

## NATURAL COLORATION OF WOOD MATERIAL BY RED BEETROOT (*Beta vulgaris*) AND DETERMINATION COLOR STABILITY UNDER UV EXPOSURE

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### ABSTRACT

This study is aimed to develop an eco-friendly wood stained extracted from beetroot (*Beta vulgaris*) and determine the color stability of this stain to UV light irradiation. Natural dyestuff were extracted from beetroot by ultrasonic-assisted method and prepared from aqueous solution with ferrous sulphate, aluminum sulphate, copper sulphate and vinegar mordant mixes. Scots pine (*Pinus sylvestris*), oriental beech (*Fagus orientalis*), oak (*Quercus petraea*) and walnut (*Juglans regia*) wood specimens were used for the study. After treatment with stain, wood specimens were exposed to UV irradiation for periods of 50, 100 and 150 hours. Results showed that the color change values for all wood specimens colored with beetroot extract had better performance compared to synthetic dye. Beetroot extracts may be used as an upper surface dyestuff for indoor application and toys. Therefore, alternative to synthetic dyes more economical and eco-friendly, wood paints may be developed.

**Keywords:** Beetroot, color changes, natural coloration, UV weathering.

### INTRODUCTION

Wood is commonly used for different purposes ranging from high-performance constructions to filigree art work. In most applications, the appearance of the surface is crucial for costumer decisions (Bechtold and Mussak 2009). Generally wood materials need to preservation and coloring by chemicals in terms of environmental health in recent years. Recently international awareness of environment, ecology and pollution control created an upsurge on the interest of people to use more environmentally products. Natural dyes derived from flora and fauna are believed to be safe because of their non-toxic, non-carcinogenic and biodegradable nature (Mirjalili *et al.* 2011). For this reason, the importance of the protection of wood material by natural sources is gaining importance day by day.

Wood-based products and decoration elements are potential sources for a number of volatile organic compounds (VOCs) that may be released indoors (Goktas *et al.* 2009c). Many chemical components are used in wood finishes and coatings industry. Salthammer *et al.* 1998, identified about 150 different VOCs. The formation of organic indoor pollutants by chemical reaction in indoor air has been extensively studied (Weschler 1997, Salthammer 2002). Moreover, many building products contain reactive compounds, which decompose under normal living conditions and can be released as secondary emission products (Salthammer 1999). These pollutants are emitted from different sources such as floor coverings, wood-based panel, furniture, solid woods, wood stains and paints (Cheng and Brown 2003). Billions of people in the world suffer from diseases resulting from low air quality, and trillions of dollars are spent to compensate for such problems (Mo *et al.* 2009). For many years, it has been known that many synthetic dyes are dangerous to human health, and the industry has been seeking environmentally friendly products (Peker *et al.* 2012).

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Goktas *et al.* investigated the development of environmentally friendly wood stains derived from laurel (*Laurus nobilis*) (2008a), oleander (*Nerium Oleander*) (2009a), and madder root (*Rubia tinctorium*) (2009b) and determined the color stability of the stains when exposed to ultra-violet (UV) light irradiation. Their results showed that the wood stains derived from laurel, oleander, and madder root extract provided some color stability after UV irradiation. Goktas *et al.* 2008b, investigated the antifungal properties and color stability under UV exposure of wood treated with aqueous solutions of *Juglans regia* extract. Results showed that the lightness values of the wood specimens did change slightly. Peker *et al.* 2012, investigated the color stability of acorn (*Quercus ithaburensis*) and reported that acorn has an aesthetic appearance as a coloring and preservative upper surface material.

Nowadays many natural roots were mentioned as a natural dye source in various scientific researches. Junqueira-Goncalves *et al.* 2011, investigated the beetroot extract as a colorant for cream cheese, they reported that beetroot extracts might be a potential alternative dye for food. Sivakumar *et al.* 2009, has been studied on the use of power ultrasound to improve the extraction of beetroot dye and application to the substrates such as leather and they reported that the beetroot is suitable for dyeing leather.

