Summer School "Molecular Methods in Energy Research" Wissenschaftspark Gelsenkirchen 2014

Lecture: Introduction to the ORCA Program System

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The Computational Environment

In order to run calculations with **ORCA**, three things are necessary:

1.Installation of the ORCA program

2.Using a text-editor to specify the calculation details, i. e. the *input file*

3.Running the ORCA program (in a cluster environment possibly controlled by a batch system)

NOTE: **ORCA** is available for all popular platforms:

- ★ Windows,
- ★ MacOS,
- ★ Linux

The Computational Environment

ORCA is available for the popular Windows, Mac OS X, and Linux platforms.

★ ORCA is distributed as an archive for all platforms. In this archive resides a directory with all executables. There are plenty of programs for extracting files from archives on all platforms (gzip, tar, zip/unzip,...).

★ On the following slides it will be shown how to run a calculation on Windows Vista/7, Mac OS X, and Linux.

★ It is assumed, that the orca executables reside in a directory 'orca'

Configuring **ORCA** under Windows Vista / 7

Setting the **PATH variable** to the **ORCA** directory (with administrator rights)

🔾 🗢 🛃 🕨 Control Panel 🕨 S	lystem	▼ 4y Search		
Tasks Image: Device Manager Image: Remote settings Image: System protection Image: Advanced system settings	View basic informat Windows edition Windows Vista™ Busin Copyright © 2007 Mic Service Pack 2 Upgrade Windows Vist	ion about your computer ress rosoft Corporation. All rights reserved.	Edit System Variable	The Path Variable can be accessed via Control Panel >
See also Windows Update Security Center Performance	System Rating: Processor: Memory (RAM): System type: Computer name, domai Computer name: Full computer name Computer descriptio Workgroup: Windows activation	System Properties Computer Name Hardware Advanced System Protection Remote You must be logged on as an Administrator to make most of these changes. Performance Visual effects, processor scheduling, memory usage, and vitual memory Settings. User Profiles Desktop settings related to your logon Settings. Startup and Recovery System startup, system failure, and debugging information Settings. OK Cancel Apply	Environment Variables User variables for frankw Variable Variable TEMP %USERPROFILE % \AppData \Local\Temp TMP %USERPROFILE % \AppData \Local\Temp New Edit Delete System variables Variable Variable NuMBER_OF_P 1 0S MMBER_OF_P 1 0S Windows_NT Path COM; EVE; BAT; CMD; VBS; VBE; JS; New Edit Delete	Advanced system > Advanced system Settings Here it is assumed that ORCA resides in c:\orca

Configuring ORCA under Windows Vista / 7

Setting the **PATH variable** to the **ORCA** directory (as normal user)

Tasks Device Manager	View basic informat	ion about your computer	Edit System V	ariable
 Remote settings System protection Advanced system settings 	Windows Vista™ Busin Copyright © 2007 Mic Service Pack 2 Upgrade Windows Vis	ess rosoft Corporation. All rights reserved.	Variable name Variable value	:: Path :: :::\orco;%SystemRoot%\system32;%Syste OK Cancel
	System Rating: Processor: Memory (RAM):	System Properties Computer Name Hardware Advanced System Protection Remote You must be logged on as an Administrator to make most of these change Performance Visual effects, processor scheduling, memory usage, and vitual memory	Environment Varia es. User variables fo Variable TEMP	ables 23 or frankw Value %USERPROFILE%\AppData\Local\Temp
See also	System type: Computer name, domai Computer name: Full computer name	User Profiles Desktop settings related to your logon Settings	System variables	%USERPROFILE%\AppData\Local\Temp New Edit Delete
Windows Update Security Center Performance	Workgroup: Windows activation	Startup and Recovery System startup, system failure, and debugging information Settings Environment Variables.	Variable NUMBER_OF_F OS Path PATHEXT	Value

The Path Variable can be accessed via

Control Panel > User Accounts and Family Safety > User Accounts > Change my environment variables

Here it is assumed that ORCA resides in **c:\orca**

Editing Textfiles under Windows Vista / 7

Editing Text using Notepad



Running ORCA under Windows Vista / 7

Opening a command window:





then cmd.exe will be executed upon pressing **RETURN**.

Running **ORCA** under Windows Vista / 7

Executing the ORCA program



Once **ORCA** is in the PATH, you can run jobs from any directory you want

Using the sequence orca myinput.inp > myinput.out'

the output will be put in a file named like that.

The TaskManager

is a handy tool to check if the calculation is still running.



Windows Task Man

Getting Started with **ORCA** under Mac OS X

TextEdit & Terminal

🕷 Grab File Edit Capture Window Help 🔢 🗿 🕴 🔿 🜒 💶 💽 (96%) Thu 9:43 Q ALC: NO. 000 Applications \bigcirc MacHD 4 1 = = ⊙ ☆ -Q **V DEVICES** 000 .bashrc MacHD Disk 🗖 export PATH=\$HOME/orca:\$PATH **V PLACES** TextEdit Stickies System Preferences T Desktop frankw Documents Time Machine Utilities VMware Fusion **V SEARCH FOR** 1 of 39 selected, 420.59 GB available M Utilities 000 4 1 ⊙ ☆ -Q **V DEVICES** MacHD IDisk **Keychain Access** Migration Assistant Network Utility sagnix% orca myinput.inp >& myinput.out & **V PLACES** top Desktop frankw Applications C Documents ODBC Administrator Podcast Capture **RAID Utility V SEARCH FOR** • Today Sesterday Remote Install Mac OS X System Profiler Past Week Terminal All Images All Movies 40 Х All Documents X11 VoiceOver Utility 1 of 26 selected, 420.59 G8 available ¥ 🛞 🔕 🥝 의 💆 🖏 II 🔘 🐼 🕊 💋 √ 🗤 🔄 🍺 Co 🐣 🛅 X 🖿 沙 🕅 📩 🔓 🌍

Configuring **ORCA** under Mac OS X

TextEdit & Terminal



ORCA path entered in the file

'.bashrc'
Or **'.bash_profile**'

using TextEdit

Running **ORCA** under Mac OS X



Configuring ORCA under Linux

Linux (Ubuntu 9.10): Setting the PATH variable



Running ORCA under Linux

Linux (Ubuntu 9.10): Text Editor (gedit) & ORCA execution

🛟 Applications Places System 🥹 🗟 🕢	🗉 🛒 🗐 Thu Aug 6, 11:08 AM 🛛 Frank Wennmohs 🕑
simple.inp (~/simple) - gedit	
<u>File Edit View Search Tools D</u> ocuments <u>H</u> elp	
New Open Save Print Undo Redo Cut Copy Paste Find Replace	Especially
simple.inp 🛛	under Linux
! RHF SVP OPT	
*xyz 0 1 frankw@MyUbunt	u: ~/simple IS an
<u>File Edit View Terminal Help</u>	
frankw@MyUbuntu:~/simple\$ orca simple.inp >& si	mple.out & number of
frankw@MyUbuntu:~/simple\$ tail -f simple.out	text-editors.
	Verv popular
	ones are
	vim, Emacs,
	NEdit, Kate ,
	etc.
🔟 🗉 frankw@MyUbuntu: ~/ 📝 simple.inp (~/simple)	

The **ORCA** Project



Philosophy of the ORCA Project

General goal: Create a powerful tool to allow connection between theory and experiment. \rightarrow Observables! (Spectra, Structures, Energies)

Design principles:

- 1. The program should be as flexible as possible
- 2. The program should be as efficient as possible (parallel, efficient algorithms, BLAS libraries)
- 3. The program should be as comprehensive as possible
- 4. The program should be as user friendly as possible
- 5. The program should be easily extendable (highly modular)
- The source code should be as clean and well structured as possible (C ++ rather than Fortran).
- 7. The program should be platform independent

Tasks to be Performed

- Calculation of single point energies
 - Hartree-Fock calculations: RHF, UHF and ROHF
 - Density functional calculations
- Optimization of molecular structures
 - Equilibrium geometries
 - Transition states and reaction rates
- Calculation of vibrational frequencies
 - Characterization of stationary points
 - Thermodynamic properties
 - Vibrational spectra
- Calculation of ground state properties
 - Looking at charge distributions and orbitals
 - IR+Raman spectra
 - NMR spectra
 - EPR spectra and exchange couplings
 - Mössbauer spectra
- Calculation of excited states and their properties
 - DFT calculation of absorption and CD spectra
 - Ab initio calculation of absorption and CD spectra
 - Advanced: Resonance-Raman, X-Ray absorption, forbidden transitions, MCD, Excited state geometry optimizations, ...

