A woman with long dark hair, wearing a black and white striped shirt and black gloves, is looking into a large, open cryogenic chamber. The chamber is filled with various scientific instruments, including a large cylindrical component and several smaller containers. The chamber is made of stainless steel and has a large handle on the door. The background shows more of the laboratory equipment, including a control panel with a red cable and a black handle.

# Short Introduction to Cryo-Pumps and Refrigerators

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## Topics

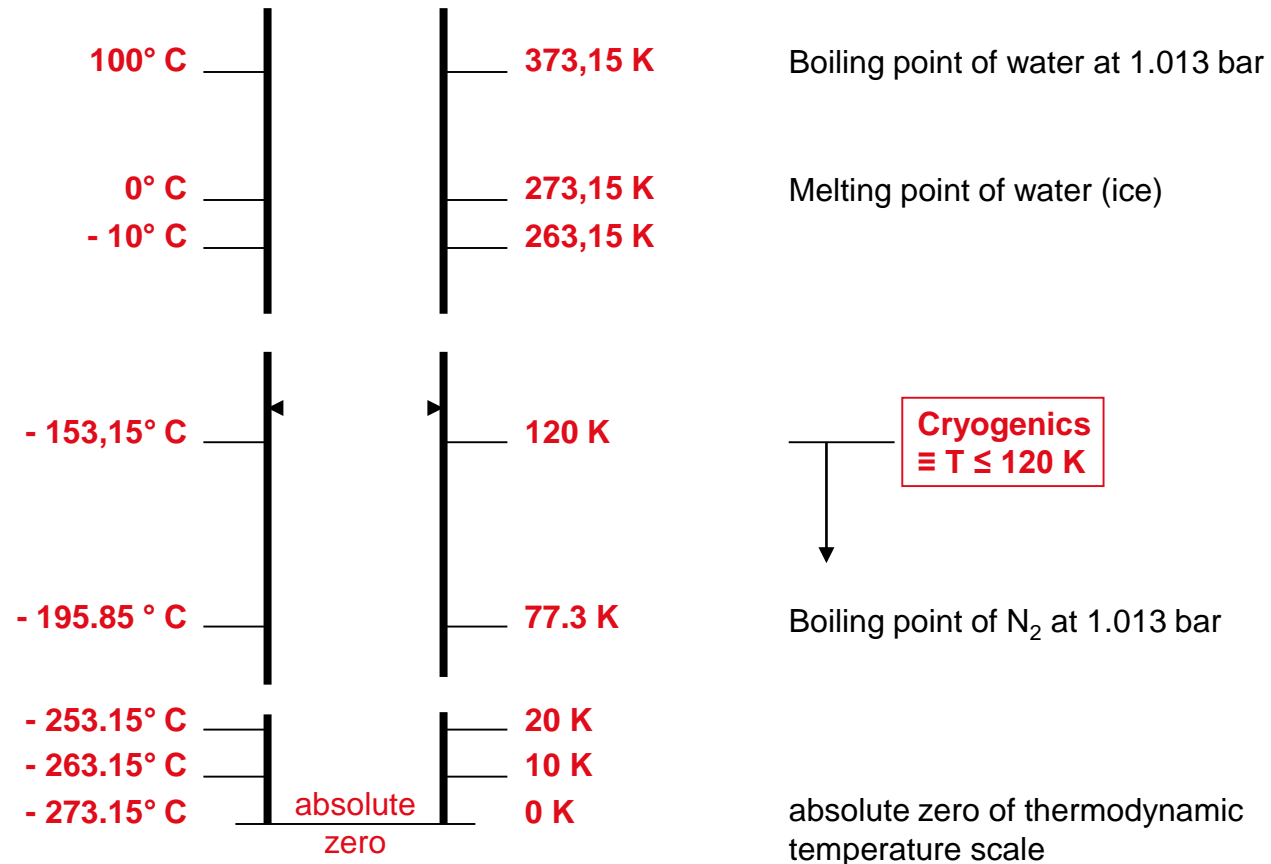
- Physical principles of cryo-pumping
- Design and control of modern refrigerator cryo-pumps
- UHV and XHV cryopumps
- Advantages – Disadvantages of cryo-pumps
- Typical applications in industry and research

# Cryogenics – Temperature Scales

Celsius Temperature Scale (t)  
Degrees Celsius (°C)

