

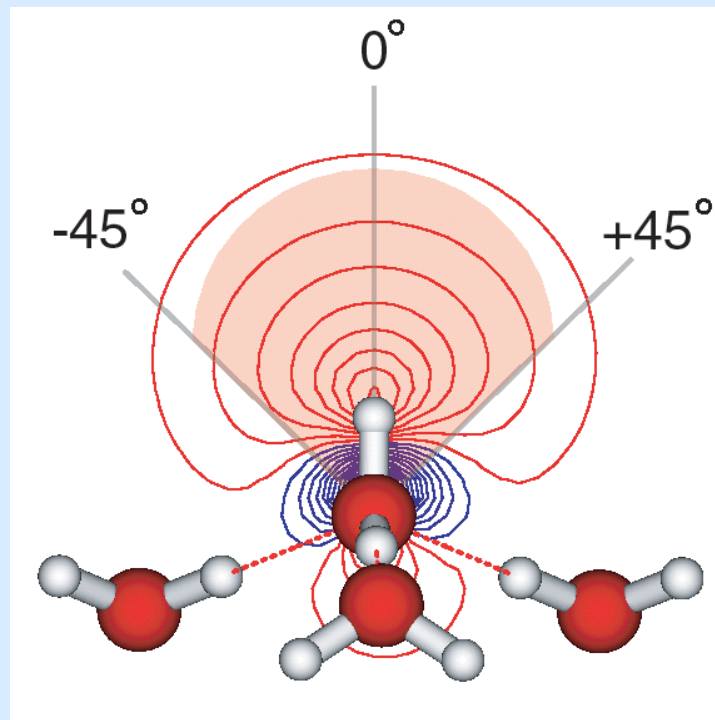
H₂O Under Pressure

Russell J. Hemley

*Geophysical Laboratory
Carnegie Institution of Washington
Washington, DC*

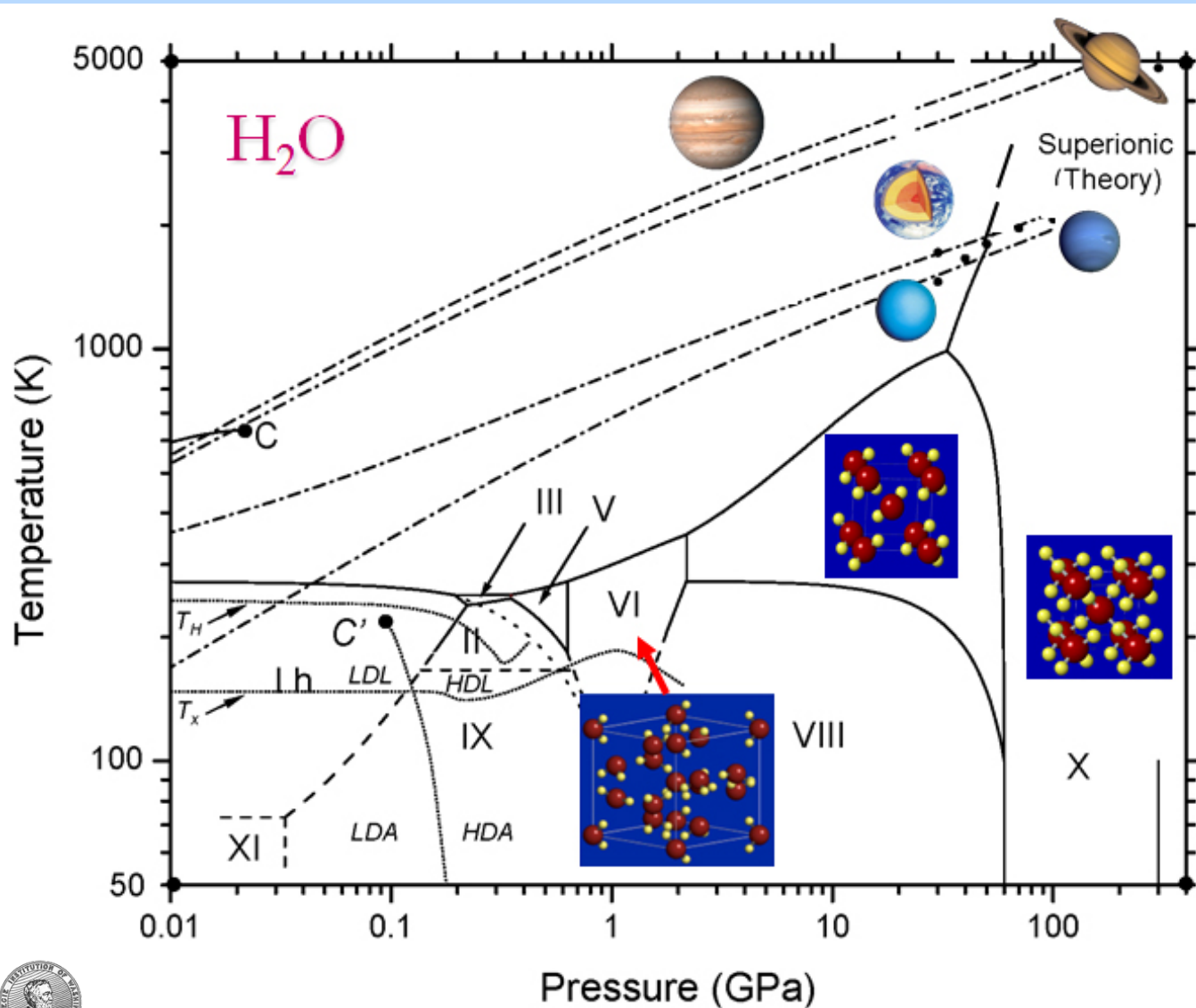


EFree



*Workshop on Fundamental Challenges in our Understanding of the
Physics and Chemistry of Water, January 16-17, 2015*

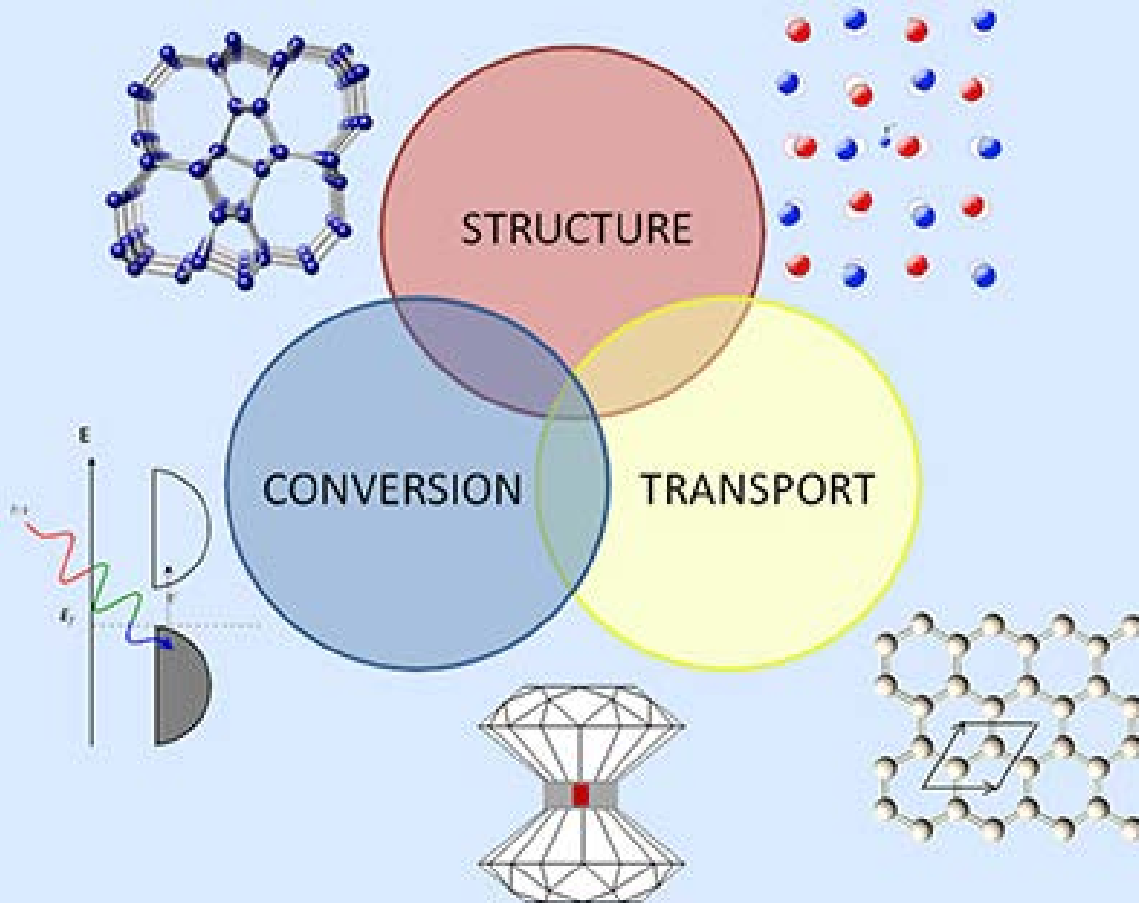
High-pressure behavior of water continues to present new questions and surprises



- Origin of stability?
~20 stable and metastable phases
- Novel transitions?
- non-molecular
- amorphization
- superionic
- liquid/liquid trans
- High P - T fluid?
- Electronic prop.?
- New chemistry?
- Breakdown of H₂O?
- Supporting life at extreme P - T ?



Mission: To accelerate the discovery and synthesis of new energy materials using extreme conditions



Carnegie:

- **Director:** R. J. Hemley
- **Assoc. Director:** T. S. Strobel
- **Admin.** S. Gramsch, M. Phillips
- **Carnegie Partners:** R. Boehler, Fei, D. Kim, Z. Liu, H. K. Mao, V. Struzhkin, W. Yang

University Partners:

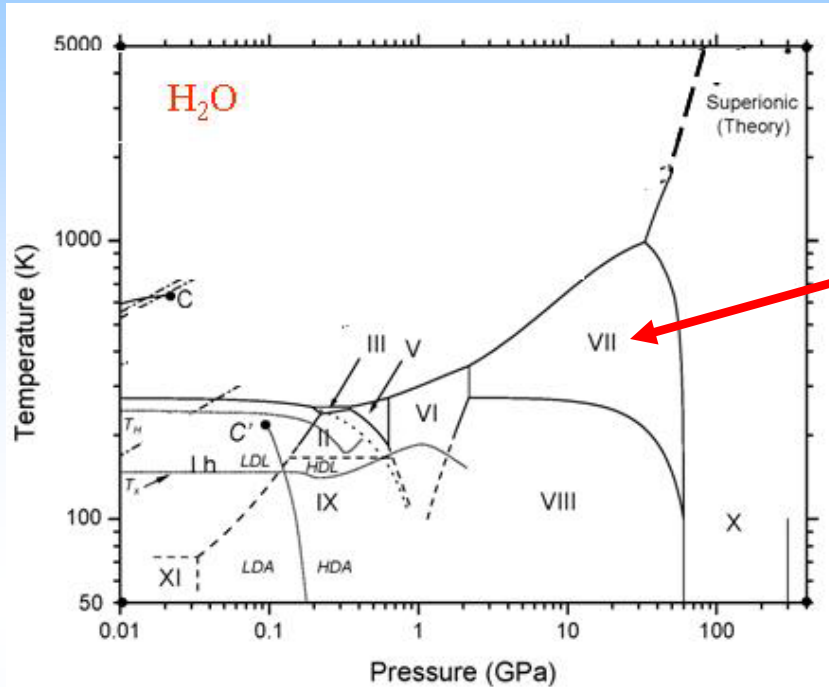
- **Penn State:** J. Badding, N. Alem, Crespi
- **Cornell:** R. Hoffmann, N. Ashcroft
- **Colorado School Mines:** C. Taylor
- **Caltech:** B. Fultz
- **Lehigh:** K. Landskron

DOE Facilities

- **APS, ANL (X-ray)**
- **SNS, ORNL (Neutron)**
- **NSLS II, BNL (IR)**

Continuing puzzles in ice VII

[Kamb & Davis, *PNAS* (1964)]