Betainin, obtained from the roots, is used industrially as red food colorants, e.g. to improve the color and flavor of tomato paste, sauces, desserts, jams and jellies, ice cream, sweets, and breakfast cereals (Grubben and Denton 2004). Beetroot dye may also be used in ink. Within older bulbs of beetroot, the color is a deep crimson, and the flesh is much softer.

In Turkey, there is a large production and consumption, while in the Aegean and Marmara regions in the Mediterranean region beetroot is produced in limited quantities. There are currently 7286 tons harvests every year in Turkey. Only pickling beetroot are used as industrial sense. Red beets are a potential source to use as a wood colorant whereof it contains substances which were very rich and beautiful colors. These substances have been used for textiles and food but have never been used as natural colorants in wood industry.

In this study we have focused on the coloring wood material by red beetroot extract and mordant mixes to determine their properties of color change after UV weathering and durability after leaching. Another aim is to utilize the potential of natural sources as a substitute colorant for synthetic dyes.

## MATERIALS AND METHODS

### Wood materials

As wood material, Scots pine (*Pinus sylvestris*), oriental beech (*Fagus orientalis*), oak (*Quercus petraea*) and walnut (*Juglans regia*) woods commonly used in furniture and decoration industries in Turkey were chosen. The samples were prepared from first-class wooden materials, which are smooth fiber, knotless, crack-free, without color and density difference, with annual rings perpendicular to the surfaces, and from parts of sapwood, in accordance with TS 2470 standards (TS 2470, 2005). The samples prepared for light fastness tests with dimensions of 150×75×0,5 mm and for leaching tests with 19×19×19 mm and were kept under suitable temperature ( $20 \pm 2^\circ\text{C}$ ) and suitable moisture (moisture of  $\pm 12\%$  and relative humidity of  $\pm 65\%$ ) conditions until they became air-dried, in order to achieve the moisture value in furniture used under interior area conditions in accordance with TS 2471, 2005 standard.

### Plant material and Mordant agents

In this study, the red beetroot was gathered from the Muğla area (Turkey). Mordant agents ferrous sulphate ( $\text{Fe}_2(\text{SO}_4)_3 \cdot 7\text{H}_2\text{O}$ ), aluminum sulphate ( $\text{Al}_2(\text{SO}_4)_3 \cdot 18\text{H}_2\text{O}$ ) and copper sulphate ( $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ ) were provided from Kimetsan Co. and vinegar was purchased from Fersan Co. The synthetic dye used for comparison; woodtex – wood colorant was supplied from Kayalar kimya Co.

### Preparation of dyestuff

A weighed amount of dry plant material was extracted with distilled water in a ultrasonic bath (Elmasonic X-tra 150 H). In the standard procedure the mass ratio of plant material to the volume of liquid was 1:20; extraction was performed for approximately 180 min at 45 °C and 180 W sonic power in a stainless ultrasonic bath. Due to the rather high liquor ratio some manual stirring was sufficient to distribute the plant material in the liquid during the extraction period. Volume loss due to evaporation was compensated by the addition of water at the end of the extraction period to obtain the initial volume. Ultrasound-assisted extraction is an inexpensive, simple and efficient alternative to conventional extraction techniques. The main benefits of using ultrasound in solid-liquid extraction include the increase of extraction yield and faster kinetics. Ultrasound can also reduce the operating temperature allowing the extraction of thermolabile compounds. Compared with other novel extraction techniques such as microwave-assisted extraction, the ultrasound apparatus is cheaper and its operation is easier (Wang and Weller 2006).

Aqueous solutions were mordanted by adding ferrous sulphate ( $\text{Fe}_2(\text{SO}_4)_3 \cdot 7\text{H}_2\text{O}$ ) 3%, aluminum sulphate ( $\text{KAl}_2(\text{SO}_4)_3 \cdot 18\text{H}_2\text{O}$ ) 5%, copper sulphate ( $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ ) 5%, and grape vinegar 10% in order to stabilize the color of extracted dyes, to ensure it to hang on to the applied material (to increase retention amount), and to create color options.

### Dyeing test samples

The air-dried wood specimens were placed into ultrasonic bath container according to their intended treatments. Treatment procedures are given in Table 1. In the treatment two different methods (immersion and ultrasonic-assisted immersion) of dyeing were used. Any extra solution left on the specimens was removed with a clean cloth. Specimens were then left to dry at  $20 \pm 3$  °C in a vertical position.