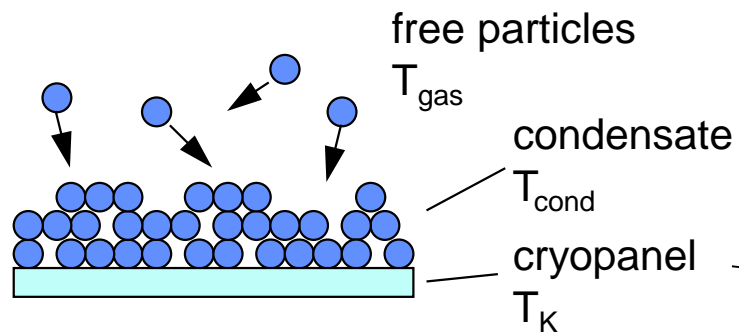
Thermodynamic  
Temperature (T)  
(Degrees) Kelvin (K)

$$\begin{aligned} \text{formula: } T \text{ [K]} &= (t \text{ [}^\circ\text{C]} + 273.15) \text{ K} \\ t \text{ [}^\circ\text{C]} &= (T \text{ [K]} - 273.15) \text{ }^\circ\text{C} \end{aligned}$$



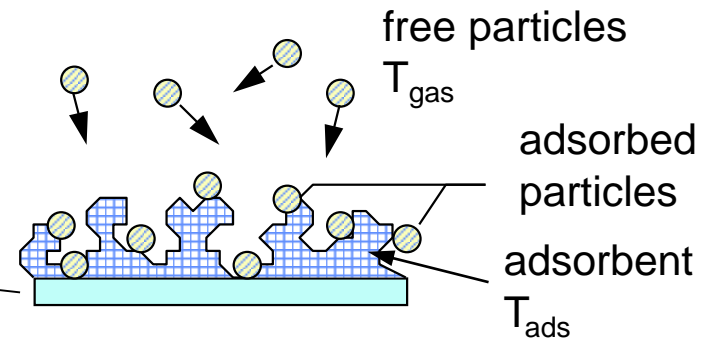
# Physical Effects Active In A Cryopump

## 1. Condensation



several layers of molecules, thickness up to > 10 nm example:  $N_2$ , Ar,  $H_2O$

## 2. Adsorption



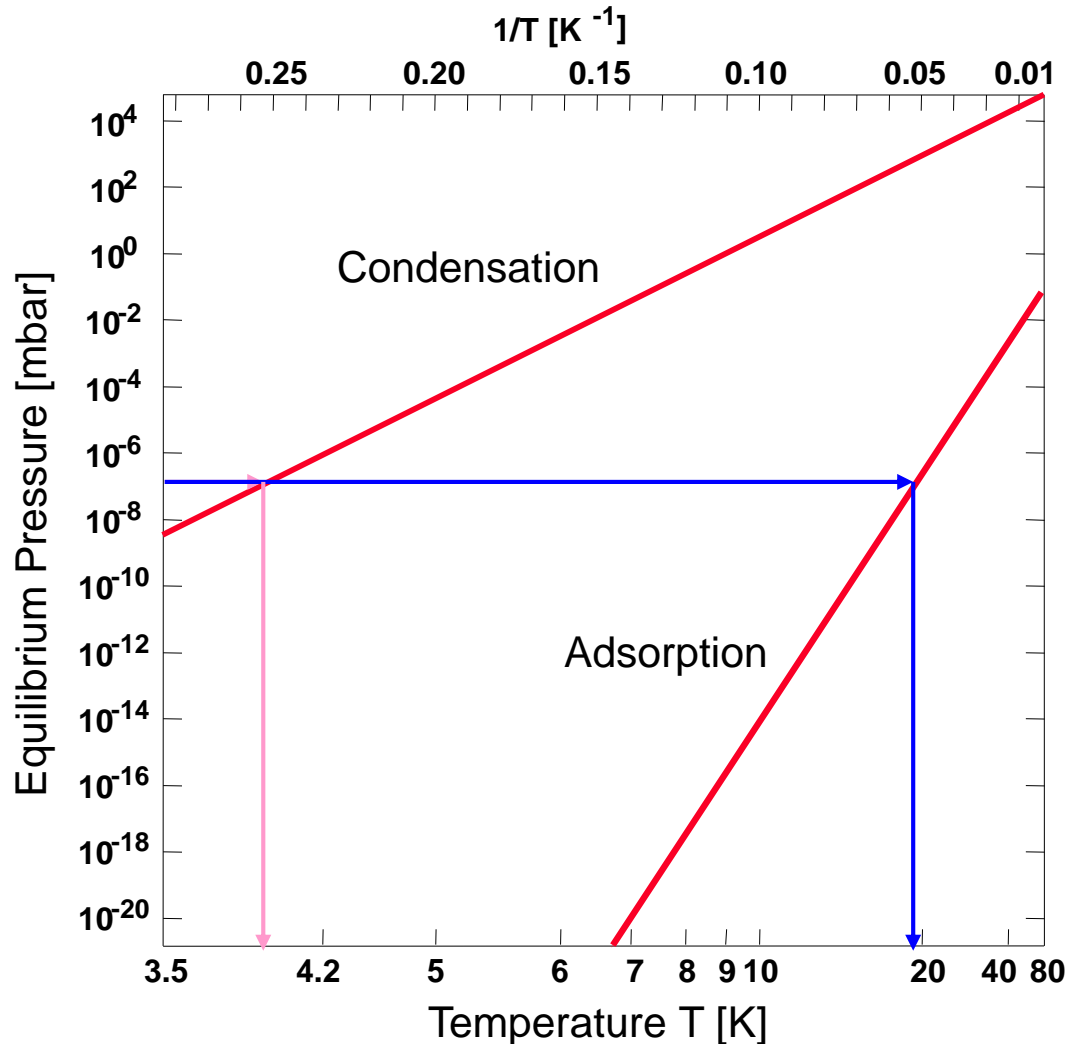
only monolayers example:  $H_2$ , He, Ne on charcoal