Common Jobs in Computational Chemistry

- Energy Calculations
- Geometry Optimizations
- Frequency Calculations
- Transition States

Common Jobs in Computational Chemistry

Energy Calculations

•Geome90%izations Of all Galculations!

Transition States

First Stop: Energy Calculations

A First ORCA Job



Output of Single Points

We first echo the input file and some references to the basis sets used. Then you get information on the job-type, the input coordinates in various formats and the basis set

CARTES	SIAN	COORDINA	TES	(ANGSTRO	EM)
c	0	.000000	0.	000000	0.000000
0	0	.000000	0.	000000	1.130000

BASIS SET INFORMATION

There are 2 groups of distinct atoms

Group	1 Type C	:	7s4p1d cont	racted to	3s2p1d	pattern	$\{511/31/1\}$
Group	2 Type O	:	7s4p1d cont	racted to	3s2p1d	pattern	${511/31/1}$

Next the one-electron integrals are calculated (and perhaps also the twoelectron integrals if **conv** if requested for "conventional SCF")

ORCA GTO INTEGRAL CALCULATION

BASIS SET STATISTICS AND STARTUP INFO

<pre># of primitive gaussian shells</pre>			24
# of primitive gaussian function	ıs		48
<pre># of contracted shell</pre>			12
<pre># of contracted basis functions</pre>			28
Highest angular momentum			2
Maximum contraction depth			5
Integral threshhold	Thresh	1	.000e-010
Primitive cut-off	TCut	3	.000e-012

Next the SCF program is taking over and commences with giving all details about the SCF settings

SCF SETTINGS			
Hamiltonian:			
Density Functional	Method		DFT (GTOs)
Exchange Functional	Exchange		B88
X-Alpha parameter	XAlpha		0.666667
Becke's b parameter	XBeta		0.004200
Correlation Functional	Correlation		LYP
LDA part of GGA corr.	LDAOpt		VWN-5
Gradients option	PostSCFGGA		off
Hybrid DFT is turned or	n		
Fraction HF Exchange	ScalHFX		0.200000
Scaling of DF-GGA-X	ScalDFX		0.720000
Scaling of DF-GGA-C	ScalDFC		0.810000
General Settings:			
Integral files	IntName		JOB-01
Hartree-Fock type	HFTvp		RHF
Total Charge	Charge		0
Multiplicity	Mult		1
Number of Electrons	NEL		14
Basis Dimension	Dim		28
Nuclear Repulsion	ENuc		22.4778902655 Eh
Convergence Tolerance:			
Energy Change	TolE		1.000e-008 Eh
Max Density Change	TolMaxP		1.000e-007
RMS Density Change	TolRMSP		1.000e-008
DIIS Error	TolErr		1.000e-007
Diagonalization of the o	overlap matrix:		
Smallest eigenvalue	-		1.911e-002
Time for diagonalization	n		0.010 sec
Time for construction of	f square roots		0.030 sec
Total time needed	······	•	0.040 sec
		•	

The integration grid is produced and the initial guess performed

DFT GRID GENERATION

General Integration Accuracy	IntAcc	 4.010	
Total number of grid points		 5057	
INITIAL GUESS: MODEL POTENTIAL			
Loading Hartree-Fock densities		done	
Now organizing SCF variables		done	
INITIAL (GUESS DONE		

Now we are ready to start the SCF iterations

SCF ITERATIONS

	-		
		! ITERATION	0
Motol Enorgy			
IOLAI Energy	•	-112.951951547451	E11
Energy Change	:	-112.951951547431	Eh
MAX-DP	:	0.674569966353	
RMS-DP	:	0.053899776162	
Actual Damping	:	0.0000	
Int. Num. El.	:	14.00002071 (UP=	7.00
Exchange	:	-10.98287769	
Correlation	:	-0.58429246	
DIIS-Error	:	0.450135783168	

Which hopefully eventually converge and:

****	***************************************	***
*	SUCCESS	*
*	SCF CONVERGED AFTER 11 CYCLES	*

The total energy and its components are printed

TOTAL SCF ENERGY

_	_	_	_	_	_	_	_	_	_	_	_	_	_	_	_

Total Energy :	-113.17273851 Eh	-3079.50944 eV
Components:		
Nuclear Repulsion :	22.47789027 Eh	611.63913 eV
Electronic Energy :	-135.65062877 Eh	-3691.14856 eV
One Electron Energy:	-198.09397475 Eh	-5390.27572 eV
Two Electron Energy:	62.44334598 Eh	1699.12715 eV
Virial components:		
Potential Energy :	-225.62136955 Eh	-6139.31540 eV
Kinetic Energy :	112.44863104 Eh	3059.80596 eV
Virial Ratio :	2.00643945	

Next you get the orbital energies printed:

ORBITAL ENERGIES

. . .

NO	OCC	E(Eh)	E(eV)
0	2.0000	-19.243280	-523.6231
1	2.0000	-10.301611	-280.3141
2	2.0000	-1.152619	-31.3636
3	2.0000	-0.568518	-15.4698
4	2.0000	-0.476766	-12.9731
5	2.0000	-0.476766	-12.9731
6	2.0000	-0.373742	-10.1698
7	0.0000	-0.026298	-0.7156
8	0.0000	-0.026298	-0.7156
9	0.0000	0.221613	6.0303
10	0.0000	0.400346	10.8937
11	0.0000	0.457384	12.4458
12	0.0000	0.457384	12.4458

Now comes the "soft science" (e.g. population analysis of the SCF density matrix

MULLIKEN A	TOMIC CH	IARGES		
0 C : 1 O : Sum of ato	0.0166 -0.0166 mic char	 569 569 cges: -0.0	000000	
MULLIKEN R	EDUCED (ORBITAL CHAR	GES	
0 C s	:	3.834569	s:	3.834569
pz	:	1.009723	р:	2.073008
px	:	0.531642		
ру	:	0.531642		
dz2	:	0.027088	d :	0.075754
dxz	:	0.024333		
dyz	:	0.024333		
dx2y	2 :	0.00000		
dxy	:	0.00000		
10 s	:	3.737862	s:	3.737862
pz	:	1.381818	р:	4.256855
рх	:	1.437518	-	
- py	:	1.437518		
dz2	:	0.008940	d :	0.021953
dxz	:	0.006506		
dyz	:	0.006506		
dx2v	2 :	0.00000		
dxv	:	0.00000		
-				

MULLIKEN OVERLAP CHARGES

B(0-C , 1-O) : 1.3340

The Löwdin analysis contains a detailed breakdown of the MOs in terms of angular momentum components of each atom. This helps a lot when you select the orbitals for plotting.

