

Synchrotron X-ray Powder Diffraction

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Outline

Diffraction – recap

X-ray crystallography

Single crystal diffraction

Powder diffraction

Conventional X-ray diffraction

Sources and limitations

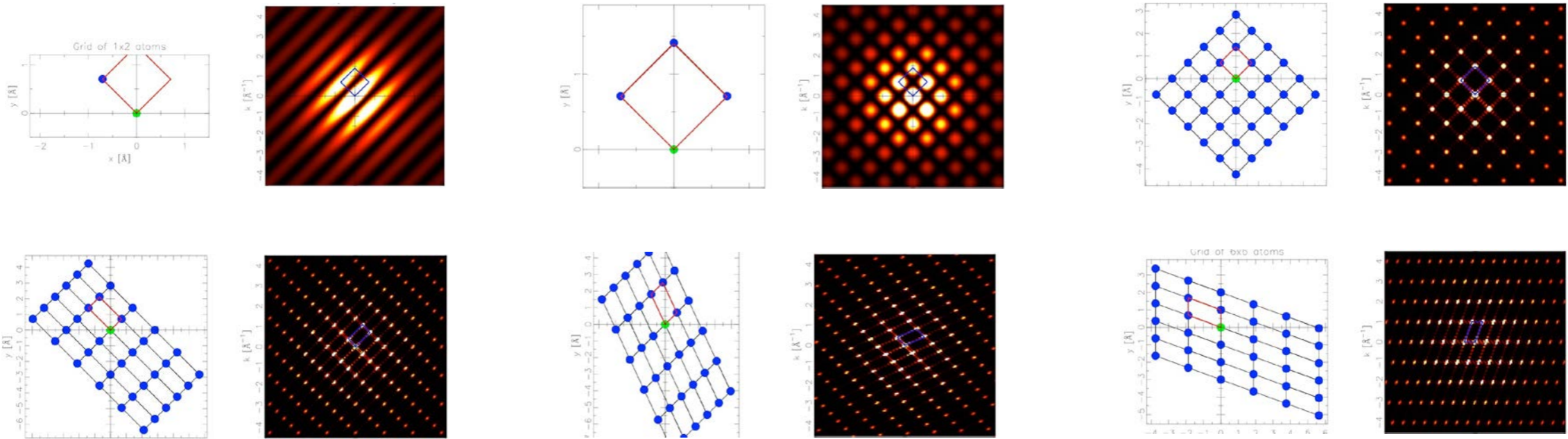
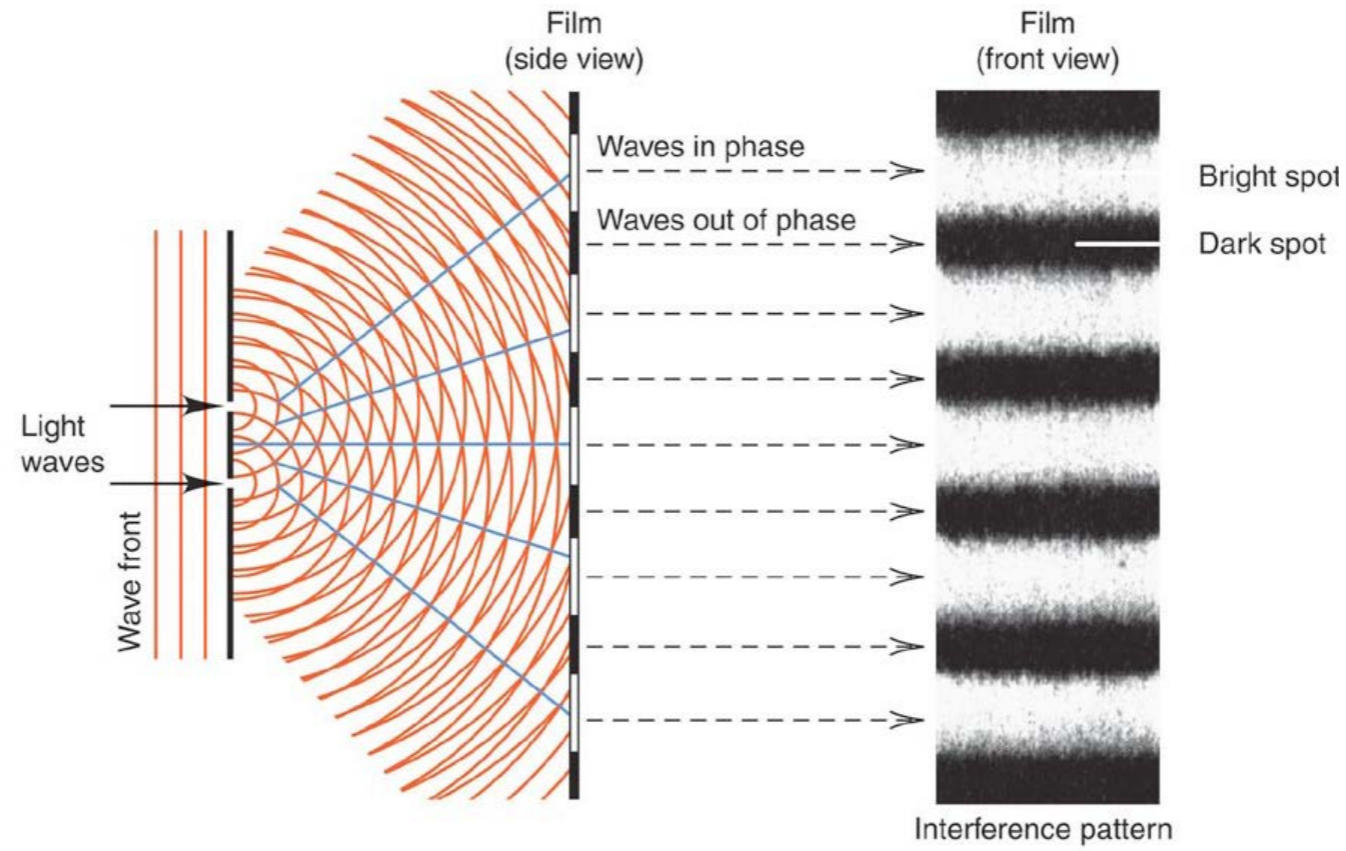
Synchrotron X-ray diffraction

Advantages and bonus features

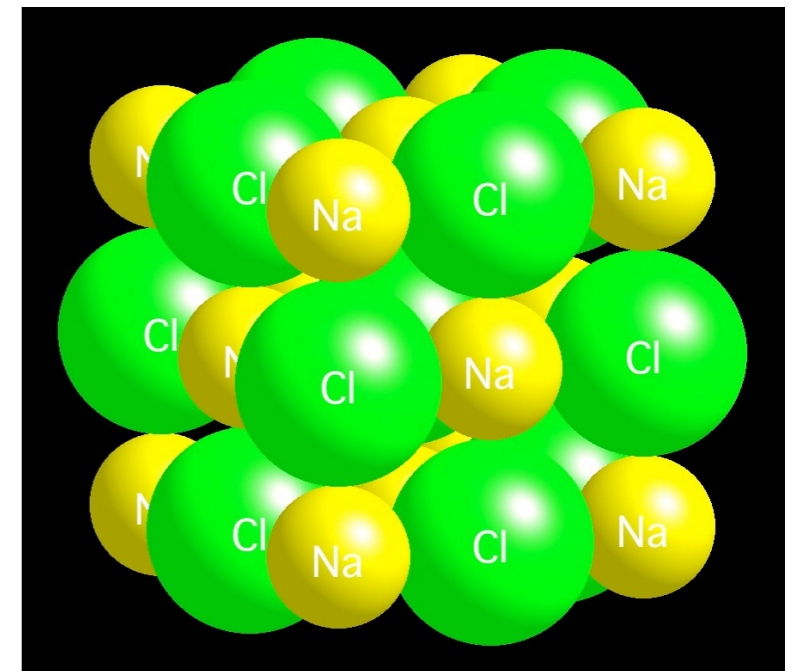
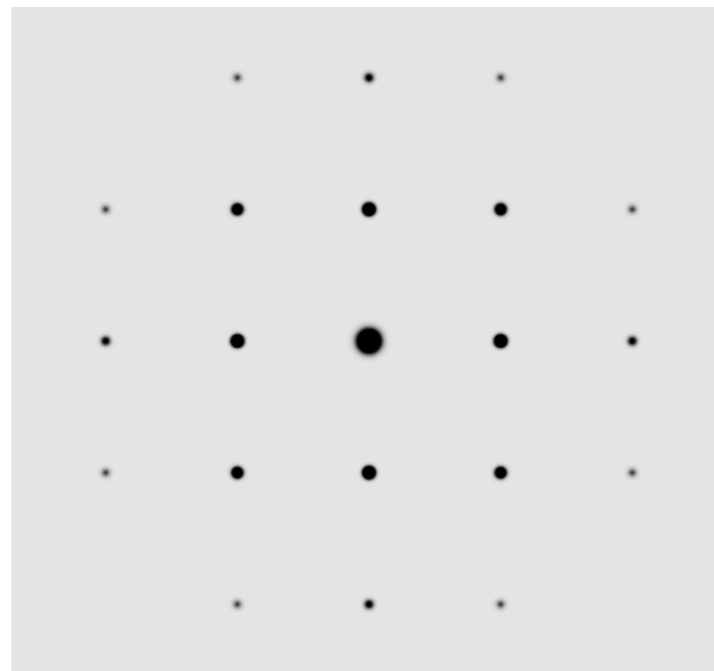
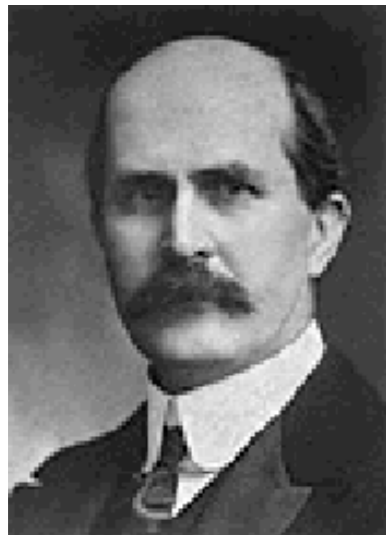
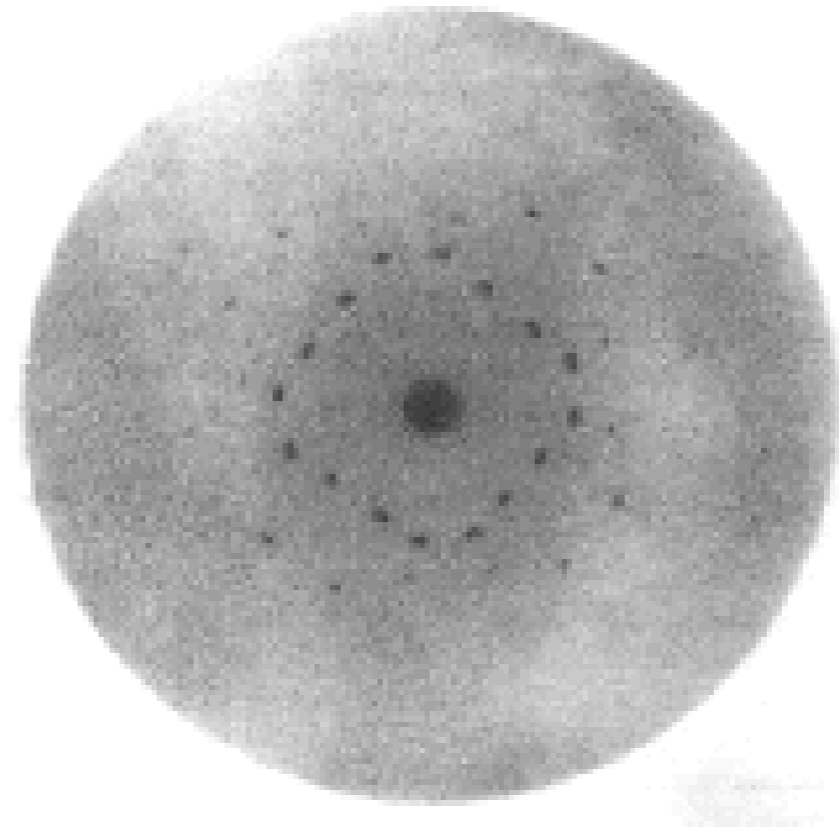
Diffraction

In three slides

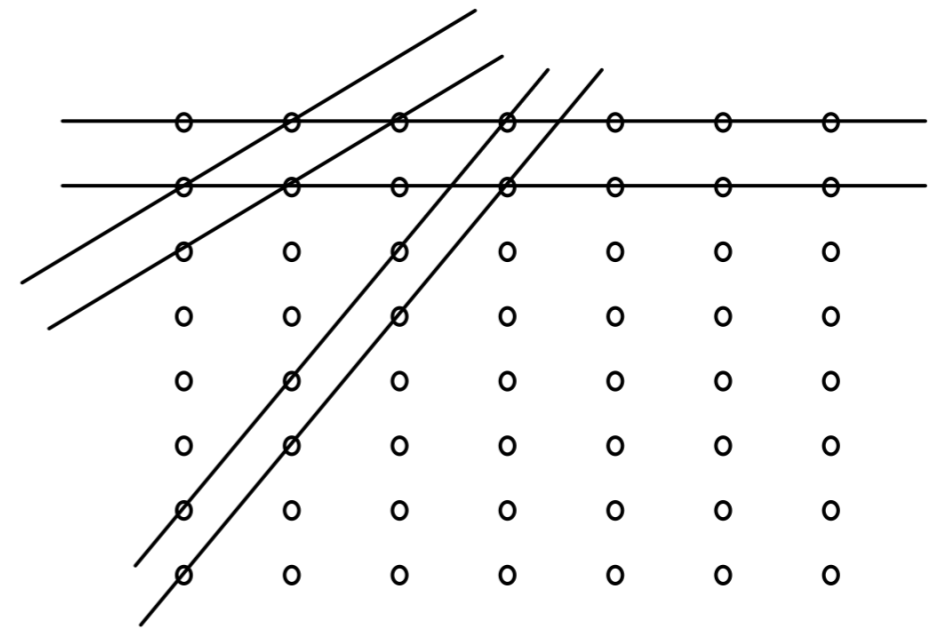
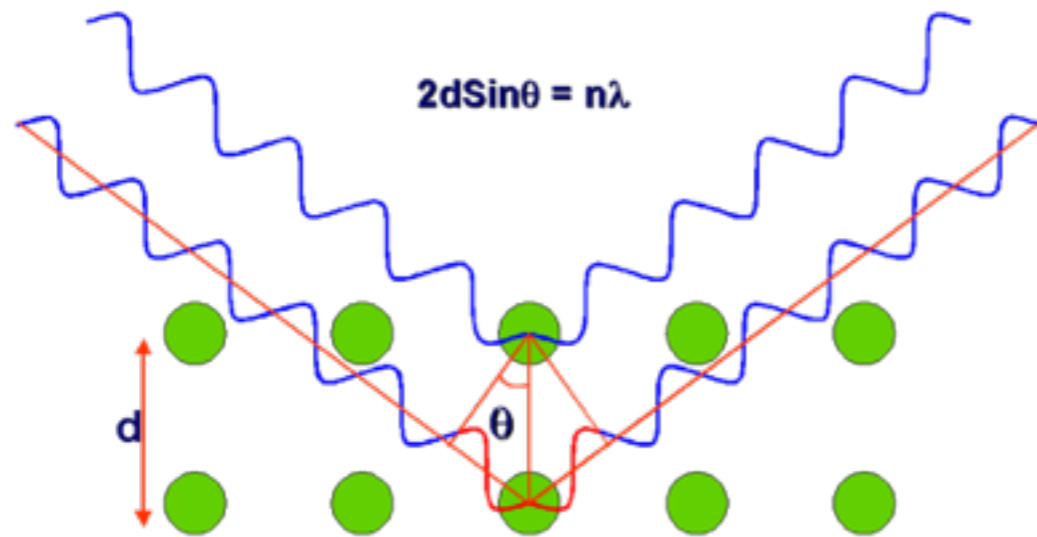
Wave diffraction



Crystals, lattice planes and X-rays



Bragg's law and crystallography



Note that the sets of lattice planes that are closer together have smaller d -spacings and therefore higher 2θ angles.