- In a cryopump, all gases are pumped by condensation at low temperatures
- The low temperatures are provided by a 2 stage cooling machine, the 'cryo-refrigerator'

The temperatures required to pump all gasses in high vacuum are explained on the next slides

# Equilibrium pressure for condensation

## And adsorption of H<sub>2</sub> on charcoal



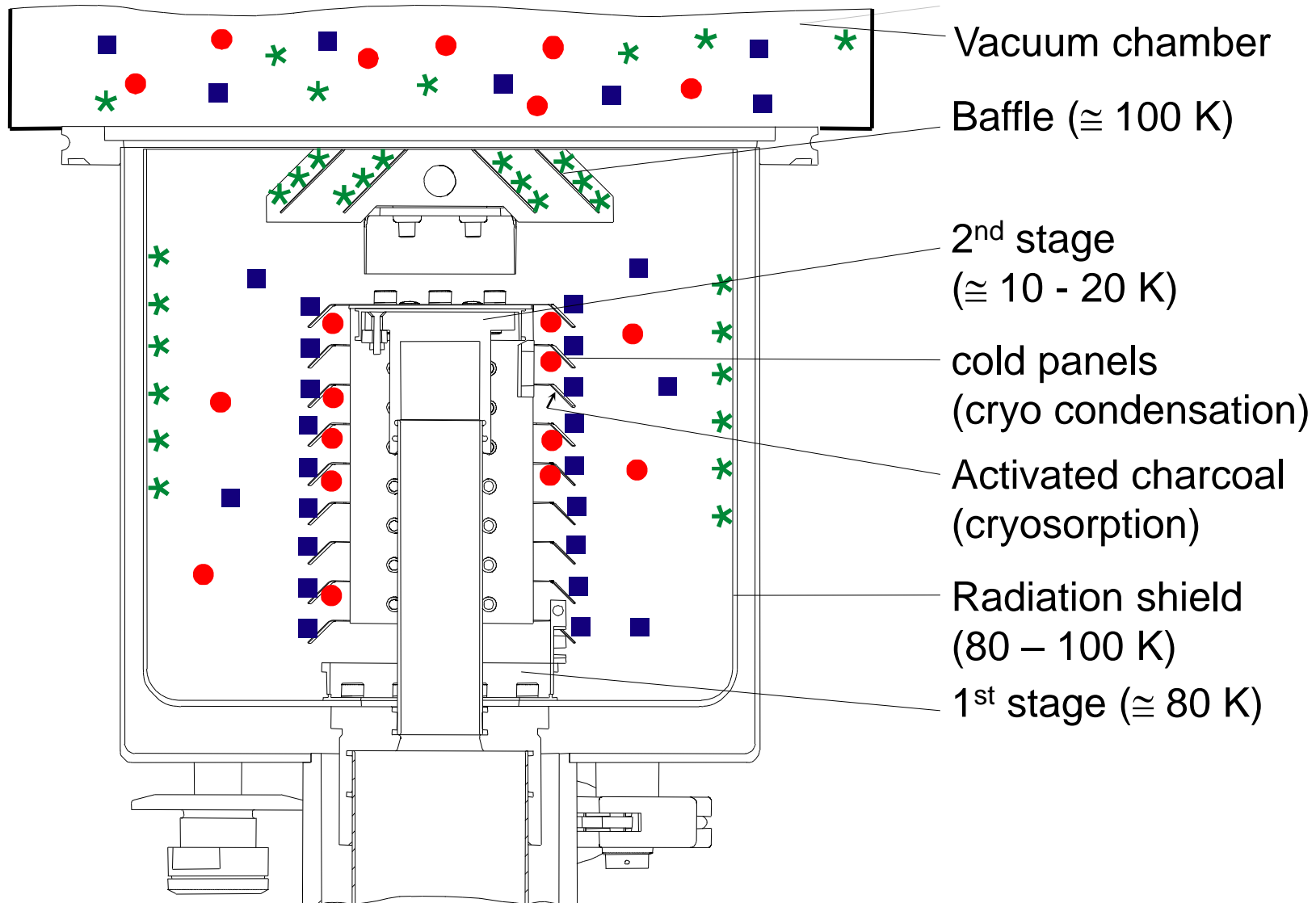
