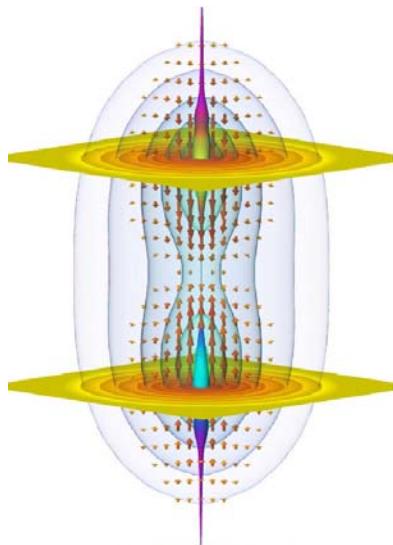




Quantum Fluxes during Chemical Reactions



Jörn Manz
Freie Universität Berlin

Workshop on Quantum Dynamic Imaging
Montreal, Canada
20.10.2009

Agenda

✖ Introduction

- Applications to **electronic fluxes**
- Applications to **nuclear fluxes**
- Applications to concerted **electronic and nuclear fluxes**

Flux densities and fluxes: Nomenclature

- Probability density ρ , dimension 1/volume

$$\rho = |\Psi|^2$$

- Flux density \mathbf{j} , dimension 1/(time x area)

- also called “current density”

$$\mathbf{j} = -\frac{i\hbar}{2m} (\Psi^* \nabla \Psi - \Psi \nabla \Psi^*)$$

Example:
one particle in 3D

- Flux F , dimension 1/time

- also called “current”

$$F(t; A_{obs}) = - \int_{A_{obs}} \mathbf{j}(t) \cdot d\mathbf{A}$$

- Yield Y , dimension 1
- also called “probability”

$$Y(t; A_{obs}) = \int_{V_{obs}} \rho(t) dV - \int_{V_{obs}} \rho(t=0) dV$$

Continuity equations and Gauss's theorem

■ Continuity equation

- for the electron
- for the k -th nucleus

$$\begin{aligned}\dot{\rho}_{el}(\mathbf{r}, t) &= -\nabla_{\mathbf{r}} \cdot \mathbf{j}_{el}(\mathbf{r}, t) \\ \dot{\rho}_{nu,k}(\mathbf{R}, t) &= -\nabla_{\mathbf{R}} \cdot \mathbf{j}_{nu,k}(\mathbf{R}, t)\end{aligned}$$

■ Gauss's theorem

$$F(t; A_{obs}) = - \int_{A_{obs}} \mathbf{j}(t) \cdot d\mathbf{A} = - \int_{V_{obs}} \nabla \cdot \mathbf{j}(t) dV = \int_{V_{obs}} \dot{\rho}(t) dV = \dot{Y}(t; A_{obs})$$

- key equation

$$F(t; A_{obs}) = \int_{V_{obs}} \dot{\rho}(t) dV$$

Heaviside function and flux operator

- Heaviside function $h(V_{obs})$
 - flux = time derivative of the mean value of $h(V_{obs})$

$$F(t; A_{obs}) = \int_{V_{obs}} \dot{\rho}(t) dV = \int_V \dot{\rho}(t) h(V_{obs}) dV = \frac{d}{dt} \langle h(V_{obs}) \rangle = \left\langle \frac{i}{\hbar} [H, h(V_{obs})] \right\rangle$$

- Nuclear flux operator by W.H. Miller

- in Schrödinger picture
 - in Heisenberg picture

$$\frac{i}{\hbar} [H, h(V_{obs})]$$
$$\frac{d}{dt} h(V_{obs})$$

W.H. Miller, J. Chem. Phys. 61, 1823 (1974)
W.H. Miller, Acc. Chem. Res. 26, 174 (1993)
W.H. Miller, J. Phys. Chem. A 102, 793 (2008)

Nuclear flux operator: Applications

- Thermal reaction rate constant $k(T)$
 - $\text{H}_2 + \text{OH} \rightarrow \text{H} + \text{H}_2\text{O}$

Quantum mechanical calculations of the rate constant for the $\text{H}_2 + \text{OH} \rightarrow \text{H} + \text{H}_2\text{O}$ reaction: Full-dimensional results and comparison to reduced dimensionality models

Uwe Manthe,^{a)} Tamar Seideman,^{b)} and William H. Miller

Department of Chemistry, University of California, and Chemical Sciences Division, Lawrence Berkeley Laboratory, Berkeley, California 94720

(Received 11 May 1994; accepted 10 June 1994)

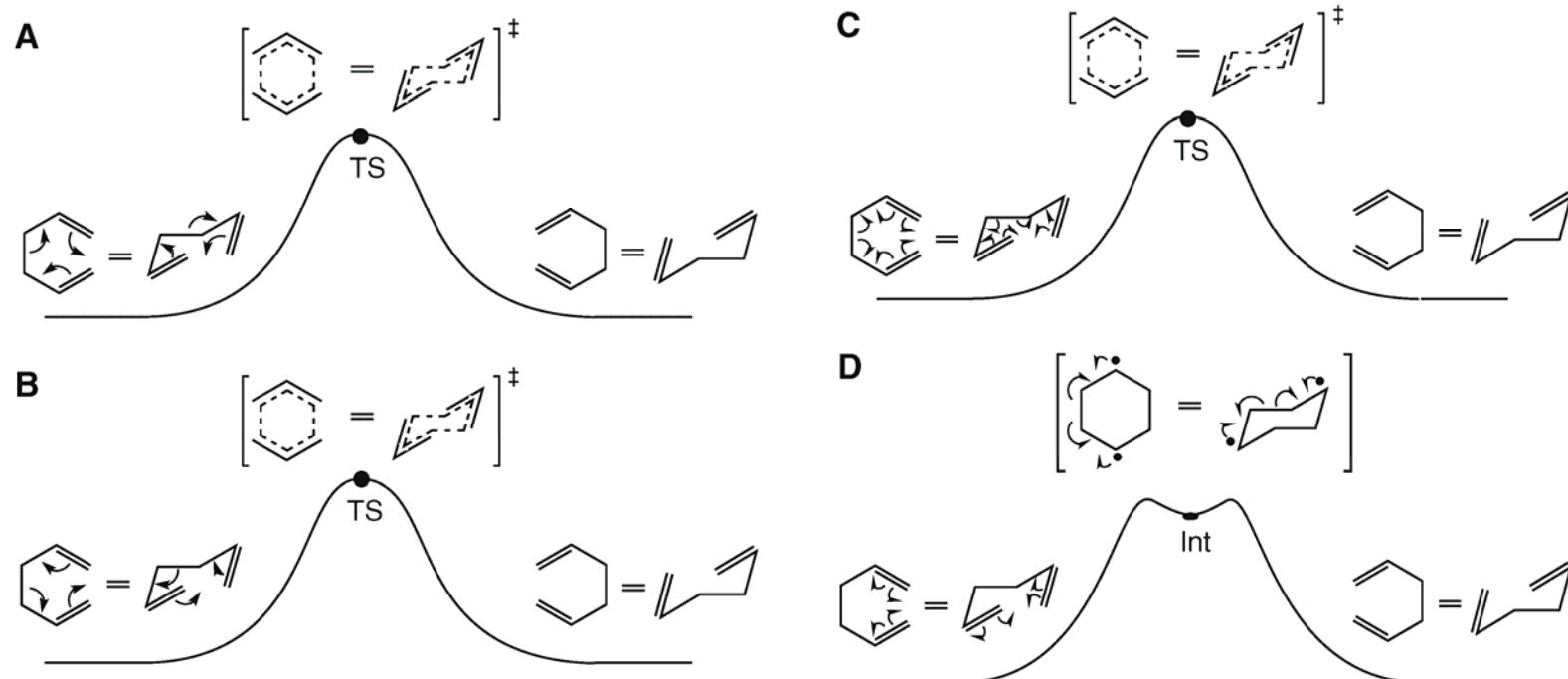
- $\text{H} + \text{CH}_4 \rightarrow \text{H}_2 + \text{CH}_3$

First-Principles Theory for the $\text{H} + \text{CH}_4 \rightarrow \text{H}_2 + \text{CH}_3$ Reaction

Tao Wu,^{1*} Hans-Joachim Werner,² Uwe Manthe^{3,1†}

Motivation: Nuclear and electron fluxes

- Cope rearrangement in organic chemistry
 - 1,5-hexadiene or derivatives



Dilemma of the BO approximation

■ Born-Oppenheimer (BO) approximation

$$\Psi_{BO}(\mathbf{q}, \mathbf{Q}, t) = \Psi_{BO,nu}(\mathbf{Q}, t) \Psi_{BO,el}(\mathbf{q}; \mathbf{Q})$$

- excellent for densities and nuclear flux densities
→ powerful for quantum chemistry and nuclear dynamics
- BUT: **electronic flux densities** are **zero** because of real electronic wavefunctions in non-degenerate states

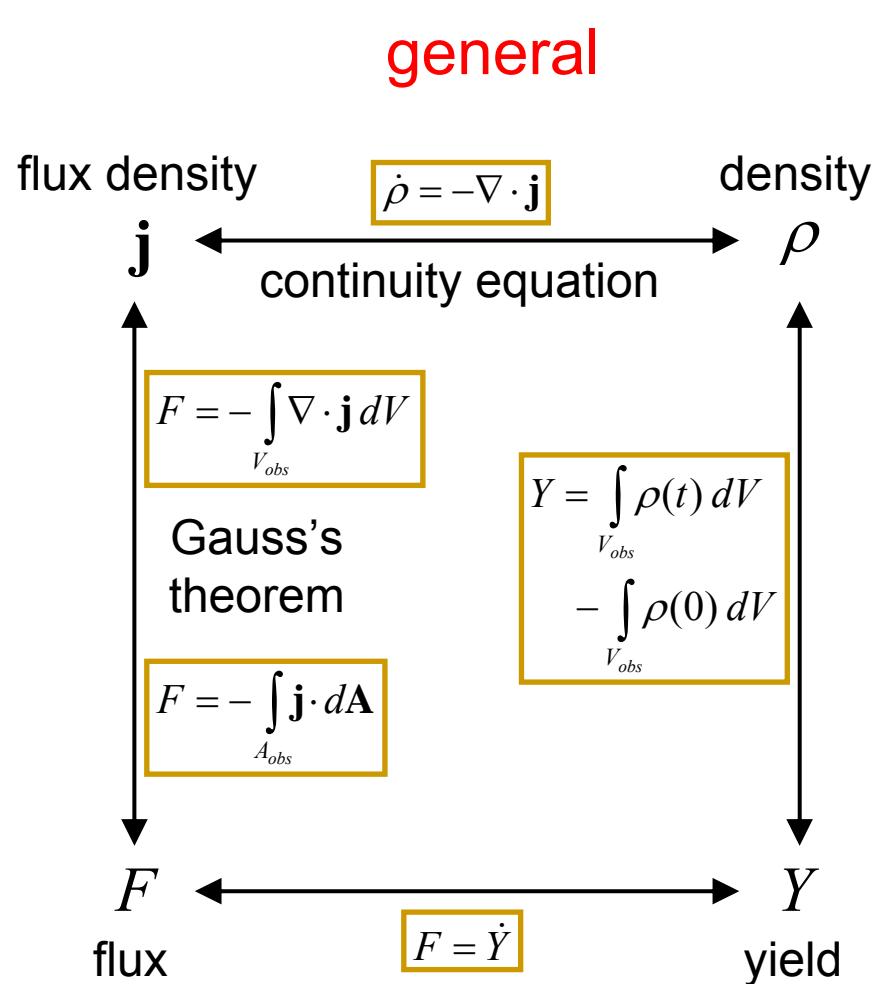
$$\mathbf{j}_{BO,el}(\mathbf{r}, t) = \mathbf{0}$$

■ Continuity equation for the electron is **not valid in BO**

$$\text{in general: } \dot{\rho}_{BO,el}(\mathbf{r}, t) \neq 0$$

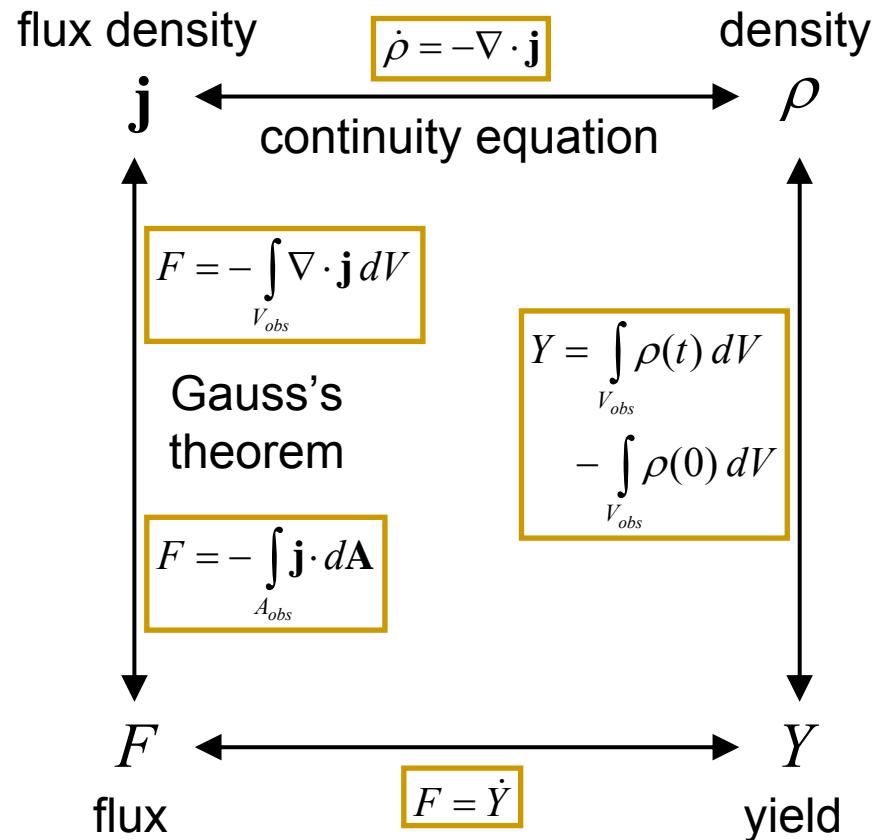
$$\dot{\rho}_{BO,el}(\mathbf{r}, t) \neq -\nabla_{\mathbf{r}} \cdot \mathbf{j}_{BO,el}(\mathbf{r}, t)$$

Schematic overview

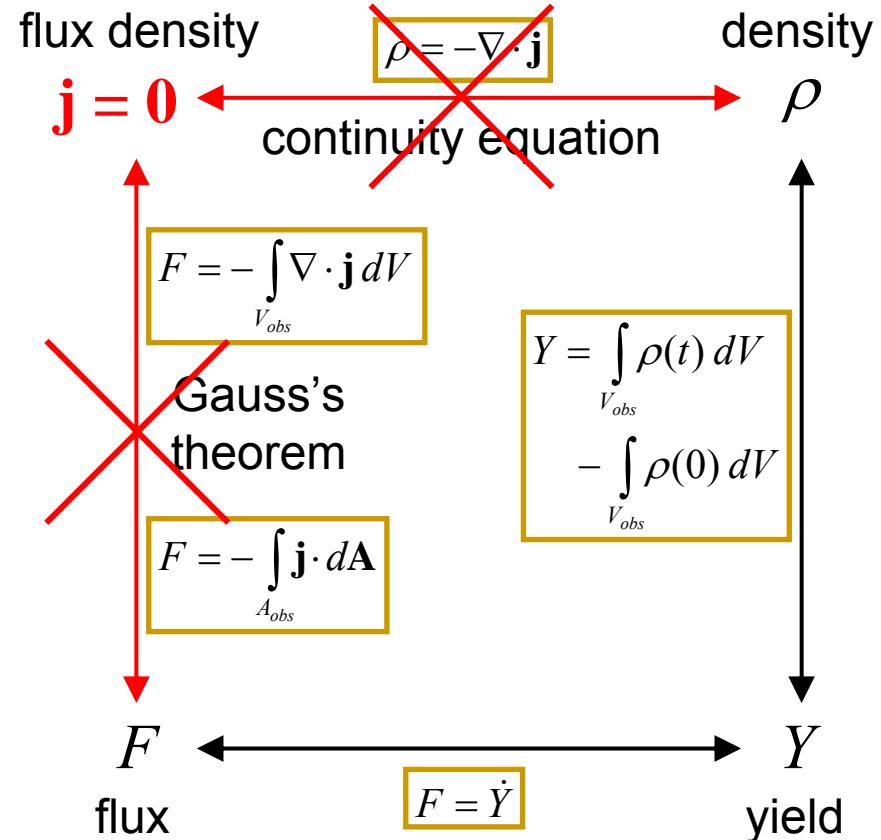


Schematic overview

general



BO for the electron



Agenda

- ✓ Introduction

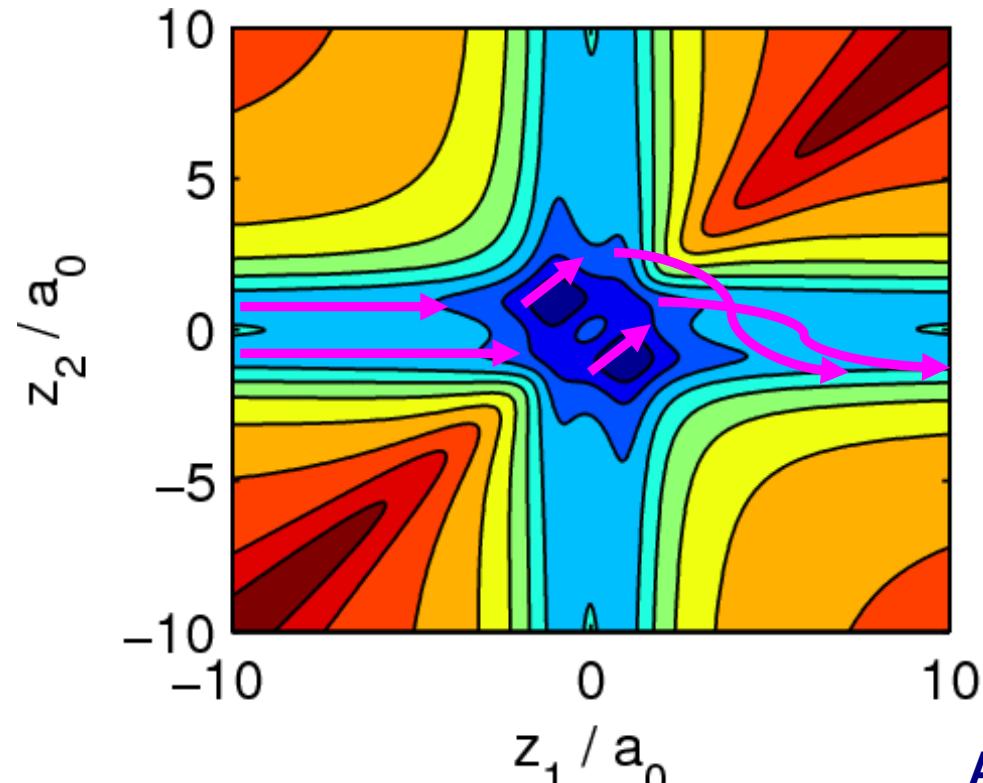
- ✗ Applications to **electronic fluxes**

