

Cardiopulmonary Anatomy & Physiology

Essentials of Respiratory Care

Fifth Edition

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CHAPTER 6

Oxygen Transport

Normal Blood Gas Value Ranges

Table 6-1

Blood Gas Value	Arterial	Venous
pH	7.35-7.4	7.30-7.40
PCO ₂	35-45 mm Hg	42-48 mmHg
HCO ₃	22-28 mEq/L	24-30 mEq/L
PO ₂	80-100 mmHg	35-45 mm Hg

OXYGEN TRANSPORT

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Oxygen Dissolved in the Blood Plasma

- Dissolve means that the gas maintains its precise molecular structure
- About .003 mL of O₂ will dissolve in 100 mL of blood for every 1 mm Hg of PO₂
- Thus, a PaO₂ of 100 = 0.3 mL

$$100 \times 0.003 = 0.3 \text{ mL}$$

Oxygen Dissolved in the Blood Plasma

- Written as 0.3 volumes percent (Vol%)
- Vol% represents amount of O₂ (in mL) that is in 100 mL of blood

$$\text{Vol\%} = \text{mL O}_2 / 100 \text{ mL bd}$$

Oxygen Dissolved in the Blood Plasma

- For example:
 - 10 vol% of O₂ means that there are 10 mL of O₂ in 100 mL of blood
 - Relatively small percentage of oxygen is transported in the form of dissolved oxygen

Oxygen Bound to Hemoglobin

- Each RBC contains about 280 million hemoglobin (Hb) molecules
- Normal adult Hb (Hb A) consists of:
 - 4 heme groups (iron portion of the Hb)
 - 4 amino acid chains: 2 alpha and 2 beta

Hemoglobin Molecule

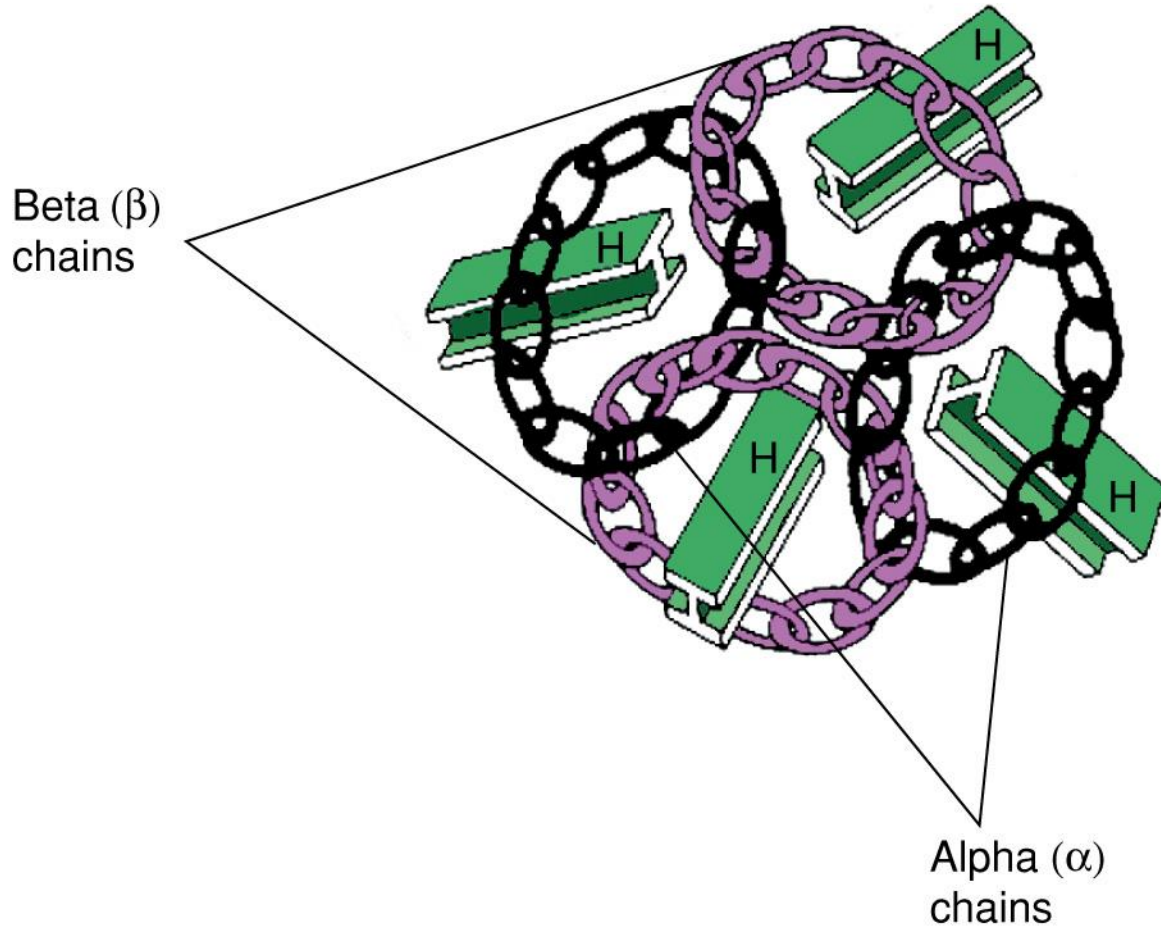
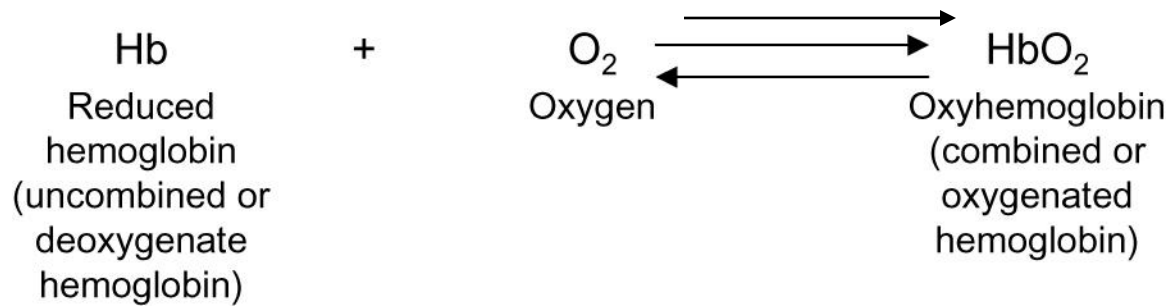


Fig. 6-1. Schematic illustration of a hemoglobin molecule.

Oxygen Bound to Hemoglobin



Oxygen Bound to Hemoglobin

- Oxyhemoglobin
 - Hemoglobin bound with oxygen
- Reduced hemoglobin or deoxyhemoglobin
 - Hemoglobin not bound with oxygen

Oxygen Bound to Hemoglobin

- Normal adult male Hb value:
 - 14-16 g/100 mL
- Normal adult female Hb value:
 - 12-15 g/100 mL

Oxygen Bound to Hemoglobin

- Clinically, the weight measurement of hemoglobin, in reference to 100 mL of blood, is referred to as either:
 - Gram percent of hemoglobin (g% Hb), or
 - Grams per deciliter (g/dL)

Quantity of Oxygen Bound to Hemoglobin

- Each g% Hb can carry 1.34 mL of oxygen
- Thus, if Hb level is 15 g%, and if Hb is fully saturated, about 20.1 vol% of O₂ will be bound to the Hb

$$\begin{aligned} \text{O}_2 \text{ bound to Hb} &= 1.34 \text{ mL O}_2 \times 15 \text{ g\% Hb} \\ &= 20.1 \text{ vol\% O}_2 \end{aligned}$$

Quantity of Oxygen Bound to Hemoglobin

- At a normal PaO_2 of 100 mm Hg, however, the Hb saturation (SaO_2) is only about 97% due to the following three normal physiologic shunts

Quantity of Oxygen Bound to Hemoglobin

- Thebesian venous drainage into the left atrium
- Bronchial venous drainage into pulmonary veins
- Alveoli that are under ventilated—dead space ventilation

Quantity of Oxygen Bound to Hemoglobin

- Thus, the amount of arterial oxygen in the preceding equation must be adjusted to 97 percent:

$$\begin{array}{r} 20.1 \text{ vol\% O}_2 \\ \times 0.97 \\ \hline 19.5 \text{ vol\% O}_2 \end{array}$$

Total Oxygen Content

- To determine the total amount of oxygen in 100 mL of blood, the following must be added together:
 - Dissolved oxygen
 - Oxygen bound to hemoglobin

Total Oxygen Content

- The following case study summarizes the calculations required to compute an individual's total oxygen content

Case Study—Anemic Patient

- 27-year-old woman
 - Long history of anemia (decreased hemoglobin concentration)
 - Showing signs of respiratory distress
 - Respiratory rate 36 breaths/min
 - Heart rate 130 beats/min
 - Blood pressure 155/90 mm Hg

Case Study—Anemic Patient

- Hemoglobin concentration is 6 g%
- PaO₂ is 80 mm Hg (SaO₂ 90%)

Case Study—Anemic Patient

- Based on this information, the patient's total oxygen content is computed as follows:

1. Dissolved O₂:

80 PaO₂

x 0.003 (dissolved O₂ factor)

0.24 vol% O₂

Case Study - Anemic Patient

2. Oxygen Bound to Hemoglobin:

6 g% Hb

x 1.34 (O₂ bound to Hb factor)

8.04 vol% O₂ (at SaO₂ of 100%)

8.04 vol% O₂

x 0.90 SaO₂

7.236 vol% O₂

Case Study—Anemic Patient

3. Total oxygen content:

7.236 vol% O₂ (bound to hemoglobin)

+ 0.24 vol% O₂ (dissolved O₂)

7.476 vol% O₂ (total amount of O₂/100 ml of blood)

Case Study—Anemic Patient

- Note:
 - Patient's total arterial oxygen content is less than 50 percent of normal
 - Her hemoglobin concentration, which is the primary mechanism for transporting oxygen, is very low
 - Once problem is corrected, respiratory distress should no longer be present

Total Oxygen Content

- Calculated for following:
 - Arterial Oxygen Content (CaO_2)
 - Mixed Venous Oxygen Content (CvO_2)
 - Oxygen Content of Pulmonary Capillary Blood (CcO_2)

Total Oxygen Content of Arterial Blood

- CaO_2 = Oxygen content of arterial blood
(Hb x 1.34 x SaO₂) + (PaO₂ x 0.003)

Total Oxygen Content of Mixed Venous Blood

- CvO_2 = Oxygen content of mixed venous blood

$$(Hb \times 1.34 \times SvO_2) + (PvO \times 0.003)$$

Total Oxygen Content of Pulmonary Capillary Blood

- CcO_2 = Oxygen content of pulmonary capillary blood

$$(Hb \times 1.34) + (PAO_2 \times 0.003)$$

Total Oxygen Content

- It will be shown later how various mathematical manipulations of the CaO_2 , CvO_2 , and CcO_2 values are used in different oxygen transport studies to reflect important factors concerning the patient's cardiac and ventilatory status.