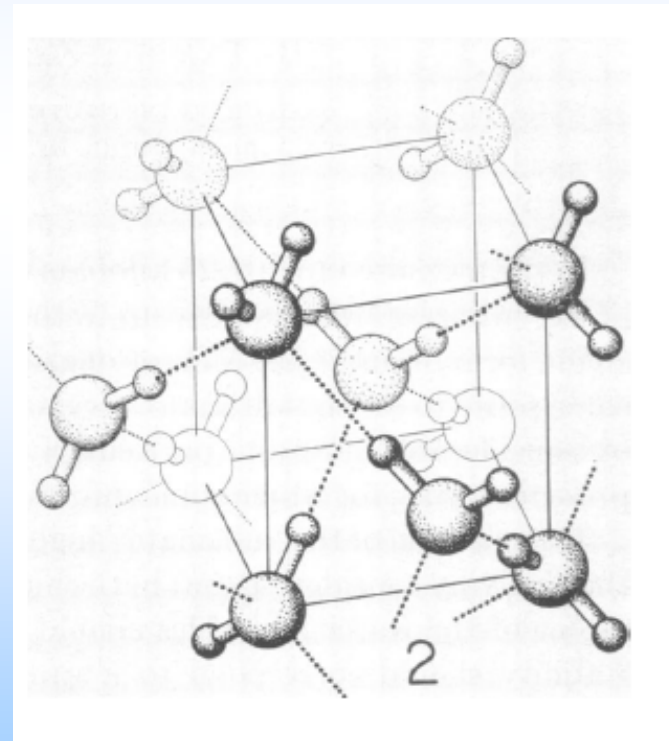
ICE VII, THE DENSEST FORM OF ICE*

BY BARCLAY KAMB AND BRIANT L. DAVIS

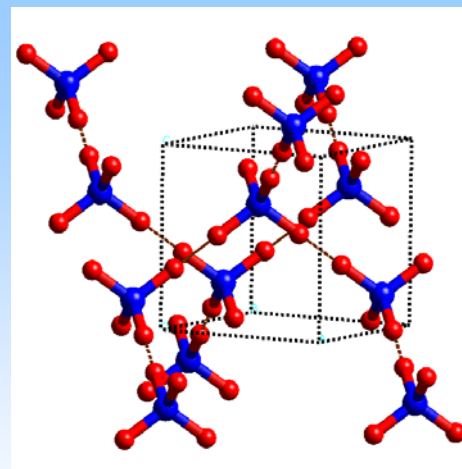
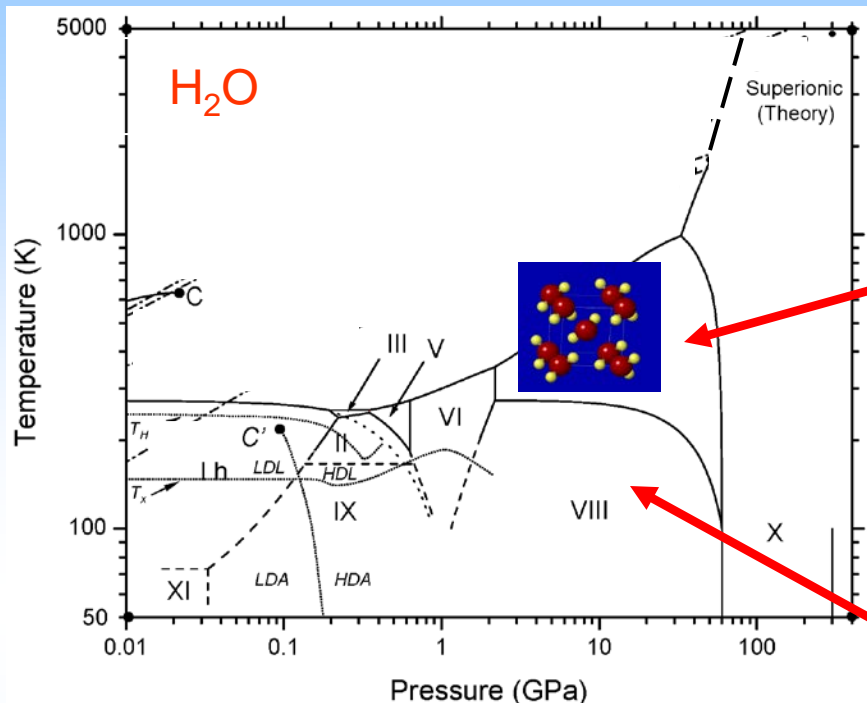
CALIFORNIA INSTITUTE OF TECHNOLOGY, PASADENA, AND
SOUTH DAKOTA SCHOOL OF MINES AND TECHNOLOGY, RAPID CITY

Communicated by Linus Pauling, October 19, 1964

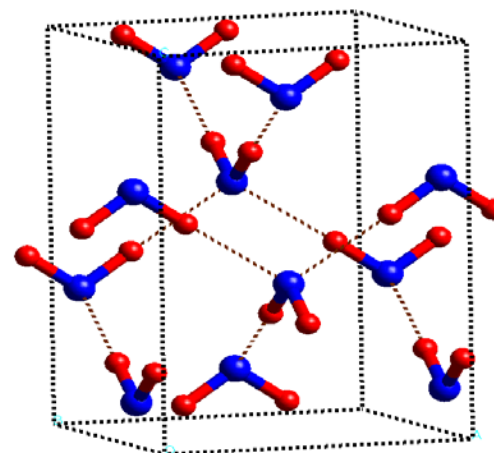
- O atoms lie on bcc lattice
- Two interpenetrating ice Ic lattices
- Retains tetrahedral motif seen in lower pressure ice phases
- Predicted H-bond symmetrization



Continuing puzzles in ice VII



**Ice VII
proton
disordered**



**Ice VIII
anti-
ferroelectric**

➤ **Nature of the proton ordering**

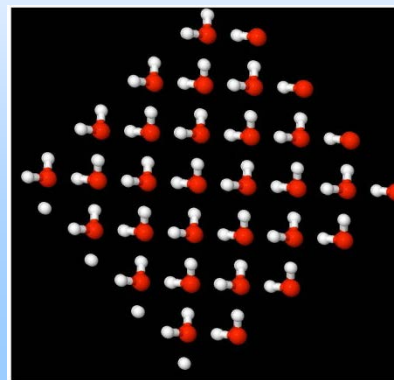
[Kuks et al., *J. Chem. Phys.* (1984)]

➤ **Structural transitions
observed in ice VII**

[Somayazulu et al. *J. Chem. Phys.* (2008)]

➤ **Is there a ferroelectric form?**

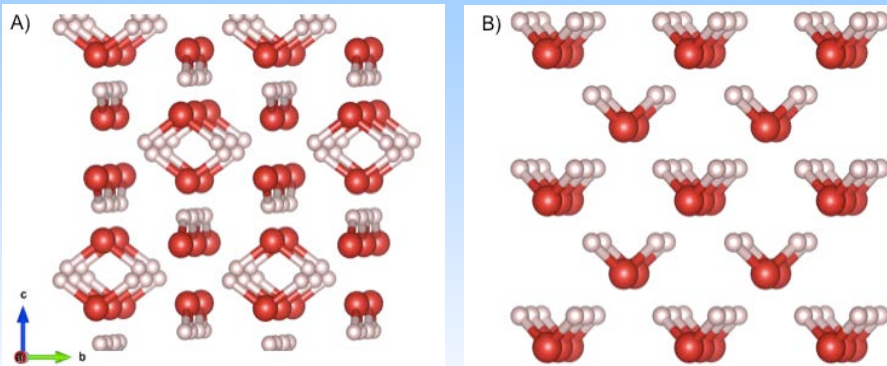
➤ **Higher P-T behavior?**



**Hypothetical
polar ice VIII
(ferroelectric)**

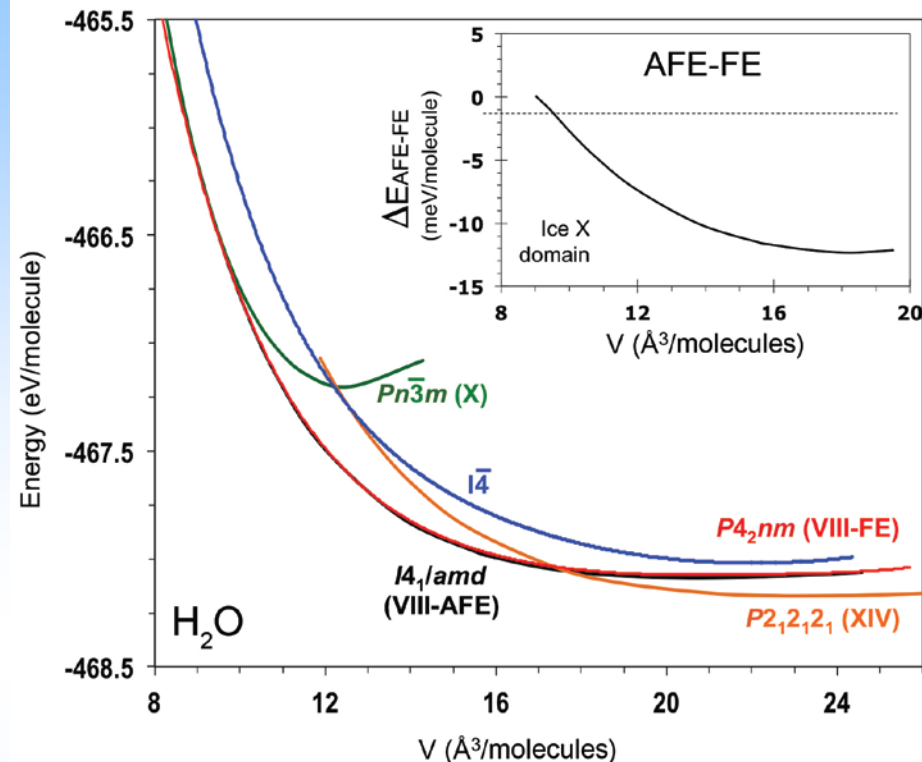
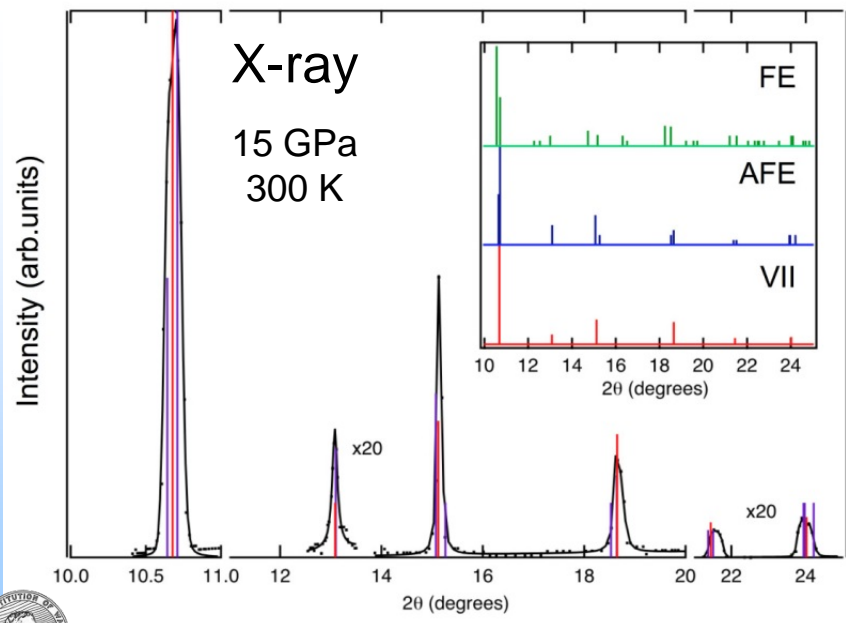


Ferroelectricity in dense H₂O



**Anti-Ferroelectric
(VIII-AFE)**

**Ferroelectric
(VIII-FE)**

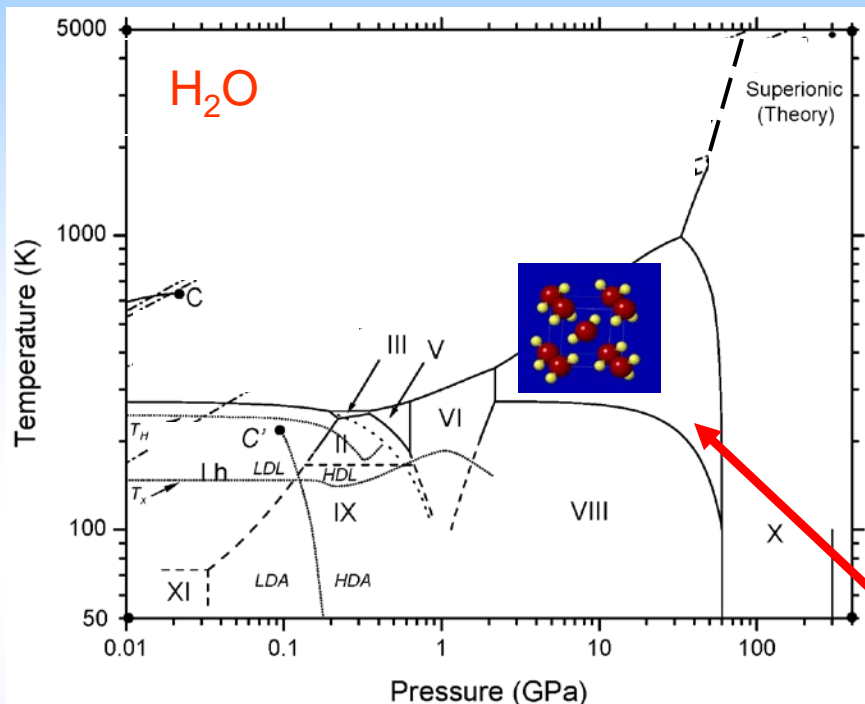


- AFE-FE nearly degenerate
- Fit to x-ray data
- Mixed domains of FE and AFE
- Stabilized by pressure and epitaxial growth?