- Electron impact dynamics: $e^- + H_2^+$ (2D)
 - Electronic ring current: Mg-porphyrin, AlCl, atoms (multi-D)

- Applications to **nuclear fluxes**

- Applications to concerted **electronic and nuclear fluxes**

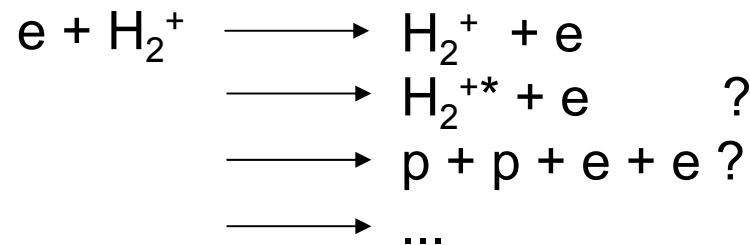
Attosecond electron impact dynamics: quantum model simulation for $e^- + H_2^+$



Axel Schild
with kind support by Ingo Barth

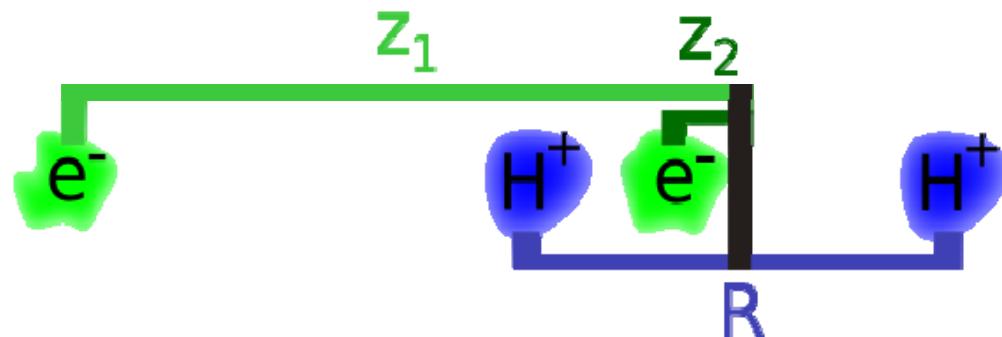
Motivation

- New method of investigation of attosecond electron dynamics:
Zewail proposed compression of ~30 keV electron pulse to 15 as with
the help of a moving intensity grating
*P. Baum, A.H. Zewail, Chem. Phys 2009, special issue Attosecond Molecular
Dynamics (Eds. Bandrauk, Manz, Vrakking)*
- Do high-energy electrons change the target?
- This work: Quantum model scattering of 0.5 – 1.5 keV electron off H_2^+



The model

one-dimensional H₂-molecule:



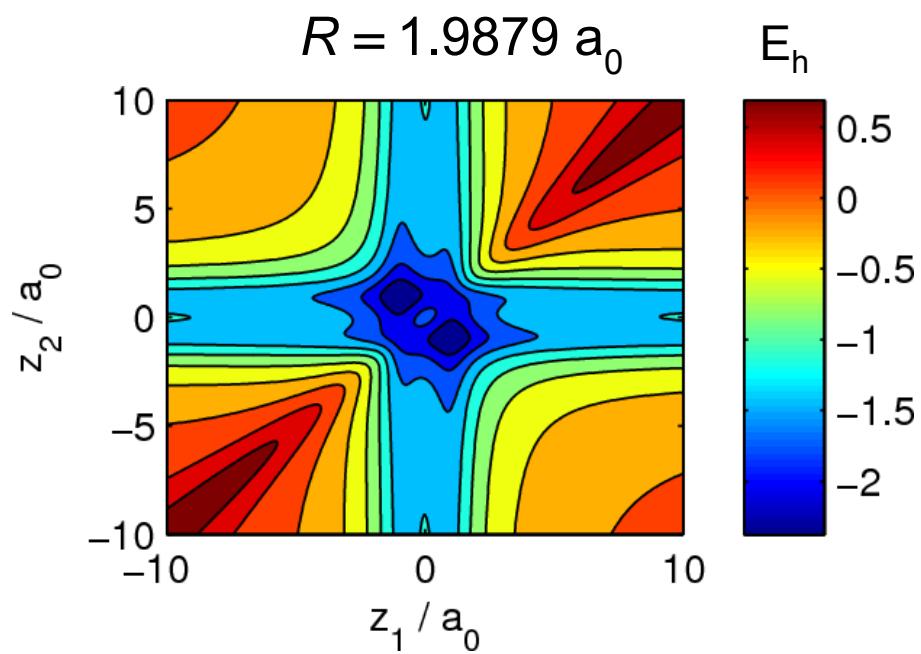
- z_1, z_2 electron distance from c.o.m.
- internuclear distance R kept fixed

$$i \frac{d}{dt} \Psi(z_1, z_2, t) = \left[-\frac{1}{2} \frac{\partial^2}{\partial z_1^2} - \frac{1}{2} \frac{\partial^2}{\partial z_2^2} + V(z_1, z_2) \right] \Psi(z_1, z_2, t)$$

- Solved with WavePacket program / Split-Operator method
(B. Schmidt, U. Lorenz, <http://wavepacket.sourceforge.net>)

The model

$$V_{H_2} = \sum_{j=1}^2 \left[-\frac{1}{\sqrt{(z_j + R/2)^2 + c}} - \frac{1}{\sqrt{(z_j - R/2)^2 + c}} \right] + \frac{1}{\sqrt{(z_1 - z_2)^2 + d}} + \frac{1}{R}$$



$R = 1.9879 a_0$

E_h

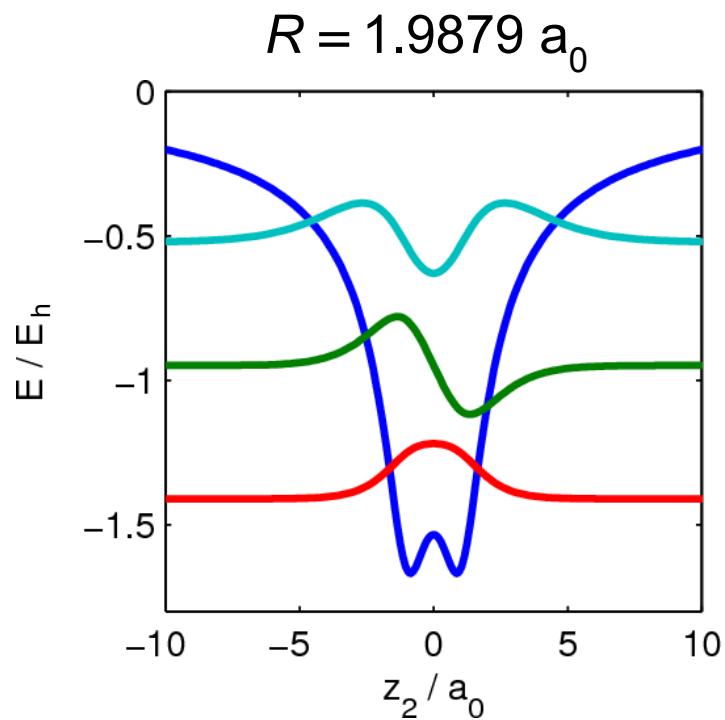
to reproduce electronic
excitation energies of H_2 :
 $c = 0.7 a_0^2$, $d = 1.2375 a_0^2$

adapted from

A.D. Bandrauk, S. Chelkowski,
S. Kawai, H. Lu,
PRL 101, 153901, 2008

The model

$$V_{H_2^+} = -\frac{1}{\sqrt{(z + R/2)^2 + c}} - \frac{1}{\sqrt{(z - R/2)^2 + c}} + \frac{1}{R}$$



$$\varepsilon_{01} = 0.461 E_h$$

compare:

$$\varepsilon_{01} = 0.435 E_h$$

(J.M. Peek, J. Chem. Phys.
43, 3004, 1965)

Initial state

Product of Gaussian in z_1 -direction
with ground state of H_2^+ in z_2 -direction:
distinguishable electrons

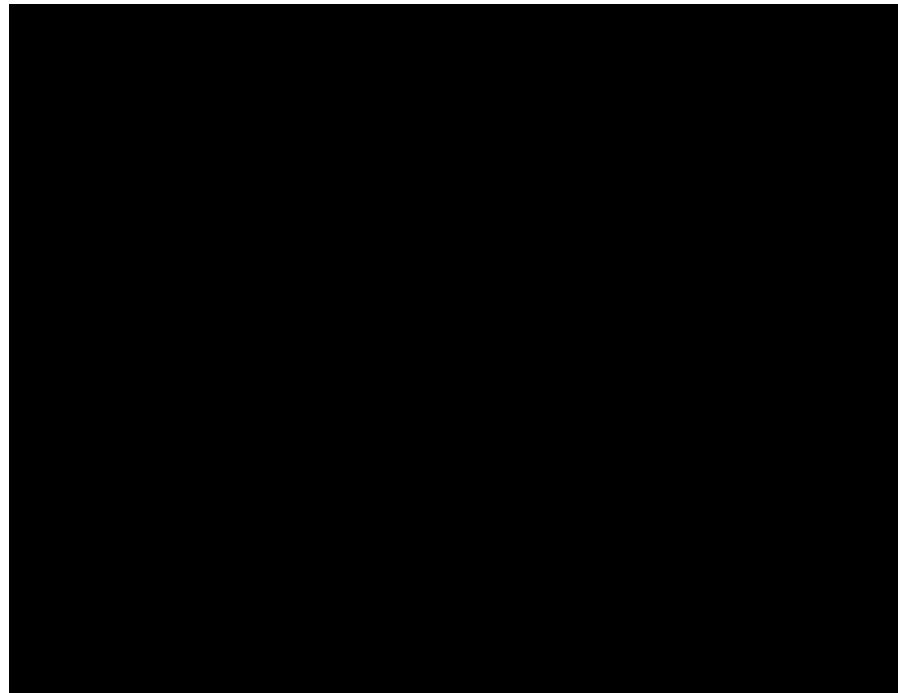
$$\Psi(z_1, z_2, t = 0) = \psi(z_1)\phi_0(z_2)$$

$$\psi = N \exp \left(-\alpha_0(z_1 - z_{1,0})^2 + \frac{i}{\hbar} p_0(z_1 - z_{1,0}) + \frac{i}{\hbar} \gamma_0 \right)$$

Kinetic energy: ~ 544 eV

D.J. Tannor, Introduction to Quantum Mechanics –
A Time Dependent Perspective
(University Science Books: USA 2007)

Electron dynamics



Time-dependent scattering theory

$$H = H^{ela} + V^{inela}$$

$$V_{H_2}^{ela} = \sum_{j=1}^2 \left[-\frac{1}{\sqrt{(z_j + R/2)^2 + c}} - \frac{1}{\sqrt{(z_j - R/2)^2 + c}} \right] + \frac{1}{R}$$

$$V^{inela} = \frac{1}{\sqrt{(z_1 - z_2)^2 + d}} \quad \text{causes electron correlation}$$

Time-dependent scattering theory

for reference: elastic scattering

Initial state

$$\Psi_j^{ela}(z_1, z_2, t = 0) = \psi_j^{ela}(z_1)\phi_j(z_2)$$

TDSE

$$i\frac{d}{dt}\Psi_j^{ela}(z_1, z_2, t) = H^{ela}\Psi_j^{ela}(z_1, z_2, t)$$

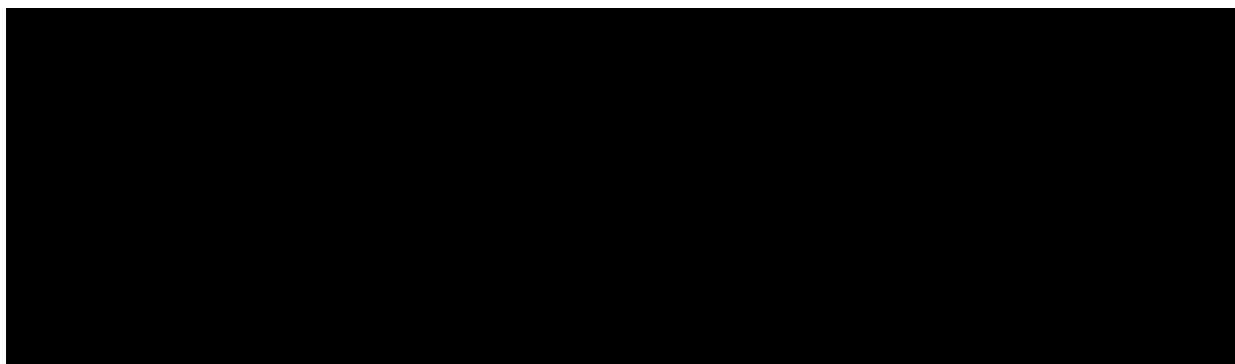
Expansion in terms of elastic wavefunctions:

$$\Psi(z_1, z_2, t) = \sum_{j=0}^{\infty} c_j^{ela}(t)\Psi_j^{ela}(z_1, z_2, t) \text{ for } t \rightarrow \infty$$

$$c_j^{ela}(t) = \int dz_1 dz_2 (\Psi_j^{ela}(z_1, z_2, t))^* \Psi(z_1, z_2, t)$$

Time evolution

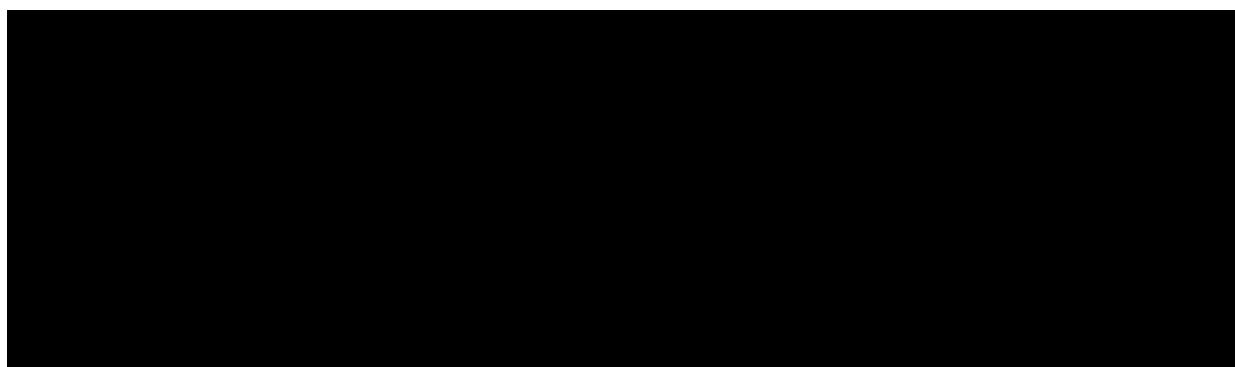
$$R = 1.9879 \text{ } a_0$$



- Dispersion, but little trace of the scattering event
→ major part is elastic scattering

