

23

Metabolism, Nutrition, and Energetics

Lecture Presentation by
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Section 1: Introduction to Cellular Metabolism

Learning Outcomes

- 23.1 Define metabolism, catabolism, and anabolism, and give an overview of cellular metabolism.
- 23.2 Describe the role of the nutrient pool in cellular metabolism.
- 23.3 Summarize the important events and products of glycolysis.
- 23.4 Describe the basic steps in the citric acid cycle.

Section 1: Introduction to Cellular Metabolism

Learning Outcomes (continued)

- 23.5 Describe the basic steps in the electron transport chain.
- 23.6 Identify the sources of ATP production and energy yield at each source during glucose catabolism.
- 23.7 Define glycogenesis, glycogenolysis, and gluconeogenesis.

Module 23.1: Metabolism is the sum of catabolic and anabolic reactions

Metabolism

Sum of all chemical reactions that occur in an organism

Catabolism

Breakdown of organic substrates in the body

Anabolism

- Synthesis of new organic molecules
- Can make use of nutrient pool from blood

Cellular metabolism

Chemical reactions within cells

Module 23.1: Overview of cellular metabolism

Metabolism (continued)

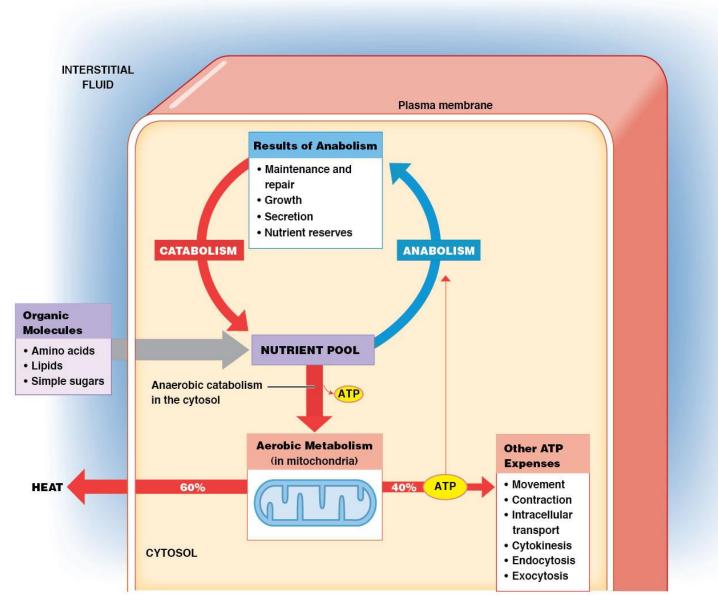
- Metabolic turnover
 - Process of continual breakdown and replacement of all cellular organic components except DNA
 - Cells obtain building blocks from:
 - Catabolic reactions
 - Absorption of organic molecules from the surrounding interstitial fluids
 - Both processes create an accessible source of organic substrates called a nutrient pool

Module 23.1: Overview of cellular metabolism

Metabolism (continued)

- Cellular catabolism (aerobic metabolism)
 - Occurs in the mitochondria
 - 40 percent of energy is captured
 - Used to convert adenosine diphosphate (ADP) to adenosine triphosphate (ATP)
 - ATP is used for anabolism and other cellular functions
 - 60 percent of energy escapes as heat
 - Warms the interior of the cell and the surrounding tissue

An overview of cellular metabolism



Module 23.1: Review

- A. Compare catabolism and anabolism.
- B. Explain the process of metabolic turnover.

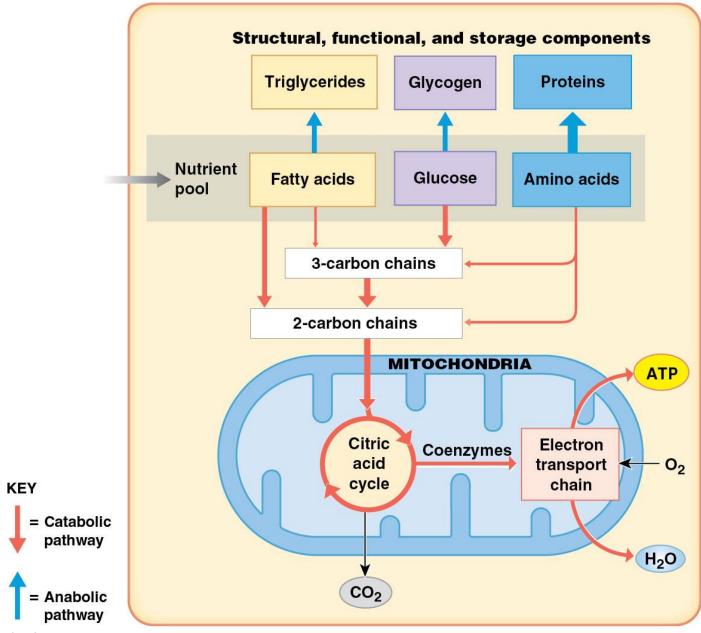
Learning Outcome: Define metabolism, catabolism, and anabolism, and give an overview of cellular metabolism.

Module 23.2: Cells use nutrients from the nutrient pool for metabolism

Nutrient pool

- Source for organic substrates (molecules) for both catabolism and anabolism
- Anabolism in the cell required for:
 - Replacing membranes, organelles, enzymes, and structural proteins
- Catabolism in the cell required for:
 - Converting substrates to a 2-carbon molecule
 - Utilized by mitochondria to produce ATP

The nutrient pool



Module 23.2: The nutrient pool

Mobilization of metabolic reserves

- Reserves are mobilized when absorption across the digestive tract is insufficient to maintain normal nutrient levels
 - Liver cells break down triglycerides and glycogen
 - Fatty acids and glucose can be released
 - Adipocytes break down triglycerides
 - Fatty acids can be released
 - Skeletal muscle cells break down contractile proteins
 - Amino acids can be released

Module 23.2: The nutrient pool

Restoration of metabolic reserves

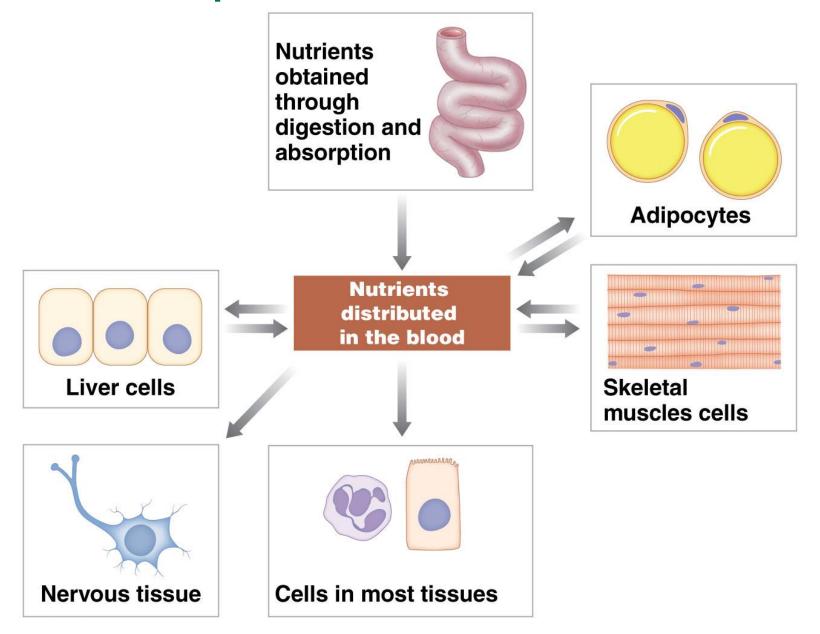
- Reserves are stocked when absorption by the digestive tract is greater than immediate nutrient needs
 - Liver cells store triglycerides and glycogen
 - Adipocytes convert excess fatty acids to triglycerides
 - Skeletal muscles build glycogen reserves and use amino acids to increase numbers of myofibrils

Module 23.2: The nutrient pool

Utilization of resources

- Cells in most tissues continuously absorb and catabolize glucose
- Nervous tissue must have a continuous supply of glucose
 - During starvation, other tissues can shift to fatty acid or amino acid catabolism
 - Conserves body's glucose for nervous tissue

The nutrient pool



Module 23.2: Review

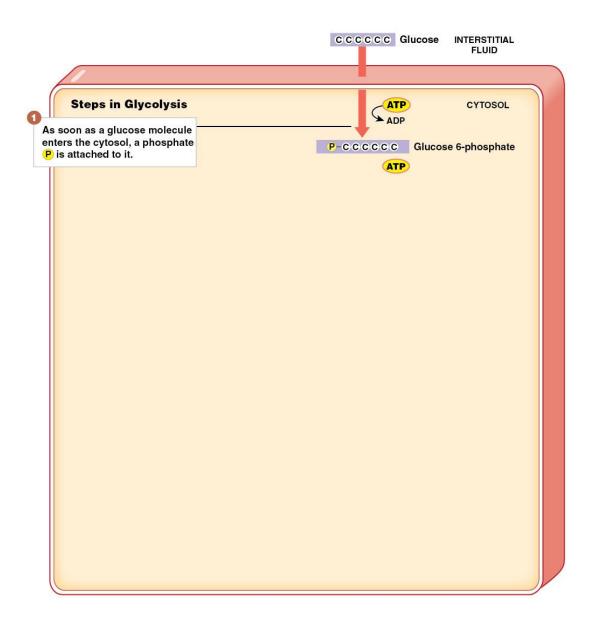
A. List the reactants required and products generated by mitochondria.

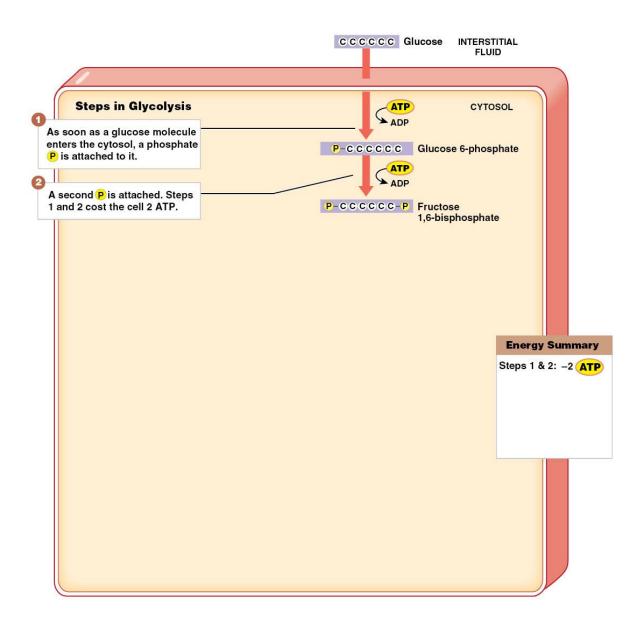
Learning Outcome: Describe the role of the nutrient pool in cellular metabolism.

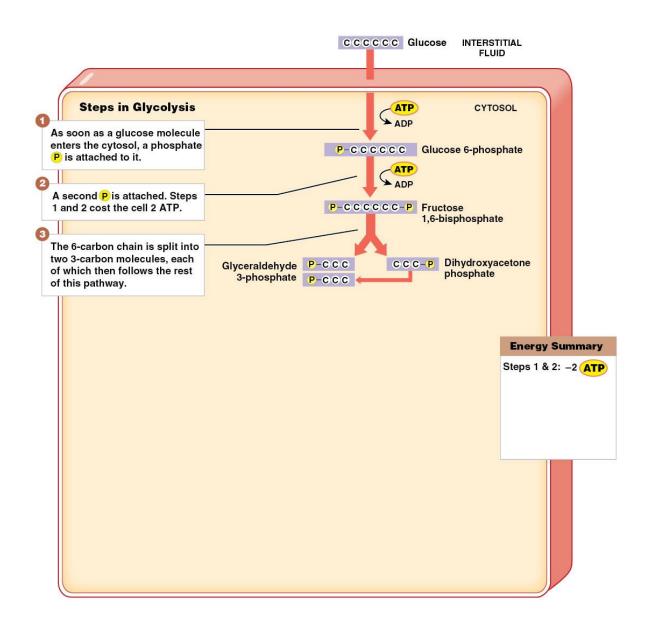
Module 23.3: Glycolysis is the first step in glucose catabolism

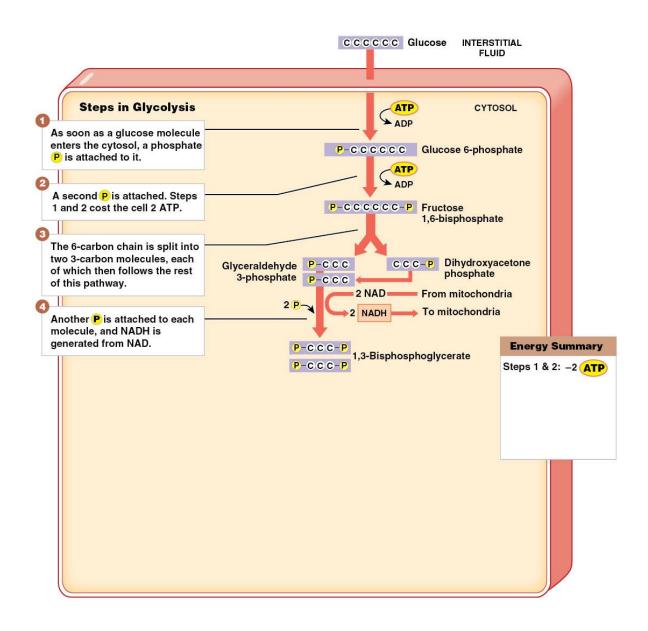
Glucose—preferred molecule for catabolism and ATP production

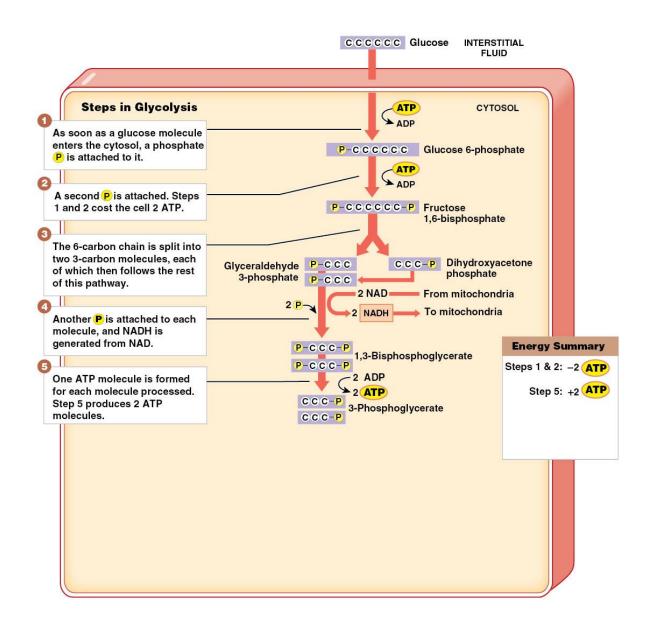
- Glycolysis—anaerobic process
 - Breaks down 6-carbon glucose to two 3-carbon pyruvates
 - Products: 2 net ATP, 2 pyruvate, 2 NADH (electron carriers)

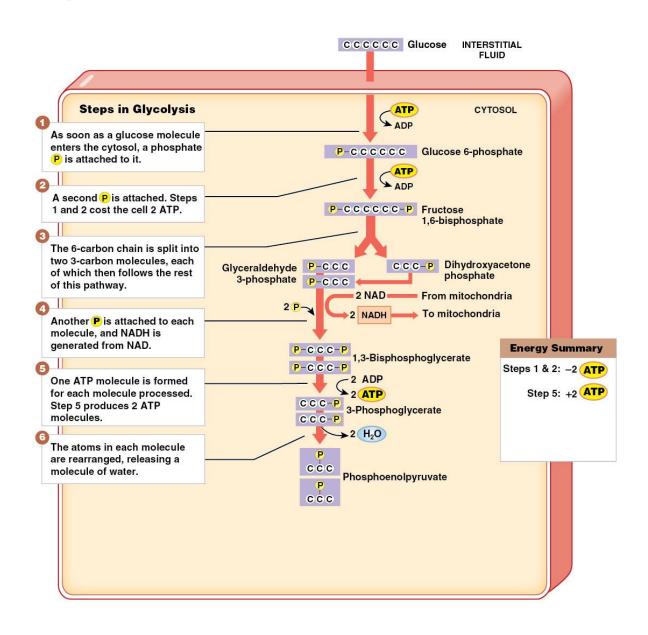


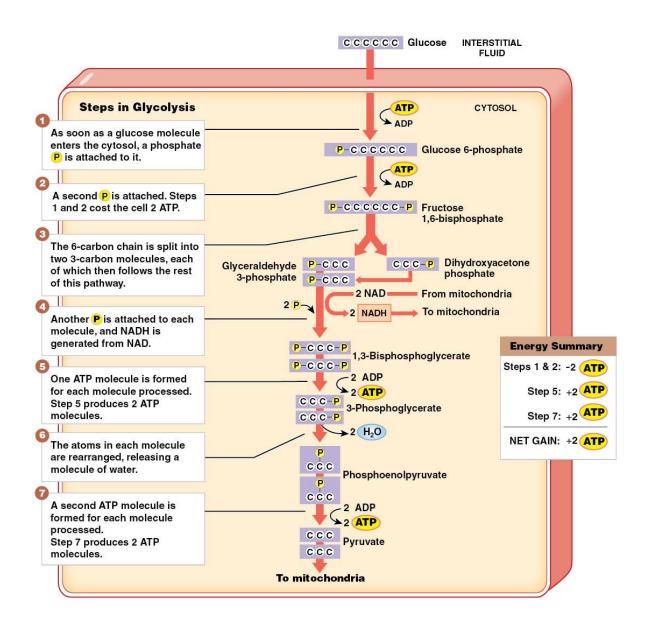












Module 23.3: Review

A. List the products of glycolysis.

Learning Outcome: Summarize the important events and products of glycolysis.