[Caracas & Hemley, to be published]



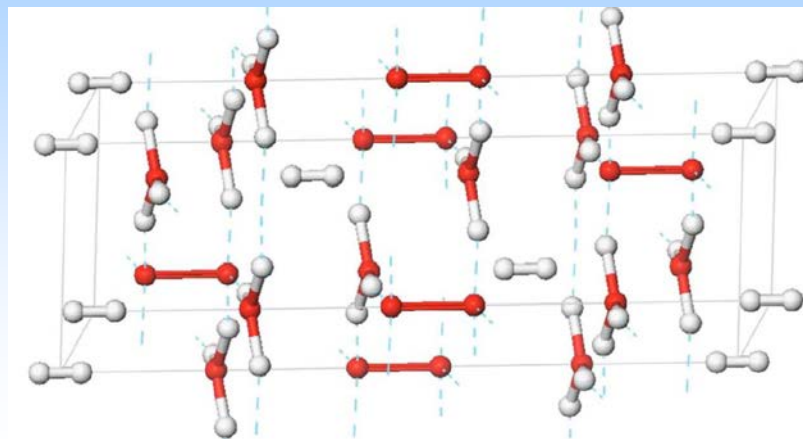
Water splitting in dense ice



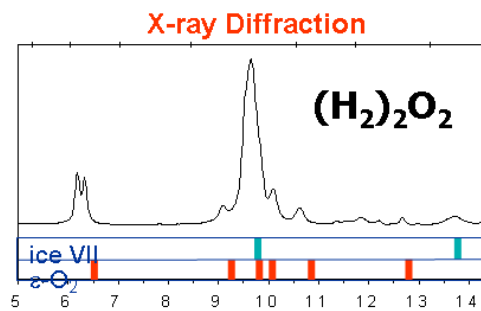
- Molecular alloy of H₂-O₂
- Metastable energetic material
- Similar results found for NH₃
[Lazor *et al.*, to be published]
- Mechanism not understood

THEORY: 'When is H₂O not water?'

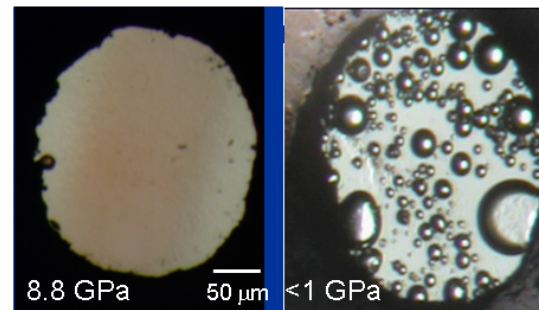
[Pickard & Needs, *J. Chem. Phys.* (2007)]



[W. Mao *et al.*, *Science* (2006)]



X-ray-induced reaction

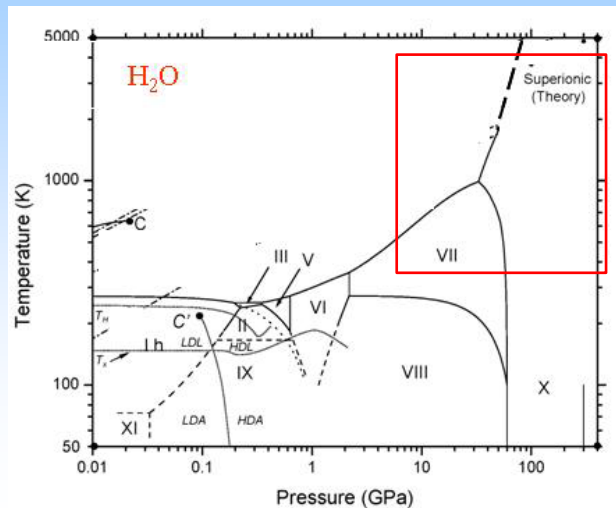


Hard (>7 keV) x-ray photons

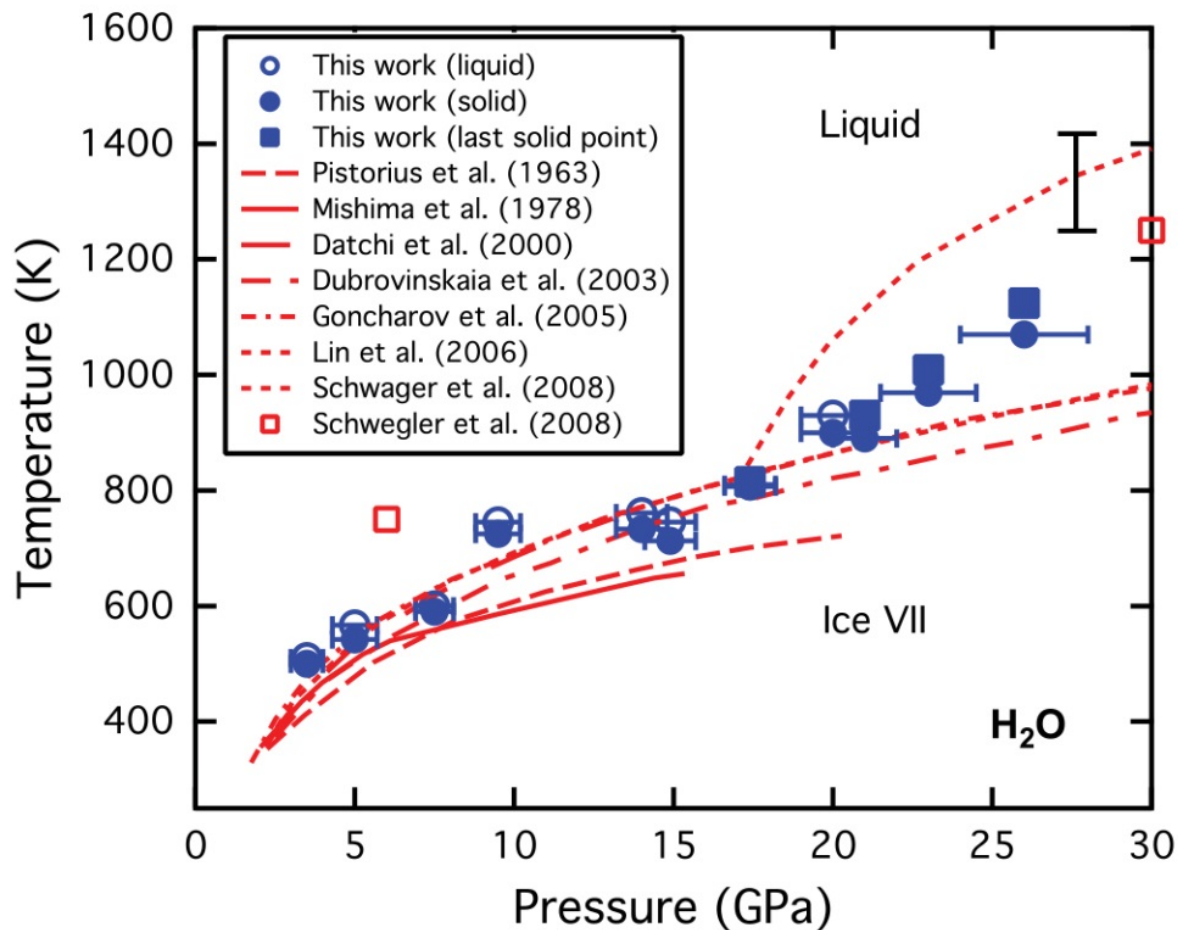
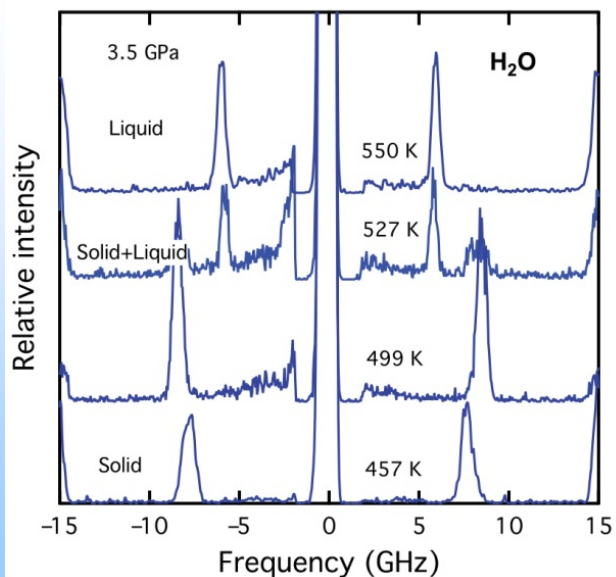


Improved measurement of high P - T melting

[Ahart et al., *High Pressure Res.* (2014)]



High P - T Brillouin Scattering

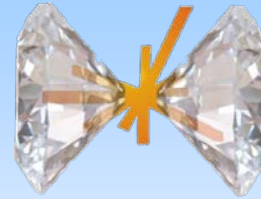


- Clear melting signature
- v_s (eos) for high P - T liquid
- High P - T relations, structures, dynamics?

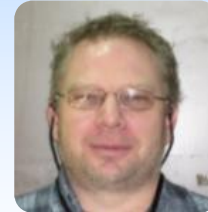
Neutron scattering at extreme P - T conditions

- Underutilized (compare x-ray)
- Low- Z cross sections (e.g, D)
- Extensively used at low P - T
- Previous limit 27 GPa
- Transitions at higher pressure

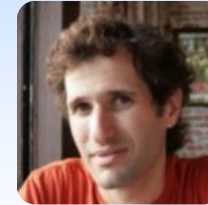
[Guthrie, *Ph.D. thesis* (2002)]