**Table 1.** Treatment Procedures.

Dye	Treatment Method	Sonic Power (W)	Temperature (°C)	Time (min)
Natural	Control (immersion)	---	45	60
	Ultrasonic-assisted immersion	300	45	60
Synthetic	Control (immersion)	---	45	60
	Ultrasonic-assisted immersion	300	45	60

### Accelerated weathering test

Operation of accelerated weathering device is composed of two periods. The first is condensation stage. This stage ensures the sample pieces to expand by changing the temperature, coolness, and moisture amount of the environment at regular intervals in order to imitate the external environment conditions and by spraying hot steam thereon. In the second stage, the test pieces were subjected to UV beams by using UV lamps. Weathering process was performed by operating the device for 4 h in the condensation stage, and for 8 h in UV period. The average irradiance was about 330 nm at 50 percent relative humidity and 20 °C. Specimens were directly exposed to UV light at a distance of 20 cm and an angle of 90°E (Kamdern and Grelier 2002). Eight replicate samples treated with each stain solution and untreated controls were run for the randomly selected irradiation times of 0 (no irradiation), 50 hours, 100 hours, and 150 hours. The color of the samples was measured after each irradiation period.

### Color measurements

In order to determine the color change values in the accelerated weathering tests, the colors of the coated parts were identified prior to weathering by using Konica Minolta CR-10, a portable color reader device. Color measurements were performed on each sample due to the non-homogenous color structure of the wooden material in four measures. The identified color values were classified according to the coordinates *Commission International de l'Eclairage-CIELAB 1976* set in ISO 2470 standards (Figure 1). The obtained colors were indicated with numerical values of L, a, and b. Here, L indicates lightness from 0% (black) and 100% (white), a from green (-a) to red (+a), and b from blue (-b) to yellow (+b). Coated sample pieces were subjected to color measure prior being exposed to accelerated weathering environment and stated as "color values prior to accelerated weathering". Afterwards, the coated samples were placed on the weathering device, exposed to UV and condensation environment for 50 h, and color measurement was carried out again from the same cross-corners. This process was repeated at the 100th and 150th h.

### Determination of color change values

Color changes due to accelerated weathering were calculated with the following formulas in accordance with ISO 2470 standards.

$$\Delta L^* = L_f^* - L_i^* \quad (1)$$

$$\Delta a^* = a_f^* - a_i^* \quad (2)$$

$$\Delta b^* = b_f^* - b_i^* \quad (3)$$

$$\Delta E^* = \sqrt{\Delta L^{*2} + \Delta a^{*2} + \Delta b^{*2}} \quad (4)$$

Here;  $\Delta L^*$ ,  $\Delta a^*$  and  $\Delta b^*$  are the changes occurring between the initial state (i) and final state (f) of the colors.  $\Delta E^*$ , indicates total changes of the colors occurring in L, a, and b. Here, the highest value shows the highest color change.

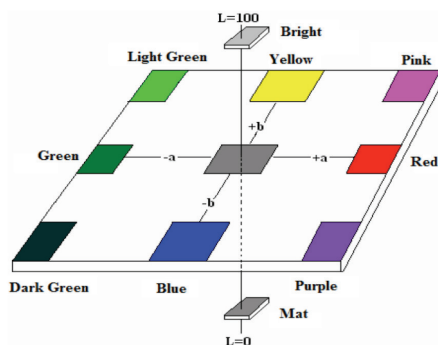


Figure 1. CIELAB-76 color system (Goktas *et al.* 2008).

## RESULTS AND DISCUSSION

Color changes values occurring on Scots pine, oriental beech, walnut and oak wood test samples stained with beetroot extracts and mordant mixes extracts and exposed to UV application for 50, 100, and 150 h are numerically represented in Tables 1 and 2. A schematic representation is given for beetroot and mordant mixes extracts in Figure 2.