OXYGEN DISSOCIATION CURVE

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Oxygen Dissociation Curve

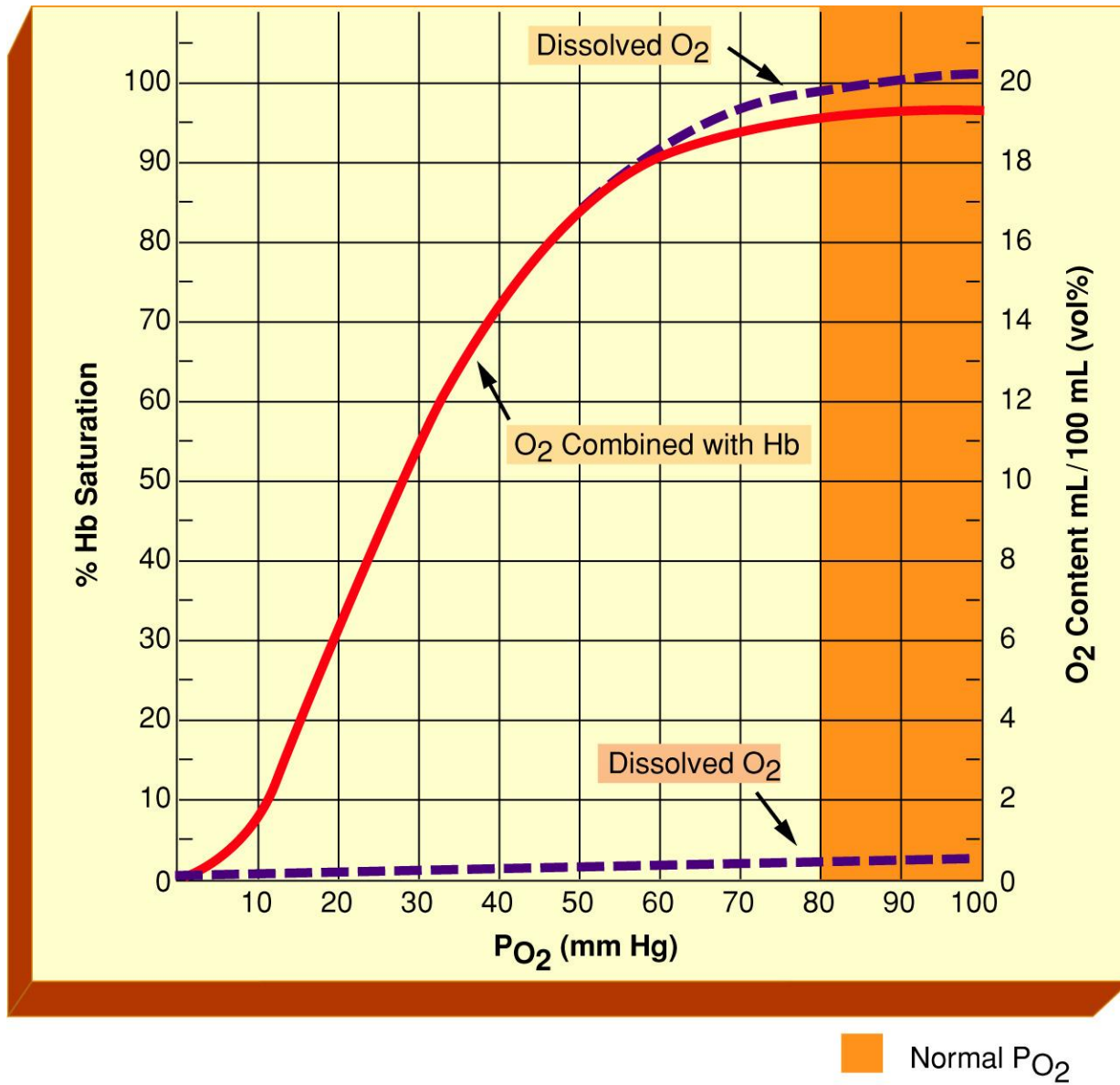


Fig. 6-2. Oxygen dissociation curve.

Clinical Significance of the Flat Portion of the Curve

- PO_2 can fall from 60 to 100 mm Hg and the hemoglobin will still be 90 percent saturated with oxygen
 - Excellent safety zone

Clinical Significance of the Flat Portion of the Curve

- As the Hb moves through the A-C system, a significant partial pressure difference continues to exist between the alveolar gas and blood, even after most O₂ has transferred
 - This enhances the diffusion of O₂

Clinical Significance of the Flat Portion of the Curve

- Increasing PO_2 beyond 100 mm Hg adds very little O_2 to the blood
 - Dissolved O_2 only
 - $(PO_2 \times 0.003 = \text{dissolved } O_2)$

Clinical Significance of the Flat Portion of the Curve

- A reduction of PO_2 below 60 mm Hg causes a rapid decrease in amount of O_2 bound to hemoglobin
- However, diffusion of oxygen from hemoglobin to tissue cells is enhanced

The P_{50}

- P_{50} represents the partial pressure at which the hemoglobin is 50 percent saturated with oxygen
- Normally, P_{50} is about 27 mm Hg

The P₅₀

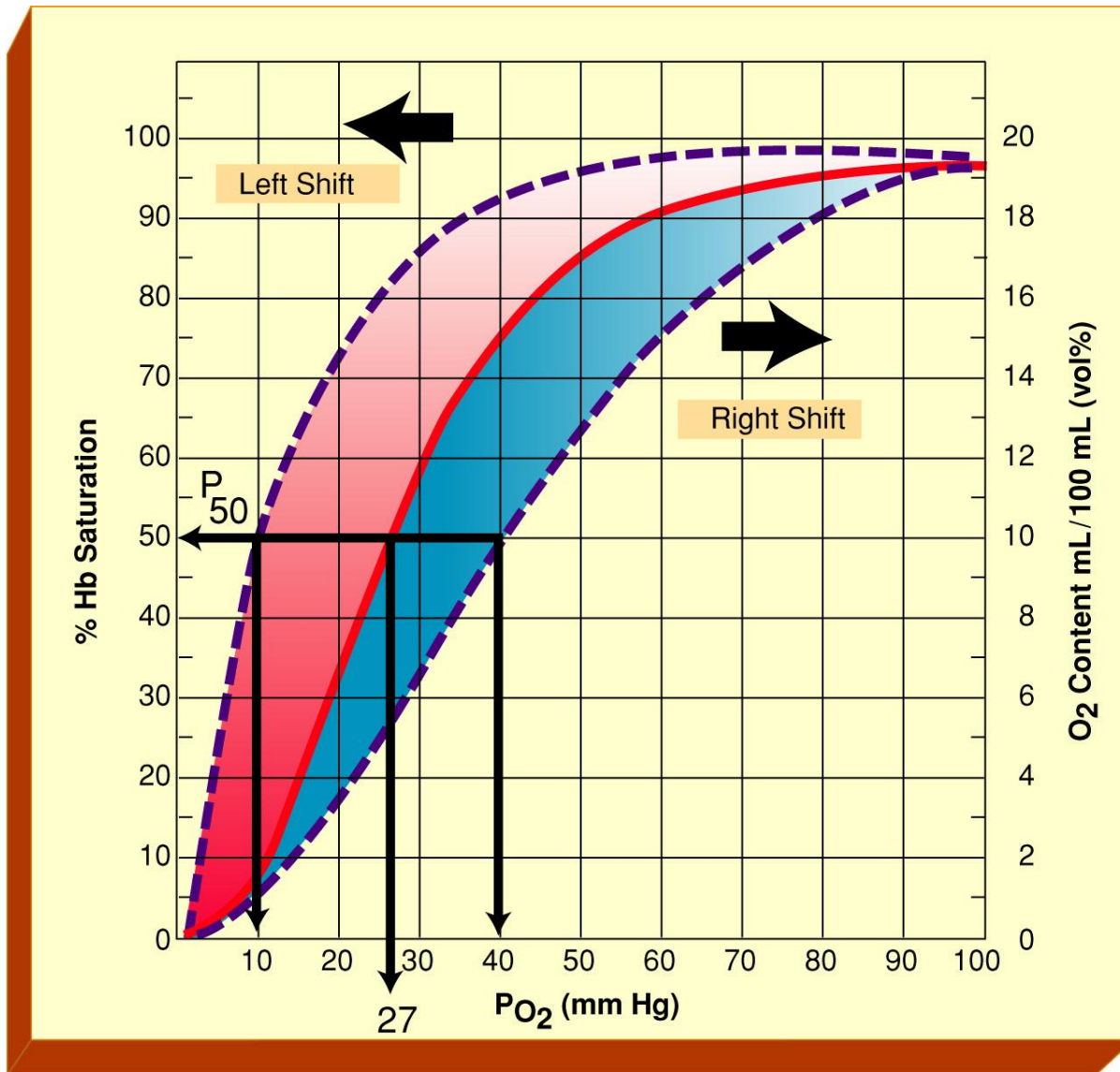


Fig. 6-3. The P₅₀ represents the partial pressure at which hemoglobin is 50 percent saturated with oxygen.

Factors that Shift Oxygen Dissociation Curve

- pH
- Temperature
- Carbon Dioxide
- 2,3-DPG
- Fetal Hemoglobin
- Carbon Monoxide Hemoglobin

Oxygen Dissociation Curve

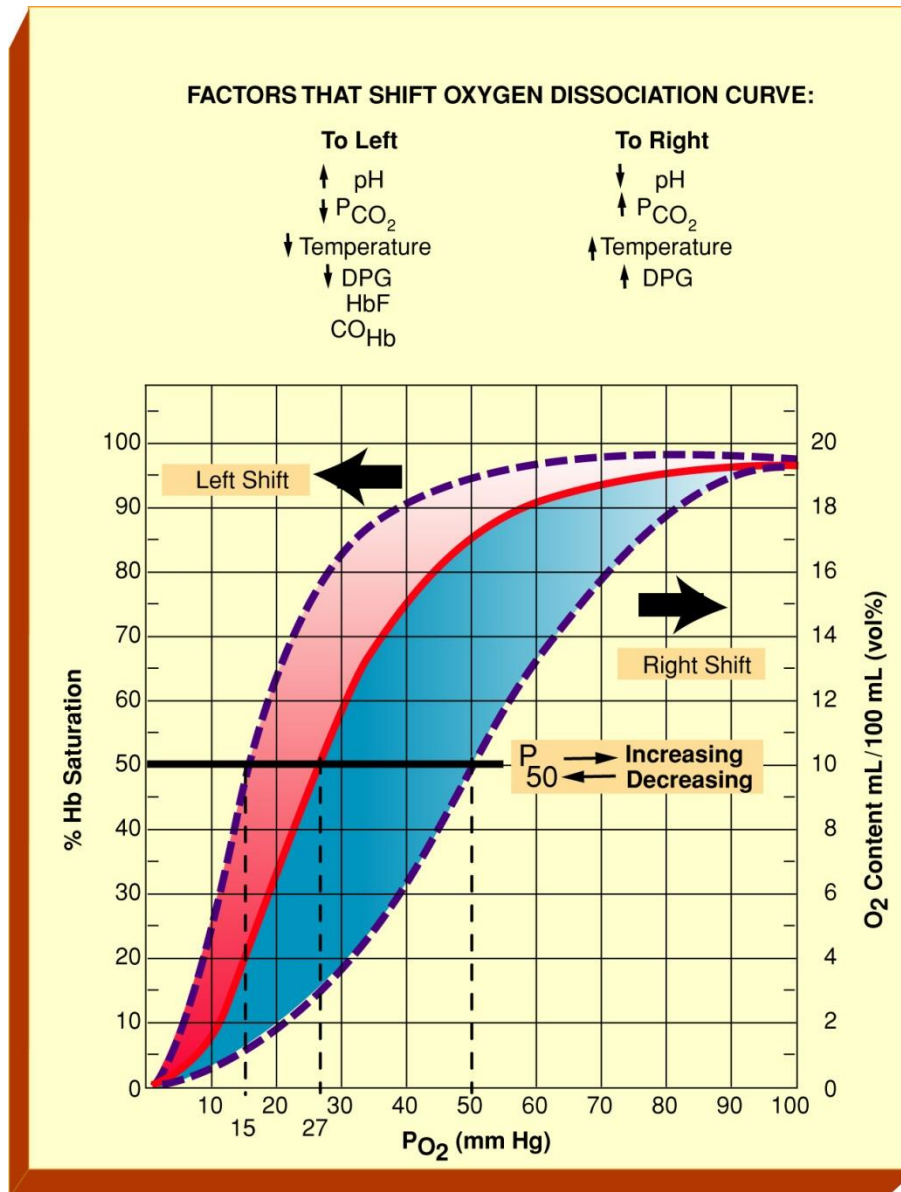


Fig. 6-4. Factors that shift the oxygen dissociation curve to the right and left.