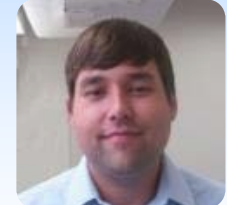
SNAP: Spallation Neutrons at Pressure



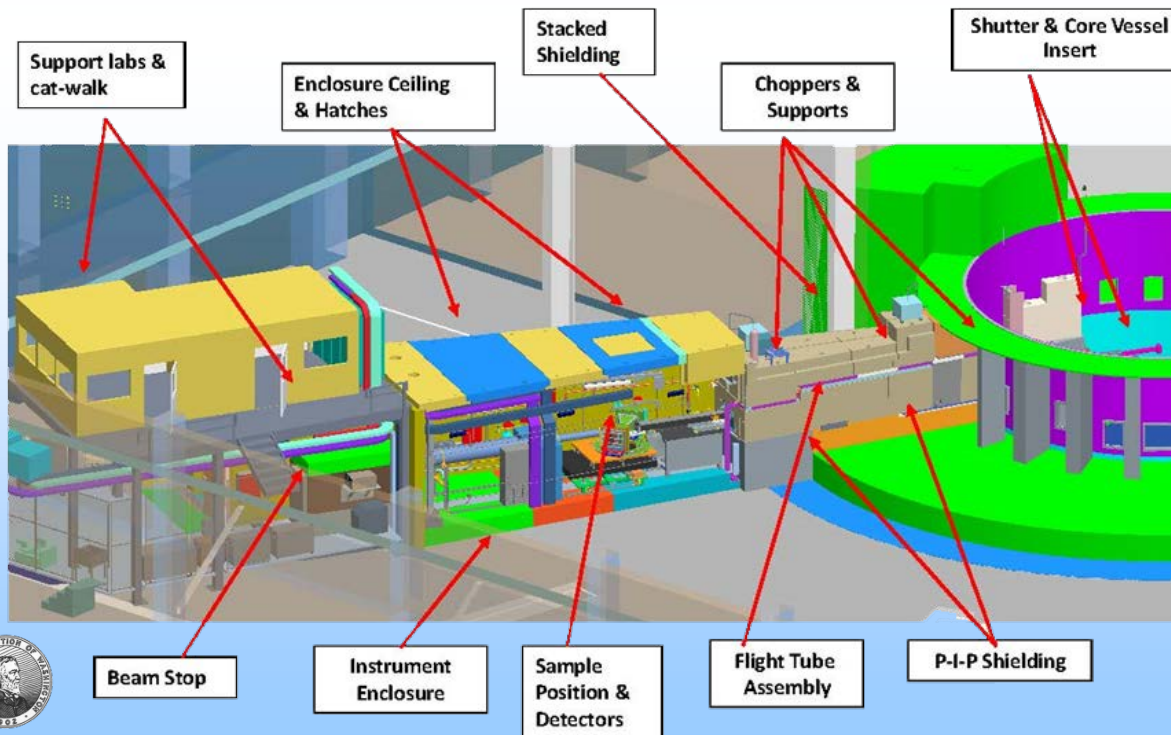
C. A. Tulk



A. M. dos Santos



J. Moliason



Spallation Neutron Source

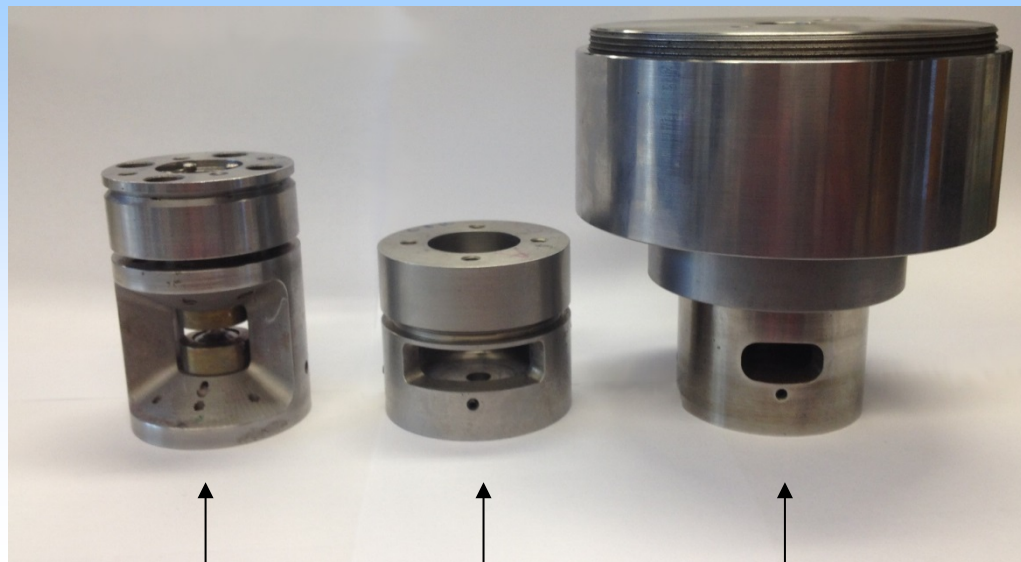
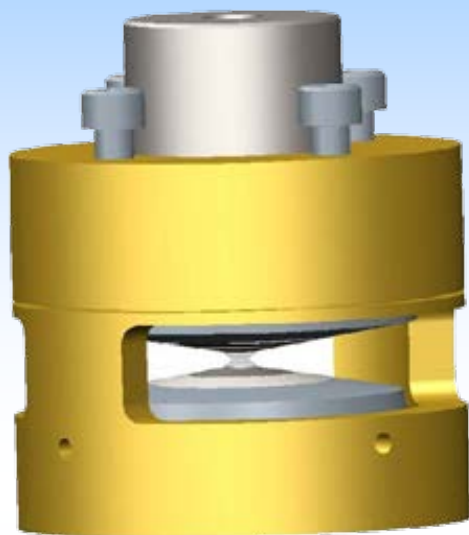
SNAP's dedicated high pressure diffractometer, came online 2006

Highly versatile: can study single crystals, powders, liquids



New cell designs of high-pressure neutron scattering

Enhancing sample volume



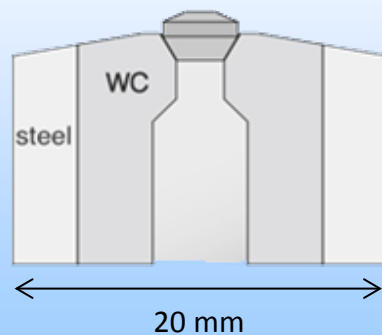
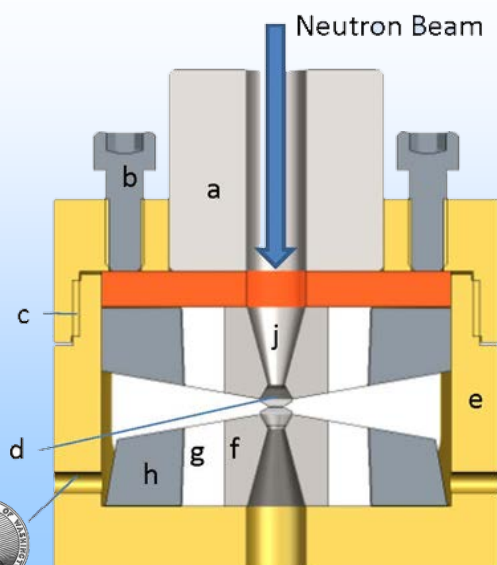
2008

2013

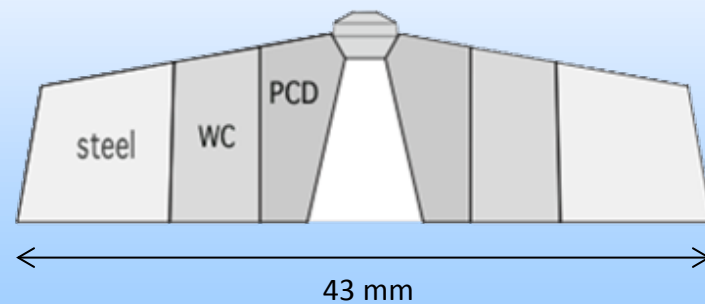
2014



R. Boehler



20 mm

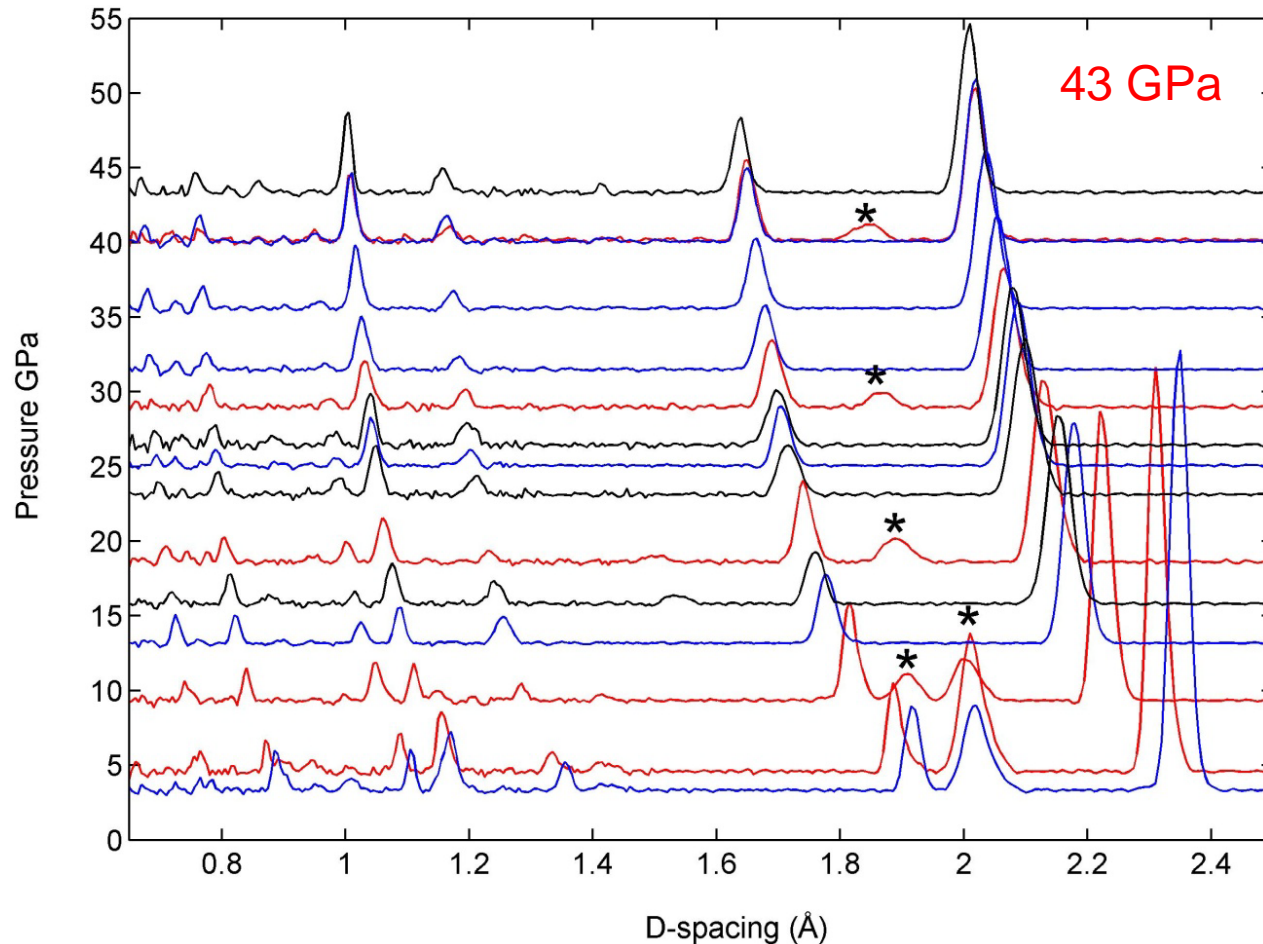
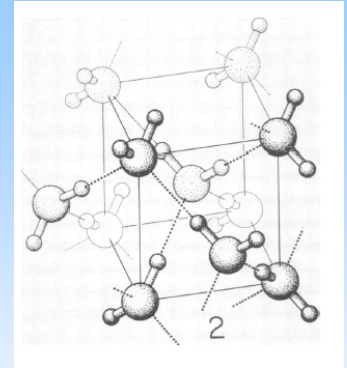


43 mm

[Boehler et al., *High Pressure Res.* (2013)]



Neutron diffraction of ice VII (300 K)



0.05 mm³
D₂O ice VII
(6 hour
datasets)

~50 ug
sample!

