Drug Metabolism

- Most metabolic products are <u>less pharmacologically active</u>
 - Important exceptions:
 - Where the metabolite is more active (Prodrugs, e.g. Erythromycin-succinate (less irritation of GI) --> Erythromycin)
 - Where the metabolite is toxic (acetaminophen)
 - · Where the metabolite is carcinogenic
 - <u>Close relationship</u> between the <u>biotransformation of drugs</u> and <u>normal biochemical</u> <u>processes</u> occurring in the body:
 - Metabolism of drugs involves many pathways associated with the synthesis of endogenous substrates such as steroid hormones, cholesterol and bile acids
 - Many of the enzymes involved in drug metabolism are principally designed for the metabolism of endogenous compounds
 - These enzymes metabolize drugs only because the <u>drugs resemble the natural compound</u>

Phases of Drug Metabolism

Phase I Reactions

- Convert parent compound into a more polar (=hydrophilic) metabolite by <u>adding or</u> <u>unmasking functional groups</u> (-OH, -SH, -NH₂, -COOH, etc.)
- Often these metabolites are inactive
- May be sufficiently polar to be excreted readily

Phase II Reactions

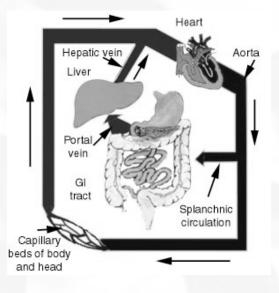
- <u>Conjugation with endogenous substrate</u> to further increase aqueous solubility
- <u>Conjugation with glucoronide</u>, sulfate, acetate, amino acid
- Phase I usually precede phase II reactions

Liver is principal site of drug metabolism:

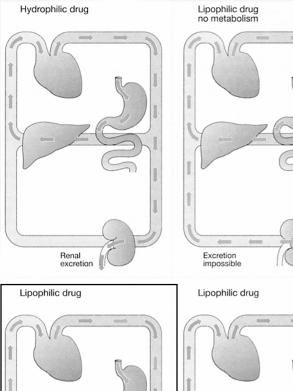
- Other sites include the gut, lungs, skin and kidneys
- For orally administered compounds, there is the

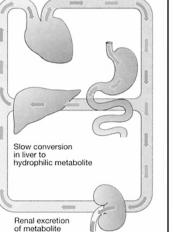
"First Pass Effect"

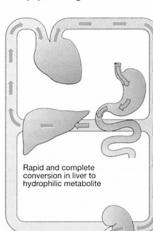
- Intestinal metabolism
- Liver metabolism
- Enterohepatic recycling
- Gut microorganisms glucuronidases



Drug Metabolism







Renal excretion

of metabolite

Drug Metabolism - Phase I

- Phase I Reactions
 - Oxidation
 - Reduction
 - Hydrolytic cleavage
 - Alkylation (Methylation)
 - Dealkylation
 - Ring cyclization
 - N-carboxylation
 - Dimerization
 - Transamidation
 - Isomerization
 - Decarboxylation

Two types of oxidation reactions:

- Oxygen is incorporated into the drug molecule (e.g. hydroxylation)
- Oxidation causes the loss of part of the drug molecule (e.g. oxidative deimination, dealkylation)

Microsomal Mixed Function Oxidases (MFOs)

- "<u>Microsomes</u>"
 - form in vitro after cell homogenization and fractionation of ER
 - <u>Rough microsomes</u> are primarily associated with <u>protein synthesis</u>
 - <u>Smooth microsomes</u> contain a class of <u>oxidative enzymes</u> called
- "Mixed Function Oxidases" or "Monooxygenases"
 - These enzymes require a reducing agent (NADPH) and molecular oxygen (one oxygen atom appearing in the product and the other in the form of water)

