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## Ionic Bonding Speed Dating!

Part I: Fill out your ion card(s) using information from the periodic table (8 minutes)
Part II: Ions speed dating

1. Mingle with your classmates and combine your ions to form as many correct compounds as you can. In some cases you may need to use one partner's ion twice to form a correct compound.
2. For each compound write name of the compound in the chart.
3. You will earn 10 points for each correct compound name. 100 points is a perfect score. Each additional compound you make will earn you extra credit points.

| Your ion symbol | Partner's ion symbol | Compound Name | Compound Formula |
| :---: | :---: | :---: | :---: |
| 1. |  |  |  |
| 2. |  |  |  |
| 3. |  |  |  |
| 4. |  |  |  |
| 5. |  |  |  |
| 6. |  |  |  |
| 7. |  |  |  |
| 8. |  |  |  |
| 9. |  |  |  |
| 10. |  |  |  |
| 11. |  |  |  |
| 12. |  |  |  |
| 13. |  |  |  |
| 14. |  |  |  |
| 15. |  |  |  |


| Element Name: Lithium <br> Symbol: $\qquad$ <br> \# of Protons: $\qquad$ \# of Electrons: $\qquad$ <br> \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ Ion symbol: $\qquad$ <br> Ion name: $\qquad$ | Element Name: Fluorine Symbol: $\qquad$ <br> \# of Protons: $\qquad$ \# of Electrons: $\qquad$ <br> \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ <br> Ion symbol: $\qquad$ <br> Ion name: $\qquad$ |
| :---: | :---: |
| Element Name: Sodium <br> Symbol: $\qquad$ \# of Protons: $\qquad$ \# of Electrons: $\qquad$ \# of Valence Electrons: $\qquad$ Will it gain or lose e-? $\qquad$ How many e- will it gain/lose? $\qquad$ Ion symbol: $\qquad$ <br> Ion name: $\qquad$ | Element Name: Chlorine Symbol: $\qquad$ <br> \# of Protons: $\qquad$ \# of Electrons: $\qquad$ <br> \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ <br> Ion symbol: $\qquad$ <br> Ion name: $\qquad$ |
| Element Name: Lithium Symbol: $\qquad$ \# of Protons: $\qquad$ \# of Electrons: $\qquad$ <br> \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ <br> Ion symbol: $\qquad$ <br> Ion name: $\qquad$ | Element Name: Fluorine Symbol: $\qquad$ <br> \# of Protons: $\qquad$ \# of Electrons: $\qquad$ <br> \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ <br> Ion symbol: $\qquad$ <br> Ion name: $\qquad$ |
| Element Name: Sodium <br> Symbol: $\qquad$ \# of Protons: $\qquad$ \# of Electrons: $\qquad$ \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ <br> Ion symbol: $\qquad$ <br> Ion name: $\qquad$ | Element Name: Chlorine Symbol: $\qquad$ <br> \# of Protons: $\qquad$ \# of Electrons: $\qquad$ <br> \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ <br> Ion symbol: $\qquad$ <br> Ion name: $\qquad$ |


| Element Name: Magnesium Symbol: $\qquad$ <br> \# of Protons: $\qquad$ \# of Electrons: $\qquad$ <br> \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ <br> Ion symbol: $\qquad$ <br> Ion name: $\qquad$ | Element Name: Oxygen Symbol: $\qquad$ <br> \# of Protons: $\qquad$ \# of Electrons: $\qquad$ \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ <br> Ion symbol: $\qquad$ <br> Ion name: $\qquad$ |
| :---: | :---: |
| Element Name: Calcium Symbol: $\qquad$ <br> \# of Protons: $\qquad$ \# of Electrons: $\qquad$ <br> \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ <br> Ion symbol: $\qquad$ <br> Ion name: $\qquad$ | $\qquad$ <br> \# of Protons: $\qquad$ \# of Electrons: $\qquad$ <br> \# of Valence Electrons: $\qquad$ <br> Will it gain or lose e-? $\qquad$ <br> How many e- will it gain/lose? $\qquad$ <br> Ion symbol: $\qquad$ <br> Ion name: $\qquad$ |

Element Name: Aluminum Symbol: $\qquad$ \# of Protons: $\qquad$ \# of Electrons: $\qquad$ \# of Valence Electrons: $\qquad$ Will it gain or lose e-? $\qquad$ How many e- will it gain/lose? $\qquad$ Ion symbol: $\qquad$ Ion name: $\qquad$


