

Radiochemical Diagnostics: A Cornerstone of Stockpile Stewardship

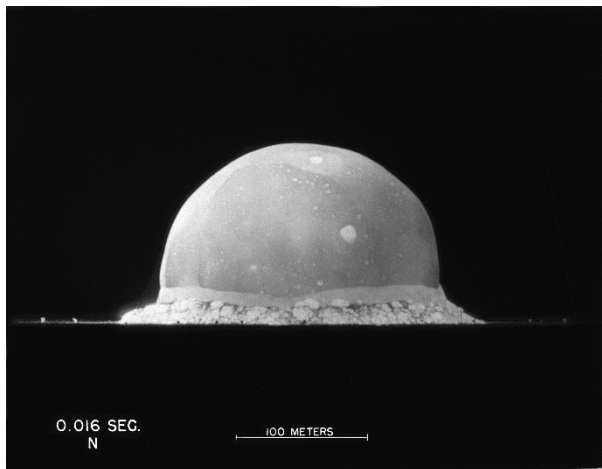
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27 February 2020

The Significance of Radiochemistry

The Trinity Fireball

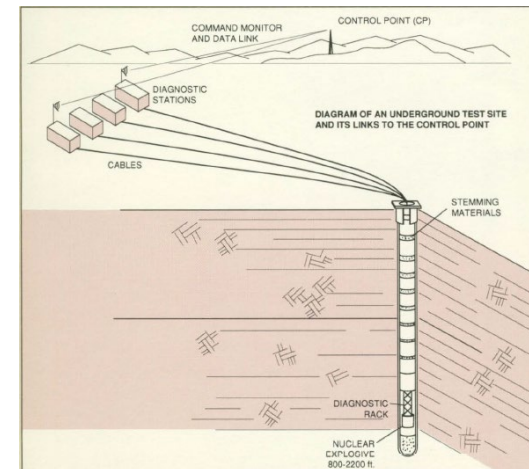


Photos Courtesy R.A Meade

Operation Crossroads B-17



Underground Testing



LALP-88-21

- **Radiochemistry established the yield of every sampled test, from Trinity to Divider**
- Radiochemistry is the gold standard for yield in today's Stockpile Stewardship Program (SSP)
- Radiochemistry plays an integral role in our ever-advancing understanding of weapons physics
- Significant R&D and programmatic investment in radiochemistry continues today
 - Basic nuclear physics
 - Sophisticated chemistry and counting methods
 - Mass spectrometry
 - Modeling and simulation

What is Radiochemical Debris Diagnostics?

Photos courtesy of W. Oldham, S. Hanson

Sample
Collection



Dissolution
and separation



Counting



- Assessment of events in support of the AAR, LEPs and experimental science campaign milestones
- Interface between radiochemical measurement and design communities

- Integrate data streams (e.g MS, counting)
- Account for chemical and physical phenomenology present in raw sample data
- Employ computational tools to determine performance from sample data
- Provide detailed assessments regarding impact of design on total performance
- The product of this work *is the Gold Standard*

How is Weapon Performance Determined?



Blink!



BOOM!



Nuclear test debris is the neutron analog to photon exposed photographic film

$$Y_x = M_x \times E_x; E \equiv \frac{F_{total}}{Fuel_x^0}$$

For plutonium:

$$E_{pu} = \frac{F_{total}}{Pu^0}; \xrightarrow{\text{sample}} \frac{f_s}{f_s + pu^{res}}$$