CLINICAL SIGNIFICANCE OF SHIFTS IN THE O₂ DISSOCIATION CURVE

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The O₂ Dissociation Curve

- When an individual's blood PaO₂ is within normal limits (80-100 mm Hg):
 - Shift of oxygen dissociation curve to the right or left does not significantly affect hemoglobin's ability to transport oxygen to the peripheral tissues.

The O₂ Dissociation Curve

- However, when an individual's blood PaO₂ falls below the normal range:
 - A shift to the right or left can have a remarkable effect on the hemoglobin's ability to pick up and release oxygen.
 - This is because shifts below the normal range occur on the steep portion of the curve.

The O₂ Dissociation Curve

- For example, consider the loading and unloading of oxygen during the following clinical conditions:

Right Shifts: Loading of Oxygen in Lungs

- Picture the loading of oxygen onto hemoglobin as blood passes through the alveolar-capillary system at a time when the alveolar oxygen tension (PaO_2) is moderately low, around 60 mm Hg.

Right Shifts: Loading of Oxygen in Lungs

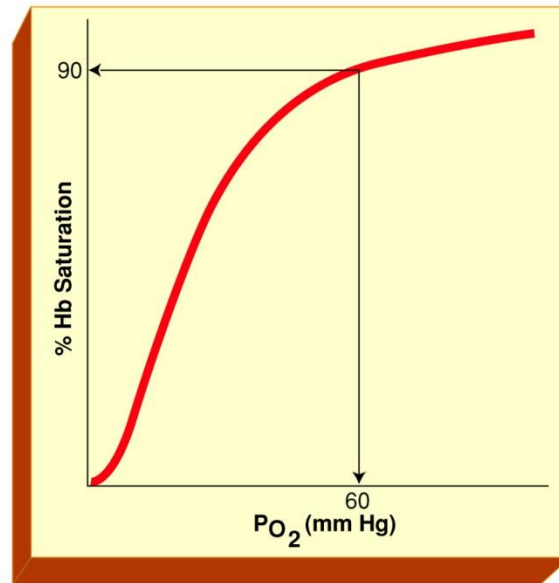
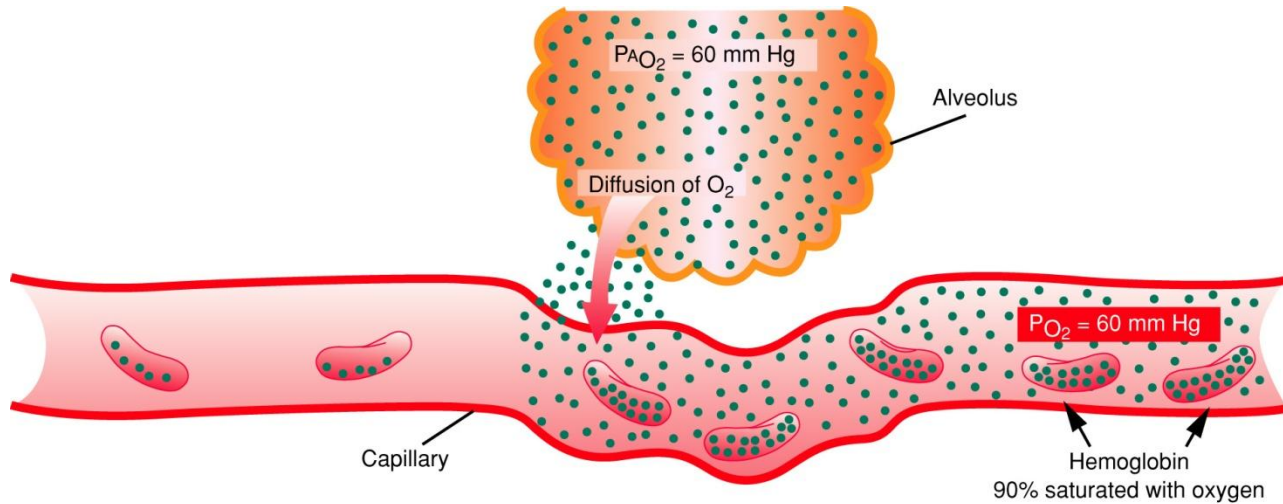


Fig. 6-5. Normally, when the PA_{O_2} is 60 mm Hg, the plasma PO_2 is about 60 mm Hg, and Hb is about 90% saturated.

Right Shifts: Loading of Oxygen in Lungs

- If, however, the oxygen dissociation curve shifts to the right, as indicated in Figure 6-6, the hemoglobin will be only about 75 percent saturated with oxygen as it leaves the alveoli.

Right Shifts: Loading of Oxygen in Lungs

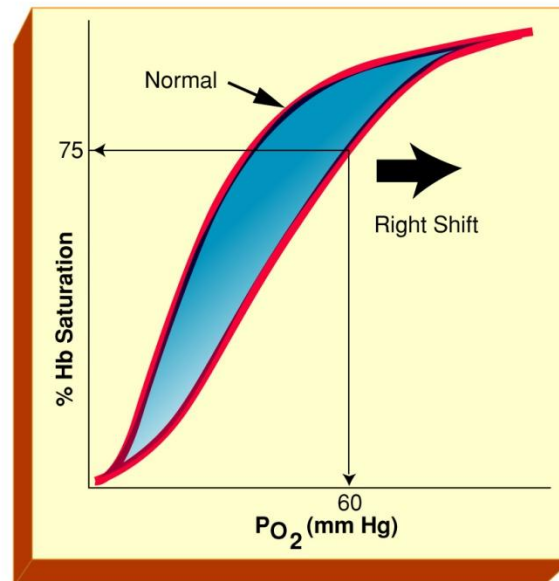
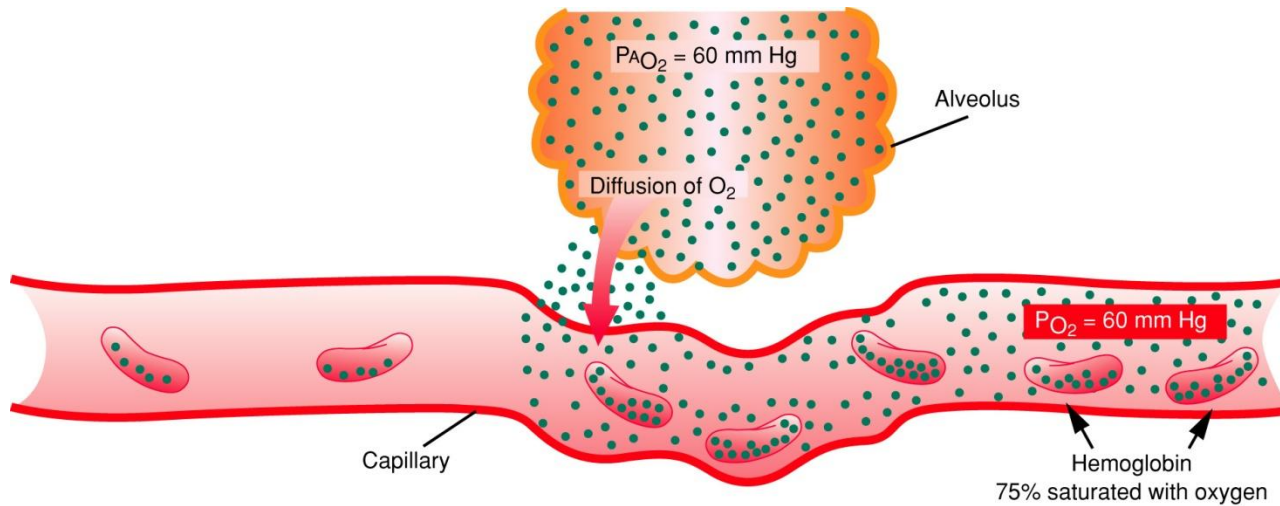


Fig. 6-6. When the PA_{O_2} is 60 mm Hg at a time when the curve has shifted to the right because of a pH of 7.1.

Right Shifts: Loading of Oxygen in Lungs

- In view of this gas transport phenomenon, it should be stressed that:
 - Total oxygen delivery may be much lower than indicated by a particular PaO_2 value when a disease process is present that causes the oxygen dissociation curve to shift to the right.

Right Shifts: Loading of Oxygen in Lungs

- Although total oxygen delivery may be decreased in the above situation:
 - Plasma PO_2 at the tissue sites does not have to fall as much to unload oxygen

Right Shifts: Unloading of Oxygen at the Tissues

- For example, if tissue cells metabolize 5 vol% oxygen at a time when the oxygen dissociation is in the normal position:
 - Plasma PO_2 must fall from 60 mm Hg to about 35 mm Hg to free 5 vol% oxygen from the hemoglobin
 - See Figure 6-7