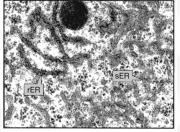
- MFO consists of two enzymes:
 - Flavoprotein, NADPH-cytochrome c reductase
 - One mole of this enzyme contains one mole each of flavin mononucleotide (FMN) and flavin adenine dinucleotide (FAD)
 - Enzyme is also called <u>NADPH-cytochrome P450 reductase</u>
 - <u>Cytochrome P450</u>
 - named based on its light absorption at 450 nm when complexed with carbon monoxide
 - is a hemoprotein containing an iron atom which can alternate between the ferrous (Fe⁺⁺) and ferric (Fe⁺⁺⁺) states
 - Electron acceptor
 - Serves as terminal oxidase
 - its relative abundance compared to NADPH-cytochrome P450 reductase makes it the <u>rate-limiting step</u> in the oxidation reactions

Humans have 18 families of cytochrome P450 genes and 43 subfamilies:

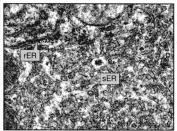
- CYP1 drug metabolism (3 subfamilies, 3 genes, 1 pseudogene)
- CYP2 drug and steroid metabolism (13 subfamilies, 16 genes, 16 pseudogenes)
- CYP3 drug metabolism (1 subfamily, 4 genes, 2 pseudogenes)
- CYP4 arachidonic acid or fatty acid metabolism (5 subfamilies, 11 genes, 10 pseudogenes)
- CYP5 Thromboxane A2 synthase (1 subfamily, 1 gene)
- CYP7A bile acid biosynthesis 7-alpha hydroxylase of steroid nucleus (1 subfamily member)
- CYP7B brain specific form of 7-alpha hydroxylase (1 subfamily member)
- CYP8A prostacyclin synthase (1 subfamily member)
- CYP8B bile acid biosynthesis (1 subfamily member)
- CYP11 steroid biosynthesis (2 subfamilies, 3 genes)
- CYP17 steroid biosynthesis (1 subfamily, 1 gene) 17-alpha hydroxylase
- CYP19 steroid biosynthesis (1 subfamily, 1 gene) aromatase forms estrogen
- CYP20 Unknown function (1 subfamily, 1 gene)
- CYP21 steroid biosynthesis (1 subfamily, 1 gene, 1 pseudogene)
- CYP24 vitamin D degradation (1 subfamily, 1 gene)
- CYP26A retinoic acid hydroxylase important in development (1 subfamily member)
- CYP26B probable retinoic acid hydroxylase (1 subfamily member)
- CYP26C probabyle retinoic acid hydroxylase (1 subfamily member)
- CYP27A bile acid biosynthesis (1 subfamily member)
- CYP27B Vitamin D3 1-alpha hydroxylase activates vitamin D3 (1 subfamily member)
- CYP27C Unknown function (1 subfamily member)
- CYP39 7 alpha hydroxylation of 24 hydroxy cholesterol (1 subfamily member)
- CYP46 cholesterol 24-hydroxylase (1 subfamily member)
- CYP51 cholesterol biosynthesis (1 subfamily, 1 gene, 3 pseudogenes) lanosterol 14-alpha demethylase

Induction of P450 enzymes:

- PPAR (peroxisome proliferator activated receptor) ligands (e.g.clofibrate)
- CYP1 family are induced by aromatic hydrocarbons (cigarette smoke; charred food)
- CYP2E enzymes induced by ethanol
- CYP2B enzymes induced 40-50 fold by barbiturates
- Polymorphisms cause differences in drug metabolism:



Normal hepatocyte



Hepatocyte after phenobarbital administration

<u>CYP2C19</u> has a polymorphism that changes the enzyme's ability to metabolize mephenytoin (a marker drug). In <u>Caucasians</u>, the polymorphism for the poor metabolizer phenotype is only seen in 3% of the population. However, it is seen in 20% of the <u>asian</u> population.
 <u>It is important to be aware of a person's race when drugs are given that are</u> metabolized differently by different populations

P450s and drug interactions:

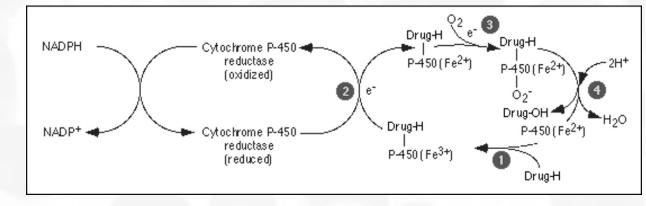
- <u>Barbiturates</u> induce CYP2B => increased metabolism of other drugs
- <u>Antifungals</u> (e.g. ketoconazole) inhibit fungal CYP51 and unintentionally also human CYP3A4
 => reduced metabolism of other drugs
- <u>Grapefruit juice</u> contains a CYP3A4 inhibitor =>12 fold increase in some drug concentrations <u>CYP3A4 Substrates</u>: • Acetominophen (Tylenol) • Codeine (narcotic) • Cyclosporin A (immunosuppressant),
 • Diazepam (Valium) • Erythromycin (Antibiotic) • Lidocaine (local anaesthetic), • Lovastatin (HMGCoA reductase inhibitor), • Taxol (cancer drug), • Warfarin (anticoagulant).

Drug oxidation requires:

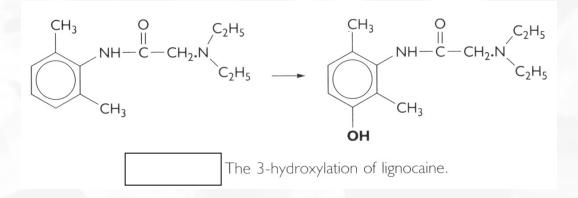
- Cytochrome P450
- Cytochrome P450 reductase
- NADPH
- Molecular oxygen

<u>The cycle involves four steps:</u>

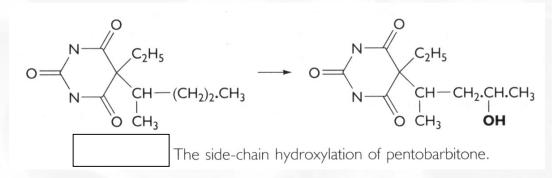
- 1. Oxidized (Fe3+) cytochrome P-450 combines with a drug substrate to form a binary complex.
- 2. NADPH donates an electron to the cytochrome P-450 reductase, which in turn reduces the oxidized cytochrome P-450-drug complex.
- 3. A second electron is introduced from NADPH *via* the same cytochrome P-450 reductase, which serves to reduce molecular oxygen and form an "activated oxygen"-cytochrome P-450-substrate complex.
- 4. This complex in turn transfers "activated" oxygen to the drug substrate to form the oxidized product. The potent oxidizing properties of this activated oxygen permit oxidation of a large number of substrates.



Aromatic hydroxylation:

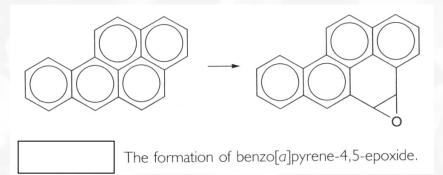


Aliphatic hydroxylation:

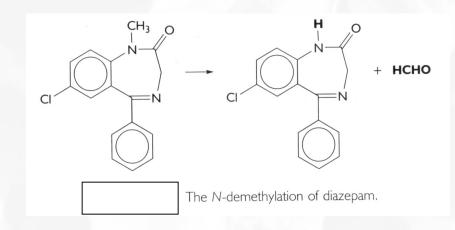


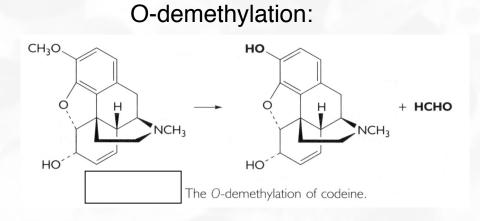
BIMM118

Epoxidation:

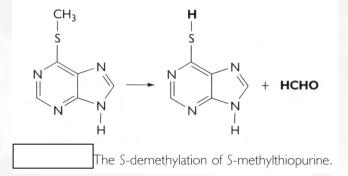


Dealkylation:

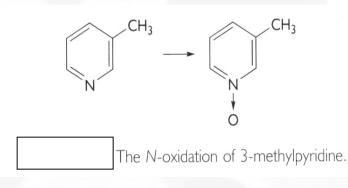




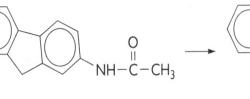
S-demethylation:

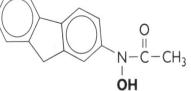


N-oxidation:



N-hydroxylation:





The N-hydroxylation of 2-acetylaminofluorene.