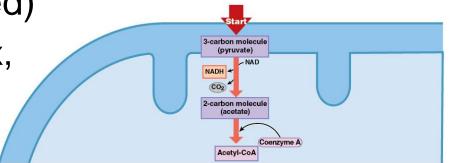
Module 23.4: The citric acid cycle transfers hydrogen atoms to coenzymes

Citric acid cycle (Krebs cycle)

- Aerobic metabolism—requires oxygen
- Overall function is to remove hydrogen atoms from specific organic molecules and transfer them to coenzymes
- Process of cellular ATP production begins in the cytoplasm
 - Organic nutrients from the nutrient pool are broken into smaller 3-carbon and 2-carbon molecules

Citric acid cycle (continued)

 In the mitochondrial matrix, pyruvate (3-carbon molecule) is converted into a 2-carbon acetate

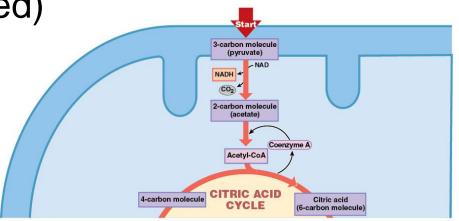


ion

- In the process, 1 NADH, and 1 CO₂ are produced
- Common substrate for mitochondria is acetate
 - Attaches to coenzyme A to form acetyl-coenzyme A (acetyl-CoA)
 - This acetyl-CoA then enters the citric acid cycle

Citric acid cycle (continued)

 The acetyl group (—CH₃CO) from acetyl-CoA attaches to a 4-carbon molecule



- Releases coenzyme A
- Produces citric acid (a 6-carbon molecule)

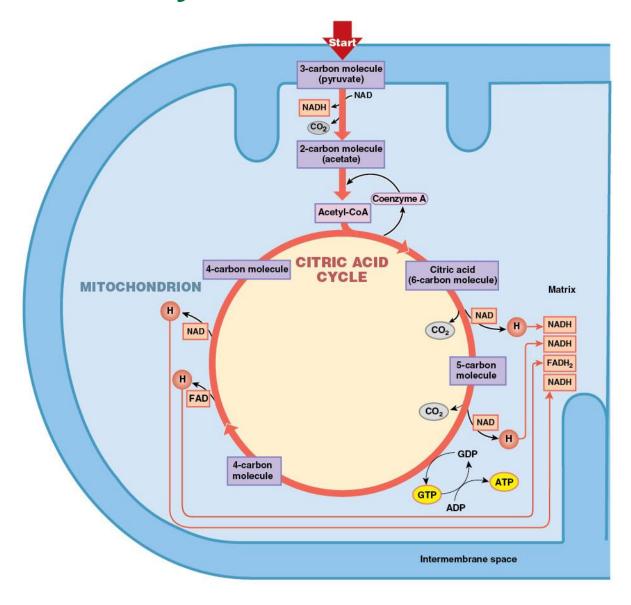
Citric acid cycle (continued)

- In a series of reactions, hydrogen atoms are removed from organic molecules
 - Hydrogen atoms (and their electrons) are transferred to coenzymes
 - Two carbon atoms are lost as CO₂
 - Remaining 4-carbon molecule is then ready to receive another 2-carbon acetyl group from acetyl-CoA
 - Each cycle turn results in production of 1 ATP
 - Cycle turns twice per 1 glucose

Citric acid cycle (continued)

- Coenzymes
 - Deliver hydrogen atoms to the electron transport system (ETS)
 - NAD (nicotinamide adenine dinucleotide)
 - Each NAD can carry 1 hydrogen atom as NADH
 - FAD (flavin adenine dinucleotide)
 - Each FAD can carry 2 hydrogen atoms as FADH₂

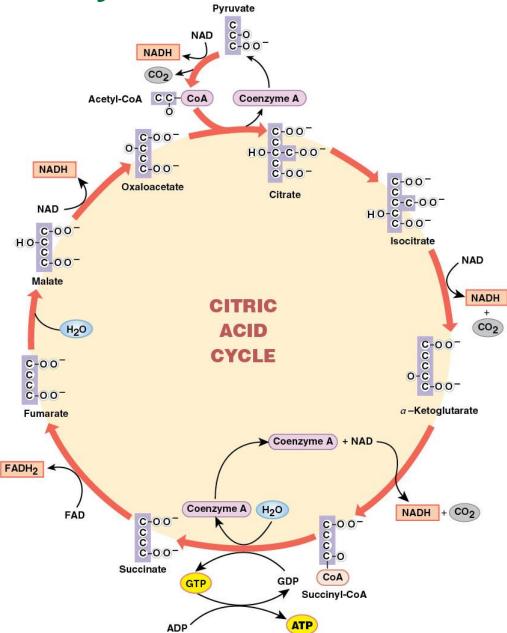
The citric acid cycle



Summary

- For each acetyl-CoA molecule entering the citric acid cycle:
 - 5 hydrogen atoms are removed and transferred to coenzymes
 - 2 molecules of CO₂ are produced
 - 2 molecules of water are consumed
 - Net energy gain of 1 ATP
 - Guanosine triphosphate (GTP) is formed from Guanosine diphosphate (GDP)
 - Transfers a phosphate group to ADP, forming ATP

The citric acid cycle



Module 23.4: Review

- A. Briefly describe the citric acid cycle, and explain its role.
- B. What molecule forms the common substrate for the citric acid cycle?
- C. What two coenzymes transfer hydrogen atoms to the electron transport system?

Learning Outcome: Describe the basic steps in the citric acid cycle.

Module 23.5: The electron transport chain establishes a proton gradient used to make ATP

Oxidative phosphorylation

- Generation of ATP within mitochondria in a reaction sequence that requires coenzymes and consumes oxygen
 - Oxidation refers to the transfer of electrons
 - Phosphorylation refers to the attachment of a phosphate to ADP producing ATP
- Produces more than 90 percent of the ATP used by body cells

Module 23.5: The electron transport chain

Oxidative phosphorylation (continued)

- Key reactions occur in the electron transport chain (ETC), or respiratory chain
- Oxygen is needed as the final electron acceptor
 - Lack of oxygen stops the ETC
 - Blocking cytochromes also stops the ETC
 - Example: poisons such as cyanide
 - With no functioning ETC, the citric acid cycle stops
 - Cells die from lack of ATP

Module 23.5: The electron transport chain

Steps in oxidative phosphorylation

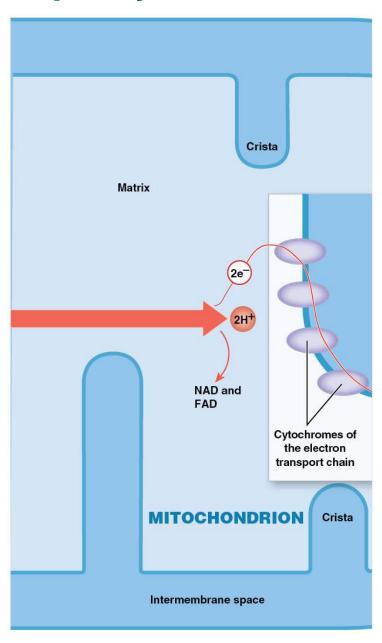
- 1. Coenzymes deliver hydrogen atoms from the citric acid cycle to the ETC
 - Series of protein-pigment molecules called cytochromes (cyto-, cell + chroma, color) embedded in inner mitochondrial membrane
- The hydrogen atom is composed of:
 - A high-energy electron (e⁻)
 - Given to the ETC
 - A hydrogen ion or proton (H+)
 - Released into the mitochondrial matrix

Module 23.5: The electron transport chain

Steps in oxidative phosphorylation (continued)

- 1. (continued)
 - Oxidation-reduction (redox) reactions
 - Oxidation is the loss of electrons
 - Reduction is the gain of electrons

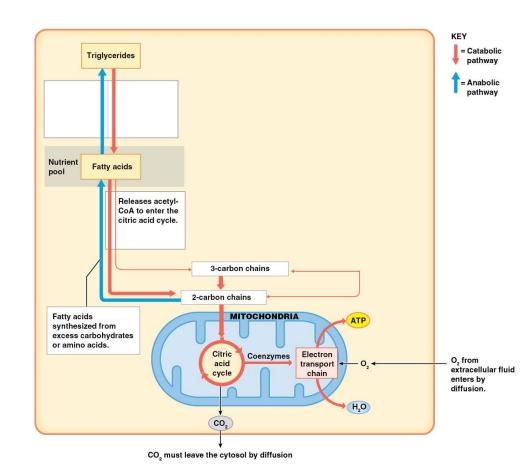
Oxidative phosphorylation



Module 23.5: The electron transport chain

Steps in oxidative phosphorylation (continued)

- Cytochromes pass electrons sequentially down the chain of cytochromes
- 3. Released energy is used to pump hydrogen ions into the intermembrane space

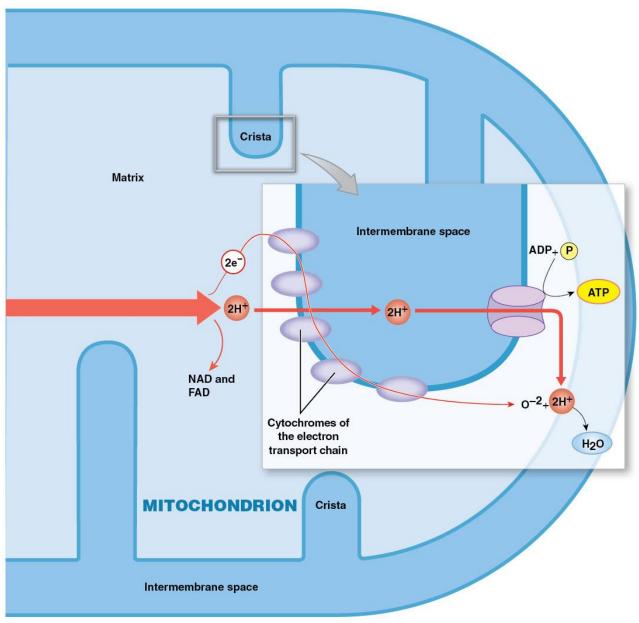


Module 23.5: The electron transport chain

Steps in oxidative phosphorylation (continued)

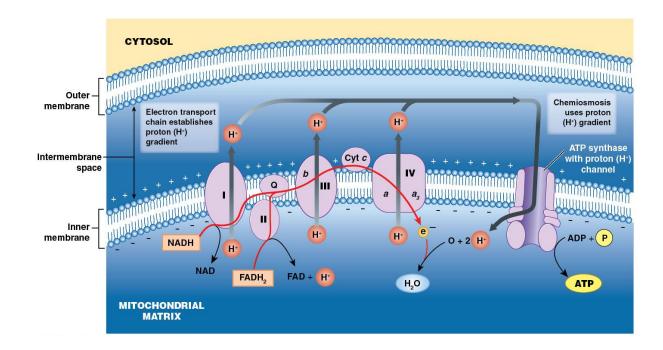
- 4. Hydrogen ions can diffuse back into the matrix only through specific hydrogen ion channels
 - Passage of the ions through these channels powers
 ATP production by ATP synthase
- 5. Oxygen is the final electron acceptor
 - Reacts with 2 hydrogen ions to form water

Oxidative phosphorylation



Module 23.5: The electron transport chain

- In oxidative phosphorylation, final electron acceptor is oxygen; water forms
- ETC has 4 respiratory complexes, coenzyme Q, and cytochromes b, c, a, and a₃
- Proton gradient generated by ETC leads to chemiosmosis



Mitochondrio

BioFlix: Cellular Respiration



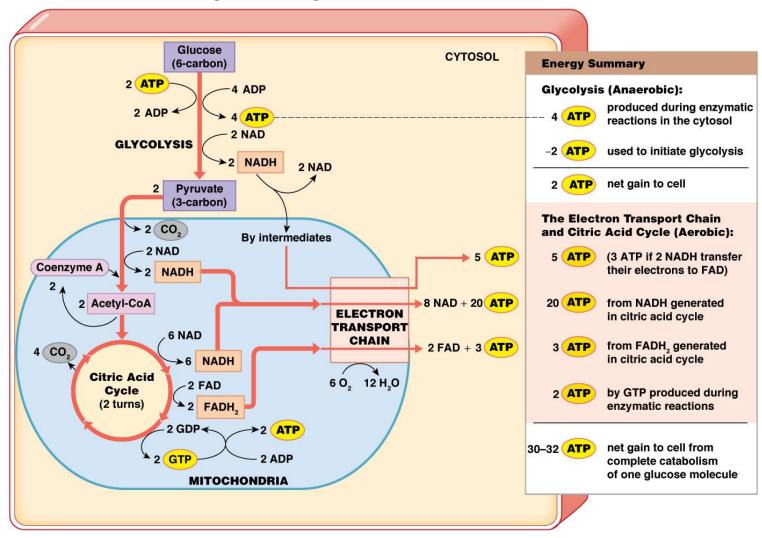
Module 23.5: Review

- A. Define oxidative phosphorylation.
- B. Compare oxidation and reduction.
- C. Describe the role that hydrogen ion channels play in the generation of ATP.

Learning Outcome: Describe the basic steps in the electron transport chain.

Glucose catabolism yields 30–32 ATP

1 molecule of glucose generates 30–32 molecules of ATP



Reasons Why Glucose Is the Primary Energy Source for Cells

- Glucose is a small, soluble molecule that is easily distributed through body fluids.
- Glucose can provide ATP anaerobically through glycolysis. Although only a small amount of ATP is produced, glycolysis is important during peak levels of physical activity, in red blood cells, or when a tissue is temporarily deprived of oxygen.
- Glucose can be stored as glycogen, which forms compact, insoluble granules.
- Glucose can be easily mobilized because the breakdown of glycogen
 (glycogenolysis) occurs very quickly and involves only a single enzymatic
 step. Mobilization of other intracellular reserves involves much more complex
 pathways and takes considerably more time.

Module 23.6: Review

- A. Identify when most of the CO₂ is released during the complete catabolism of glucose.
- B. Explain when glycolysis is important in cellular metabolism.

Learning Outcome: Identify the sources of ATP production and energy yield at each source during glucose catabolism.