Right Shifts: Unloading of Oxygen at the Tissues

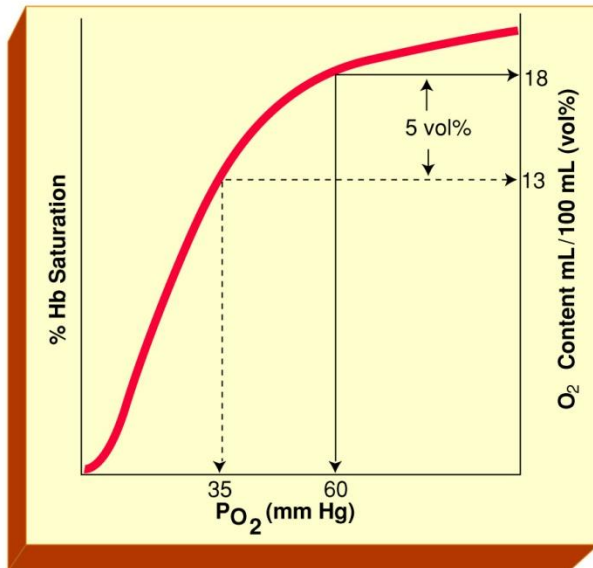
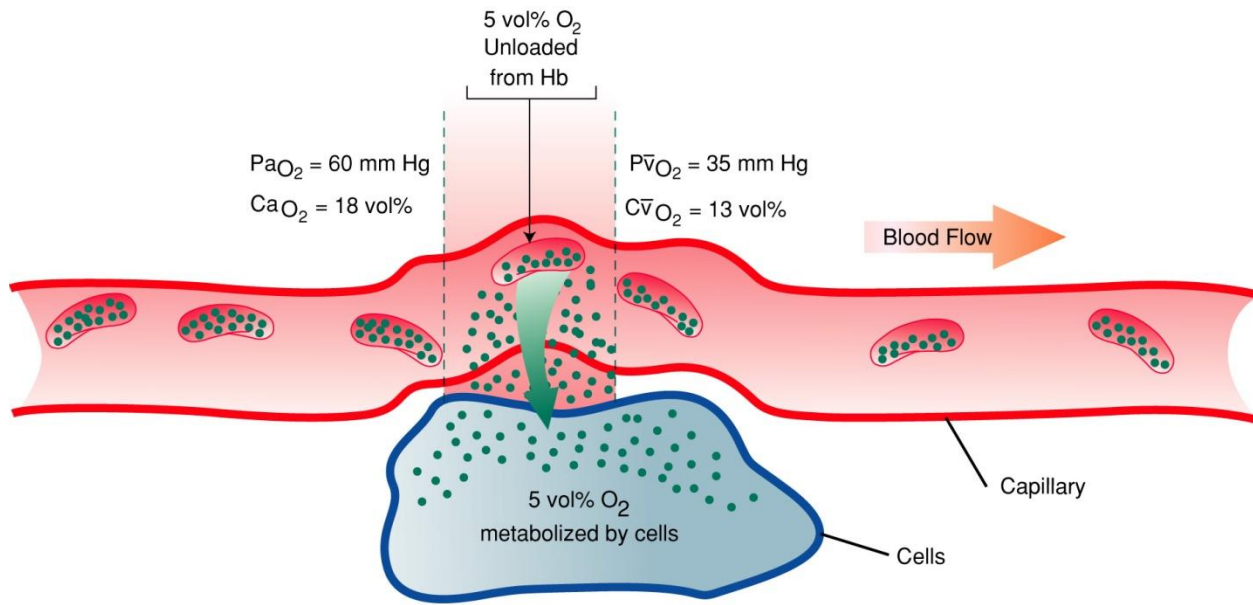


Fig. 6-7. Normally, when the plasma PO₂ is 60 mm Hg, the PO₂ must fall to about 35 mm Hg to free 5 vol% oxygen for metabolism.

Right Shifts: Unloading of Oxygen at the Tissues

- If, however, the curve shifts to the right in response to a pH of 7.1:
 - Plasma PO_2 at tissue sites would only have to fall from 60 mm Hg to about 40 mm Hg to unload 5 vol% oxygen from the hemoglobin
 - See Figure 6-8

Right Shifts: Unloading of Oxygen at the Tissues

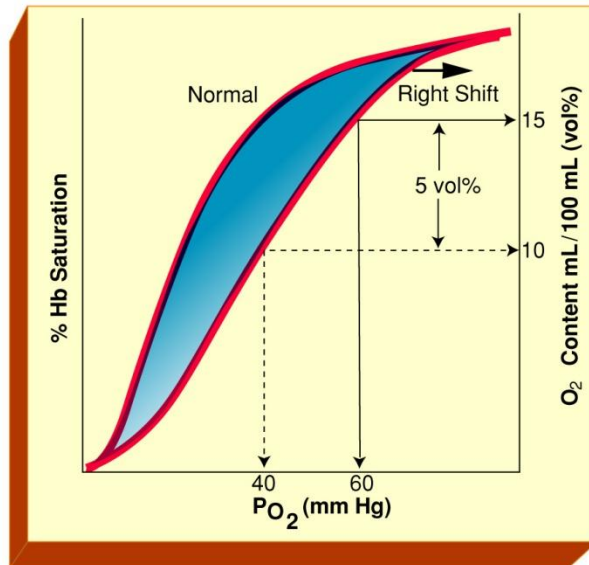
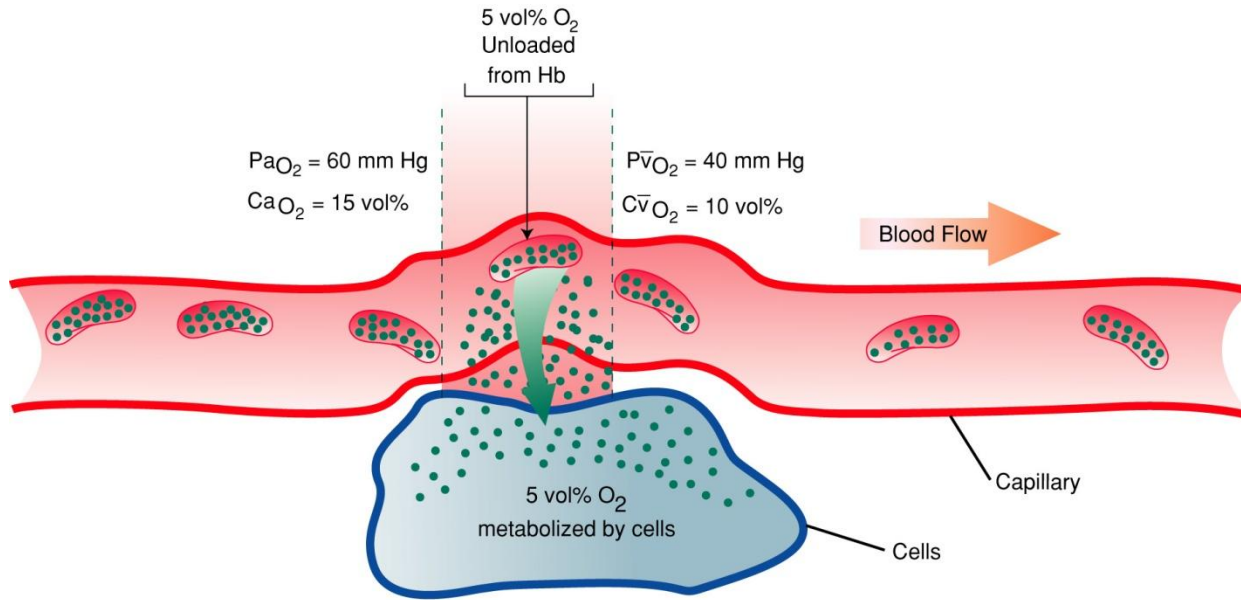


Fig. 6-8. When the plasma PO₂ is 60 mm Hg at a time when the curve is to the right because of pH of 7.1, the PO₂ must fall to about 40 mm Hg to free 5 vol% oxygen for metabolism.

Left Shifts: Loading of Oxygen in the Lungs

- If the oxygen dissociation curve shifts to left when the PAO_2 is 60 mm Hg at a time when the curve has shifted to the left because of a pH of 7.6:
 - Hemoglobin will be about 95 percent saturated with oxygen
 - See Figure 6-9

Left Shifts: Loading of Oxygen in the Lungs

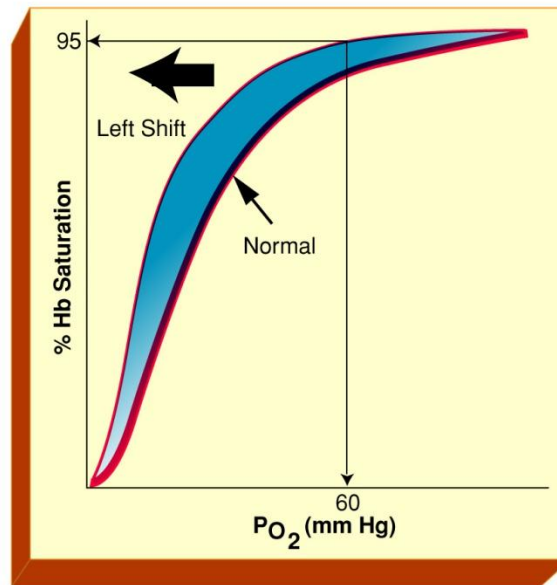
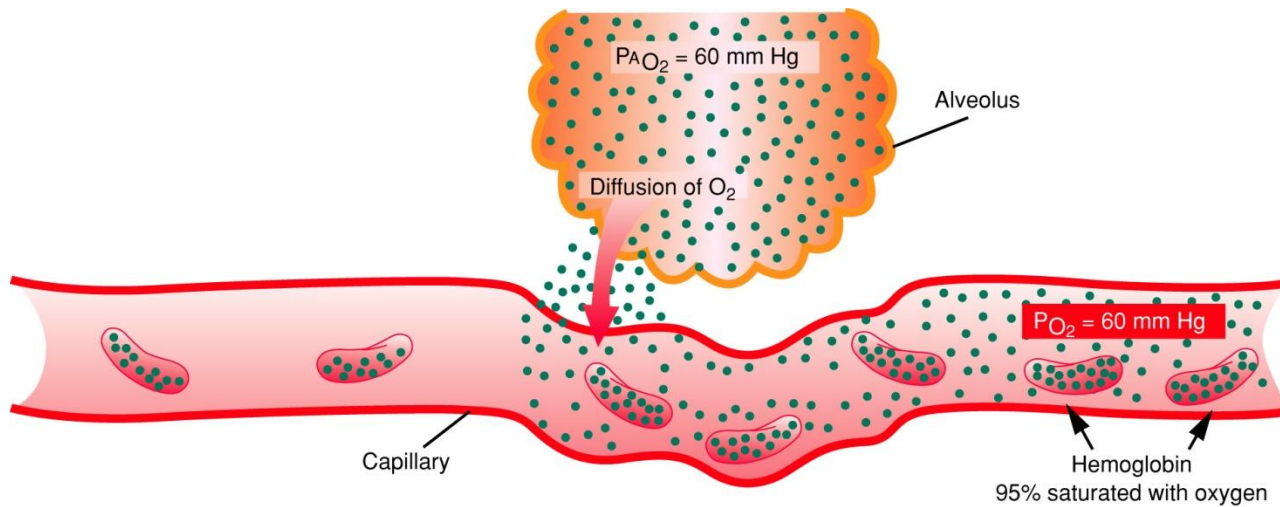


Fig. 6-9. When the PA_{O_2} is 60 mm Hg at a time when the curve has shifted to the left because of a pH of 7.6.