Oxidation reactions NOT catalyzed by Cytochrome P450:

Flavin containing monoxygenase system

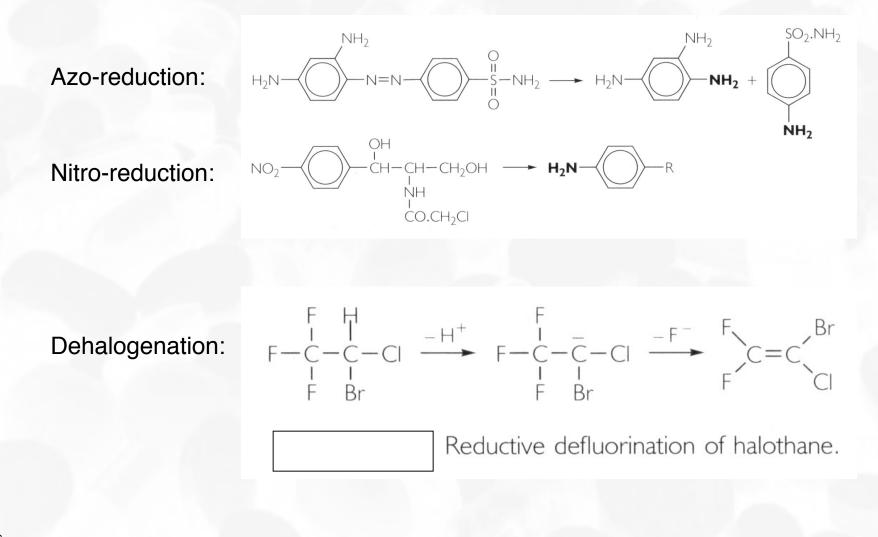
- Present mainly in liver but some is expressed in gut and lung
- Located in smooth endoplasmic reticulum
- Oxidizes compounds containing sulfur and nitrogen
- Uses NADH and NADPH as cofactors
- <u>Alcohol dehydrogenase</u> (cytosol)
- <u>Aldehyde oxidation</u> (cytosol)
- Xanthine oxidase
- <u>Amine oxidases</u>
 - Monoamine oxidase (nerve terminals, mitochondria)
 - Diamine oxidase found in liver microsomes
 - Primarily endogenous metabolism

Monoamine Oxidases (MAO):

- Catalyze oxidative deamination of endogenous catecholamines (epinephrine)
- Located in nerve terminals and peripheral tissues
- Substrates for catecholamine metabolism found in foods (tyramine) can cause a drug/food interaction
- Inhibited by class of antidepressants called <u>MAO inhibitors</u>

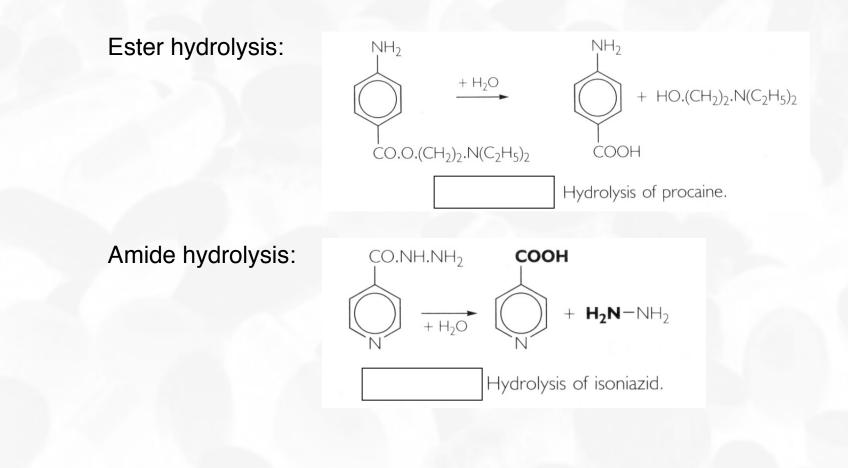
(Inhibition of MAO isoforms in the CNS also effects levels of serotonin - Tranylcypromine) These drugs can cause <u>severe or fatal drug/drug interactions</u> with drugs that increase release of catecholamines or inhibit their reuptake in nerve terminals (Meperidine, pentazocine, <u>dextromethorphan</u>, SSRI antidepressants)

