

Distributed Bio-Oil Reforming



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Overview

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- Start date: 2005
- End date: 2012
- Percent complete: 60%

• FY 2005: \$100K

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- FY 2006: \$300K
- FY 2007: \$350K
- FY 2008: \$700K
- FY 2009: \$0K

Production Barriers

- A. Fuel processor capital
- C. Operation & maintenance
- D. Feedstock issues
- F. Control & safety

2012 Targets

- \$3.80/gallon gasoline equivalent
- 72% energy efficiency (bio-oil to H2)

- Colorado School of Mines -Oxidative cracking
- University of Minnesota -
- Catalyst Development
- Chevron Feedstock Effects (3 year CRADA)

Biomass fast pyrolysis produces high yields of liquid product, bio-oil, which can be stored and shipped to a site for renewable hydrogen production.

NREL is investigating the low-temperature, partial oxidation, and catalytic autothermal reforming of biooil for this application.



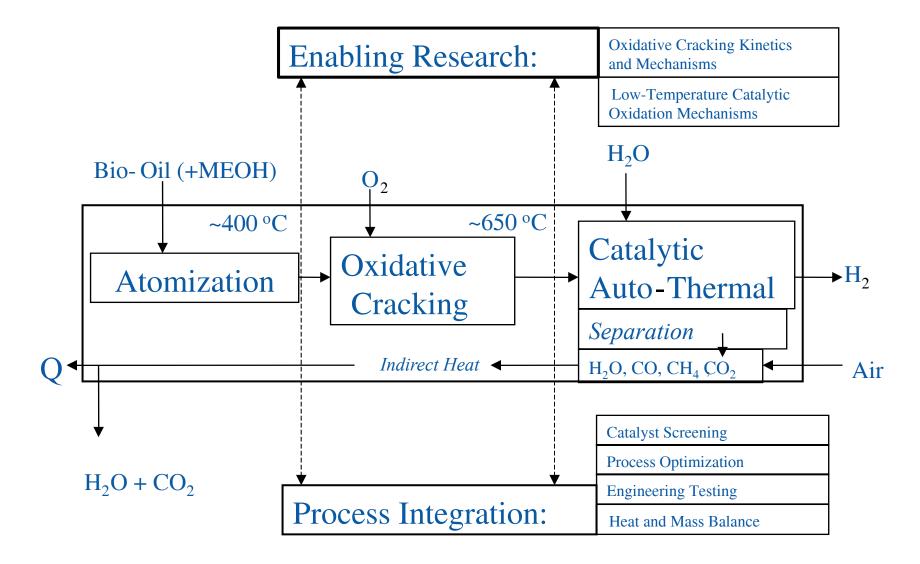
Pyrolysis: $CH_{1.46}O_{0.67} \rightarrow 0.71CH_{1.98}O_{0.76} + 0.21CH_{0.1}O_{0.15} + 0.08CH_{0.44}O_{1.23}$ BiomassBio-Oil (75%)Char (13%)Gas (12%)

Catalytic Steam Reforming of Bio-oil:Bio-oil (74 wt% $CH_{1.28}O_{0.41}$, 26 wt % H_2O)- 90 wt% of feed CH_3OH - 10 wt% of feed H_2O (2 mole ratio steam to carbon)

Overall Reaction: $CH_{2.18}O_{0.78} + 0.51O_2 + 0.19 H_2O \rightarrow CO_2 + 1.28 H_2$

Estimated Practical Yield: 9.3 wt % Energy Efficiency Estimates are in Progress based on Aspen Modeling

Distributed Bio-Oil Reforming Approach



Objectives

Overall

Develop the necessary understanding of the process chemistry, compositional effects, catalyst chemistry, deactivation, and regeneration strategy as a basis for process definition for automated distributed reforming; demonstrate the process

• FY 2009

Improve bio-oil atomization with less MeOH addition,

Demonstrate non-catalytic partial oxidation of bio-oil at bench scale

Demonstrate catalytic conversion of bio-oil to syngas at bench scale

Provide mass balance data for H2A

Technical Accomplishments

- FY 2006
 - Bio-oil volatilization method developed
 - Oxidative cracking to CO with minimal CO₂
- FY 2007
 - Demonstrated equilibrium catalytic conversion to syngas at low temperature and low H_2O/C
- FY2008
 - Demonstrated catalyst performance
 - Designed and built a bench scale reactor system
- FY2009
 - Demonstrated operation of a bench-scale reactor system using 90 wt% bio-oil/10 wt% methanol mixture