Left Shifts: Unloading of Oxygen at the Tissues

- Although total oxygen increases in the previously mentioned situation:
 - Plasma PO_2 at the tissue sites must decrease more than normal in order for oxygen to dissociate from the hemoglobin

Left Shifts: Unloading of Oxygen at the Tissues

- For example, if the tissue cells require 5 vol% oxygen at a time when the oxygen dissociation curve is normal, the plasma PO_2 will fall from 60 mm Hg to about 35 mm Hg to free 5 vol% of oxygen from the hemoglobin
 - See Figure 6-7

Oxygen for Metabolism

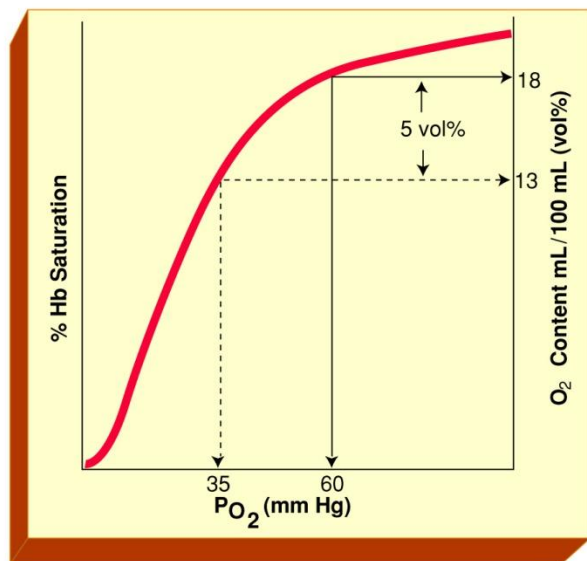
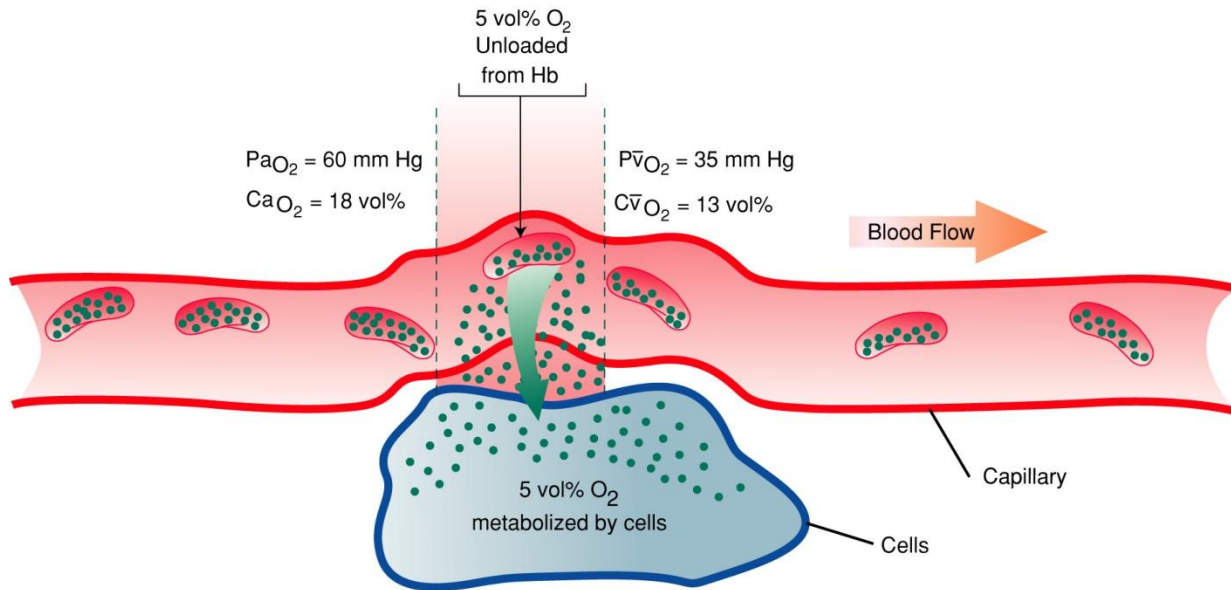


Fig. 6-7. Normally, when the plasma PO_2 is 60 mm Hg, the PO_2 must fall to about 35 mm Hg to free 5 vol% oxygen for metabolism.

Left Shifts: Unloading of Oxygen at the Tissues

- If, however, the curve shifts to the left because of a pH of 7.6:
 - Plasma PO_2 at tissue sites would have to fall from 60 mm Hg to about 30 mm Hg to unload 5 vol% oxygen from the hemoglobin
 - See Figure 6-10

Left Shifts: Unloading of Oxygen at the Tissues

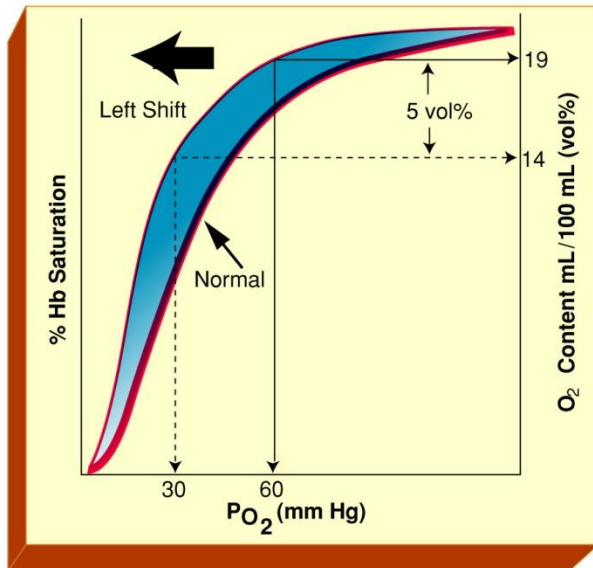
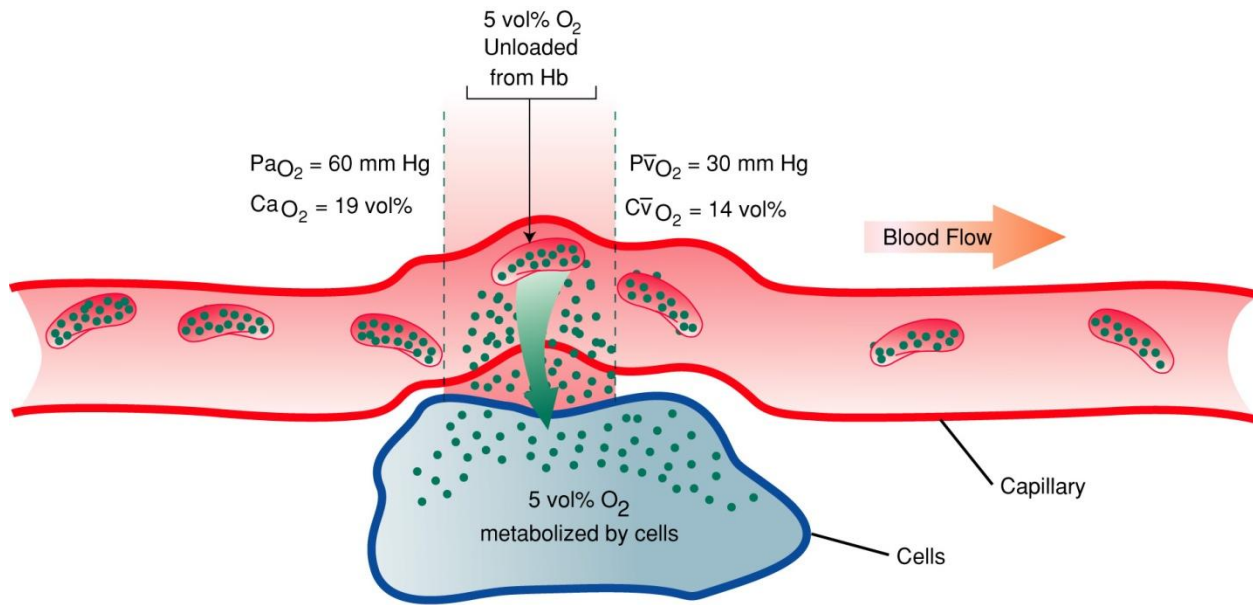


Fig. 6-10. When the plasma PO₂ is 60 mm Hg at a time when the curve is to the left because of pH of 7.6, the PO₂ must fall to about 30 mm Hg to free 5 vol% oxygen for metabolism.

Oxygen Transport Calculations

- Total Oxygen Delivery
- Arterial-Venous Oxygen Content Difference
- Oxygen Consumption
- Oxygen Extraction Ratio
- Mixed Venous Oxygen Saturation
- Pulmonary Shunting

Total Oxygen Delivery: $DO_2 = QT \times (CaO_2 \times 10)$

- The total amount of oxygen delivered or transported to the peripheral tissues is dependent on
 1. The body's ability to oxygenate blood
 2. The hemoglobin concentration
 3. The cardiac output

Total Oxygen Delivery (DO₂) is calculated as follows:

$$DO_2 = QT \times (CaO_2 \times 10)$$

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Total Oxygen Delivery

- For example:
 - If a patient has a cardiac output of 5 L/min and a CaO_2 of 20 vol%
 - DO_2 will be about 1000 mL of oxygen per minute:

Total Oxygen Delivery

$$\begin{aligned} \text{DO}_2 &= Q_T \times (\text{CaO}_2 \times 10) \\ &= 5 \text{ L/min} \times (20 \text{ vol\%} \times 10) \\ &= 1000 \text{ ml O}_2/\text{min} \end{aligned}$$

Note: The normal DO_2 is about 1000 ml/min

Total Oxygen Delivery

- DO_2 decreases in response to:
 - Low blood oxygenation
 - Low PaO_2
 - Low SaO_2
 - Low hemoglobin concentration
 - Low cardiac output

Total Oxygen Delivery

- DO_2 increases in response to increased blood oxygenation
 - Increased PaO_2
 - Increased SaO_2
 - Increased hemoglobin concentration
 - Increased cardiac output

Arterial-Venous Oxygen Content Difference

$$C(a-v)O_2 = CaO_2 - CvO_2$$

- The $C(a-v)O_2$ is the difference between the CaO_2 and the CvO_2

Arterial-Venous Oxygen Content Difference

- Normally, the CaO_2 is about 20 vol% and the CvO_2 is 15 vol%.
- Thus, the $C(a-v)O_2$ is about 5 vol%:

Arterial-Venous Oxygen Content Difference

$$\begin{aligned}C(a-v)O_2 &= CaO_2 - CvO_2 \\ &= 20 \text{ vol\%} - 15 \text{ vol\%} \\ &= 5 \text{ vol\%}\end{aligned}$$