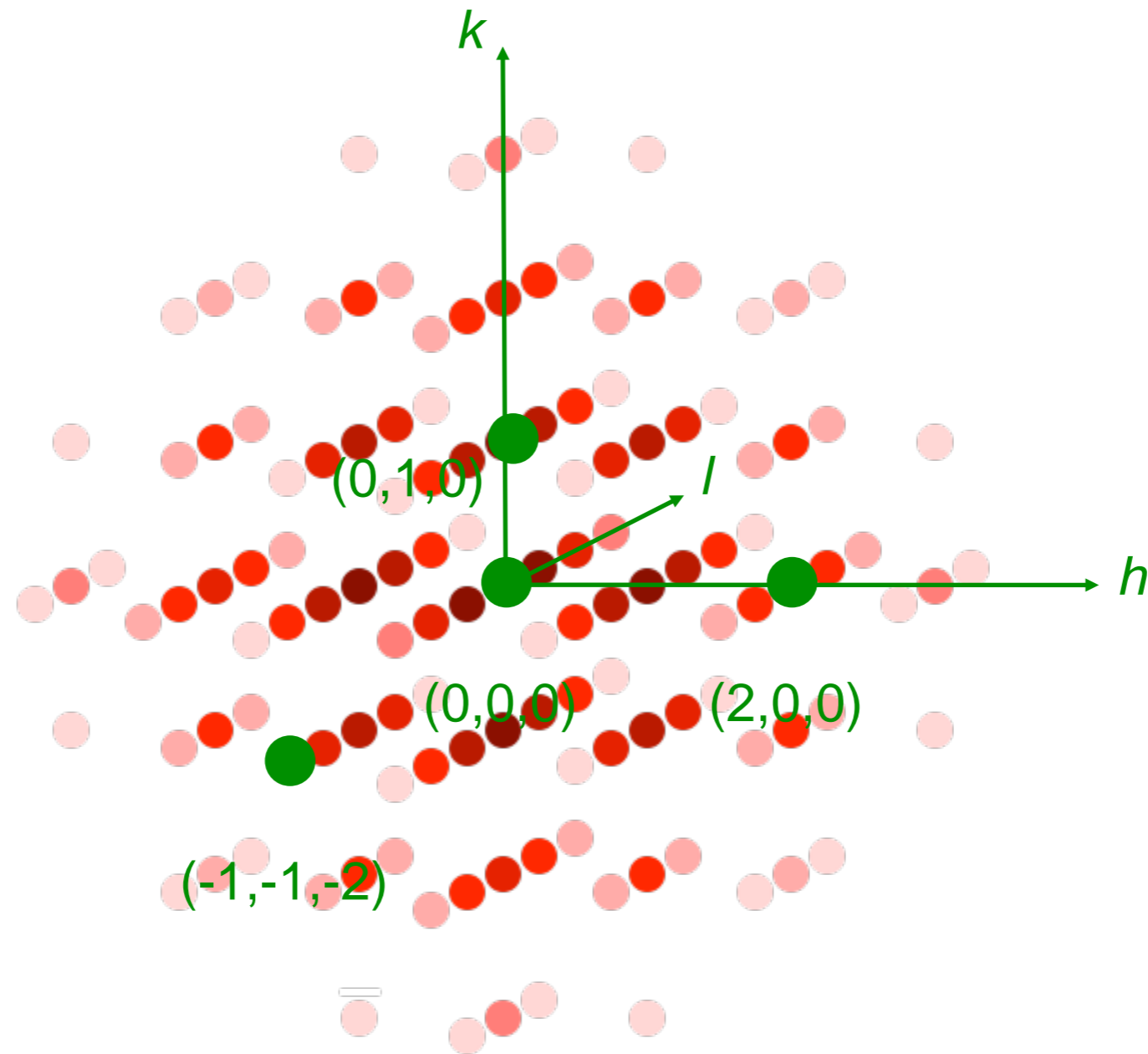
These planes produce less coherent diffraction, because the *size* of the atoms (hence the uncertainty in their positions) becomes more significant relative to the interlayer spacing d . Therefore, reflections get weaker as 2θ angle increases.

X-ray crystallography

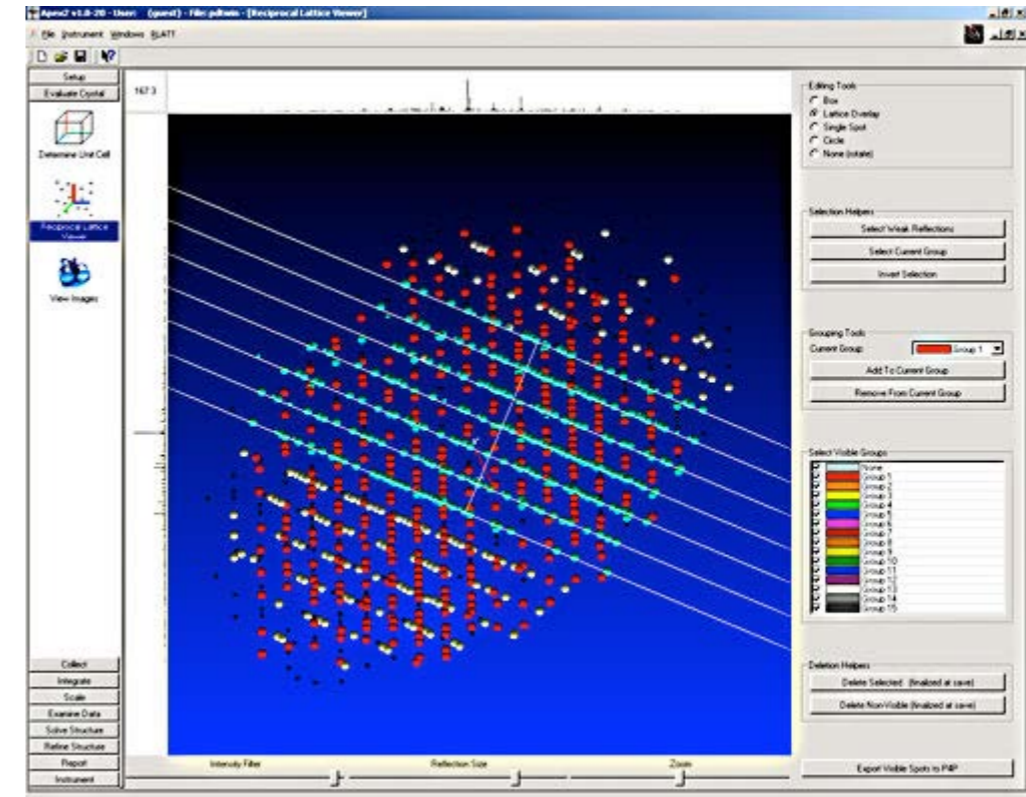
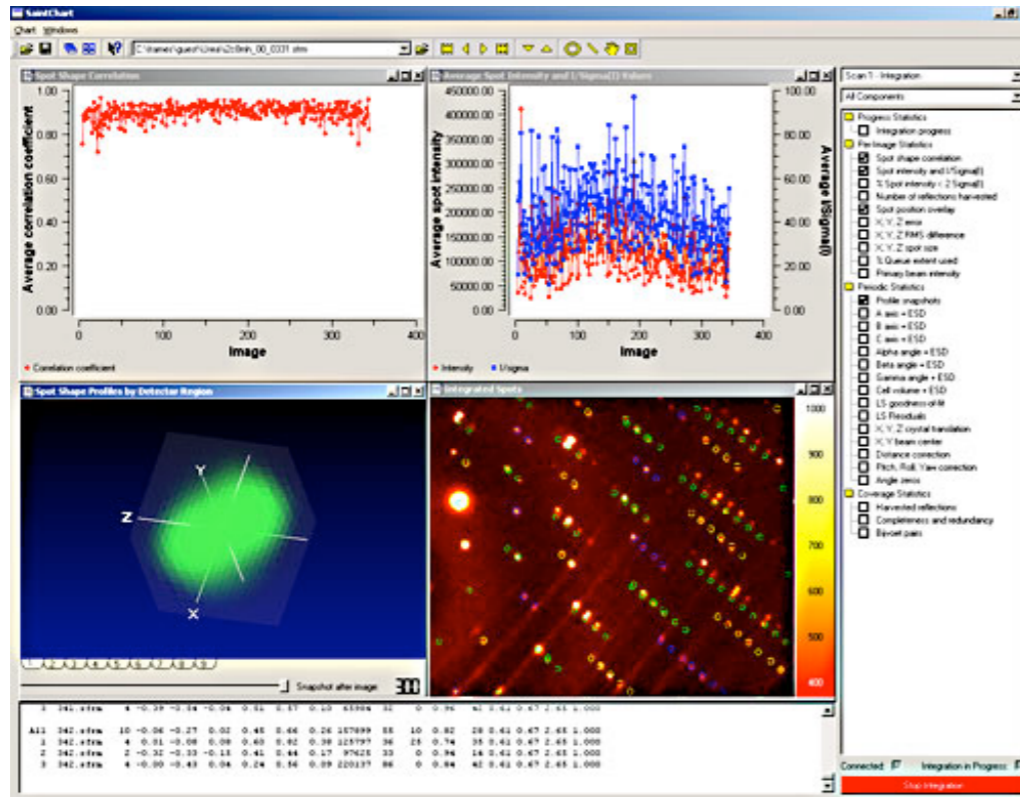
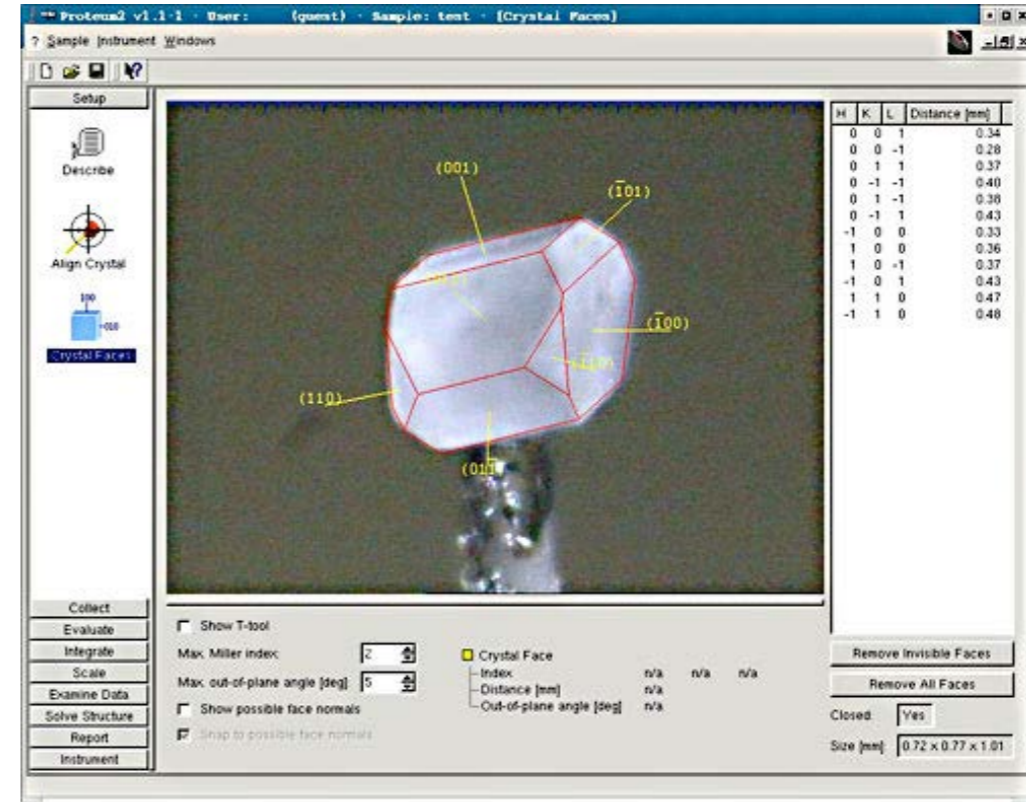
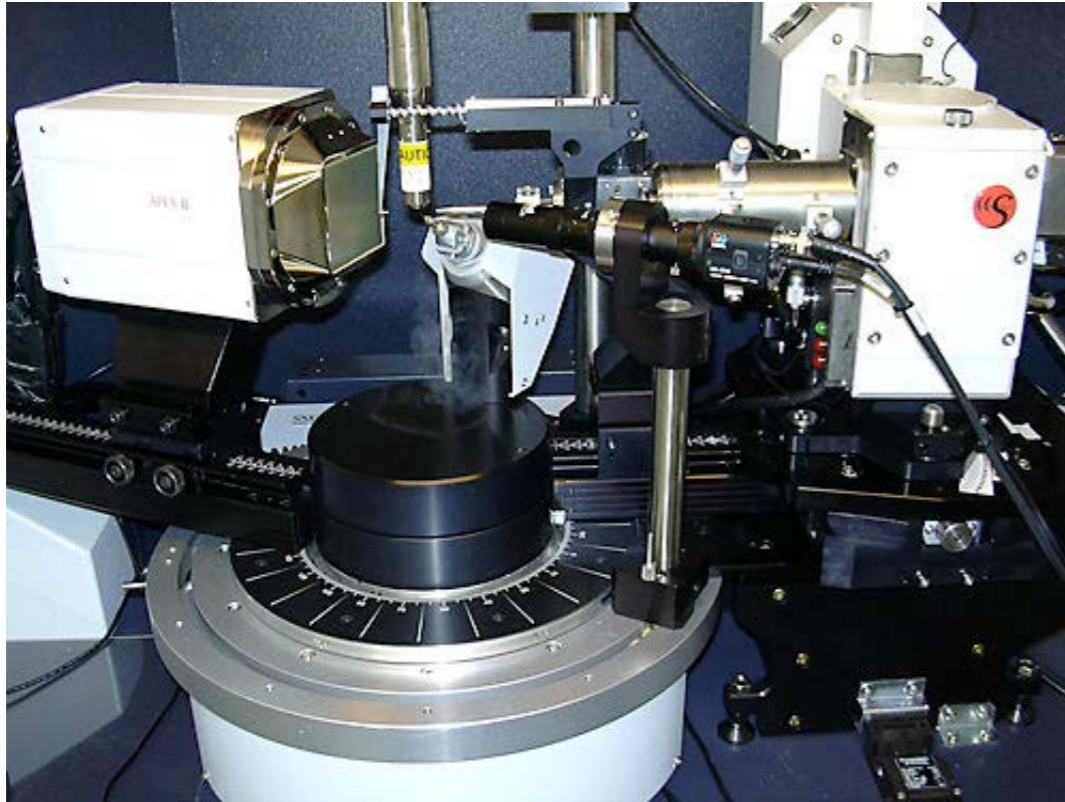
*Modern laboratory instruments and techniques for
crystal structure determination*