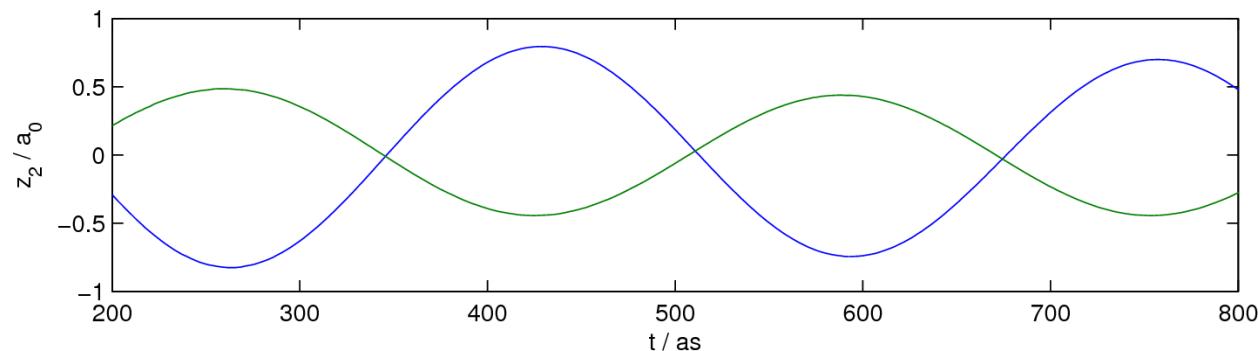
Time evolution

$$\Psi(z_1, z_2, t) - c_0^{ela}(t)\Psi_0^{ela}(z_1, z_2, t)$$



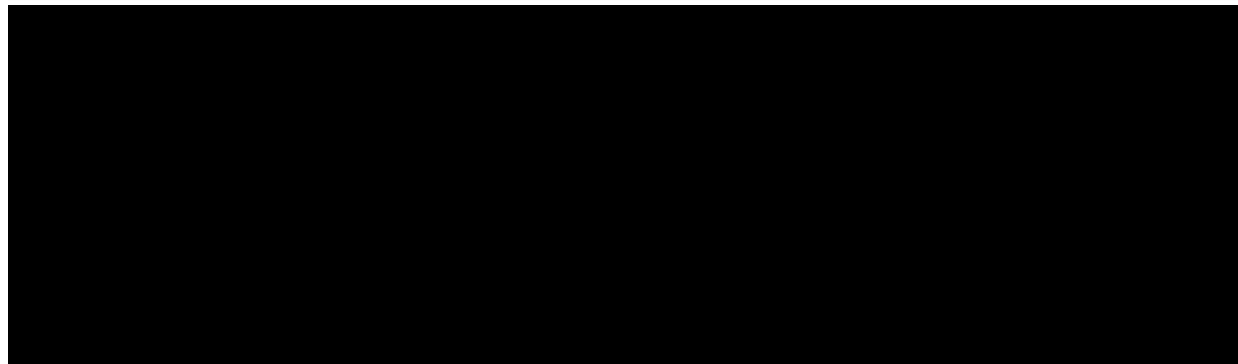
→ Two oscillating parts, head and tail

Period $T = 329$ as, $h\nu = 0.462 E_h$ compare $\varepsilon_{01} = 0.461 E_h$



Time evolution

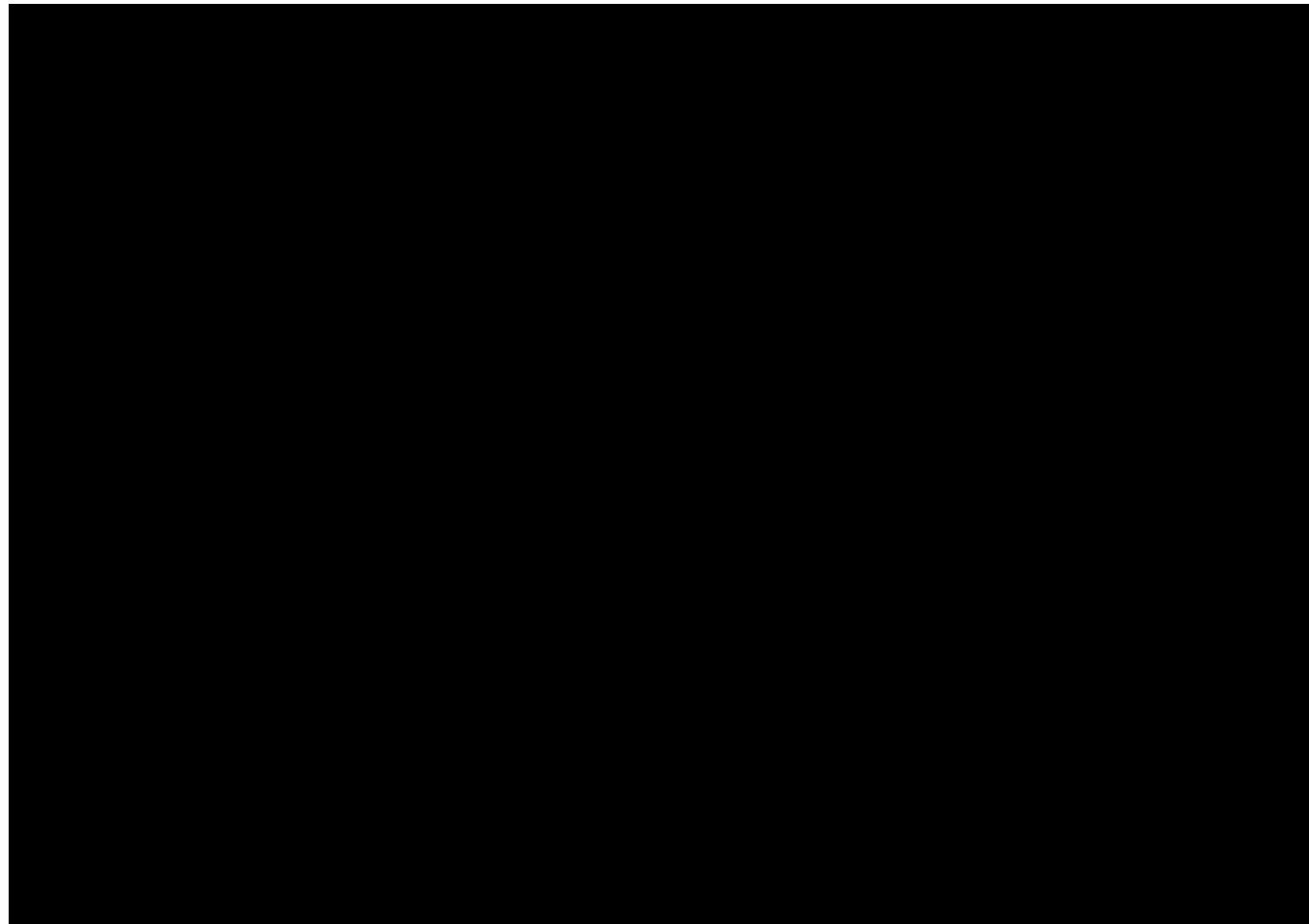
$$\Psi(z_1, z_2, t) - \sum_{j=0}^1 c_j^{ela}(t) \Psi_j^{ela}(z_1, z_2, t)$$



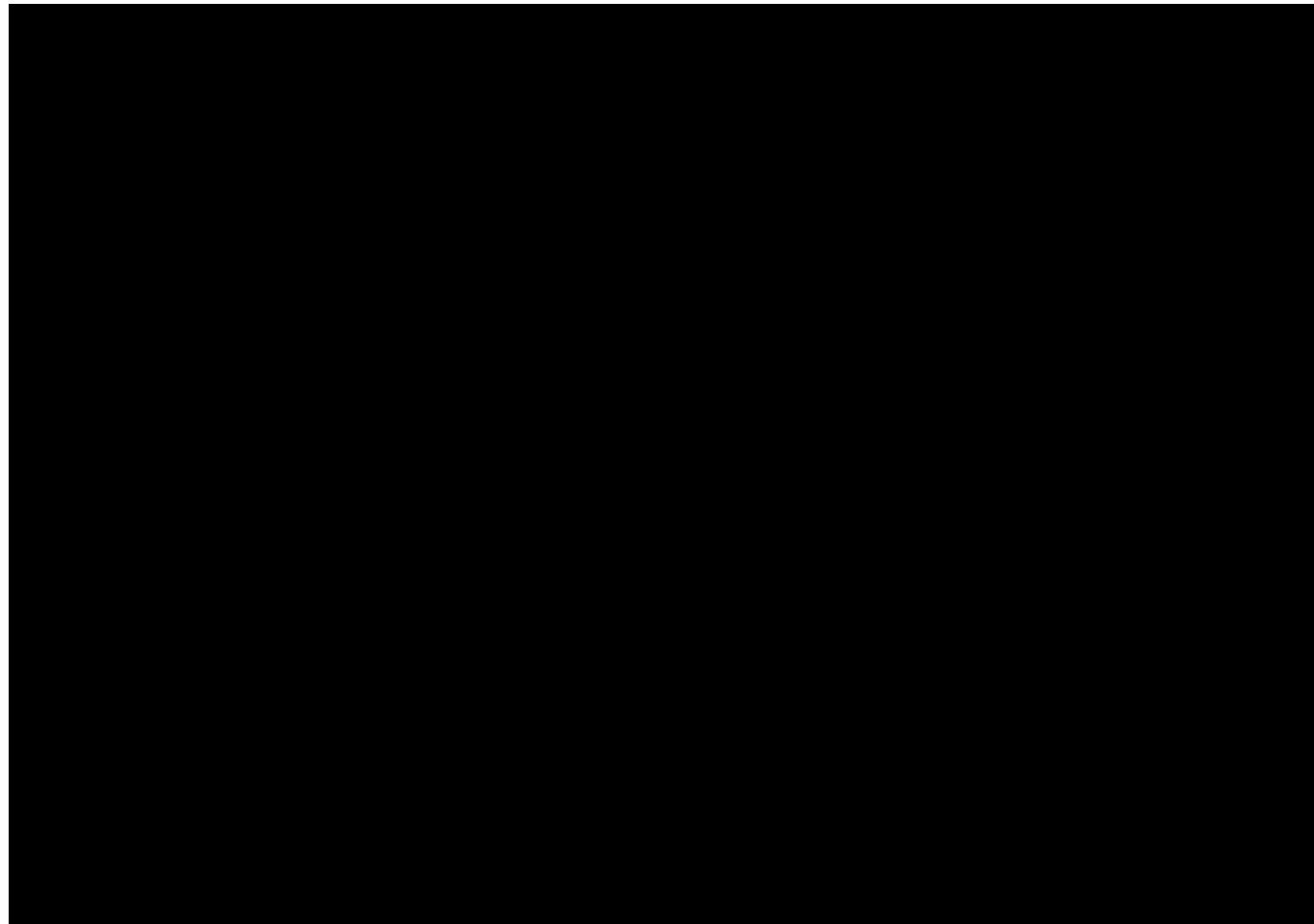
→ no oscillation

removing further contributions $c_j^{ela}(t) \Psi_j^{ela}(z_1, z_2, t)$
for $j > 1$ has no visible effect on the density