Normally, 5 vol%

Oxygen Dissociation Curve

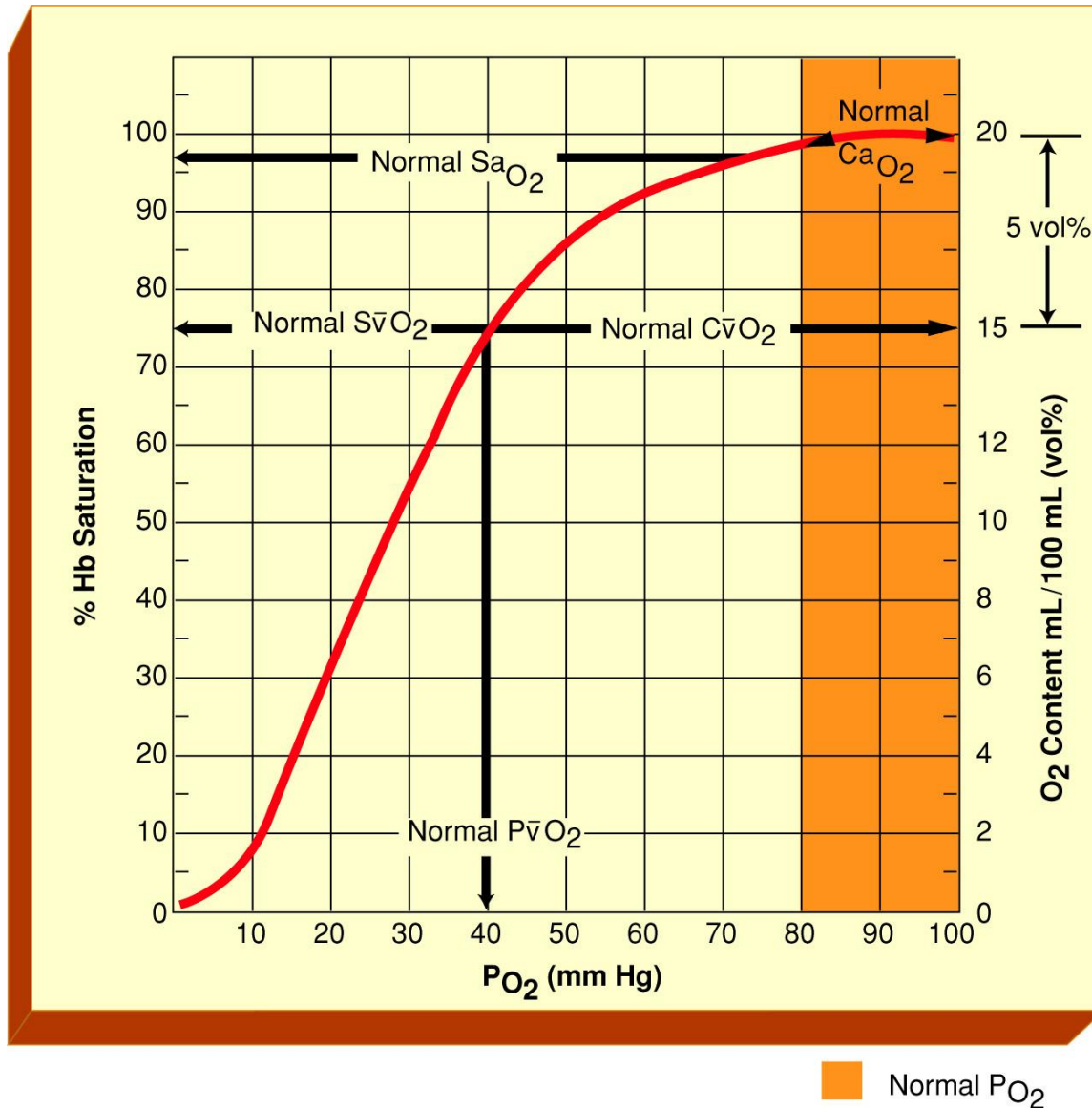


Fig. 6-11. Oxygen dissociation curve. Summary of important values.

Factors that Increase the $C(a-v)O_2$

- Decreased cardiac output
- Periods of increased oxygen consumption
 - Exercise
 - Seizures
 - Shivering
 - Hyperthermia

Factors that Decrease the $C(a-v)O_2$

- Increased cardiac output
- Skeletal relaxation
 - Induced by drugs
- Peripheral shunting
 - Sepsis, trauma

Factors that Decrease the $C(a-v)O_2$

- Certain poisons
 - Cyanide
- Hypothermia

Oxygen Consumption

- Amount of oxygen extracted by the peripheral tissues during the period of one minute
- Also called oxygen uptake (VO_2)

Oxygen Consumption

- Calculated as follows:

$$VO_2 = Q_T [C(a-v)O_2 \times 10]$$

- Case: If a patient has a cardiac output of 5 L/min and a $C(a-v)O_2$ of 5 vol%:
 - What is the total amount of oxygen consumed by the tissue cells in one minute?

Oxygen Consumption

- For example:
- If an individual has a cardiac output of 5 L/min and a $C(a-v)O_2$ of 5 vol%
 - Total amount of oxygen metabolized by the tissue cells in one minute will be 250 mL:

Oxygen Consumption

$$\begin{aligned} \text{VO}_2 &= Q_T [\text{C(a-v)}\text{O}_2 \times 10] \\ &= 5 \text{ L/min} \times 5 \text{ vol\%} \times 10 \\ &= 250 \text{ ml O}_2/\text{min} \end{aligned}$$

Note: The VO_2 is normally about 250 ml O_2/min

Factors that Increase VO_2

- Exercise
- Seizures
- Shivering
- Hyperthermia
- Body Size

Factors that Decrease $\dot{V}O_2$

- Skeletal Muscle Relaxation
 - Induced by drugs
- Peripheral shunting
 - Sepsis, trauma
- Certain poisons
 - Cyanide
- Hypothermia

Oxygen Extraction Ratio

- Oxygen extraction ratio (O_2ER) is the amount of oxygen extracted by the peripheral tissues divided by the amount of oxygen delivered to the peripheral cells
- Also called:
 - Oxygen coefficient ratio
 - Oxygen utilization ratio

Oxygen Extraction Ratio Calculated as Follows:

$$O_2ER = \frac{CaO_2 - CvO_2}{CaO_2}$$

Oxygen Extraction Ratio Calculated as Follows:

- In considering the normal CaO_2 of 20 vol% and the normal CvO_2 of 15 vol%:
- O_2ER is about 25 percent

Oxygen Extraction Ratio

$$\begin{aligned} O_2ER &= \frac{CaO_2 - CvO_2}{CaO_2} \\ &= \frac{20 \text{ vol\%} - 15 \text{ vol\%}}{20 \text{ vol\%}} \\ &= \frac{5 \text{ vol\%}}{20 \text{ vol\%}} \\ &= 0.25 \end{aligned}$$

Oxygen Extraction Ratio

- O_2ER provides an important view of the oxygen transport status when O_2 consumption remains the same
- For example, consider the following two cases with the same $C(a-v)O_2$ (5 vol%), but with different DO_2

Normal CaO_2 and CvO_2

$$\begin{array}{r} \text{CaO}_2 \quad 20 \text{ vol\%} \\ - \text{CvO}_2 \quad 15 \text{ vol\%} \\ \hline \text{C(a-v)O}_2 \quad 5 \text{ vol\%} \end{array}$$

The $\text{O}_2\text{ER} = 25\%$

Decreased CaO_2 and CvO_2

$$\begin{array}{r} CaO_2 \quad 10 \text{ vol\%} \\ - CvO_2 \quad 5 \text{ vol\%} \\ \hline C(a-v)O_2 \quad 5 \text{ vol\%} \end{array}$$

The $O_2ER = 50\%$

Factors that Increase O_2ER

- Decreased cardiac output
- Periods of increased O_2 consumption
 - Exercise
 - Seizures
 - Shivering
 - Hyperthermia
 - Anemia

Factors that Decrease O₂ER

- Increased cardiac output
- Skeletal muscle relaxation
 - Drug induced
- Peripheral shunting (e.g., sepsis)

Factors that Decrease O₂ER

- Certain poisons
 - Cyanide
- Hypothermia
- Increased Hb
- Increased arterial oxygenation (PaO₂)

Mixed Venous Oxygen Saturation (SvO₂)

- Changes in the SvO₂ can be used to detect changes in the:
 - C(a-v)O₂
 - VO₂
 - O₂ER

Factors that Decrease the SvO₂

- Decreased cardiac output
- Exercise
- Seizures
- Shivering
- Hyperthermia

Factors that Increase the SvO₂

- Increased cardiac output
- Skeletal muscle relaxation
 - Drug induced
- Peripheral shunting
 - Sepsis

Factors that Increase the SvO₂

- Certain poisons
 - Cyanide
- Hypothermia

Oxygen Transport Calculations

Table 6-10

Clinical Factors	DO_2	VO_2	$C(a-v)O_2$	O_2ER	SvO_2
$\uparrow O_2$ loading	\uparrow	Same	Same	\downarrow	\uparrow
$\uparrow Hb$					
$\uparrow PaO_2$					
$\downarrow PaCO_2$					
$\uparrow pH$					
\uparrow Temperature					

Oxygen Transport Calculations

Table 6-10

Clinical Factors	DO₂	VO₂	C(a-v)O₂	O₂ER	SvO₂
↓ O ₂ loading	↓	Same	Same	↑	↓
↓ Hb					
↓ PaCO ₂					
↓ pH					
↓ PaO ₂					
Anemia					
↓ Temperature					

Oxygen Transport Calculations

Table 6-10

Clinical Factors	DO₂	VO₂	C(a-v)O₂	O₂ER	SvO₂
↑ Metabolism	Same	↑	↑	↑	↓
Exercise					
Seizures					
Hyperthermia					
Shivering					

Oxygen Transport Calculations

Table 6-10

Clinical Factors	DO₂	VO₂	C(a-v)O₂	O₂ER	SvO₂
↓ Metabolism	Same	↓	↓	↓	↑
Hypothermia					
Skeletal muscle relaxation					

Oxygen Transport Calculations

Table 6-10

Clinical Factors	DO₂	VO₂	C(a-v)O₂	O₂ER	SvO₂
↓ Cardiac Output	↓	Same	↑	↑	↓

Oxygen Transport Calculations

Table 6-10

Clinical Factors	DO₂	VO₂	C(a-v)O₂	O₂ER	SvO₂
↑ Cardiac Output	↑	Same	↓	↓	↑

Oxygen Transport Calculations

Table 6-10

Clinical Factors	DO_2	VO_2	$C(a-v)O_2$	O_2ER	SvO_2
Peripheral shunting	Same	↓	↓	↓	↑

Oxygen Transport Calculations

Table 6-10

Clinical Factors	DO₂	VO₂	C(a-v)O₂	O₂ER	SvO₂
Certain Poisons	Same	↓	↓	↓	↑

Pulmonary Shunting

- Portion of cardiac output that moves from the right side to the left side of the heart without being exposed to alveolar oxygen (PAO_2).

Pulmonary Shunting

- Clinically, pulmonary shunting can be subdivided into:
 - Absolute Shunt
 - Also called True Shunt
- Relative Shunt
 - Also called shunt-like effects

Absolute Shunt

- An anatomic shunt (true shunt)
 - When blood flows from the right side of heart to the left side without coming in contact with an alveolus for gas exchange
 - See Figure 6-12, A and B