### Example: H<sub>2</sub>

To reach partial pressures in the high vacuum regime by condensation, temperatures of <4 K are required.

The same pressure regime can be reached much easier by cryo-sorption on activated charcoal at significantly higher temperatures

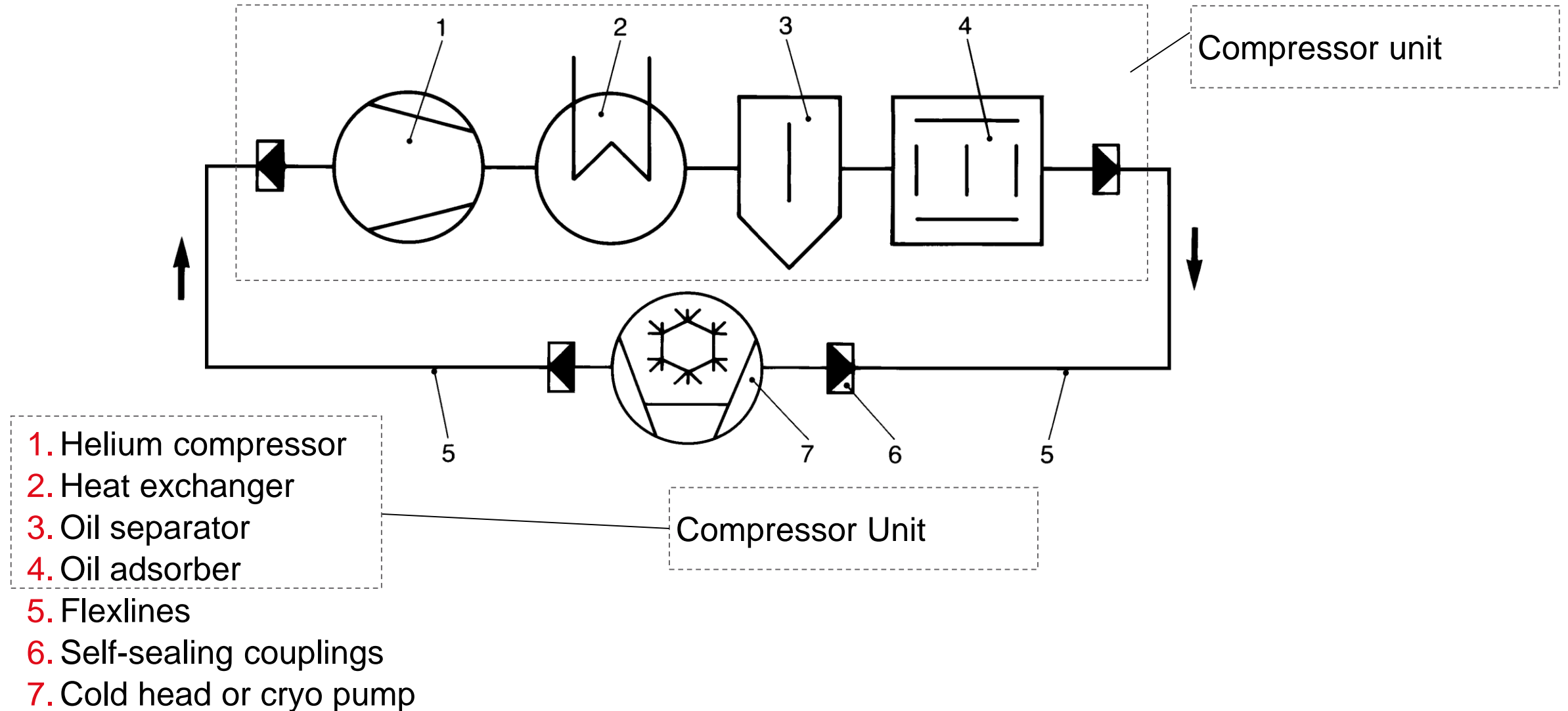
Temperatures which can be reached with standard cryo-refrigerators: ≤10 K