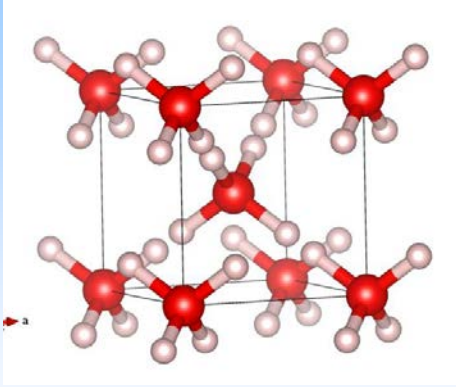
5.3 GPa

- Excellent S/N to 50 GPa
- Shifts in positions and intensities

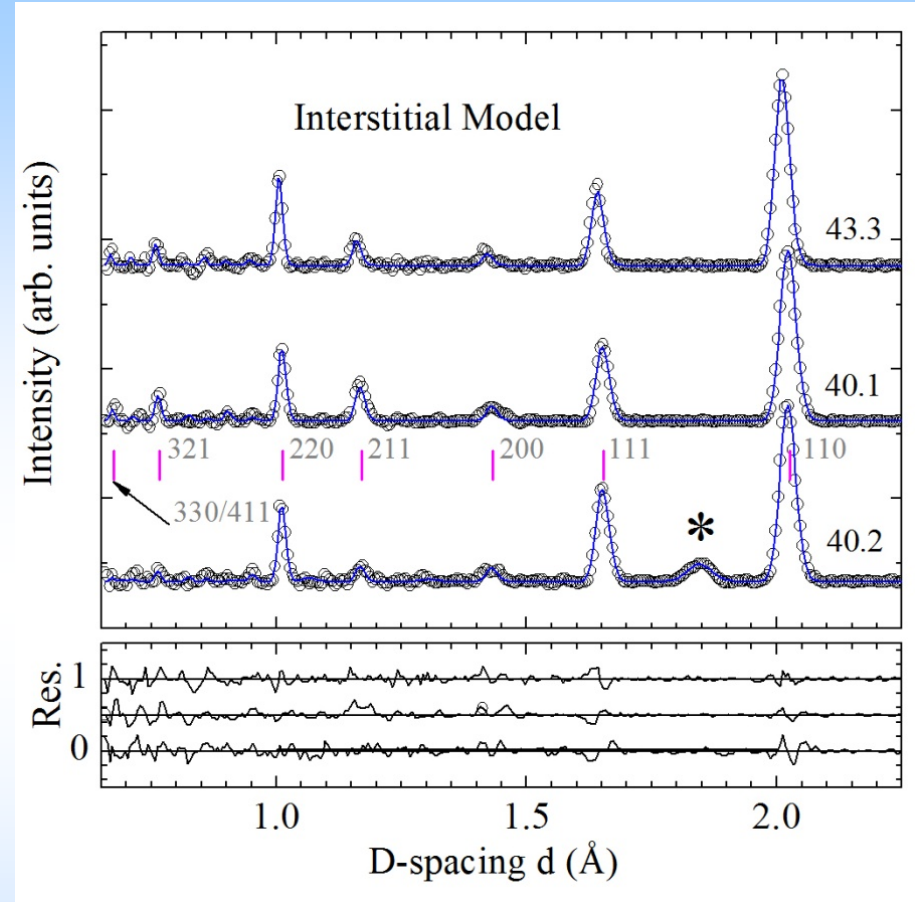
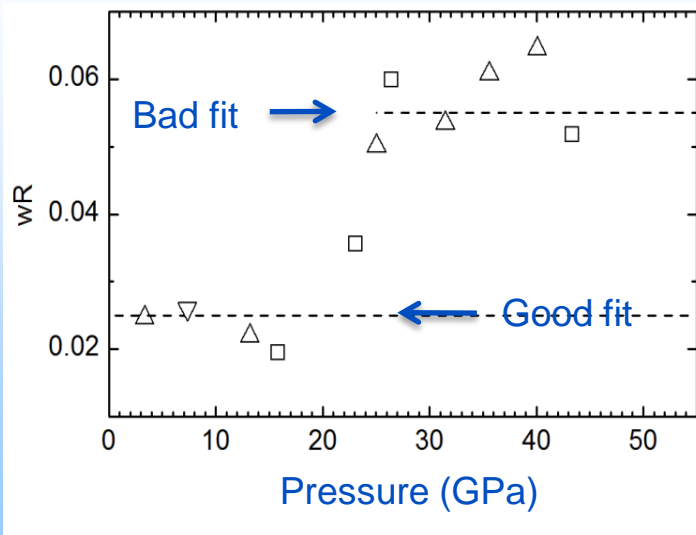
[Guthrie et al., *PNAS* (2013)]



Failure of the conventional model for 'proton centering'

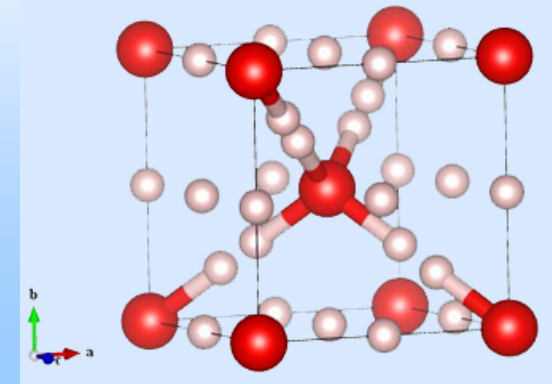


- Rietveld refinement
- Low pressure data agree well with previous work
- Abrupt reduction in quality of fit above 25

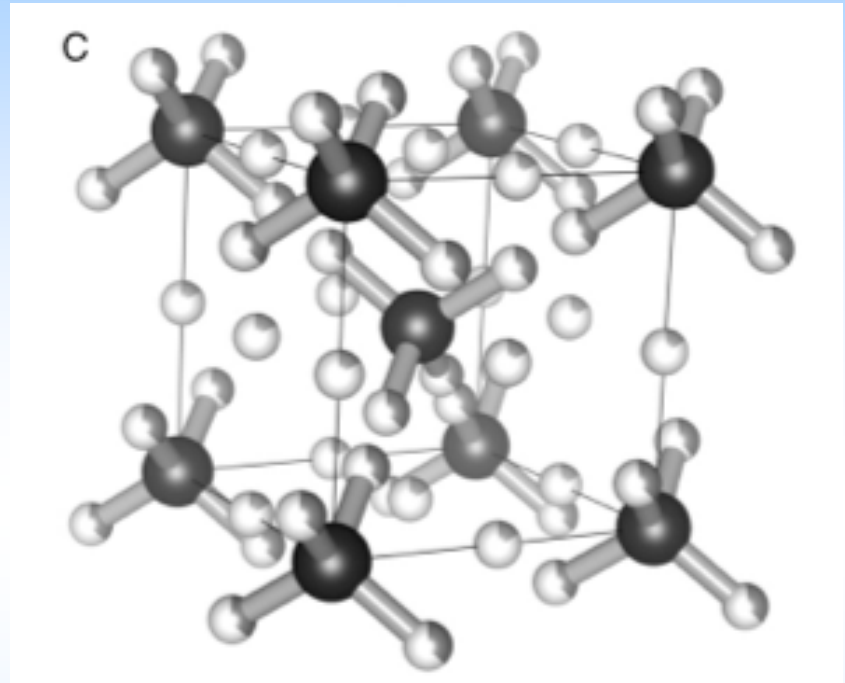
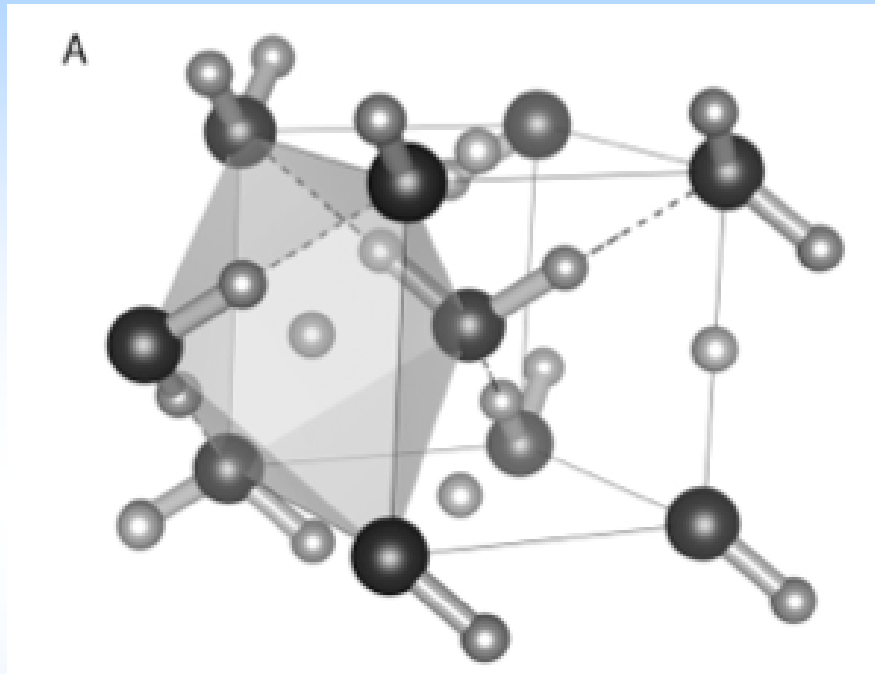


Observed intensities consistent with scattering density in the octahedral voids of O lattice

[Guthrie et al., *PNAS* (2013)]



Interstitial protons in ice VII above 30 GPa (300 K)

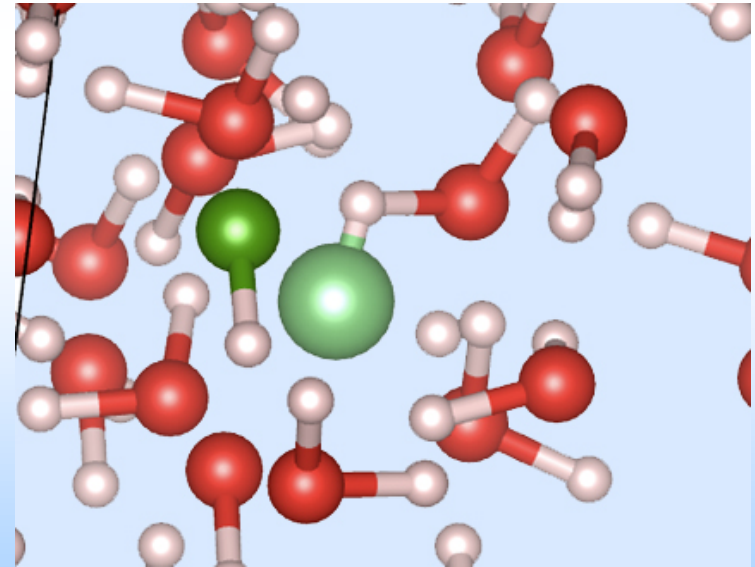
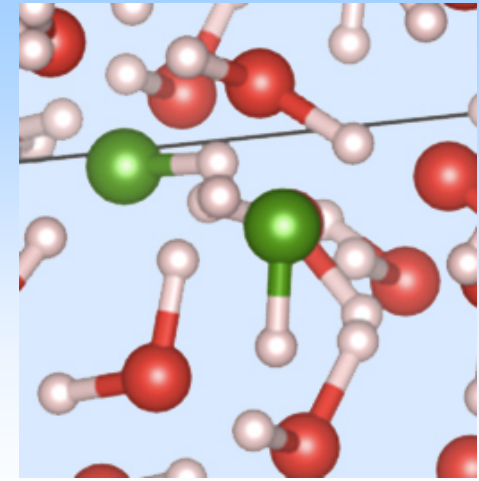
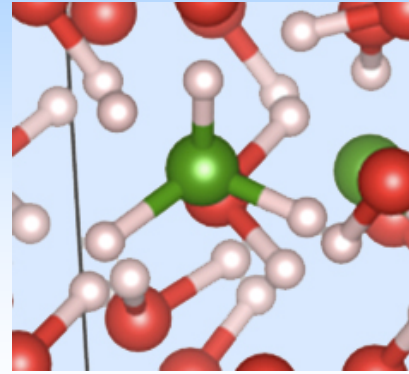
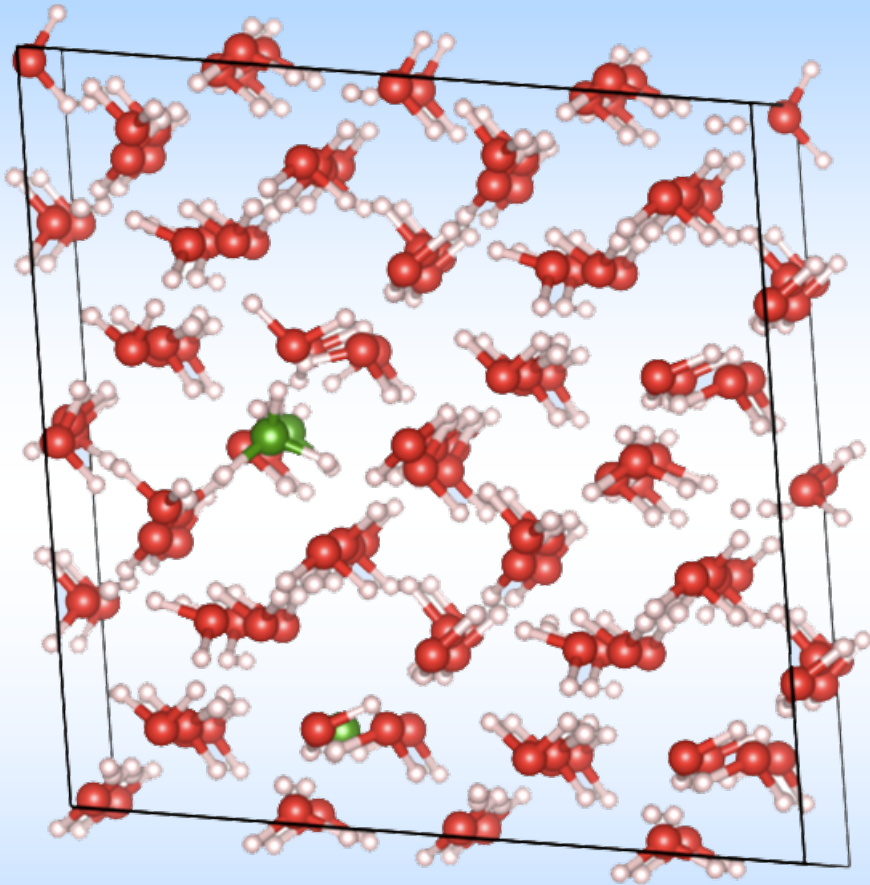


[Guthrie *et al.* PNAS (2013)]

- Not reproduced by DFT calculations (classical nuclei)
- Quantum diffusion? (path integrals?)
- Improved theory or other models?

