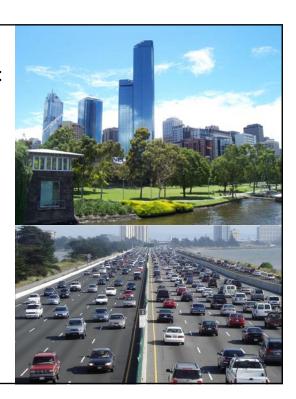
Chapter 14 Indoor Air Quality

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Indoor Environment and Built Environment

- Buildings
- Transportations vehicles



Indoor Air Quality (IAQ)

Indoor air quality (IAQ) is the quality of air in an indoor environment.

- Thermal comfort
 - Temperature,
 - Relative Humidity
- Indoor Air Pollutant

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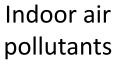
Why is IAQ Important?

- We spend over 90% of our time in indoor environments
- IAQ is much poorer than outdoor air
 - 2-100 times worse in USA/Canada
 - National Institute for Occupational Safety and Health (NIOSH) ranked <u>Occupational</u> lung dysfunctions (including lung cancer, pneumonoconioses, and occupational asthma) the top occupational diseases and injuries.
- 30% of newly constructed or remodeled facilities have IAQ problems. Illnesses related to indoor air pollution have been classified into two categories:
 - Sick building syndrome (SBS)
 - Building related illness (BRI)
- Indoor contaminants are responsible for half of all illnesses
- Liability issues

Illnesses related to indoor air pollution

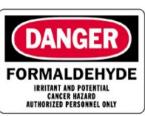
- SBS: Sick building syndrome
 - SBS is defined as the discomfort or sickness associated with poor indoor air quality with clear identification of the source substances.
 - Examples: irritation to eyes, noses, or throat, fatigue, and nausea.
- BRI: Building related illness
 - BRI is defined as a recognized disease caused by known agents that can be clinically identified.
 - Examples: asthma, legionella, hypersensitivity, and humidifier fever.
- Approximately one million buildings in the United States are sick buildings with 70 million occupants.

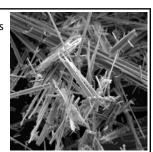
Asbestos



- Dust/Aerosol/PM
- Asbestos
- Combustion related contaminants
- Formaldehyde
- VOCs: Volatile Organic Compounds
- Bio-aerosol: Allergen, mold, dust mite
- Radon



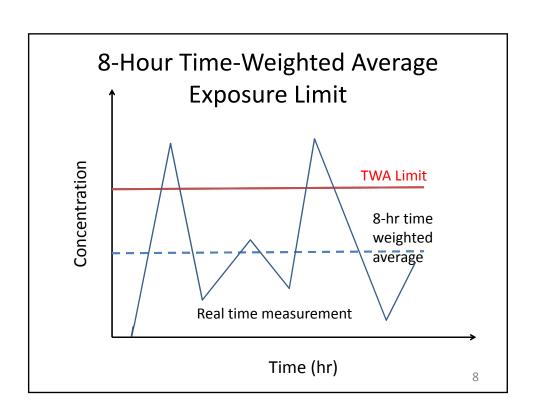






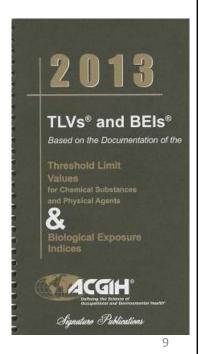
Threshold Limit Values

- Threshold limit values (TLVs) refer to upper limit of the concentrations of indoor air pollutants under which it is believed to be safe for all working occupants without impacting their health.
 - 1. Time-Weighted Average Threshold Limit Value (TLV-TWA)
 - Usually 8 hours 24 hours or Annual
 - 2. Short -Term Exposure Limit Threshold Limit Value (TLV-STEL):
 - Say 15 minutes
 - 3. Threshold Limit Value Ceiling (TLV-C):
 - At any instance



Exposure Limit Guidelines

- American Conference of Governmental Industrial Hygienists (ACGIH) also publishes
 - Threshold Limit Values for Chemical Substances and Physical Agents and Biological Exposure Indices.
 - Updated annually
- WHO (2010) guidelines for indoor air quality



