Morphological Characterization and Analysis of Ion-

Containing Polymers Using Small Angle X-ray Scattering

Mingqiang Zhang

Dissertation submitted to the faculty of the Virginia Polytechnic Institute and State University in partial fulfillment of the requirements for the degree of

DOCTOR OF PHILOSOPHY

In

Chemistry

Robert B. Moore, Chair Timothy E. Long Hervé Marand Judy S. Riffle

October, 2014 Blacksburg, Virginia

Keywords: perfluorosulfonic acid ionomer, Nafion[®], ionomer, proton exchange membrane, block copolymer, morphology, small angle X-ray scattering, solution procedure

Figure 1.3. Yarusso-Copper's modified hard-sphere model (Adapted with permission from *Macromolecules* **1983**, *16*, 1871. Copyright, The American Chemistry Society, 1983)



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Figure 2.5. Correlations between SAXS, DMA, and NMR (Reprinted with permission from *Macromolecules* **2005**, *38*, 6472. Copyright, The American Chemical Society, 2005)





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Figure 2.6. Small-angle X-ray scattering (SAXS) profiles of TMA⁺(A) and TBA⁺–Nafion (B) subjected to thermal annealing at 100 and 200 $^{\circ}$ C for 10 min. Each plot contains two dimensional SAXS images before (left) and after (right) thermal annealing at 200 $^{\circ}$ C (Reprinted with permission from *Polymer* **2009**, *50*, 5720. Copyright, Elsevier, 2009)

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Figure 2.7. Cluster-network model for Nafion membranes developed by Gierke and coworkers (Adapted with permission from the *Journal of Membrane Science* **1983**, *13*, 307. Copyright, Elsevier, 1983)



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Figure 2.10. Core-Shell Model by Haubold and co-workers (Adapted with permission from *Polymer* 2000, *41*, 5829. Copyright, Elsevier, 2000)



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Figure 2.11. Change in morphology in Nafion as a function of the volume fraction of water as described by Gebel (Adapted with permission from *Polymer* **2000**, *41*, 5829. Copyright, Elsevier, 2000)



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Figure 2.12. Fibrillar Structure of Nafion developed by Rubatat and co-workers (Adapted with permission from *Macromolecules* 2004, *37*, 7772. Copyright, The American Chemical Society, 2004)



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Figure 2.13. Schematic of Nafion worm-like model as described by Kim and co-workers (Adapted with permission from *Macromolecules* **2006**, *39*, 4775. Copyright, American Chemical Society, 2004)



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Figure 2.16. Schematic representation of micelle arrangement within Nafion cast onto different substrates (Reprinted with permission from *Macromolecules* **2011**, *44*, 2893. Copyright, The American Chemistry Society, 2011)



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Figure 2.17. Sketch of fibrillar structure of Nafion proposed by Heijden and co-workers (Reprinted with permission from *Macromolecules* 2004, *37*, 5327.Copyright, The American Chemistry Society, 2004)



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Figure 2.18. Variable temperature SAXS of TMA⁺(A), TEA⁺(B), TPA⁺(C), TBA⁺(D)-Nafion (Reprinted with permission from *Macromolecules* **2006**, *39*, 3939. Copyright, The American Chemistry Society, 2006)



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Figure 8.3. Common lattice structure. (a) Lamellae; (b) hexagonal packed cylinders; (c) BCC; (d) FCC; (e) hexagonal close-packed spheres; (f) primitive cubic; (g) double gyroid; (h) double diamond (DD); (i) Pm3n; and (j) Fddd (Reprinted with permission from Burger, C.; Hsiao, B. S.; Chu, B. In *Polymer Science: A Comprehensive Reference*; Matyjaszewski, K., Möler, M., Eds.; Elsevier: Amsterdam, 2012, p 363. Copyright, Elsevier, 2012)

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Figure 8.5. SAXS profiles of BPSH-BPS MBC ionomer (Reprinted with permission from *Polymer* **2011**, 52, 3963. Copyright, Elsevier, 2011)