Partial dissociation to form additional defects

MD DFT-GGA (50 GPa, 300 K)



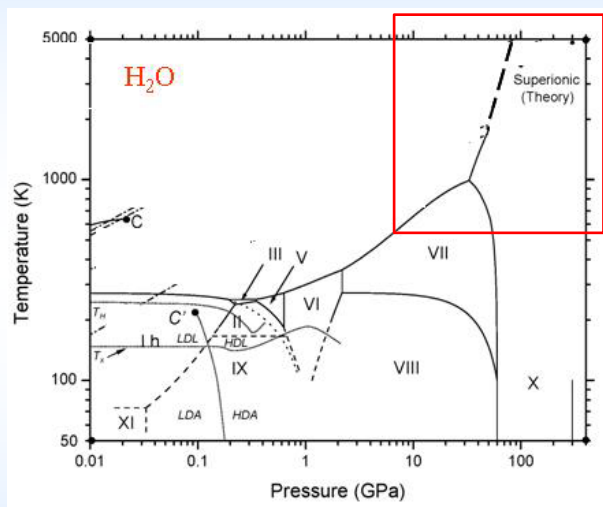
- Partial breakdown of H_2O
- $\text{H}_3\text{O}\dots\text{OH}$ defects
- H-H-O-H molecules

[Caracas et al. *in preparation*]



Interstitial protons and superionicity

(Meta)stability of interstitial H defects at high temperatures



**Need confirmation
by neutron scattering**

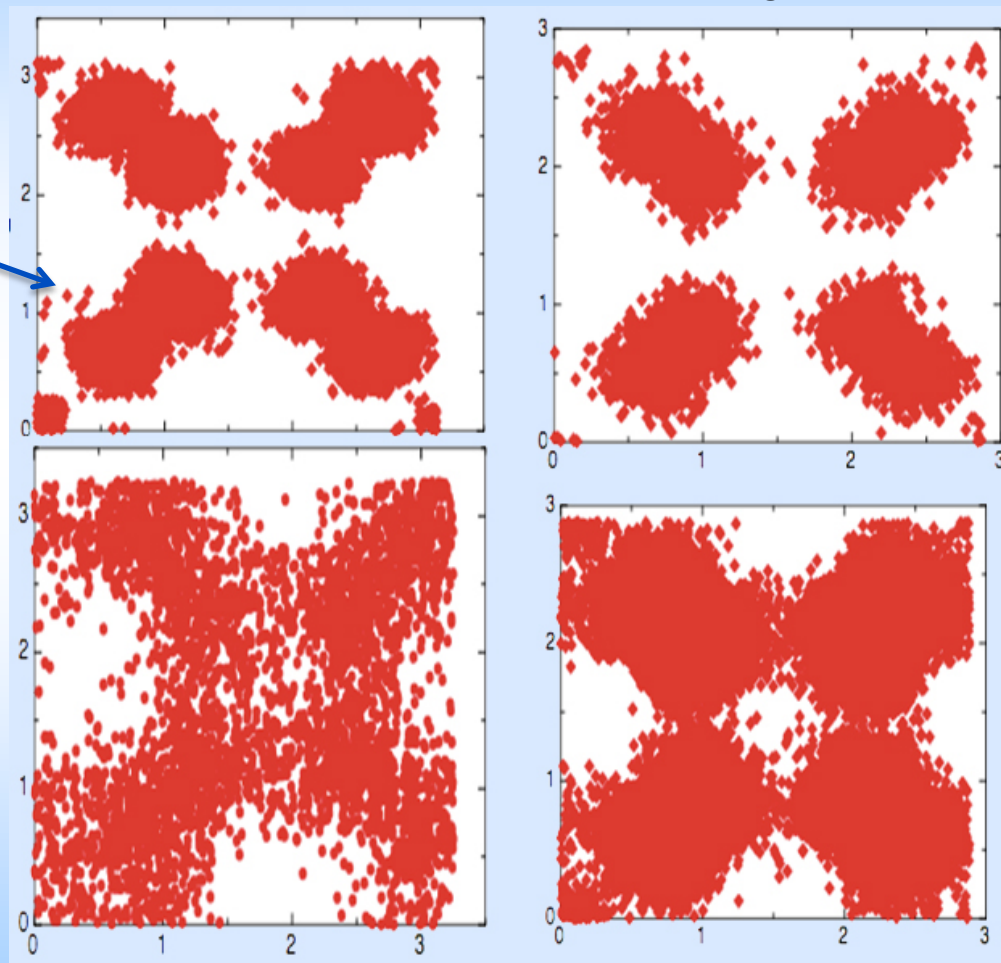
300 K:
Interstitial
H defect ?

1000 K:

Melt/
Super-
ionic?

low-P

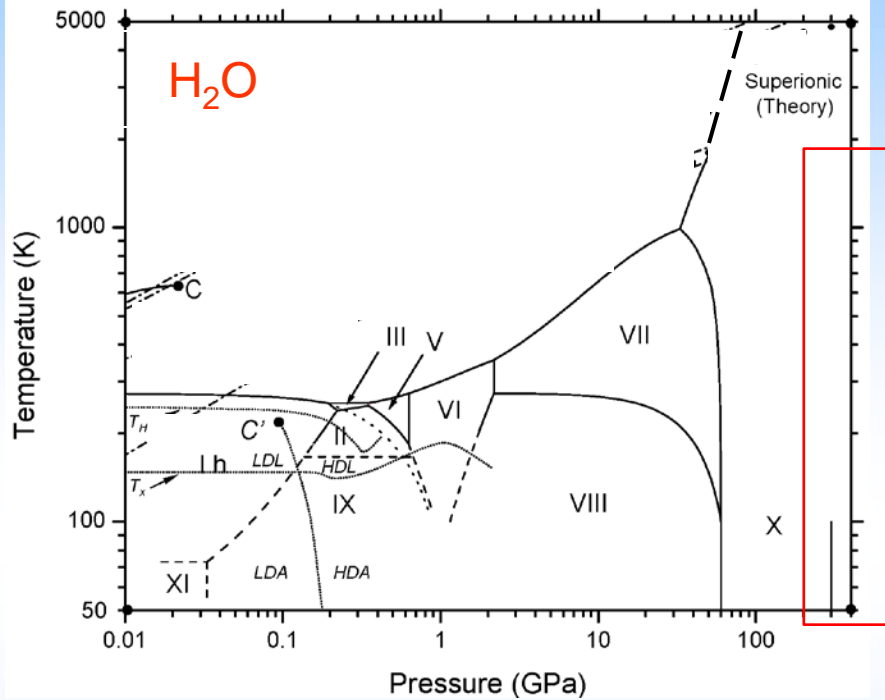
high-P



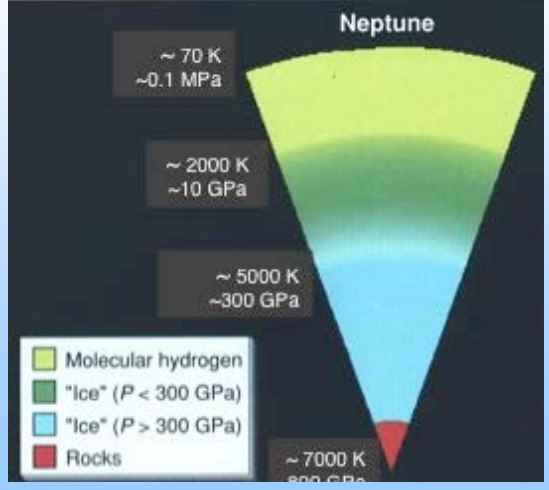
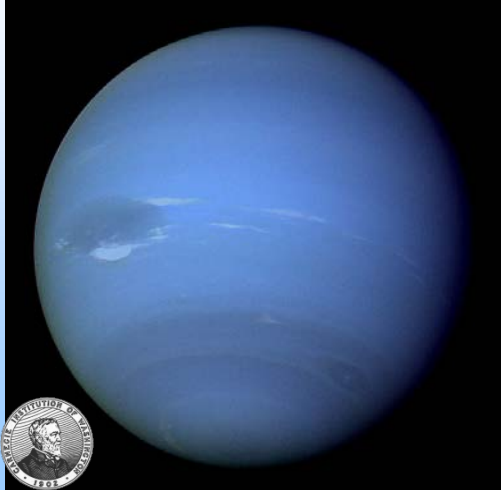
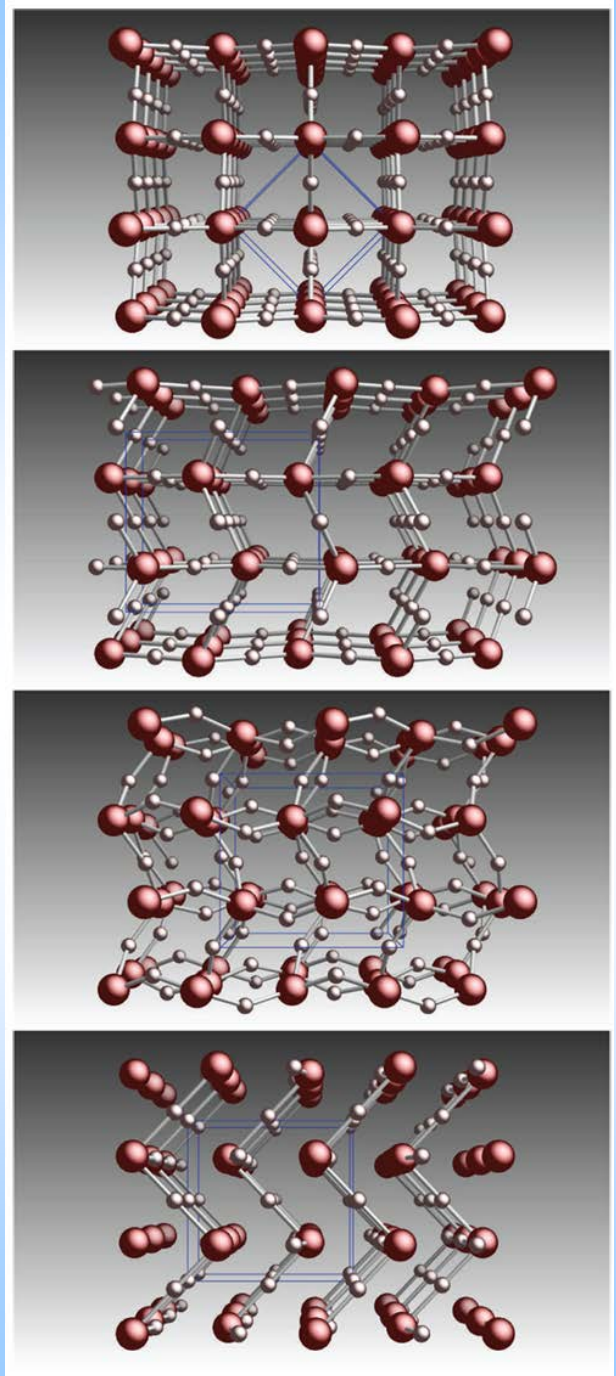
MD DFT-GGA (50 GPa, 300 K)

[Caracas et al. *in preparation*]

