

Grade 8 FCAT Science Sample Questions

The intent of these sample test materials is to orient teachers and students to the types of questions on FCAT tests. By using these materials, students will become familiar with the types of items and response formats that they will see on the actual test. The sample test materials are not intended to demonstrate the length of the actual test, nor should student responses be used as an indicator of student performance on the actual test. Additional information about test items can be found in the *FCAT Test Item Specifications* at <http://fcats.fldoe.org/fcatis01.asp> and previously released FCAT tests at <http://fcats.fldoe.org/fcatrelease.asp>.

Directions for Answering the Science Sample Questions

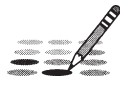
Mark your answers on the Sample Answer Sheet, which begins on page 13. If you don't understand a question, ask your teacher to explain it to you. Your teacher has the answers to the sample questions.

The sample questions for students and the sample answers for teachers will only be available online at: <http://fcats.fldoe.org/fcatsmpl.asp>.

You may need the Reference Sheet or the Periodic Table to help you answer some of the questions. You may refer to the Reference Sheet (page 5) or the Periodic Table (page 6) as often as you like.

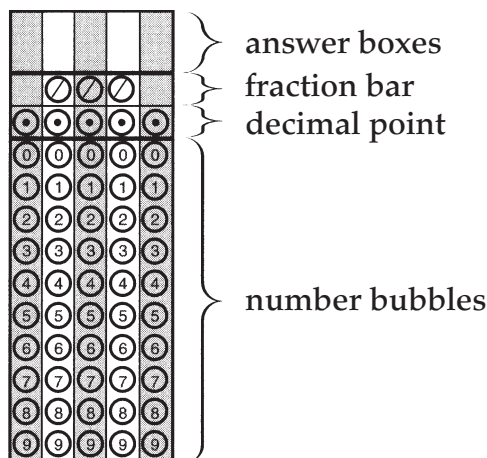
Use the space in this booklet to do your work on the multiple-choice and gridded-response questions, but be sure to put your answers on the Sample Answer Sheet.

How to Complete the Response Grids

Science test questions with this symbol  require that you fill in a grid in your answer book. There may be more than one correct way to fill in a response grid. This section shows you different ways the response grid may be completed.

Parts of a Response Grid

For Grade 8, response grids have the following parts:



Directions

1. Work the problem and find an answer or solution.
2. Write your answer in the answer boxes at the top of the grid.
 - Print your answer with the first digit in the left answer box OR with the last digit in the right answer box.
 - Print only one digit or symbol in each answer box. Do NOT leave a blank answer box in the middle of an answer.
 - Be sure to write a decimal point or fraction bar in the answer box if it is a part of the answer.

3. Fill in a bubble under each box in which you wrote your answer.

- Fill in one and ONLY one bubble for each answer in an answer box. Do NOT fill in a bubble under an unused answer box.
- Fill in each bubble by making a solid black mark that completely fills the circle.
- You MUST fill in the bubbles accurately to receive credit for your answer.

Examples

Whole Number

$60 + 10 =$

7	0			
/	/	/		
•	•	•	•	•
0	•	0	0	0
1	1	1	1	1
2	2	2	2	2
3	3	3	3	3
4	4	4	4	4
5	5	5	5	5
6	6	6	6	6
•	7	7	7	7
8	8	8	8	8
9	9	9	9	9

OR

			7	0
/	/	/		
•	•	•	•	•
0	0	0	0	•
1	1	1	1	1
2	2	2	2	2
3	3	3	3	3
4	4	4	4	4
5	5	5	5	5
6	6	6	6	6
7	7	7	•	7
8	8	8	8	8
9	9	9	9	9