Single crystal diffraction

3-D reciprocal lattice



Single crystal diffraction

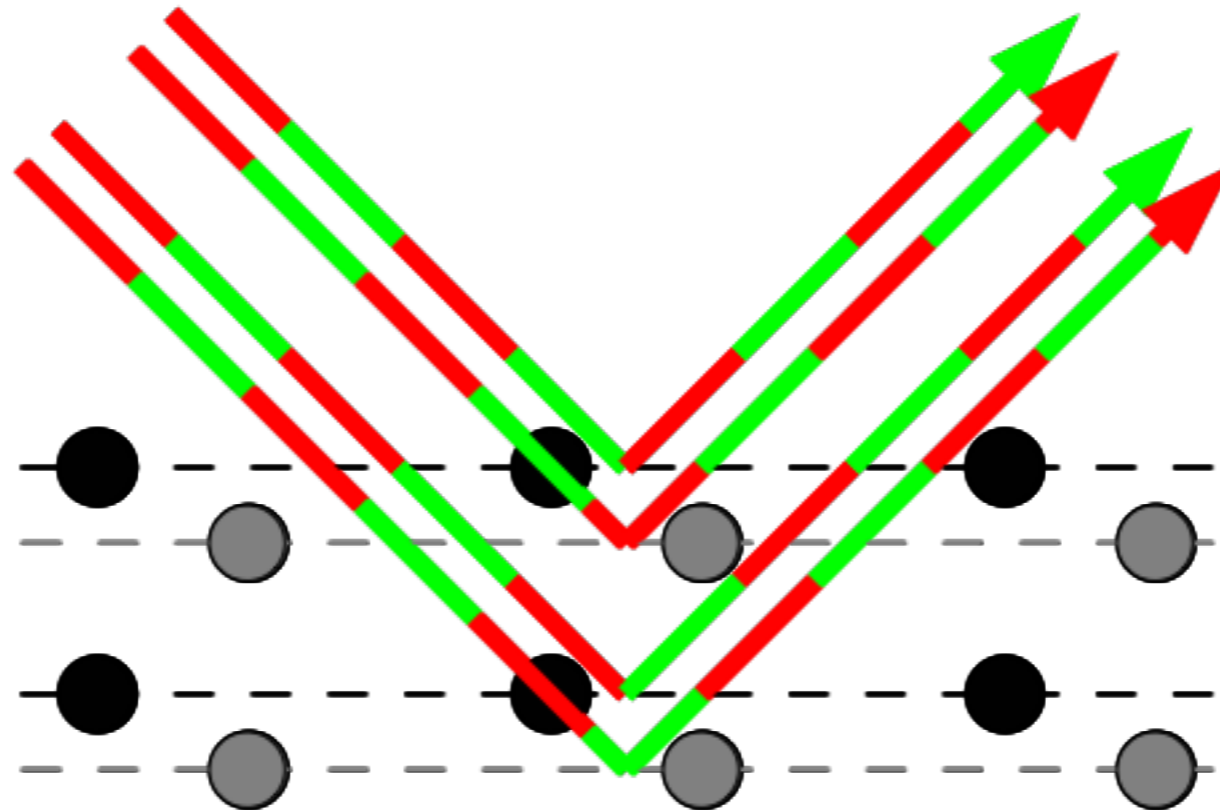


Structure solution and refinement

Diffraction as a Fourier transform:

$$\rho(x, y, z) = \frac{1}{V_c} \sum_{hkl=-\infty}^{\infty} \sum_{hkl=-\infty}^{\infty} F(h, k, l) e^{-i2\pi\{hx+ky+lz\}} = \frac{2}{V_c} \sum_{h=0,kl=-\infty}^{\infty} \sum_{h=0,kl=-\infty}^{\infty} |F(h, k, l)| \cos(2\pi\{hx + ky + lz\} - \alpha_{hkl})$$

The phase problem:



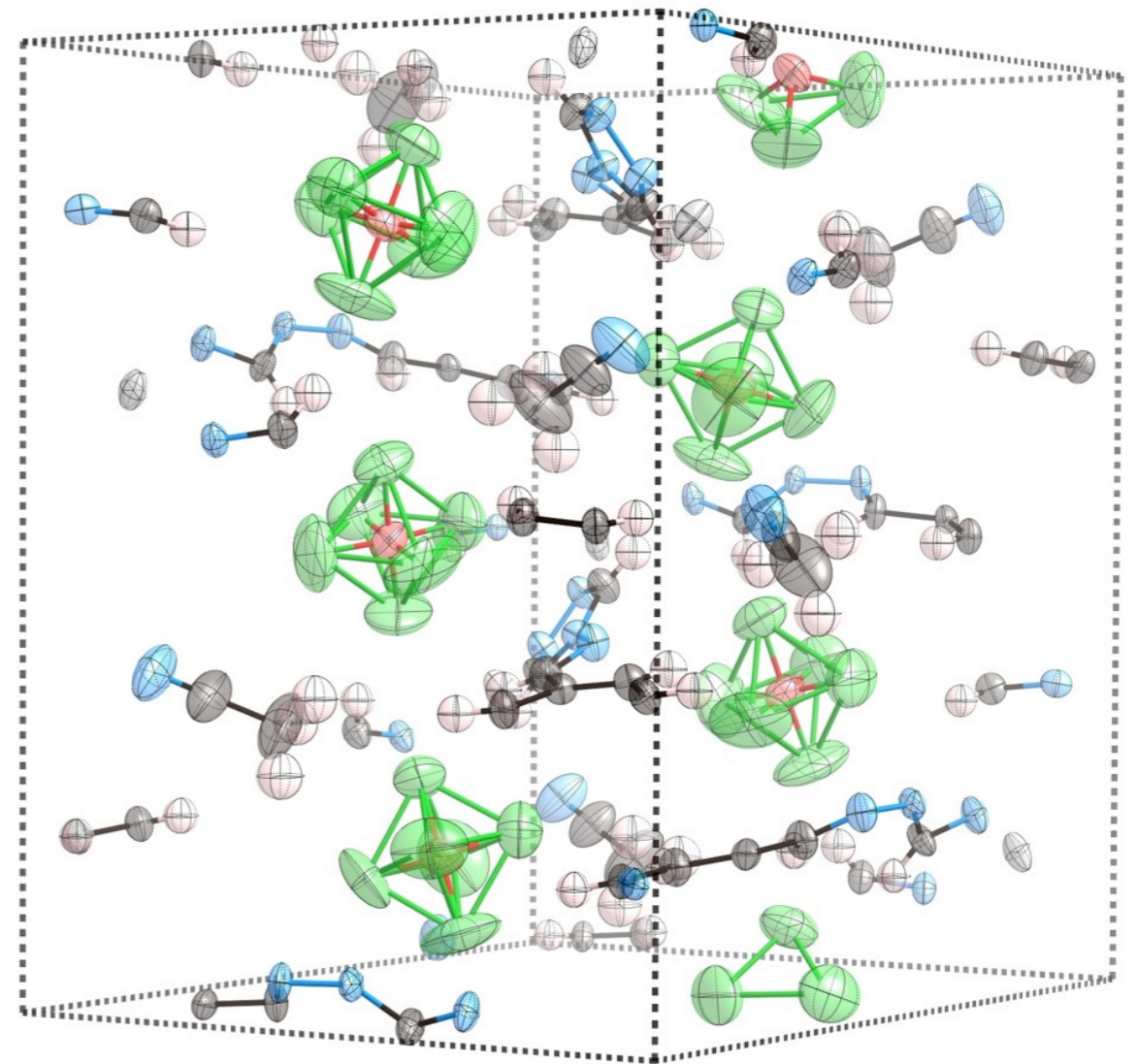
Structure solution and refinement

After applying various number-crunching statistical methods we expect to be able to solve and refine:

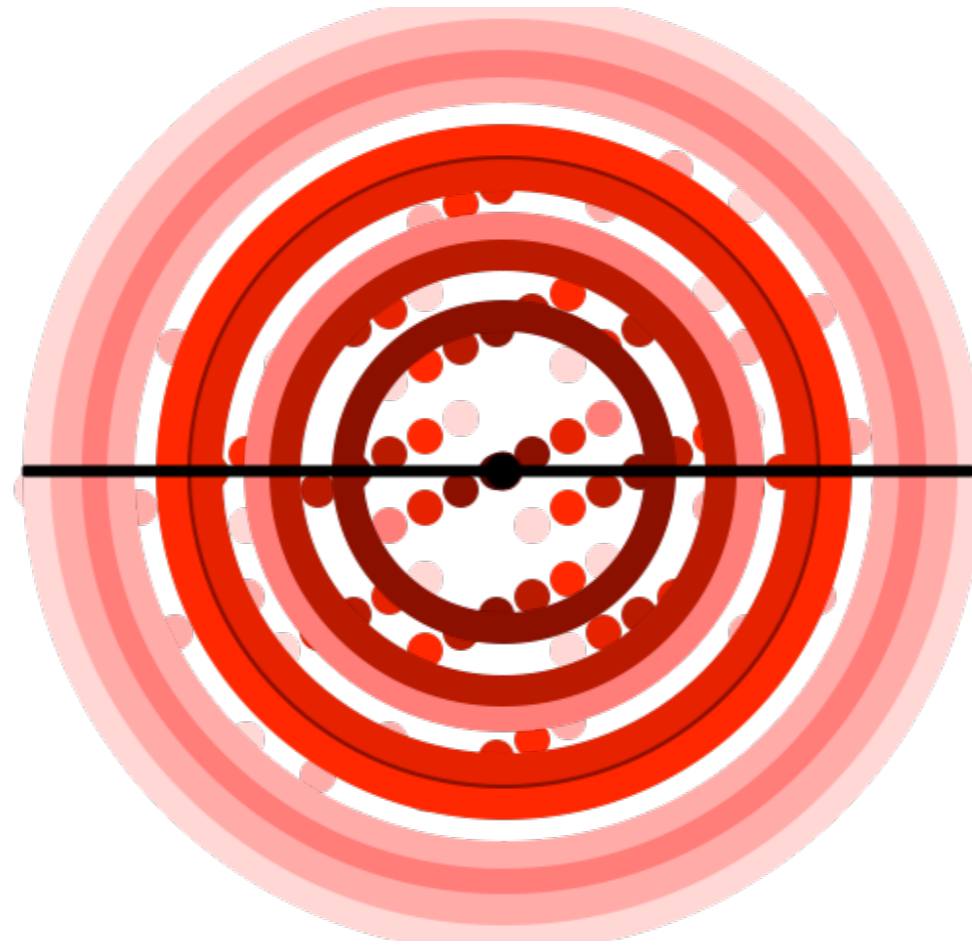
Unit cell and space group symmetry

Atomic positions

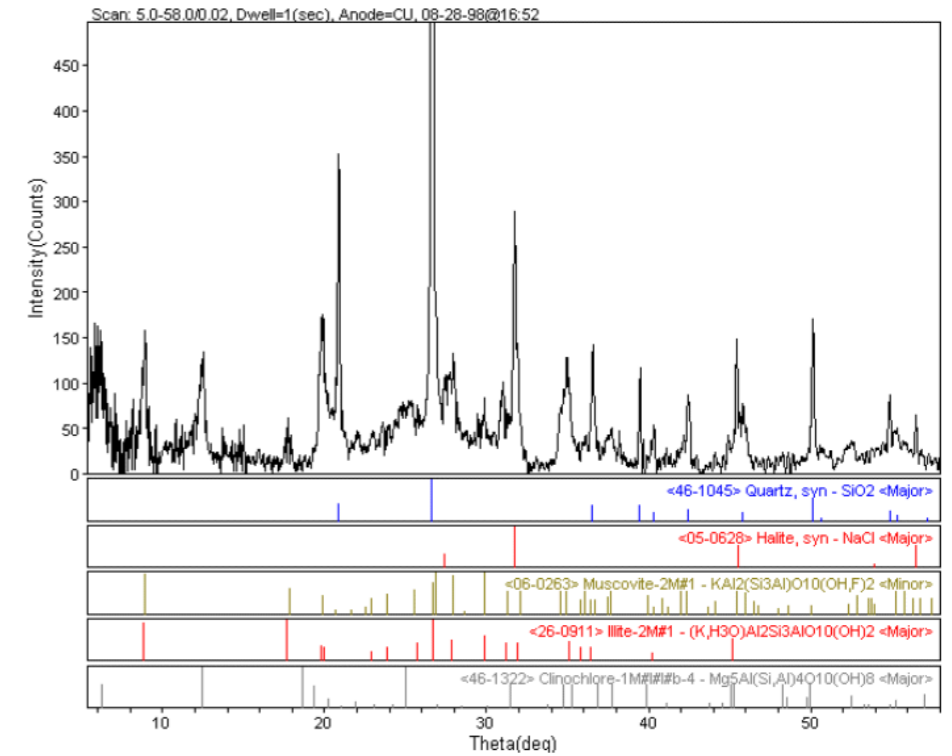
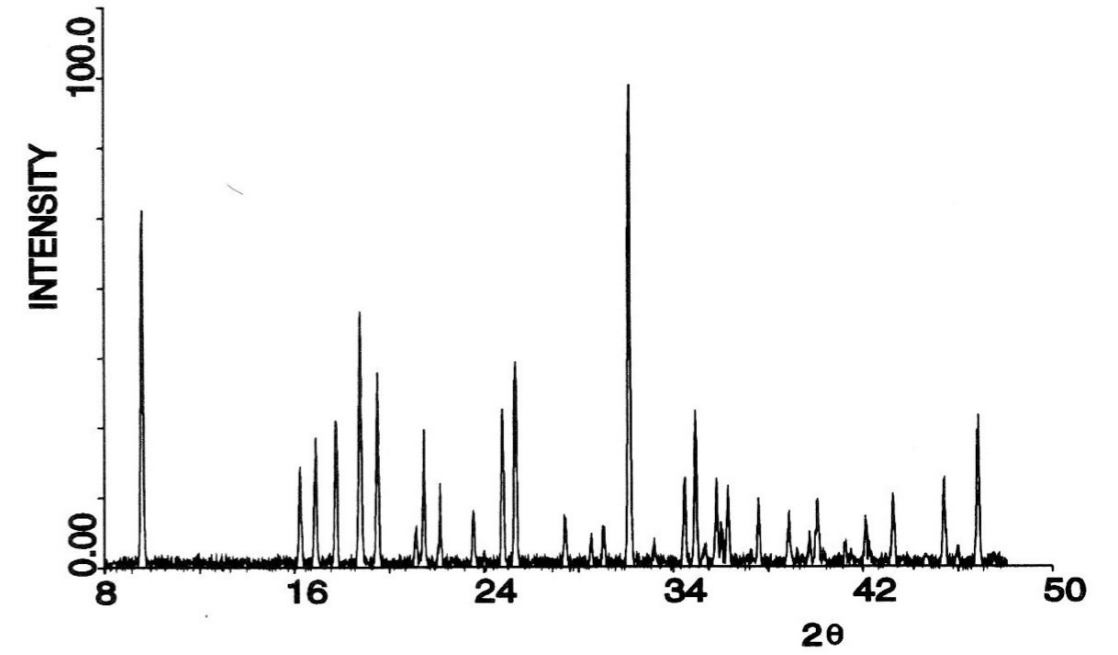
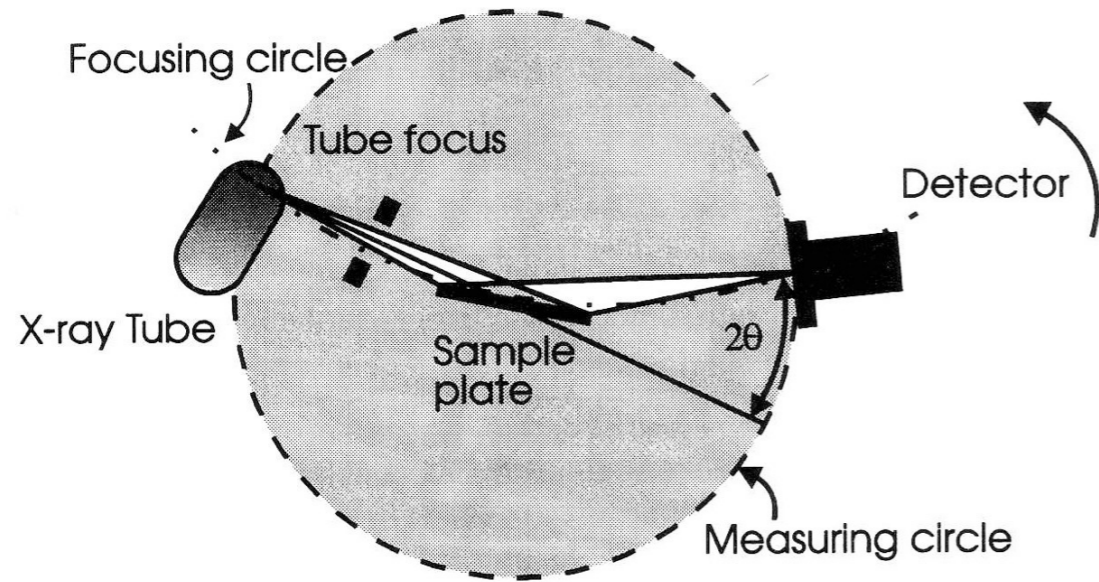
Anisotropic atomic displacements