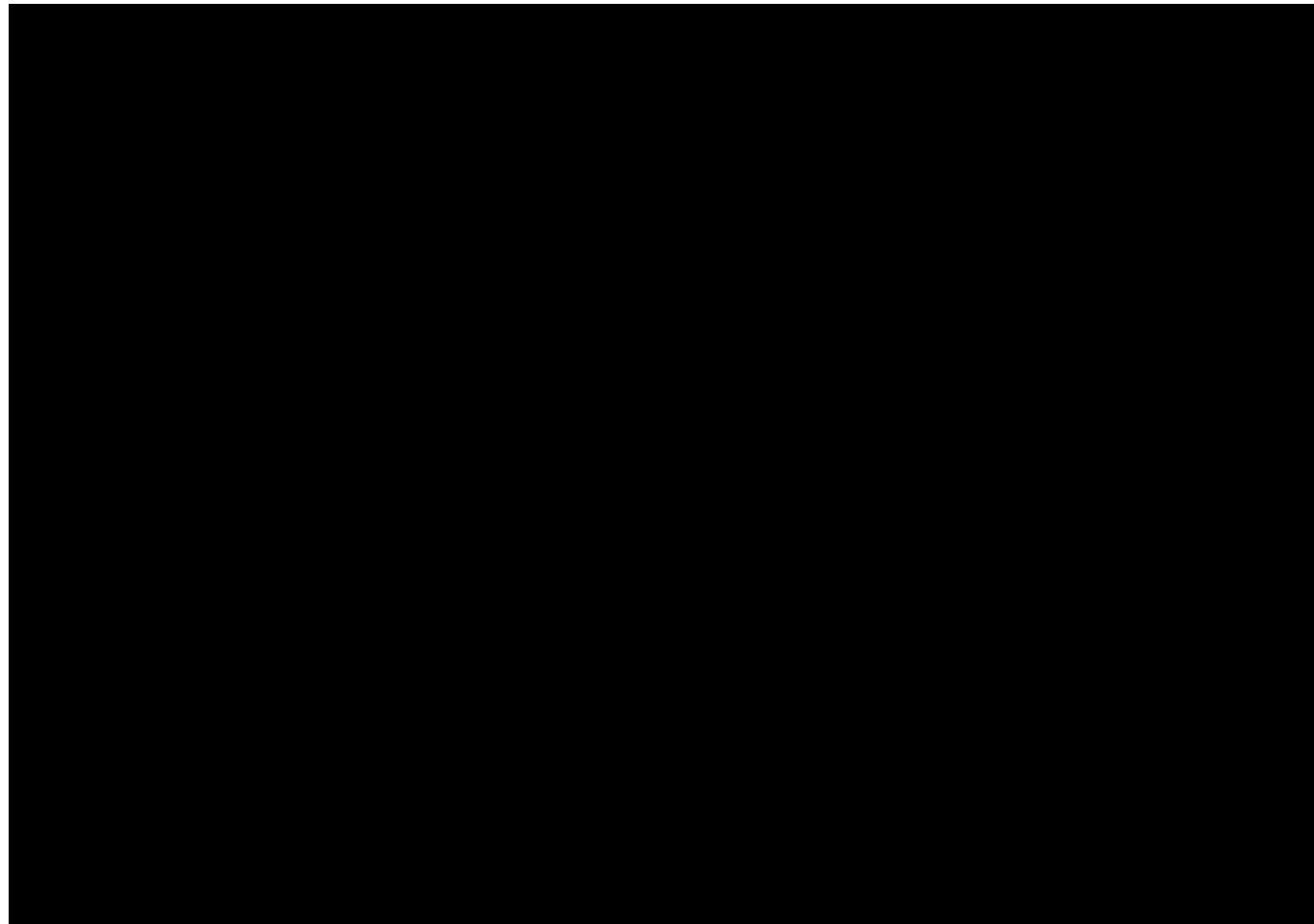
Electronic flux density



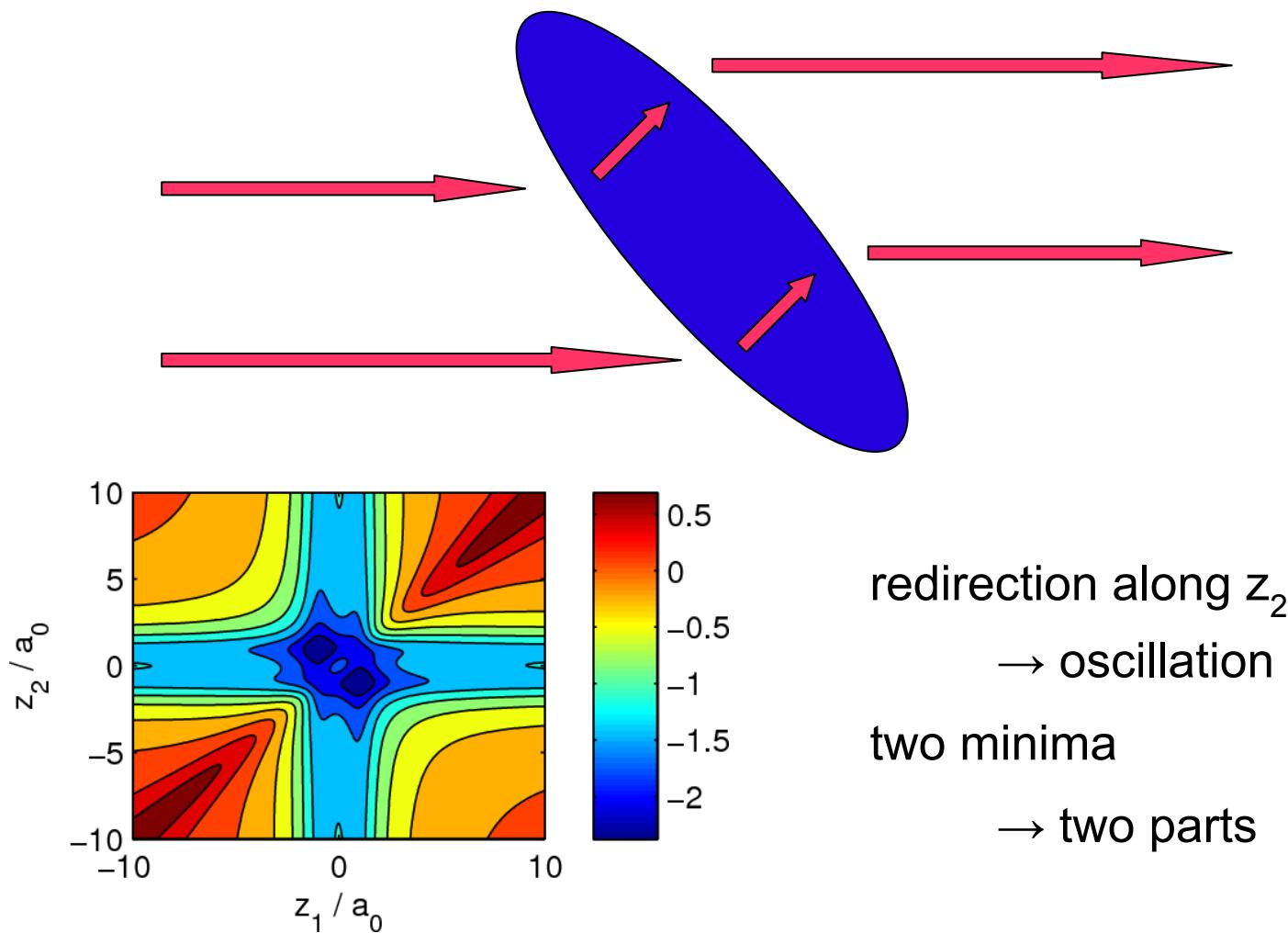
Electronic flux density



Electronic flux density



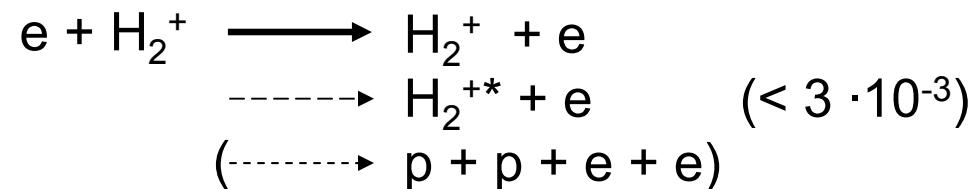
Mechanism



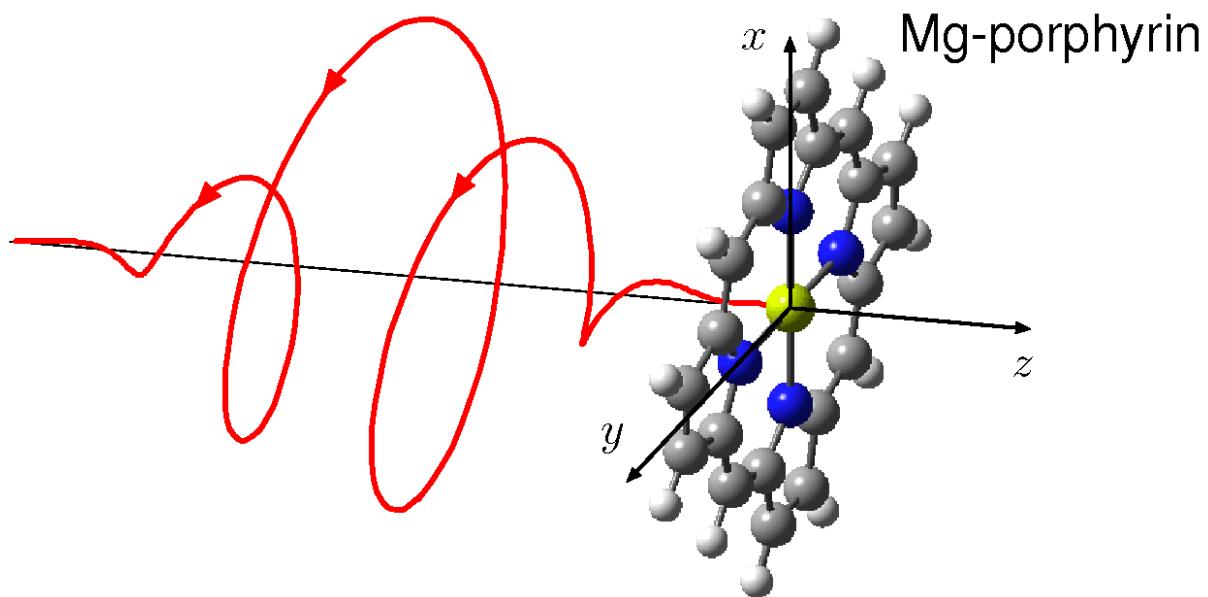
Conclusion

Take home message:

- 1) Analysis of mechanisms by flux density
- 2) Experiment by Baum & Zewail is promising!



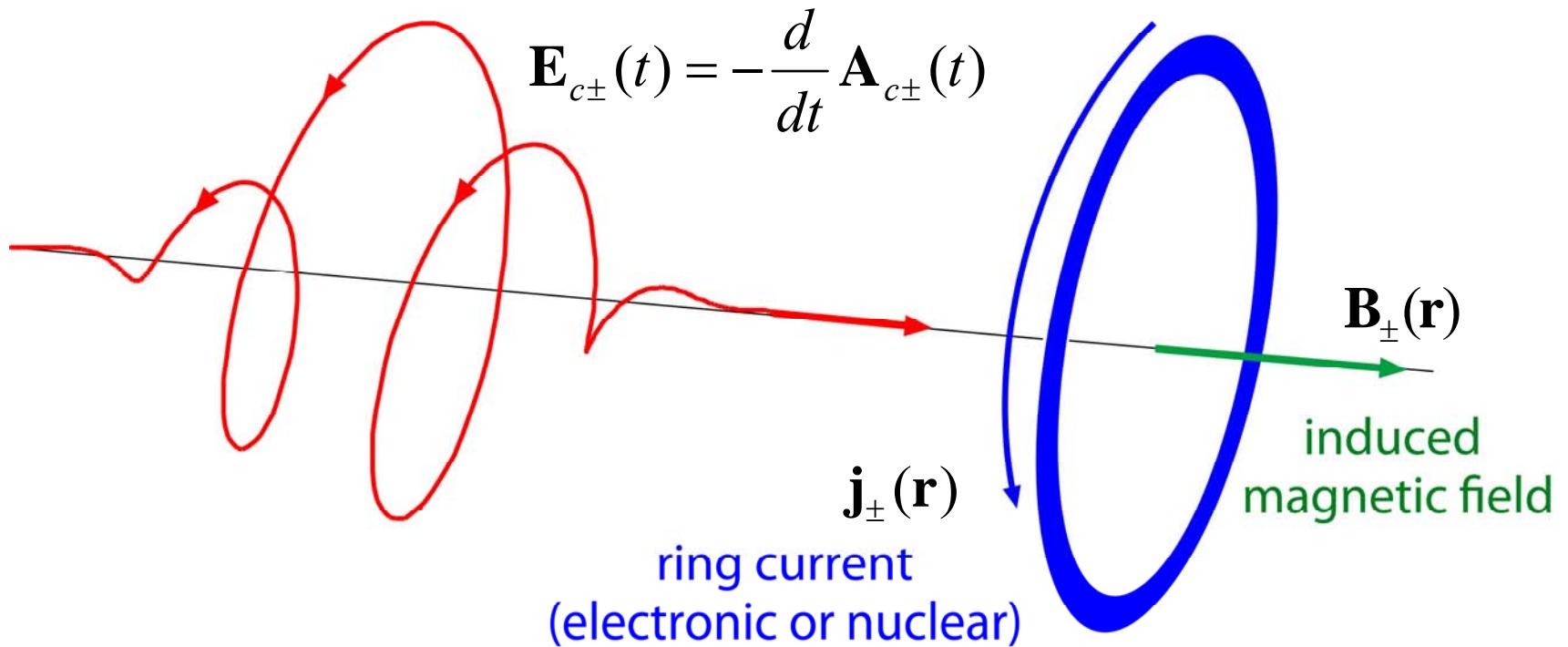
Control of electronic ring currents by circularly polarized laser pulses



- I. Barth, J. Manz, Y. Shigeta, K. Yagi, *J. Am. Chem. Soc.* 128, 7043 (2006)
- I. Barth, J. Manz, *Angew. Chem. Int. Ed.* 45, 2962 (2006)
- I. Barth, J. Manz, *Phys. Rev. A* 75, 012510 (2007)
- I. Barth, J. Manz, L. Serrano-Andrés, *Chem. Phys.* 347, 263 (2008)
- I. Barth, L. Serrano-Andrés, T. Seideman, *Chem. Phys.* 347, 263 (2008)
- I. Barth, PhD thesis, FU Berlin (2009)

Concept

circularly polarized laser pulse



related: inverse Faraday effect

Model assumptions for ring currents

- Fixed orientation of molecule

H. Stapelfeldt, T. Seideman, Rev. Mod. Phys. 75, 543 (2003)

M. Leibscher, I. S. Averbukh, H. Rabitz, Phys. Rev. A 69, 013402 (2004)

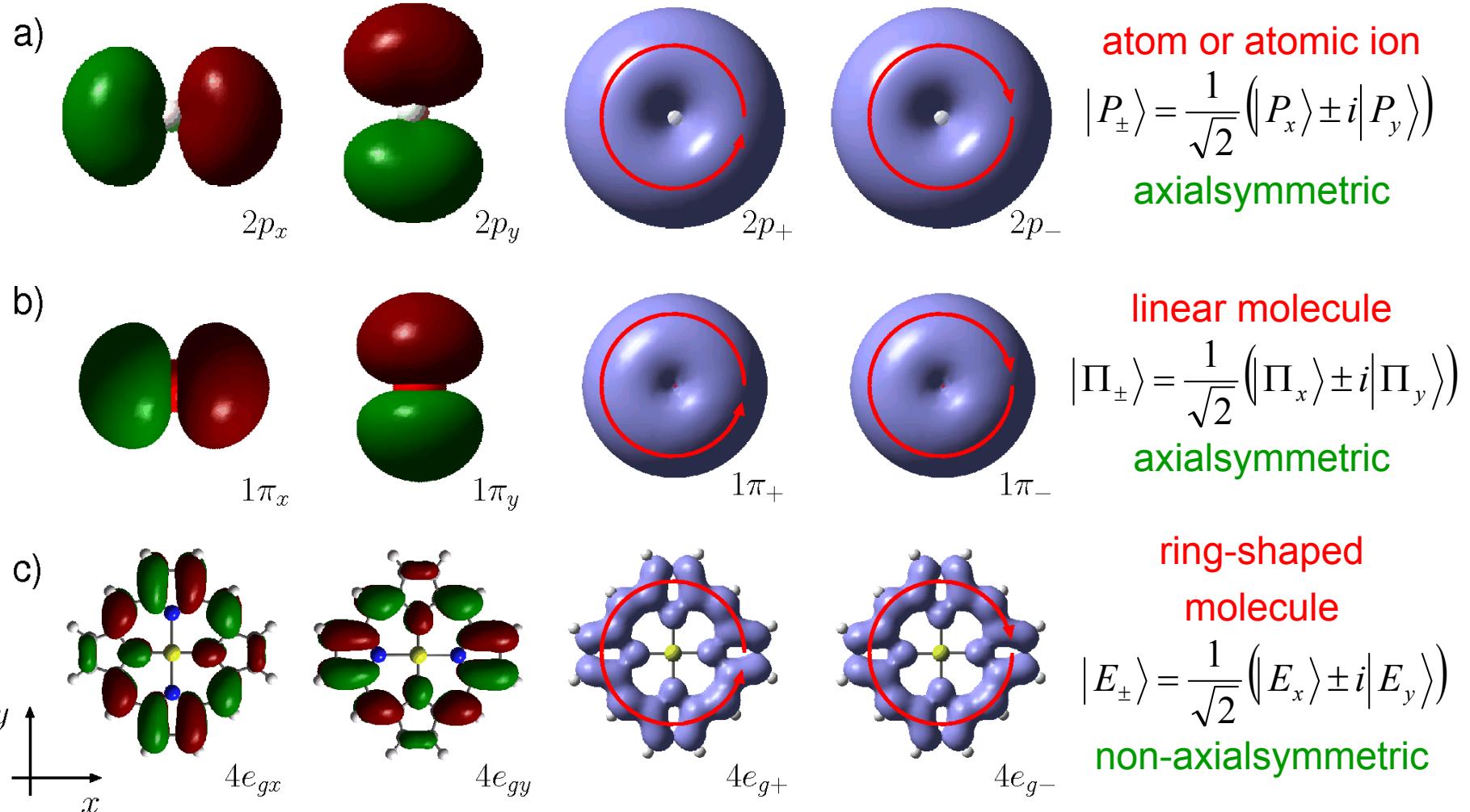
E. Hamilton, T. Seideman, T. Ejdrup, M. D. Poulsen, C. Z. Bisgaard, S. S. Viftrup, H. Stapelfeldt, Phys. Rev. A 72, 043402 (2005)

I. Barth, L. Serrano-Andrés, T. Seideman, Chem. Phys. 347, 263 (2008)

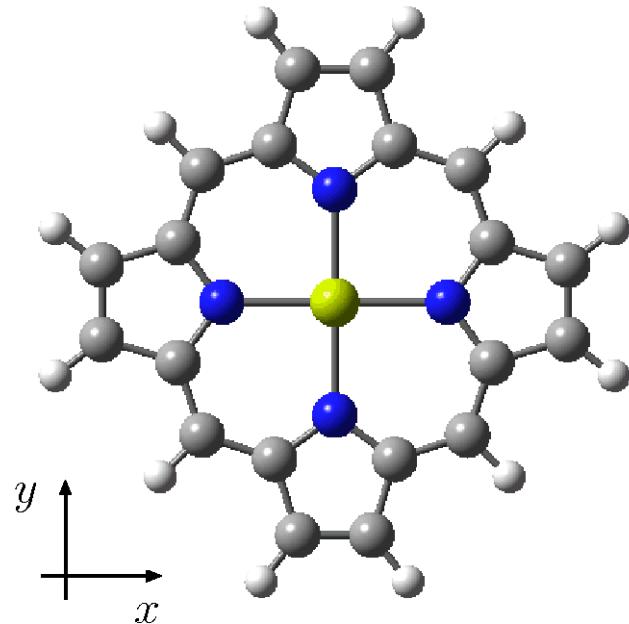
- Transition from ground state

- No spin-orbit interaction, spin conservation

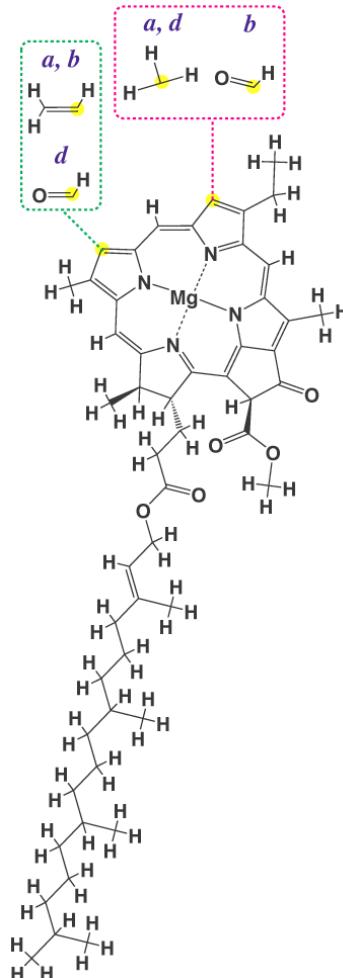
Electronic degenerate orbitals/states



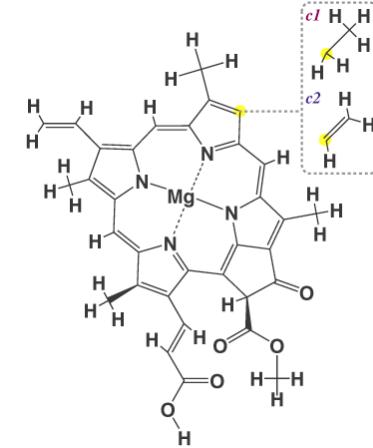
Model: Magnesium porphyrin (MgP)



- Ground state: X^1A_{1g}
- Symmetry: D_{4h}



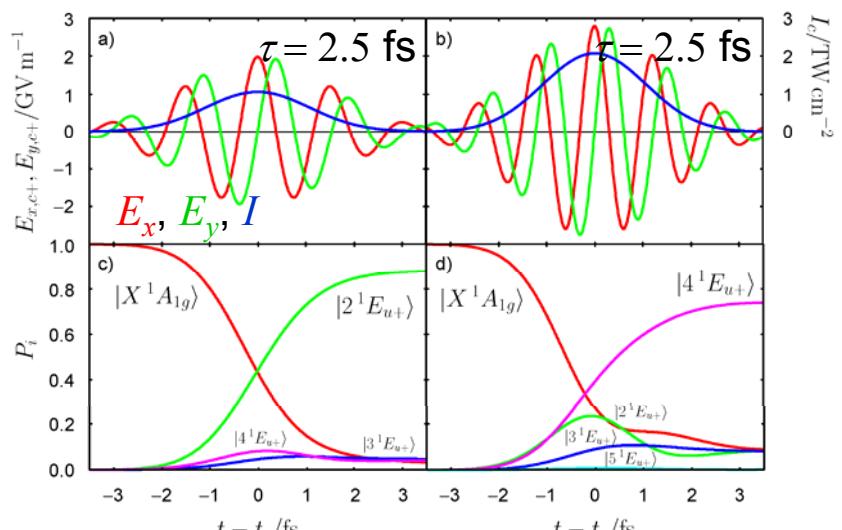
Chlorophyll c1, c2



Chlorophyll a, b, d

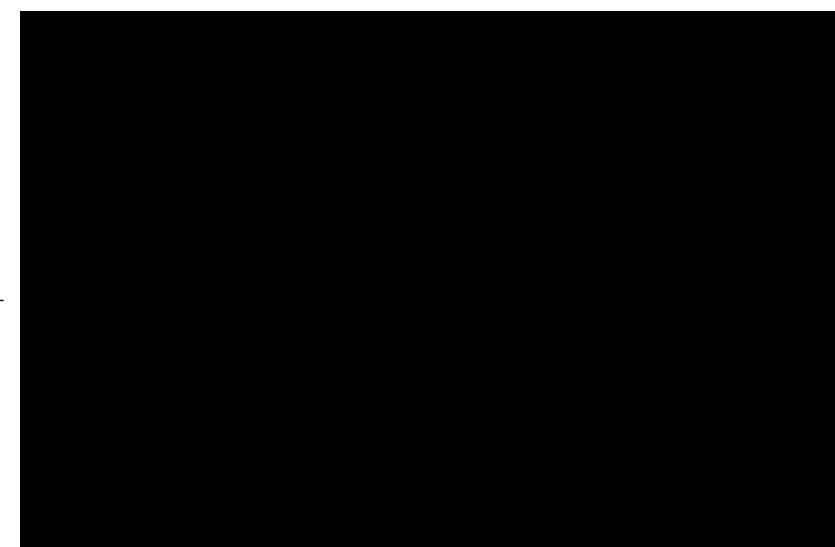
Electronic ring currents in MgP

- Selective population transfer $|X^1A_{1g}\rangle \rightarrow |n^1E_{u+}\rangle$ by means of a right circ. pol. optimized π laser pulse with \cos^{20} envelope



