

General and Dilution Ventilation



General Ventilation - Purpose

- General ventilation
 - Provide heating or cooling
 - Provide make-up air
 - Provide dilution and reduction of contaminants such as CO₂ and body odor
- Dilution ventilation
 - Provide dilution of contaminants to safe levels (<TLV or LEL)
 - Constrained by comfort and other factors
 - Usually initial cost: DV cost << LEV cost
 - Usually for operation: DV cost >> LEV cost



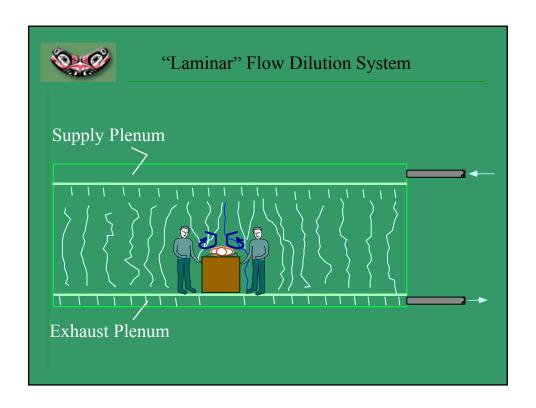
Dilution Ventilation - applications

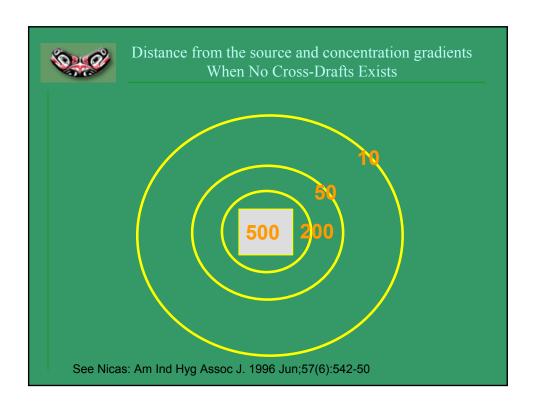
- Toxicity of contaminant is low to moderate (High TLV)
- Velocity and generation rate of contaminant low to moderate – must consider periodic generation too
- Sources are not well localized or identifiable
- Mobile sources or variable work process
- Energy costs are not a significant concern

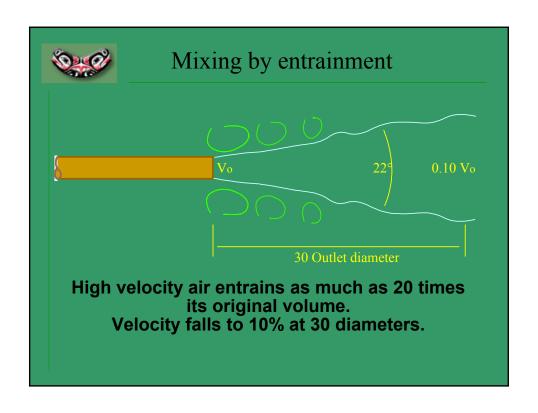


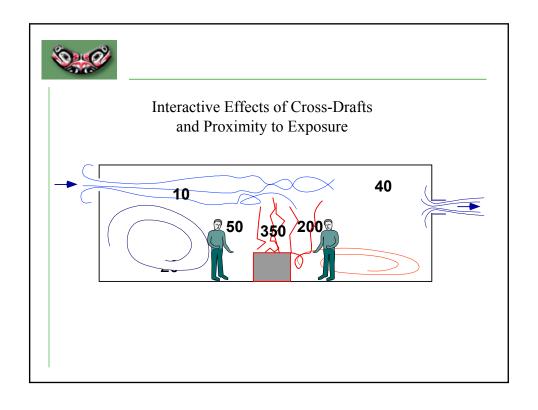
Dilution Ventilation

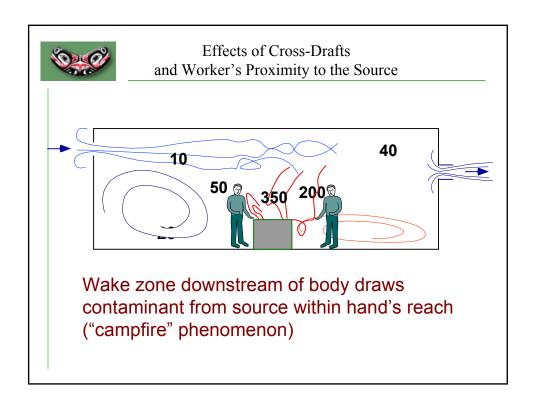
- The solution to pollution is dilution?
- Do you want to move a lot of air?
- What happens in the winter?
- How do you get a sweeping effect?
- Why bother with local exhaust if there are too many sources to vent them all?
- To have effective DV we need to:
 - Mix contaminated air with large volume of fresh air
 - Have sufficient air changes/hour to prevent build-up
 - Create air movement and mixing at all required locations