Decimal

Show the decimal equivalent of $5\frac{6}{100}$

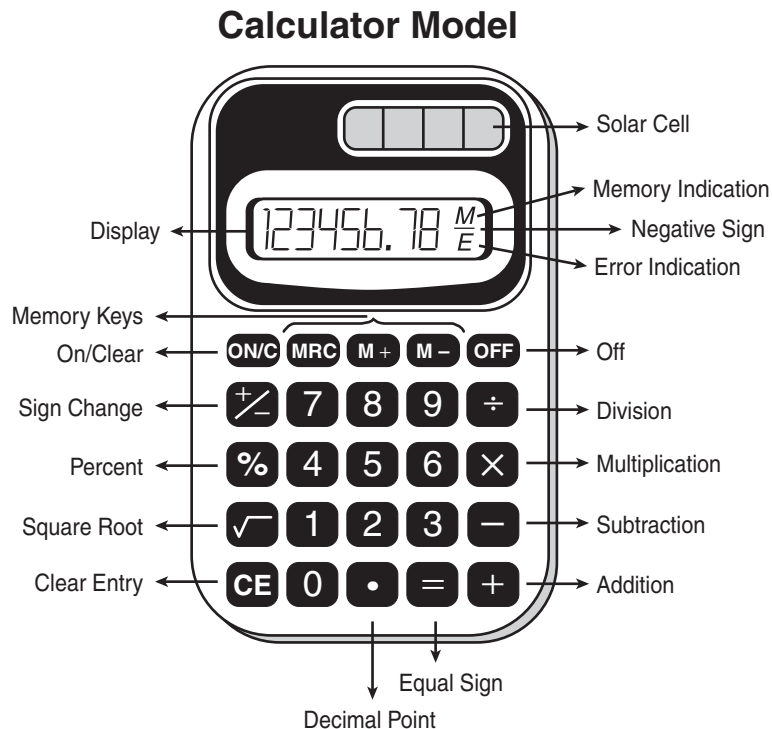
5	.	0	6	
/	/	/		
•	•	•	•	•
0	0	•	0	0
1	1	1	1	1
2	2	2	2	2
3	3	3	3	3
4	4	4	4	4
•	5	5	5	5
6	6	6	•	6
7	7	7	7	7
8	8	8	8	8
9	9	9	9	9

OR

5	.	0	6	
/	/	/		
•	•	•	•	•
0	0	0	0	•
1	1	1	1	1
2	2	2	2	2
3	3	3	3	3
4	4	4	4	4
5	•	5	5	5
6	6	6	6	•
7	7	7	7	7
8	8	8	8	8
9	9	9	9	9

CALCULATOR INSTRUCTIONS

1. Read the problem very carefully. Then decide whether or not you need the calculator to help you solve the problem.
2. When starting a new problem, always clear your calculator by pressing the on/clear key.
3. If you see an **E** in the display, clear the error before you begin.
4. If you see an **M** in the display, clear the memory and the calculator before you begin.
5. If the number in the display is not one of the answer choices, check your work. Remember that when computing with certain types of fractions, you may have to round the number in the display.
6. Remember, your calculator will NOT automatically perform the algebraic order of operations.
7. Calculators might display an incorrect answer if you press the keys too quickly. When working with calculators, use careful and deliberate keystrokes, and always remember to check your answer to make sure that it is reasonable.
8. The negative sign may appear either to the left or to the right of the number.
9. Always check your answer to make sure that you have completed all of the necessary steps.



Grade 8 FCAT Science Reference Sheet

Equations

Acceleration (a)	= $\frac{\text{change in velocity (m/s)}}{\text{time taken for this change (s)}}$	a = $\frac{v_f - v_i}{t_f - t_i}$
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Average speed (v)	= $\frac{\text{distance}}{\text{time}}$	v = $\frac{d}{t}$
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Density (D)	= $\frac{\text{mass (g)}}{\text{Volume (cm}^3\text{)}}$	D = $\frac{m}{V}$
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Percent Efficiency (e)	= $\frac{\text{Work out (J)}}{\text{Work in (J)}} \times 100$	%e = $\frac{W_{\text{out}}}{W_{\text{in}}} \times 100$
------------------------	---------------------------------------------------------------	--------------------------------------------------------