Threshold values of typical indoor air pollutants in work places								
Pollutant	Formula	Odor	TWA	STEL				
Gases (In ppmv unless stated otherwise)								
Ammonia	NH ₃	Sharp pungent	25	35				
Carbon dioxide	CO ₂	None	5,000	30,000				
Carbon monoxide	СО	None	25					
Formaldehyde	НСНО	Pungent	-	0.3				
Hydrogen sulfide	H ₂ S	Rotten eggs nauseating	10	15				
Methanol	CH ₃ OH	Alcohol	200	250				
Ozone	03	Odorous	0.05-0.2	-				
Particulates (Unit in mg/m³ unless stated otherwise)								
Asbestos		0.1 fiber/ml						
Coal dust, anthracite	9	0.4						
Coal dust, bitumino	JS	0.9						
Grain dust (Oat, who	eat, barley)	4.0						
Graphite (non fiber)		2.0						
Iron oxide particles	and fume, in	5.0						
Lead		0.05						
Welding fumes		5.0						

Normalized Air Contaminant Concentration

In order to quantify the effect of the combined effect, the normalized concentration

$$C_N = \sum \frac{C_i}{TLV_i} \frac{\text{Measured concentrations}}{\text{Threshold limits}}$$

 C_N >1: Do something!

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Example 14.1

- In a welding shop, the measured concentrations of CO, CO₂ and welding fumes are 10 ppmv, 1,500 ppmv and 3.5 mg/m³, respectively, each below the recommended TLV-TWA.
- Is this working environment safe to the workers daily based on normalized concentration?

Solution

Air Pollutant	TLV-TWA	Measured concentration	C _i /TLV _i
CO	25	10	0.40
CO ₂	5,000	1500	0.30
Welding fumes	5	3.5	0.70

$$C_N = \sum \frac{C_i}{TLV_i} = 0.40 + 0.30 + 0.70 = 1.4 > 1.0$$

Since the $C_N > 1$, it is not safe for the workers to be there daily (8 hours).

Clean room





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Clean room ISO 146441-1

$$c_p^* = 10^N \left(\frac{0.1}{d_p^*}\right)^{2.08}$$

- c_p^* =upper limit of particle number concentration, number/m³
- N = the clean room class number
 - N=1, 2, ...9
- $d_p^* = \text{threshold particle diameter in } \mu\text{m}$,
 - $-d_p^* = 0.1, 0.2, 0.3, 0.5, 1$ and 5 μ m.
- According to ISO 146441-1, in a class N cleanroom, the number concentration of particles greater than d_p^* cannot exceed c_p^* .

Example 14.2: Cleanroom class

- An ISO 146441–1 Class 2 cleanroom is 3 meter heigh with a total area of 100 m², what is the maximum amount of particles that are greater than 100 nm?
- Solution:

Substitute
$$d_p^* = 100 \text{ nm} = 0.1 \text{ } \mu\text{m}; N = 2$$

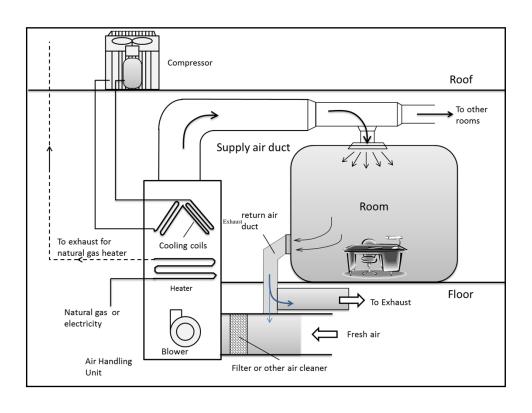
$$c_p^* = 10^N \left(\frac{0.1}{d_p^*}\right)^{2.08} = 10^2 \left(\frac{0.1}{0.1}\right)^{2.08} = 100$$

• Since the volume of the room is 300 m³, the total number of particles larger than 100 nm in diameter cannot exceed 30,000 in total.