LOEWDIN ATOMIC CHARGES

0 C : -0.050462 1 O : 0.050462

etc.

LOEWDIN REDUCED ORBITAL POPULATIONS PER MO

THRESHOLD FOR PRINTING IS 0.1%

		0	1	2	3	4	5
ORB-E	N	-19.24328	-10.30161	-1.15262	-0.56852	-0.47677	-0.47677
occ		2.00000	2.00000	2.00000	2.00000	2.00000	2.00000
0 C	s	0.0	99.5	18.6	11.5	0.0	0.0
0 C	pz	0.1	0.0	16.7	4.8	0.0	0.0
0 C	px	0.0	0.0	0.0	0.0	6.9	20.9
0 C	py	0.0	0.0	0.0	0.0	20.9	6.9
0 C	dz2	0.1	0.0	2.4	0.0	0.0	0.0
0 C	dxz	0.0	0.0	0.0	0.0	0.5	1.5
0 C	dyz	0.0	0.0	0.0	0.0	1.5	0.5
10	s	99.8	0.1	54.7	20.8	0.0	0.0
10	pz	0.0	0.4	7.3	62.3	0.0	0.0
10	рх	0.0	0.0	0.0	0.0	17.4	52.4
10	ру	0.0	0.0	0.0	0.0	52.4	17.4
10	dz2	0.0	0.1	0.3	0.6	0.0	0.0
10	dxz	0.0	0.0	0.0	0.0	0.1	0.3
10	dyz	0.0	0.0	0.0	0.0	0.3	0.1
		6	7	8	9	10	11
		6 -0.37374	7 -0.02630	8 -0.02630	9 0.22161	10 0.40035	11 0.45738
		6 -0.37374 2.00000	7 -0.02630 0.00000	8 -0.02630 0.00000	9 0.22161 0.00000	10 0.40035 0.00000	11 0.45738 0.00000
		6 -0.37374 2.00000 	7 -0.02630 0.00000	8 -0.02630 0.00000	9 0.22161 0.00000	10 0.40035 0.00000	11 0.45738 0.00000
0 C	s	6 -0.37374 2.00000 49.2	7 -0.02630 0.00000 0.0	8 -0.02630 0.00000 0.0	9 0.22161 0.00000 33.5	10 0.40035 0.00000 60.9	11 0.45738 0.00000 0.0
0 C 0 C	s pz	6 -0.37374 2.00000 49.2 39.3	7 -0.02630 0.00000 0.0 0.0	8 -0.02630 0.00000 0.0 0.0	9 0.22161 0.00000 33.5 46.5	10 0.40035 0.00000 60.9 35.3	11 0.45738 0.00000 0.0 0.0
0 C 0 C 0 C	s pz px	6 -0.37374 2.00000 49.2 39.3 0.0	7 -0.02630 0.00000 0.0 0.0 15.2	8 -0.02630 0.00000 0.0 0.0 54.7	9 0.22161 0.00000 33.5 46.5 0.0	10 0.40035 0.00000 60.9 35.3 0.0	11 0.45738 0.00000 0.0 0.0 15.5
0 C 0 C 0 C 0 C	s pz px py	6 -0.37374 2.00000 49.2 39.3 0.0 0.0	7 -0.02630 0.00000 0.0 0.0 15.2 54.7	8 -0.02630 0.00000 0.0 0.0 54.7 15.2	9 0.22161 0.00000 33.5 46.5 0.0 0.0	10 0.40035 0.00000 60.9 35.3 0.0 0.0	11 0.45738 0.00000 0.0 0.0 15.5 81.4
0 C 0 C 0 C 0 C	s pz px py dz2	6 -0.37374 2.00000 49.2 39.3 0.0 0.0 0.8	7 -0.02630 0.00000 0.0 0.0 15.2 54.7 0.0	8 -0.02630 0.00000 0.0 0.0 54.7 15.2 0.0	9 0.22161 0.00000 33.5 46.5 0.0 0.0 0.0 7.8	10 0.40035 0.00000 60.9 35.3 0.0 0.0 1.2	11 0.45738 0.00000 0.0 0.0 15.5 81.4 0.0
0 C 0 C 0 C 0 C 0 C	s pz px py dz2 dxz	6 -0.37374 2.00000 49.2 39.3 0.0 0.0 0.0 0.8 0.0	7 -0.02630 0.00000 0.0 0.0 15.2 54.7 0.0 0.4	8 -0.02630 0.00000 0.0 0.0 54.7 15.2 0.0 1.5	9 0.22161 0.00000 33.5 46.5 0.0 0.0 7.8 0.0	10 0.40035 0.00000 60.9 35.3 0.0 0.0 1.2 0.0	11 0.45738 0.00000 0.0 0.0 15.5 81.4 0.0 0.1
0 C 0 C 0 C 0 C 0 C 0 C	s pz px py dz2 dxz dyz	6 -0.37374 2.00000 49.2 39.3 0.0 0.0 0.0 0.8 0.0 0.0	7 -0.02630 0.00000 0.0 0.0 15.2 54.7 0.0 0.4 1.5	8 -0.02630 0.00000 0.0 0.0 54.7 15.2 0.0 1.5 0.4	9 0.22161 0.00000 33.5 46.5 0.0 0.0 7.8 0.0 0.0 0.0	10 0.40035 0.00000 60.9 35.3 0.0 0.0 1.2 0.0 0.0	11 0.45738 0.00000 0.0 0.0 15.5 81.4 0.0 0.1 0.3
0 C 0 C 0 C 0 C 0 C 0 C 0 C 1 0	s pz px dz2 dxz dyz s	6 -0.37374 2.00000 49.2 39.3 0.0 0.0 0.8 0.0 0.0 0.0 2.0	$7 \\ -0.02630 \\ 0.00000 \\ \\ 0.0 \\ 0.0 \\ 15.2 \\ 54.7 \\ 0.0 \\ 0.4 \\ 1.5 \\ 0.0 \\$	8 -0.02630 0.00000 0.0 0.0 54.7 15.2 0.0 1.5 0.4 0.0	9 0.22161 0.00000 33.5 46.5 0.0 0.0 7.8 0.0 0.0 7.8 0.0 0.0 12.0	10 0.40035 0.00000 60.9 35.3 0.0 0.0 1.2 0.0 0.0 0.0 0.0	11 0.45738 0.00000 0.0 0.0 15.5 81.4 0.0 0.1 0.3 0.0