Force (F)	= mass (kg) \times acceleration (m/s ²)	F = ma
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Frequency (f)	= $\frac{\text{number of events (waves)}}{\text{time (s)}}$	f = $\frac{n \text{ of events}}{t}$
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Momentum (p)	= mass (kg) \times velocity (m/s)	p = mv
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Wavelength (λ)	= $\frac{\text{velocity (m/s)}}{\text{frequency (Hz)}}$	λ = $\frac{v}{f}$
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Work (W)	= Force (N) \times distance (m)	W = Fd
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Units of Measure

m = meter	g = gram	s = second
cm = centimeter	kg = kilogram	Hz = hertz (waves per second)
J = joule (newton-meter)		
N = newton (kilogram-meter per second squared)		

Periodic Table of the Elements

(based on $^{12}_6\text{C} = 12.0000$)

Representative Elements

Period	Group 1		Transition Metals										Representative Elements					
	1A	2A	3B	4B	5B	6B	7B	8B		1B	2B	3A	4A	5A	6A	7A	8A	
1	H 1 Hydrogen 1.008																He 2 Helium 4.003	
2	Li 3 Lithium 6.941	Be 4 Beryllium 9.012											B 5 Boron 10.81	C 6 Carbon 12.011	N 7 Nitrogen 14.007	O 8 Oxygen 15.999	F 9 Fluorine 18.998	Ne 10 Neon 20.180
3	Na 11 Sodium 22.990	Mg 12 Magnesium 24.305											Al 13 Aluminum 26.982	Si 14 Silicon 28.086	P 15 Phosphorus 30.974	S 16 Sulfur 32.06	Cl 17 Chlorine 35.453	Ar 18 Argon 39.948
4	K 19 Potassium 39.098	Ca 20 Calcium 40.078	Sc 21 Scandium 44.956	Ti 22 Titanium 47.88	V 23 Vanadium 50.942	Cr 24 Chromium 51.996	Mn 25 Manganese 54.938	Fe 26 Iron 55.847	Co 27 Cobalt 58.933	Ni 28 Nickel 58.693	Cu 29 Copper 63.546	Zn 30 Zinc 65.39	Ga 31 Gallium 69.723	Ge 32 Germanium 72.61	As 33 Arsenic 74.922	Se 34 Selenium 78.96	Br 35 Bromine 79.904	Kr 36 Krypton 83.80
5	Rb 37 Rubidium 85.468	Sr 38 Strontium 87.62	Y 39 Yttrium 88.906	Zr 40 Zirconium 91.224	Nb 41 Niobium 92.906	Mo 42 Molybdenum 95.94	Tc 43 Technetium 98	Ru 44 Ruthenium 101.07	Rh 45 Rhodium 102.906	Pd 46 Palladium 106.42	Ag 47 Silver 107.868	Cd 48 Cadmium 112.411	In 49 Indium 114.82	Sn 50 Tin 118.710	Sb 51 Antimony 121.757	Te 52 Tellurium 127.60	I 53 Iodine 126.905	Xe 54 Xenon 131.29
6	Cs 55 Cesium 132.905	Ba 56 Barium 137.327	La 57 Lanthanum 138.905	Hf 72 Hafnium 178.49	Ta 73 Tantalum 180.948	W 74 Tungsten 183.85	Re 75 Rhenium 186.207	Os 76 Osmium 190.2	Ir 77 Iridium 192.22	Pt 78 Platinum 195.08	Au 79 Gold 196.967	Hg 80 Mercury 200.59	Tl 81 Thallium 204.383	Pb 82 Lead 207.2	Bi 83 Bismuth 208.980	Po 84 Polonium 208.982	At 85 Astatine 210	Rn 86 Radon 222
7	Fr 87 Francium 223	Ra 88 Radium 226.025	Ac 89 Actinium 227.028	Rf 104 Rutherfordium (261)	Db 105 Dubnium (262)	Sg 106 Seaborgium (263)	Bh 107 Bohrium (264)	Hs 108 Hassium (265)	Mt 109 Meitnerium (268)									