Progress in Process Development

1. Bio-Oil Volatilization

- The new ultrasonic nozzle can produce a fine mist even from high-viscosity liquids.
- Successful tests with 90 wt% bio-oil/10 wt% methanol mixture.
- Promising attempts of feeding neat bio-oil though 10 wt% methanol addition will likely be used to homogenize and stabilize bio-oil.
- Ultrasonic nozzle will likely be replaced by a high-pressure injector in larger-scale units.

Progress in Process Development 2. Oxidative Cracking

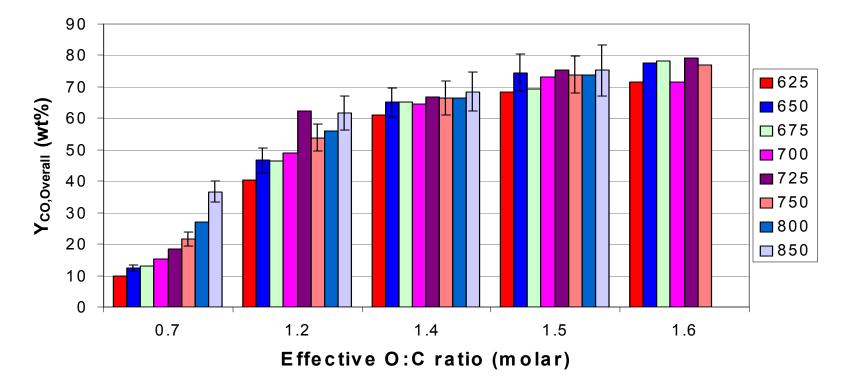
40 35 625 30 YH2, Overall (wt%) 650 25 675 **700** 20 **725** 15 **750** 800 10 850 5 0 0.7 1.2 1.4 1.5 1.6 Effective O:C ratio (molar)

Overall H₂ Yield

Hydrogen yields are 25-30% of the stoichiometric potential; Those yields will significantly increase after completion of water-gas shift.

Progress in Process Development 2. Oxidative Cracking

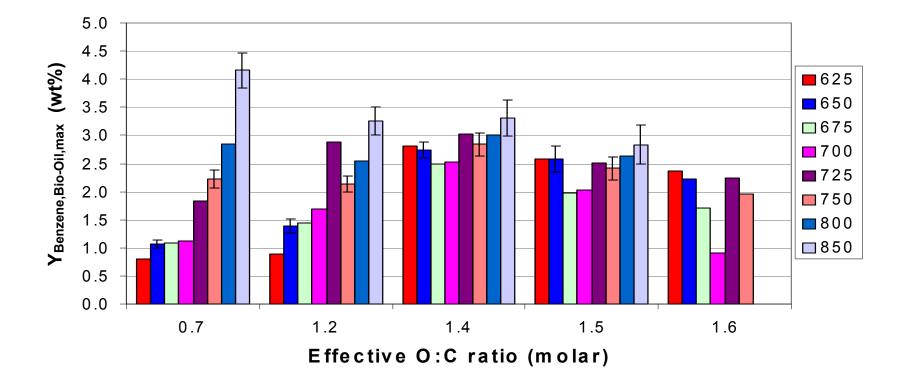
Overall CO Yield



>70% carbon can be converted to CO by non-catalytic POX; O:C ratio has much stronger effect than temperature on CO yields