0 C 0 C 0 C 0 C 0 C 0 C 1 0 1 0	s pz py dz2 dxz dyz s pz	6 -0.37374 2.00000 49.2 39.3 0.0 0.0 0.0 0.8 0.0 0.0 0.0 2.0 8.8	$\begin{array}{r} 7 \\ -0.02630 \\ 0.00000 \\ \hline \\ 0.0 \\ 0.0 \\ 15.2 \\ 54.7 \\ 0.0 \\ 0.4 \\ 1.5 \\ 0.0 \\ 0.0 \\ 0.0 \\ \end{array}$	8 -0.02630 0.00000 0.0 0.0 54.7 15.2 0.0 1.5 0.4 0.0 0.0	9 0.22161 0.00000 33.5 46.5 0.0 0.0 7.8 0.0 0.0 7.8 0.0 0.0 12.0 0.2	10 0.40035 0.00000 60.9 35.3 0.0 0.0 1.2 0.0 0.0 0.0 0.0 0.6 1.0	11 0.45738 0.00000 0.0 0.0 15.5 81.4 0.0 0.1 0.3 0.0 0.0
0 C 0 C 0 C 0 C 0 C 0 C 1 0 1 0	s pz py dz2 dxz dyz s pz px	6 -0.37374 2.00000 49.2 39.3 0.0 0.0 0.0 0.8 0.0 0.0 0.0 2.0 8.8 0.0	$\begin{array}{c} 7\\ -0.02630\\ 0.00000\\ \hline \\ 0.0\\ 0.0\\ 15.2\\ 54.7\\ 0.0\\ 0.4\\ 1.5\\ 0.0\\ 0.0\\ 5.9\\ \end{array}$	8 -0.02630 0.00000 0.0 0.0 54.7 15.2 0.0 1.5 0.4 0.0 0.0 21.2	9 0.22161 0.00000 33.5 46.5 0.0 0.0 7.8 0.0 0.0 7.8 0.0 0.0 12.0 0.2 0.0	$ \begin{array}{r}10\\0.40035\\0.00000\\\hline\\60.9\\35.3\\0.0\\0.0\\1.2\\0.0\\0.0\\1.2\\0.0\\0.0\\1.2\\0.0\\0.0\\0.0\\0.0\\0.0\\0.0\end{array} $	11 0.45738 0.00000 0.0 0.0 15.5 81.4 0.0 0.1 0.3 0.0 0.0 0.3
0 C 0 C 0 C 0 C 0 C 0 C 1 0 1 0 1 0	s pz py dz2 dxz dyz s pz px py	6 -0.37374 2.00000 49.2 39.3 0.0 0.0 0.0 0.8 0.0 0.0 2.0 8.8 0.0 0.0 0.0	$\begin{array}{c} 7\\ -0.02630\\ 0.00000\\ \hline \\ 0.0\\ 0.0\\ 15.2\\ 54.7\\ 0.0\\ 0.4\\ 1.5\\ 0.0\\ 0.0\\ 5.9\\ 21.2 \end{array}$	8 -0.02630 0.00000 0.0 0.0 54.7 15.2 0.0 1.5 0.4 0.0 1.5 0.4 0.0 21.2 5.9	9 0.22161 0.00000 33.5 46.5 0.0 0.0 7.8 0.0 0.0 12.0 0.2 0.0 0.0	$ \begin{array}{r} 10\\ 0.40035\\ 0.00000\\ \hline 60.9\\ 35.3\\ 0.0\\ 0.0\\ 1.2\\ 0.0\\ 0.0\\ 0.6\\ 1.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0$	11 0.45738 0.00000 0.0 0.0 15.5 81.4 0.0 0.1 0.3 0.0 0.0 0.3 1.5
0 C 0 C 0 C 0 C 0 C 0 C 0 C 1 0 1 0 1 0	s pz py dz2 dxz dyz s pz pz px PY dz2	6 -0.37374 2.00000 49.2 39.3 0.0 0.0 0.0 0.8 0.0 0.0 2.0 8.8 0.0 0.0 0.0 0.0	$\begin{array}{r} 7 \\ -0.02630 \\ 0.00000 \\ \hline \\ 0.0 \\ 0.0 \\ 15.2 \\ 54.7 \\ 0.0 \\ 0.4 \\ 1.5 \\ 0.0 \\ 0.0 \\ 5.9 \\ 21.2 \\ 0.0 \end{array}$	$ \begin{array}{r} 8 \\ -0.02630 \\ 0.00000 \\ \hline 0.0 \\ 54.7 \\ 15.2 \\ 0.0 \\ 1.5 \\ 0.4 \\ 0.0 \\ 21.2 \\ 5.9 \\ 0.0 \\ \end{array} $	$\begin{array}{r} 9\\ 0.22161\\ 0.00000\\ \hline \\ 33.5\\ 46.5\\ 0.0\\ 0.0\\ 7.8\\ 0.0\\ 0.0\\ 12.0\\ 0.2\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ $	$ \begin{array}{r} 10\\ 0.40035\\ 0.00000\\ \hline 60.9\\ 35.3\\ 0.0\\ 0.0\\ 1.2\\ 0.0\\ 0.0\\ 0.6\\ 1.0\\ 0.0\\ 0.0\\ 0.0\\ 0.9\\ \end{array} $	11 0.45738 0.00000 0.0 0.0 15.5 81.4 0.0 0.1 0.3 0.0 0.0 0.3 1.5 0.0
0 C 0 C 0 C 0 C 0 C 0 C 0 C 0 C 1 0 1 0 1 0 1 0	s pz py dz2 dxz dyz s pz px py dz2 dxz	$\begin{array}{c} 6\\ -0.37374\\ 2.00000\\\\ 49.2\\ 39.3\\ 0.0\\ 0.0\\ 0.0\\ 0.8\\ 0.0\\ 0.0\\ 2.0\\ 8.8\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0$	$\begin{array}{c} 7\\ -0.02630\\ 0.00000\\ \hline \\ 0.0\\ 15.2\\ 54.7\\ 0.0\\ 0.4\\ 1.5\\ 0.0\\ 0.0\\ 5.9\\ 21.2\\ 0.0\\ 0.2\\ \end{array}$	$ \begin{array}{r} 8 \\ -0.02630 \\ 0.00000 \\ \hline 0.0 \\ 54.7 \\ 15.2 \\ 0.0 \\ 1.5 \\ 0.4 \\ 0.0 \\ 21.2 \\ 5.9 \\ 0.0 \\ 0.9 \\ \end{array} $	$\begin{array}{r} 9\\ 0.22161\\ 0.00000\\ \hline \\ 33.5\\ 46.5\\ 0.0\\ 0.0\\ 7.8\\ 0.0\\ 0.0\\ 12.0\\ 0.2\\ 0.0\\ 0.2\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ $	$ \begin{array}{c} 10\\ 0.40035\\ 0.00000\\ \hline 60.9\\ 35.3\\ 0.0\\ 0.0\\ 1.2\\ 0.0\\ 0.0\\ 0.6\\ 1.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0\\ 0.0$	11 0.45738 0.00000 0.0 0.0 15.5 81.4 0.0 0.1 0.3 0.0 0.1 0.3 1.5 0.0 0.1