Positive values of  $\Delta L^*$  show whitening, and negative values of  $\Delta L^*$  indicate the color turning grey. Positive values of  $\Delta a^*$  indicate reddening of the colors, and negative values of  $\Delta a^*$  show a shift towards green. Positive values of  $\Delta b^*$  represent yellowing in color, and negative values of  $\Delta b^*$  represent the color turning blue.

In the process of wood photodegradation, the first phenomenon observed is the color change, followed by the loss of fibers and surfaces erosion. The erosion process is slow and depends on several factors such as anatomy of wood, density, proportion of earlywood and latewood, grain angle and angle of exposure (Williams 2005, Garcia *et al.* 2014). According to the results, all wood specimens exposed to UV radiation of 350 nm for all the exposure periods generally showed negative values of  $\Delta L^*$ . This was attributed to chemical changes which occurred, especially in lignin, due to photo-degradation resulting from UV exposure and, consequently, to a darkening of color on the wood (Peker *et al.* 2012, Ozen *et al.* 2014). The high negative values of  $\Delta L^*$  indicate the sensitivity of that wood type against UV radiation and the surface quality thereof (Feist and Hon 1984).

In general, it is realized that color changes occurs quickly in 50h and 100h periods; during the following period less changes occurred. The first part of the color change process is because of the formation of chromophoric groups as carbonyl and carboxyl groups resulting mainly from degradation of  $\alpha$ -carbonyl, biphenyl and ring-conjugated double bond structures in lignin and moving of the extractives towards to wood surface from inside of wood (Lin and Kringstad 1970, Hon and Feist 1992). Relatively more quick change is reported even if the wood material is exposed to UV light in external environment for short period or under accelerated weathering (Feist and Hon 1984, Peker *et al.* 2012).

The best color stability ( $\Delta E^*$  28,31) was attained for scots pine wood with beetroot + vinegar mixes and dyed by classical method after 150 h of exposure.

The biggest color change was seen on scots pine species dyed with beetroot + ferrous sulphate mixes. Generally color changes on scots pine species showed similar results with all dyestuff types (except beetroot + ferrous sulphate). When we look at the color changes on beech wood species the lowest change was ( $\Delta E^*$  18,72) acquired for control (without mordant) and the biggest change ( $\Delta E^*$  35,40) for beetroot + ferrous sulphate mixes.

The lowest color change value ( $\Delta E^*$  12,07) is obtained from the application of control (without mordant) on walnut wood species, and the highest color change value ( $\Delta E^*$  27,80) from the application of beetroot + ferrous sulphate mixes.