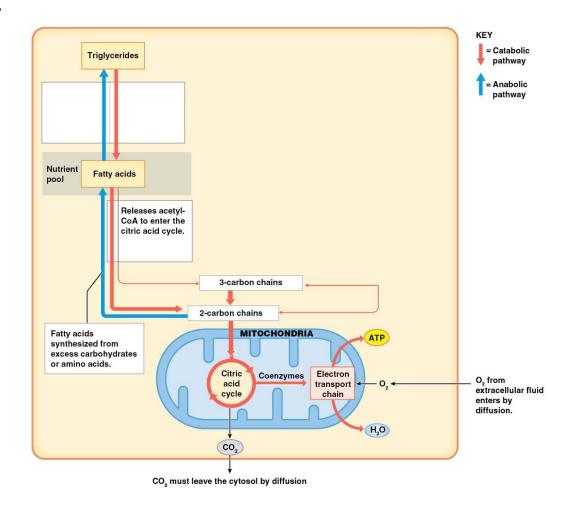
Module 23.7: Nutrient metabolism follows several pathways

The diet and nutrient pools do not provide all building blocks for proteins, carbohydrates, and lipids

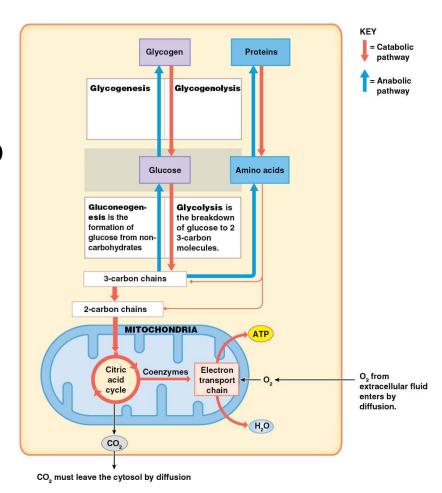
Cellular enzymes synthesize them

Fatty acids stored as triglycerides can be broken down into glycerol and fatty acids

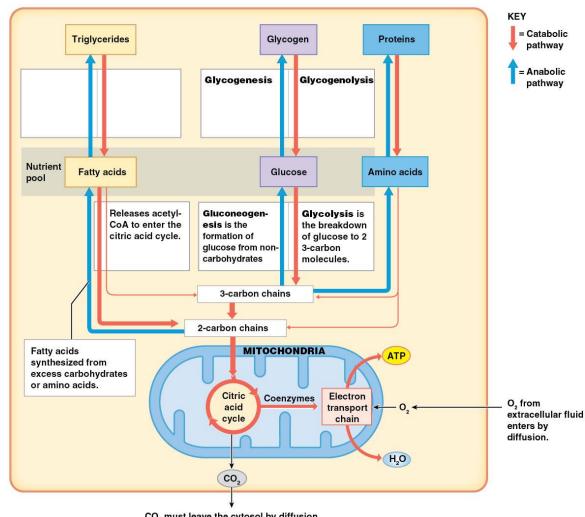
 Breakdown of fatty acid releases acetyl- CoA, which enters citric acid cycle



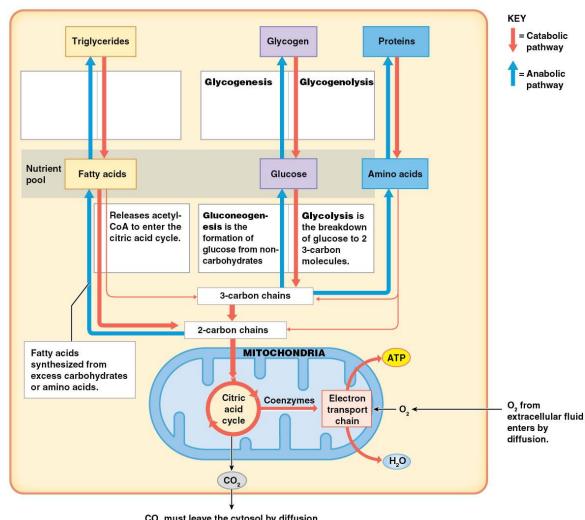
- Glycogenesis—formation of glycogen from glucose
- Glycogenolysis breakdown of glycogen into glucose
- Gluconeogenesis formation of glucose from noncarbohydrates
- Glycolysis—breakdown of glucose to two 3-carbon molecules



- Amino acids are used for protein synthesis
- In starvation, muscle proteins are mobilized



- CO₂ leaves cytosol by diffusion
- O₂ is delivered by diffusion
- Both require respiratory function



CO, must leave the cytosol by diffusion

Module 23.7: Review

- A. Compare glycogenesis and glycogenolysis.
- B. Why do cells make new compounds?

Learning Outcome: Define glycogenesis, glycogenolysis, and gluconeogenesis.

Section 2: Digestion and Metabolism of Organic Nutrients

Learning Outcomes

- 23.8 Outline the steps involved in digestion, and list the nutrients used by the body.
- 23.9 Describe carbohydrate metabolism.
- 23.10 Describe the mechanisms of lipid transport and distribution.
- 23.11 Describe the fate of fatty acids in lipid metabolism.

Section 2: Digestion and Metabolism of Organic Nutrients

Learning Outcomes (continued)

- 23.12 Summarize the main features of protein metabolism and the use of proteins as an energy source.
- 23.13 Differentiate between the absorptive and postabsorptive metabolic states, and summarize the characteristics of each.
- 23.14 Explain the role of fat-soluble vitamins and water-soluble vitamins in metabolic pathways.

Section 2: Digestion and Metabolism of Organic Nutrients

Learning Outcomes (continued)

- 23.15 Explain what constitutes a balanced diet and why such a diet is important.
- 23.16 Clinical Module: Describe several metabolic disorders resulting from nutritional or biochemical problems.

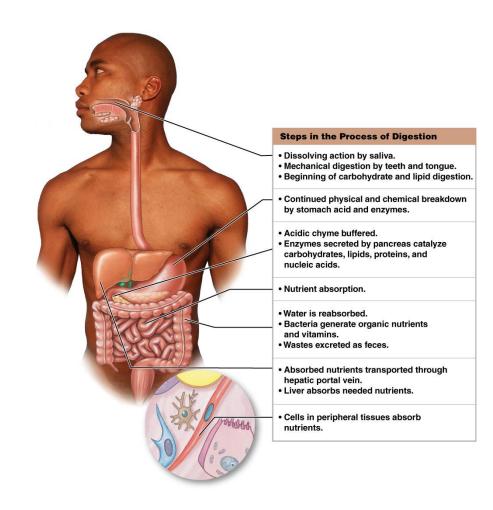
Module 23.8: Digestion involves a series of steps to make nutrients available to the body

Digestion

- Breakdown of the physical structure and chemical bonds of organic compounds
- Resulting compounds are absorbed in the digestive tract and distributed through the bloodstream
- Cells throughout the body utilize these organic molecules for energy and to restock the intracellular nutrient pool

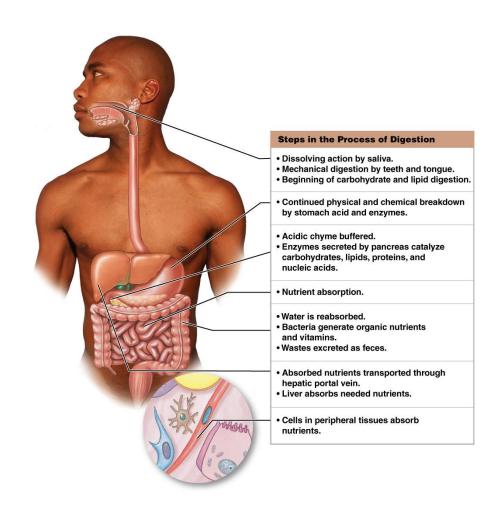
Oral cavity

- Saliva dissolves some organic molecules
- Mechanical processing by teeth and tongue disrupts physical structures (provides access for digestive enzymes)
- Chemical digestion of complex carbohydrates and lipids



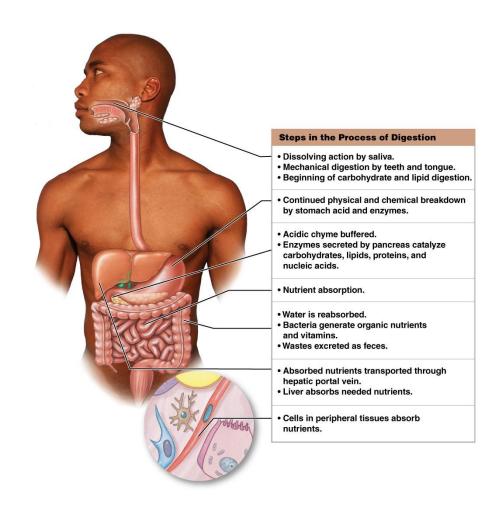
Stomach

- Material further broken down physically and chemically by stomach acid and enzymes
- Protein digestion begins



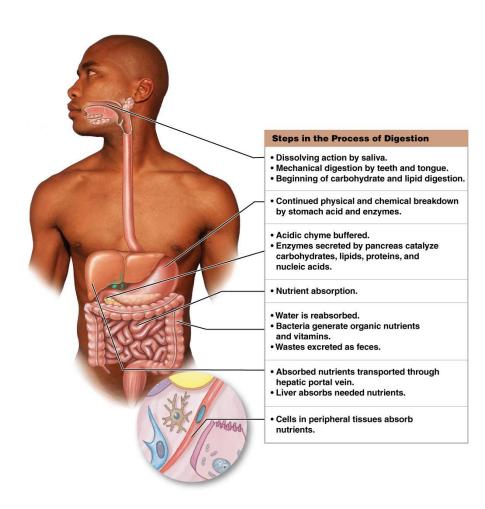
Duodenum

- Buffers from the pancreas and liver moderate the pH of the arriving chyme
- Digestive enzymes catalyze catabolism of all organic molecules needed by cells



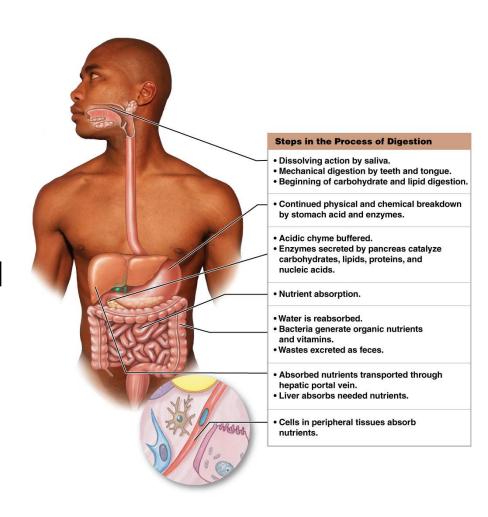
Jejunum and ileum

- Most nutrient absorption occurs here
- Nutrients delivered to the liver by the hepatic portal vein, where they are absorbed as needed



Large intestine

- Water reabsorbed
- Bacterial action generates organic nutrients and vitamins
- Indigestible material and wastes are excreted as feces



Module 23.8: Review

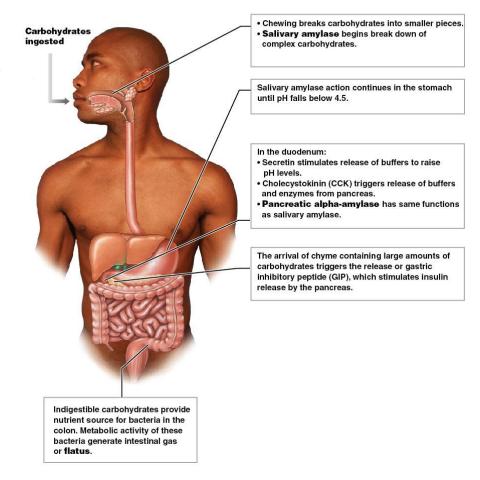
- A. Why is digestion important?
- B. Where does most nutrient absorption occur?
- C. Most of the absorbed nutrients enter into which blood vessel?

Learning Outcome: Outline the steps involved in digestion, and list the nutrients used by the body.

Module 23.9: Carbohydrates are usually the preferred substrates for catabolism and ATP production under resting conditions

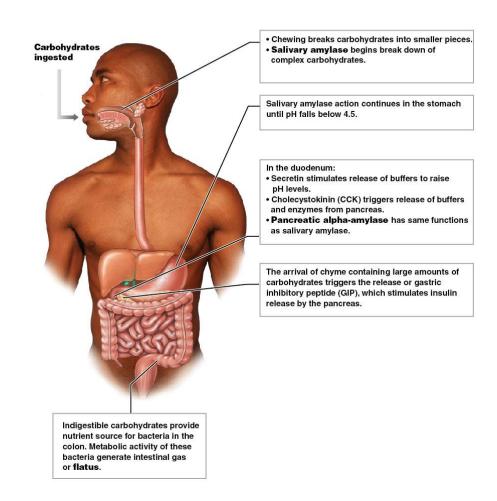
Steps of carbohydrate digestion

- Begins in the mouth
 - Chewing mechanically breaks apart the bolus and mixes with saliva
 - Salivary amylase
 breaks down complex
 carbohydrates into
 disaccharides and
 trisaccharides



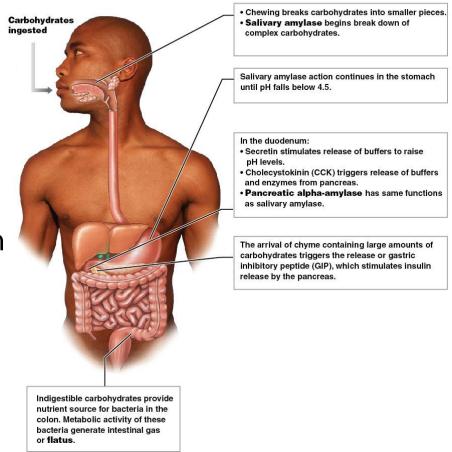
Steps of carbohydrate digestion (continued)

- Amylase is deactivated by the acidity in the stomach
- Active for about
 1–2 hours, until pH
 falls below 4.5



Steps of carbohydrate digestion (continued)

- Chyme arrives in the duodenum
 - Secretin signals buffer release
 - CCK triggers release of pancreatic enzymes
 - Pancreatic alphaamylase continues carbohydrate digestion

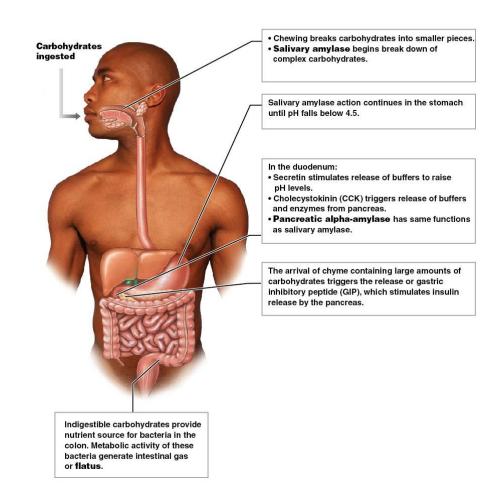


Steps of carbohydrate digestion (continued)

Arrival of chyme with carbohydrates triggers release

of gastric inhibitory protein (GIP)

 Stimulates insulin release



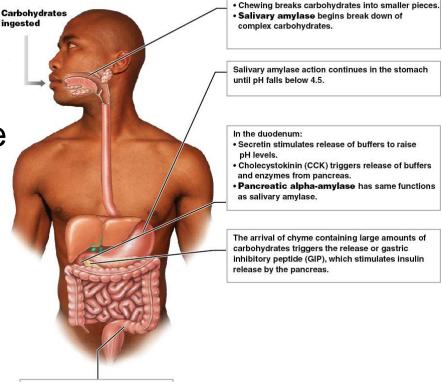
Steps of carbohydrate digestion (continued)

In the jejunum, brush border enzymes finish

carbohydrate digestion down to simple sugars (monosaccharides)

 Maltase (digests maltose into glucose + glucose)

- Sucrase (digests sucrose into glucose + fructose)
- Lactase (digests lactose into glucose + galactose)



Indigestible carbohydrates provide nutrient source for bacteria in the colon. Metabolic activity of these bacteria generate intestinal gas

or flatus.

Carbohydrate absorption and transport

- Simple sugars are transported across the small intestine epithelial cells
 - Leave cells by facilitated diffusion through basolateral surface
- Diffuse into capillaries
- Transported to the liver by the hepatic portal vein
- Released into blood to maintain normal glucose levels (~90 mg/dL)
- Liver converts excess glucose to glycogen

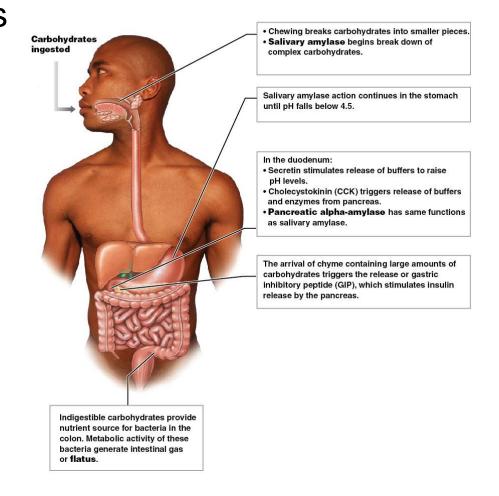
Carbohydrate digestion Liver Simple sugars are absorbed and Excess stored as converted to glucose as needed. glycogen or used Glucose is released into the blood in other ways sufficient to maintain normal glucose levels (~90 mg/dL). Hepatic portal vein **GLUCOSE** (6-carbon) · Epithelial cells lining the jejunum Disaccharides secrete enzymes. · Maltase, sucrase, and lactase break down disaccharides into simple sugars. Enzymes Simple sugars (such as glucose or fructose) · Monosaccharides are transported across plasma membrane and Epithelial cell into cytosol then across basolateral surface and Into capillaries. Simple sugars

Steps of carbohydrate digestion (continued)

In the large intestine, remaining indigestible

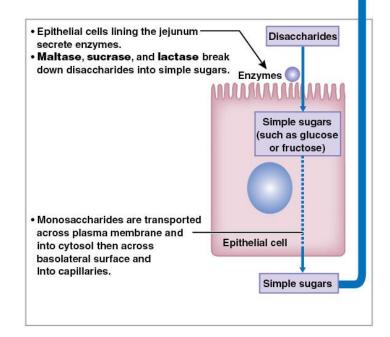
carbohydrates (such as cellulose) provide nutrient source for bacteria in the colon

 Bacterial metabolic activity generates intestinal gas (flatus)