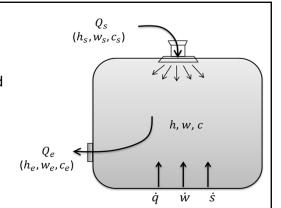
ISO Class Number	Maximum particle number concentration (#/m³) for particle sizes							
(N)	>0.1 μm	>0.2 μm	>0.3 μm	>0.5 μm	>1 μm	>5 μm		
ISO Class 1	10	2						
ISO Class 2	100	24	10	4				
ISO Class 3	1,000	237	102	35	8			
ISO Class 4	10,000	2,370	1,020	352	83			
ISO Class 5	100,000	23,700	10,200	3,520	832	29		
ISO Class 6	1,000,000	237,000	102,000	35,200	8,320	293		
ISO Class 7				352,000	83,200	2,930		
ISO Class 8				3,520,000	832,000	29,300		
ISO Class 9				35,200,000	8,320,000	293,000		

IAQ Control

- Source control
 - Very important to IAQ
- Ventilation: Ventilation controls the indoor air quality by
 - bringing fresh air into an indoor environment
 - supply or reduce the heat and moisture and
 - dilute gaseous and particulate pollutants indoors.
- IAQ control devices
 - Indoor air cleaner
 - Indoor (de)humidifier



- In general, three types of variables are of concern and can be controlled by ventilation in an indoor environment:
 - temperature,
 - relative humidity, and
 - air pollutants.
- Ideally a minimum ventilation rate must be maintained in order to control all these parameters at desired levels for an indoor environment.



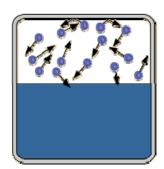
 Practically many buildings minimum ventilation rates are based on the temperature control due to the energy concern.

Moist Air

- Moist Air=Dry air + Water vapor
- For air at certain pressure and temperature
- Water vapor pressure
 - Partial pressure of water vapor
- Saturated air

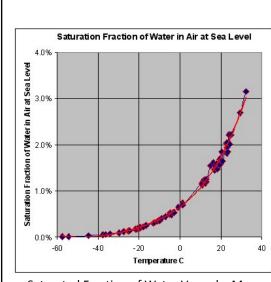
Dew Point

 The dew point is the temperature at which the water vapor in air (at constant pressure) condenses into liquid water at the same rate at which it evaporates.



- At temperatures below the dew point, water vapor condenses and leaves the air.
- The condensed water is called dew when it forms on a solid surface.

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Saturated Fraction of Water Vapor by Mass

 The maximum saturation pressure of the water vapor in moist air varies with the temperature of the air vapor mixture

Humidity

- Absolute humidity: the mass of water vapor per unit volume of moist air
 - Not useful for heat and mass transfer
- Relative humidity is the ratio of the partial pressure of water vapor in an air-water mixture to the saturated vapor pressure of water at a prescribed temperature.
 - The relative humidity of air depends on temperature and the pressure of the system of interest.

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How Relative Humidity Affects IAQ

- Too low, dry air, evaporation from surface (including your body)
 - Dry skin, dry eyes, dry throat.... Discomfort!
 - 50-60% most comfort
- Too high, condensation, mold



Acceptable Temperature Ranges ASHRAE 62.1-2004

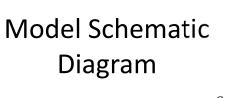
• Dry bulb at 30% relative humidity:

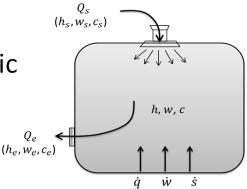
• Winter: 68.5 °F – 76.0°F (20.28 oC- 24.44 oC)

• Summer: 74.0°F – 80.0°F (23.33 oC-26.67 oC)

Minimum Ventilation Rate

- Consider a control volume, where temperature, relative humidity and air pollutant levels are uniform.
 - This assumption is acceptable if the control volume is small enough or the error is acceptable.
- This control volume could be an entire room or a zone within.
- When the control volume is a room, the air within is assumed completely mixed.
- Admittedly this is a bold assumption, but it has been widely used in guiding the HVAC industry.