Powder diffraction



Powder diffraction



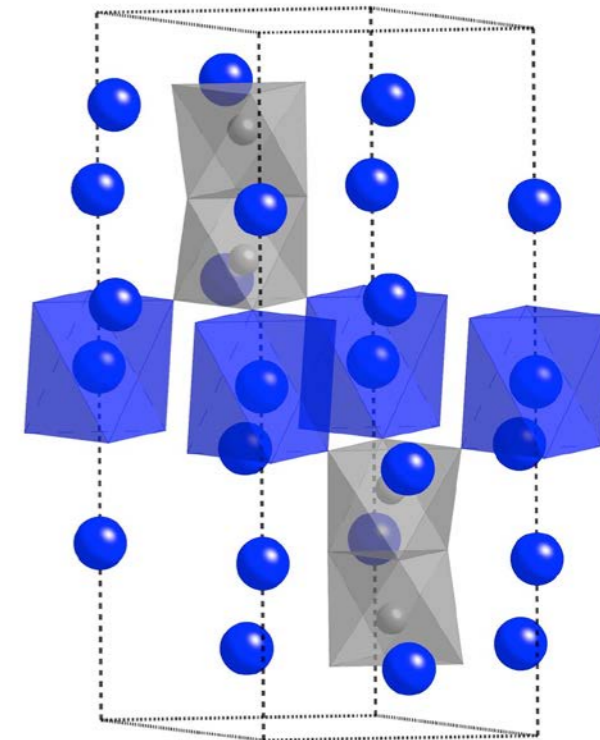
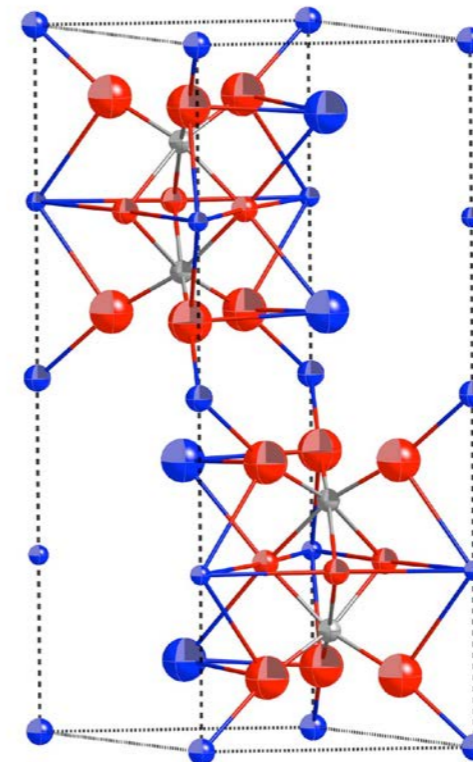
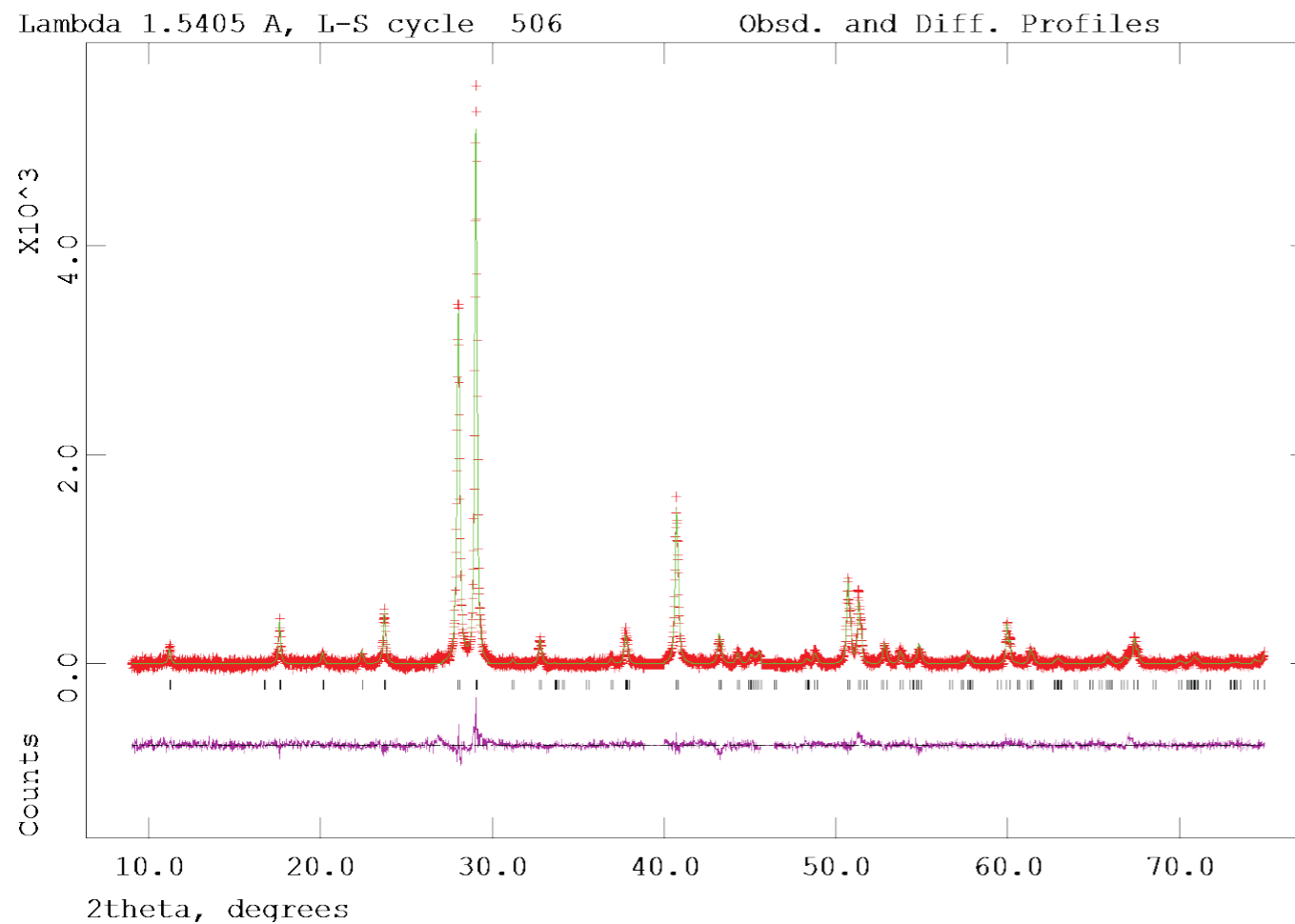
Structure solution and refinement

With full-pattern “Rietveld” methods we expect to refine:

Unit cell and space group symmetry

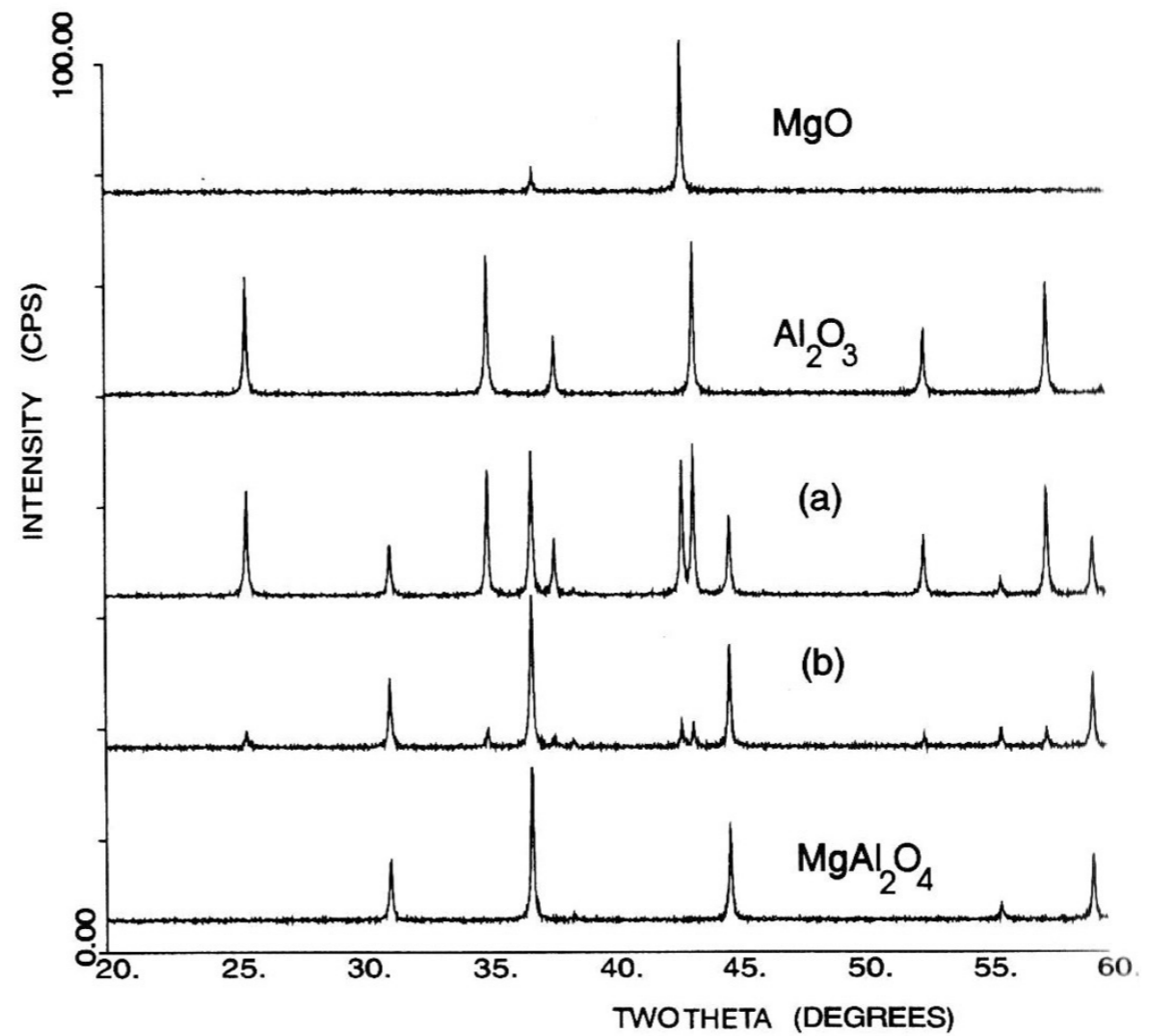
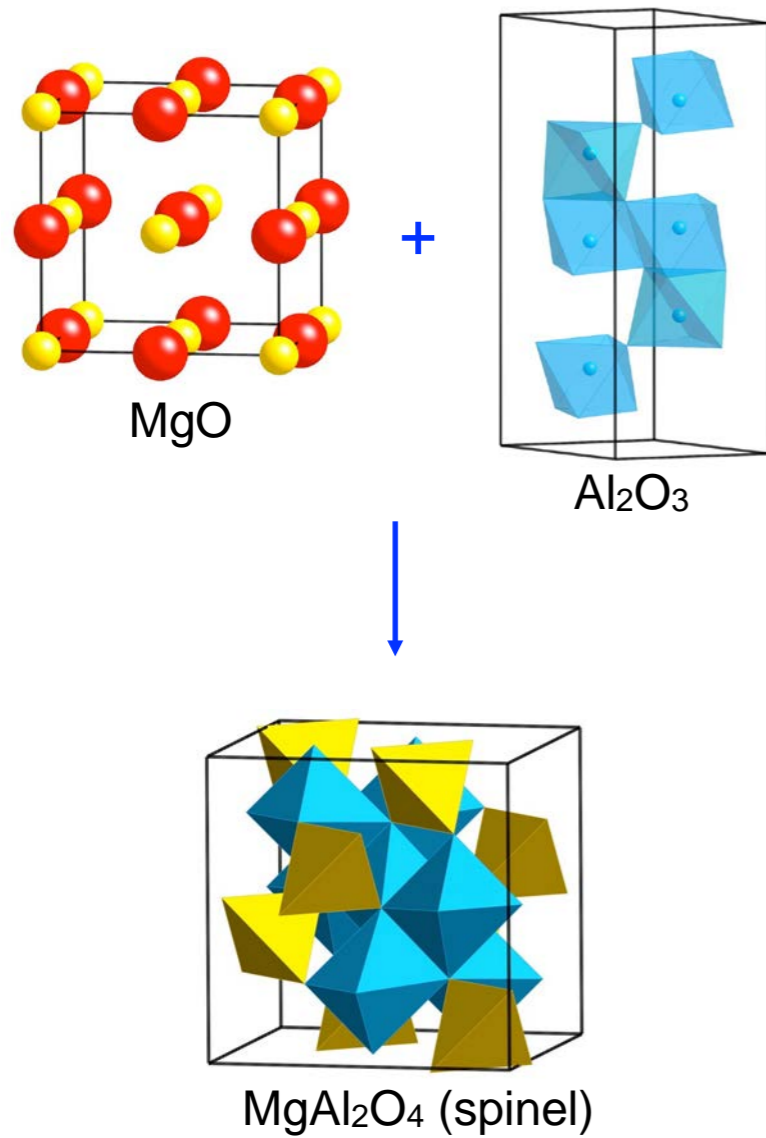
Atomic positions