14 — Atomic number
Si — Symbol
 Silicon — Name
 28.086 — Average Atomic Mass

← Metals | Nonmetals →

Inner Transition Metals

Lanthanide series

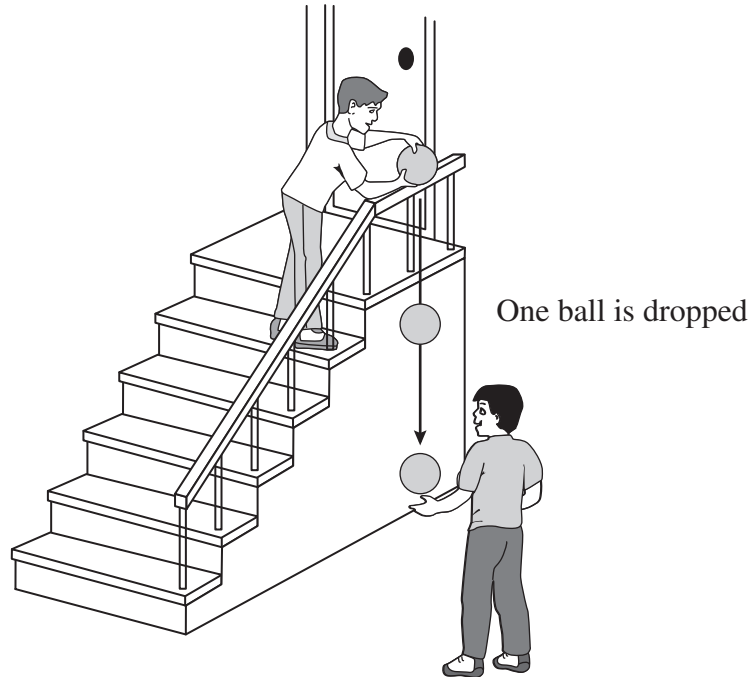
58 Ce Cerium 140.12	59 Pr Praseodymium 140.908	60 Nd Neodymium 144.24	61 Pm Promethium 144.913	62 Sm Samarium 150.36	63 Eu Europium 151.96	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.50	67 Ho Holmium 164.930	68 Er Erbium 167.26	69 Tm Thulium 168.934	70 Yb Ytterbium 173.04	71 Lu Lutetium 174.967
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90 Th Thorium 232.038	91 Pa Protactinium 231.036	92 U Uranium 238.029	93 Np Neptunium 237.048	94 Pu Plutonium 244.064	95 Am Americium 243.061	96 Cm Curium 247.070	97 Bk Berkelium 247.070	98 Cf Californium 251.080	99 Es Einsteinium 252.083	100 Fm Fermium 257.095	101 Md Mendelevium 258.099	102 No Nobelium 259.101	103 Lr Lawrencium 260.105
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Actinide series

- 1 A calorimeter is a device used to measure the energy content of food. In a calorimeter, a measured amount of food is burned, heating a known mass of water. The temperature change of the water is measured with a thermometer. Which type of energy present in food can be measured by the calorimeter?
- A. chemical
 - B. electric
 - C. light
 - D. mechanical
- 2 Mason and his friends were playing baseball outside his house when they saw lightning and heard thunder rumbling in the distance. They wanted to know how far away the storm was, so they went inside, and as soon as they saw the next lightning flash, they started counting seconds until they heard the thunder. For every three seconds they counted, they knew the storm was about one kilometer away. Which of the following properties did Mason and his friends use to determine the distance of the storm?
- F. speed of light
 - G. speed of sound
 - H. speed of the storm
 - I. speed of the lightning
- 3 The Milky Way is a spiral galaxy, yet when Gordon looked at the stars one night, the Milky Way Galaxy looked like a flat band of stars across the sky. Why does the Milky Way Galaxy appear to be flat?
- A. Earth is inside the Galaxy.
 - B. Earth is close to the other stars in the Galaxy.
 - C. Earth is rotating too quickly to view the Galaxy.
 - D. Earth's atmosphere distorts light from the Galaxy.