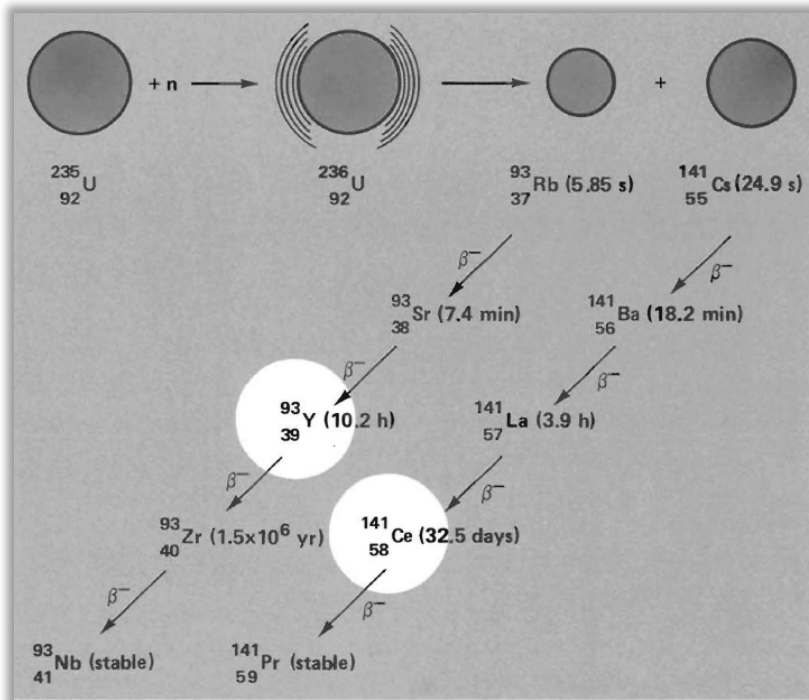
- “Inventory” equations
- Sample interferences must be accounted for
 - Chemical fractionation
 - Uranium blank
- Note: it is unnecessary to know how much of the device was collected

Each Inventory Term is a *Measurement* – Consider Sample Fissions

$$f_s = N_{fp,s} \div CY_{fp}$$

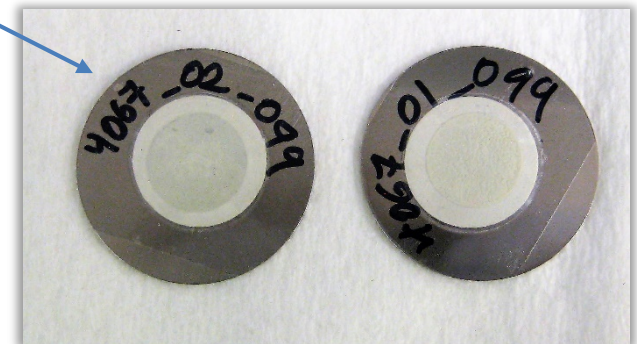
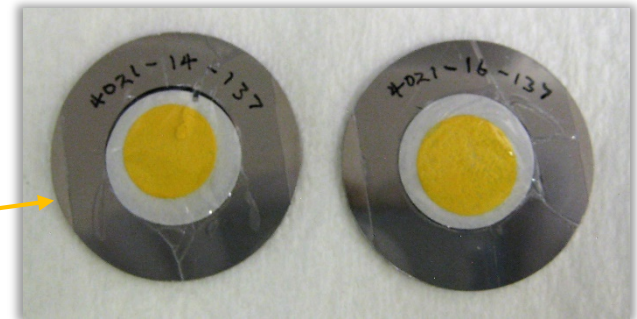
Each sample is yielded gravimetrically and mounted

Example mount forms:



Knight and Sattizahn, Los Alamos Science, 1983

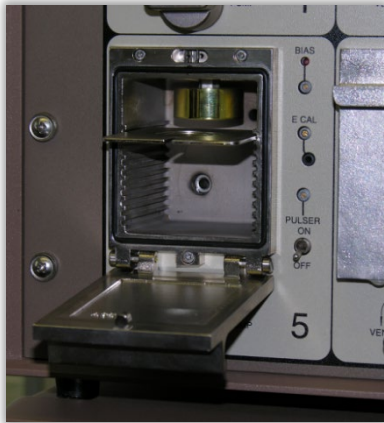
- SrCO₃
- Y₂O₃
- Cs₂PtCl₆
- MoO₃
- BaCrO₄
- Nd₂O₃
- Ag metal
- Cd metal
- Tb₄O₇