Isotropic atomic displacement parameters



Phase identification

Tracking the progress of reactions



Reagents

1st firing

2nd firing

3rd firing

Conventional X-ray diffraction

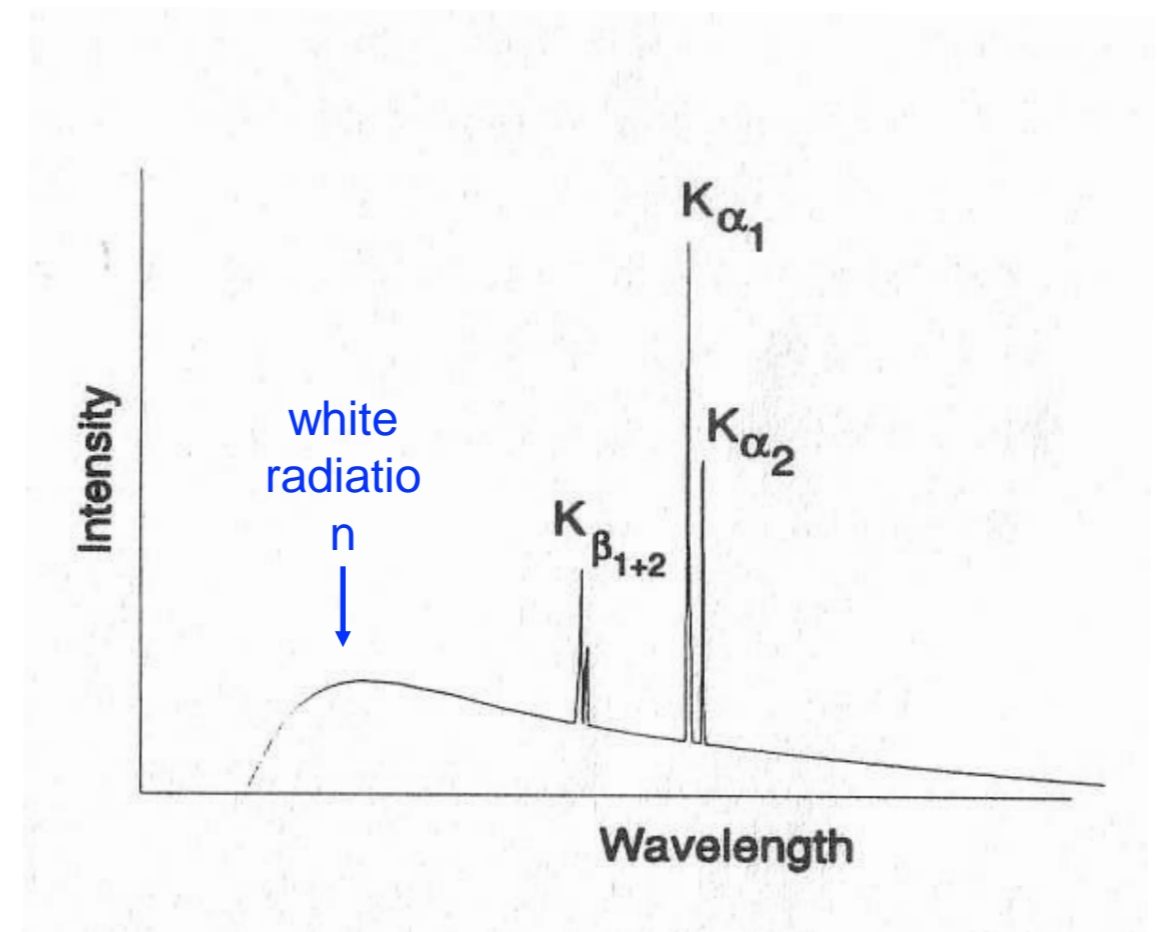
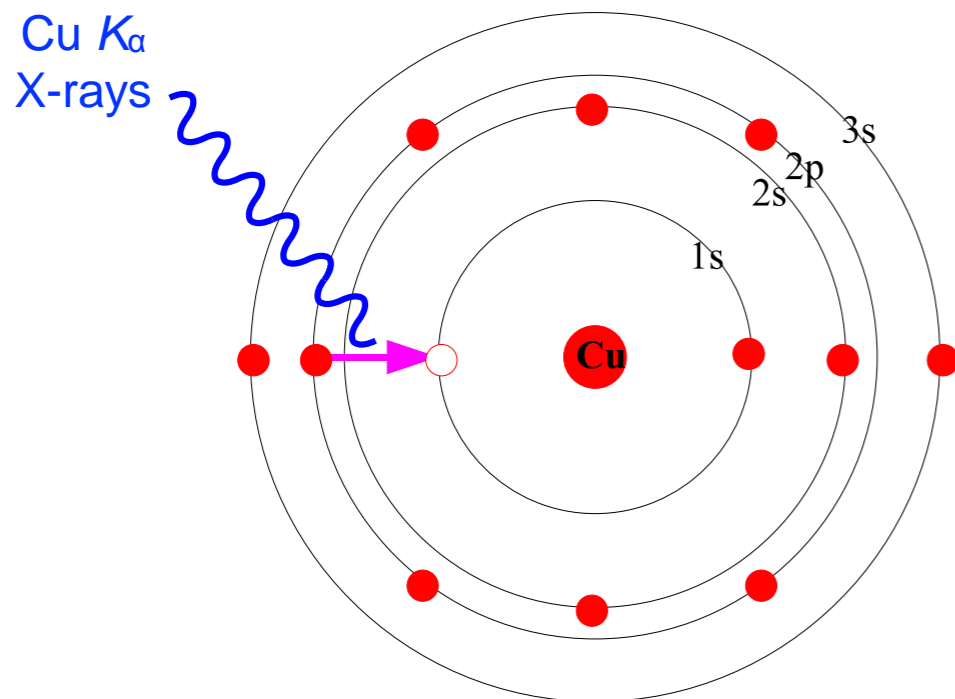
Cheap, convenient and user-friendly

Laboratory X-ray sources

Laboratory XRD instruments generate X-rays by hitting a pure element with a beam of electrons accelerated through ~30 keV.

Standard sources use sealed tubes

~3x higher fluxes can be achieved using rotating anodes



Practical limitations of lab sources

Limited intensity means that:

Requires sample sizes that are not always achievable synthetically (especially single crystals)

Trade-off between divergence and intensity → limited resolution

Limited minimum wavelength means that:

The accessible Q range is restricted → a limited number of reflections can be collected

Samples are more absorbing → maximum sample size is limited

Together this means:

The effective maximum dynamic range (weak vs. strong reflections) is limited

Synchrotron X-ray diffraction

~~Expensive, inconvenient and frustrating~~
Intense, energetic and tuneable

Advantages of synchrotron X-rays

High intensity

Small samples (powder or single crystal)

High speed (fast *in situ* processes, unstable compounds...)

High signal-to-noise → better data

→ weak features (impurities, satellite reflections, diffuse scattering)

High resolution → complex structures with big unit cells

→ fine peak splitting (phase transitions, decomposition)

→ peak shapes defined by sample properties (strain, particle size)

Tuneable energy

Highlight elements of interest

Minimise absorption

High energy

High Q → more accurate ADPs, potential for PDF analysis

Compressed patterns → can use restrictive sample environments (pressure, fields)

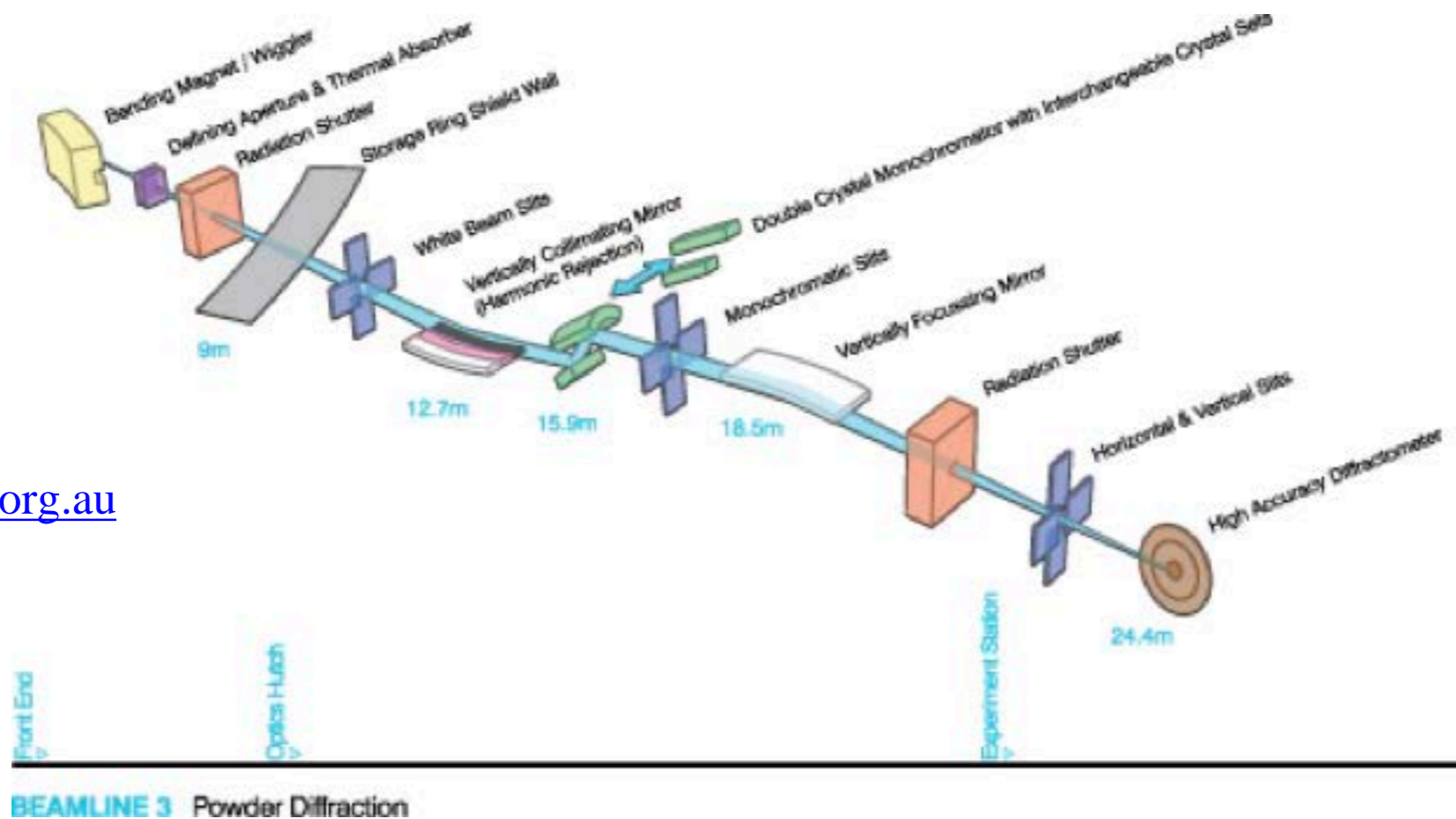
High penetration → minimise surface effects

Diffraction beamline design

The “white” beam must be monochromated for diffraction

Single-crystal monochromators, typically Si(111), select a single x-ray energy

“Bent” monochromators, mirrors and slits can be used to focus the beam



www.synchrotron.org.au

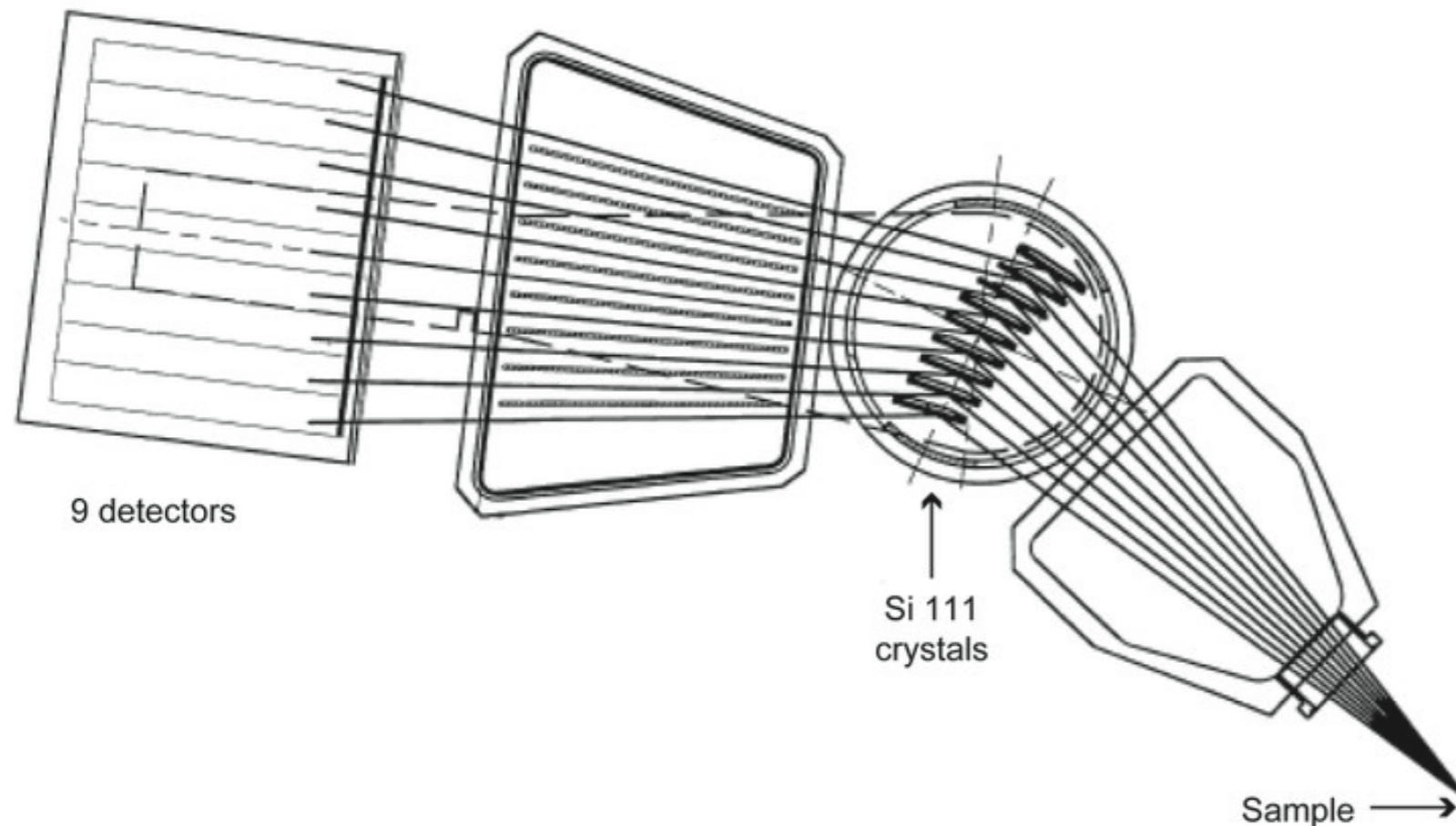
Typical flux at sample: $\sim 10^{12} - 10^{13}$ photons $\text{mm}^{-2} \text{s}^{-1}$

cf $\sim 10^{10} - 10^{11}$ photons $\text{mm}^{-2} \text{s}^{-1}$ (after focussing) from a modern sealed tube (rotating anode) source

High-resolution detection

Analyser crystals

Signal-to-noise can be improved further by using a second monochromator crystal to direct only x-rays of the correct energy to the detector.