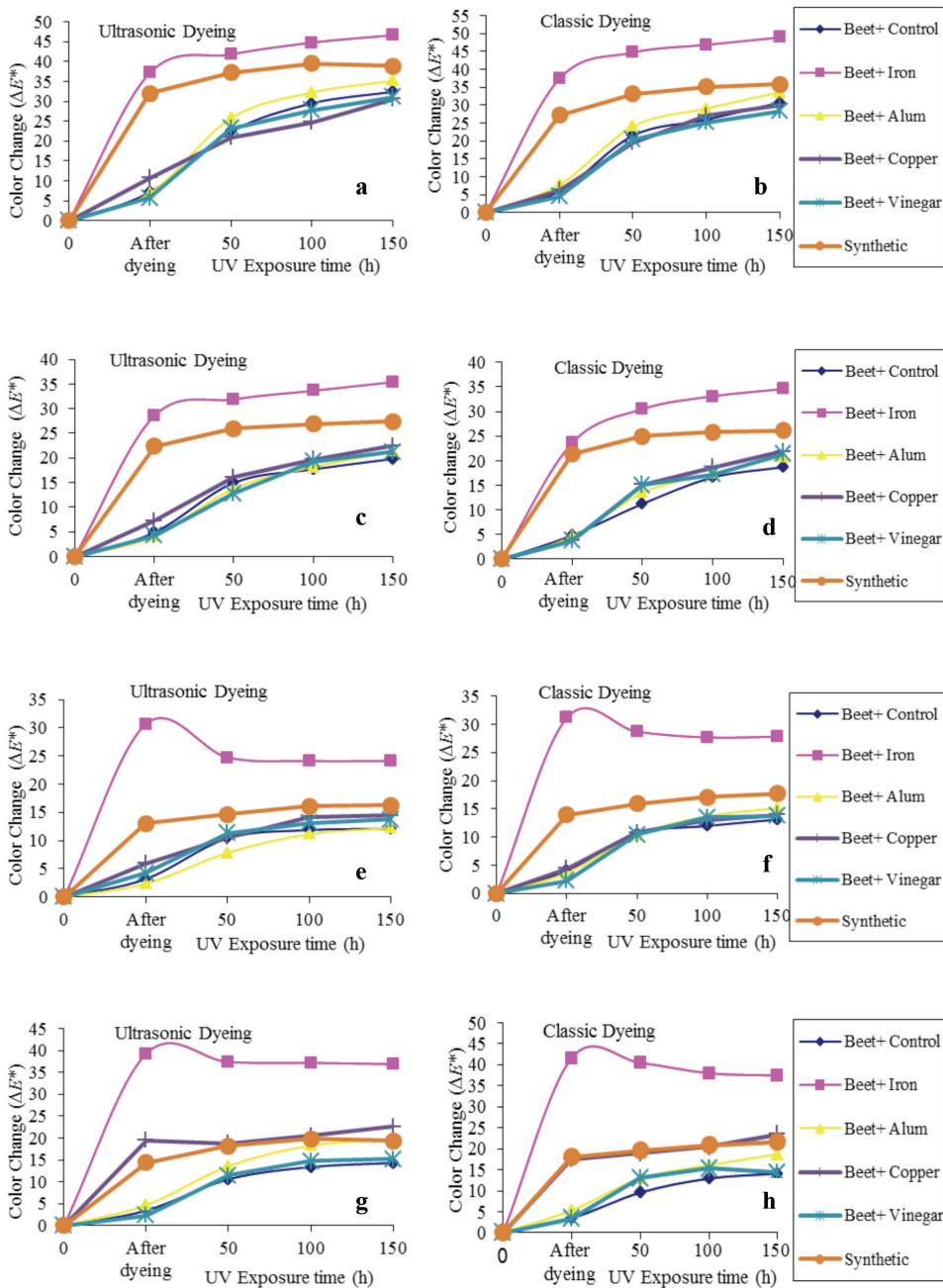
According to the color changes of the oak wood species, the lowest change was ( $\Delta E^*$  14,11) acquired for control (without mordant) and the highest color change was ( $\Delta E^*$  37,44) performed from the application of beetroot + ferrous sulphate mixes.

Generally all wood species have darker color with beetroot + ferrous sulphate mixes and wood species colored by beetroot + ferrous sulphate mixes had large color change values compared to other components. These changes may be explained with interaction of ferrous sulphate ions and wood components. Metal mordant have been pointed out as the reason for color stability. Because metal ions, promotes free radical formation (Feist and Hon 1984, Peker *et al.* 2012) of wood components even when they are exposed to light. The stabilization of lignin by ferrous was reported to occur through the formation of complex (Kamdem and Grelier 2002).

The reason for the differences between wood types may be due to the differences of chemical composition of the wood types, and interaction of beetroot and mordant mixes extract compounds with wood component, resulting in the different photo-degradation effects of UV irradiation. Feist and Hon (1984) reported that higher color changes were found in samples having higher lignin content. Lignin is the compound absorbing UV of 80 to 95% among the three main components of wood (Peker *et al.* 2012). Besides, generally soft woods have 2-10% more lignin than hard woods (Tereza *et al.* 2004, Goktas *et al.* 2009b). Lignin has aromatic, phenolic and carboxylic groups that absorb rays of different energy levels. Although cellulose is not sensitive to UV light of wavelengths longer than 340 nm (Feist and Hon 1984). Nevertheless, the roughness of color stabilized wood is a complex phenomenon because wood is an anisotropic and heterogeneous material. Several factors, such as anatomical differences, growing characteristics, machining properties, pre-treatments (e.g. steaming, drying, etc.), can affect the color stability (Temiz *et al.* 2005, Goktas *et al.* 2009b). Also, the treatments parameters as treatment time, percentage of dyer materials, application temperature and percentage of mordant may affect the color stability.