$|2^1E_{u+}\rangle$

$|4^1E_{u+}\rangle$



Difference of the electronic probability density

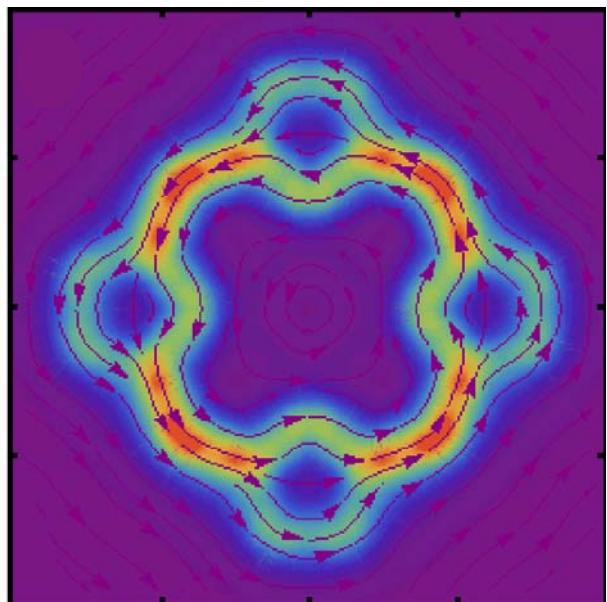
$\Delta\rho(\mathbf{r}, t) > 0$ (red)

$\Delta\rho(\mathbf{r}, t) < 0$ (green)

-
- I. Barth, J. Manz, Y. Shigeta, K. Yagi, J. Am. Chem. Soc. 128, 7043 (2006)
 - I. Barth, C. Lasser, J. Phys. B (2009), in press
 - I. Barth, PhD thesis, FU Berlin (2009)

Electronic ring currents in MgP

laser pulse

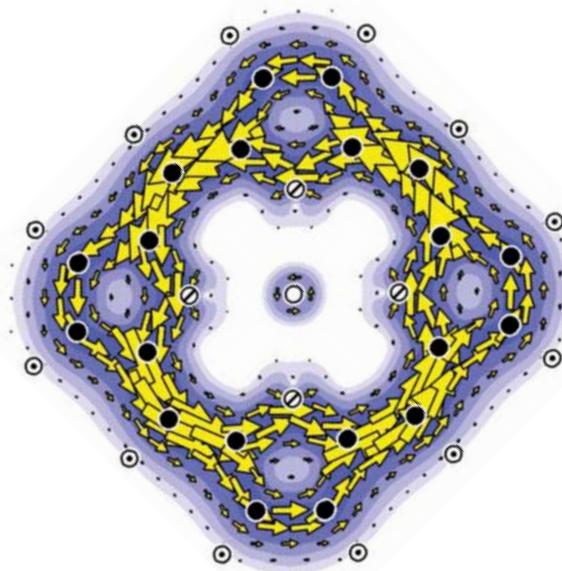


$$\langle r \rangle = 6.32 a_0$$

$$I = 84.5 \mu\text{A}$$

$$B_{ind} = 0.159 \text{ T}$$

permanent magnetic field



$$\langle r \rangle = 6.85 a_0$$

$$I = 84.5 \mu\text{A} \quad \text{if } B = 8048 \text{ T}$$

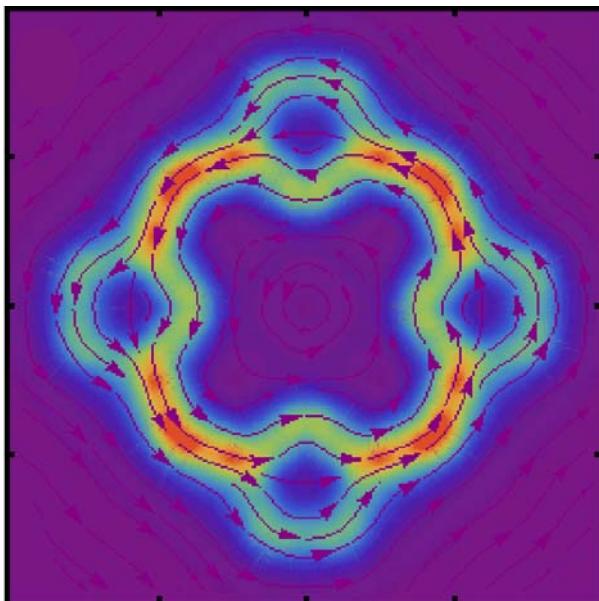
Present technology:
< 100 T (permanent), 34000 T (10 ps)
(Rossendorf / Dresden, Rutherford Appleton)

I. Barth, J. Manz, Y. Shigeta, K. Yagi,
J. Am. Chem. Soc. 128, 7043 (2006)

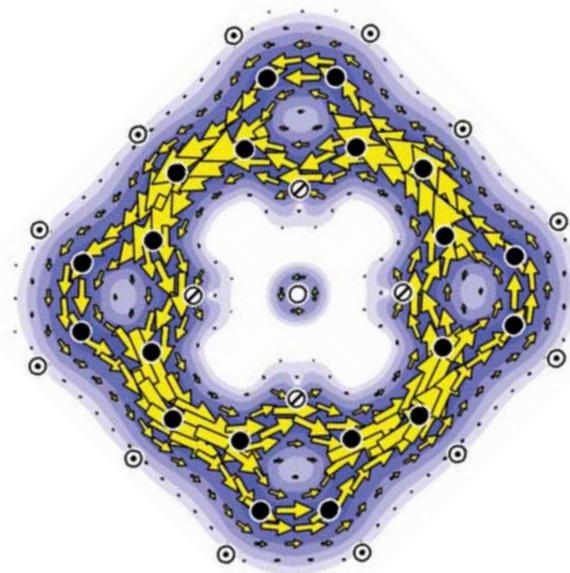
E. Steiner et al, Org. Biomol. Chem. 3, 4053 (2005)
J. Jusélius et al, J. Org. Chem. 65, 5233 (2000)

Electronic ring currents in MgP

laser pulse



permanent magnetic field



strong ring current
active control

$$\langle r \rangle = 6.85 a_0$$

$$I = 84.5 \mu\text{A} \quad \text{if } B = 8048 \text{ T}$$

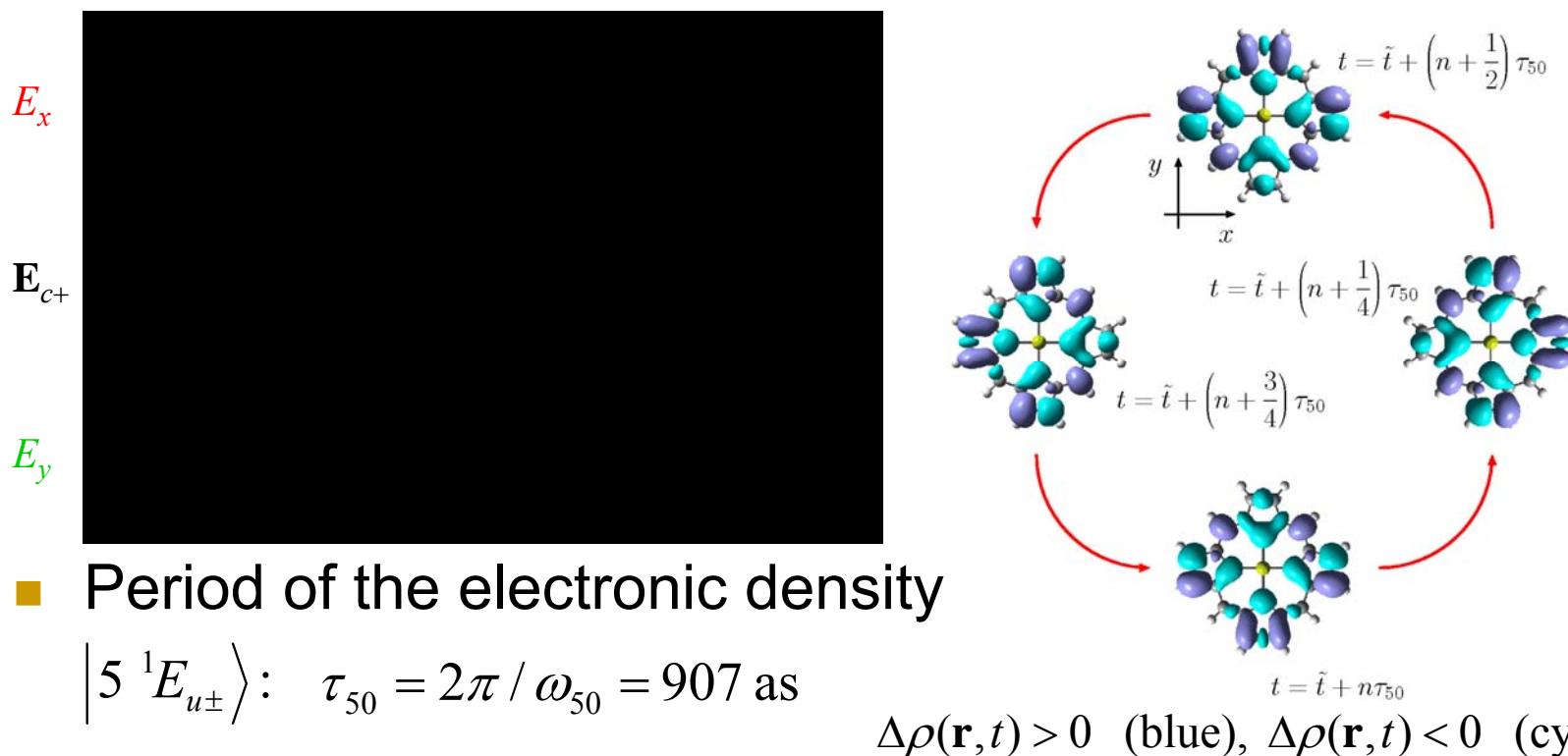
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I. Barth, J. Manz, Y. Shigeta, K. Yagi,
J. Am. Chem. Soc. 128, 7043 (2006)

E. Steiner et al, Org. Biomol. Chem. 3, 4053 (2005)
J. Jusélius et al, J. Org. Chem. 65, 5233 (2000)

Electron circulation in MgP

- Half population transfer $|X\ ^1A_{1g}\rangle \rightarrow |n\ ^1E_{u+}\rangle$ by means of a right circ. pol. optimized $\pi/2$ laser pulse with \cos^{20} envelope

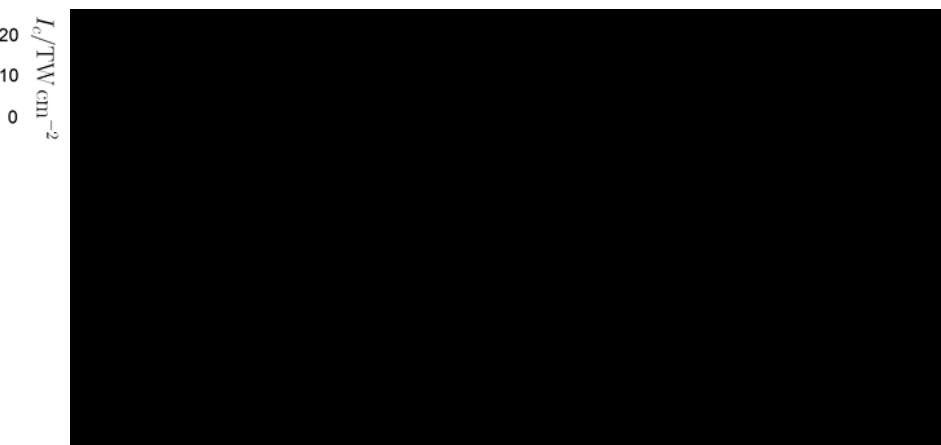
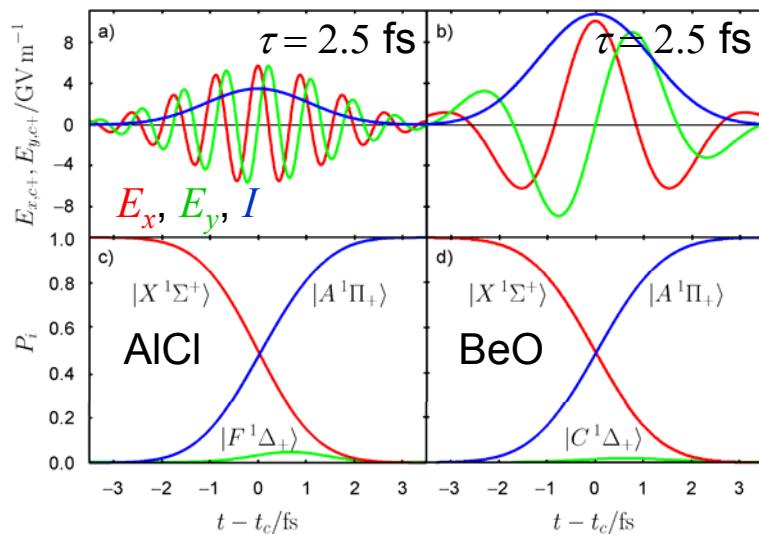


I. Barth, J. Manz, Angew. Chem. Int. Ed. 45, 2962 (2006), Angew. Chem. 118, 3028 (2006)

I. Barth, J. Manz, in: A.W. Castleman, Jr., M.L. Kimble (eds.), *Femtochemistry VII: Fundamental Ultrafast Processes in Chemistry, Physics, and Biology* (Elsevier, Amsterdam, 2006), p. 441

Electronic ring currents in AlCl and BeO

- Complete population transfer $|X^1\Sigma^+\rangle \rightarrow |A^1\Pi_+\rangle$ by means of a right circ. pol. optimized π laser pulse with \cos^{20} envelope



Difference of the electronic probability density

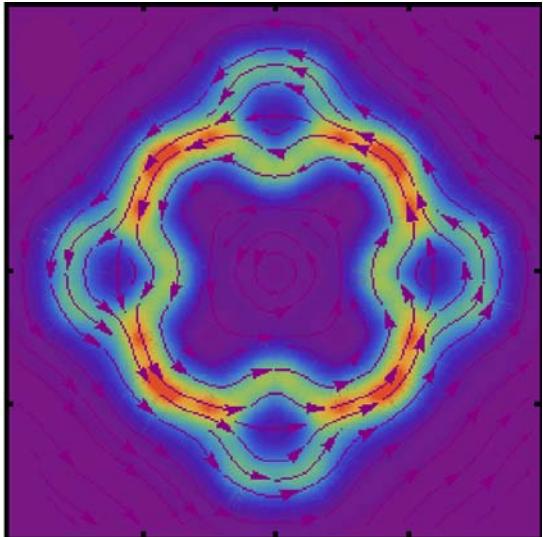
$$\Delta\rho(\mathbf{r},t) = \rho(\mathbf{r},t) - \rho(\mathbf{r},t_0)$$
$$\Delta\rho(\mathbf{r},t) > 0 \quad (\text{red})$$
$$\Delta\rho(\mathbf{r},t) < 0 \quad (\text{green})$$

I. Barth, J. Manz, L. Serrano-Andrés, Chem. Phys. 347, 263 (2008)

I. Barth, L. Serrano-Andrés, T. Seideman, J. Chem. Phys. 129, 164303 (2008); 130, 109901(E) (2009)

Comparison: Electronic ring currents

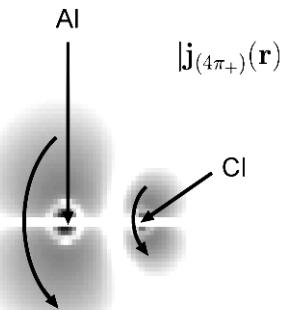
MgP



$$\begin{aligned}\langle r \rangle &= 6.32 a_0 \\ I &= 84.5 \mu\text{A} \\ T &= 1.90 \text{ fs} \\ B_{ind}(\text{Mg}) &= 0.16 \text{ T} \\ \langle L_{z,el} \rangle &\approx 2.5 \hbar\end{aligned}$$

I. Barth, J. Manz, Y. Shigeta, K. Yagi,
J. Am. Chem. Soc. 128, 7043 (2006)

AlCl



$$\begin{aligned}\langle r^{-1} \rangle^{-1} &= 0.18 a_0 \\ I &= 405 \mu\text{A} \\ T &= 396 \text{ as} \\ B_{ind}(\text{Al}) &= 7.68 \text{ T} \\ \langle L_{z,el} \rangle &= 1 \hbar\end{aligned}$$