# Cryo-Pump – Schematic



- \* gases condensated at high temperatures ( $H_2O$ ,  $CO_2$ )
- gases condensated at low temperatures ( $N_2$ , Ar,  $O_2$ )
- gases adsorbed on activated charcoal ( $H_2$ , He, Ne)

# Gifford McMahon Refrigerator System – Simplified Schematic For Helium Circuit



## Starting pressure (mbar)

- Basically a start at atmosphere would be possible, but the heat load for the cold panels would be too high and the pump panels would be coated by frozen air (water vapour and nitrogen), which disables them to pump e.g. hydrogen to reach low ultimate pressure
- **If pressure in the lower UHV regime need to be reached, it is recommended to start cryo-pumps at pressures  $< 10^{-3}$  mbar.**
- Else evacuation of the cryopump to typically 0.1 ... 0.05 mbar is sufficient before starting the cooldown.



How does it look like in the cryopump, when it is loaded with gas, and when it needs to be regenerated?



- **Cooldown time**

Time to cool the internal parts by the refrigerator (60 – 180 minutes)

- **Regeneration interval**

Operation time for an individual gas until the pump is saturated and has to be regenerated (warm to room temperature to evaporate the condensate, evacuate the cryopump with a roughing pump and cool down again)

pump operated at  $10^{-08}$  mbar: approx. 300 years

$10^{-06}$  mbar: approx. 3 years

$10^{-04}$  mbar: approx. 10 days

- **Cryopumps** have higher pumping speeds than TMP or diffusion pumps with comparable flanges sizes (4 times higher for H<sub>2</sub>O!)

**The cryopump is ideal for low pressure applications where large pumping speeds especially for water vapour are required**

## Classic cryopumps

Ultimate pressure limited by

- Restricted baking possibilities of chamber
- Leak rate of O-rings of safety valve, feed-throughs etc.
- Degassing from wiring of diode, heaters

$$\Rightarrow P_{\text{ult}} < 1 \times 10^{-9} \text{ mbar}$$

## UHV cryopumps

Measures: all metal-sealed (rupture disk, feed-through, special diode)

Ultimate Pressure still limited by

- Unbaked cryopump
- Limited baking possibilities of chamber

$$\Rightarrow P_{\text{ult}} = 1 - 3 \times 10^{-11} \text{ mbar}$$

## Special solution suitable to reach XHV:

Cryopump with additional LN2-cooling during baking

- Allows baking of chamber and pump housing to >180 C

- During bakeout

$$P_{\text{ult}} 1 \times 10^{-9} \dots 1 \times 10^{-10} \text{ mbar}$$

- After bakeout

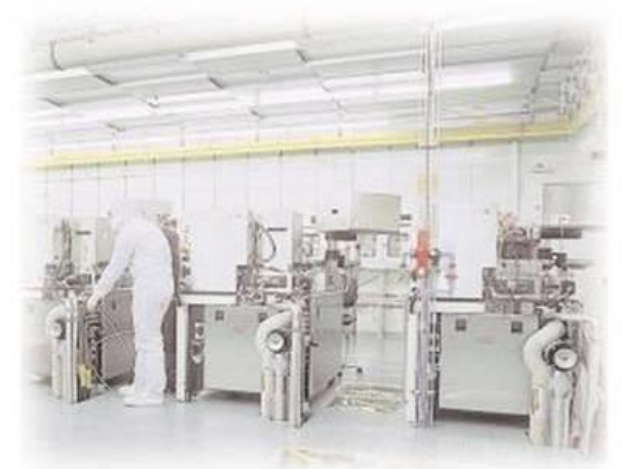
$$P_{\text{ult}} < 1 \times 10^{-12} \text{ mbar}$$

Cryopump is used during bakeout

⇒ much more effective to pump water vapour

## Semiconductor Industry

- Sputtering (PVD) Systems
- Ion Implanters
- Transfer Chambers
- Load Lock Chambers



## Coating

### Precision Optics



- Lenses
- Optical Filters
- Mirrors
- Displays

### Electronics



- Hard Disks
- Magneto-optical (MO) Disks
- Magnetic Heads

## Process Industry



- Electron Beam Welding
- Furnace

## Research and Development



- Space Simulation Chambers
- Big Testing Chambers

A large, industrial-grade stainless steel chamber with its door open, revealing a complex internal structure with various ports and a perforated floor. A dark grey semi-transparent banner is overlaid across the middle of the image, containing the text 'Thank you for your attention!' in a bold, red, sans-serif font.

**Thank you for your attention!**