Looking at Orbitals

There is a utility program orca_plot which lets you generate graphics information. The information about the MOs, the geometry and the basis set is stored in the so called gbw-File (,geometry-basis-wavefunction').

To generate the plot information interactively use:

orca_plot myjob.gw -i

You will get a "stone-age" menu which you can use to generate the necessary files.

- 1. Press 5 ENTER to choose the output formation. (press 7 ENTER for gaussian cube, the preferred format)
- Press 4 ENTER to choose the number of grid intervals. Something like 40 will be o.k. For high resolution on larger molecules choose 65-75.
- 3. To plot an orbital from a closed shell calculation press 3 ENTER and choose 0 ENTER. For spin-up from UHF/UKS the same. For spin down from UHF/UKS choose 1 ENTER.
- 4. Press 2 ENTER and enter the number of the MO that you want to plot. NOTE THAT COUNTING STARTS WITH 0!!!
- 5. Press 10 ENTER to generate the output file.
- 6. Repeat steps 4 and 5 until all desired files are produced

The orca_plot program automatically produces an ,xyz' file which contains the molecular coordinates.

We presently like the *Chimera* program for vizualization. There are many free alternatives such as *Molekel, gOpenMol, Molden,...* any program that reads .xyz and .cube files

With **Chimera**, do the following:

- 1. Start Chimera
- 2. Choose "File Open" and navigate to open the .xyz file of interest
- 3. Choose "**Presets** → **Publication 1**"
- 4. Choose "Actions → Atoms&Bonds → ball & stick"
- 5. Choose "Actions → Color → By element"
- 6. Choose **"Tools → Volume Data → Volume Viewer**"
- Choose "File → Open Map" and Gaussian Cube format and open the cube file of interest
- Enter in "Level" 0.03 and press ENTER. Then click on Color and choose red (or whatever you like)
- Hold the Ctrl-key and click on the negative part of the contours. Enter in "Level" -0.03 and press ENTER, then go to color and choose yellow (or whatever you like)
- Orient the contour in the way you like and go to "File → Save Image". Choose PovRay "true" and go ahead.



What You Need for an Energy Calculation

Energy calculations make up the simplest computational task available

To actually start a calculation three things have to be determined before:

- Method
- Basis set
- Coordinates

Method: Ab initio wavefunction based or DFT?

Accuracy versus Computing Time Tradeoff

Hartree Fock	Fast, but not accurate enough. Usually only as a basis for higher methods
DFT	Fast, <i>usually</i> good accuracy. But is the accuracy also good in my case? Very dependent on the Functional
Correlated, wavefunction based methods	Very high accuracy, with Coupled Pair Theories up to Chemical Accuracy. BUT: Calculation scales formally at least with the size of the basis set to N ⁵

Method: Ab initio wavefunction based or DFT?

Roughly there are three main branches available:



The Functional Zoo

Name	Туре	Comments
BLYP	GGA	One of the earliest GGA functionals. Usually inferior to BP86 and PBE. Predicts too long bonds.
BP86	GGA	Excellent geometries and vibrational frequencies. Energetics is usually not highly accurate but performs often well in spectroscopic investigations.
PW91	GGA	One of the older GGA functionals with excellent accuracy for exchange couplings.
PBE	GGA	A GGA version designed to replace PW91. Very popular in physics. Often similar to BP86.
OLYP	GGA	Violates the uniform electron gas limit but gives improved results for molecules
B3LYP	Hybrid	De facto standard in chemistry for structures, energies and properties. See discussion in the text.
PBE0	Hybrid	Excellent accuracy; competitive with B3LYP
TPSS	Meta-GGA	Improvement over PBE. Includes the kinetic energy density and obeys more constraints known from rigorous theory.
TPSSh	Hybrid meta- GGA	Probably improvement over PBE0; perhaps increase fraction of HF to 25% (TPSS0)
B2PLYP	Double hybrid	First (and prototypical) member of the double hybrid class of functionals. So far been proven excellent for energies and geometries. More exploration needed.

Efficient DFT: The RI Approximation

As long as there is NO Hartree-Fock exchange present (no hybrid functionals), a very efficient approximation can be used to speed up (factor 10-100) DFT calculations: the **RI approximation** (also called **density fitting** method)

In this method the electron density is fit to an auxiliary basis set which must be provided by the user. The effect of the approximation on structures and frequencies is barely visible. Absolute energies are affected to a few kcal/mol, relative energies much less.



TIP: For geometry and frequency calculations the BP86 and PBE functionals together with the RI approximation is recommended. Basis sets of TZVP quality are appropriate for good accuracy! SV(P) is already good enough for a first orientation. Use keywords **QuickOpt, NormalOpt or GoodOpt**! For <u>energy calculations</u> I recommend the B3LPY or PBE0 functionals and larger basis sets (TZVPP if possible or even aug-TZVPP). Use Keyword **DFTEnergy**

Efficient HF and Hybrid DFT: RIJCOSX

Unfortunately, RI does not smoothly carry over to Hartree-Fock and hybrid DFT calculations. One attempt to do so is the **RI-JK** approximation that needs to be invoked together with ,JK' fitting bases. RI-JK cannot be used for optimizations

! B3LYP RI-JK def2-SVP def2-SVP/JK TightSCF

A more efficient approximation that leads to large speedups is the **RIJCOSX** approximation. it uses **RI-J** for the Coulomb part and a special approximation ,**COSX**' for the exchange part. It is available throughout the program and leads to large speedups at very little loss in accuracy

! B3LYP RIJCOSX def2-SVP def2-SVP/J TightSCF

If combined with **RI-MP2** you need to give two auxiliary basis sets for optimal speed and accuracy:

```
! RI-MP2 RIJCOSX def2-SVP def2-SVP/J def2-SVP/C TightSCF
```

Neese, F.; Wennmohs, F.; Hansen, A.; Becker, U. (2009) Chem. Phys., <u>356</u>, 98–109

The Crux of DFT: Which Functional to Choose?

Always first thing to do:

RI BP86

The BP86 functional in combination with the Resolution of Identity approximation (RI) is the fastest way to calculate by far, with reasonable results with respect to geometries, etc.

For a more quantitative result, for example, the calculation of energy differences, the following hybrid functionals are known to provide good results:

B3LYP RIJCOSX (PBE0 RIJCOSX)

To keep in mind:

Usually the results are good, but this does not rule out the possibility, that it might produce wrong results in other cases!

Modern DFT: Double Hybrid Functionals

- A Special feature of ORCA are ,double hybrid functionals' proposed in 2006 by Stefan Grimme. Here part of the DFT correlation is replaced by a MP2 like term and part of the DFT exchange by Hartree-Fock exchange.
- ★ For main group thermodynamics and kinetics it is more accurate than standard functionals
- ★ ORCA is the only program to have analytic gradients and excited states for these functionals
- ★ The prototypical member is "B2PLYP"

! RI-B2PLYP def2-SVP def2-SVP/C TightSCF Opt

★ It can be combined with RIJCOSX

```
! RI-B2PLYP RIJCOSX def2-SVP def2-SVP/J def2-SVP/C TightSCF Opt
```

★ And together with empirical van der Waals corrections (also proposed by Stefan Grimme) it is particularly accurate:

! RI-B2PLYP D3BJ RIJCOSX def2-SVP def2-SVP/J def2-SVP/C TightSCF Opt

Choice of Basis Set

6-31G 6-31G* 6-31G** 6-31++G** 6-3111++G**

> DZP TZP

Ahlrichs SVP Ahlrichs TZVP

cc-pVDZ cc-pVTZ aug-cc-pVTZ Lots of basis sets to choose from

Which one is the best for one's purpose?

Choice of Basis Set



- Core orbitals are represented by a contraction of six primitive gaussian functions. The valence shell is double- ζ with 3 primitives in first contraction, one in second.
- There is a set of diffuse orbitals (p,d) on heavy atoms as well as on hydrogen atoms (diffuse s-orbitals). There is one additional set of d-functions on heavy atoms and one additional set of p-polarization functions on hydrogen atoms.
- <u>note:</u> Core orbitals have the lowest energy. Thus, an accurate description is necessary (otherwise the variational principle makes the valence orbitals "core like").
 - → Deep contraction for core orbitals, but only one function per AO (because the core-radial function does not change much in the molecule)!

Choice of Basis Set

6-31G 6-31G* 6-31G** 6-31++G** 6-3111++G**

> DZP TZP

Ahlrichs SVP Ahlrichs TZVP

cc-pVDZ cc-pVTZ aug-cc-pVTZ These Pople style basis sets are very popular in the English-speaking community. They are also very popular due to their availability in the Gxx series of programs.

Better:

The Ahlrichs basis sets provide better results at the same computational cost.

Tip:

Always start calculations at the SVP level, especially geometry optimizations. Only when precise energies are requested, switch to TZVP (or even: def2-TZVPP)
In principle:

Only a texteditor is needed to enter a molecule's coordinates

Small molecules can be entered by specification of internal coordinates and their values. This is usually called a

Z-Matrix

But this is only feasible for molecules of a few (<5) atoms.