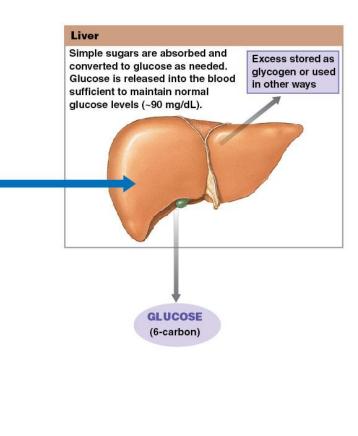


 Simple sugars are absorbed and converted to glucose as needed

Excess sugar is stored as glycogen

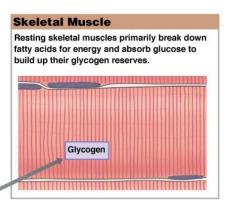


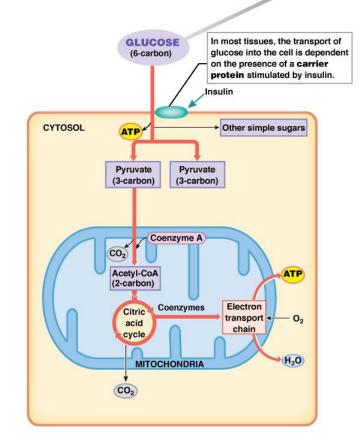
Hepatic portal vein



Module 23.9: Carbohydrates

 Resting skeletal muscle breaks down fatty acids for energy and absorbs glucose to build glycogen reserves



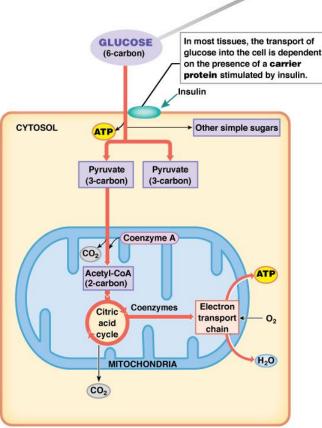


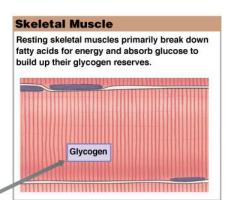
Module 23.9: Carbohydrates

In most tissues, glucose is transported into cell by **carrier protein** (regulated by insulin)

May be converted to ribose or

glycerol





Module 23.9: Carbohydrates

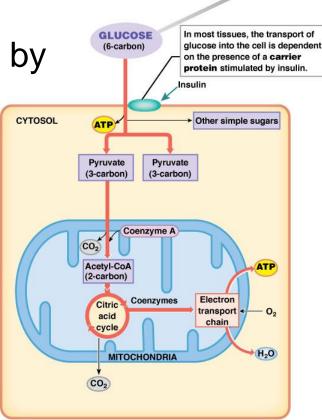
 If needed for energy, glucose is converted to 2 pyruvate molecules in glycolysis

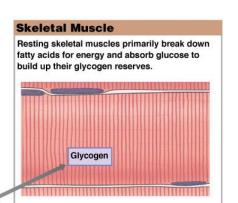
Produces 2 ATP for every glucose molecule

Pyruvates are used by

mitochondria

 For each pair of pyruvates, cell uses 6 O₂ and generates 6 CO₂, 12 H₂O, 30–32 ATP





Module 23.9: Review

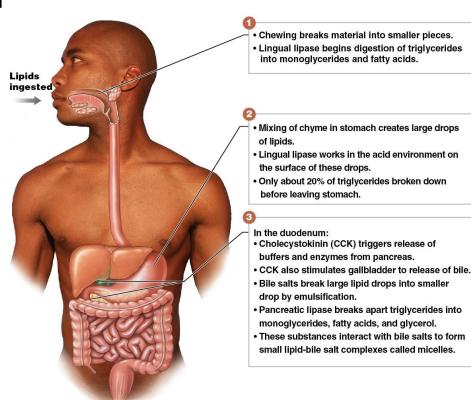
- A. Describe the source of intestinal gas.
- B. Explain the role of glycogen in cellular metabolism.
- C. Explain why carbohydrates are preferred over proteins and lipids as an energy source.

Learning Outcome: Describe carbohydrate metabolism.

Module 23.10: Lipids reach the bloodstream in chylomicrons; the cholesterol is then extracted and released as lipoproteins

Lipid digestion

- Begins in the mouth
 - Mechanical breakdown into smaller chunks
 - Lingual lipase breaks down triglycerides into monoglycerides and fatty acids

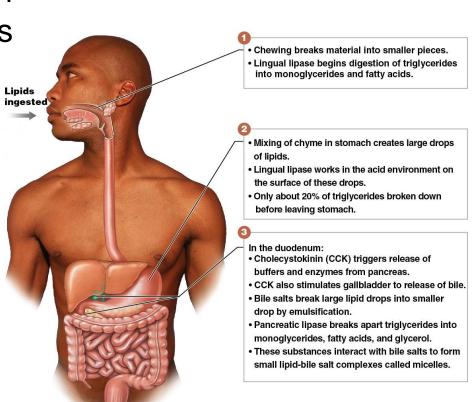


Lipid digestion (continued)

- Continues in the stomach
 - Mixing of chyme in the stomach creates large drops containing a variety of lipids

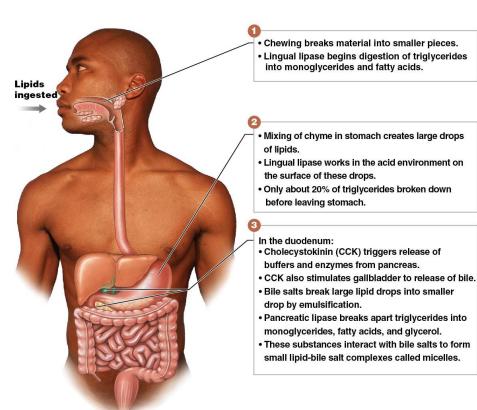
 Lingual lipase continues to function

 Only about 20 percent of triglycerides are broken down by the time chyme leaves the stomach



Lipid digestion (continued)

- Completes in the duodenum
 - Release of CCK triggers contraction of gallbladder
 - Ejects bile into the duodenum
 - Bile salts break up lipid drops into smaller droplets in a process called emulsification



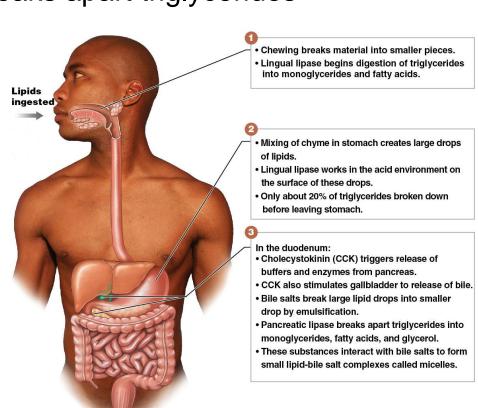
Lipid digestion (continued)

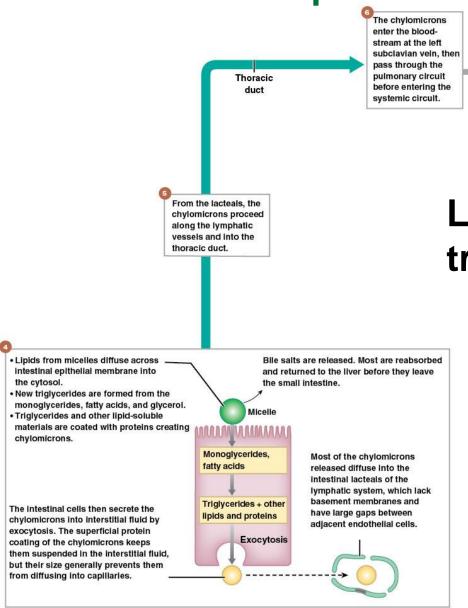
- Completes in the duodenum (continued)
 - CCK stimulates secretion of pancreatic enzymes

Pancreatic lipase breaks apart triglycerides

to form mixture of fatty acids, monoglycerides, and glycerol

Released molecules form micelles
 (lipid-bile salt complexes)





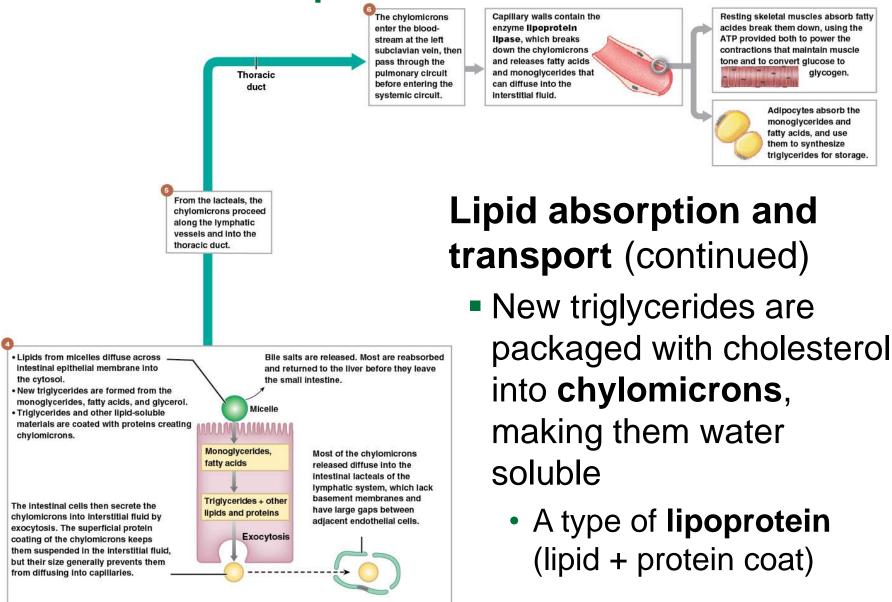
Capillary walls contain the enzyme lipoprotein lipase, which breaks down the chylomicrons and releases fatty acids and monoglycerides that can diffuse into the interstitial fluid.

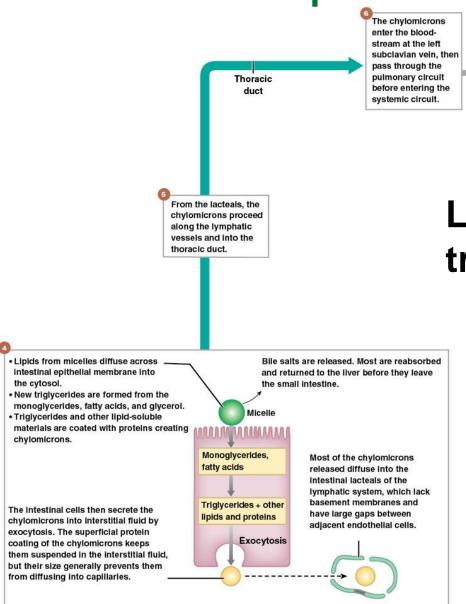
Resting skeletal muscles absorb fatty acides break them down, using the ATP provided both to power the contractions that maintain muscle tone and to convert glucose to glycogen.

Adipocytes absorb the monoglycerides and fatty acids, and use them to synthesize triglycerides for storage.

Lipid absorption and transport

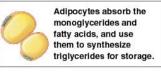
- Micelle contacts the intestinal epithelium
 - Lipids diffuse across the plasma membrane
 - Bile salts are released
- Cells synthesize new triglycerides from the absorbed lipids





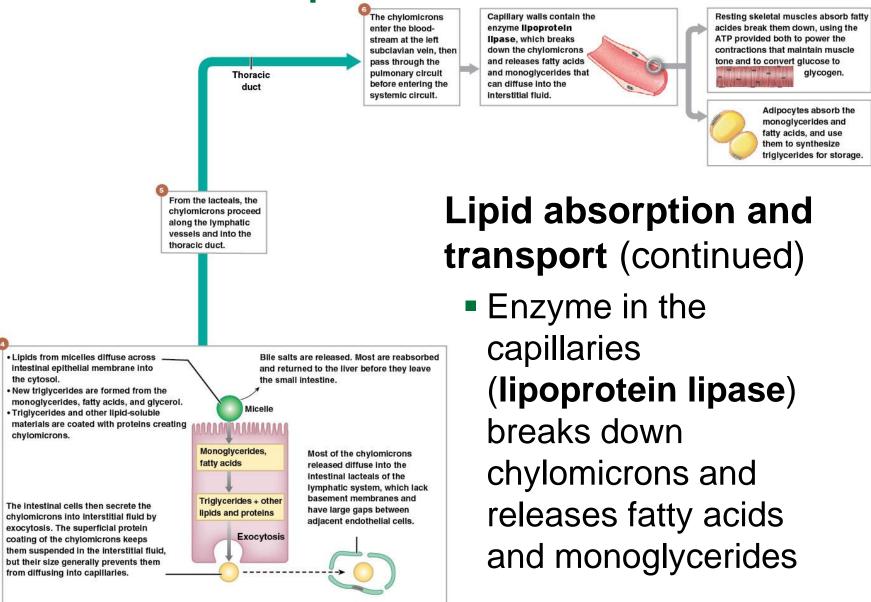
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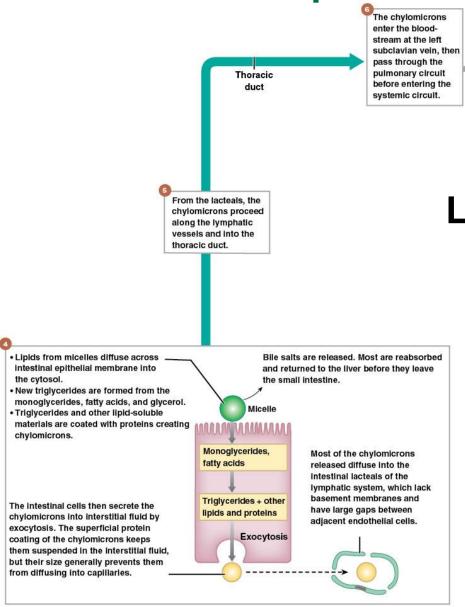
Resting skeletal muscles absorb fatty acides break them down, using the ATP provided both to power the contractions that maintain muscle tone and to convert glucose to glycogen.



Lipid absorption and transport (continued)

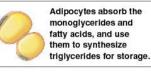
- Chylomicrons are secreted by exocytosis
 - Then diffuse into lacteals (lymphatic capillaries)
 - Carried by the lymphatic vessels into the thoracic duct and then into the bloodstream





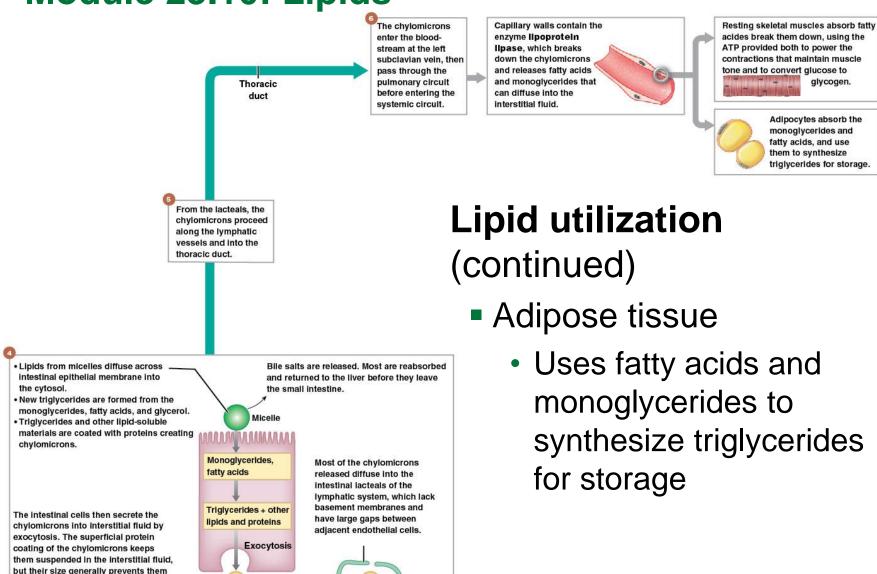
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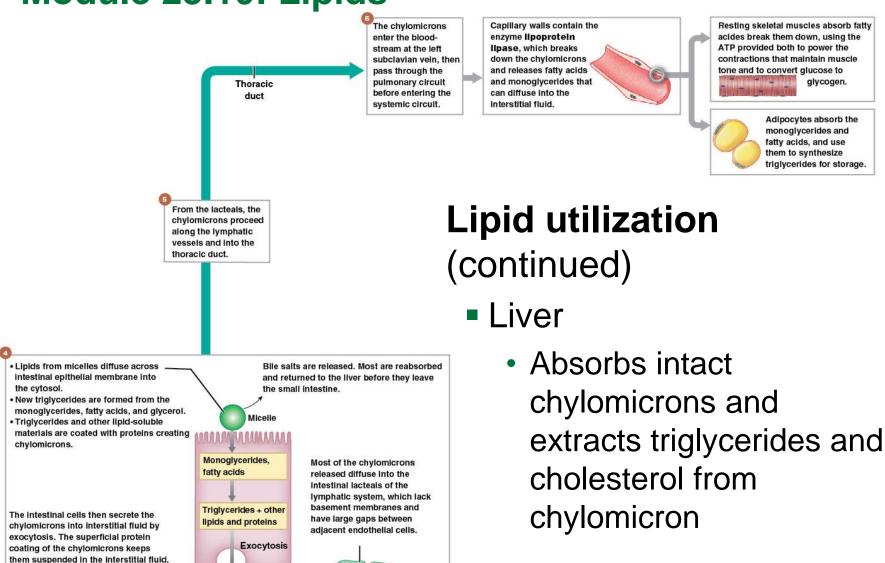
Lipid utilization

- Tissues that use or process digested lipids
 - Skeletal muscles
 - Use fatty acids to generate ATP for contraction and to convert glucose to glycogen



glycogen.

from diffusing into capillaries.