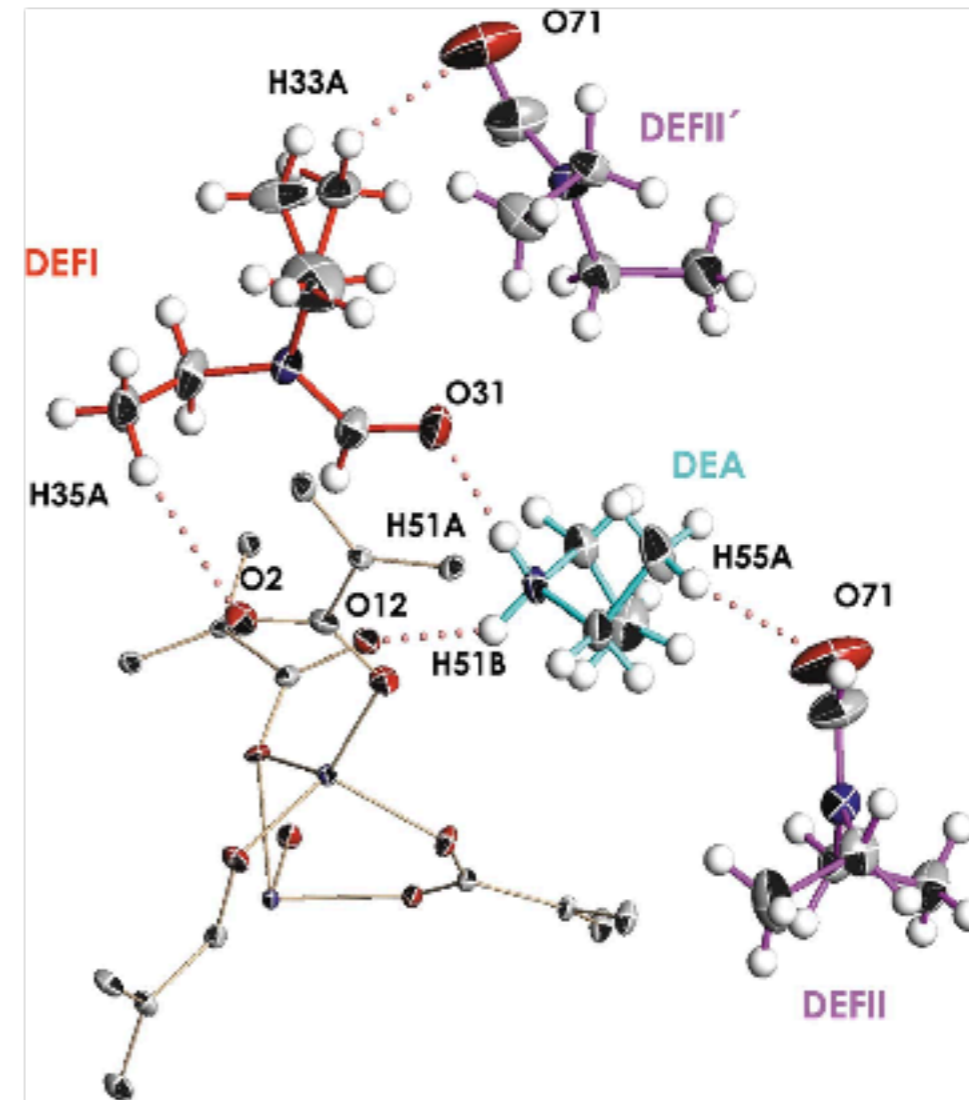
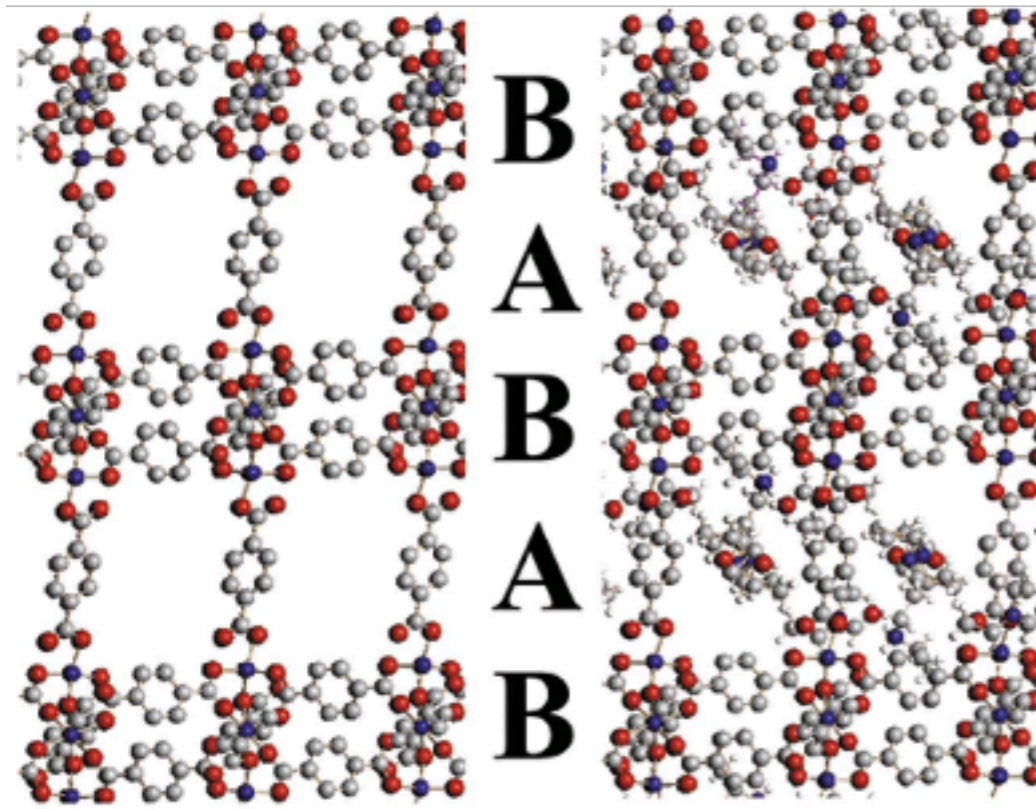
High Intensity and Coherence

Smaller samples

Air-sensitive, hygroscopic etc.

Expensive reagents

Difficult to synthesise

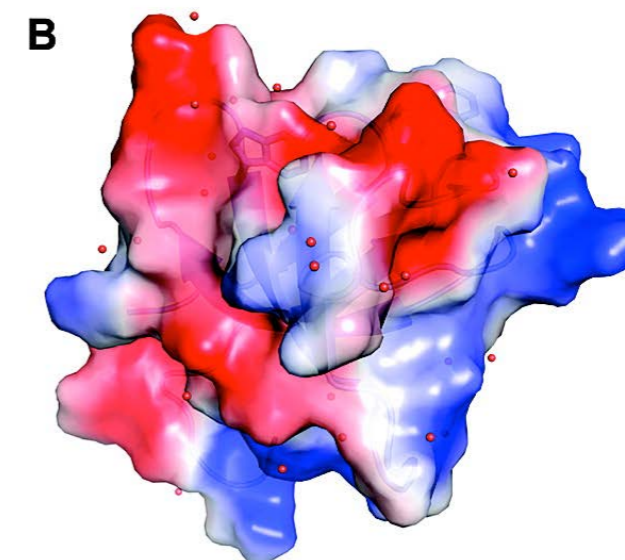
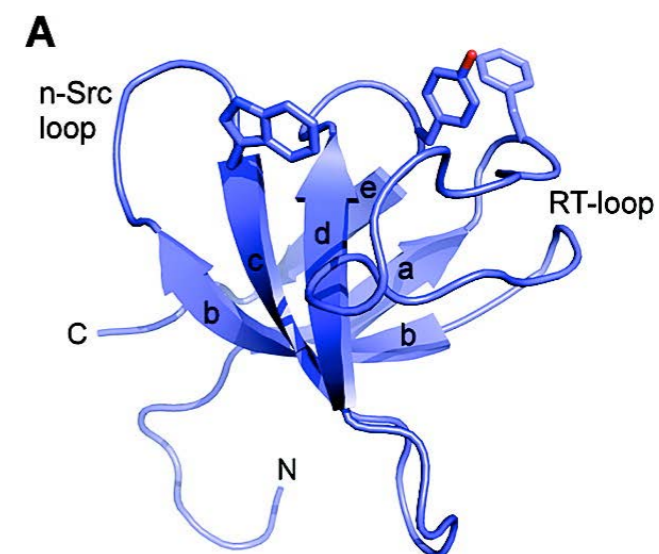
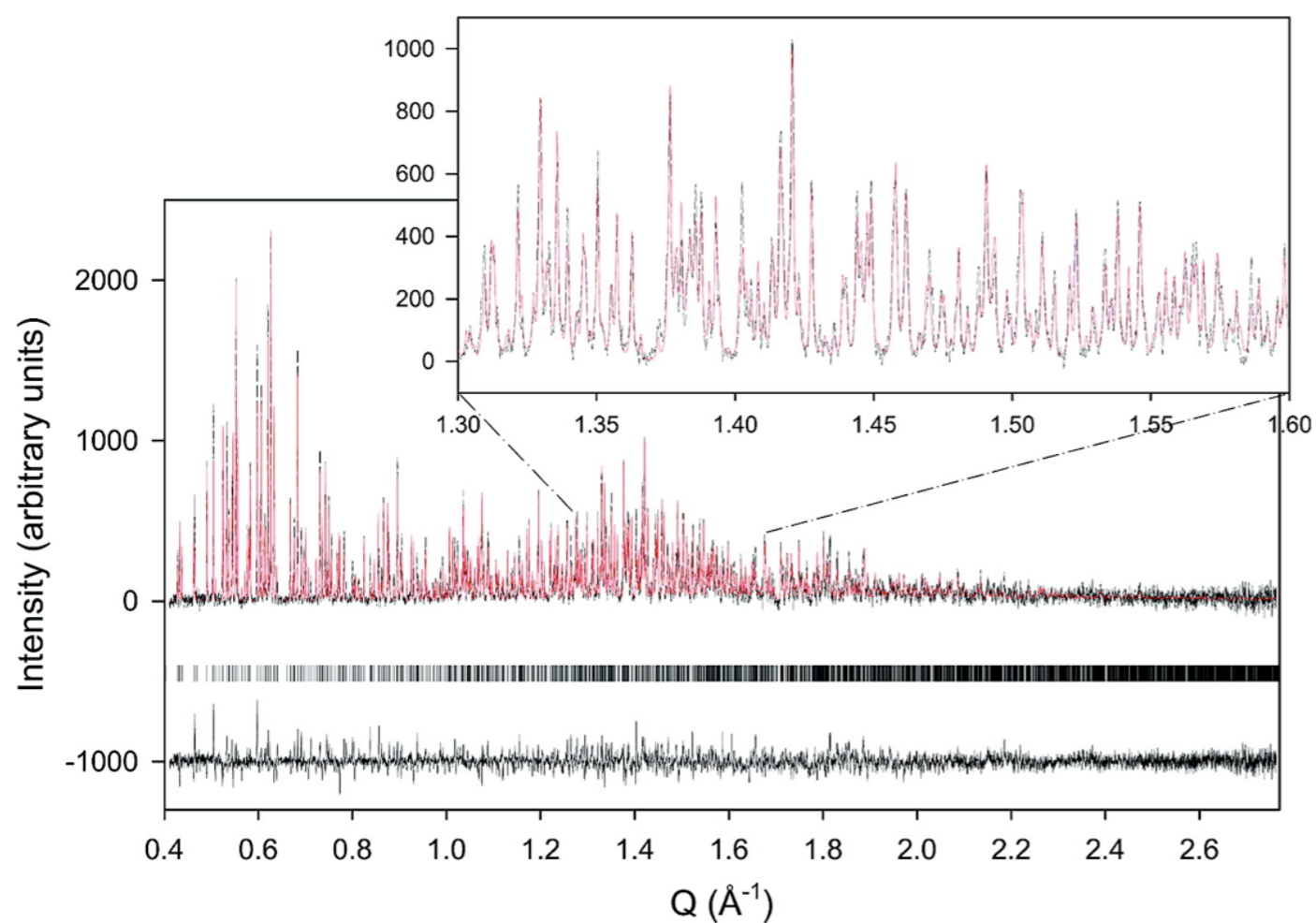
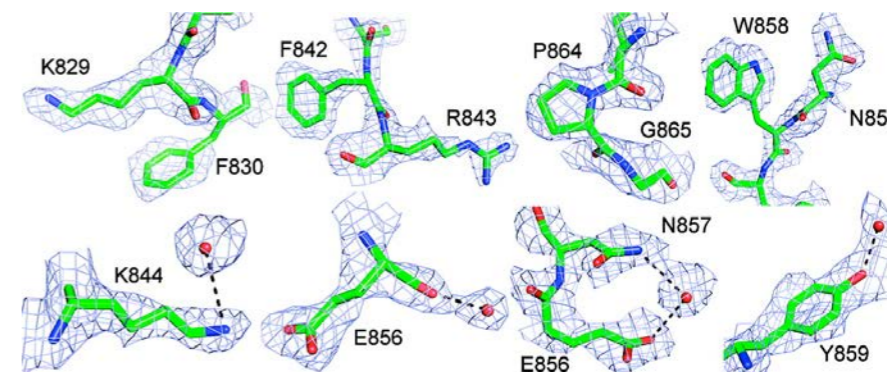