Progress in Process Development 2. Oxidative Cracking

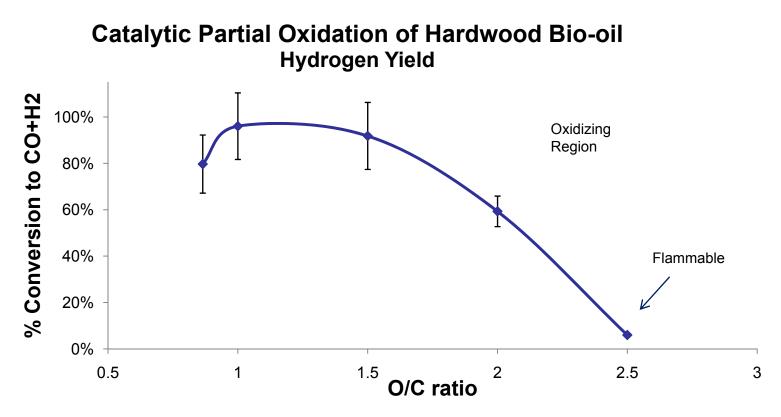




At O:C>1.4 temperature has little effect on benzene (aromatics) formation

Progress in Process Development 3. Catalytic Conversion

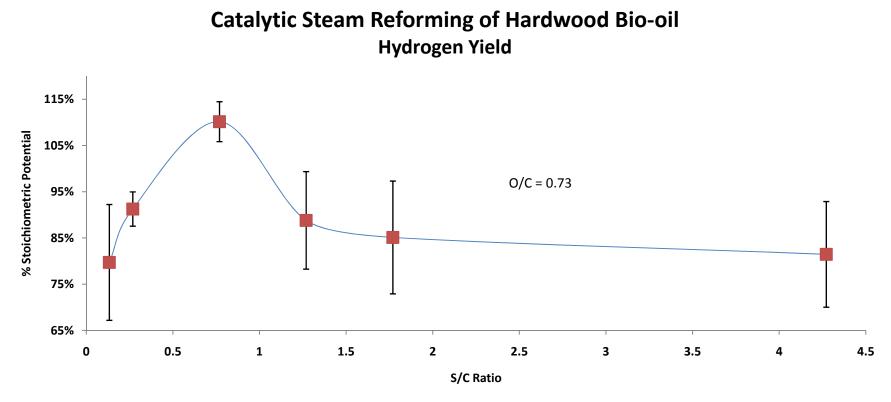
University of Minnesota catalyst (Lanny Schmidt's group): 1 wt % Rh and 1 wt % ceria on alumina



1 - 1.5 is the optimal range of O/C for catalytic POX of bio-oil

Progress in Process Development 3. Catalytic Conversion

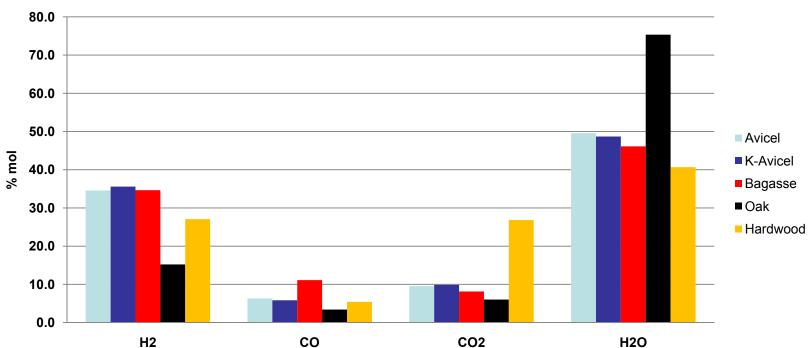
University of Minnesota catalyst (Lanny Schmidt's group): 1 wt % Rh and 1 wt % ceria on alumina



80-90% of stoichiometric yield of hydrogen was produced by catalytic steam reforming of bio-oil for the whole S/C range

Progress in Process Development 3. Catalytic Conversion

Product Gas Composition from Different Feedstocks



Syn-gas Composition 650 C; O/C=1.3; S/C=2.5

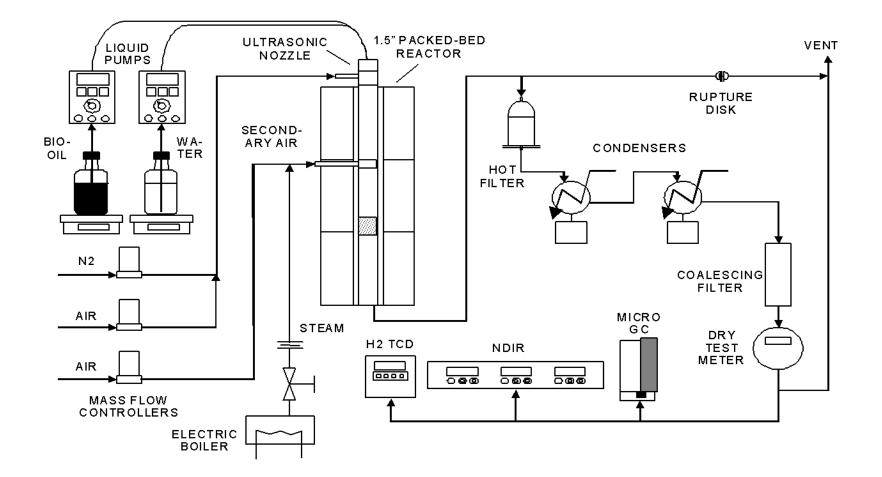
Progress in Process Development 4. Bench-scale Reactor



