less than 10–20 ug of water.**

Coulometric Karl Fischer titration: cell without diaphragm



Recoveries with diaphragm-free KF titration with a METTLER TOLEDO DL39 coulometer for a HYDRANAL 100 ppm standard solution
(electrolyte: HYDRANAL Coulomat AG/CG)

Applications: Coulometry of soluble samples

- Samples that are **easily soluble**:
Solvents (alcohols, ethers, esters, hydrocarbons, halogenated hydrocarbons, nitro-compounds, etc.)
Use **standard electrolytes** for cell with/without diaphragm, e.g. Riedel-de Haën HYDRANAL, Merck APURA, and J.T. Baker HYDRA-POINT
- Samples that **do not easily dissolve**:
Edible oils, etheric oils, ointments, etc.
Add **hexanol**, **octanol** or **decanol** to the electrolyte, or use special electrolytes, e.g. HYDRANAL Coulomat AG-H
- Mineral oils, transformer oils, silicon oils, etc.
→ **Add chloroform to the electrolyte**, or use special electrolytes, e.g. HYDRANAL Coulomat Oil

Applications: Coulometry of soluble samples

Ketones and aldehydes:

they react with methanol, producing H_2O as a by-product.

Special KF reagents for ketones are required, e.g. HYDRANAL Coulomat AK/(CG-K) [cell with/without diaphragm]

Note:

Short aldehydes (e.g. acetaldehyde) are oxidized at the anode, with H_2O as a by-product.

Long-chain and aromatic aldehydes are no problem.

Applications: Special samples

Special samples:

Hardly-soluble or **non-soluble** samples

Samples that undergo **side-reactions** with the Karl Fischer electrolyte

Non-soluble samples that release water only at **elevated temperatures**

Samples with **high viscosity**

→ **Extraction of water by external extraction**

→ **Extraction of water with an oven**, with transfer of the carrier gas to the Karl Fischer cell

DO307 oven

Stromboli sample changer oven

Instrumentation: DO307 and Stromboli Sample changer oven

DL39 - DO307



DL39 - Stromboli



Instrumentation: LabX titration software

LabX - [Titrator 39]

Datei Bearbeiten Ansicht Extras Analyse Hilfsfunktionen Datensicherung Fenster ?

RUN [Icons]

Titratoren

- Titratör 39
- Methoden**
 - 112 Stromboli, 26.11.200
 - 311 **Standard 1.0, 17.12**
 - 312 Stromboli, 13.10.200
 - 320 Stromboli, 22.10.200
 - 612 Stromboli, 26.11.200
 - 613 Stromboli, 26.11.200
 - 614 Stromboli, 26.11.200
- Proben**
 - 112
 - 320
 - 612
 - 613
 - 614

Standby **Methode: 311** **Probe: 2**

Verbrauch / mC

Zeit / s

Probe 1

R1 =	717.4750	ug
R2 =	7174.75	ppm

Probe 2

R1 =	1760.1133	ug
R2 =	14667.61	ppm

Verbr. mC	Gehalt µg	Online-Drift µg/min	Signal mV	Zeit min:s
17803.45	1660.8	1872	456.0	1:33
18186.69	1696.5	1560	397.0	1:35
18247.26	1702.2	847	431.0	1:37
18351.78	1711.9	223	423.0	1:39
18403.23	1716.7	213	405.0	1:41
18454.69	1721.5	154	394.0	1:43
18506.15	1726.3	144	384.0	1:45
18557.60	1731.1	144	354.0	1:47
18609.06	1735.9	144	354.0	1:49
18651.94	1739.9	134	322.0	1:51
18694.82	1743.9	124	284.0	1:53
18737.81	1747.9	120	233.0	1:55
18780.90	1751.9	120	226.0	1:57
18811.13	1754.8	91	183.0	1:59
18818.85	1755.5	64	156.0	2:01
18826.57	1756.2	25	156.0	2:03
18836.43	1757.1	22	157.0	2:05
18844.15	1757.8	22	152.0	2:07
18854.01	1758.8	26	173.0	2:09
18860.66	1759.4	24	170.0	2:11
18870.52	1760.3	26	152.0	2:13
18876.09	1760.8	21	147.0	2:15
18881.67	1761.3	15	144.0	2:17
18884.88	1761.7	11	148.0	2:19
18890.46	1762.2	13	153.0	2:21
18893.67	1762.5	10	135.0	2:23
18894.96	1762.6	8	129.0	2:25

Analyse Protokolle Konfiguration

Titrator 39 311 17.12.2...