Pulmonary Shunting

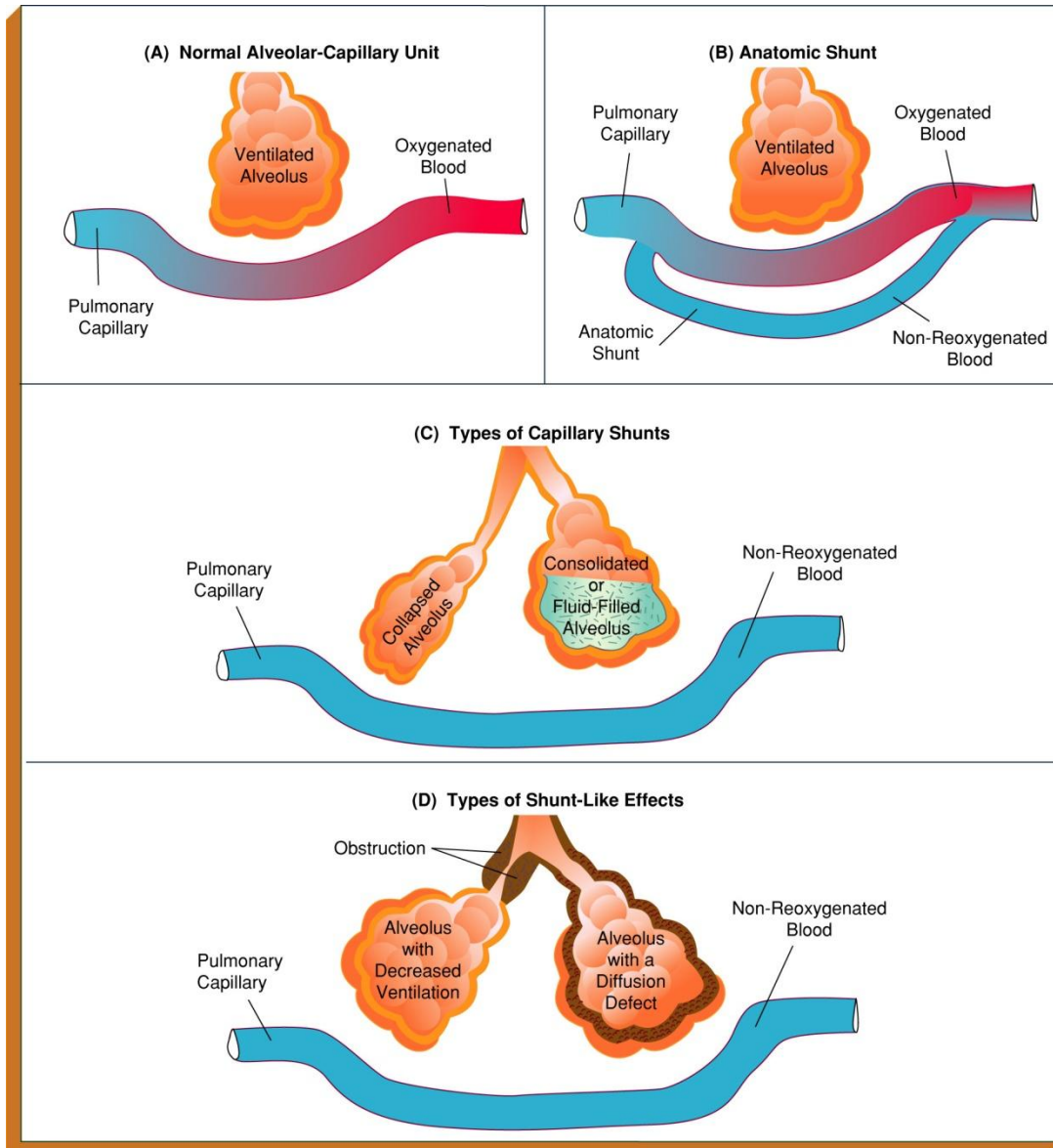


Fig. 6-12. Pulmonary shunting.

Common Causes of Absolute Shunting

- Congenital heart disease
- Intrapulmonary fistula
- Vascular lung tumors

Common Causes of Absolute Shunting

- Capillary shunting is commonly caused by:
 - Alveolar collapse or atelectasis
 - Alveolar fluid accumulation
 - Alveolar consolidation
 - See Figure 6-12, C

Common Causes of Absolute Shunting

- When pulmonary capillary perfusion is in excess of alveolar ventilation, a relative or shunt-like effect is said to exist
 - See Figure 6-12, D

Common Causes of This Form of Shunting

- Hypoventilation
- Ventilation/perfusion mismatches
 - Chronic emphysema, bronchitis, asthma
- Alveolar-capillary diffusion defects
 - Alveolar fibrosis or alveolar edema

Venous Admixture

- Venous mixture is the mixing of shunted, non-reoxygenated blood with reoxygenated blood distal to the alveoli
 - Downstream in the pulmonary venous system
 - See Figure 6-13

Venous Admixture

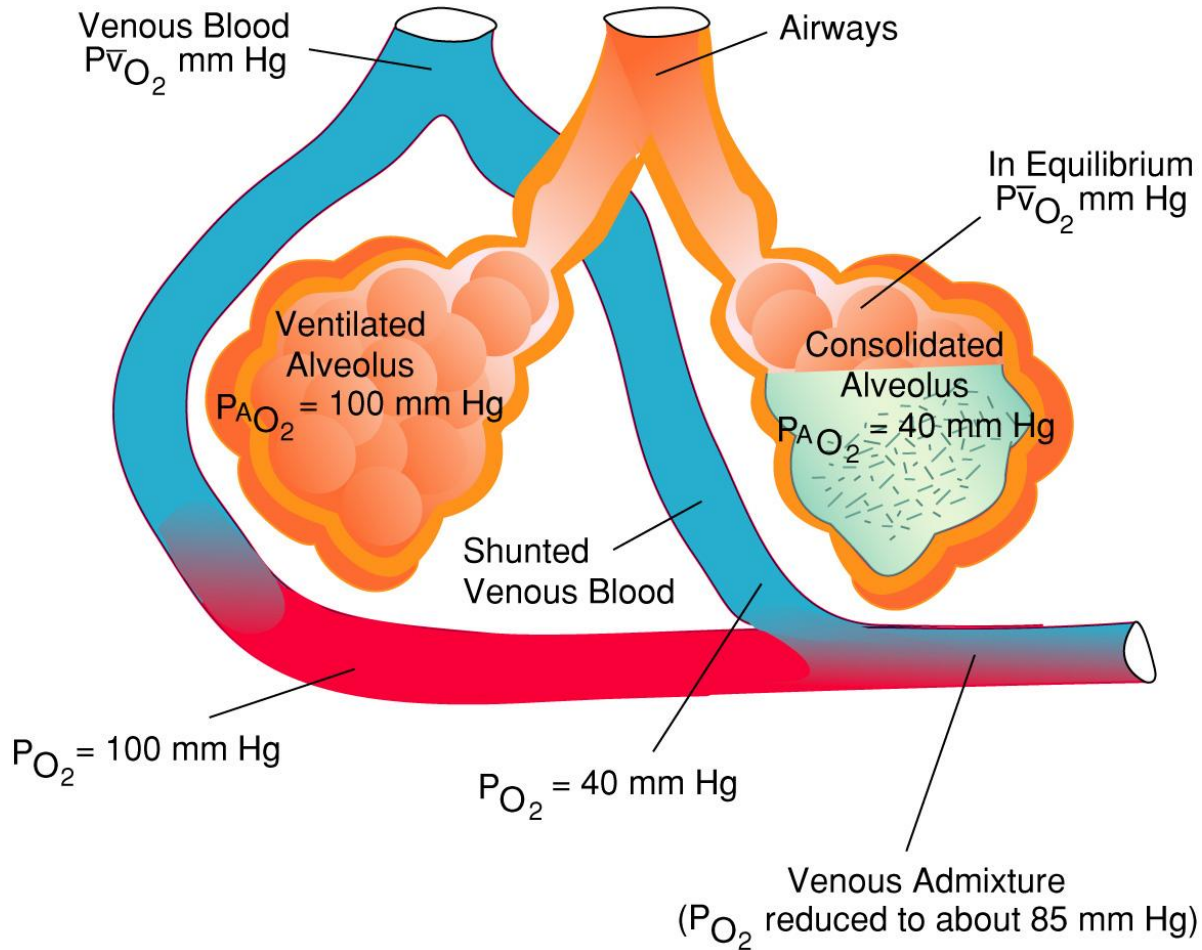


Fig. 6-13. Venous admixture occurs when reoxygenated blood mixes with non-reoxygenated blood.

Pulmonary Equation

$$\frac{Q_s}{Q_T} = \frac{CcO_2 - CaO}{CcO_2 - CvO_2}$$

Shunt Equation Clinical Information Needed

- PB
- PaO₂
- PaCO₂
- PvO₂
- Hb
- PAO₂
- FIO₂

Case Study: Motorcycle Accident Victim

- A 38-year-old man is on a volume-cycled mechanical ventilator on a day when the barometric pressure is 750 mm Hg
- Patient is receiving an FIO_2 of .70
 - The following clinical data are obtained:

Case Study: Motorcycle Accident Victim

- Hb: 13 g%
- PaO₂: 50 mm Hg (SaO₂ = 85%)
- PaCO₂: 43 mm Hg
- PvO₂: 37 mm Hg (SvO₂ = 65%)

Case Study: Motorcycle Accident Victim

- With this information, the patient's PAO_2 , CcO_2 , CaO_2 , and CvO_2 can now be calculated

Case Study: Motorcycle Accident Victim

$$\begin{aligned} 1. \text{ PAO}_2 &= (\text{PB} - \text{PH}_2\text{O}) \text{ FIO}_2 - \text{PaCO}_2 (1.25) \\ &= (750 - 47) 0.70 - 43 (1.25) \\ &= (703) 0.70 - 53.75 \\ &= 492.1 - 53.75 \\ &= 438.35 \text{ mm Hg} \end{aligned}$$

Case Study: Motorcycle Accident Victim

$$\begin{aligned} 2. CcO_2 &= (Hb \times 1.34) + (PAO_2 \times 0.003) \\ &= (13 \times 1.34) + (438.35 \times 0.003) \\ &= 17.42 + 1.315 \\ &= 18.735 \text{ (vol\% O}_2\text{)} \end{aligned}$$

Case Study: Motorcycle Accident Victim

$$\begin{aligned} 3. \text{CaO}_2 &= (\text{Hb} \times 1.34 \times \text{SaO}_2) + (\text{PaO}_2 \times 0.003) \\ &= (13 \times 1.34 \times .85) + (50 \times 0.003) \\ &= 14.807 + 0.15 \\ &= 14.95 \text{ (vol\% O}_2\text{)} \end{aligned}$$

Case Study: Motorcycle Accident Victim

$$\begin{aligned} 4. \text{CaO}_2 &= (\text{Hb} \times 1.34 \times \text{SvO}_2) + (\text{PvO}_2 \times 0.003) \\ &= (13 \times 1.34 \times .65) + (37 \times 0.003) \\ &= 11.323 + 0.111 \\ &= 11.434 \text{ (vol\% O}_2\text{)} \end{aligned}$$

Case Study: Motorcycle Accident Victim

- Based on the previous calculation the patient's degree of pulmonary shunting can now be calculated:

$$\begin{aligned}\frac{Q_s}{Q_T} &= \frac{CcO_2 - CaO_2}{CcO_2 - CvO_2} \\ &= \frac{18.735 - 14.957}{18.375 - 11.434} \\ &= \frac{3.778}{7.301} \\ &= 0.515\end{aligned}$$

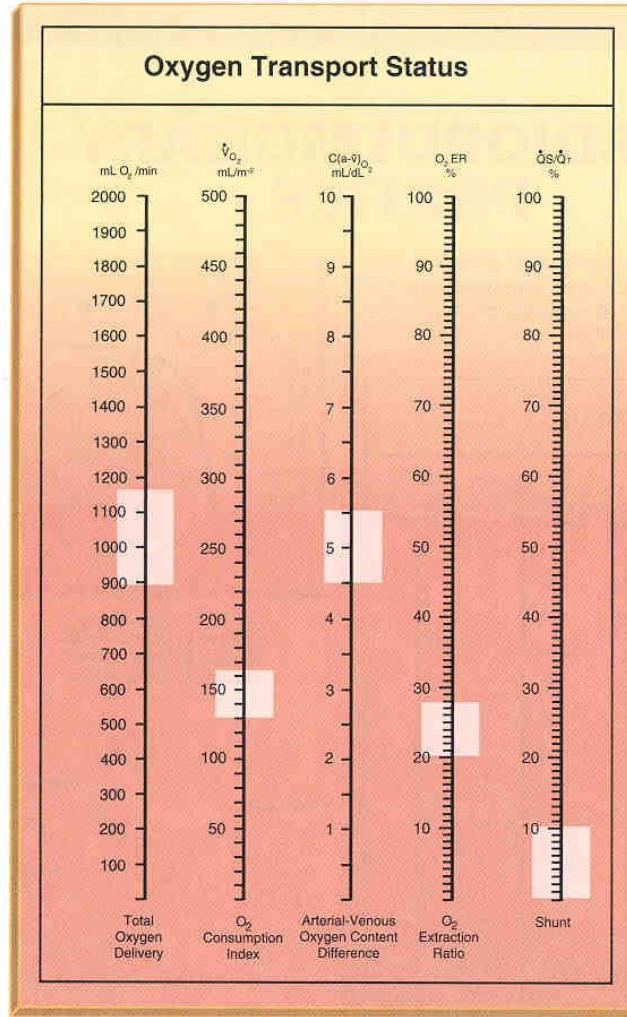
Clinical Significance of Pulmonary Shunting

- <10%
 - Normal status
- 10 to 20%
 - Indicates intrapulmonary abnormality

Clinical Significance of Pulmonary Shunting

- 20 to 30%
 - Significant intrapulmonary diseases
- > 30%
 - Potentially life-threatening

Appendix V



Blood Gas Values

pH _____
 PaCO₂ _____
 HCO₃⁻ _____
 PaO₂ _____ PvO₂ _____
 SaO₂ _____ % SvO₂ _____ %
 FiO₂ _____ Hb _____

Mode(s) of Ventilatory

Support: _____

Shaded areas represent normal ranges.

