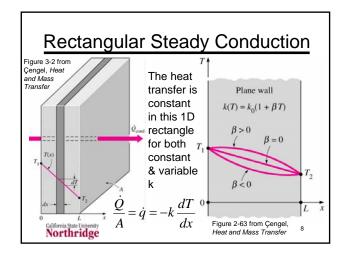
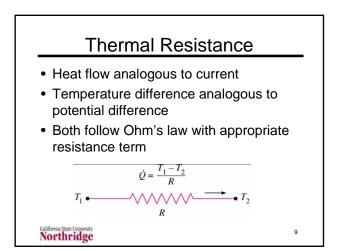


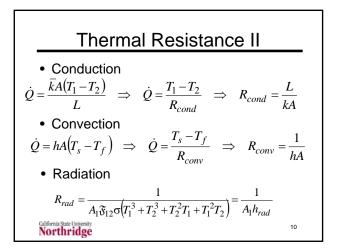
Steady Heat Transfer Definition

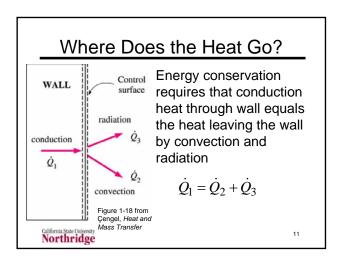
- In steady heat transfer the temperature and heat flux at any coordinate point do not change with time
- Both temperature and heat transfer can change with spatial locations, but not with time
- Steady energy balance (first law of thermodynamics) means that heat in plus heat generated equals heat out