Table 8.2. SAXS domain space comparison for the data in Figure 8.5 (Reprinted with permission from *Polymer* **2011**, 52, 3963. Copyright, Elsevier, 2011)

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Figure 8.9. SAXS profiles of 6FPAEB-BPS100 and 6FPAEB-HQS100 multiblock copolymers and 6FPAEB35 random copolymer membrane (Reprinted with permission from *Polymer* **2013**, *54*, 6305.Copyright, Elsevier, 2013)

Figure 8.10. TEM images of 6FPAEB-BPSH (top left: 7K-7K, top right: 15K-15K) and 6FPAEB-HQSH (bottom 7K-7K) multiblock copolymer membranes (Reprinted with permission from *Polymer* **2013**, *54*, 6305.Copyright, Elsevier, 2013)



Figure 8.11. Chemical structures of poly(4VIM-*b*-DEGMEMA-*b*-4VIM) triblock copolymers, and copolymers were synthesized and by Dr. Michael H. Allen from Prof. Timothy E. Long's research group (Reprinted with permission from *Polymer Chemistry* **2013**, *4*, 2333. Copyright, RSC Publishing, 2013)

Figure 8.12. SAXS profiles of scattering intensity versus scattering vector for poly(4VIM*b*-DEGMEMA-*b*-4VIM) triblock copolymers (Reprinted with permission from *Polymer Chemistry* **2013**, *4*, 2333. Copyright, RSC Publishing, 2013)

Table 8.5. SAXS q Values and Bragg spacings (Reprinted with permission from Polymer Chemistry 2013, 4, 2333. Copyright, RSC Publishing, 2013)

Figure 8.14. AFM phase image in tapping mode and TEM image of 40 wt. % 4VIMcontaining ABA triblock copolymer (Reprinted with permission from *Polymer Chemistry* **2013**, *4*, 2333. Copyright, RSC Publishing, 2013)

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Figure 8.15. Chemical structure of PC-PDMS multiblock copolymers (Reprinted with permission from *Polymer Engineering & Science* **2013**. Copyright, Wiley Online Library, 2013)

Figure 8.16. SAXS profiles for PC-PDMS multiblock copolymers. The SAXS profiles have been vertically shifted to facilitate a comparison of the peak positions (Reprinted with permission from *Polymer Engineering & Science* **2013**. Copyright, Wiley Online Library, 2013)

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Figure 9.1. (A) SAXS pattern from the block copolymer sample after alignment in the electric field. The arrow shows that X-ray beam was orthogonal to the applied field direction. (B) Schema of the orientation of slice planes under the applied electric field (Reprinted with permission from *Macromolecules* 1993, *26*, 2698. Copyright, The American Chemistry Society, 1993)



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Figure 9.2. (A) Theoretical WAXS pattern of a main chain liquid crystalline block copolymer annealed in a magnetic field at elevated temperature. X-ray beam is perpendicular to the magnetic field direction (B) Experimental WAXS pattern clearly shows the rod-like structure are aligned parallel to the magnetic field (C) SAXS pattern clearly demonstrates the alignment of the blocks within the block copolymer is perpendicular to the magnetic field direction (Reprinted with permission from *Nano Letters* **2007**, *7*, 2742. Copyright, The American Chemistry Society, 2007)



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Figure 9.3. SAXS patterns as a function of the applied magnetic field strength. The magnetic field direction is horizontal with respect to the orientation of the X-ray detector. The curves show the peak intensities and peak shapes of the microdomain scattering at q=0.07 Å⁻¹ (triangles) and scattering at q=0.18 Å⁻¹ (circles), respectively as a function of field strength (Reprinted with permission from *Journal of the American Chemical Society* **2010**, *132*, 17516. Copyright, The American Chemistry Society, 2010)

Figure 9.4. The average conductivity of 120:1 EO:Li⁺ sample aligned in 5 T magnetic field in two directions under room temperature, with conductivity of nonaligned sample shown for comparison (Reprinted with permission from *Journal of the American Chemical Society* **2010**, *132*, 17516. Copyright, The American Chemistry Society, 2010)



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