Q= Volumetric flow rate; V= Volume of the pace h= total enthalpy w= moisture content c= pollutant concentration s= supply air e= exhaust air

- <u>dot</u> stands for generation rate
- Perfect mixing \rightarrow the properties in the exhaust air are the same as in the room air. $v = v_e$; $h = h_e$, $c = c_e$

Mass balance for dry air

$$\frac{d}{dt}\left(\frac{V}{v}\right) = \frac{Q_s}{v_s} - \frac{Q_e}{v_e}$$

- where v is specific volume of moist air, which is defined herein as the volume of the moist air (dry air plus the water vapor) containing one unit of mass of "dry air".
- we use specific volume of the air instead of the density because specific volume the of the air defines the volume of the mixture (dry air plus the water vapor) containing one unit of mass of "dry air".

the energy balance

$$\frac{d}{dt}\left(\frac{Vh}{V}\right) = \dot{q} + \frac{Q_s}{V_s}h_s - \frac{Q_e}{V_e}h_e$$

- h = sensible heat for supply air and exhaust air (kJ/kg of dry air). It implies that the difference between the sensible heat of the supply air and that of the exhaust air equals to the total sensible heat production plus the change of sensible heat in the room air.
- The total sensible heat transfer rate, \dot{q} , of the ventilated airspace is the sum of all heat loss or gain through and within the airspace, including the sensible heat production rate by occupants and indoor equipment (e.g. stove, lights), and the heat transfer through building envelope. \dot{q} cam be positive or negative, which are referred to as heating load or cooling load.

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Moisture mass balances

$$\frac{d}{dt}\left(\frac{Vw}{v}\right) = \dot{w} + \frac{Q_s}{v_s}w_s - \frac{Q_e}{v_e}w_e$$

where w = humidity ratio (kg of water vapor per kg of <u>dry</u> air, kg/kg).

Subscripts 's' and 'e' stand for supply and exhaust air, respectively. \dot{w} is the water vapor production rate in kg/s.

Specific Pollutant Mass Balance

$$\frac{d}{dt}(Vc) = \dot{s} + Q_s c_s - Q_e c_e$$

- In most engineering practice, the unit of gases is in ppmv Therefore, unit conversions are necessary for accurate calculation of ventilation rate with a unit of m³ air/s
- $\dot{s}_{CO2} = \rho_{CO2} \frac{THP}{24,600}$

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CO₂ Production Rate

- From large scale statistical analyses, it has been determined that the carbon dioxide production rate of average human and animals is
- 1 liter of CO₂ ~24,600 J of total heat production (THP) under standard indoor conditions.
- Converting to the mass production rate of carbon dioxide in kg/s is

$$\dot{s}_{CO_2} = \rho_{CO_2} \frac{THP}{24600}$$

- ρ_{CO_2} =the density of carbon dioxide (1.83 kg/m³ at 20 °C and 1 atm)
- THP = the total heat production of occupants (kJ/s)

$$v = v_e; h = h_e; c = c_e$$

•
$$\frac{d}{dt}\left(\frac{V}{V}\right) = \frac{Q_S}{V_S} - \frac{Q_e}{V_e}$$
 • $\frac{Q_S}{V_S} - \frac{Q_e}{V_e} = 0$

$$\frac{Q_S}{v_S} - \frac{Q_e}{v_e} = 0$$

•
$$\frac{d}{dt}\left(\frac{Vh}{V}\right) = \dot{q} + \frac{Q_S}{V_S}h_S - \frac{Q_e}{V_e}h_e$$
 • $\frac{\dot{q}}{h_e - h_S} = \frac{Q_S}{V_S} = \frac{Q_e}{V_e}$

$$\bullet \quad \frac{\dot{q}}{h_e - h_S} = \frac{Q_S}{v_S} = \frac{Q_e}{v_e}$$

•
$$\frac{d}{dt} \left(\frac{Vw}{v} \right) = \dot{w} + \frac{Q_S}{v_S} w_S - \frac{Q_e}{v_e} w_e$$
 • $\frac{\dot{w}}{w_e - w_S} = \frac{Q_S}{v_S} = \frac{Q_e}{v_e}$

$$\bullet \quad \frac{\dot{w}}{w_e - w_s} = \frac{Q_s}{v_s} = \frac{Q_e}{v_e}$$

•
$$\frac{d}{dt}(Vc) = \dot{s} + \frac{Q_S}{v_S}c_S - \frac{Q_e}{v_e}c_e$$
 • $\dot{s} + Q_Sc_S = Q_ec_e$

•
$$\dot{s} + Q_s c_s = Q_e c_e$$

$$\dot{s} = Q_e c_e \left(1 - \frac{v_s c_s}{v_e c_e} \right) = Q_s c_s \left(\frac{v_e c_e}{v_s c_s} - 1 \right)$$

Psychrometric chart

- http://www.uigi.com/UIGI SI.PDF
- See handout

Example 14.3: Ventilation rate calculation

- A dinning room have 20 people
- each produces 200 W total heat.
- The carbon dioxide concentration in the supply air is 500 ppmv.
- Assume the supply air temperature is 15 °C and 50% relative humidity.
- The room air is 22 °C and 60% relative humidity.
- If the required maximum CO₂ concentration in the room is 1000 ppmv,
- estimate the minimum ventilation rate based on the CO₂ concentration.