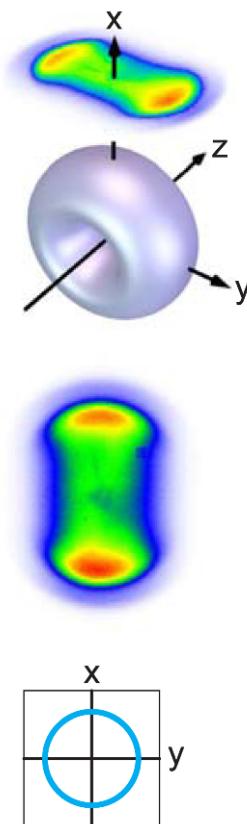
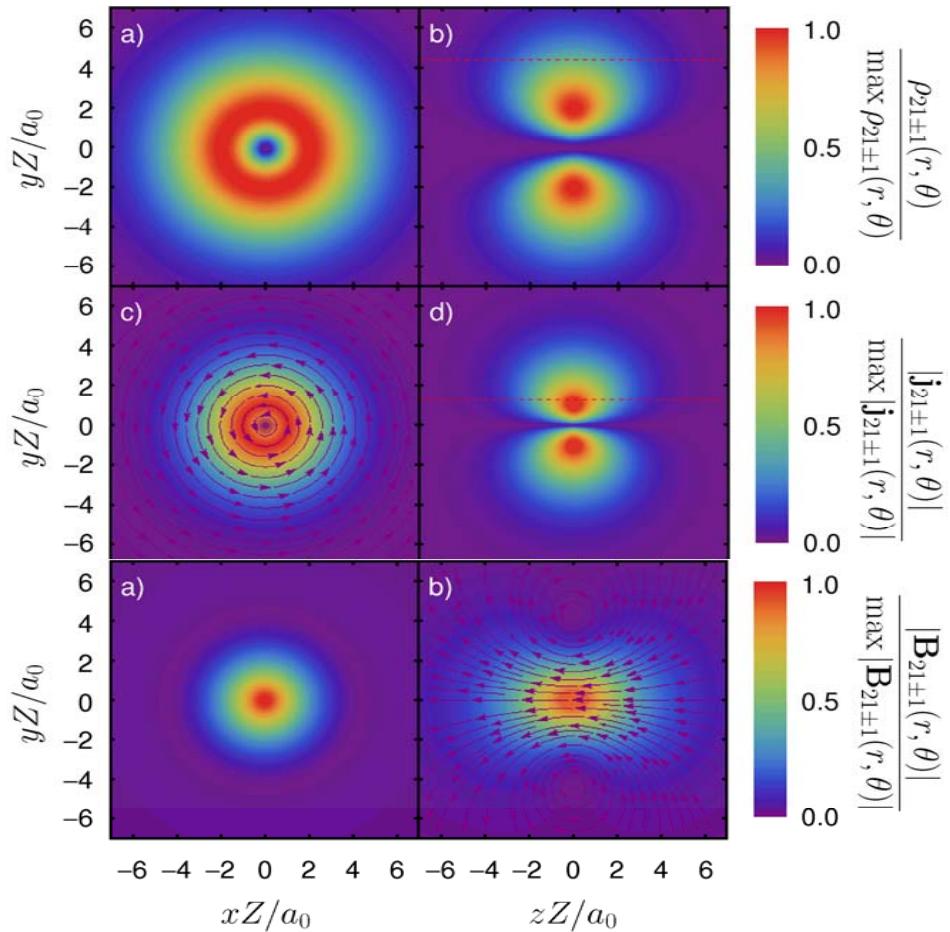
I. Barth, J. Manz, L. Serrano-Andrés,
Chem. Phys. 347, 263 (2008)

Biot-Savart law
for ring loop model:

$$\mathbf{B}(\mathbf{r} = \mathbf{0}) \approx -\frac{\mu_0 I}{2R} \mathbf{e}_z$$

$$I = \frac{Q}{T}$$

Electronic ring currents in atomic orbitals



ionized
 $f(m=\pm 3)$ orbital
of the K atom

**experimentally
observed**

by means of
circularly polarized
laser pulses

I. Barth, J. Manz, Phys. Rev. A 75, 012510 (2007)
I. Barth, PhD thesis, FU Berlin (2009)

M. Wollenhaupt et al,
Appl. Phys. B 95, 245 (2009)

Agenda

✓ Introduction

✓ Applications to **electronic fluxes**

✗ Applications to **nuclear** fluxes

- Double proton transfer: **porphyrin** (2D)

- Applications to concerted **electronic and nuclear** fluxes

From Synchronous to Sequential Double Proton Transfer: Quantum Dynamics Simulation for the Model Porphine

A. Accardi, I. Barth, O. Kühn, J. Manz

MOTIVATION

System:

- Porphyrins and metalloporphyrins, as well known as the “pigments of life”, play a decisive role in biological processes^[1], such as:

- Photosynthesis (*chlorins*)
- Oxygen transport (*hemoglobin and myoglobin*)
- Oxygen activation (*Cytochromes*)

Processes:

- Double Proton Transfer is important (for example, it may cause mutation of DNA base pair)^[2].

[1] H.-H. Limbach, J.T. Hynes, J.P. Klinman, R. L. Schowen, *Hydrogen-Transfer Reaction, Vol. 1*, Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim, 2007.

[2] M. Meuwly, A. Müller and S. Leutwyler, *PCCP*, **5**, 2663-2672 (2003)

MODEL [3]

$$H = T(X_1, X_2) + V(X_1, X_2)$$

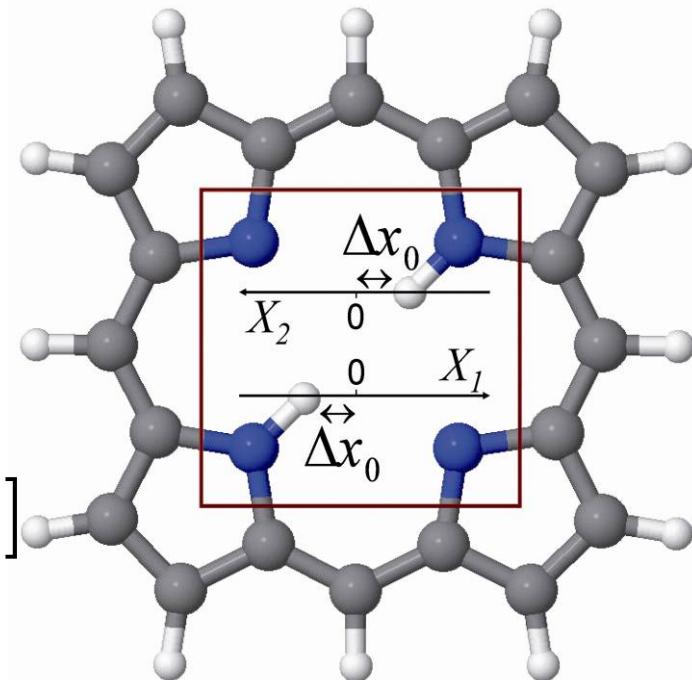
$$T = \frac{P_1^2}{2m_{H_1}} + \frac{P_2^2}{2m_{H_2}}$$

$$V = \frac{U_0}{\Delta x_0^4} \left[(X_1^2 - \Delta x_0^2)^2 + (X_2^2 - \Delta x_0^2)^2 - 4G\Delta x_0^2 X_1 X_2 \right]$$

$$U_0 = 0.473 eV , \Delta x_0 = 1.251 a_0 \text{ and } G = 0.063$$

Parameters based on:

- Experimental value U_0 [5], [6], [7] → NMR, Laser-Induced Fluor.-Spect.
- Quantum chemistry for other values [8] → DFT/B3LYP/6-31G*



**Two dimensional model
supported in [4]: not
RRKM, slow IVR.**

[3] Z. Smedarchina, W. Siebrand, A. Fernandez-Ramos, *J. Chem. Phys.* **127**, 174513 (2007)

[4] T. J. Butenhoff, R. S. Chuck, H.-H. Limbach, C. B. Moore *J. Phys. Chem.* **1990**, *94*, 7847-7851.

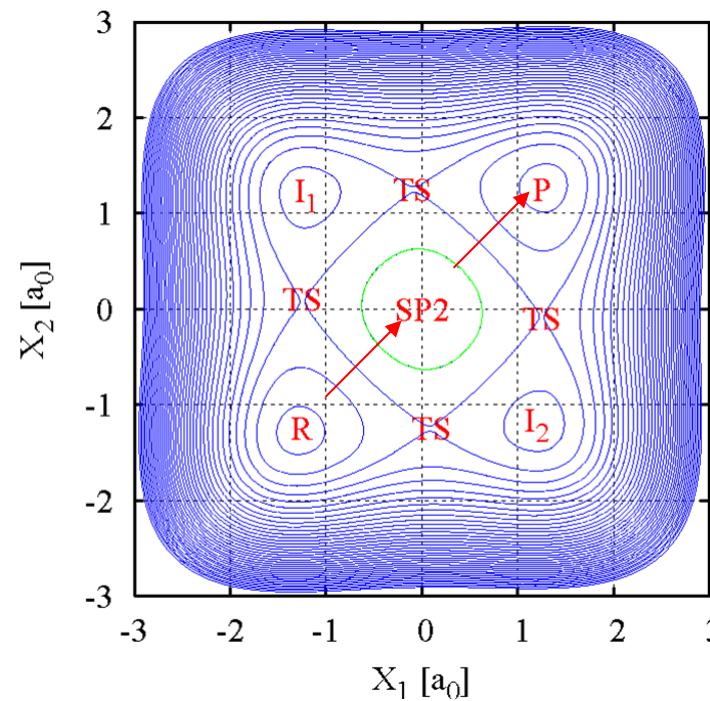
[5] J. Braun, M. Schlabach, B. Wehrle, M. Kocher, E. Vogel, and H.-H. Limbach, *J. Am. Chem. Soc.* **116**, 6593 (1994)

[6] J. Braun, H.-H. Limbach, P. G. Williams, H. Moromoto, and D. E. Wemmer, *J. Am. Chem. Soc.* **118**, 7241 (1996)

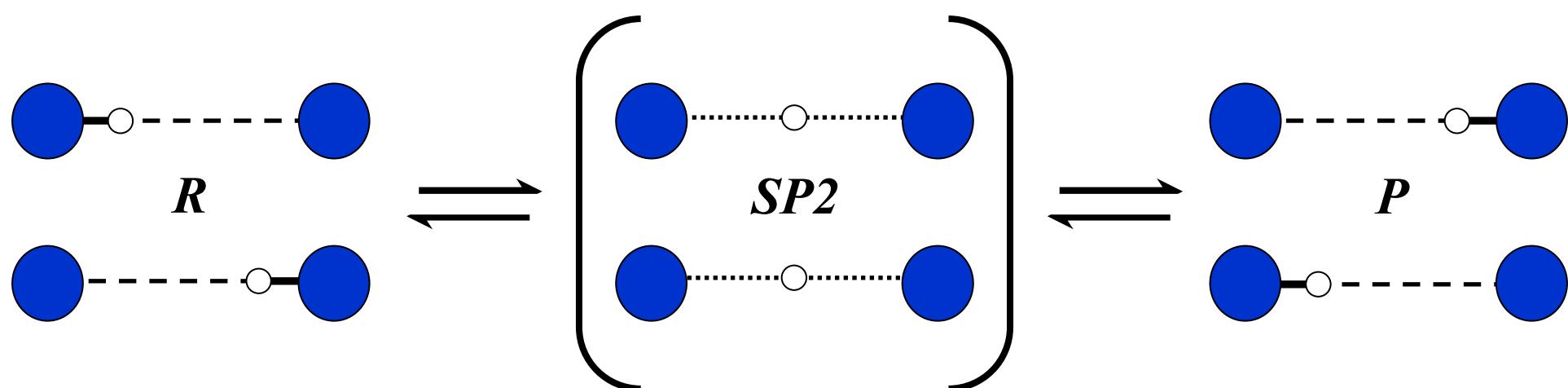
[7] T. J. Butenhoff, C. B. Moore *J. Am. Chem. Soc.* **1988**, *110*, 8336.

[8] Z. Smedarchina, M. Z. Zgierski, W. Siebrand, and P. M. Kozlowsji, *J. Chem. Phys.* **109**, 1014 (1998)