Drug Metabolism - Reduction



Drug Metabolism - Reduction

Hydrolysis reactions



Drug Metabolism - Phase I

- Almost <u>any drug</u> can undergo modifications by drug-metabolizing enzyme systems
- Drugs can be subject to <u>several Phase I pathways</u>
- These reactions create functional groups that place the drugs in a correct chemical state to be acted upon by Phase II conjugative mechanisms
- <u>Main function</u> of phase I reactions is to prepare chemicals for phase II metabolism and subsequent excretion
- <u>Phase II</u> is the true "detoxification" step in the metabolism process.

Drug Metabolism - Phase II

Conjugation reactions

- <u>Glucuronidation</u> by UDP-Glucuronosyltransferase:
 - (on -OH, -COOH, -NH₂, -SH groups)
- Sulfation by Sulfotransferase:
 - (on -NH2, -SO₂NH₂, -OH groups)
- Acetylation by acetyltransferase: (on -NH₂, -SO₂NH₂, -OH groups)
- Amino acid conjugation (on -COOH groups)
- Glutathione conjugation by Glutathione-S-transferase: (to epoxides or organic halides)
- Fatty acid conjugation (on -OH groups)
- Condensation reactions

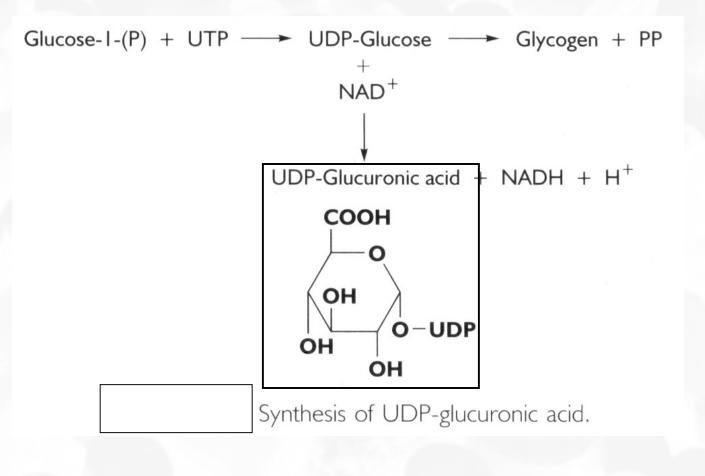
• <u>Glucuronidation (= conjugation to α -d-glucuronic acid)</u>

- Quantitatively the most important phase II pathway for drugs and endogenous compounds
- Products are often excreted in the bile.
- Enterohepatic recycling may occur due to gut glucuronidases
- Requires enzyme <u>UDP-glucuronosyltransferase (UGT)</u>:
 - Genetic family of enzymes
 - Metabolizes a broad range of structurally diverse endogenous and exogenous compounds
 - Structurally related family with approximately 16 isoforms in man