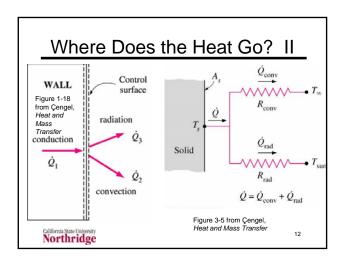
California State University Northridge

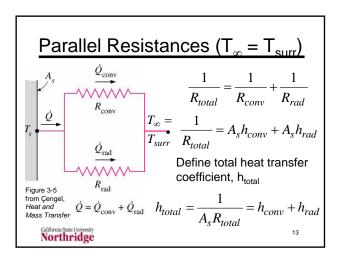


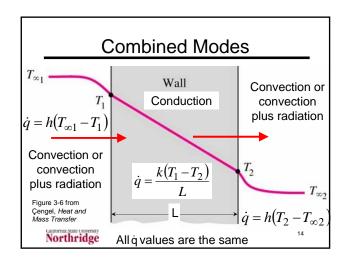


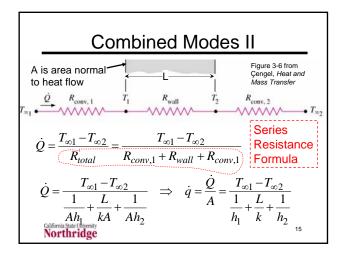


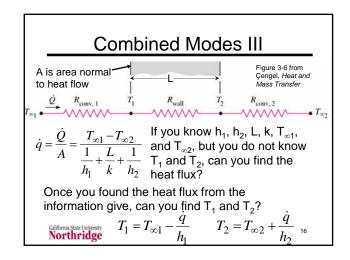


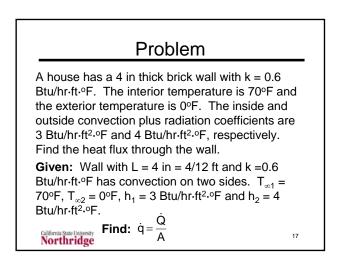


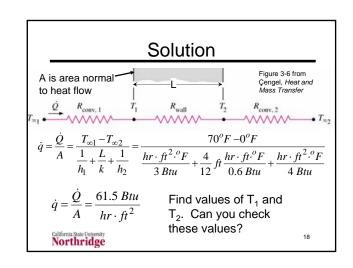


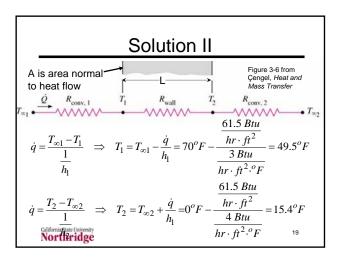


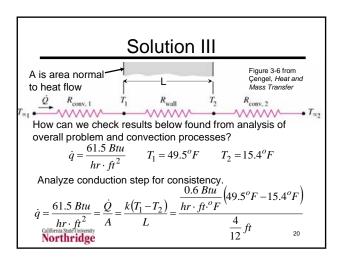


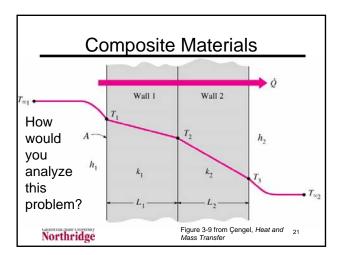


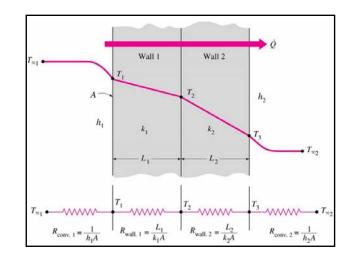


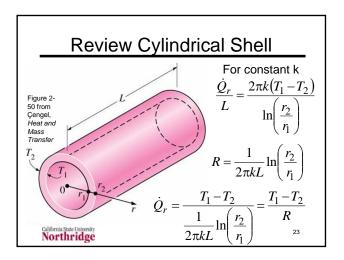


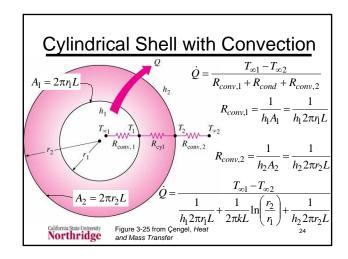


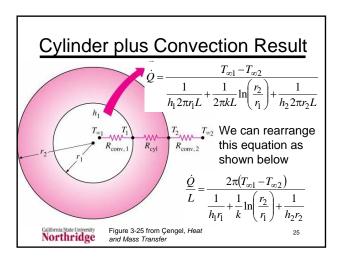




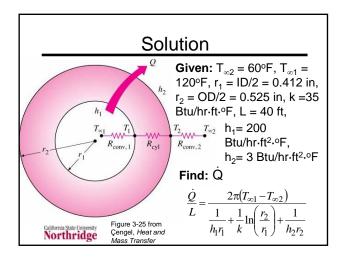


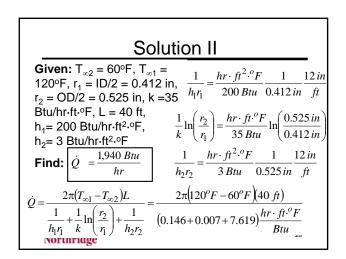


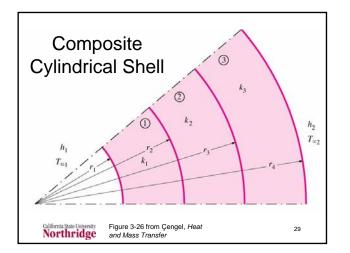


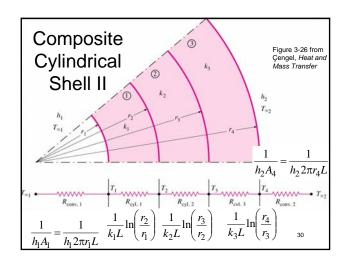


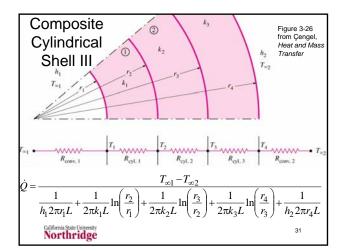
Problem
 A hot-water pipe (k = 35 Btu/hr·ft·°F) in a house, made of ³⁄₄ inch schedule 40 pipe (OD = 1.050 in; ID = 0.824 in) is 40 ft long and contains water at 120°F. The air around the pipe is at 60°F. The heat transfer coefficients inside and outside the pipe are, respectively, 200 and 3 Btu/hr·ft²·°F. Determine the heat loss
from the pipe. $\frac{\dot{Q}}{L} = \frac{2\pi(T_{\infty_1} - T_{\infty_2})}{\frac{1}{h_1r_1} + \frac{1}{k}\ln\left(\frac{r_2}{r_1}\right) + \frac{1}{h_2r_2}} {}_{26}$

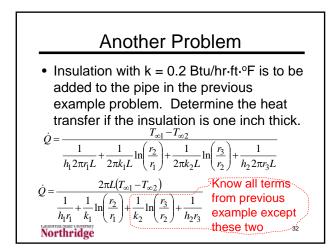


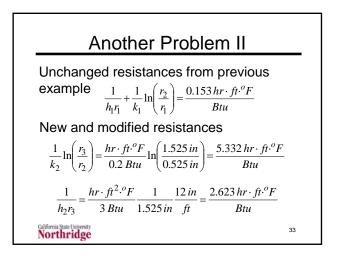


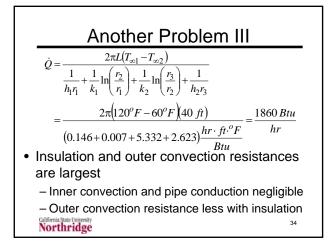


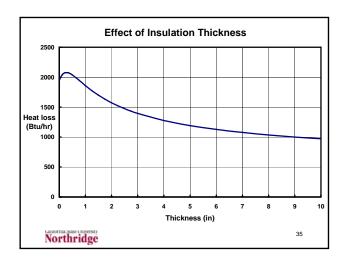


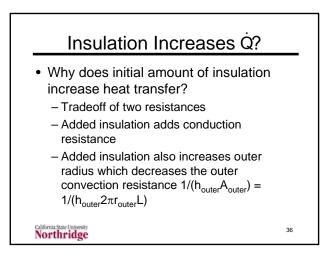


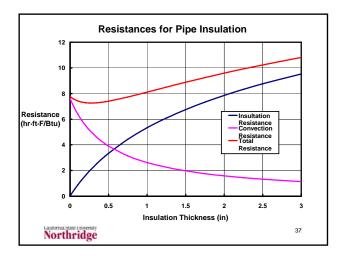


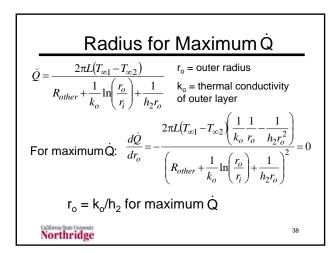


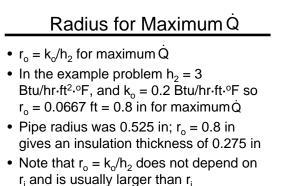












There is no radius for minimum Q
 Morthridge

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