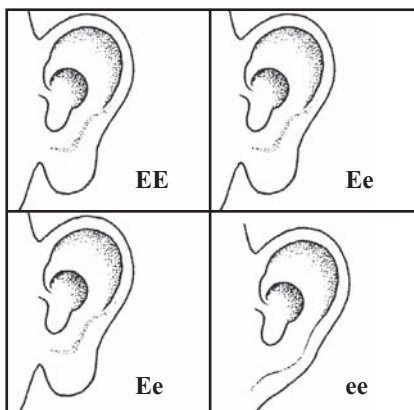
- 4 Sam drops a rubber ball from the top of a staircase to his friend, as shown in the picture below. The speed of the ball increases as it gets farther away from Sam's hand.



What is the **best** explanation for the increase in the speed of the ball?

- F. Electric forces are acting on the ball.
 - G. Air resistance is accelerating the ball.
 - H. Magnetic forces are pushing the ball.
 - I. Gravitational force is pulling on the ball.
- 5 The Rocky Mountains are known for beautiful scenery. Which processes of the Rocky Mountains **most likely** required the longest period of time to occur?
- A. growth of plants and trees
 - B. formation of valleys from glaciers
 - C. large uplift of the land from tectonic forces
 - D. large accumulation of snow on the mountaintops

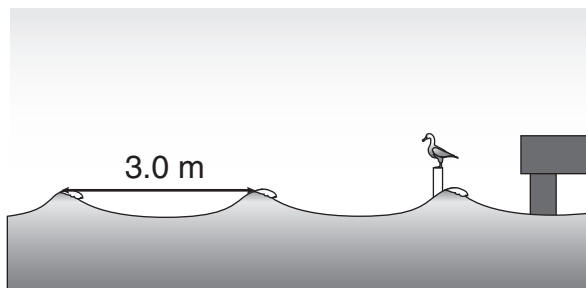
- 6 Elena has been studying the inheritance of traits in humans. She learned that unattached earlobes (E) are dominant over attached earlobes (e). The Punnett square shown below represents four children in one family.



According to the results shown in the Punnett square, what are the genotypes of the parents?

- F. EE and EE
- G. Ee and Ee
- H. EE and ee
- I. ee and ee

- 7 The picture below shows a bird sitting on a post near an ocean pier. Water waves are moving past the bird. Every 20 seconds (s), 10 waves pass by the bird. Each wave has a wavelength of 3.0 meters (m).



What is the frequency, in **hertz (Hz)**, of the water waves as they pass the bird?

8

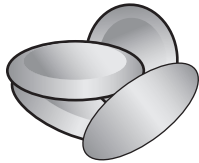
Leah is using a 3.4-meter (m) ladder to paint the exterior of her house. Leah has a mass of 50 kilograms (kg). The acceleration due to gravity is equal to 9.8 meters per second squared (m/s^2). What force, **in newtons (N)**, does Leah exert on the ladder when she is standing on it?



9

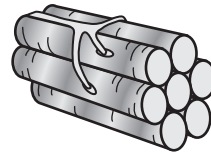
The human body has many different types of cells. Which of the following cells carries genetic information from parents to their children?

A.



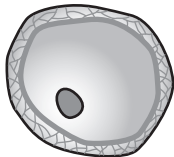
red blood cells

C.



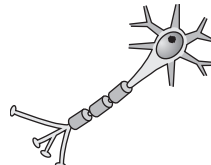
muscle cells

B.



reproductive cell

D.



nerve cell

- 10 The table below shows the distance from the Sun and the period of revolution of the first three planets in the Solar System.