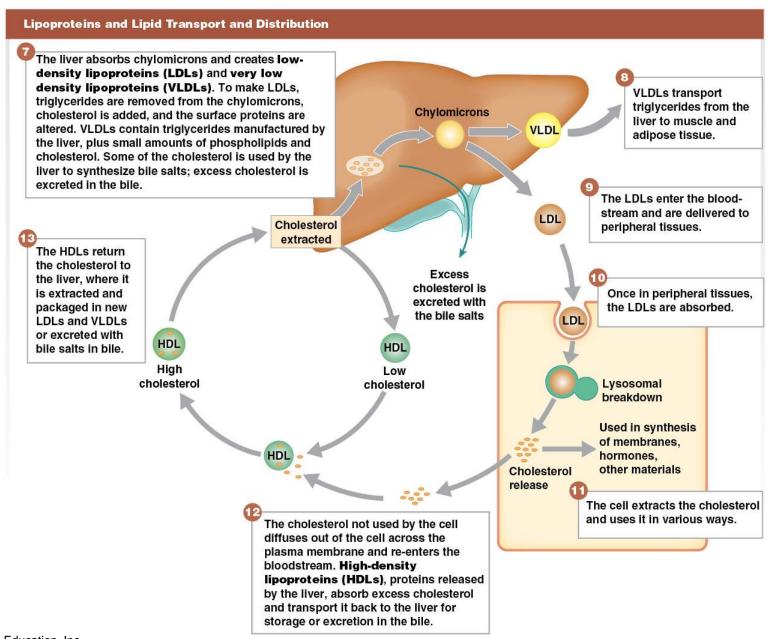
glycogen.

but their size generally prevents them from diffusing into capillaries.

Lipid processing in the liver

- From absorbed chylomicrons, the liver removes the triglycerides, adds cholesterol, and alters the surface proteins
- Creates low-density lipoproteins (LDLs) and very low density lipoproteins (VLDLs)
 - VLDLs transport triglycerides to muscle and adipose tissue
 - LDLs deliver cholesterol to peripheral tissues
 - Cells extract and use the cholesterol to build membranes, hormones, and other materials
 - Excess cholesterol diffuses back into the bloodstream

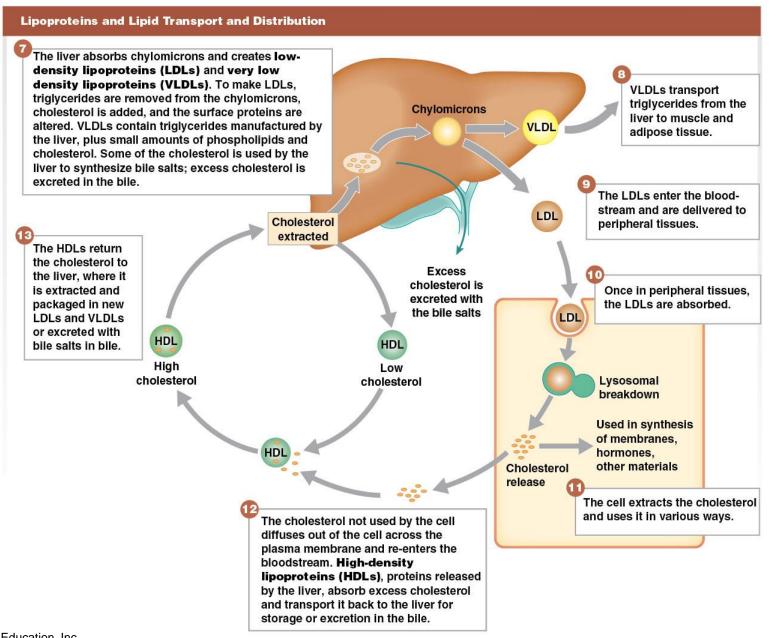
Lipoproteins and lipid transport



Lipid processing in the liver (continued)

- High-density lipoproteins (HDLs)
 - Proteins released by the liver
 - Absorb excess cholesterol in the bloodstream and return it to the liver
 - Cholesterol used to synthesize more LDLs or VLDLs
 - Also used to synthesize bile salts

Lipoproteins and lipid transport (continued)



Applications to health

- Indicators of potential cardiovascular problems
 - Total cholesterol above 200 mg/dL
 - High LDL:HDL ratio
- These values show:
 - 1. High levels of cholesterol in circulation
 - 2. Most of the cholesterol is going into the tissues and staying instead of returning to the liver
- Consequences
 - Excess cholesterol can accumulate as plaques in blood vessels, causing heart attacks and strokes

Module 23.10: Review

- A. What is the difference between a micelle and a chylomicron?
- B. What does the liver do with the chylomicrons it receives?
- C. Describe the roles of LDLs and HDLs.

Learning Outcome: Describe the mechanisms of lipid transport and distribution.

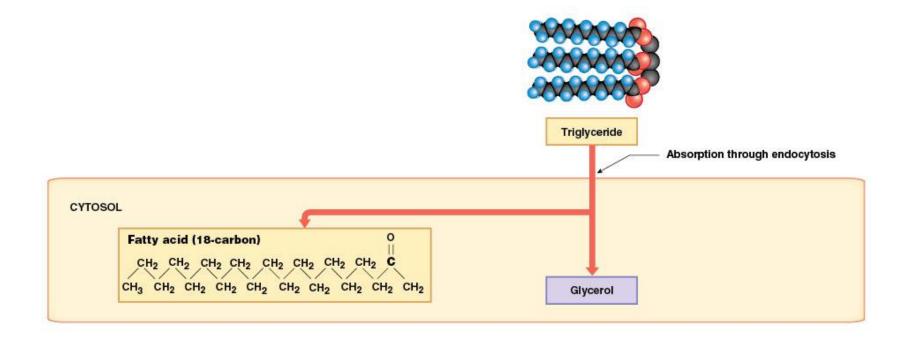
Module 23.11: Fatty acids can be broken down to provide energy or converted to other lipids

Lipolysis

- Lipid catabolism in which lipids are broken down into pieces that:
 - Can be converted to pyruvate
 - Can be channeled directly into the citric acid cycle
- More efficient than glucose catabolism
 - 6-carbon glucose yields 36 ATP
 - 6 carbons from fatty acids yield 51 ATP

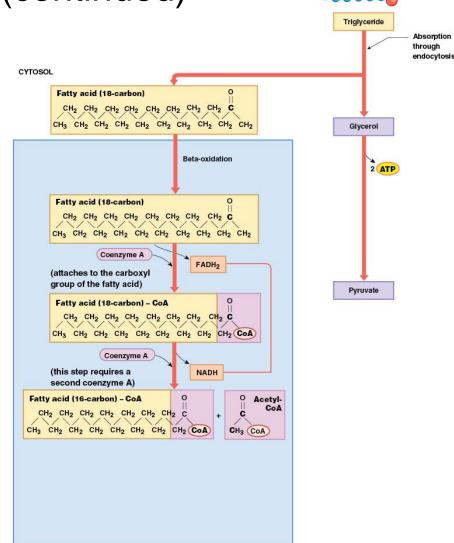
Triglyceride catabolism

- Lysosomal enzymes break down triglycerides into:
- 1 glycerol molecule
- 3 fatty acid molecules



Triglyceride catabolism (continued)

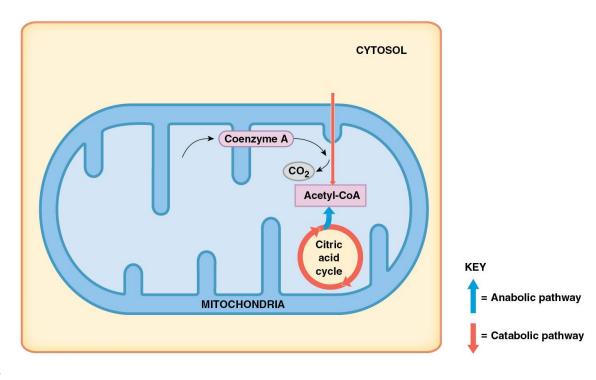
- Glycerol is converted to pyruvate in glycolysis (yields 2 ATP)
- Fatty acids are catabolized to acetyl-CoA through beta-oxidation



Fatty acid metabolism Triglyceride Absorption through endocytosis CYTOSOL Fatty acid (18-carbon) CH2 CH2 CH2 CH2 CH2 CH2 CH2 CH2 C CH3 CH2 CH2 CH2 CH2 CH2 CH2 CH2 CH2 CH2 Glycerol Beta-oxidation 2 ATP Fatty acid (18-carbon) CH2 CH2 CH2 CH2 CH2 CH2 CH2 CH2 C CH3 CH2 CH2 CH2 CH2 CH2 CH2 CH2 CH2 CH2 Coenzyme A FADH₂ (attaches to the carboxyl group of the fatty acid) Pyruvate Fatty acid (18-carbon) - CoA CH2 CH2 CH2 CH2 CH2 CH2 CH2 CH2 C CH3 CH2 CH2 CH2 CH2 CH2 CH2 CH2 COA Coenzyme A Coenzyme A (this step requires a NADH CO2 K second coenzyme A) 13 ATP Fatty acid (16-carbon) - CoA Acetyl-Acetyl-CoA CoA CH2 CH2 CH2 CH2 CH2 CH2 CH2 C CH3 CH2 CH2 CH2 CH2 CH2 CH2 CH2 COA CH3 COA Electron Coenzymes acid 02 transport cycle chain

Lipogenesis

- Synthesis of lipids
- Begins with acetyl-CoA
 - Almost any organic substrate (lipids, amino acids, carbohydrates) can be converted to acetyl-CoA

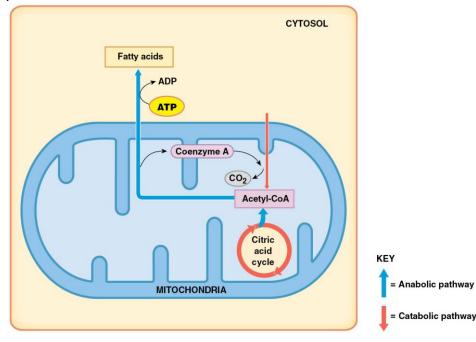


Lipogenesis (continued)

- Some fatty acids are synthesized from acetyl-CoA
 - Involves a series of enzymatic steps different from beta-oxidation, using different enzymes

Essential fatty acids

- Cannot be synthesized; must be obtained from diet
- Examples: linolenic acid (omega-3 fatty acid) and linoleic acid (omega-6 fatty acid)



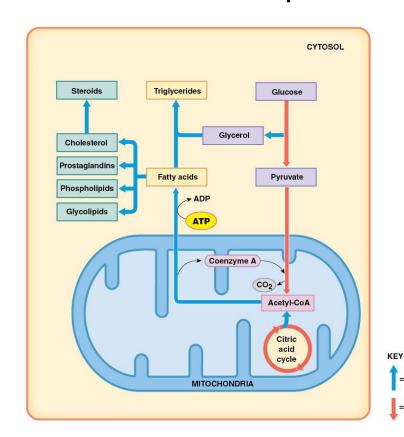
Lipogenesis (continued)

 Structural and functional lipids are created from fatty acids

Glycerol is synthesized from intermediate products

of glycolysis

Fatty acids + glycerol = triglycerides



= Anabolic pathway

= Catabolic pathway

Lipids as energy reserves

- Useful because:
 - Beta-oxidation is very efficient
 - Excess lipids can be easily stored as triglycerides
- However:
 - Cannot provide large amounts of ATP quickly
 - Difficult for water-soluble enzymes to access the insoluble droplets
 - Well suited for chronic energy demands during stress or starvation

Module 23.11: Review

- A. Define beta-oxidation.
- B. Which molecule is a key reactant in both ATP production from fatty acids and lipogenesis?
- C. Identify the fates of fatty acids.

Learning Outcome: Describe the fate of fatty acids in lipid metabolism.

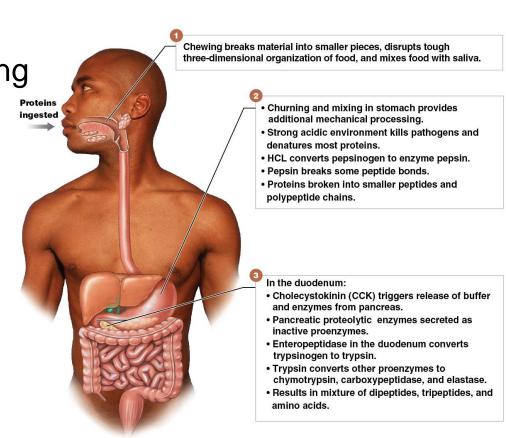
Module 23.12: An amino acid not needed for protein synthesis may be broken down or converted to a different amino acid

Protein digestion

In the mouth, mechanical processing occurs

In the stomach

 Mechanical processing continues with churning and mixing

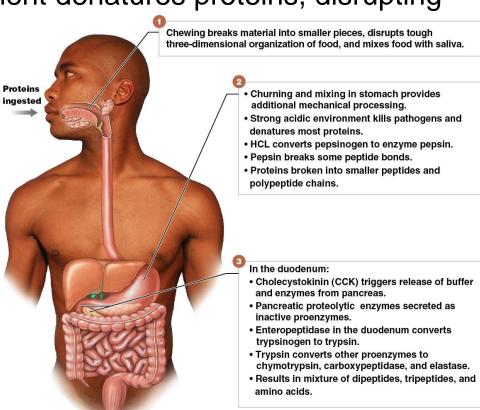


Protein digestion (continued)

- In the stomach (continued)
 - Chemical digestion begins

- Strong acid environment denatures proteins, disrupting

secondary and tertiary structures



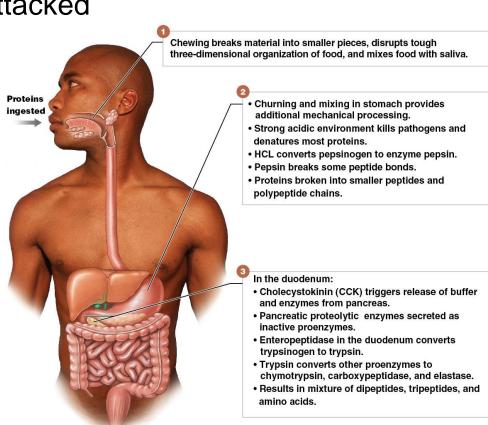
Protein digestion (continued)

- In the stomach (continued)
 - Chemical digestion (continued)

Peptide bonds are attacked

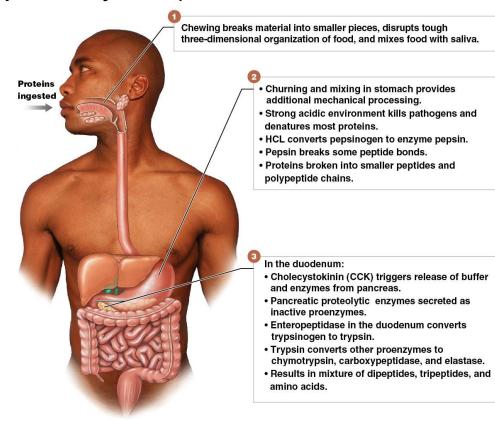
by **pepsin** (secreted from chief cells)

 Digests proteins into smaller peptide and polypeptide chains



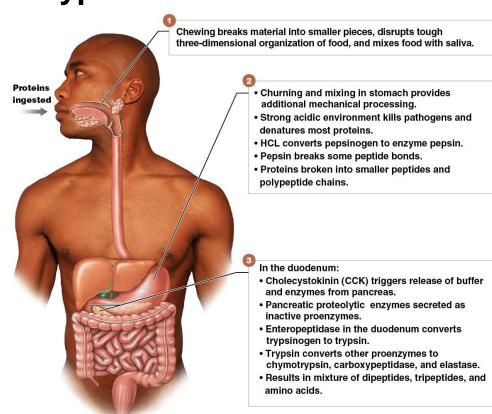
Protein digestion (continued)

- In the duodenum
 - CCK stimulates secretion of pancreatic enzymes (released as inactive proenzymes)



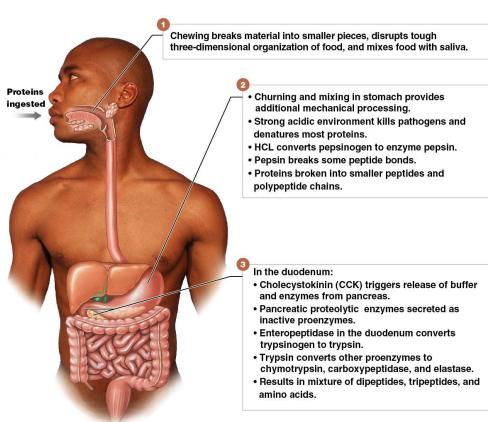
Protein digestion (continued)

- In the duodenum (continued)
 - Enteropeptidase (from the duodenal epithelium) converts trypsinogen to trypsin



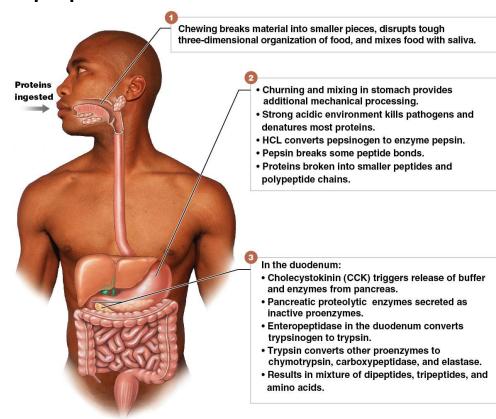
Protein digestion (continued)

- In the duodenum (continued)
 - Trypsin activates other pancreatic proenzymes to their active forms
 - Chymotrypsin,
 carboxypeptidase,
 and elastase



Protein digestion (continued)

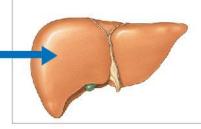
- In the duodenum (continued)
 - Activated pancreatic enzymes digest specific peptide bonds, producing short peptides and amino acids



- Peptidases from the small intestine epithelium
 - Break peptide bonds and release amino acids
 - Example: dipeptidases break dipeptides into single amino acids

Short peptides, amino acids The epithelial surfaces of the small Dipeptidase intestine contain several peptidaseenzymes that break peptide bonds Facilitated diffusion notably dipeptidases that break and cotransport dipeptides apart and release individual Amino acids amino acids. These amino acids, as well as those released by the action of pancreatic enzymes, are absorbed through both facilitated diffusion and cotransport mechanisms. After diffusing to the basal surface of the cell, the amino acids are released into Epithelial cell interstitial fluid by facilitated diffusion and Facilitated diffusion cotransport. Once in the interstitial fluid. and cotransport the amino acids enter interstitial capillaries for transport to the liver in the hepatic Amino acids portal vein.