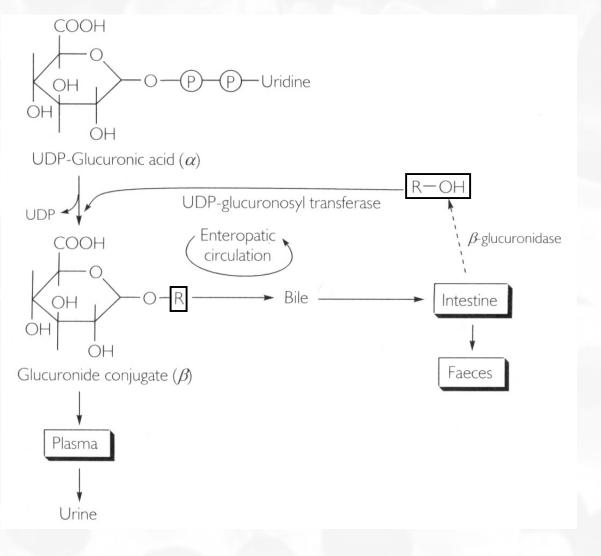
| | UGT activity ^a | | | | | | | | | |
|------------------|---------------------------|-----|-----|-------|------|--------------|-----|------|------|------|
| Substrate | ΙΑΙ | IA3 | 1A4 | 1A6 | 1A8 | 2 A I | 2B4 | 2B7 | 2B15 | 2B17 |
| Simple phenols | 1900 | 239 | 30 | 2400 | 5300 | 735 | 0.4 | 5 | 167 | 38 |
| Bilirubin | 400 | 0 | 2 | 0 | 0 | nd | 0 | 0 | 0 | 0 |
| Carboxylic acids | 0 | 121 | 0 | nd | 170 | 68 | 0 | 2 | 0 | nd |
| Primary amines | I | 84 | 540 | 10600 | 1800 | 22 | nd | 3 | 0 | nd |
| Opioids | 0 | 130 | 0 | 0 | 0 | 73 | 0 | 3462 | 0 | nd |

^a Data are maximum specific enzyme activities (pmol/min/mg protein). nd, not determined. Adapted from Tukey and Strassburg (2000) *Ann. Rev. Pharmacol. Toxicol.* **40**, 581–616. Note that these substrate specificities have yet to be further refined (see text).

Glucuronidation – requires creation of high energy intermediate: <u>UDP-Glucuronic Acid</u>:

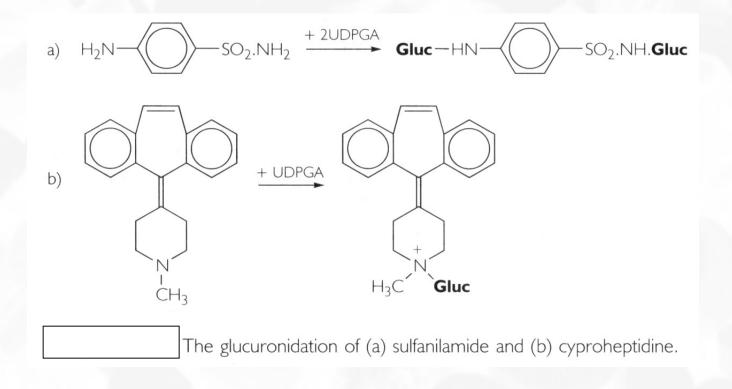


Glucuronidation Pathway and Enterohepatic Recirculation



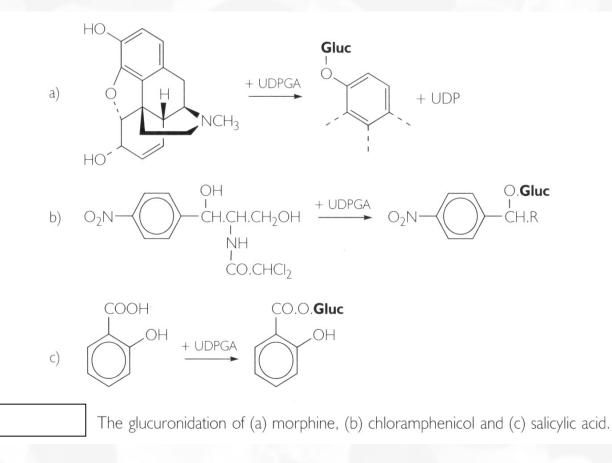
<u>N-glucuronidation:</u>

- Occurs with <u>amines</u> (mainly aromatic)
- Occurs with <u>amides</u> and <u>sulfonamides</u>



<u>O-glucuronidation:</u>

- Occurs by ester linkages with carboxylic acids
- Occurs by ether linkages with phenols and alcohols