Photos courtesy of W. Oldham, S. Hanson

Measurement of Actinide Fuels

- Actinide fuels are the first term in $y = mass * efficiency$
- LANL point of excellence
 - World-class scientists and instrumentation
 - Alpha spectroscopy and mass spectrometry are primary tools

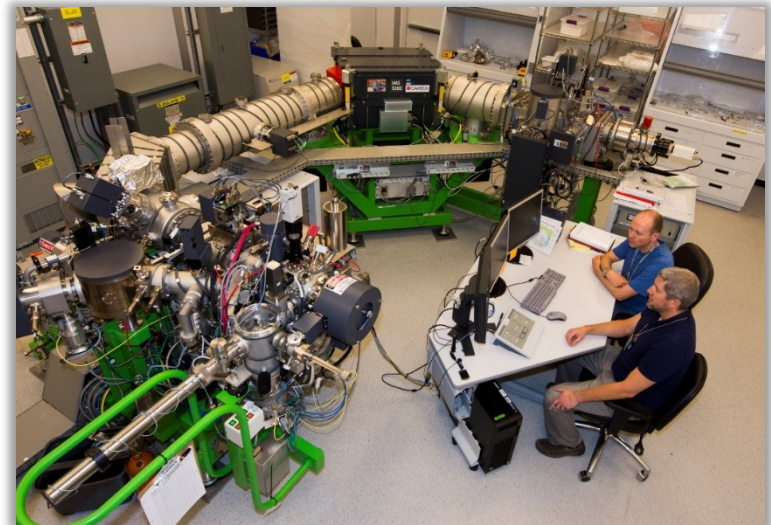


High resolution α -counting
($^{238-240}\text{Pu}$, ^{241}Am , ^{242}Cm)

Photos courtesy of D. Dry, F. Taw, R. Steiner



Thermal Ionization MS
(long-lived U, Pu)



Secondary Ion MS (Spatially resolved actinide isotopic analysis)

So, About Those Nuclear Tests...

- **Nuclear testing ended in 1992**
- **The stockpile did not**
 - How do we ensure the function of our weapons with no testing?
 - How does radiochemistry remain relevant?
- **Scientific Stockpile Stewardship was born in the mid-1990s**
 - Based on advanced computing and simulation of tests
- **Requires more and higher quality radiochemical assessments with much better nuclear data and uncertainty than was possible during testing**
 - Radiochemical assessment has advanced far beyond testing and its requirements have advanced nuclear science
 - Radiochemistry underpins several missions in addition to Stewardship



Radiochemistry at Los Alamos Today: Chemistry Division's Nuclear and Radiochemistry Group

Primary capabilities:

Major program thrusts:

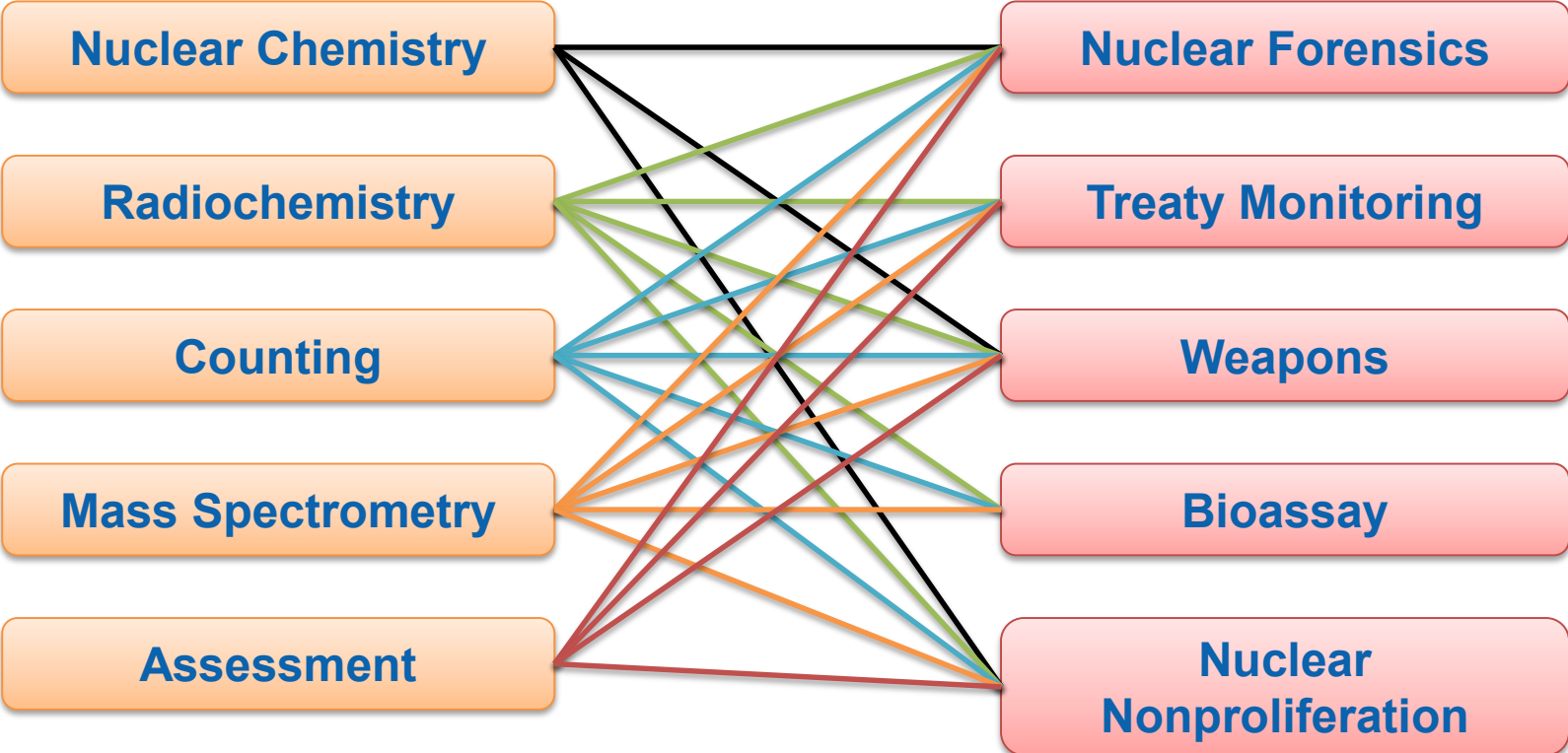


Figure courtesy of F. Taw

Over 100 Personnel Executing Mission and R&D

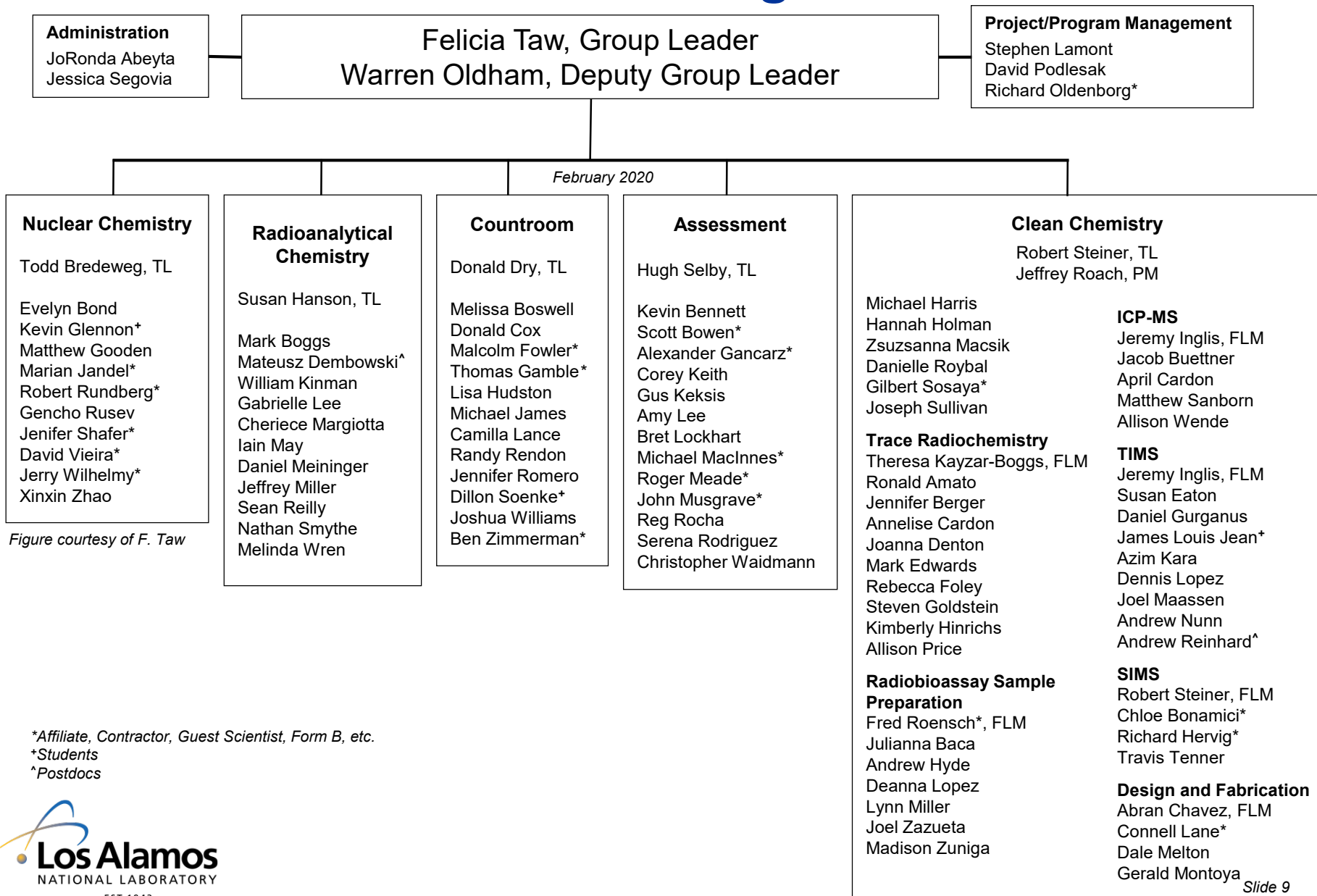


Figure courtesy of F. Taw

*Affiliate, Contractor, Guest Scientist, Form B, etc.

*Students

^Postdocs