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Solution

- Based on the supply air temperature of 15 °C and 50% relative humidity we can get the specific volume of supply air using a psychrometric chart, v_s =0.822 m³/kg dry air
- The specific volume of exhaust air at 22 $^{\circ}$ C and 60% relative humidity is v_e =0.855 m^3 /kg dry air
- The total heat production rate by 20 people in the dining room is, $THP=20\times200~W=4000W~or~4~kJ/s$
- Then the mass production rate of CO_2 is estimated using Equation (14-9) $\dot{s}_{CO2} = \rho_{CO2} \frac{THP}{24600} = 1.83 \times \frac{4}{24600} = 2.98 \times 10^{-4} \, (kg/s)$
- Then the minimum ventilation rate fan is determined using Equation (14-6)

$$\dot{s}_{CO2} = Q_e c_e \left(1 - \frac{v_s c_s}{v_e c_e} \right) = Q_s c_s \left(\frac{v_e c_e}{v_s c_s} - 1 \right)$$

Solution (continued)

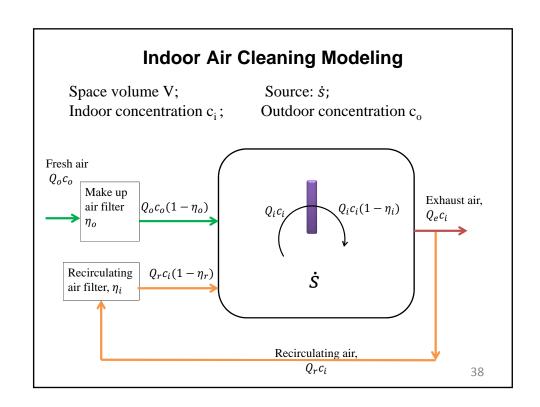
$$\dot{s}_{CO2} = Q_e c_e \left(1 - \frac{v_s c_s}{v_e c_e} \right) = Q_s c_s \left(\frac{v_e c_e}{v_s c_s} - 1 \right)$$

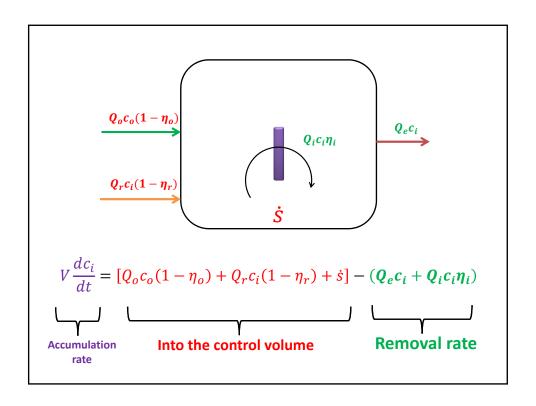
Since $v_s < v_e$, the minimum ventilation rate should be calculated based on the exhaust air. However, we need to convert the unit of concentration c from ppmv to kg CO_2/m^3 air.

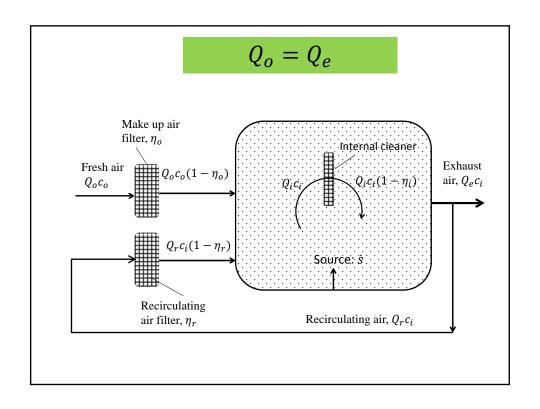
$$\begin{split} c_e &= \frac{1000 \, m^3 CO_2 \times 1.83 kg \, CO_2 / m^3 \, CO_2}{10^6 \, m^3 \, air} \\ &= 1.83 \times 10^{-3} \, kg \, CO_2 / m^3 \, air \end{split}$$