- Built out of quartz:
 d = 30 mm; h = 450 mm
- Connected to the feeding and condensation systems
- Feed rate 1-2 g/min 90 wt% bio-oil/10 wt% methanol solution
- On-line product gas composition monitoring
- 1-4 hour runs
- Detailed product analysis
- Improved mass balance

Progress in Process Development

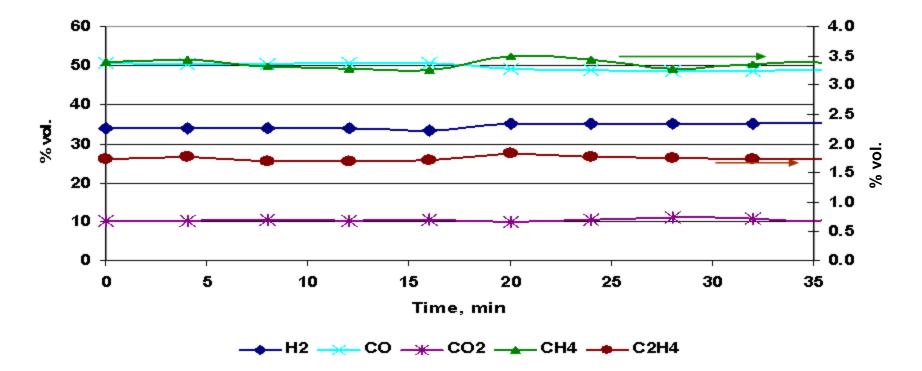
4. Bench Scale Reactor System



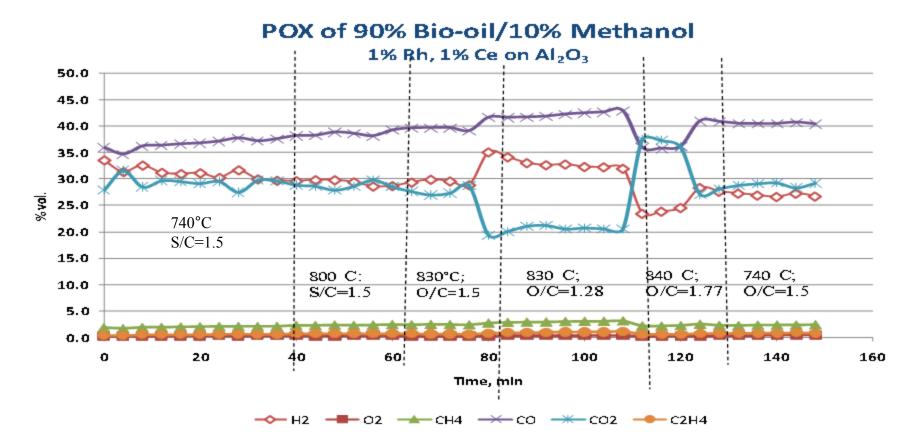
Progress in Process Development 4. Bench Scale Reactor System

Gas Composition from POX of Bio-oil/Methanol

POX of Bio-oil/Methanol Mixture t = 730C, O:C = 1.5



Gas Composition from Catalytic POX of Bio-oil/Methanol at different process conditions