**Table 2.** Color Change values UV Application for 50h, 100h and 150 h for dyed test sample.

Wood type	Dyeing Method	Dye extract	After dyeing				After 50 h				After 100 h				After 150 h			
			$\Delta L^*$	$\Delta a^*$	$\Delta b^*$	$\Delta E^*$	$\Delta L^*$	$\Delta a^*$	$\Delta b^*$	$\Delta E^*$	$\Delta L^*$	$\Delta a^*$	$\Delta b^*$	$\Delta E^*$	$\Delta L^*$	$\Delta a^*$	$\Delta b^*$	$\Delta E^*$
Walnut ( <i>Juglans regia</i> )	Ultrasonic	Control (Without Mordant)	-2,29	1,00	1,99	3,19	-4,67	3,61	8,57	10,40	-6,06	-6,06	8,96	11,87	-7,38	4,82	8,25	12,07
		Beet + Ferrous	-25,46	-6,29	-15,94	30,68	-21,72	-6,95	-9,63	24,75	-21,24	-6,88	-9,18	24,14	-21,81	-6,61	-7,96	24,14
		Beet + Alum	-2,33	0,04	-0,42	2,37	-4,98	2,76	5,39	7,83	-8,38	4,44	5,87	11,15	-9,47	4,59	6,18	12,20
		Beet + Copper	-5,95	-0,16	-0,15	5,95	-8,82	1,87	5,66	10,64	-12,72	2,36	5,81	14,18	-13,39	2,50	5,01	14,51
		Beet + Vinegar	-3,57	1,44	1,83	4,26	-5,29	4,50	8,99	11,36	-6,91	5,34	9,74	13,08	-7,87	5,64	9,77	13,75
		Synthetic Dye	-11,52	5,00	3,57	13,05	-11,58	6,63	5,97	14,62	-12,85	7,63	5,87	16,05	-11,90	8,71	6,94	16,29
	Classic	Control (Without Mordant)	-2,94	0,87	2,69	4,08	-5,36	3,69	8,31	10,55	-7,27	5,10	8,04	11,98	-8,91	6,01	7,51	13,11
		Beet + Ferrous	-25,33	-7,23	-16,65	31,16	-25,09	-6,76	-12,17	28,69	-24,75	-6,67	-10,45	27,68	-25,34	-6,29	-9,56	27,80
		Beet + Alum	-3,46	-3,46	1,06	-0,18	-7,05	3,96	6,47	10,35	-10,45	5,25	7,14	13,70	-12,06	6,19	6,75	15,14
		Beet + Copper	-4,42	0,16	0,16	4,42	-8,73	1,62	6,10	10,77	-11,04	2,30	6,35	12,93	-12,47	2,80	5,65	13,97
		Beet + Vinegar	-1,94	0,85	0,81	2,26	-3,24	3,08	9,40	10,40	-5,31	5,40	11,13	13,46	-7,55	6,01	9,83	13,78
		Synthetic Dye	-11,41	5,15	6,08	13,91	-11,20	6,41	9,30	15,90	-11,56	6,805	10,58	17,08	-10,61	8,23	11,55	17,71
Oak ( <i>Quercus petraea</i> )	Ultrasonic	Control (Without Mordant)	-2,13	0,65	2,69	3,49	-1,81	0,20	3,13	3,62	-4,52	-4,52	11,91	13,39	-6,70	5,18	11,56	14,33
		Beet + Ferrous	-32,86	-8,16	-19,87	39,26	-33,12	-7,43	-15,56	37,33	-33,64	-7,25	-14,05	37,17	-33,52	-7,02	-13,64	36,86
		Beet + Alum	-4,56	0,99	0,59	4,70	-9,66	4,78	8,30	13,60	-12,18	7,80	11,20	18,29	-15,10	8,51	9,44	19,73
		Beet + Copper	-18,85	-2,01	-4,40	19,46	-18,72	-1,33	0,08	18,77	-20,28	3,08	0,31	20,51	-22,55	0,67	1,58	22,61
		Beet + Vinegar	-2,09	0,52	0,93	2,35	-5,51	3,69	9,43	11,52	-6,84	4,48	12,20	14,69	-8,86	6,68	10,33	15,16
		Synthetic Dye	-10,68	6,85	6,76	14,38	-13,37	8,83	8,56	18,17	-14,46	8,975	10,18	19,83	-14,69	9,20	8,57	19,33
	Classic	Control (Without Mordant)	-1,36	0,11	3,22	3,50	-1,28	-0,13	3,35	3,59	-5,03	3,87	11,27	12,93	-7,74	5,03	10,68	14,11
		Beet + Ferrous	-34,05	-8,26	-22,46	41,62	-33,42	-8,51	-21,14	40,45	-33,48	-8,15	-16,00	37,98	-33,51	-7,68	-14,84	37,44
		Beet + Alum	-5,35	-5,35	-0,34	0,11	-9,96	3,31	7,56	12,93	-11,94	5,48	9,28	16,08	-15,41	6,68	8,31	18,74
		Beet + Copper	-17,05	-0,30	-4,10	17,54	-18,79	-0,14	1,90	18,89	-20,37	0,23	2,50	20,52	-23,08	1,85	3,43	23,41
		Beet + Vinegar	-1,03	0,52	3,21	3,41	-5,40	4,07	11,10	12,99	-9,83	4,73	10,81	15,36	-7,37	6,04	10,66	14,29