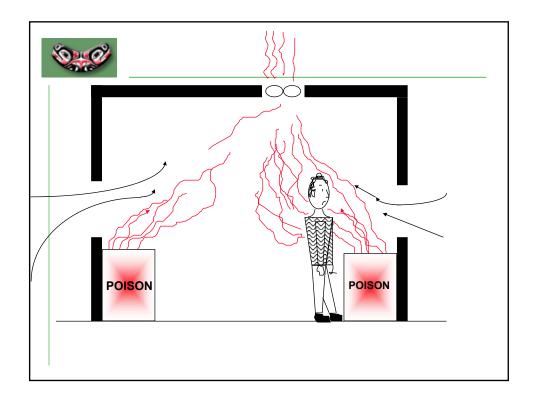


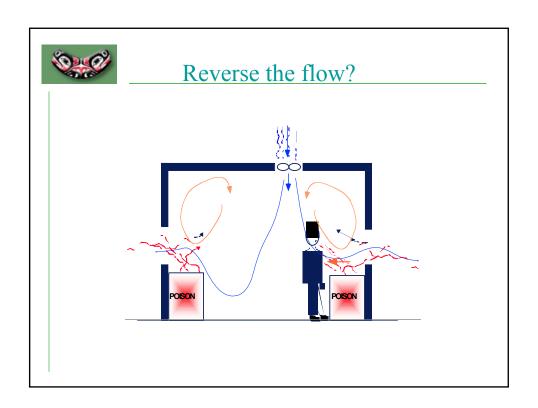


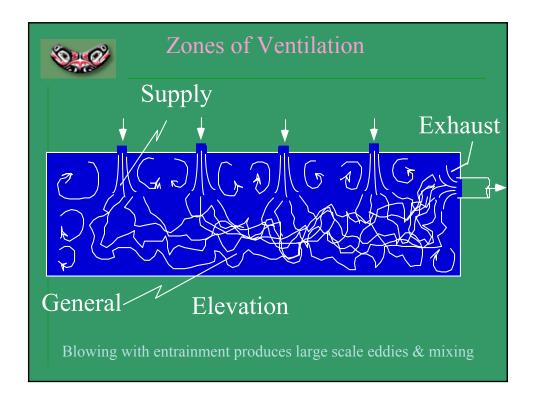














Computing Generation Rate

- We are concerned about dynamic conditions
- Assume constant generation rate

$$G = \frac{d(Vol_{vapor})}{dt}$$

$$G = \frac{d(Vol_{vapor})}{dt} \qquad G = \frac{Amount \ Evaporated}{t_2 - t_1}$$



Calculating dilution volumes

$$VolumeCont = \frac{MassOfTheLiquid}{MassForOneMole} * VolumeForOneMole$$

$$VaporVol = \frac{MassOfTheLiquid}{MW} * 24.04 L* \left(\frac{273.15C+T}{293.15}\right) \left(\frac{760_{\textit{mmHgO}}}{P_{\textit{atm}}}\right)$$

$$VaporVol = \frac{(sp.grav. * \rho_{H2O} * Vol_{liquid})}{MW} * 24.04 L * \left(\frac{273.15C + T}{293.15}\right) \left(\frac{760_{mmHgO}}{P_{atm}}\right)$$



Target Concentration

moderately toxic somewhat toxic slightly toxic

TLV ppm

100-200 > 200

local exhaust only



Concentration if perfect mixing

$$C = \frac{G}{Q}$$

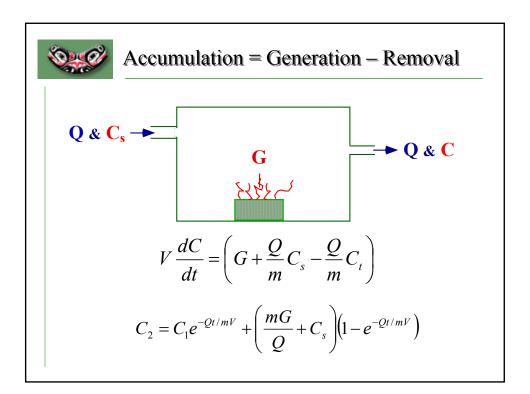
Concentration if not perfect mixing

$$C = \frac{G}{Q / m}$$

Mixing factors
$$m_{i} = \frac{C_{i}}{C_{exhaust}}$$

$$C_{avg} = \frac{1}{8 hours} \sum_{i}^{n} C_{i}t_{i} \qquad C_{avg} = \frac{C_{exhaust}}{8 hours} \sum_{i}^{n} m_{i}t_{i}$$

$$M_{avg} = \frac{1}{8 hrs} \sum_{i}^{n} m_{i}t_{i} \qquad M_{peak} = \frac{\overline{C}_{15 \, \text{min}}}{C_{exhaust}}$$





ROOM VOLUME

$$C_2 = C_1 e^{-Qt/mV} + \left(\frac{mG}{Q} + C_s\right) \left(1 - e^{-Qt/mV}\right)$$