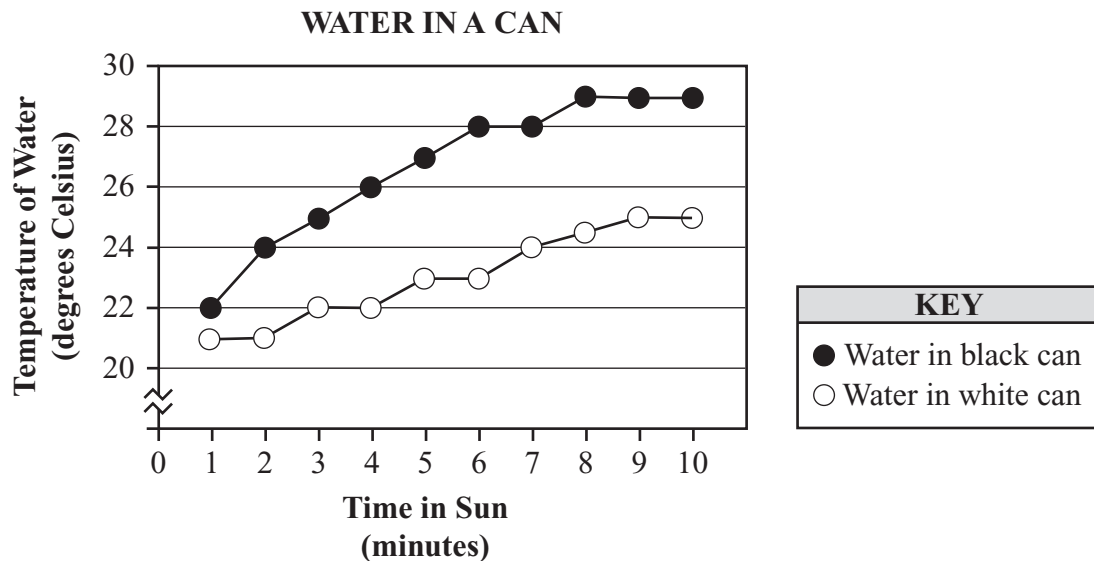
THE SOLAR SYSTEM

Name of Planet	Average Distance from Sun (millions of km)	Period of Revolution (Earth days)
Mercury	58	88
Venus	108	225
Earth	150	365

Based on the data in the table, approximately how many planetary years does Mercury complete during the length of time that Earth has one planetary year?

- F. $\frac{1}{4}$ of a year
 - G. 4 years
 - H. 6 years
 - I. 24 years
- 11 A layer of small rock and mineral particles can provide a surface for plant growth. Plants can eventually grow to cover this surface, while their roots spread out beneath the surface. Which natural process can **most likely** be reduced or prevented by the introduction of such plant growth?
- A. erosion
 - B. flooding
 - C. weathering
 - D. soil formation

- 12 Kevin hypothesized that dark colors absorb more heat than light colors. He put equal quantities of water in a can covered with black paper and in a can covered with white paper. Kevin measured the temperature of the water in each can at regular intervals on several sunny days. He constructed a graph to display the average temperature of each can over the course of his experiment.



Which of the following actions made Kevin's experiment **more** valid?

- F. He graphed his results.
 - G. He used a lengthy procedure.
 - H. He repeated his experiment on several days.
 - I. He recorded the temperature in metric units.
- 13 The population of the world continues to grow, creating an increased need for fuel to heat our homes, schools, and workplaces. Which of the following would **best** conserve Earth's natural resources?
- A. using more electricity as a source of heat
 - B. using more solar energy as a source of heat
 - C. using more coal or charcoal as a source of heat
 - D. using more oil or natural gas as a source of heat

Name _____

Answer all the Science Sample Questions on this Sample Answer Sheet.

1 A B C D

2 F G H I

3 A B C D

4 F G H I

5 A B C D

6 F G H I

7

	/	/	/	
<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>
0	0	0	0	0
1	1	1	1	1
2	2	2	2	2
3	3	3	3	3
4	4	4	4	4
5	5	5	5	5
6	6	6	6	6
7	7	7	7	7
8	8	8	8	8
9	9	9	9	9

8

	/	/	/	
<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>	<input type="radio"/>
0	0	0	0	0
1	1	1	1	1
2	2	2	2	2
3	3	3	3	3
4	4	4	4	4
5	5	5	5	5
6	6	6	6	6
7	7	7	7	7
8	8	8	8	8
9	9	9	9	9

9 A B C D

10 (F) (G) (H) (I)

11 (A) (B) (C) (D)

12 (F) (G) (H) (I)

13 (A) (B) (C) (D)



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