Blood amino acid levels normally range between 35 and 65 mg/dL, but they may become elevated after a protein-rich meal. The liver itself uses many amino acids for synthesizing plasma proteins, and it has all of the enzymes needed to synthesize, convert, or catabolize amino acids. In addition, amino acids that can be broken down to 3-carbon molecules can be used for gluconeogenesis when other sources of glucose are unavailable.



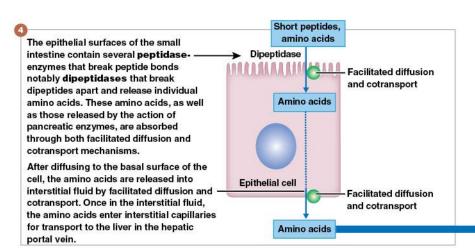
Hepatic -

portal vein

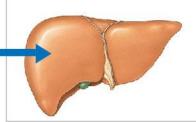
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Protein absorption and transport

- Short peptides and amino acids are absorbed into epithelial cells by:
 - Facilitated diffusion
 - Cotransport



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Hepatic portal

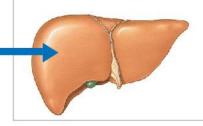
vein

Protein absorption and transport (continued)

 Amino acids are transported to liver through intestinal capillaries to hepatic portal vein

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Hepatic portal

vein

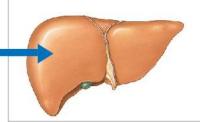
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Protein absorption and transport (continued)

- In the liver
 - Control of plasma amino acid levels is less precise than glucose

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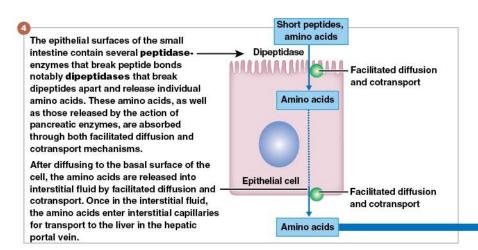
Hepatic portal

vein

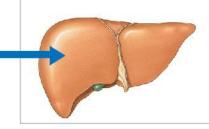
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Protein absorption and transport (continued)

- In the liver (continued)
 - Liver uses amino acids to:
 - Synthesize plasma proteins
 - Create 3-carbon molecules for gluconeogenesis



Blood amino acid levels normally range between 35 and 65 mg/dL, but they may become elevated after a protein-rich meal. The liver itself uses many amino acids for synthesizing plasma proteins, and it has all of the enzymes needed to synthesize, convert, or catabolize amino acids. In addition, amino acids that can be broken down to 3-carbon molecules can be used for gluconeogenesis when other sources of glucose are unavailable.

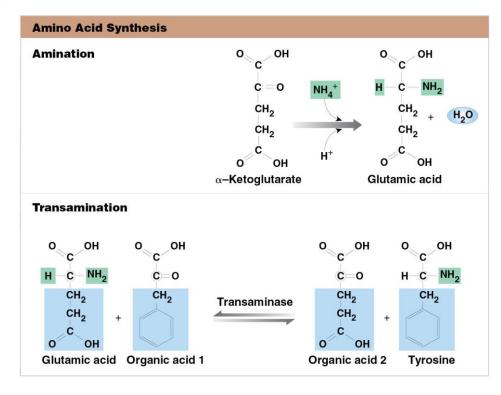


Hepatic portal

vein

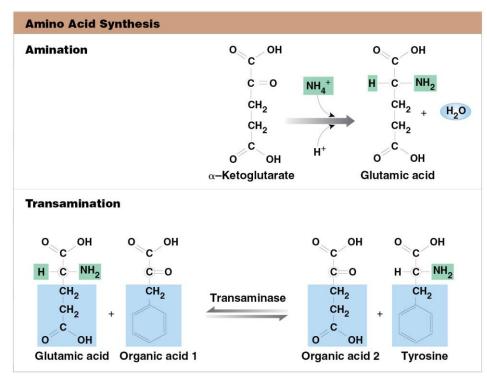
Amino acid synthesis

- Liver and body cells can produce 10 of the 20 amino acids required to synthesize needed proteins
 - Remaining 10 are essential amino acids
 - Cannot be produced



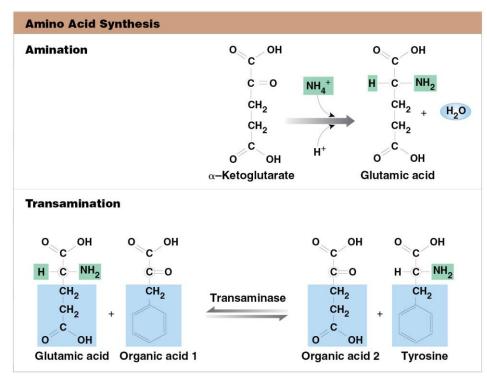
Amino acid synthesis (continued)

- Two methods of forming amino acids
 - Amination
 - Using an ammonium ion to form an amino group and attaching that group to a molecule



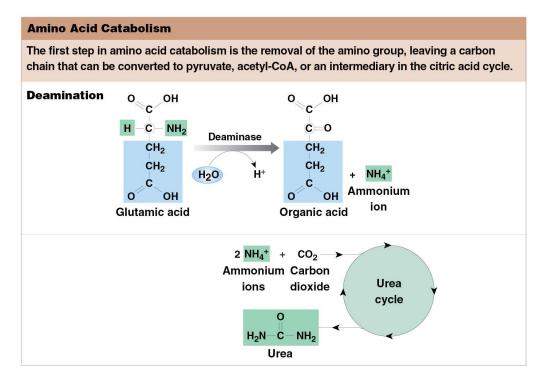
Amino acid synthesis (continued)

- Two methods of forming amino acids (continued)
 - Transamination
 - Transferring an amino group to another molecule, yielding a different amino acid



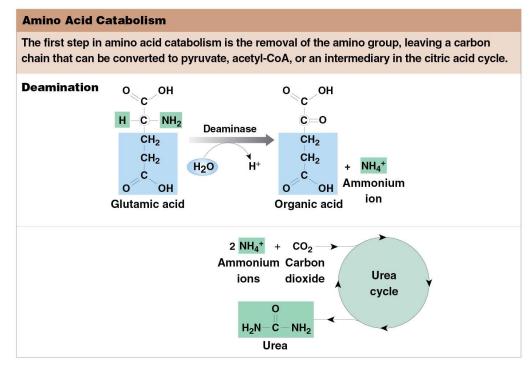
Amino acid catabolism

- Requires the removal of the amino group
 - Removal can be by transamination (transfer to another molecule)
 - Actual process of removal is called deamination



Amino acid catabolism (continued)

- Releases ammonium ions, which are toxic
- Enzymes in liver cells use ammonium ions to synthesize urea
 - Process called the urea cycle
 - Urea is then excreted in urine



Module 23.12: Review

- A. Describe the role of CCK release and its effects on proteins.
- B. In amino acid metabolism, identify the processes by which the amino group is removed.
- C. What happens to the ammonium ions that are removed from amino acids during deamination?

Learning Outcome: Summarize the main features of protein metabolism and the use of proteins as an energy source.

Module 23.13: There are two general patterns of metabolic activity: the absorptive and postabsorptive states

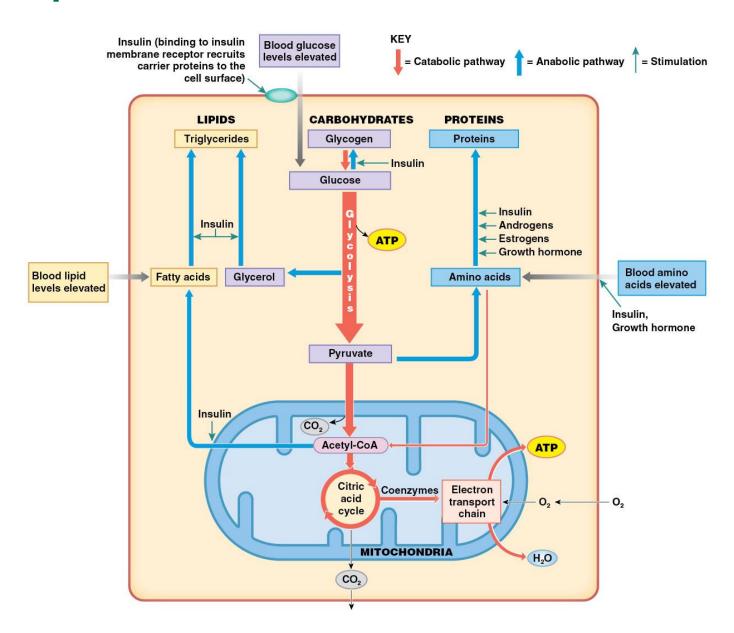
Absorptive state

- Time following a meal, when nutrient absorption is occurring
- Typically continues for ~4 hours
- Insulin is primary regulating hormone

In the Absorptive State

- Insulin stimulates
 (1) glucose uptakeand glycogenesis,(2) amino acid uptakeand protein synthesis,and (3) triglyceridesynthesis.
- Androgens, estrogens, and growth hormone also stimulate protein synthesis.
- Glycolysis and aerobic metabolism provide the ATP needed to power cellular activities as well as the synthesis of lipids and proteins.

Absorptive state



Module 23.13: Absorptive and postabsorptive states

Postabsorptive state

- Time when nutrient absorption is not occurring and body relies on internal energy reserves
- Metabolic activity focused on mobilizing energy reserves and maintaining blood glucose
- Coordinated by several hormones

Glucagon, epinephrine, glucocorticoids, growth hormone

In the Postabsorptive State

- Glucocorticoids stimulate the mobilization of lipid and protein reserves; these effects are enhanced by growth hormone.
- Glucagon stimulates glycogenolysis and gluconeogenesis, primarily in the liver. The release of glucose by the liver and the shift away from glucose metabolism by other tissues stabilizes blood glucose levels.
- Epinephrine is important in stimulating glycogenolysis in skeletal and cardiac muscle, and lipolysis in adipocytes.

Module 23.13: Absorptive and postabsorptive states

Postabsorptive state (continued)

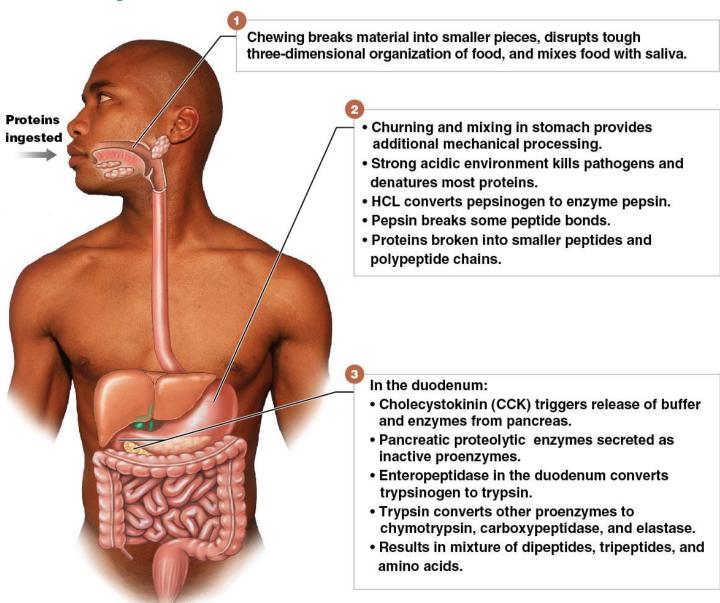
- Blood lipid levels decrease
 - Response is release of fatty acids by adipocytes
- Blood amino acid levels decrease
 - Response is amino acid release by skeletal muscles and other tissues
- Blood glucose levels decrease
 - Response is glucose release by liver

Module 23.13: Absorptive and postabsorptive states

Postabsorptive state (continued)

- Catabolism of lipids and amino acids in liver produce acetyl-CoA
 - Leads to formation of ketone bodies
 - Diffuse into blood and are used by other cells as energy source

Postabsorptive state



Module 23.13: Review

- A. Define and describe the absorptive state.
- B. When and how do ketone bodies form?
- C. How do the absorptive and postabsorptive states maintain normal blood glucose levels?

Learning Outcome: Differentiate between the absorptive and postabsorptive metabolic states, and summarize the characteristics of each.