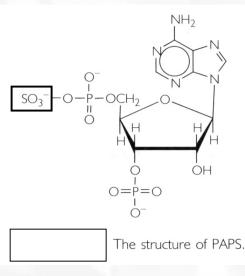
Drug Metabolism - Sulfation

Sulfation:

- Major pathway for phenols but also occurs for alcohols, amines and thiols
- Energy rich donor required:

PAPS (3'-Phosphoadenosine-5'-phosphosulfate)

- Sulfation and glucuronidation are competing pathways:
 - Sulfation predominates at low substrate concentrations
 - Glucuronidation predominates at higher concentrations
 - There is relatively less PAPS in cell cytosol compared to UDPGA



- Sulfotransferases (=SULTs) catalyze transfer of sulfate to substrates:
 - Phenol, alcohol and arylamine sulfotransferases are fairly non-specific
 - Steroid sulfotransferases are very specific

Drug Metabolism - Acylation

Acetylation:

- Common reaction for <u>aromatic amines</u> and <u>sulfonamides</u>
- Requires co-factor <u>acetyl-CoA</u>
- Responsible enzyme is N-acetyltransferase
- Takes place mainly in the liver
- Important in sulfonamide metabolism because acetyl-sulfonamides are less soluble than the parent compound and may cause <u>renal toxicity</u> due to precipitation in the kidney

Fatty Acid Conjugation:

- <u>Stearic</u> and <u>palmitic</u> acids are conjugated to drug by esterification reaction
- Occurs in liver microsomal fraction

(Cannabinols are metabolized in this fashion => long half-life)

Drug Metabolism - Other conjugations

Amino Acid Conjugation:

- ATP-dependent acid:CoA ligase forms active CoA-amino acid conjugates which then react with drugs by N-Acetylation:
 - Usual amino acids involved are:
 - · Glycine. Glutamine, Ornithine, Arginine

Glutathione Conjugation:

- Tripeptide Gly-Cys-Glu; conjugated by glutathione-S-transferase (GST)
- Glutathione is a protective factor for removal of potentially toxic compounds
- Conjugated compounds can subsequently be attacked by <u>γ-glutamyltranspeptidase</u> and a <u>peptidase</u> to yield the cysteine conjugate => product can be further acetylated to N-acetylcysteine conjugate

The further metabolism of a glutathione conjugate.

Drug Metabolism - Phase I & II

Phase I and II - Summary:

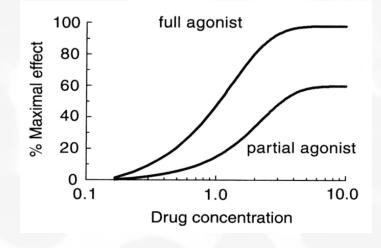
- Products are generally more water soluble
- These reactions products are ready for (renal) excretion
- There are many complementary, sequential and competing pathways
- Phase I and Phase II metabolism are a coupled interactive system interfacing with endogenous metabolic pathways

Drug Action: Receptor Theory

Many drugs act by binding to receptors (see Lecture 4) where they either provoke or inhibit a biological response.

Agonists:

- Can be <u>drugs</u> or <u>endogenous ligands</u> for the receptor
- <u>Increasing concentrations</u> of the agonist will produce an <u>increase in the</u> <u>biological response</u>:
 - <u>Full Agonist:</u> Evokes 100% of the maximum possible effect
 - <u>Partial Agonist</u>: Produces the same type of biological response, but cannot achieve 100% even at very high doses



Drug Action: Receptor Theory

Antagonists:

- Block or reverse the effects of agonists. They have no effects on their own
 - <u>Competitive Antagonists</u>: Compete with agonist for receptor binding => Agonist appears less potent, but <u>can still achieve 100% effect</u> (but at higher concentrations)
 - <u>Non-competitive Antagonists</u>: Bind to receptor at different site and either prevent agonist binding or the agonist effect => <u>maximal achievable response reduced</u>
 - <u>Inverse Agonists</u>: <u>Not the same as antagonists</u>! Inverse agonists trigger a negative response (= reduce baseline) (e.g. diazepam = full agonist = anticonvulsant BUT inverse agonists of benzodiazepin receptor are convulsants)