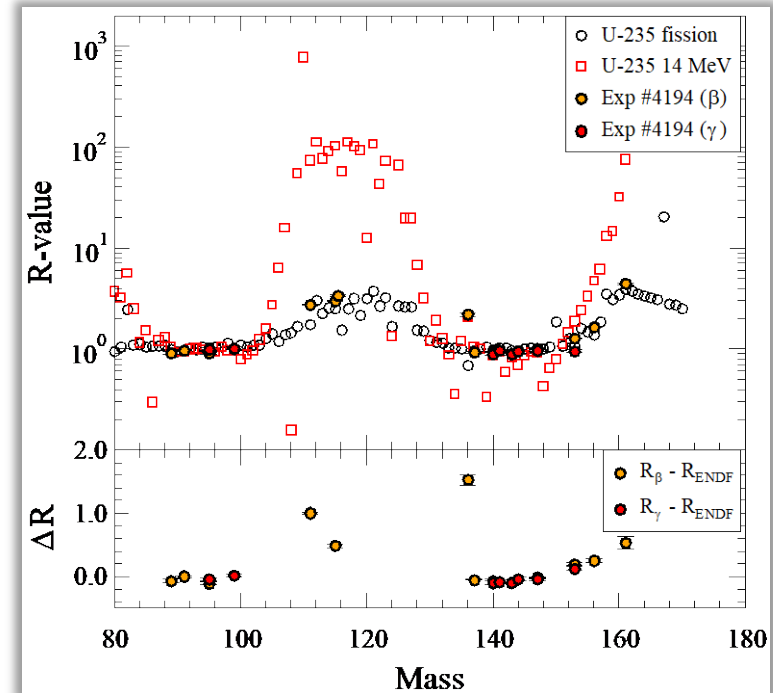
Research Highlights, CY_{fp} (Cumulative Fission Product Yields)

Objective: Measure integral cumulative fission product yields, $R_i^{j,k}$ and $Y_i^{j,k}$, in relevant neutron fields for several major and minor actinides.