Ultrahigh-pressure phases of H₂O: *theoretical predictions*



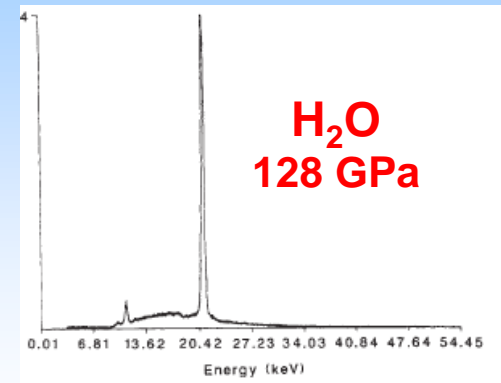
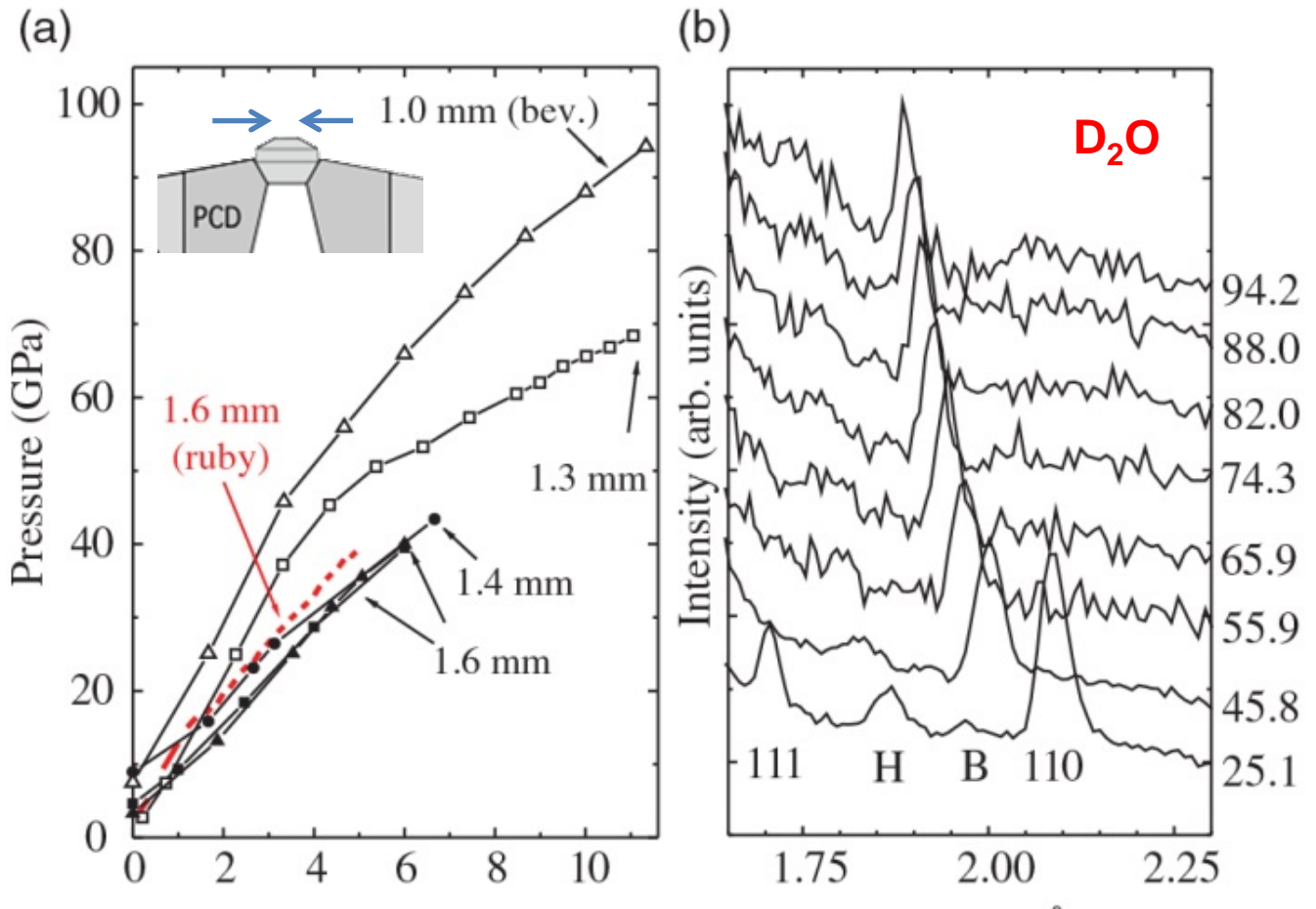
[Militzer et al. *Phys. Rev. Lett.* (2010); see also Caracas, *ibid.* (2008); Hermann et al., *PNAS* (2012)]



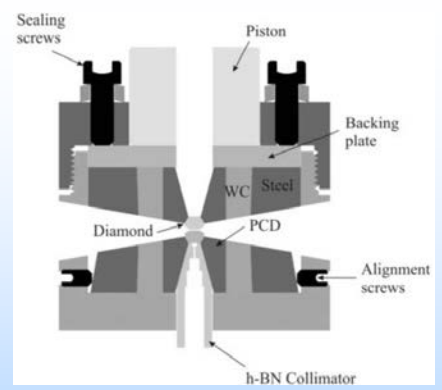
Neutron diffraction at megabar pressures



X-ray H₂O



[Hemley et al., *Nature* (1987)]



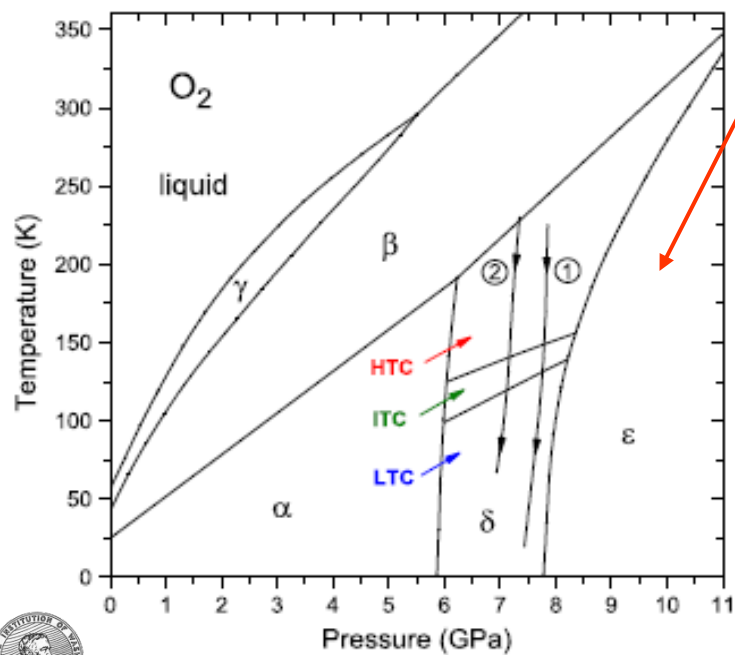
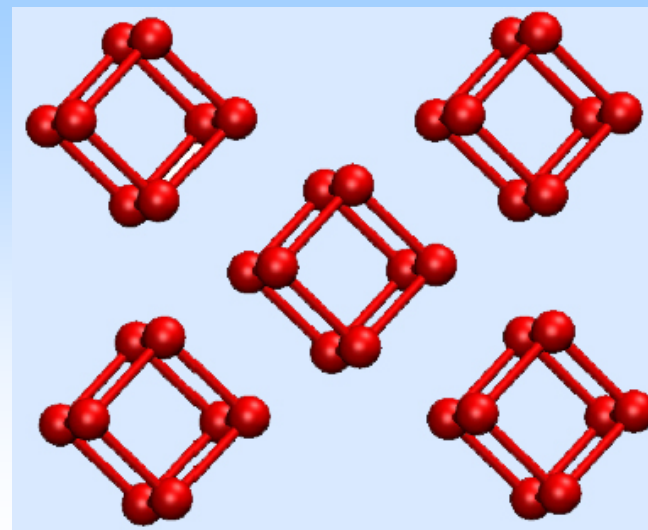
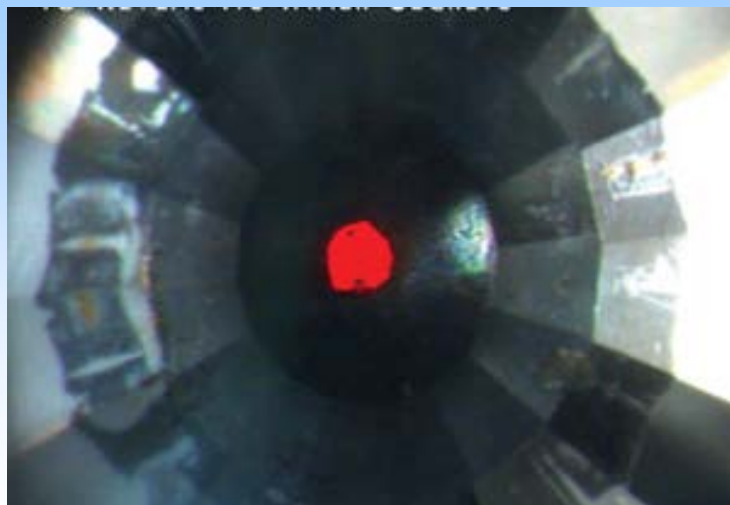
Increase sample volume by x100 at 100 GPa
(10^{-4} mm^3 to $\sim 2 \times 10^{-2} \text{ mm}^3$)

[Boehler et al., *High Pressure Res.* (2013)]

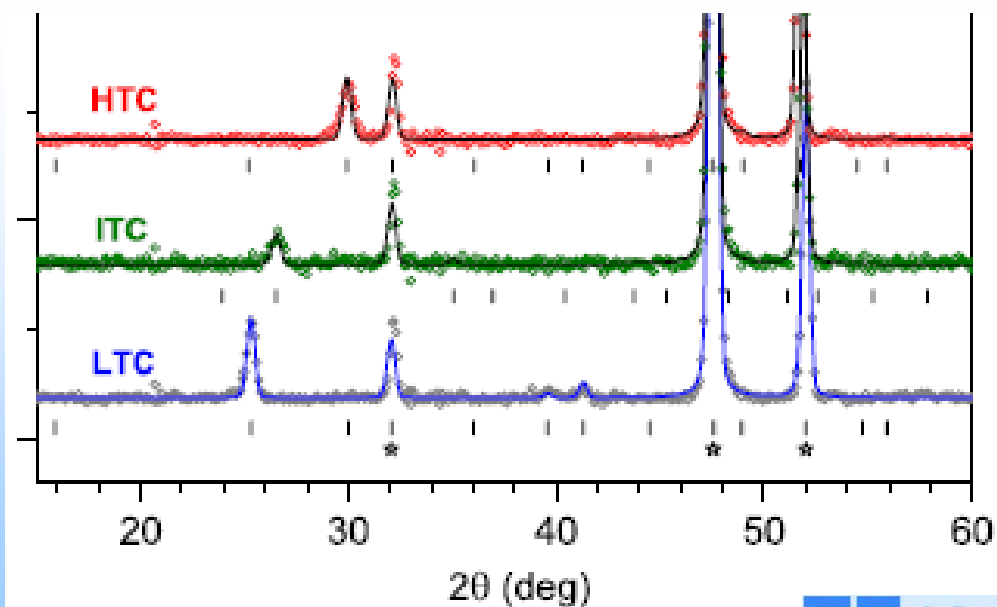


Oxygen under pressure

ϵ -oxygen: $(O_2)_4$ clusters (>8 GPa)