• The exhaust ventilation rate is

$$\begin{split} Q_e &= \frac{\dot{s}_{CO2}}{c_e \left(1 - \frac{v_s c_s}{v_e c_e}\right)} \\ &= \frac{2.98 \times 10^{-4} \ kg \ CO_2/s}{\left(\frac{1000 \ m^3 CO_2 \times 1.83 \ kg \ CO_2/m^3 \ CO_2}{10^6 \ m^3 \ air}\right) \left(1 - \frac{0.822}{0.855} \times \frac{500}{1000}\right)} \\ &= 0.31 \ m^3 \ dry \ air/s \end{split}$$







$$Q_{o} = Q_{e}$$

$$V \frac{dc_{i}}{dt} = [Q_{o}c_{o}(1 - \eta_{o}) + Q_{r}c_{i}(1 - \eta_{r}) + \dot{s}] - (Q_{e}c_{i} + Q_{i}c_{i}\eta_{i})$$

$$V\frac{dc_{i}}{dt} = [Q_{o}c_{o}(1 - \eta_{o}) + Q_{r}c_{i}(1 - \eta_{r}) + \dot{s}] - (Q_{o}c_{i} + Q_{i}c_{i}\eta_{i})$$

$$\begin{aligned} c_i &= \left[c_{i0} - \frac{Q_o c_o (1 - \eta_o) + \dot{s}}{Q_o - Q_r (1 - \eta_r) + Q_i \eta_i} \right] \exp \left[- \frac{Q_o - Q_r (1 - \eta_r) + Q_i \eta_i}{V} t \right] \\ &+ \frac{Q_o c_o (1 - \eta_o) + \dot{s}}{Q_o - Q_r (1 - \eta_r) + Q_i \eta_i} \end{aligned}$$

$$c_{i} = \left[c_{i0} - \frac{Q_{o}c_{o}(1 - \eta_{o}) + \dot{s}}{Q_{o} - Q_{r}(1 - \eta_{r}) + Q_{i}\eta_{i}}\right] \exp\left[-\frac{Q_{o} - Q_{r}(1 - \eta_{r}) + Q_{i}\eta_{i}}{V}\right] + \frac{Q_{o}c_{o}(1 - \eta_{o}) + \dot{s}}{Q_{o} - Q_{r}(1 - \eta_{r}) + Q_{i}\eta_{i}}$$

$$c_{i\infty} = \frac{Q_{o}c_{o}(1 - \eta_{o}) + \dot{s}}{Q_{o} - Q_{r}(1 - \eta_{r}) + Q_{i}\eta_{i}} \quad (t \to \infty)$$

$$\tau = \frac{V}{Q_{o} - Q_{r}(1 - \eta_{r}) + Q_{i}\eta_{i}}$$

$$c_{i}(t) = (c_{i0} - c_{i\infty}) \exp\left(-\frac{t}{\tau}\right) + c_{i\infty}$$

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Example 14.4

- Consider a kitchen with a volume of 80 m³
- where natural gas is used as a cooking fuel
- Assume after cooking the indoor fume concentration is $1000 \mu g/m^3$.
- The HVAC system works at a fresh flow rate of 1 m³/s
- a recirculating flow rate 1/3 of that of the fresh air.
- An internal air cleaner has a flow rate of 0.1 m³/s;
- Ignore the outdoor and indoor source.
- The intake and recirculating filter efficiencies are the same as 90%
- the internal air cleaner has a flow rate of 1 m³/s
- and an efficiency of 95%.
- Plot the indoor fume concentration over time.

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Solution

•
$$\dot{s}$$
=0, and $c_o = 0$

•
$$c_{i\infty} = \frac{Q_o c_o (1 - \eta_o) + \dot{s}}{Q_o - Q_r (1 - \eta_r) + Q_i \eta_i} = 0$$

$$\bullet = \frac{V}{Q_o - Q_r(1 - \eta_r) + Q_i \eta_i}$$

•
$$\tau = \frac{800}{10 - \frac{10}{3}(1 - 0.9) + 1 \times 0.95} = 75.35 (s)$$

•
$$c_i(t) = 1000 \exp\left(-\frac{t}{75.35}\right)$$