Today

Molecules are created with full specification in an xyz space using common programs.

Today

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Building molecules using PyMol

Today

Molecules are created with full specification in an xyz space using common programs. (Molden, PyMol, GaussView, etc.)

There are two ways to get coordinates into ORCA, once those have been created and exported into an .xyz file:

- 1. Copy the contents of the .xzy file into the ORCA input file, using copy/ paste facilities of the texteditor or operating system
- 2. Read in the .xyz file via ORCA input, using the *xyzfile command

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Although the first method looks somewhat more cumbersome at first, it usually is the safer way:

By this, all information reside in your input file, reducing the error margin and making the file more easily transportable.

Other Sources of Molecular Structures

Today there are many more sources for acquiring the molecular structure of molecules. It is quite easy, to generate a small organic molecule from scratch. This is definitely not the case with large biomolecular structures like proteins or DNA !

The structures deposited in publicly available databanks might have different origins:

Crystal Structures:

Structures generated by analyzing X-Ray diffraction patterns. Very good resolution with organic/inorganic molecules, varying with large biomolecules

• NMR Structures:

Structures, or usually an ensemble of structures, by applying NMR structural information into the generation of structures.

Very good for accessing structures in solution.

• Theoretical Models

Sometimes there are neither X-Ray nor NMR information available. In that case people try to come up with a theoretical model, which is usually refined by techniques like simulated annealing etc. to make certain the overall structure reaches a state of low energy.

It can be hard to assess the quality of these structures at time. The difficulty of generating good theoretical models gets worse with increasing molecular size.

SCF Convergence

Sooner or later it happens: The SCF cycles does not converge!

What now?

There are serveral reasons causing the SCF not to converge.

- Molecular Structure contains errors
 For example, missing atoms, the electronic state is wrong, etc.
- Very Large Systems Due to the accumulation of numerical noise the SCF might not converge
- Difficult Chemical Bonding Bonding situations containing transition metals might be hard to converge due, especially if the specified state is far away from the ground state



What now?

There are many ways to influence the SCF convergence, for example:

- Damping
- Level Shift
- DIIS
- SOSCF
- Newton Raphson



What now?

There are many ways to influence the SCF convergence, for example:

- Damping
- Level Shift
- DIIS
- SOSCF
- Newton Raphson

First thing to try: Specifying **!slowconv**

Otherwise:

- Increase Level Shift
- Increase DIIS Length

Transition Metal Calculations

For open-shell transition metals convergence of the SCF is often a problem. There are a few ways in ORCA to help this process:



... Actually, this particular job converges without any damping or shifting. However, in many situations, the damping and shifting is necessary as you will undoubtedly find out yourself once you start calculations in the "real world".

In the present example the convergence "aids" acctually strongly slow down convergence. It is quite uniform since the energy is always decreasing. Thus, these "helpers" more or less "babysit" the job to a converged solution.

Restarting Calculations

In many cases it is a very good idea to start from the convergence MOs of a previous job.



Note that the calculation can be started from a gbw file which is from a nearby geometry, uses a different basis set of theoretical method or HFType (UHF,RHF or ROHF). The program will take care to translate the orbitals to the present situation.

Second Stop: Geometry Optimizations

Geometry Optimizations

From a formal point of view:

A geometry optimizations minimizes the overall energy of the system by changing the atomic coordinates.

Minimum: All gradients have to vanish





Problem: Leads to the *nearest* local minimum, not the *global* minimum of the system!

Geometry Optimizations

Things to keep in mind when performing geometry optimizations:

- The SCF wavefunction has to be quite precise to generate a smooth potential surface without 'bumps'
- Geometry optimizations might be tough if the molecule contains many soft potentials, for example dihedrals with very low barriers

Geometry Optimization

To optimize the geometry of the molecule, simply include the keyword Opt

```
! RKS SV(P) B3LYP TightSCF Opt
* int 0 1
C 0 0 0 0 0 0 0
0 1 0 0 1.15 0 0
H 1 2 0 1.00 115 0
H 1 2 3 1.00 115 180
*
```

The program will first produce a set of "redundant internal coordinates" which are used in the calculation.

			Redund	dant Internal Cod	ordinates
Defint				Initial Value	Approx d2E/dq
1. B(O 2. B(H 3. B(H 4. A(H 5. A(H 6. A(H 7. I(O	1,C 2,C 3,C 2,C 3,C 3,C 3,C 1,H	0) 0) 0,0 0,0 0,0 0,H 3,H	1) 1) 2) 2,C	1.1500 1.0000 1.0000 115.0000 115.0000 130.0000 0) 0.0000	1.351281 0.501167 0.501167 0.425466 0.425466 0.323418 0.151694

TIP: Always use TightSCF or VeryTightSCF in geometry optimizations. Otherwise the gradients are somewhat noisy.

After calculating the SCF energy and the gradient of the energy, a And a new geometry is proposed: relaxation is step is carried out:

		(ORCA GEOMETRY	RELAXA	TION ST	TEP	
Number o	of at	coms			4		
Number o	of in	nternal coordin	nates		7		
Current	Enei	rgy			-114.3	317745134	Eh
Current	grad	dient norm			0.2	207887808	Eh/bohr
Maximum	allo	wed component	of the step		0.300		
Current	trus	st radius			0.300		
Evaluati	.ng t	the initial hes	ssian		(Almlo	oef) done	
Projecti	.ng t	the Hessian			done		
Forming	the	augmented Hess	sian		done		
Diagonal	izir	ng the augmente	ed Hessian		done		
Last ele	ement	t of RFO vector	r		0.9579	975075	
Lowest e	eiger	nvalues of augr	mented Hessian	h :			
-0.0571	.747(0.15169387	0.360949845	5 0.42	5465740	0.50116	56791
Length c	of th	ne computed ste	ep		0.2994	435193	
The fina	1 le	ength of the in	nternal step		0.2994	435193	
Converti	.ng t	the step to can	rtesian space:	:			
Transfor	minq	g coordinates:					
Iter	0:	RMS(Cart)=	0.0824560429	RMS (In	it)=	0.1127292	2561
Iter	1:	RMS(Cart)=	0.0024951498	RMS (In	it)=	0.0033197	7589
Iter	2:	RMS(Cart)=	0.0002432861	RMS (In	it)=	0.0003067	7902
Iter	3:	RMS(Cart)=	0.0000185328	RMS (In	it)=	0.0000231	161
Iter	4:	RMS(Cart)=	0.000012690	RMS (In	it)=	0.000015	5779
Iter	5:	RMS(Cart)=	0.000000832	RMS (In	it)=	0.000001	L034
Iter	6:	RMS(Cart)=	0.000000054	RMS (In	it)=	0.000000	067
done							
Storing	new	coordinates			Done		

The status of the geometry convergence is printed:

	Ge	ometry convergen	nce	
Item	value		Tolerance	Converged
RMS gradient	0.078	83145	0.00010000	NO
MAX gradient	0.142	33649	0.00030000	NO
RMS step	0.113	17586	0.00200000	NO
MAX step	0.185	11086	0.00400000	NO
Max(Bonds)	0.0980	Max(Angles)	4.89	
Max(Dihed)	0.00	Max(Improp)	0.00	

The optimization has not yet converged - more geometry cycles are needed

Redundant Internal Coordinates

(Angstroem and degrees)

Defir	nition				Value dE/dq	Step	New-Value	
1 B (0	1 0	•••••			1 1600 0 1422		1 2025	
1. Б(О	1,0	0)			1.1500 -0.1425	56 0.0555	1.2035	
2. B(H	2,C	0)			1.0000 -0.1033	55 0.0980	1.0980	
3. В(Н	3,C	0)			1.0000 -0.1033	55 0.0980	1.0980	
4. A(H	2,C	0,0	1)		115.00 -0.0176	85 2.44	117.44	
5. A(H	3,C	0,0	1)		115.00 -0.0176	85 2.44	117.44	
6. A(H	3,C	О,Н	2)		130.00 0.0353	70 -4.89	125.11	
7. I(O	1,H	З,Н	2,C	0)	0.00 -0.0000	00 0.00	0.00	

GEOMETRY OPTIMIZATION CYCLE 2

Then the next SCF is done and the next gradient calculated, a new geometry is proposed until (hopefully) finally:



Following this statement one more energy calculation is performed in order to make sure that the energy and properties are really done at the stationary point of the PES.