		Synthetic Dye	-14,97	7,53	6,65	18,02	-13,52	8,47	11,33	19,56	-14,49	9,145	11,85	20,83	-15,72	9,43	11,44	21,60
Scots pine ( <i>Pinus sylvestris</i> )	Ultrasonic	Control (Without Mordant)	-6,07	1,90	3,33	7,17	-14,67	9,01	14,30	22,38	-20,05	-20,05	17,14	29,52	-22,84	14,83	17,58	32,41
		Beet + Ferrous	-34,61	-2,94	-13,37	37,22	-40,76	-1,45	-9,29	41,83	-43,68	-1,21	-10,00	44,82	-45,52	0,02	-10,21	46,65
		Beet + Alum	-6,45	2,44	0,61	6,92	-18,86	12,01	12,80	25,76	-24,60	15,07	14,43	32,25	-27,67	16,94	13,62	35,18
		Beet + Copper	-10,33	-2,83	0,01	10,71	-17,29	2,65	11,50	20,93	-21,31	4,36	11,84	24,76	-26,09	8,95	13,23	30,59
		Beet + Vinegar	-4,70	2,73	1,95	5,77	-14,57	11,34	13,96	23,14	-18,33	13,93	15,53	27,77	-22,40	15,02	15,52	31,12
		Synthetic Dye	-23,26	18,56	11,95	32,06	-27,52	20,50	14,41	37,21	-30,42	21,37	13,29	39,48	-30,86	20,21	12,23	38,86
	Classic	Control (Without Mordant)	-4,80	1,04	3,31	5,92	-13,27	8,02	14,88	21,49	-16,40	10,28	17,32	25,97	-20,60	13,55	18,36	30,74
		Beet + Ferrous	-34,67	-0,70	-13,88	37,35	-43,70	1,71	-9,03	44,66	-45,75	1,17	-10,06	46,85	-47,72	2,44	-10,39	48,90
		Beet + Alum	-7,36	-7,36	2,20	0,95	-17,81	10,71	12,12	24,06	-21,92	13,56	13,46	29,07	-26,39	16,11	13,17	33,60
		Beet + Copper	-5,10	-3,59	-0,26	6,25	-13,60	1,11	13,90	19,48	-21,04	4,78	16,40	27,10	-23,57	9,02	16,26	30,02
		Beet + Vinegar	-3,46	2,06	2,14	4,55	-12,77	8,86	13,27	20,43	-16,93	10,60	15,30	25,15	-19,50	13,29	15,65	28,31
		Synthetic Dye	-20,12	15,46	9,54	27,11	-24,95	17,96	12,25	33,08	-27,23	18,52	11,99	35,04	-28,91	18,35	10,41	35,79
Oriental Beech ( <i>Fagus orientalis</i> )	Ultrasonic	Control (Without Mordant)	-4,09	0,70	2,59	4,89	-9,17	4,33	11,03	14,98	-10,54	-10,54	12,95	17,72	-13,44	6,94	12,82	19,83
		Beet + Ferrous	-23,46	-10,05	-12,73	28,52	-29,09	-9,20	-9,33	31,90	-31,35	-8,01	-9,25	33,65	-33,20	-7,78	-9,50	35,40
		Beet + Alum	-4,09	0,35	-0,34	4,11	-9,49	5,38	8,49	13,82	-12,38	7,48	11,05	18,20	-16,43	8,56	10,56	21,32
		Beet + Copper	-7,01	-1,71	0,45	7,23	-13,55	1,86	8,52	16,11	-17,81	3,21	7,91	19,75	-19,83	5,06	9,24	22,45
		Beet + Vinegar	-2,35	1,12	3,35	4,24	-5,34	3,58	11,15	12,86	-8,59	7,40	15,66	19,33	-12,25	8,85	15,00	21,29
		Synthetic Dye	-16,40	12,66	8,56	22,41	-19,95	13,95	9,03	25,96	-20,89	14,2	9,13	26,85	-21,41	13,94	9,84	27,37
	Classic	Control (Without Mordant)	-4,05	0,64	2,55	4,83	-7,09	2,50	8,37	11,25	-8,92	5,66	12,84	16,63	-11,74	6,66	12,99	18,72
		Beet + Ferrous	-19,25	-8,65	-10,57	23,60	-28,79	-7,37	-6,92	30,51	-31,89	-5,80	-6,78	33,11	-33,43	-5,47	-7,01	34,59
		Beet + Alum	-4,61	-4,61	0,63	-0,13	-9,75	5,04	7,82	13,47	-12,22	7,17	10,26	17,49	-16,14	8,37	10,12	20,81
		Beet + Copper	-3,26	-1,67	-0,66	3,72	-11,56	2,20	9,60	15,19	-15,32	3,29	10,25	18,72	-18,52	5,51	10,29	21,89
		Beet + Vinegar	-1,39	1,37	3,25	3,79	-7,26	5,04	12,19	15,05	-9,10	6,20	13,21	17,20	-11,81	9,23	15,22	21,36
		Synthetic Dye	-14,33	11,69	10,68	21,35	-18,30	13,33	10,66	25,02	-18,91	13,63	11,12	25,83	-19,41	13,78	10,78	26,13