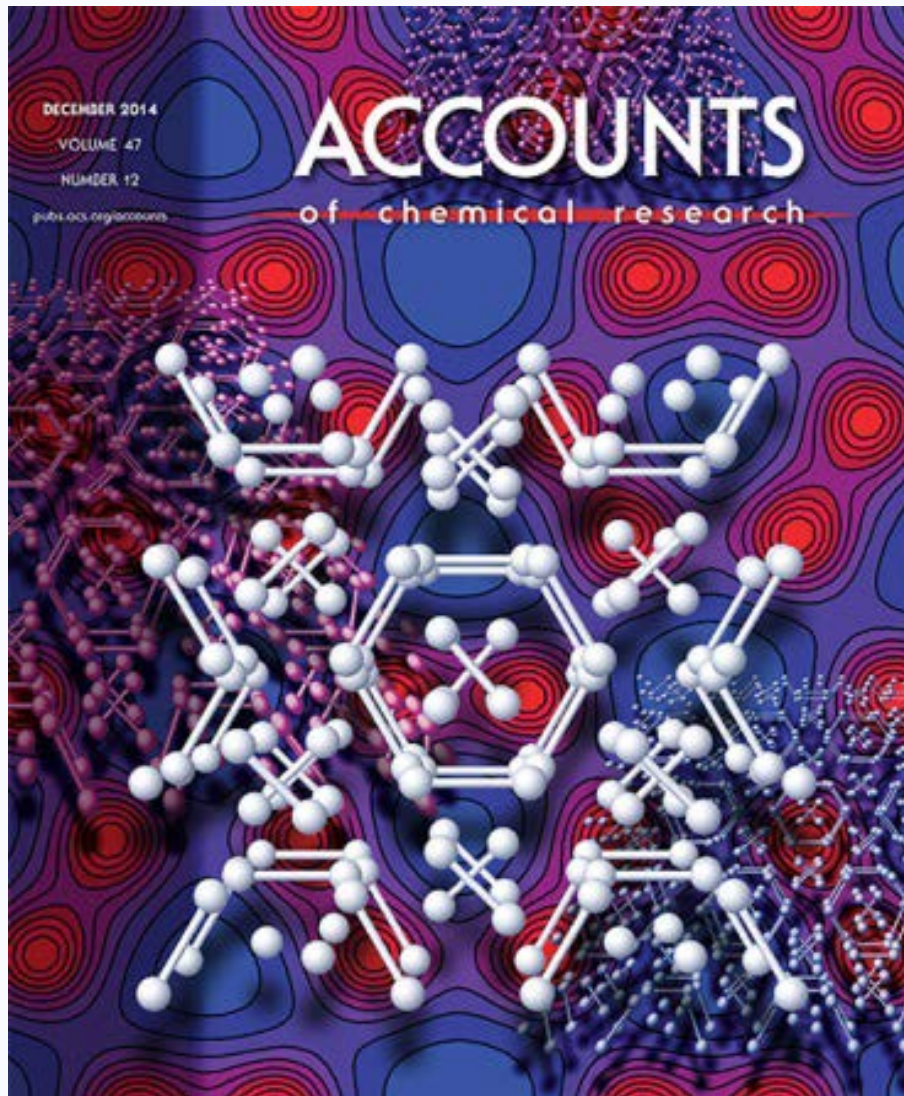


Neutron scattering shows magnetic collapse

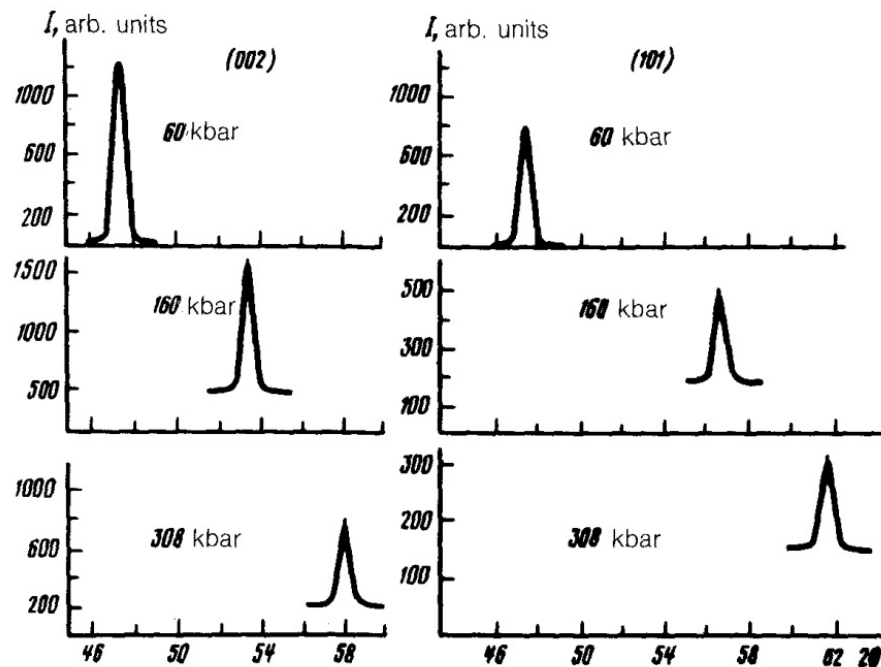


[Klotz et al., *Phys. Rev. Lett.* (2010)]

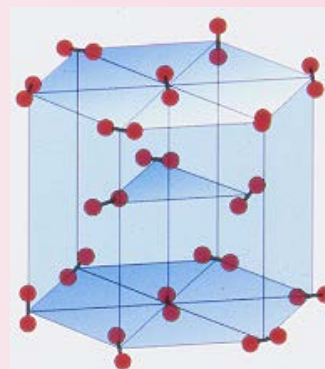
Hydrogen under pressure



Neutron diffraction of D_2 to 30 GPa (phase I)



[Glazkov et al., *JETP Lett.* (1988)]



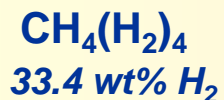
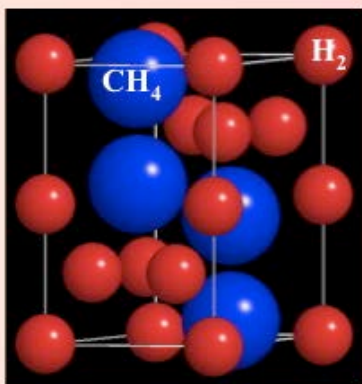
*Rotational
disordered hcp*

'Graphenic' hydrogen at 230 GPa (phase IV)

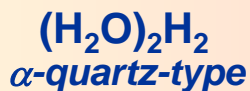
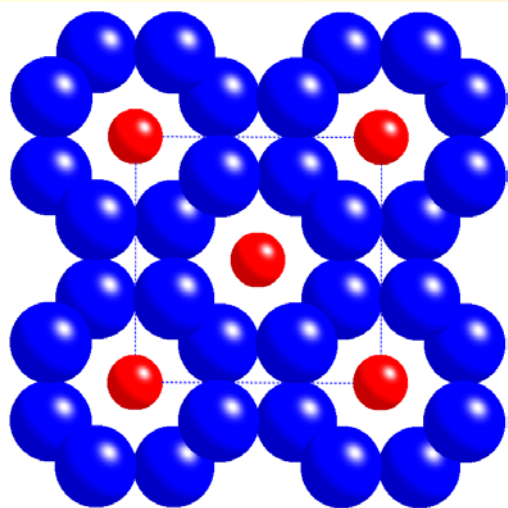
[Naumov & Hemley, *Accts. Chem. Res.* (2014)]



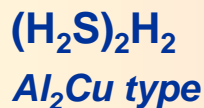
Novel Dense Molecular Compounds



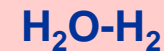
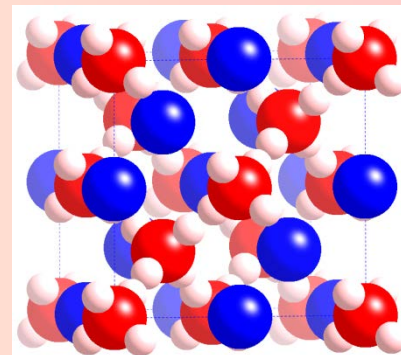
[Somayazulu et al., *Science* (1996);
W. Mao et al. *Chem. Phys. Lett.* (2005)]



[Strobel et al.,
J. Phys. Chem.
(2011)]

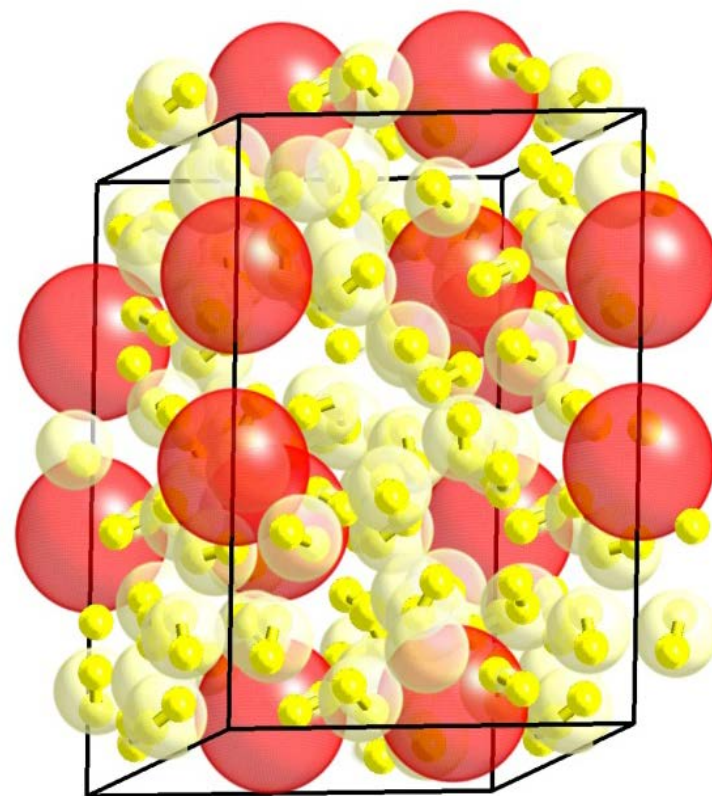


[Strobel et al.,
Phys. Rev. Lett.
(2010)]



11.3 wt% H₂

[Vos et al., *Phys. Rev. Lett.* (1993)]



Insulating to >255 GPa

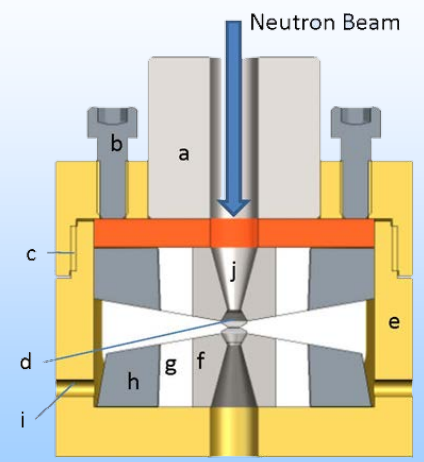
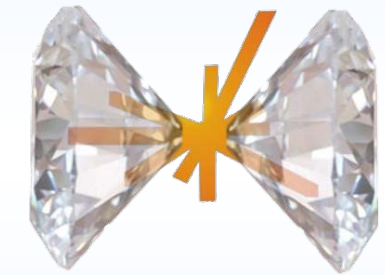
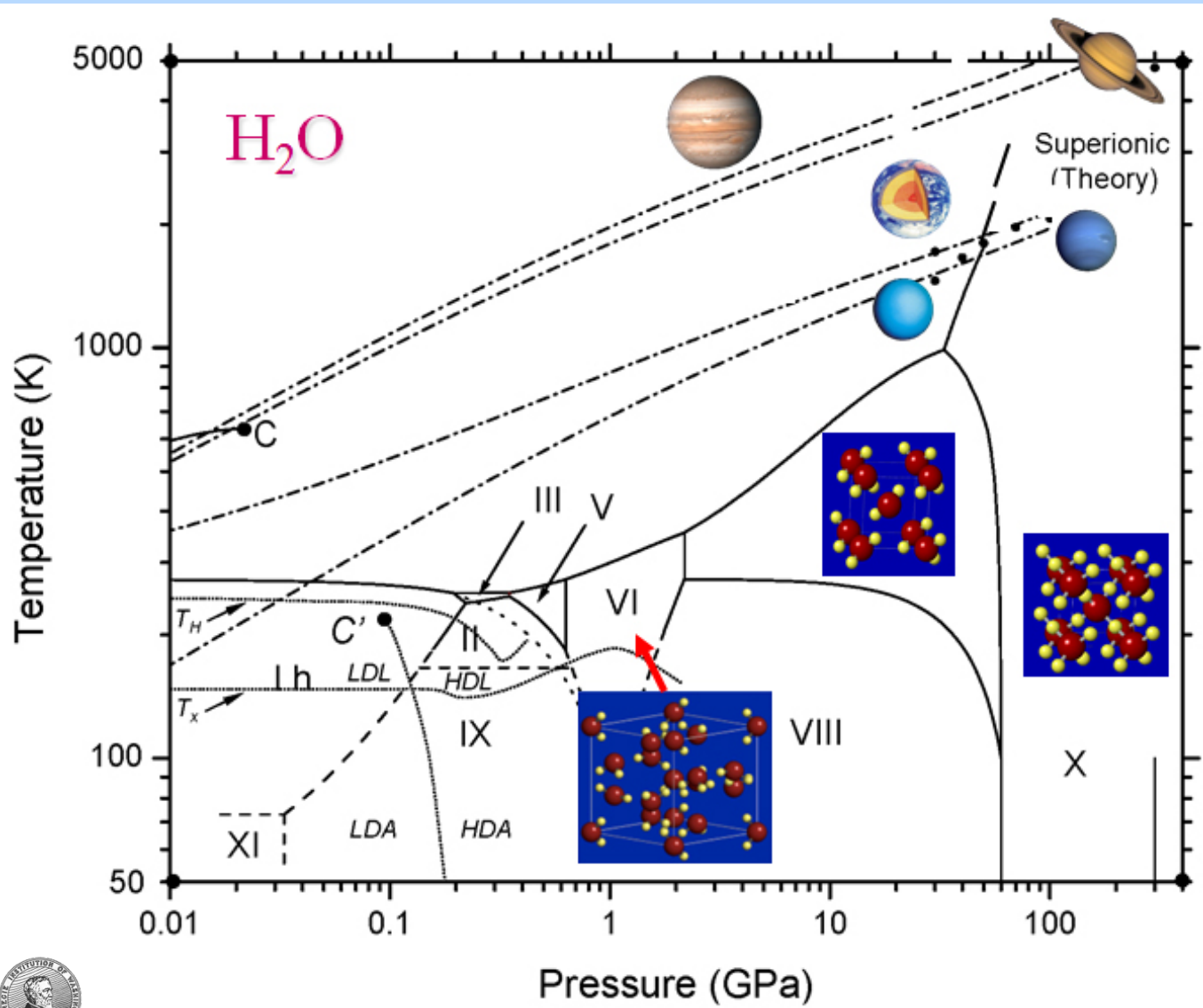
[Somayazulu et al., *Nature Chem.* (2009)]

Higher pressure: Superconductor: T_c = 190 K (!)

[Eremets et al., *to be published*]



A bright future for studies of the behavior of water under extreme conditions



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Collaborators

CARNEGIE INSTITUTION

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Timothy Strobel	Stephen Gramsch
Zhenxian Liu	Changsheng Zha
Yingwei Fei	Kuo Li
Zhenxian Liu	S. Sinogeikin
S. Michida	

OTHER INSTITUTIONS

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Jamie Molaison (ORNL)
A.M. dos Santos (ORNL)
N. Pradhan (ORNL)
Razvan Caracas (Lyon)
Neil Ashcroft (Cornell)
Roald Hoffman (Cornell)
A. Hermann (Cornell)

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CDAC