Constraints and Relaxed Scans

You can "freeze" certain geometrical parameters in an optimization:

```
! RKS B3LYP/G SV(P) TightSCF Opt
%geom Constraints
        { B 0 1 1.25 C }
        { A 2 0 3 120.0 C }
        end
      end
* int 0 1
     0 0 0 0.0000
                      0.000
    С
                               0.00
   0 1 0 0 1.2500
                      0.000
                               0.00
   H 1 2 0 1.1075 122.016
                               0.00
   H 1 2 3 1.1075 122.016
                             180.00
*
```

... Or freeze some and vary others (one frequently used possibility is to only optimize hydrogen positions **OptimizeHydrogens true**). Constrained surfaces are calculated as:

```
%geom Scan
        B 0 1 = 1.35, 1.10, 12 # C-O distance that will be scanned
        end
        end
```

Third Stop: Frequency Calculations

Frequencies? What Frequencies?

Behind the simple term 'Frequency Calculation' quite a few useful properties are hidden.

Calculation of the mass weighted Hessian matrix, i. e. the second derivative of the energy with

respect to all coordinates

$$H_{AB} = \frac{\partial^2 E^{(I)}}{\partial X_A \partial X_B} \bigg|_{X_A = X_B = \dots \bar{X}}$$

Upon diagonalization we get the frequencies of the normal modes:

$$\omega = \frac{1}{2\pi c} \sqrt{\frac{k}{m}}$$

Following the strict meaning, it describes the calculation of frequencies, usually in the domain of Infrared, in the harmonic approximation of all coordinates of the molecule.



Transition State?

At times the geometry optimization might not lead to a minimum on the PES, but to a transition state.



Two-dimensional potential energy surface for the system H-H-H in a linear arrangement. A transition state is observed around the H_1 - H_2 and H_2 - H_3 distances being both 1.0 Angström.

A transition state is defined as the point, where all coordinates are at a minimum level, with the exception of one, which is at its maximum.

Transition State?

At times the geometry optimization might not lead to a minimum on the PES, but to a transition state.



Analyzing the Normal Modes and Frequencies:

A Transition State has exactly one negative frequency!

Two-dimensional potential energy surface for the system H-H-H in a linear arrangement. A transition state is observed around the H_1 - H_2 and H_2 - H_3 distances being both 1.0 Angström.

A transition state is defined as the point, where all coordinates are at a minimum level, with the exception of one, which is at its maximum.

Zero Point Energy (ZPE)

As known from basic Physical Chemistry, even at 0K a molecule still has some vibrational energy.

Having calculated all normal modes of the molecule, as well as the frequency of vibration, it is fairly easy to calculate the zero point energy using the harmonic approximation.

In addition, taking into account a finite Temperature, the vibrational part of the ZPE accounts to

$$H_{vib} = R \sum_{i=1}^{3N-6(7)} \left(\frac{hv_i}{2k_B} + \frac{hv_i}{k_B} \cdot \frac{1}{e^{hv_i/k_BT} - 1} \right)$$

For the translational and rotational degrees of freedom, the statistical equipartition of energy is applied, meaning

 $H_{trans} = H_{rot} = 3/2 \text{ RT}$

Frequency Calculations

There are several good reasons for calculating the harmonic frequencies:

- 1. Characterize stationary points as minima (no negative frequencies), transition states (one negative frequency) or higher-order saddle point (more negative frequencies
- 2. Predict vibrational spectra (IR, Raman)
- **3.** Calculate thermodynamic properties (zero-point energy, finite temperature correction)

ORCA presently calculates harmonic frequencies through (one- or two-sided) numeric differentiation of analytic frequencies

<pre>! RKS BP86 RI SV(P) SV/J Grid4 TightSCF SmallPrint ! TightOpt NumFreq</pre>	
%freq CentralDiff true	 Run a numerical frequency calculation
Increment 0.005 end * xyz 0 1	 Two sided differences (twice as expensive but more accurate! Be careful – numerical frequencies can be quite noisy)
c 0.000000 0.000000 -0.533905 o 0.000000 0.000000 0.682807 H 0.000000 0.926563 -1.129511 H 0.000000 -0.926563 -1.129511 *	 Increment for displacements (in Bohrs)

The first thing that is printed are the vibrational frequencies. The first six modes are translations and rotations and these are zero because they are projected out.

VIBRATI	ONAL FREG	QUENCIES
0:	0.00	cm**-1
1:	0.00	cm**-1
2:	0.00	cm**-1
3:	0.00	cm**-1
4:	0.00	cm**-1
5:	0.00	cm**-1
6:	1140.72	cm**-1
7:	1230.49	cm**-1
8:	1498.92	cm**-1
9:	1812.75	cm**-1
10:	2773.72	cm**-1
11:	2805.59	cm**-1

Then the program prints the normal modes. This is usually not very revealing. The program produces a **BaseName.hess** file which you can run through **orca_vib** to get additional information.

XYZ files to be used for animation of vibrational modes are produces by the program **orca_pltvib**.

Then you get the IR spectrum:

IR SPECTRUM

F 066	
(· ·	
- ja	
	V
	V
l Isr	



The IR spectrum can be plotted through the orca_mapspc utility

(orca_mapspc BaseName.out IR)

1000

050

Finally, the thermodynamic properties at 298.15 K are printed (assuming ideal gas behaviour)

Electronic energy		-114.414	435858 Eh	•
Zero point energy	• • •	16.10	kcal/mol	
Thermal vibrational correct	tion	0.03	kcal/mol	
Thermal rotational correct	ion	0.89	kcal/mol	
Thermal translational corr	ection	0.89	kcal/mol	
				_
Total thermal energy		-114.385	582646 Eh	L
Then enthalpy+entropy and	finally the	e free er	nergy:	
Total enthalpy	-114.384882	225 Eh		
Total entropy correction	-15.36 kca	al/mol		
Final Gibbs free enthalpy	-114.409358	358 Eh		
$G \Pi (GT) = 0.0000000 \Pi \Pi =$	J.IT ACAL/I			

Mode	freq (cm**	-1) T**2	ТХ	TY	TZ
6:	1140.72	1.465985	(-1.210739	-0.008963	-0.004000)
7:	1230.49	10.082152	(0.004428	-3.175237	-0.001184)
8:	1498.92	4.648016	(-0.000197	0.001760	-2.155925)
9:	1812.75	105.821353	(-0.000043	0.002847	-10.286950)
10:	2773.72	73.358541	(0.000042	-0.020336	-8.564936)
11:	2805.59	205.230431	(-0.000359	-14.325864	0.007790)

Solvation

- ★ The effects of solvation can be quite important for many properties (and in particular for charged molecules for cations less than for anions).
- Solvation effects can be (relatively crudely) modelled by polarizable continuum models.
 In ORCA the Conductor Like Screening (COSMO) model by Klamt is implemented in an efficient way for energies and structures (don't do frequencies!)