Hilfe erhalten Sie durch Drücken von F1

Angemeldeter Benutzer ist Development Administrator mit Profil Default.

NUM

Foodstuffs: Sample addition

Solid samples: external extraction/dissolution, drying oven

Liquid samples: direct injection

Sample	Method	KF-Reagents	Water content
Sucrose Surface water	External extraction in chloroform (15 min at 25°C)	Coulomat AG/ Coulomat CG	71.5 ppm
Sucrose Total water	External dissolution in formamide (15 min at 50°C)	Coulomat AG/ Coulomat CG	533.7 ppm
Olive oil	Direct injection with syringe	Coulomat AG with 40 vol.% decanol/ Coulomat CG	836 ppm
Rape oil	Direct injection with syringe	Coulomat AG with 40 vol.% decanol/ Coulomat CG	424 ppm
Cooking salt	Heating with DO307 oven, 300°C	Coulomat AG Oven/ Coulomat CG	359.6 ppm
Cinnamon powder	Heating with Stromboli oven, 180°C	Coulomat AG Oven/ Coulomat CG	4.8%
Garlic powder	Heating with Stromboli oven, 180°C	Coulomat AG Oven/ Coulomat CG	10.1%

Foodstuffs: Edible oils

Sample	Sample amount	Method	KF reagents	Water content, cell with diaphragm	Water content, cell without diaphragm
Olive oil	0.1–0.3 g	Direct injection with syringe	With diaphragm: Hydranal Coulomat AG-H/CG Without diaphragm: Hydranal Coulomat AD + 20% Octanol	507 ± 7 ppm	504 ± 13 ppm
Corn seed oil	0.1–0.3 g	Direct injection with syringe	With diaphragm: Hydranal Coulomat AG-H/CG Without diaphragm: Hydranal Coulomat AD + 20% Octanol	378 ± 4 ppm	376 ± 6 ppm
Sunflower oil	0.1–0.3 g	Direct injection with syringe	With diaphragm: Hydranal Coulomat AG-H/CG Without diaphragm: Hydranal Coulomat AD + 20% Octanol	698 ± 22 ppm	657 ± 2 ppm

Sunflower oil:
dissolution
few sample injections.

Difference in water content may be due to slower
leading to turbidity of electrolyte after a
Turbidity disappears after long stirring.

Foodstuffs: Spices, teas

Sample	Sample amount	Method	KF reagents	Water content, cell with diaphragm	Water content, cell without diaphragm
Marjoram spice	0.05 g	Heating with Stromboli oven, 180°C, 15 min.	Baker Hydra-Point Coulometric Oven	8.58 ± 0.17 %	8.9 ± 0.2 %
Marjoram spice	5 g	External extraction with methanol	Coulomat AG/CG Coulomat AG	8.33 ± 0.14 %	8.33 ± 0.15 %
Black tea	0.05 g	Heating with Stromboli oven, 180°C, 15 min.	Baker Hydra-Point Coulometric Oven	6.35 ± 0.18 %	6.3 ± 0.2 %
Black tea	2 g	External extraction with methanol	Coulomat AG/CG Coulomat AG	6.10 ± 0.16 %	6.10 ± 0.15 %

Difference in water recoveries may be due to **sample preparation procedure**. Optimisation of the sample preparation procedure is needed for each sample type.

Conclusions

Coulometric Karl Fischer titration is well suited for water determination in foods with low water content.

A diaphragm-free coulometric Karl Fischer titration cell is easy to use (easy cleaning and refilling, no clogging of the diaphragm), shows a lower drift and may replace a cell with diaphragm in most cases.

Diaphragm-free Karl Fischer coulometry has been optimized even for very low water content determinations and allows precise water determinations down to less than 10–20 μg of water.