40µm crystal of
 $2\text{C}_4\text{H}_{12}\text{N}^+ \cdot [\text{Co}_3(\text{C}_8\text{H}_4\text{O}_4)_4]^{2-} \cdot 3\text{C}_5\text{H}_{11}\text{NO}$

High Intensity and Coherence

Complex structures

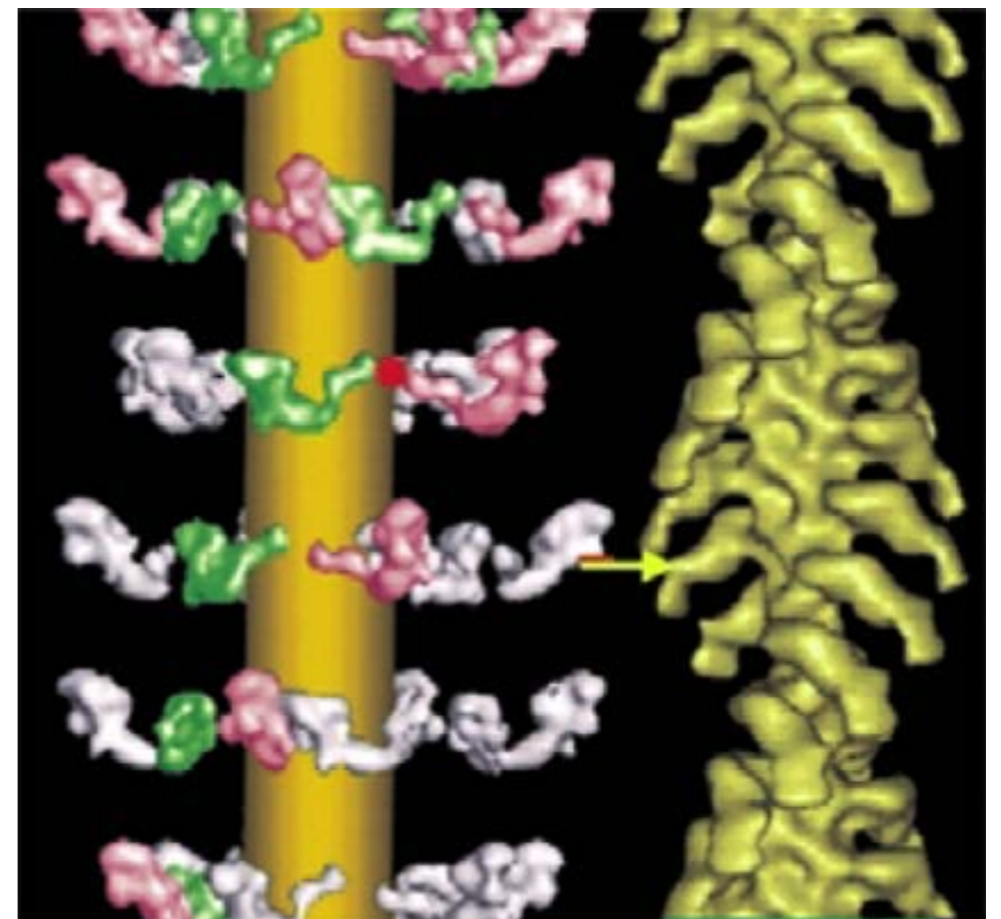
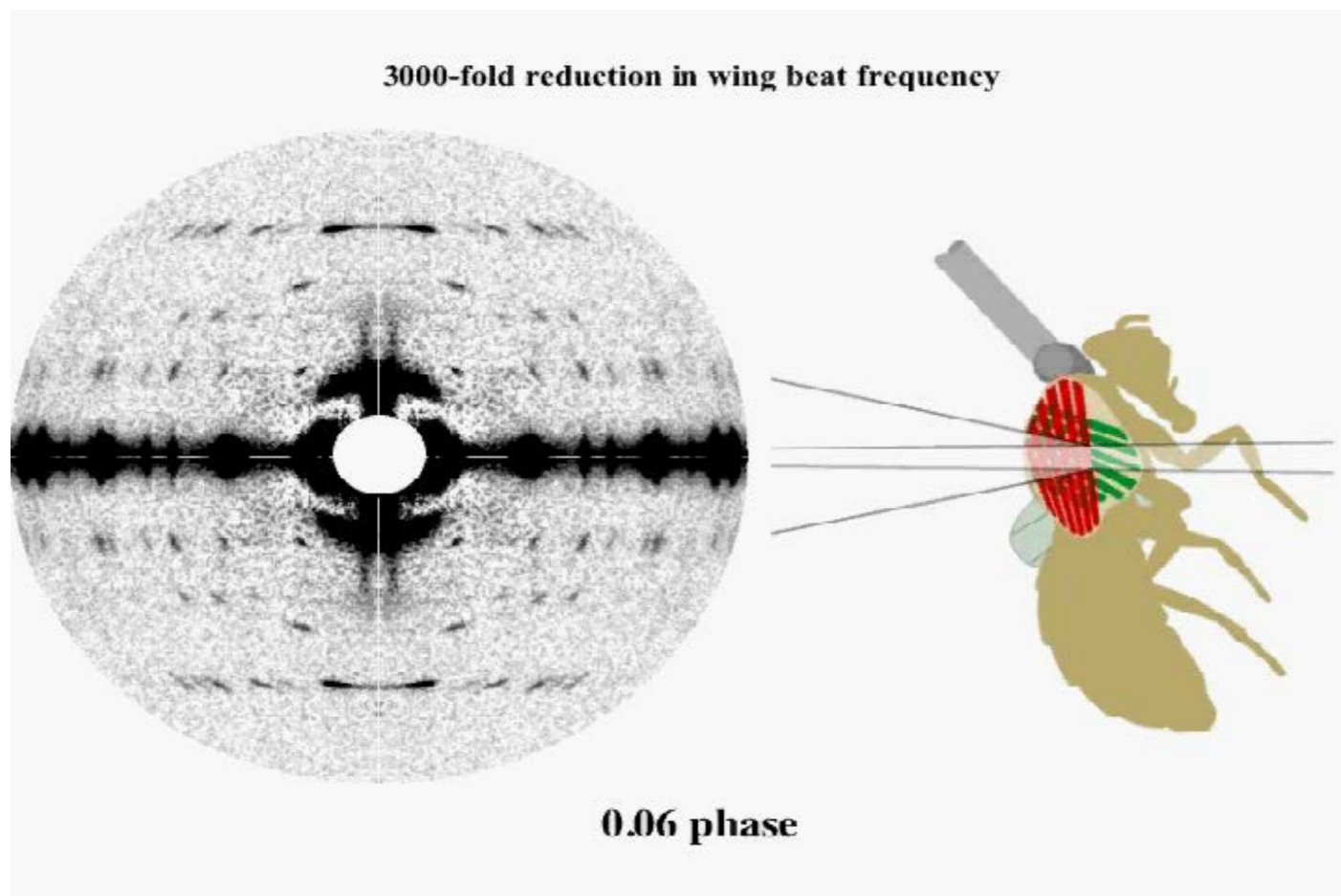
Rietveld-refinement of a 67-residue protein domain crystal structure with a cell volume of 64 879 Å³