Important for transient conditions
Irrelevant for steady state

$$C_t = \frac{m G}{Q}$$



Application of Equations

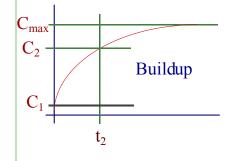
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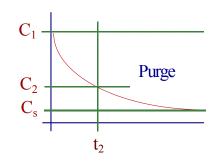
$$C_t = \frac{m G}{Q}$$
 "Steady"-State Generation

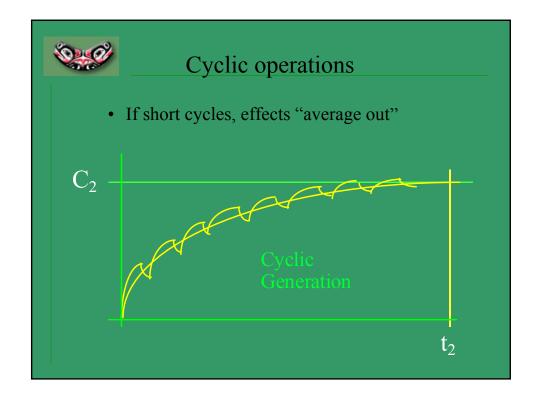


Application of Equations

$$C_2 = C_1 e^{-Qt/mV} + \left(\frac{mG}{Q} + C_s\right) \left(1 - e^{-Qt/mV}\right)$$







When conditions change with time

- Solve for time interval during which all conditions are constant
- If conditions change continuously, make interval one minute
- · Use result as initial conditions for next interval

$$C_{i} = C_{i-1} e^{-Q_{i} \Delta t_{i}/m_{i}R} + \left(\frac{m_{i}G_{i}}{Q_{i}} + C_{Si}\right) \left(1 - e^{-Q_{i} \Delta t_{i}/m_{i}R}\right)$$



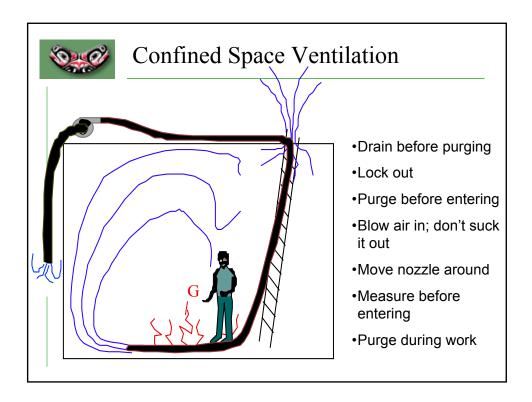
DESIGNING NEW SYSTEM

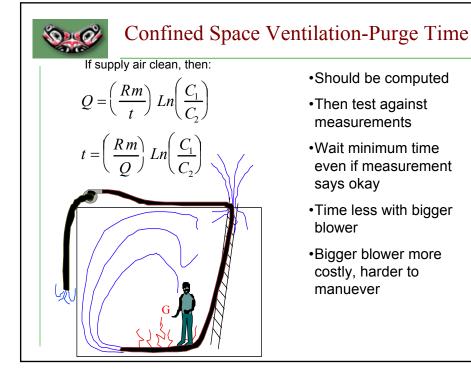
Locate sources near exhaust fans

- Locate supply air outlets to direct air away from face and towards exhaust fans
- Separate sources from traffic using barriers
- Block undesirable cross drafts and competing sources of motion using barriers.

EXISTING SYSTEM IMPROVEMENTS

- Substitute less volatile or toxic chemicals
- Install or improve local exhaust hoods
- Reduce incidence of spills and leaks
- Relocate supply and exhaust points
- Relocate workers or the sources or both
- Increase airflow







Example Problem

Initial measurements indicate 10,000 ppm of xylene in a confined space. Assuming that Ct = 0.25 * TLV, how much should Q be to allow entry in 30 minutes if:

 $R=1000 \text{ ft}^3, M=3, C_s=0$

$$Q = \left(\frac{Rm}{\Delta t}\right) \ln \left[\frac{G + QC_S/m - QC_O/m}{G + QC_S/m - QC_2/m}\right] = \left(\frac{1000 \, \text{ft}^3 * 3}{30 \, \text{min}}\right) \ln \left[\frac{010^{-2}}{025 * 10^{-6}}\right] = 599 \, \text{fts/min}$$

b. $R=2000 \text{ ft}^3, M=3, C_s=0$: Solution: $Q = 1198 \text{ ft}^3/\text{min}$

c. $R=1000 \text{ ft}^3$, M=6, $C_s=0$: Solution: $Q=1198 \text{ ft}_3/\text{min}$

d. R=1000 ft³, M=6, $C_s = 15$ ppm: Solution: Q = 1381 ft³/min



Summary

- · Estimating G and m is difficult
- Reduce G as much as possible
- Reduce greatest contributors to exposure and perceived exposure first
- Use sweeping, but be realistic about it
- Complement local exhaust systems
- Provide winter and summer
- Purge before and during confined space entry