Patient's Name _____

Date _____

Time _____

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Hypoxemia

- Abnormally low arterial oxygen tension (PaO₂)
- Frequently associated with hypoxia
 - Which is an inadequate level of tissue oxygenation

Hypoxemia Classifications

Classifications	PaO₂ (rule of thumb)
Normal	80-100 mm Hg
Mild hypoxemia	60-80 mm Hg
Moderate hypoxemia	40-60 mm Hg
Severe hypoxemia	<40 mm Hg

Hypoxia

- Low or inadequate oxygen for cellular metabolism

- There are four main types of hypoxia:
 - Hypoxic
 - Anemic
 - Circulatory
 - Histotoxic

Types of Hypoxia

- Hypoxic hypoxia
 - Inadequate oxygen at tissue cells caused by low arterial oxygen tension (PaO_2)
 - Common Causes
 - Low PaO_2 caused by
 - Hypoventilation
 - High altitude

Types of Hypoxia

- Hypoxic hypoxia
 - Diffusion defects
 - Ventilation-perfusion mismatch
 - Pulmonary shunting

Types of Hypoxia

- Anemic hypoxia
 - PaO₂ is normal, but the oxygen carrying capacity of the hemoglobin is inadequate

Types of Hypoxia

- Anemic hypoxia
 - Common Causes
 - Decreased hemoglobin
 - Anemia
 - Hemorrhage
 - Abnormal hemoglobin
 - Carboxyhemoglobinemia
 - Methemoglobinemia

Types of Hypoxia

- Circulatory hypoxia
 - Stagnant hypoxia or hypoperfusion
 - Blood flow to the tissue cells is inadequate
 - Thus, oxygen is not adequate to meet tissue needs

Types of Hypoxia

- Circulatory hypoxia
 - Common causes
 - Slow or stagnant (pooling) peripheral blood flow
 - Arterial-venous shunts

Types of Hypoxia

- Histotoxic hypoxia
 - Impaired ability of the tissue cells to metabolize oxygen
 - Common cause
 - Cyanide poisoning

Cyanosis

- Blue-gray or purplish discoloration seen on the mucous membranes, fingertips, and toes
 - Blood in these areas contain at least 5 g% of reduced hemoglobin

Cyanosis

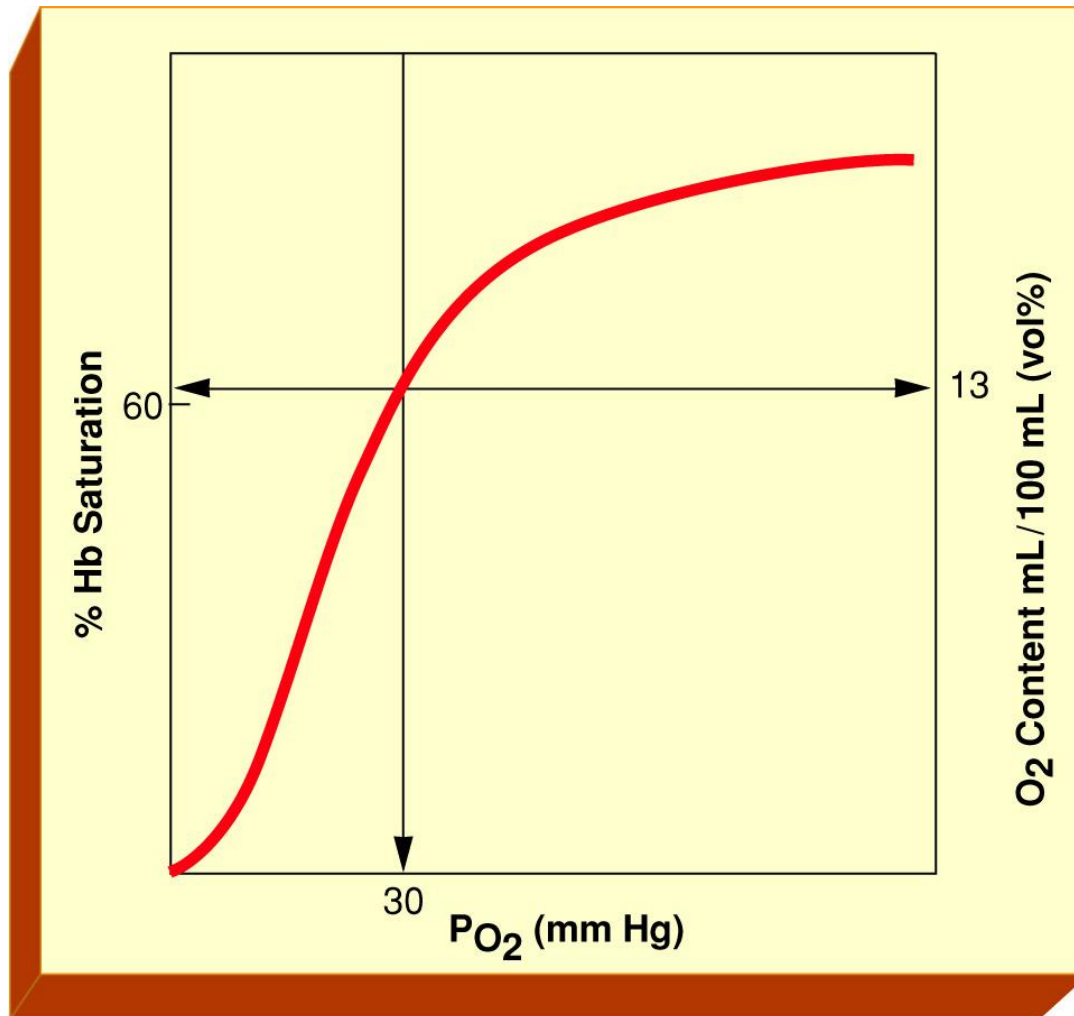


Fig. 6-14. Cyanosis may appear whenever the blood contains at least 5 g% of reduced hemoglobin.

Polycythemia

- An increased level of RBCs
- An adaptive mechanism designed to increase the oxygen-carrying capacity of the blood

Clinical Application 1 Discussion

- How did this case illustrate ...
 - The importance of hemoglobin in the oxygen transport system

Asthma

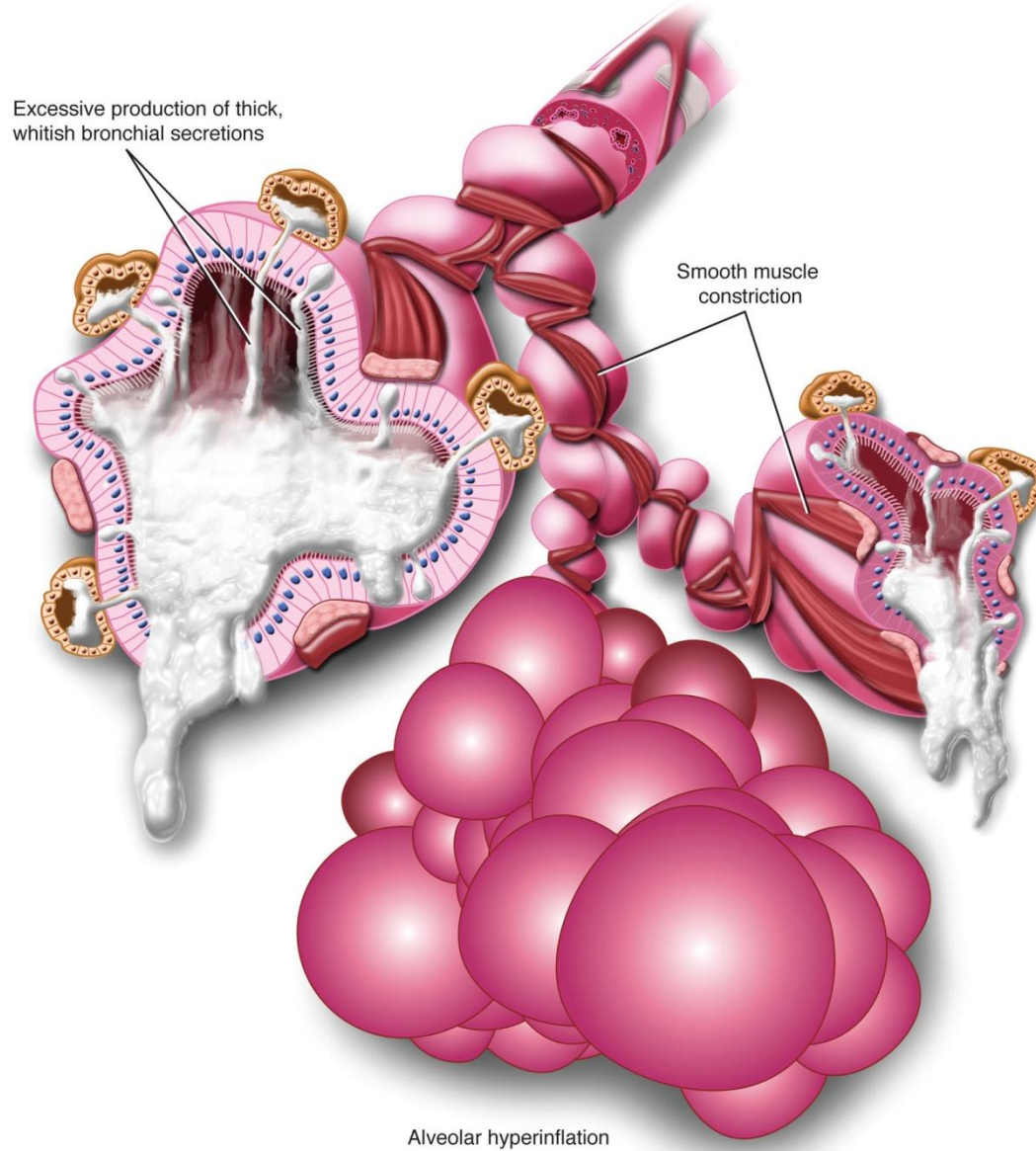


Fig. 6-15. Asthma. Pathology includes (1) bronchial smooth muscle constriction, (2) inflammation and excessive production of thick, whitish bronchial secretions, and (3) alveolar hyperinflation.

Clinical Application 2 Discussion

- How did this case illustrate ...
 - The loading of oxygen on hemoglobin in the lung?
 - The patient's total oxygen delivery (DO_2)?