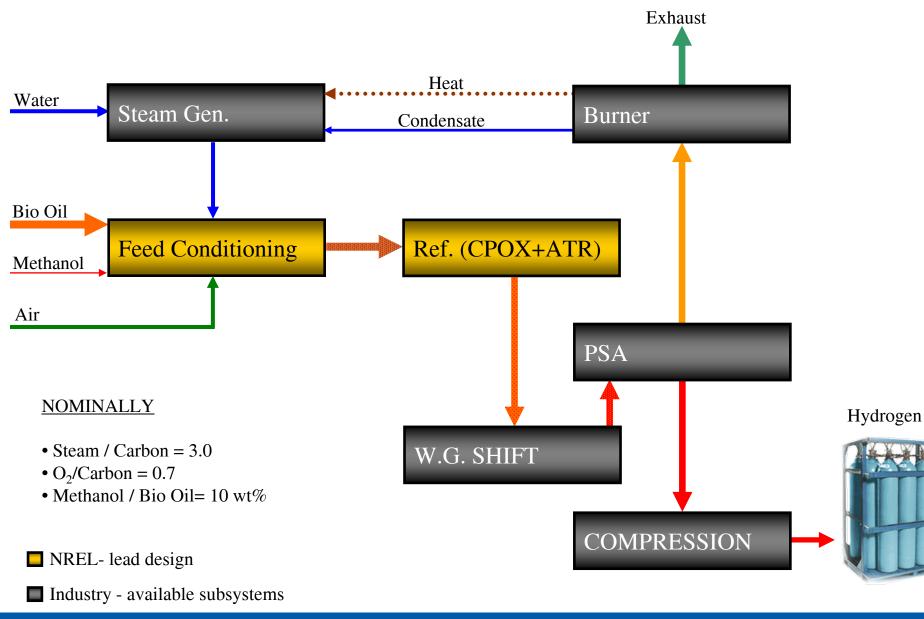
Hydrogen Cost

1500 kg/day station used for H2A analysis.

Capital Costs \$1,660,000 (\$2.03/gge).

Total cost of delivered hydrogen **\$4.48/gge** \$2.59/gge for the production \$1.88/gge for compression, distribution, and dispensing.

Process Subsystems Outline



Summary

- Bench-scale reactor system tests of non-catalytic and catalytic partial oxidation of bio-oil were performed using 90 wt% biooil/10 wt% methanol mixtures
- Carbon-to-gas conversion at bench-scale was slightly less than that achieved in micro scale system
- Rhodium catalyst enhanced bio-oil to syngas conversion by POX with and without added steam
- Bio-oils from herbaceous feedstock were more difficult to process and left more deposits than wood bio-oils
- Experimental results from bench-scale system will be used as to validate ASPEN simulations based on micro-scale data.

Project Timeline

ID	Task Name	2005	2006	2007	2008	2009	2010	2011	2012
1	Bio-Oil Volatilization								
2	Processing Options			արի հեր					
3	Modification and Characterization				-				
4	Injector Development								
5	Coking Studies								
6	Go / No Go on Bio-Oil performance			5/3	1				
7	Oxidative Cracking					~			
8	Proof of Concept								
9	Reduce Catalyst Loading by 50%		6/3	0					
10	Partial Oxidation Database								
11	Modeling and Optimization								
12	Jon Marda Thesis					12/31			
13	Catalytic Auto-Thermal Reforming								
14	Catalyst Screening								
15	Catalyst Process optimization								
16	Demonstrate catalyst performance consistent with \$3.80/gge				5/30)			
17	Catalyst Mechanistic Studies								
18	Integrated Separation								
19	Concept Evaluation				ц.				
20	Materials Evaluation								
21	Integrated Laboratory System Experiment								
22	Go / No Go on Conceptual Design						۴ 🗹	/31	
23	Systems Engineering								\sim
24	Oxygen, Steam and Heat Integration								
25	Engineering Design and Construction								
26	Prototype System Developed							5/31	
27	Heat and Mass Balances								
28	Process Upsets								
29	Long Duration Runs								
30	Demonstrate Distributed Hydrogen Production from								9
	Bio-Oil for \$3.8/gge								
31	Safety Analysis								
32	Review and Analysis of Pressure, O2, H2								
33	Systems Integration							1	

Future Work

- FY 2009: Produce process performance data as a function of process conditions (temperature, catalyst, O/C, S/C) using the bench-scale system
 - Integrated laboratory experiment
 - Optimization work
 - Long-term catalyst performance test
 - Assess the impact of the bench-scale results on the process design and on hydrogen production cost
- FY 2010: "Go/no-go" on conceptual design
- FY 2011: Prototype system
- FY 2012: Long duration runs to validate the process