Module 23.14: Vitamins are essential to the function of many metabolic pathways

Nutrition

Absorption of nutrients from food

Vitamins

 Organic compounds required in very small quantities for essential metabolic activities

- Two classes
 - Fat-soluble vitamins
 (A, D₃, E, and K)
 - Water-soluble vitamins
 (B vitamins and C)



Minerals

 Inorganic nutrients that serve as electrolytes in body fluids



Fat-soluble vitamins

- Absorbed primarily from digestive tract with lipid contents of micelles
- Sources
 - Vegetables for vitamins A, E, K
 - Vitamin D₃ is produced in skin when exposed to sunlight
 - Vitamin K is produced by intestinal bacteria
- Stored in lipid deposits
 - Give large bodily reserves
 - Normal metabolic activities can continue for months without a dietary source

Vitamin	Significance	Sources	Recommended Daily Allowance (RDA) in mg	Effects of Deficiency	Effects of Excess
A	Maintains epithelia; required for synthesis of visual pigments; supports immune system; promotes growth and bone remodeling	Leafy green and yellow vegetables, liver, dairy products, and fish	0.7–0.9	Retarded growth, night blindness, deterioration of epithelial membranes	Liver damage, skin paling, CNS effects (nausea, anorexia)
D	Required for normal bone growth, intestinal calcium and phosphorus absorption, and retention of these ions by the kidneys	Synthesized in skin exposed to sunlight; fortified dairy products and egg yolks	0.005-0.015*	Rickets, skeletal deterioration	Calcium deposits in many tissues, disrupting functions
E	Prevents breakdown of vitamin A and fatty acids	Meat, milk, vegetables	15	Anemia, other problems suspected	Nausea, stomach cramps, blurred vision, fatigue
К	Essential for liver synthesis of prothrombin and other clotting factors	Vegetables; production by intestinal bacteria	0.09-0.12	Bleeding disorders	Liver dysfunction, jaundice

Hypovitaminosis

- Vitamin deficiency disease
- Rarely occurs with fat-soluble vitamins (except for vitamin D₃)

Hypervitaminosis

- Dietary intake exceeds the body's ability to store, utilize, or excrete vitamins
- Can occur because metabolism from lipid reserves takes time
- Potentially harmful

Water-soluble vitamins

- Most are components of coenzymes
- Nutritional sources
 - B vitamins are found in meat, eggs, and dairy products
 - Vitamin C is found in citrus fruits
- Rapidly exchanged between digestive tract and blood
- Excess amounts readily excreted in urine
 - Hypervitaminosis rarely occurs unless ingesting megadoses of vitamin supplements

Water-soluble vitamins (continued)

- Most water-soluble vitamins are not stored in the body in significant amounts
 - Vitamins B₁₂ and C are the exceptions
- All but one are absorbed easily across the intestinal epithelium
 - Vitamin B₁₂ has to bind to intrinsic factor to be absorbed
- Insufficient intake can lead to initial symptoms of vitamin deficiency in days to weeks
- Intestinal bacteria produce small amounts of four vitamins (B₅, B₇, B₉, B₁₂)

The Water-Solu	The Water-Soluble Vitamins						
Vitamin	Component or Precursor of	Sources	Recommended Daily Allowance (RDA) in mg	Effects of Deficiency	Effects of Excess		
B ₁ (thiamine)	Coenzyme in many pathways	Milk, meat, bread	1.1–1.2	Muscle weakness, CNS and cardiovascular problems, including heart disease; called <i>beriberi</i>	Hypotension		
B ₂ (riboflavin)	Part of FAD, involved in multiple pathways, including glycolysis and citric acid cycle	Milk, meat, eggs and cheese	1.1–1.3	Epithelial and mucosal deterioration	Itching, tingling		
B ₃ (niacin)	Part of NAD, involved in multiple pathways	Meat, bread, potatoes	14–16	CNS, GI, epithelial, and mucosal deterioration; called <i>pellagra</i>	Itching, burning; vasodilation; death after large dose		
B ₅ (pantothenic acid)	Coenzyme A, in multiple pathways	Milk, meat	10	Retarded growth, CNS disturbances	None reported		
B ₆ (pyridoxine)	Coenzyme in amino acid and lipid metabolism	Meat, whole grains, vegetables, orange juice, cheese and milk	1.3–1.7	Retarded growth, anemia, convulsions, epithelial changes	CNS alterations, perhaps fatal		
B ₉ (folic acid)	Coenzyme in amino acid and nucleic acid metabolism	Leafy vegetables, some fruits, liver, cereal and bread	0.2-0.4	Retarded growth, anemia, gastrointestinal disorders, developmental abnormalities	Few noted, except at massive doses		
B ₁₂ (cobalamin)	Coenzyme in nucleic acid metabolism	Milk, meat	0.0024	Impaired RBC production, causing pernicious anemia	Polycythemia		
B ₇ (biotin)	Coenzyme in many pathways	Eggs, meat, vegetables	0.03	Fatigue, muscular pain, nausea, dermatitis	None reported		
C (ascorbic acid)	Coenzyme in many pathways	Citrus fruits	75–90; smokers add 35 mg	Epithelial and mucosal deterioration; called scurvy	Kidney stones		

Module 23.14: Review

- A. Define nutrition.
- B. Identify the two classes of vitamins.
- C. If vitamins do not provide a source of energy, what is their role in nutrition?

Learning Outcome: Explain the role of fat-soluble vitamins and water-soluble vitamins in metabolic pathways.

Module 23.15: Proper nutrition depends on eating a balanced diet

Balanced diet

- Contains all ingredients needed to maintain homeostasis
 - Substrates for ATP production
 - Essential amino acids
 - Fatty acids
 - Vitamins
 - Electrolytes
 - Water

Malnutrition

Unhealthy state from inadequate or excessive nutrient absorption

Know the Limits on Fats, Sugars, and Salt (Sodium)

- Get most of your fat from fish, nuts, and vegetable oils, which are key sources of essential fatty acids.
- Limit solid fats like butter, shortening, and lard that are high in saturated fats and cholesterol.
- Check the Nutrition Facts label to keep saturated fats, trans fats, and sodium low.
- Choose food and beverages low in added sugars. Added sugars contribute to excessive caloric intake.

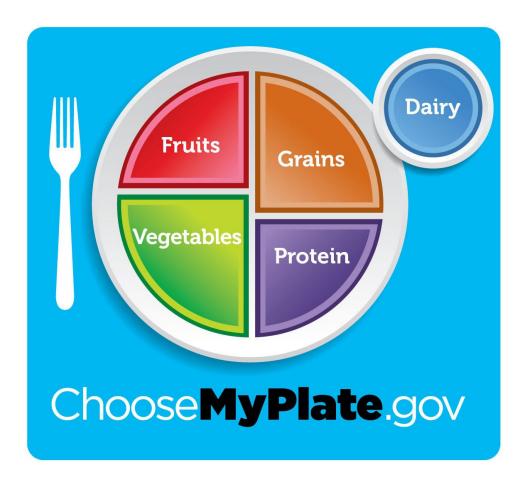
ChooseMyPlate.gov

- U.S. Department of Agriculture personalized eating plans based on current Dietary Guidelines for Americans
- Color-coded food groups indicate recommended proportions
 - Grains (orange)
 - Vegetables (green)
 - Fruits (red)
 - Dairy products (blue)
 - Proteins (purple)
 - Oils are not included in the MyPlate graphic and should be used sparingly

Find Your Balance between Food and Physical Activity

- Stay within your daily calorie needs.
- Adults should be physically active for at least 30 minutes most days of the week.
- About 60 minutes a day of physical activity may be needed to prevent weight gain.
- To sustain weight loss, at least 60 to 90 minutes a day of physical activity may be required.
- Children and teenagers should be physically active for 60 minutes most days.

Choose MyPlate.gov



GRAINS

Make half your grains whole

VEGETABLESVary your veggies

FRUITS
Focus on fruits

0 I L S

DAIRY

Get your calcium-rich foods

PROTEINS

Go lean with proteins

Food energy content

- Common units are calories
 - 1 calorie = energy to raise temperature of 1 g of water by 1°C
- Kilocalorie (kcal) or Calorie (Cal) is used

when referring to metabolism of the entire body

 1 Cal = energy to raise temperature of 1 kg of water by 1°C

Energy Content of Some Food Types				
Food Type	Serving	Cal		
Breakfast bar	1 bar	368		
Long-grain rice	1.5 cup	308		
Bread, whole wheat	4 slices	277		
Butter	0.5 tbsp	51		
Beer, regular	12 fl oz	160		
Cola, regular	12 fl oz	140		

Food energy content (continued)

- Energy yield of different nutrients varies
 - Carbohydrates: 4.18 Cal/g
 - Proteins: 4.32 Cal/g
 - Lipids: 9.46 Cal/g
- Average adult needs 2000–3000 Cal daily

Nutrient Group	Provides	Health Effects	
Grains (recommended: at least half of total eaten should be whole grains)	Carbohydrates; vitamins E, thiamine, niacin, folate; calcium; phosphorus; iron; sodium; dietary fiber	Whole grains prevent rapid rise in blood glucose levels and consequent rapid rise in insulin levels	
Vegetables (recommended: especially dark- green and orange vegetables)	Carbohydrates; vitamins A, C, E, folate; dietary fiber; potassium	Reduce risk of cardiovascular disease; protect against colon cancer (folate) and prostate cancer (lycopene in tomatoes)	
Fruits (recommended: a variety of fruit each day)	Carbohydrates; vitamins A, C, E, folate; dietary fiber; potassium	Reduce risk of cardiovascular disease; protect against colon cancer (folate)	
Dairy (recommended: low-fat or fat-free milk, yogurt, and cheese)	Complete proteins; fats; carbohydrates; calcium; potassium; magnesium; sodium; phosphorus; vitamins A, B ₁₂ , pantothenic acid, thiamine, riboflavin	Good source of calcium, which strengthens bones. Whole milk: High in calories, may cause weight gain; saturated fats associated with heart disease	
Meat, Fish, Eggs, and Legumes (recommended: lean meats, fish, poultry, eggs, beans, peas, lentils)	Complete proteins; fats; calcium; potassium; phosphorus; iron; zinc; vitamins E, thiamine, B ₆	Fish and poultry lower risk of heart disease and colon cancer (compared to red meat). Consumption of up to one egg per day does not appear to increase incidence of heart disease; nuts and legumes improve blood cholesterol ratios, lower risk of heart disease and diabetes	

Complete/incomplete proteins

- Complete proteins
 - Provide all the essential amino acids in sufficient quantities
 - From beef, fish, poultry, eggs, and dairy products



Complete/incomplete proteins (continued)

- Incomplete proteins
 - Lack one or more essential amino acids
 - Combinations of these can meet all amino acid requirements
 - Consideration for vegetarians and vegans



Module 23.15: Review

- A. Define balanced diet.
- B. Distinguish between a complete protein and an incomplete protein.
- C. Of these—carbohydrates, lipids, or proteins which releases the greatest amount of energy per gram during catabolism?

Learning Outcome: Explain what constitutes a balanced diet and why such a diet is important.

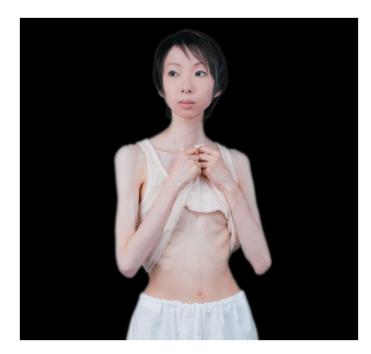
Module 23.16: Clinical Module: Metabolic disorders may result from nutritional or biochemical problems

Eating disorders

 Psychological problems resulting in inadequate or excessive food consumption

Anorexia nervosa

- Self-induced starvation or lack/loss of appetite
- Most common in adolescent Caucasian females
 - Weights commonly 30 percent below normal
- Death rates in severe cases range from 10 to 15 percent



Eating disorders (continued)

Bulimia

- Condition in which individual "binges," eating 20,000 calories or more, then induces vomiting
- Often uses laxatives and diuretics as well
- More common than anorexia nervosa
- Generally affects adolescent females

Eating disorders (continued)

- Bulimia (continued)
 - Health risks from
 - Cumulative damage to the stomach, esophagus, oral cavity, and teeth from repeated exposure to stomach acids
 - 2. Electrolyte imbalances
 - 3. Edema
 - 4. Cardiac arrhythmias

Obesity

- Defined as a condition of being >20 percent over ideal weight
- Linked with serious health risks (diabetes, hypertension, hypercholesterolemia)



Obesity (continued)

- U.S. Centers for Disease Control (CDC) estimates obesity percentages at:
 - 35.7 percent of U.S. adults
 - ~17 percent of U.S. children
- Caused by food energy intake greater than energy usage



Obesity (continued)

Two major categories

1. Regulatory obesity

- Failure to regulate food intake
- Most common form

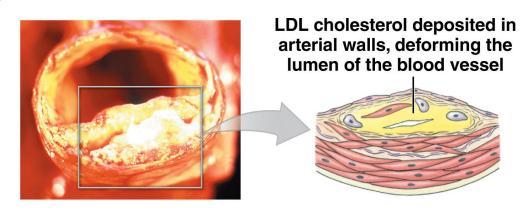
2. Metabolic obesity

- Secondary to underlying malfunction in cell/tissue metabolism
- Relatively rare



Elevated cholesterol levels

- Associated with development of atherosclerosis and coronary artery disease
- Recommended cholesterol intake <300 mg/day
- High LDL levels can lead to deposits in peripheral tissues such as blood vessels
- Low HDL levels (<35 mg/dL) also a problem
 - Excess cholesterol delivered to tissues is not easily returned to the liver



Nutritional/metabolic disorders

Phenylketonuria, or PKU

Nutritional/Metabolic Disorders

Phenylketonuria

Several inherited metabolic disorders result from an inability to produce specific enzymes involved in amino acid metabolism. People with **phenylketonuria** (fen-il-kē-tō-NŪ-rē-uh), or **PKU**, for example, cannot convert phenylalanine to tyrosine. This reaction is an

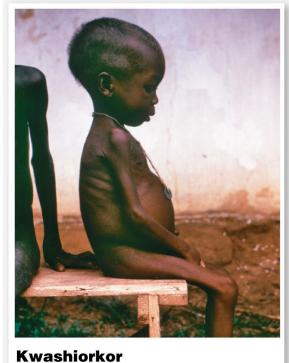
essential step in the synthesis of norepinephrine, epinephrine, dopamine, and melanin. If PKU is not detected in infancy, central nervous system development is inhibited, and severe brain damage results. The condition is common enough that a warning is printed on the packaging of products, such as diet drinks, that contain phenylalanine.

Nutritional/metabolic disorders (continued)

- Protein deficiency disease
 - Protein synthesis decreases throughout the body
 - Plasma protein synthesis in the liver also affected
 - Reduced levels of plasma proteins lead to reduced plasma osmolarity
 - Fluid shifts out of capillaries and accumulates in the interstitial spaces (edema) and peritoneal cavity (ascites)

Nutritional/metabolic disorders (continued)

- Protein deficiency disease (continued)
 - Example: kwashiorkor
 - Occurs in children whose protein intake is inadequate,
 - even if total caloric intake is sufficient
 - Complications include damage to the developing brain



Nutritional/metabolic disorders (continued)

Ketoacidosis

- Acidification of blood due to ketone body production
 - Leads to ketosis
- Occurs when glucose supplies are limited
 - Fatty acid and amino acid catabolism in liver leads to acetyl-CoA production and generation of ketones
- In extreme cases, blood pH drops below 7.05
 - May cause coma, cardiac arrhythmias, and death

Nutritional/metabolic disorders (continued)

- Uncontrolled diabetes mellitus increases risk
 - No glucose available for the liver, so the liver responds as if starving, catabolizing proteins and lipids

Nutritional/metabolic disorders (continued)

Gout

- Condition involving insoluble uric acid crystals
- Purine bases of RNA catabolized into uric acid
 - Categorized as **nitrogenous**waste (contains nitrogen)
 - Normal uric acid levels2.7–7.4 mg/dL
 - With higher concentrations,
 body fluids are saturated,
 and insoluble crystals form



Nutritional/metabolic disorders (continued)

- Gout (continued)
 - Crystals accumulate in joints, especially the great toe, causing gouty arthritis
 - Painful condition may persist for several days then disappear for days or years



Module 23.16: Review

- A. Identify and briefly define two eating disorders.
- B. Briefly describe phenylketonuria (PKU).
- C. Define *protein deficiency disease*, and cite an example.

Learning Outcome: Describe several metabolic disorders resulting from nutritional or biochemical problems.

Section 3: Energetics and Thermoregulation

Learning Outcomes

- 23.17 Explain energetics and the role of thermoregulation in maintaining homeostasis.
- 23.18 Describe the roles of the satiety center and the feeding center in the regulation of food intake.
- 23.19 Discuss the mechanisms involved in heat gain and heat loss.
- 23.20 Discuss the homeostatic mechanisms that maintain a constant body temperature.

Module 23.17: Energetics is the study of energy changes, and thermoregulation involves heat balance

Energetics

- Study of energy flow and energy conversion
- Basal metabolic rate (BMR)
 - Common benchmark for energetics studies
 - Defined as the minimum resting energy expenditure of awake, alert person

Module 23.17: Energetics and thermoregulation

BMR can be measured directly

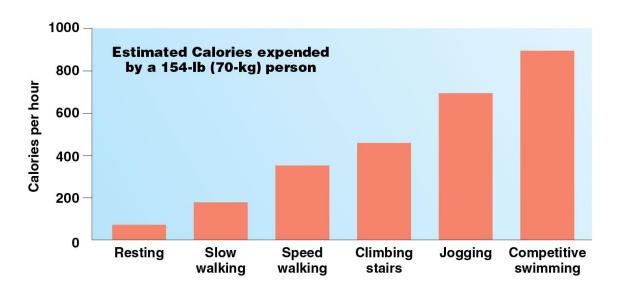
- Monitoring respiratory activity
- Assuming 4.825 Cal expended per liter of oxygen consumed



Module 23.17: Energetics and thermoregulation

Basal metabolic rate (continued)

- Average person has BMR of 70 Cal per hour (1680 Cal per day)
- Various factors can affect BMR
 - Size, weight, level of physical activity
- Food intake must be adequate to support activities



Module 23.17: Energetics and thermoregulation

Thermoregulation

- Homeostatic control of body temperature
- The body's metabolic activities generate heat
 - 40 percent of energy is used to form ATP
 - 60 percent is released as heat
 - Enzymes require specific range of temperature to function
 - Thermoregulation is required to maintain that range

Module 23.17: Review

- A. Define energetics.
- B. What is basal metabolic rate?
- C. Define thermoregulation.

Learning Outcome: Explain energetics and the role of thermoregulation in maintaining homeostasis.

Module 23.18: The control of appetite is complex and involves both short-term and long-term mechanisms

Appetite control

- Involves two areas of the hypothalamus with opposite effects (stimulating one inhibits the other)
 - 1. Feeding center (involved with hunger)
 - 2. Satiety center (involved with food satisfaction)

Control of appetite

Short-Term Regulation of Appetite

Stimulation of Satiety Center

Elevated blood glucose levels depress appetite, and low blood glucose stimulates appetite. The likely mechanism is glucose entry stimulating the neurons of the satiety center.