a. Scots pine/ultrasonic dyeing, b. Scots pine/Classic dyeing, c. Oriental beech/ultrasonic dyeing, d. Oriental beech/Classic dyeing, e. Walnut/ultrasonic dyeing, f. Walnut/Classic dyeing, g. Oak/ultrasonic dyeing, h. Oak/Classic dyeing.

Figure 2. Color changes of wood species exposed to 50 h, 100 h and 150 h UV irradiation.



## CONCLUSIONS

Vinegar mordant applications provide resistance as much as other mordant applications; thus, a 100% natural dyestuff for wood surfaces is developed by using natural mordant in natural dye application. Ferrous sulphate is observed to be the mordant type with the highest color change among all the wood types in general. Generally beetroot extracts and mordant mixes were showed better color change performance compared to synthetic dye. Dyeing method has not affected the color changes results.

Generally mordants in use to have color variation and to retain the stain on wood material. Wood materials colored with metal mordants mixes had darker colors, but it seems metal mordant had negative effect on color change performances. These may have explained with the reactions between wood material and mordant mixes and definitely the effect of UV irradiation on different colors. Dark colors could absorption the light maximum in ratio 98%, and transparent colors 11%. Therefore darker wood species could have been much more color changes. As a matter of fact wood species colored with beet root + mordant mixes after weathering hold much more esthetic appearances. Mordant mixture extracts that can be preferred in places of the color change may be omitted.

In conclusion of these study, natural plant sources that have a great potential, but are not benefited will be able to be activated through ensuring and disseminating usage of natural dye plant extracts harmless to environmental and human health as colorant and protective materials in surface treatments for furniture products, and by means of proliferation thereof, new line of business will emerge. Therefore, more economical and eco-friendly compared to synthetic dyes, wood paints may be developed.

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