- 1. Create a folder / directory named 'water'
- Open your text-editor (Windows: Notepad, MacOS: TextEdit)
- 3. Enter the 'water' example in the text-editor
- 4. Save the water example as '**water.inp**' in the '**water**' folder that has been created before
- 5. Open the command-line window

On Windows: Type '**cmd.exe**' in the program-search field

On MacOS: Open the 'Terminal' program from

'Application' -> 'Utilities'

- In the command-line window, move to the water folder/directory: Windows: cd C:\Documents\water MacOS: cd /Users/MyUSERNAME/Documents/water
- 7. Execute the ORCA program with the water example, type: orca water.inp

A simple water
! RKS B3LYP SVP OPT
* xyz 0 1
0 0 0 0
H 0 0 1.2
H 0 0 -1.2
*

A simple water Timings for individual modules: ! RKS B3LYP SVP OPT Sum of individual times 0.226 min) 13.587 sec (= . . . * xyz 0 1 GTO integral calculation 1.927 sec (= 0.032 min) 14.2 % . . . SCF iterations 9.339 sec (= 0.156 min) 68.7 % 0000 . . . SCF Gradient evaluation 2.097 sec (= 0.035 min) 15.4 % . . . H 0 0 1.2 Geometry relaxation 0.224 sec (= 0.004 min) 1.7 % . . . H 0 0 -1.2 ****ORCA TERMINATED NORMALLY**** * TOTAL RUN TIME: 0 days 0 hours 0 minutes 13 seconds 767 msec

C:\Documents\water>

٠

Where did my output go?



C:\Documents\water>

Where did my output go?

Starting ORCA for the advanced user: **orca water.inp > water.out**

This will put the *output* of the calculation into the file '**water.out**'. This file can be opened with the text-editor and examined at leisure.

	e o o water.out
	The step convergence is overachieved with reasonable convergence on the gradient Convergence will therefore be signaled now

Where is my	Redundant Internal Coordinates Optimized Parameters (Angstroem and degrees)
result?	Definition OldVal dE/dq Step FinalVal
(Or is water really linear?)	1. B(H 1,0 0) 0.9378 0.000333 -0.0003 0.9375 2. B(H 2.0 0) 0.9378 0.000333 -0.0003 0.9375
(OF is water really linear !)	3. L(H 1,0 0,H 2,2) 180.00 0.000000 -0.00 180.00 4. L(H 1,0 0,H 2,1) 180.00 0.000000 -0.00 180.00
Look for ' HURRAY ': Below it you will find the optimized structure	CARTESIAN COORDINATES (ANGSTROEM) 0 8.000000 -0.000000 -0.000000 H -0.000000 0.000000 0.937514 H -0.000000 0.000000 -0.937514
	NO LB ZA FRAG MASS X Y Z 0 0 8.0000 0 15.999 0.000000030868530 -0.000000034755270 -0.000000002434515 1 H 1.0000 0 1.008 -0.000000015434265 0.000000017377635 1.771645136378302 2 H 1.0000 0 1.008 -0.000000015434265 0.000000017377635 -1.771645133943787
	INTERNAL COORDINATES (ANGSTROEM)
	0 0 0 0 0 0.000000 0.000 0.000 H 1 0 0 0.937514 0.000 0.000 H 1 2 0 0.937514 180.000 0.000
	INTERNAL COORDINATES (A.U.)
	H 1 2 0 1.771645 180.000 0.000

*

Sometimes it might be difficult to imagine the structure by looking at the xyz coordinates...



Open the molecule with a molecule-viewer, e.g. Chimera...

Open the molecule with a molecule-viewer, e.g. Chimera...



Ahm..... what happened?!

Slight modification:

A simple water ! RKS B3LYP SVP TightOPT Freq * xyz 0 1 0 0 0 0 H 0 0 1.2 H 0 0 -1.2 *

This will result on the calculation of vibrational frequencies

00	🗋 water.o	ut			
CP-SCF ITERATION 0: CP-SCF ITERATION 1: 0.000407361 CP-SCF ITERATION 2: 0.000004867 CP-SCF ITERATION 3: 0.000000005 Forming perturbed density Hessian contributions Making the perturbed densities 2nd integral derivative contribs Exchange-correlation Hessian Dipol derivatives Total SCF Hessian time: 0 days 0 hours 0 min 3 sec					
Warning: T+R mode no 5 is zero VIBRATIONAL FREQUENCIES					
0: 0.00 cm** 1: 0.00 cm** 2: 0.00 cm** 3: 0.00 cm** 4: 0.00 cm** 5: -1571.41 cm** 6: -1571.41 cm** 7: 4095.52 cm** 8: 4480.15 cm**	-1 -1 -1 -1 -1 ***imaginary mode*** -1 ***imaginary mode*** -1 -1				

Negative frequencies: The optimization has converged to a **saddle-point**!!!

Solution: Start from a different structure

A si RKS	Lm <u>r</u> B3	ple water BLYP SVP	r, 90 degrees TightOPT Freq										
xyz	0	1											
0	0	0											
0	0	1.2											
0 1.	. 2	0											
	A si RKS 0 0 0 1	A simp RKS B3 xyz 0 0 0 0 0 0 1.2	A simple water RKS B3LYP SVP xyz 0 1 0 0 0 0 0 1.2 0 1.2 0										
0	0		,		🗋 water_	90.0	out						
----------------------------	--	---	----------------------------------	----------------------------------	--	-------	---	------------------	------------------------------	-------------	--	------------	--
	**************************************						***** K*** File Select Actions		esets Tools	Favorites	UCSF Chimera Help		
		***	*****	(AFTER	6 CYCLES)	*****	***						
CARTESIA	AN COORDI	NATES (AN	IGSTROEM)										
0 H H	0.00000 0.00000 0.00000	0.000000 0.116774 0.000000 0.005866 0.000000 1.077360		0.116774 L.077360 0.005866									
CARTESIA	AN COORDI	NATES (A.	U.)									6 7	
NO LB 0 0 1 H 2 H	ZA 8.000 1.000 1.000	FRAG 10 0 10 0 10 0	MASS 15.999 1.008 1.008	x	Y 0.00000000000000000000000000000000000	Z	0.220670558659413 0.011085364462620 2.035915437583469						
INTERNAL	COORDIN	ATES (ANG	STROEM)										
0 Н Н	0 0 1 0 1 2	0 0.000 0 0.966 0 0.966	968 6 968 6 968 183	0.000 0.000 3.172	0.000 0.000 0.000		Forming perturbed density Hessian	contributions	done	er_90. (out 1.1 sec) 0.0 sec)		
INTERNAL 0	COORDIN	WATES (A.U	1.) 1000 6	0.000	0.000		Making the perturbed densities 2nd integral derivative contribs Exchange-correlation Hessian Dipol derivatives		done done done done		0.0 sec) 0.3 sec) 0.7 sec) 0.1 sec)		
н	1 0	0 1.827 0 1.827	7384 6 7384 183	3.172	0.000		Total SCF Hessian time: 0 days 0 h	nours 0 min 3 se	c				

Same procedure as before... water is **not** linear!

ATIONAL	FREQUEN	ICIES			
	0.00	cm**-1			
	0.00	cm**-1			
l.	0.00	cm**-1			
6	0.00	cm**-1			
	0.00	cm**-1			
	0.00	cm**-1			
	1638.00	cm**-1			
:	3787.36	cm**-1			
	3883.54	cm**-1			

These modes are the cartesian displacements weighted by the diagonal matrix

The End