Several hormones of the digestive tract, including CCK, suppress appetite during the absorptive state.

Stimulation of stretch receptors along the digestive tract, especially in the stomach, causes a sense of satiation and suppresses appetite.

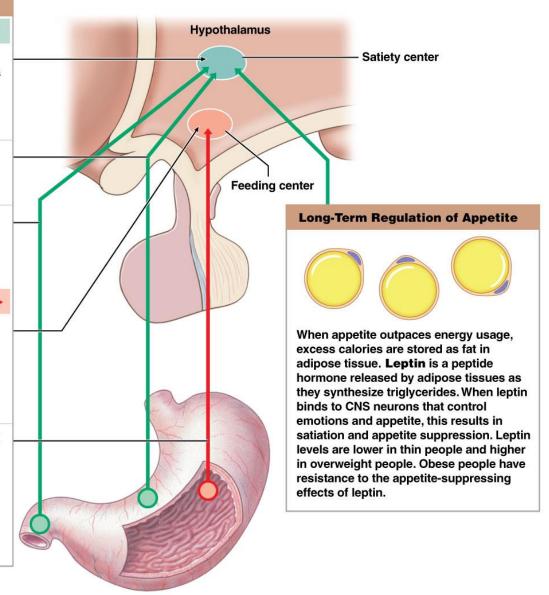
Stimulation of Feeding Center

Several neurotransmitters have been linked to appetite regulation.

Neuropeptide Y (NPY), for example,

is a hypothalamic neurotransmitter that (among other effects) stimulates the feeding center and increases appetite.

The hormone **ghrelin** (GREL-in), secreted by the gastric mucosa, stimulates appetite. Ghrelin levels are high when the stomach is empty and decline as the stomach fills. Some research suggests that leptin plays a role in regulating ghrelin, showing that these "hunger hormones" are complex and warrant further study.



Module 23.18: Control of appetite

Appetite control

- Affected by multiple factors, including social factors, psychological pressures, and dietary habits
- Regulation of appetite can occur on two levels
 - 1. Short-term regulation
 - 2. Long-term regulation

Module 23.18: Control of appetite

Short-term regulation of appetite

- Stimulation of satiety center
 - Elevated blood glucose levels
 - Hormones of digestive tract (such as CCK)
 - Stretching of digestive tract wall
- Stimulation of feeding center
 - Neurotransmitters
 - Example: neuropeptide Y, or NPY, from hypothalamus
 - Ghrelin
 - Hormone secreted by gastric mucosa when stomach is empty

Module 23.18: Control of appetite

Long-term regulation of appetite

- Stimulation of the satiety center
 - Leptin
 - Peptide hormone secreted by adipocytes
 - Stimulates satiety center and suppresses appetite
 - Effects are gradual

Control of appetite

Short-Term Regulation of Appetite

Stimulation of Satiety Center

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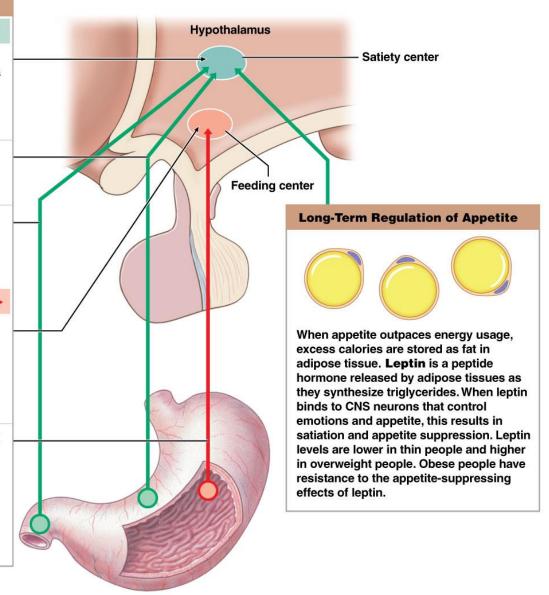
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Module 23.18: Review

- A. Which hormone inhibits the satiety center and stimulates appetite in the short term?
- B. Describe leptin and its effect on appetite.
- C. How might a lack of neuropeptide Y in the hypothalamus affect the control of appetite?

Learning Outcome: Describe the roles of the satiety center and the feeding center in the regulation of food intake.

Module 23.19: To maintain a constant body temperature, heat gain and heat loss must be in balance

Mechanisms of heat transfer

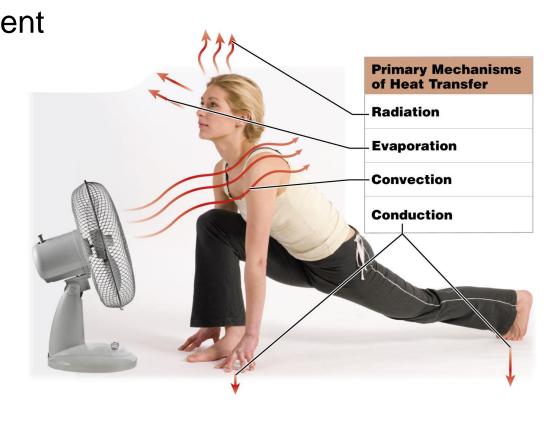
- Required to balance heat loss and heat production in order to maintain body temperature
- Four primary mechanisms

Mechanisms of heat transfer (continued)

1. Radiation

- Heat energy transfer as infrared radiation
- Example: the heat from the sun

 More than 50 percent of body heat loss indoors is via radiation

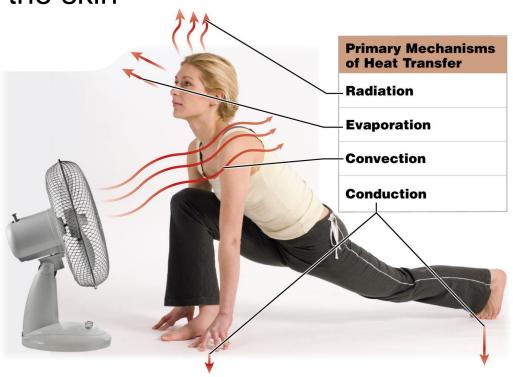


Mechanisms of heat transfer (continued)

2. Evaporation

 Water changing from liquid to vapor absorbs 0.58 Cal per gram of water

Cools the surface of the skin



Mechanisms of heat transfer (continued)

- 2. Evaporation (continued)
 - Insensible perspiration (from alveoli and skin)
 - About 20–25 mL per hour (relatively constant)
 - Accounts for ~20 percent of body's average indoor heat loss
 - Sensible perspiration (from sweat glands)
 - Can excrete up to 2–4 liters per hour

Mechanisms of heat transfer (continued)

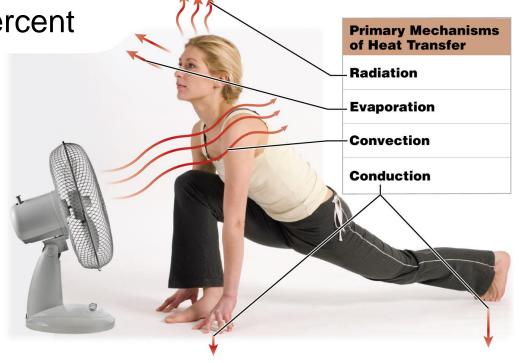
3. Convection

Result of heat loss due to air movement

 Warmer air rises away from the body, replaced by cooler air

 Accounts for ~15 percent of body's heat loss

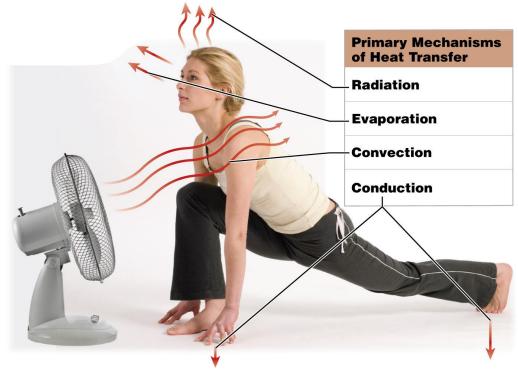
indoors



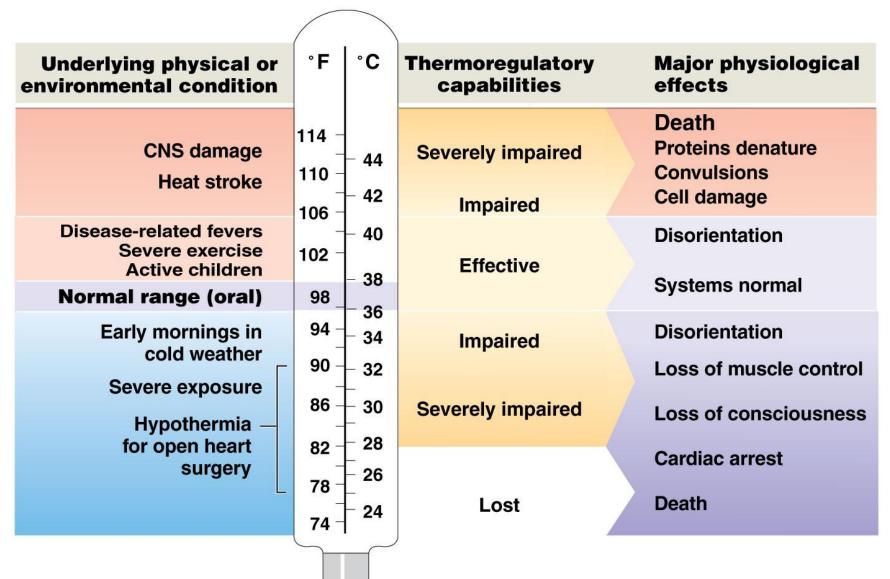
Mechanisms of heat transfer (continued)

4. Conduction

- Direct transfer of energy through physical contact
- Generally not an effective mechanism of gaining/losing heat



Effects of failure to control body temperature



Module 23.19: Review

- A. Define insensible perspiration.
- B. What heat transfer process accounts for about one-half of a person's heat loss when indoors?
- C. How is heat loss different between conduction and convection?

Learning Outcome: Discuss the mechanisms involved in heat gain and heat loss.

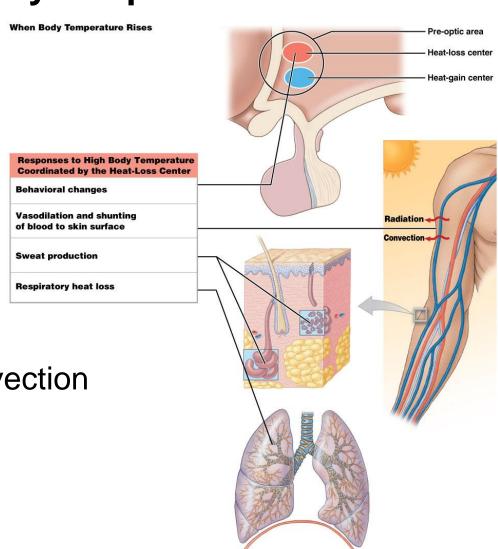
Module 23.20: Thermoregulatory centers in the hypothalamus adjust heat loss and heat gain

Thermoregulation

- Heat loss and heat gain involve many systems
- Coordinated by two centers in hypothalamus pre-optic area
 - 1. Heat-loss center
 - 2. Heat-gain center

Responses to high body temperature

- Behavioral changes
 - Moving to shade, water, or other voluntary steps
- Vasodilation and shunting of blood to skin surface
 - Increases heat loss by radiation and convection



Responses to high body temperature

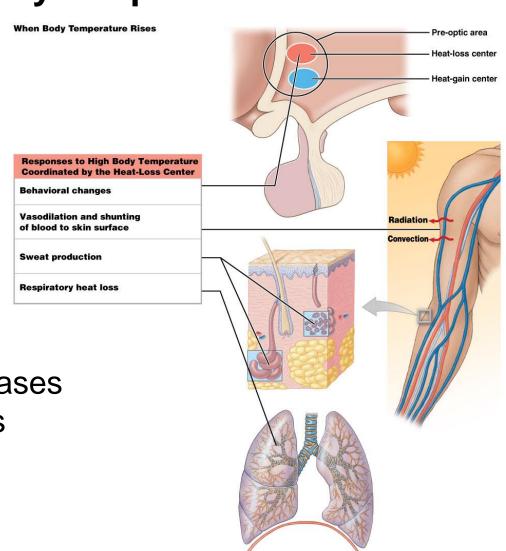
(continued)

Sweat production

 Accelerates heat loss by evaporation

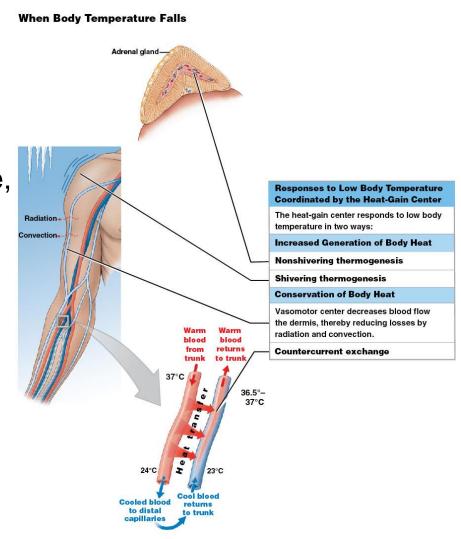
Respiratory heat loss

 Breathing more deeply and through an open mouth increases evaporative heat loss



Responses to low body temperature

- Increased generation of body heat
 - Nonshivering thermogenesis
 - Hormones (epinephrine, thyroid hormone) increase the metabolic activity of tissues



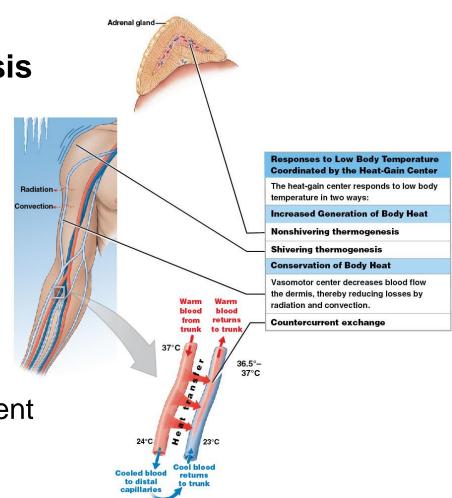
Responses to low body temperature (continued)

When Body Temperature Falls

Increased generation of body heat (continued)

Shivering thermogenesis

- Increase in muscle tone increases energy consumption
- Increases to the point of involuntary contractions of skeletal muscles (shivering)
- Can increase heat generation by 400 percent



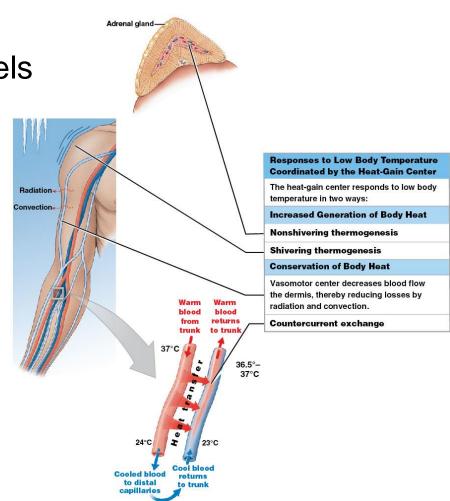
Responses to low body temperature (continued)

When Body Temperature Falls

Conservation of body heat

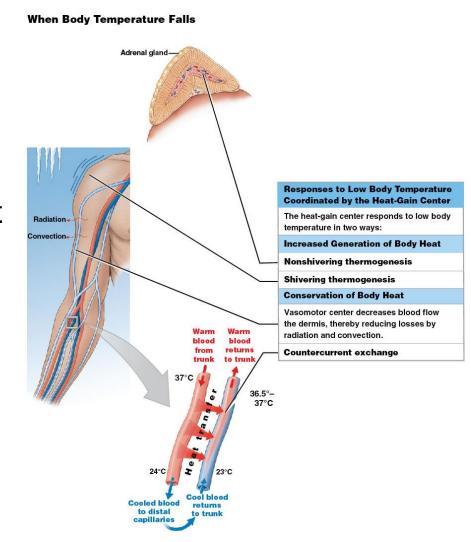
> Vasoconstriction of vessels near body surface

 Reduces heat loss by radiation and convection



Responses to low body temperature (continued)

- Countercurrent exchange of heat
 - Transfer of heat from deep arteries (bringing warmer blood from the body's core) to adjacent deep veins (bringing cooler blood from the periphery)
 - Traps heat close to the body core and reduces heat loss



Module 23.20: Review

- A. Predict the effect of peripheral vasodilation on a person's body temperature.
- B. Describe the role of nonshivering thermogenesis in regulating body temperature.
- C. Name the heat conservation mechanism that conducts heat from deep arteries to adjacent deep veins in the limbs.

Learning Outcome: Discuss the homeostatic mechanisms that maintain a constant body temperature.