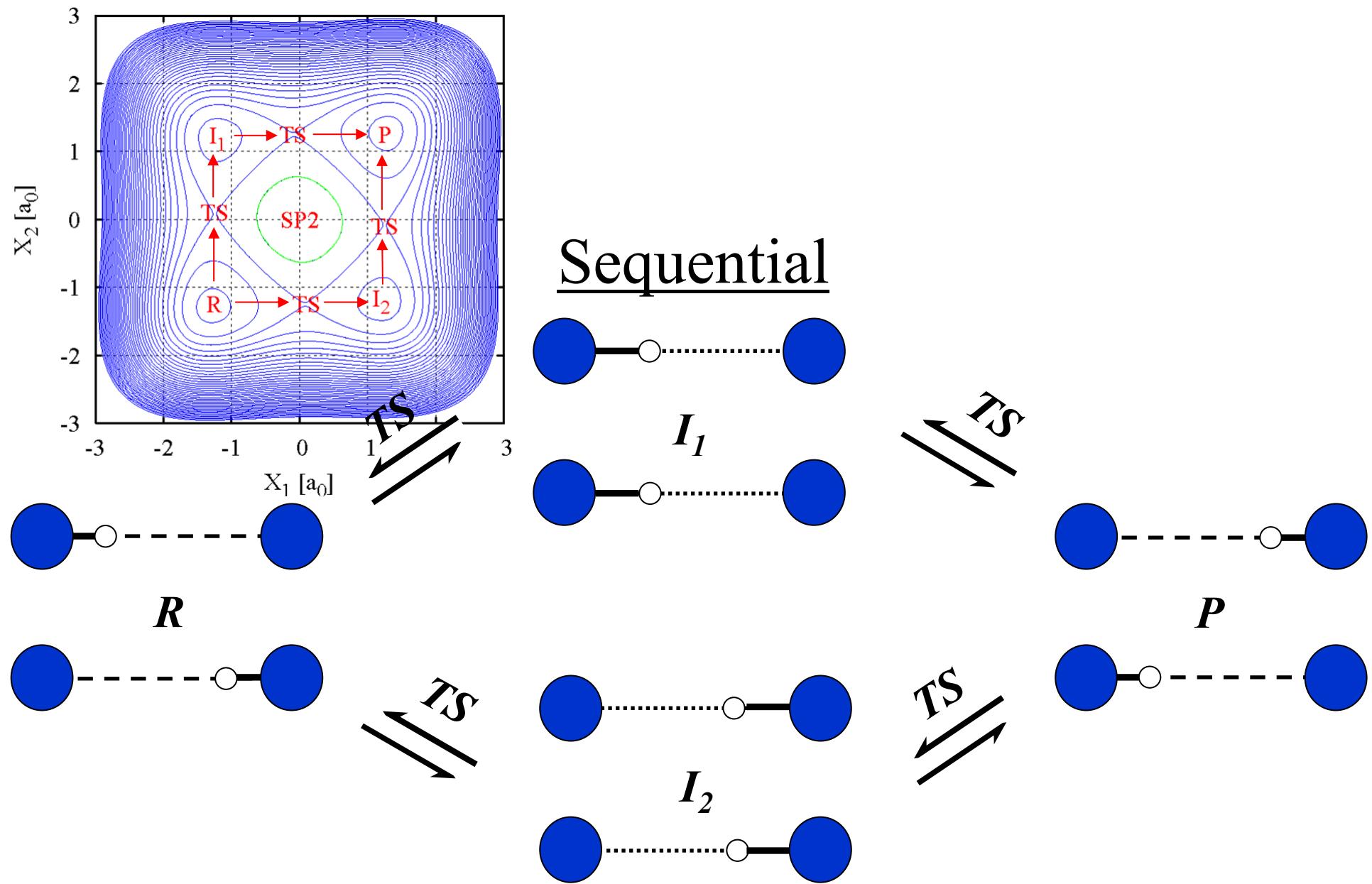
DOUBLE PROTON TRANSFER



Synchronous

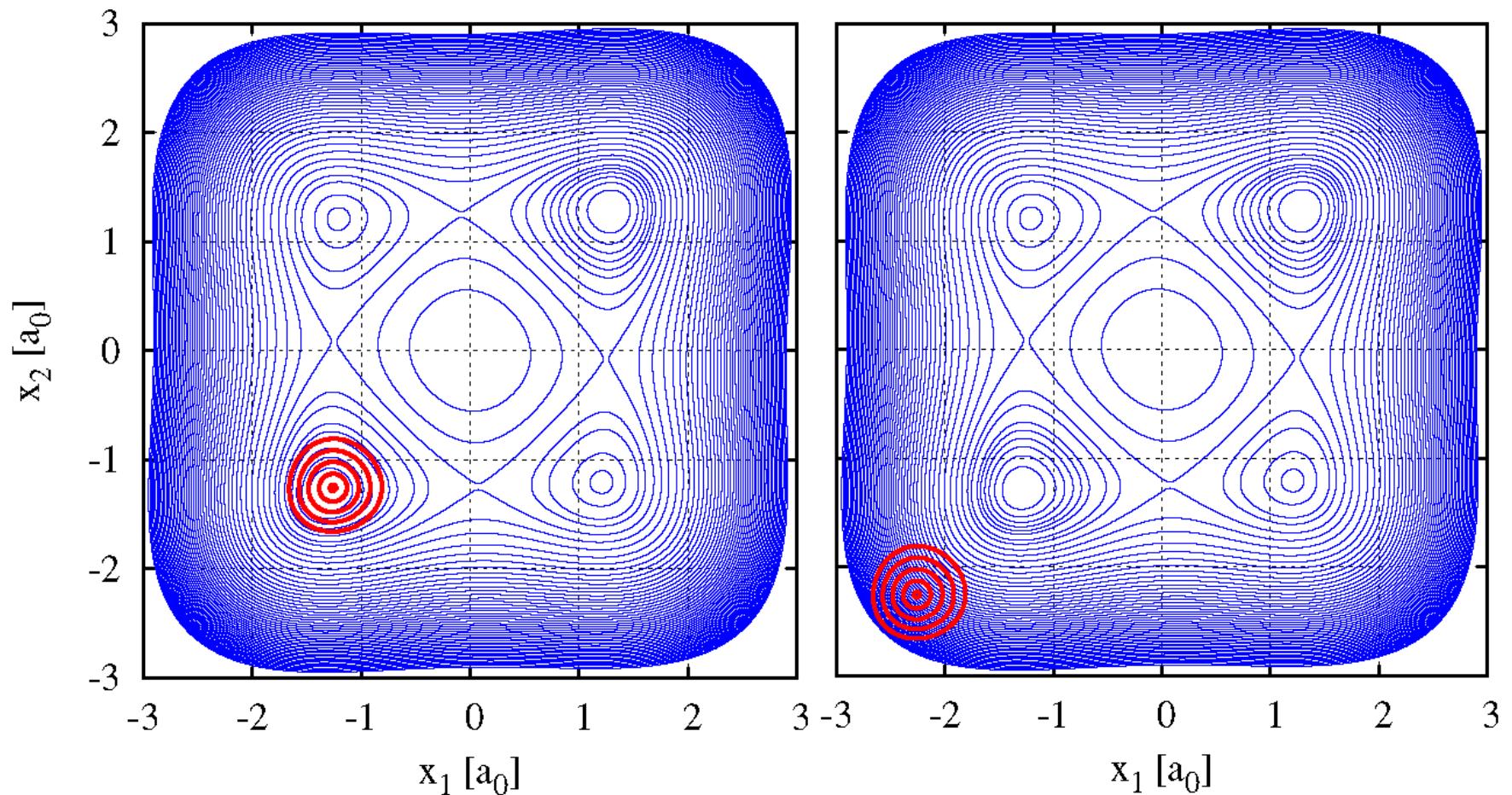


DOUBLE PROTON TRANSFER



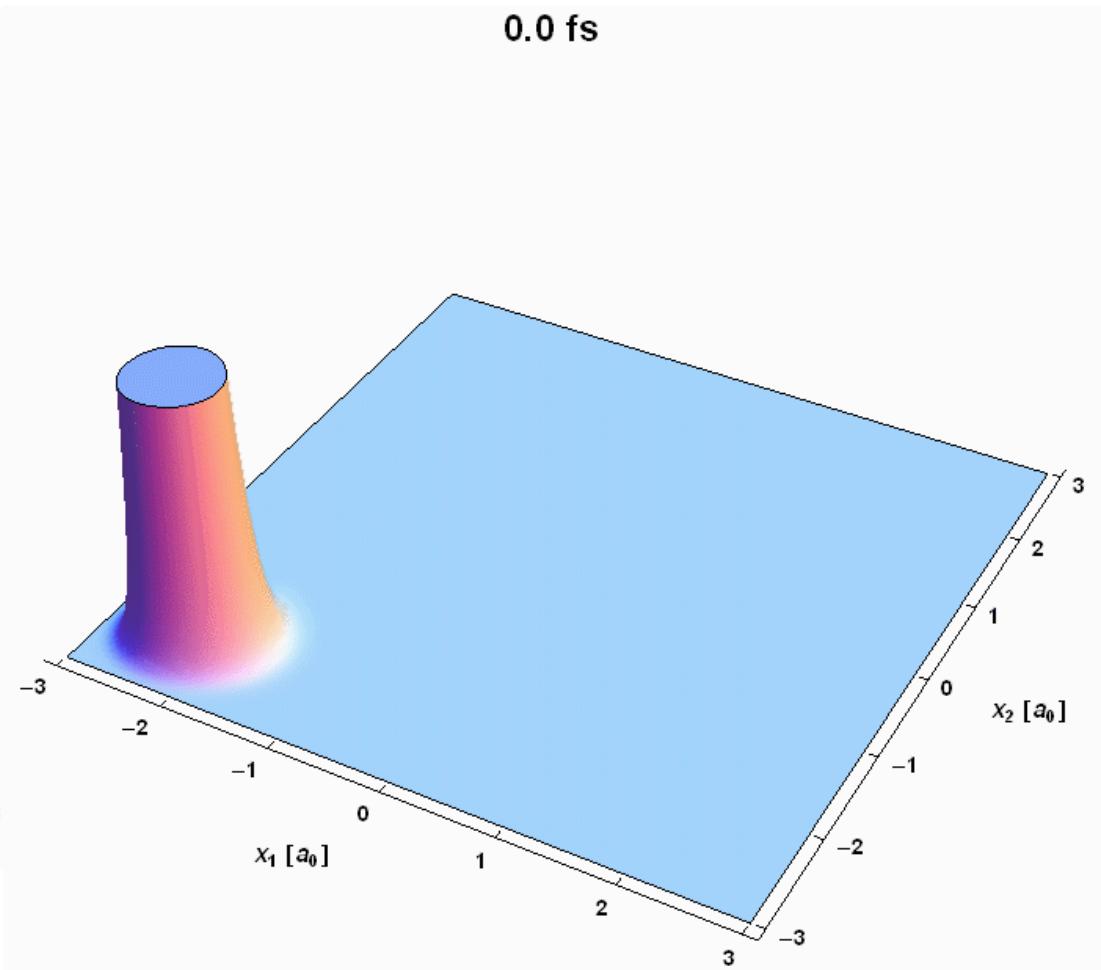
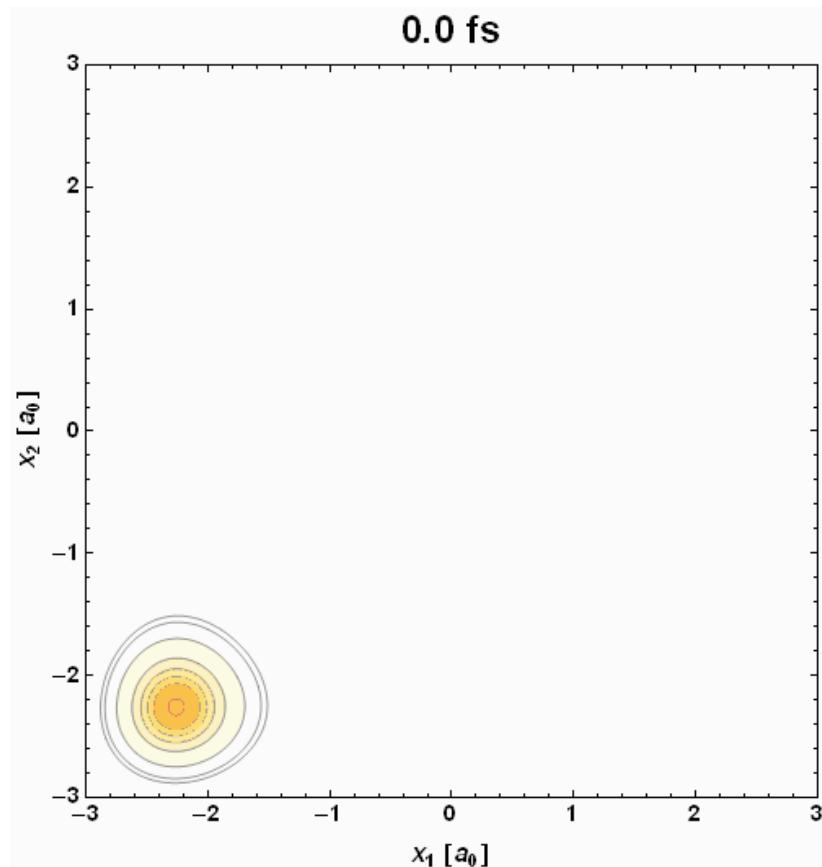
INITIAL CONDITION

Initial displacement of the wave packet^[11]

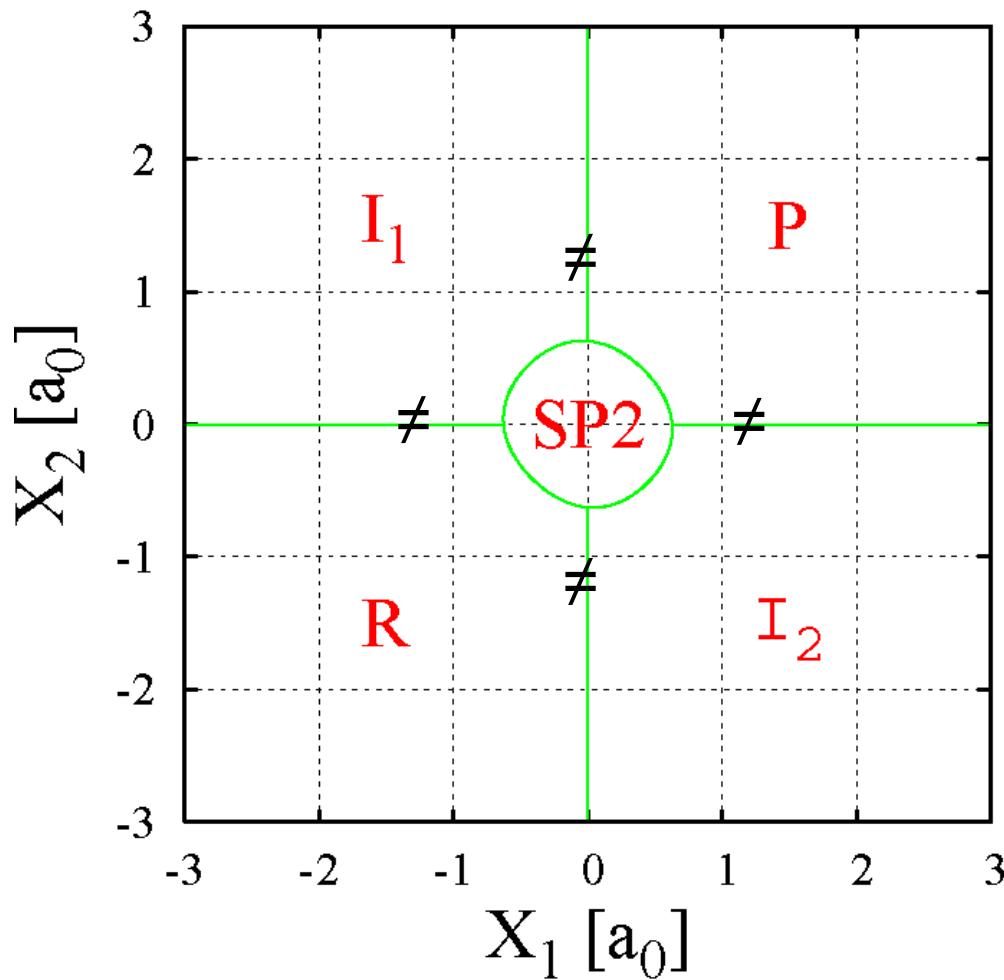


[11] W. Li, X. Zhou, R. Lock, S. Patchovskii, A. Stolow, H.C. Kapteyn, M.M. Murnane, Science, **322**, 1207 (2008)

NUCLEAR DYNAMICS



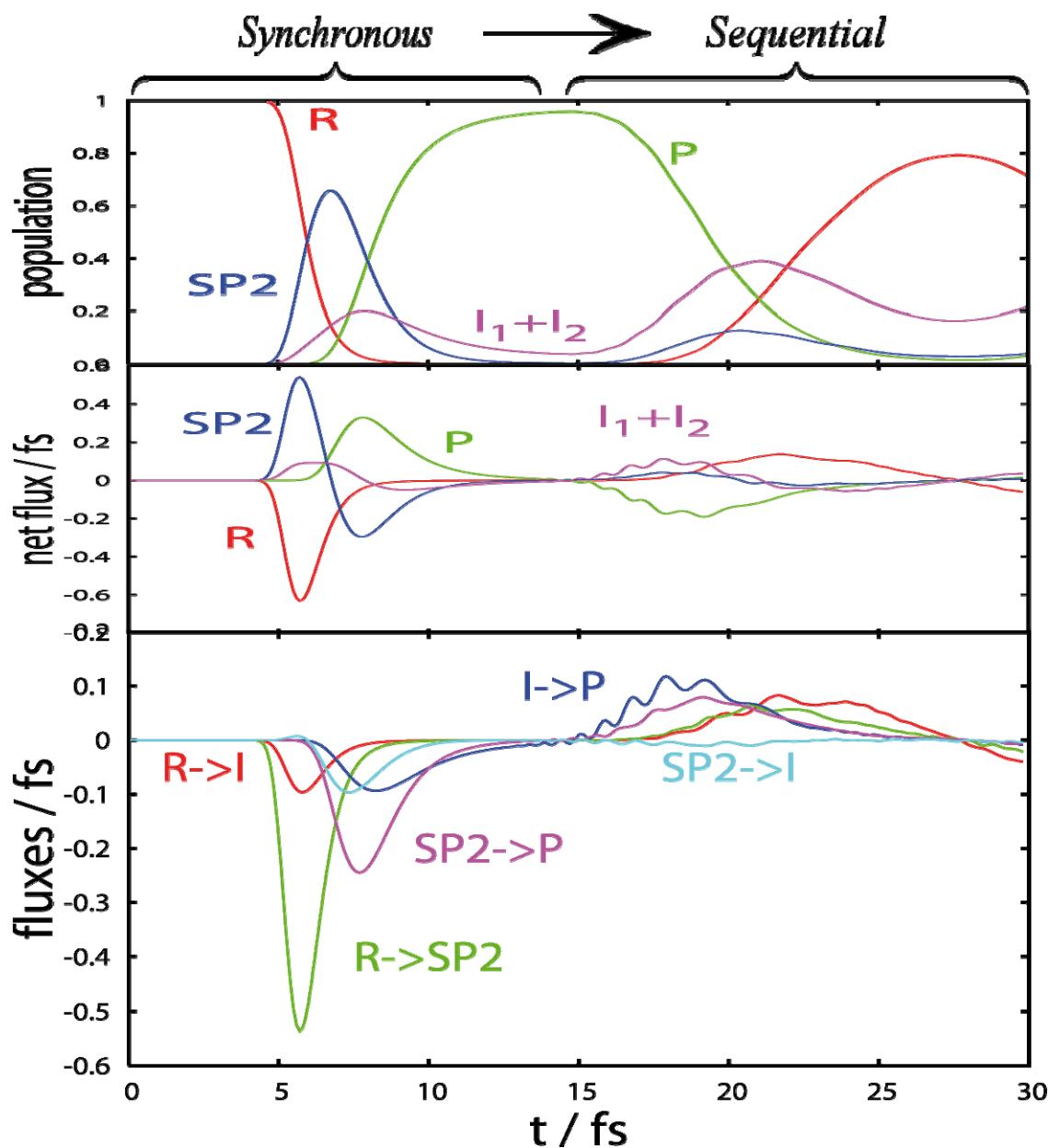
DOMAINS OF THE POTENTIAL



Time-dependent population
of domains:

$$P_{Domain}(t) = \iint_{Domain} \rho(X_1, X_2, t) dX_1 dX_2$$

POPULATION AND FLUX DYNAMICS



CONCLUSIONS

Fluxes → Analysis of mechanism

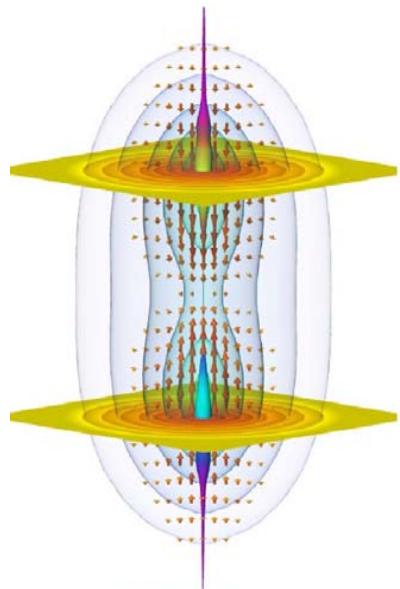
Double Proton Transfer

- 1) Synchronous → sequential
- 2) in general:
quantum chemistry \otimes quantum reaction dynamics

Agenda

- ✓ Introduction
- ✓ Applications to **electronic fluxes**
- ✓ Applications to **nuclear fluxes**
- ✗ Applications to concerted **electronic and nuclear fluxes**
 - Vibration dynamics: H_2^+ (3D+1D)
 - Cope rearrangement: **Semibullvalene** (multi-D+1D)

Concerted electronic and nuclear fluxes in molecules

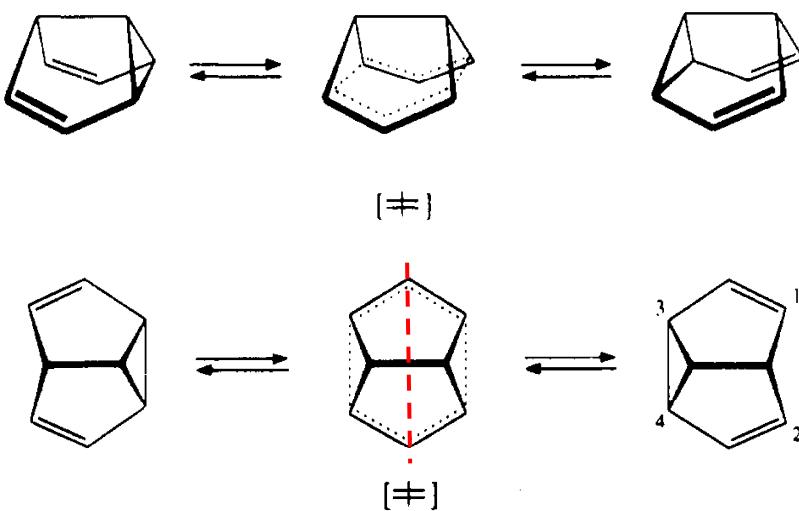


Vibration dynamics in H_2^+

I. Barth, H.-C. Hege, H. Ikeda, A. Kenfack,
M. Koppitz, J. Manz, F. Marquardt,
G.K. Paramonov,
Chem. Phys. Lett. 481, 118 (2009)

Cope rearrangement in Semibullvalene (SBV)