Relevance: Fission product yields represent important nuclear fission observables for basic science as well as numerous nuclear applications. This and related work provides experimental data to validate and improve differential nuclear data, nuclear physics modeling and operational tools.

Approach: Make use of critical assemblies and other neutron sources to perform fission chamber measurements in parallel with historical radiochemical analysis on select actinides to determine relative ($R_i^{j,k}$) and absolute ($Y_i^{j,k}$) fission product yields.

Collaborations: Most of these experiments have included participation from LLNL and/or PNNL.



Figures courtesy of T. Bredeweg

Accomplishments/Results: Executed 1-2 irradiations per year since 2012 on various actinide samples including ^{233}U , ^{235}U , ^{238}U , ^{237}Np and ^{239}Pu . For each actinide sample we extracted relative fission product yields, $R_i^{j,k}$, by radiochemical analysis.

Tested and fielded prototype fission chambers with ^{235}U and ^{237}Np reference and macro-foils on the Flattop assembly to extract absolute fission product yields, $Y_i^{j,k}$.

Research Highlights, Extinct Radionuclides

Project Goal: Reconstruct the entire suite of diagnostically useful radionuclides by measuring perturbations in stable element isotope ratios and actinide composition.

LANL LDRD-DR

PI: Hugh Selby (C-NR)

Co-PI for modeling: John Scott (XTD-IDA)

Co-PI for measurements: Warren Oldham (C-NR)

Measurements of extinct fission products in nuclear bomb debris: Determination of the yield of the Trinity nuclear test 70 y later

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Edited by Thure E. Cerling, University of Utah, Salt Lake City, UT, and approved May 16, 2016 (received for review February 19, 2016)

Abstract: This paper describes an approach to measuring extinct fission products that would allow for the characterization of a nuclear test at any time. The isotopic composition of molybdenum in five samples of glassy debris from the 1945 Trinity nuclear test has been measured, reflecting both the short-lived fission products and the long-lived fission products that have decayed to stable elements. Nonnatural molybdenum isotopic compositions were observed, reflecting an input from the decay of the short-lived fission products ⁹²Zr and ⁹²Nb. By measuring the total amount of ⁹²Mo/⁹⁶Mo and ⁹⁴Mo/⁹⁶Mo isotopic ratios and the total amount of ⁹²Mo/⁹⁶Mo and ⁹⁴Mo/⁹⁶Mo isotopic ratios, it is possible to calculate the original concentration of the ⁹²Zr and ⁹²Nb isotopes in the nuclear debris. Together with a determination of the amount of plutonium in the debris, these measurements of extinct fission products allow for new estimates of the efficiency and yield of the historic Trinity test.

Significance: This work describes an approach to postdetonation nuclear forensics involving isotopic measurements that allows for characterization of a nuclear detonation at any time. By performing high-precision measurements of stable isotope perturbations in nuclear bomb debris, it is possible to quantify short-lived fission products long after they have decayed below radiometric detection limits and become extinct. The extinct fission product products can be used to reconstruct details of the nuclear test months to years after the detonation occurred. The approach is demonstrated by analysis of debris from the Trinity nuclear test and new estimates of the efficiency and yield of the nuclear test are presented.

Author contributions: S.K.H., A.D.P., C.R.W., W.S.K., W.J.O., and H.D.S. designed research; S.K.H., A.D.P., C.R.W., W.S.K., A.M.W., J.L.M., and H.A.B. performed research; A.M.W., S.K.H., A.D.P., C.R.W., W.S.K., A.M.W., J.L.M., and H.A.B. analyzed data; W.S.K.H., C.R.W., W.J.O., and H.D.S. wrote the paper.

Supplementary Information: This article contains supporting information online at www.pnas.org/lookup/suppl/doi:10.1073/pnas.1602797113/-DCSupplemental.

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⁹² Mo 14.84	⁹³ Mo 3.5E3a ε	⁹⁴ Mo 9.25	⁹⁵ Mo 15.92	⁹⁶ Mo 16.68	⁹⁷ Mo 9.55	⁹⁸ Mo 24.13	⁹⁹ Mo 2.75d β	¹⁰⁰ Mo 9.63	¹⁰¹ Mo 14.61m β	¹⁰² Mo 11.3m β	¹⁰³ Mo 1.13m β	¹⁰⁴ Mo 1.00m β
⁹¹ Nb 7E2a ε	⁹² Nb 3.5E7a ε	⁹³ Nb 100	⁹⁴ Nb 2.0E4a β	⁹⁵ Nb 34.99d β	⁹⁶ Nb 23.4h β	⁹⁷ Nb 1.23h β	⁹⁸ Nb 2.9s β	⁹⁹ Nb 15.0s β	¹⁰⁰ Nb 5s β	¹⁰¹ Nb 7.1s β	¹⁰² Nb 1.3s β	¹⁰³ Nb 1.5s β
⁹⁰ Zr 51.45	⁹¹ Zr 11.22	⁹² Zr 17.15	⁹³ Zr 1.5E6a β	⁹⁴ Zr 17.38	⁹⁵ Zr 64.02d β	⁹⁶ Zr 2.8	⁹⁷ Zr 1.68h β	⁹⁸ Zr 30.7s β	⁹⁹ Zr 0.2s β	¹⁰⁰ Zr 1s β	¹⁰¹ Zr 2.4s β	¹⁰² Zr 2.9s β
⁸³ Y 100	⁹⁰ Y 2.67d β	⁹¹ Y 58.5d β	⁹² Y 3.64h β	⁹³ Y 10.2h β	⁹⁴ Y 16.7m β	⁹⁵ Y 1.3m β	⁹⁶ Y 1.3s β	⁹⁷ Y 3.76s β	⁹⁸ Y 0.59s β	⁹⁹ Y 1.47s β	¹⁰⁰ Y 6.73s β	¹⁰¹ Y 0.43s β
⁸⁸ Sr 82.58	⁸⁹ Ru 50.52d β	⁹⁰ Sr 28.8a β	⁹¹ Sr 9.5h β	⁹² Sr 2.11h β	⁹³ Sr 7.41m β	⁹⁴ Sr 1.25m β	⁹⁵ Sr 2.51s β	⁹⁶ Sr 1.07s β	⁹⁷ Sr 0.43s β	⁹⁸ Sr 0.65s β	⁹⁹ Sr 2.69ms β	¹⁰⁰ Sr 2.11ms β

Hanson, S.K.; Pollington, A.D.; Waidmann, C.R.; Kinman, W.S.; Wende, A.M.; Miller, J.L.; Berger, J.A.; Oldham, W.J.; Selby, H.D. *Proc. Nat. Acad. Sci.*, **2016**, *113*, 8104-8108.



Why Consider the Weapons Laboratories?



Images from Reuters.com

- **Relevance**
 - The world is not a safe place
 - The Laboratories' primary mission is Science in Defense of the Nation
 - Diverse programs, from Stewardship to Nuclear Forensics, work to this theme
- **Challenge**
 - No one comes out of school trained for this work
 - It is an honor to be trusted with ensuring our deterrent
- **Science**
 - The highest concentration of the best experimental facilities and scientists provide opportunity to explore unlike anywhere else

Acknowledgments

- **The Radiochemical Assessment Team (RATs)**
 - Amy S. Lee, August L. Keksis, Christopher R. Waidmann, Corey C. Keith, Kevin T. Bennett, Reginaldo C. Rocha, Bret L. Lockhart, Elyza J. Ortiz, Serena R. Rodriguez, Roger A. Meade, John A. Musgrave
- **DOE-NNSA Science Campaign 4.1**

Thank You for Your Interest!