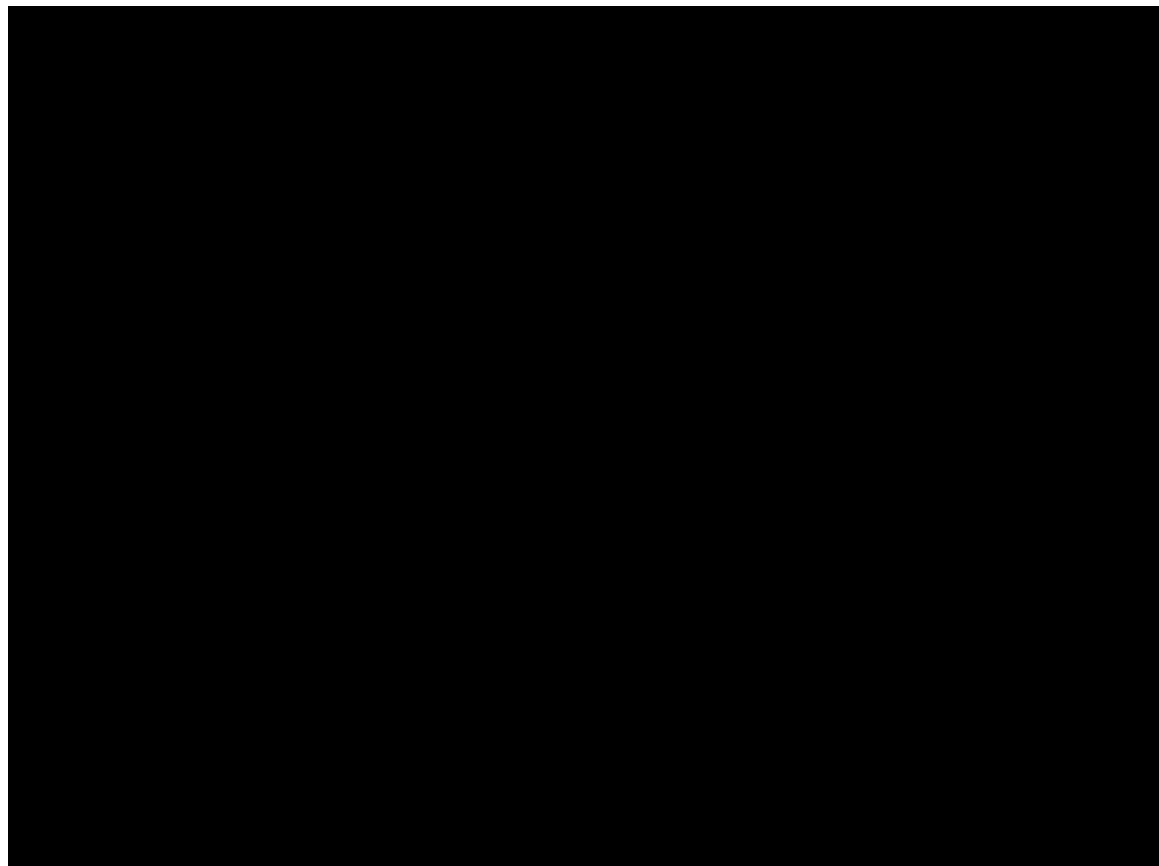
D. Andrae, I. Barth,
T. Bredtmann, J. Manz, B. Paulus,
in preparation



- A.D. Bandrauk, S. Chelkowski, P.B. Corkum, J. Manz, G.L. Yudin, J. Phys. B 42, 134001 (2009)
M. Okuyama, K. Takatsuka, Chem. Phys. Lett. 476, 109 (2009)
K. Nagashima, K. Takatsuka, J. Phys. Chem. A (2009), in press

Concerted electronic and nuclear fluxes in molecules: Vibration dynamics in H_2^+

- *Example:*
*Vibration dynamics
in H_2^+*



- *Initialization:*

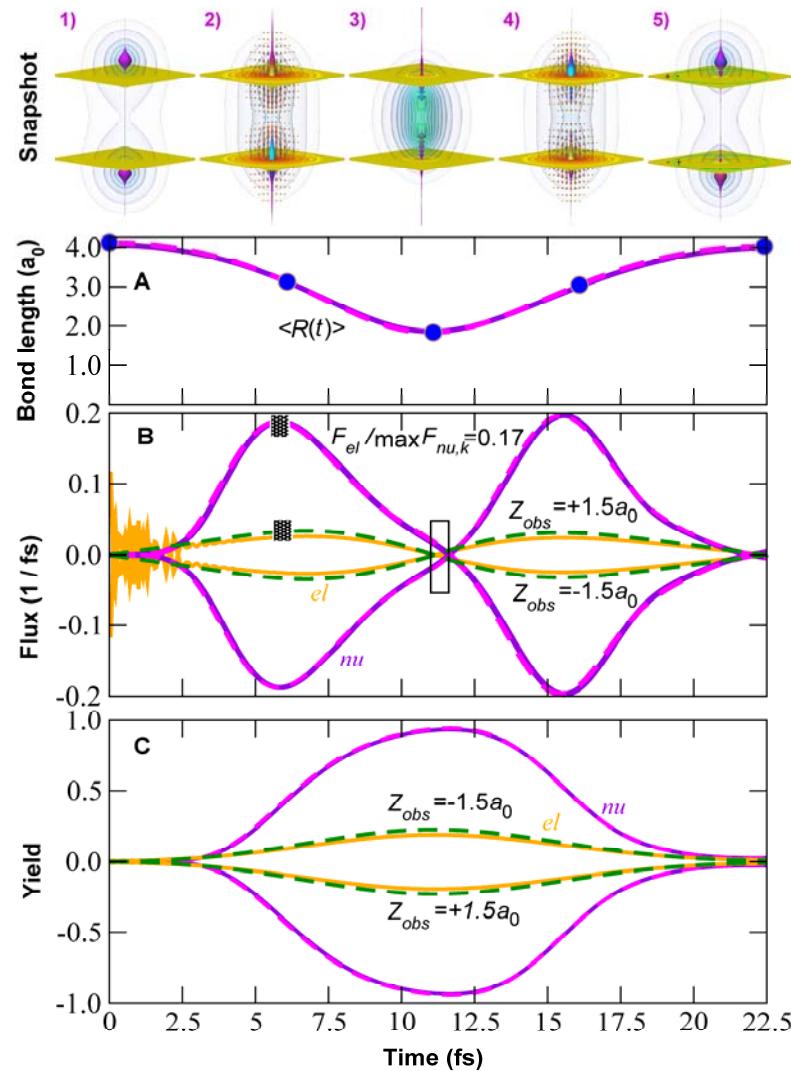
W. Li, X. Zhou, R. Lock, S. Patchkovskii,
A. Stolow, H.C. Kapteyn, M.M. Murnane,
Science 322 (2008) 1207.

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Chem. Phys. Lett. 481, 118 (2009)

Concerted electronic and nuclear fluxes in molecules: SBV

- Example:
Cope rearrangement in Semibullvalene (SBV)

K. Bergmann, S. Görtler, J. Manz, and H. Quast,
J. Am. Chem. Soc. **1993**, 115, 1490-1495.

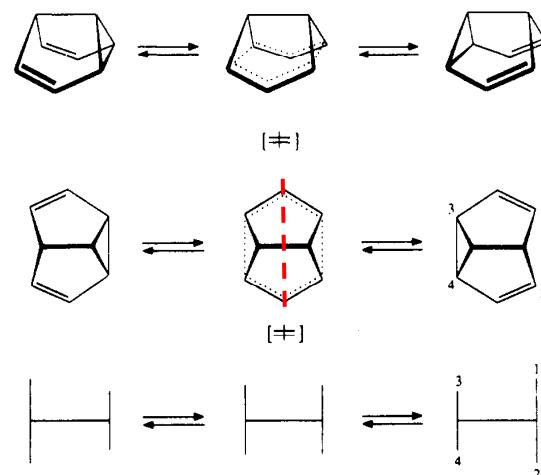
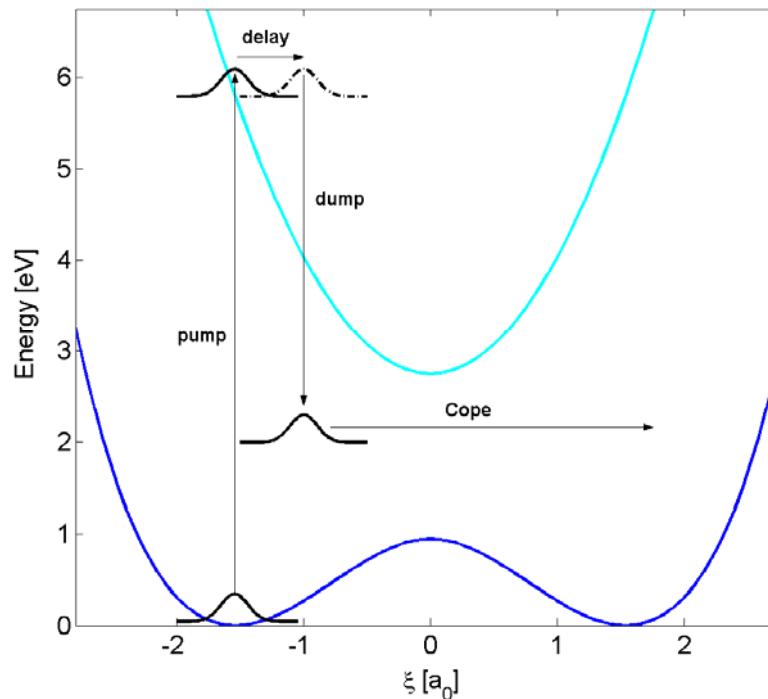
M. Dohle, J. Manz, G.K. Paramonov, H. Quast,
Chem. Phys. **1995**, 197, 91-97.

M. Dohle, J. Manz, G.K. Paramonov,
Ber. Bunsenges. Phys. Chem. **1995**, 99,
478-484.

- Initialization:

W. Li, X. Zhou, R. Lock, S. Patchkovskii,
A. Stolow, H.C. Kapteyn, M.M. Murnane,
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**D. Andrae, I. Barth, T. Bredtmann,
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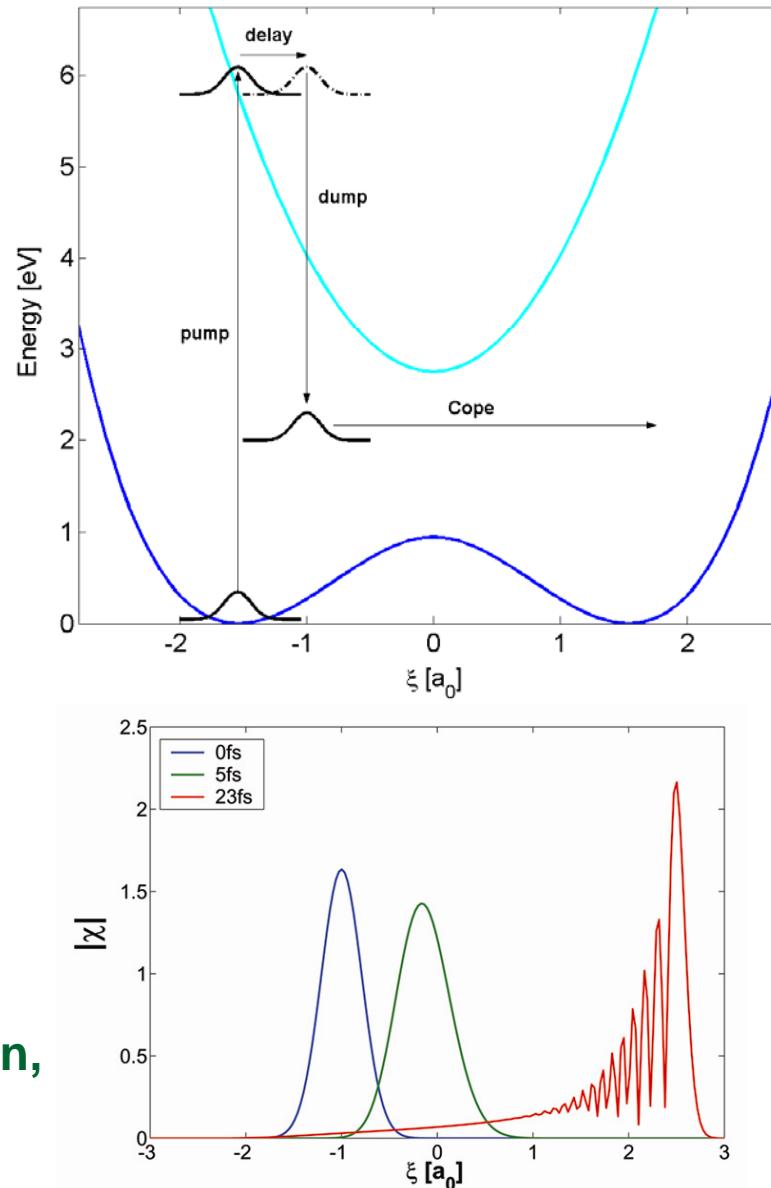
M. Dohle, J. Manz, G.K. Paramonov, H. Quast,
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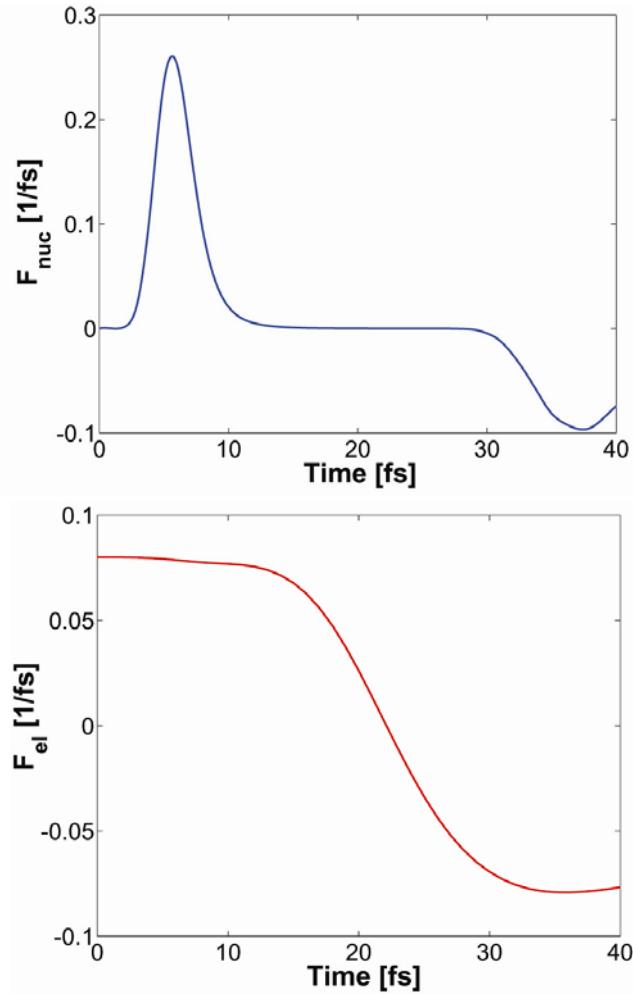
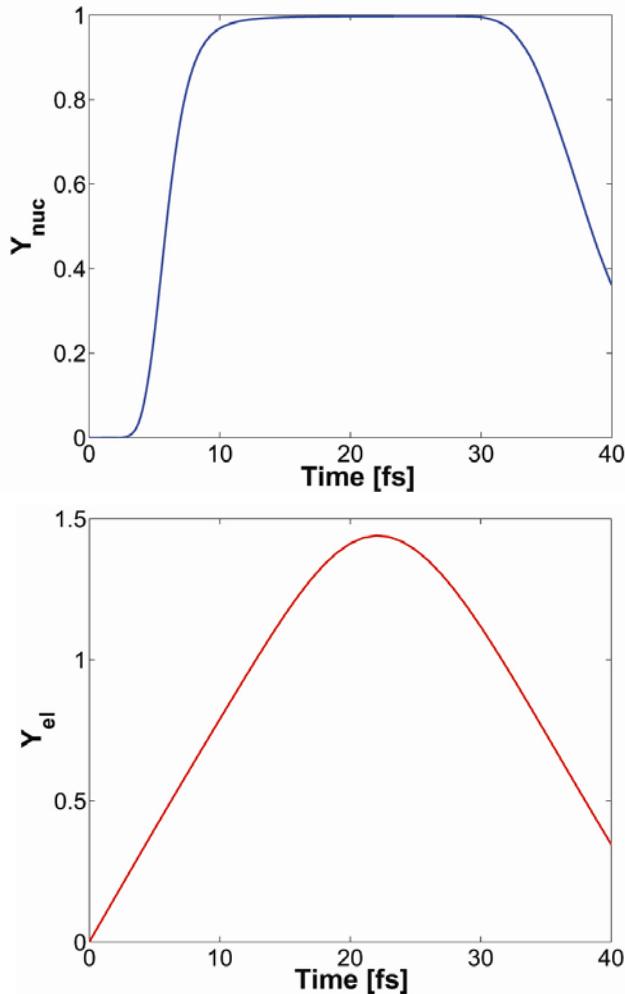
- *Initialization:*

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A. Kaushik, A. Kenfack,
J. Manz, B. Paulus**



Concerted electronic and nuclear fluxes in molecules: Cope rearrangement in SBV



Conclusions and outlook

Flux: → analysis of reaction mechanism
→ concerted effects of coupled nuclei and electrons

Challenges for future

Calculation of the time-dependent electronic flux density in large molecules

Experiments for coupled quantum effects of electron and nuclear dynamics

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- international partners



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GK 788 (project A1)



CSS



FCI

Group at FU Berlin