High Intensity and Coherence

High-speed diffraction

Variable wavelengths permit ultra-fast “white-beam” Laue diffraction



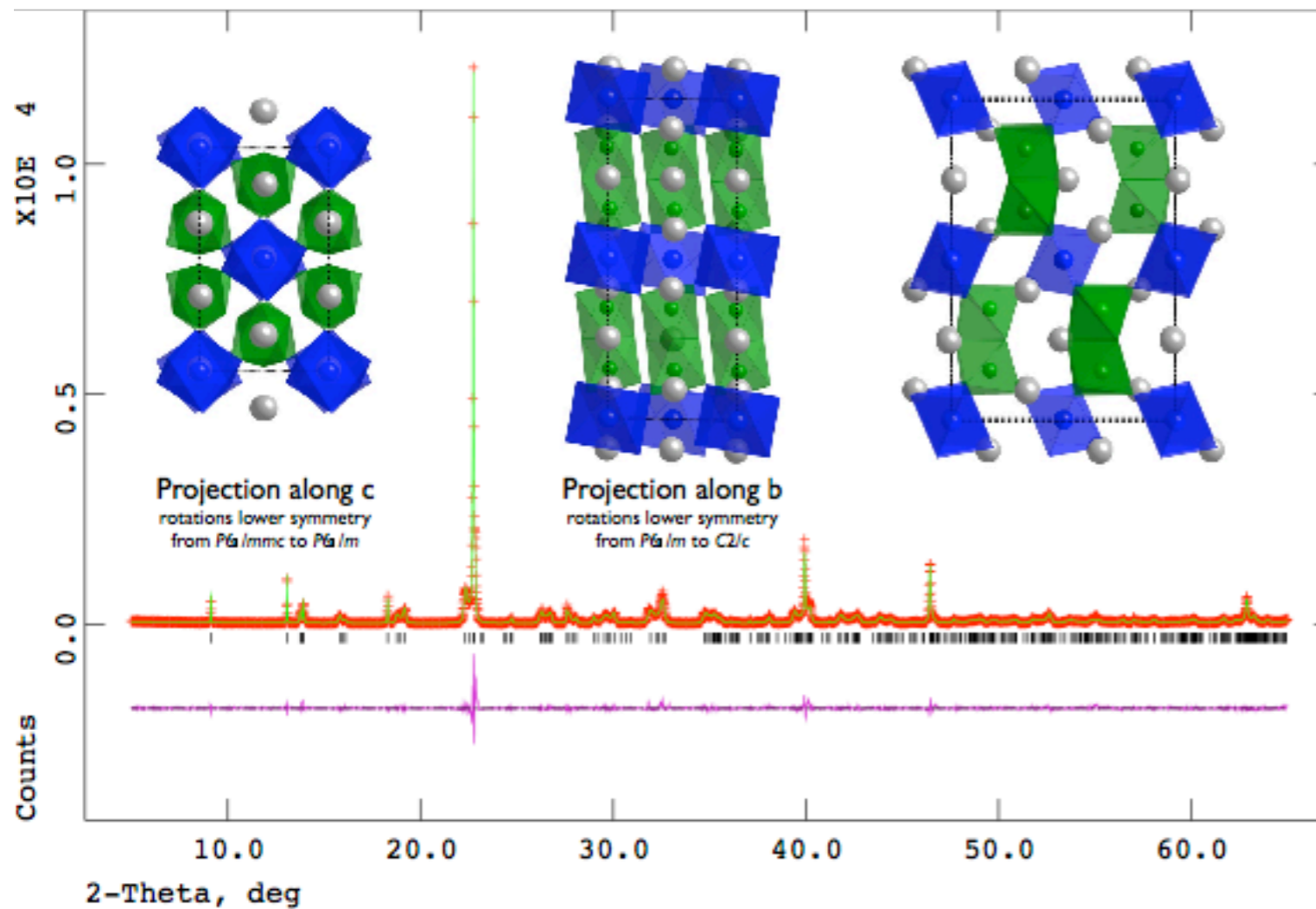
High Intensity and Coherence

Lower beam divergence → higher resolution

Large, low-symmetry unit cells

Sample-dependent peak shape functions, e.g., lattice strain

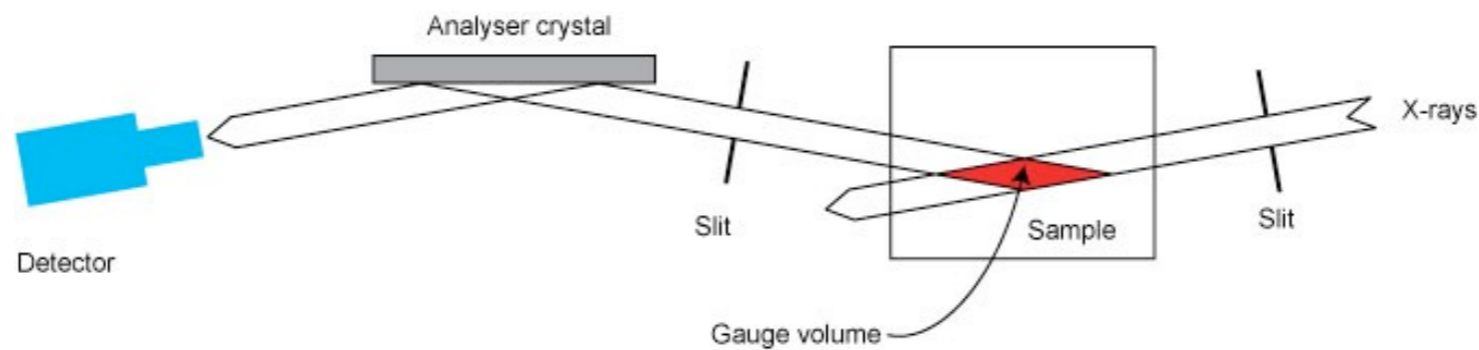
Subtle symmetry lowering



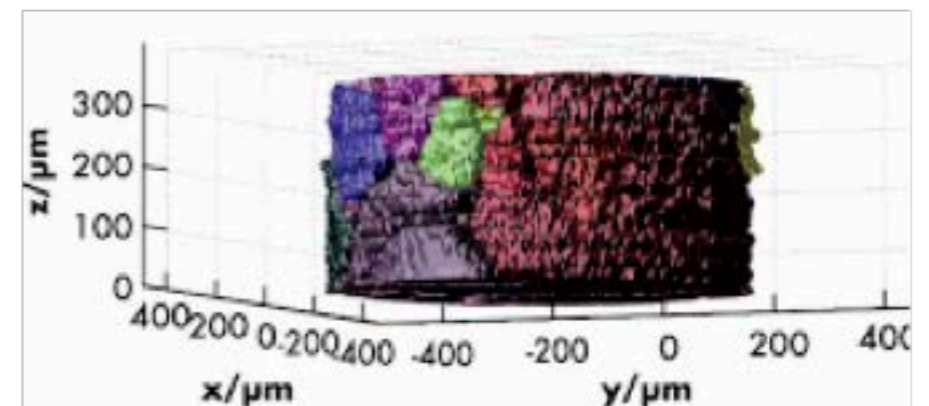
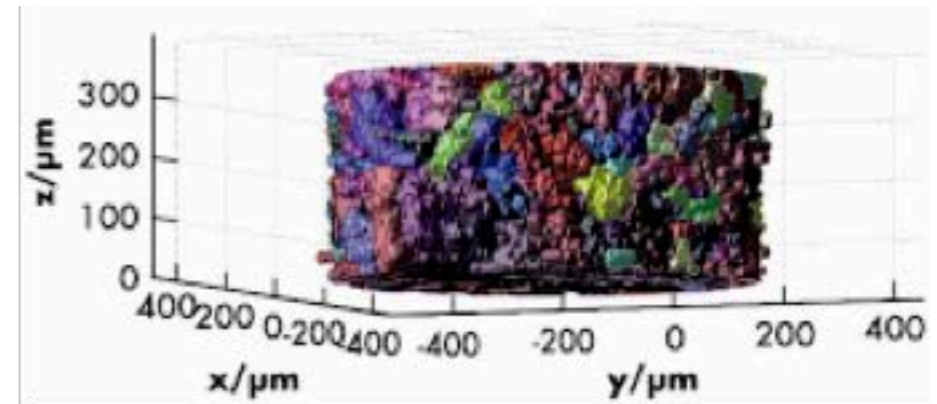
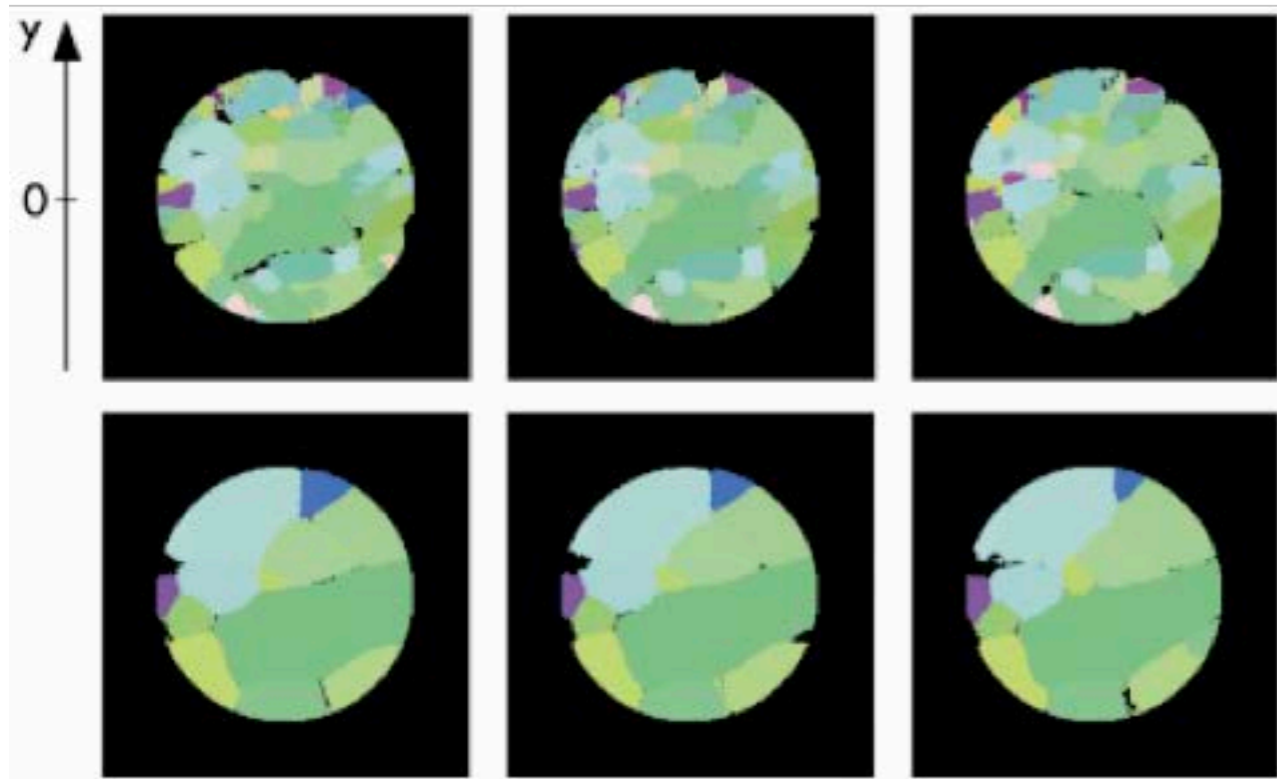
Tuneable Energy/Wavelength

Higher x-ray energy \rightarrow higher penetration

3D mapping of crystal orientation and/or strain (peak width) in different gauge volumes



Al-0.1%Mn alloy before (top) and after annealing

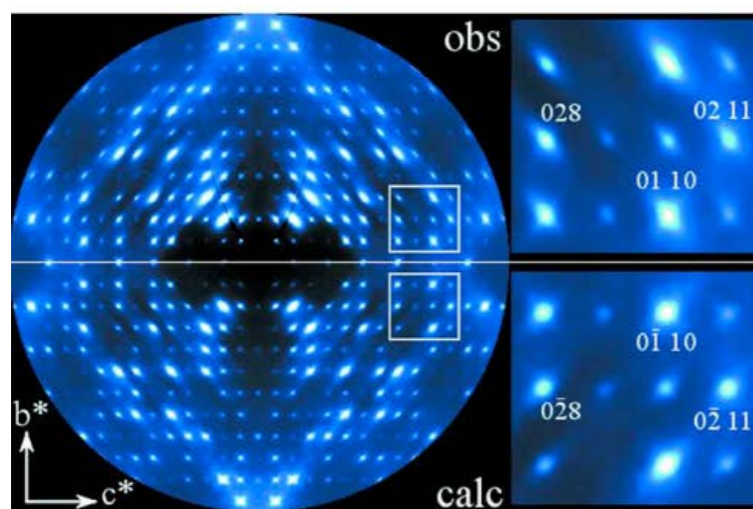


High Intensity and Coherence

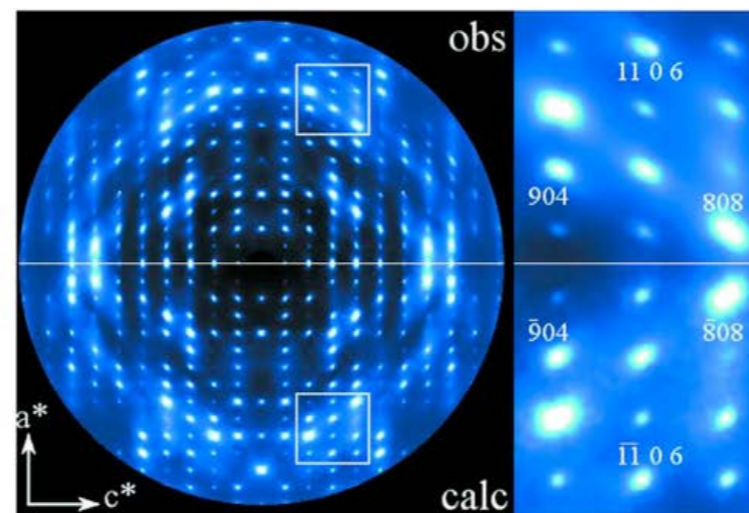
Greater accessible dynamic range

Diffuse scattering: structural disorder

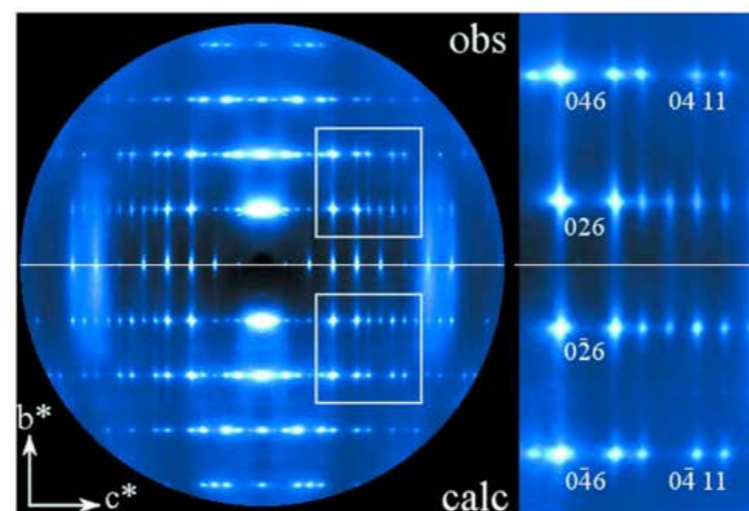
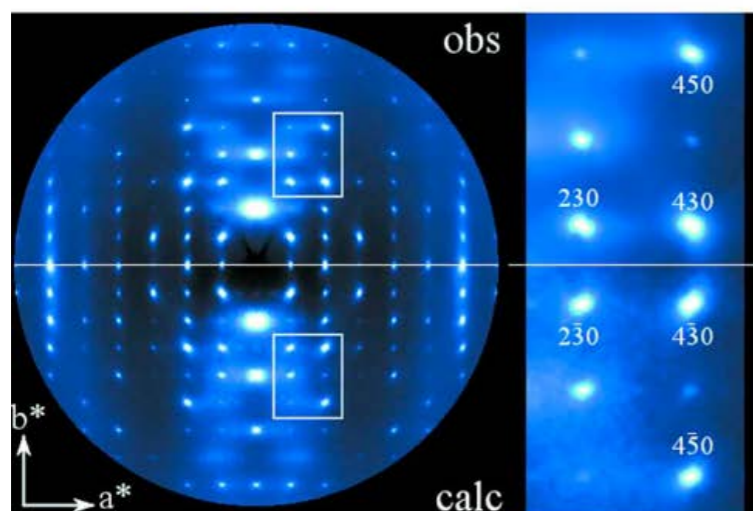
Modulated structures



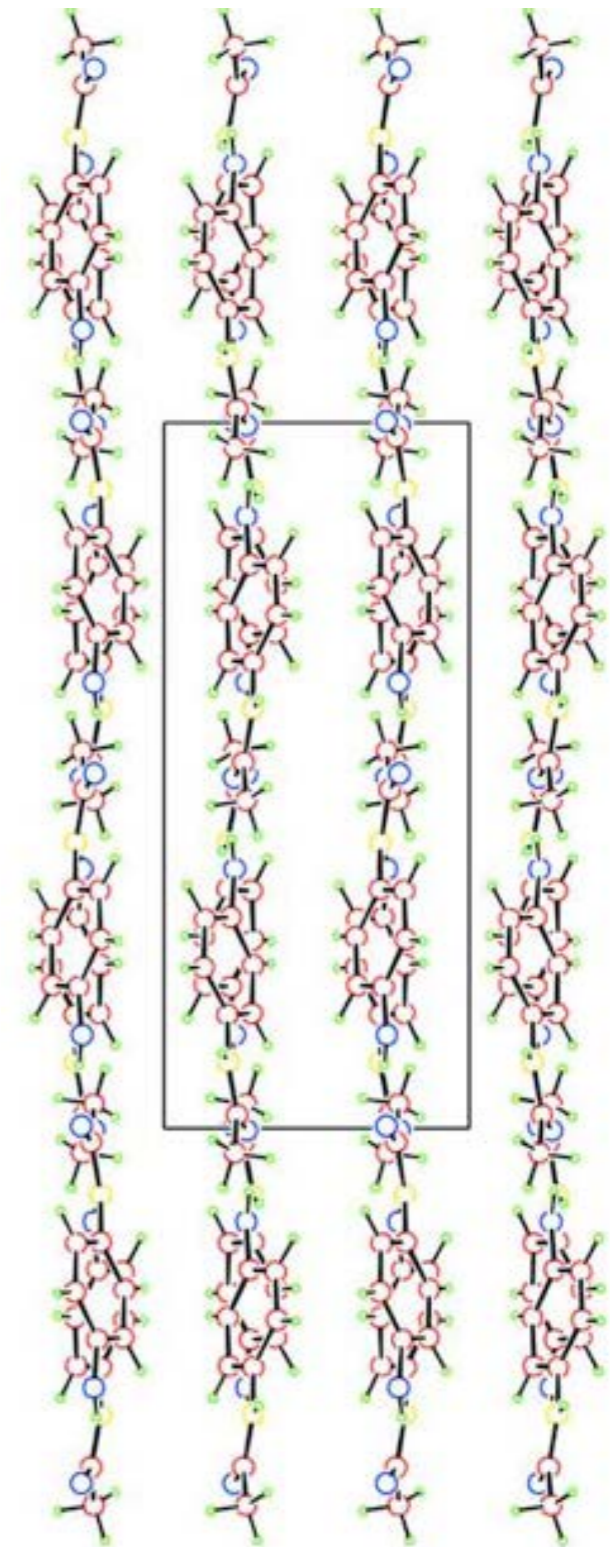
(c)



(d)



Paracetamol

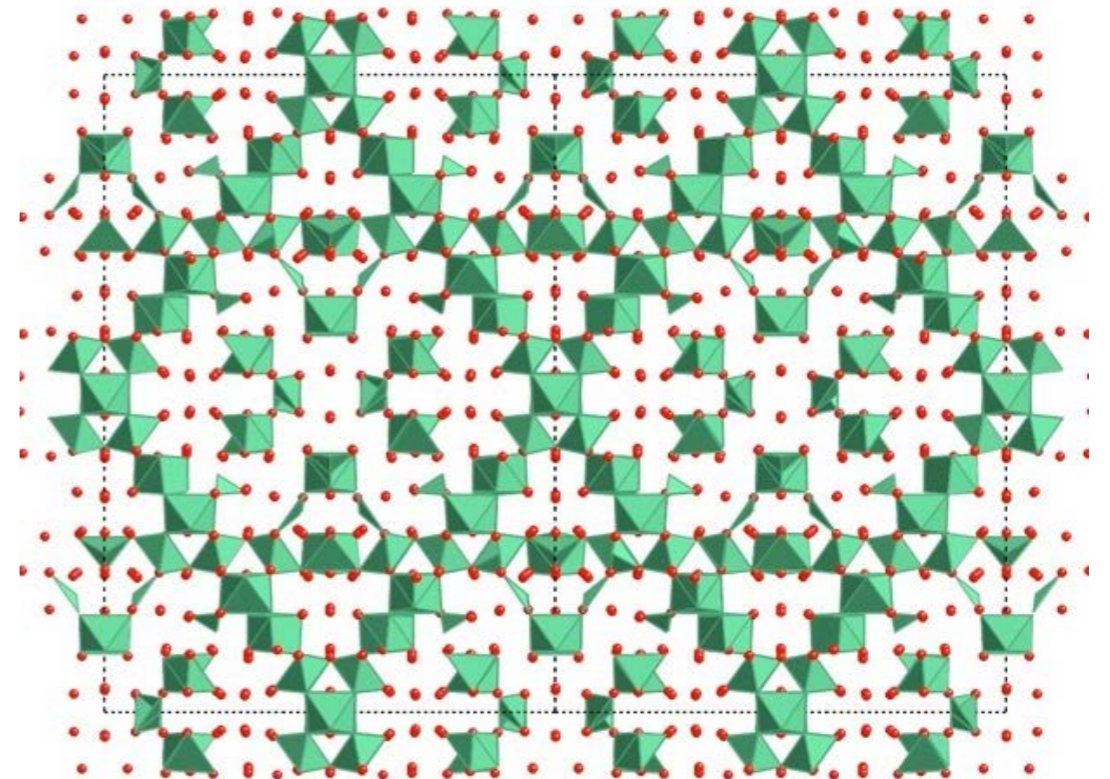
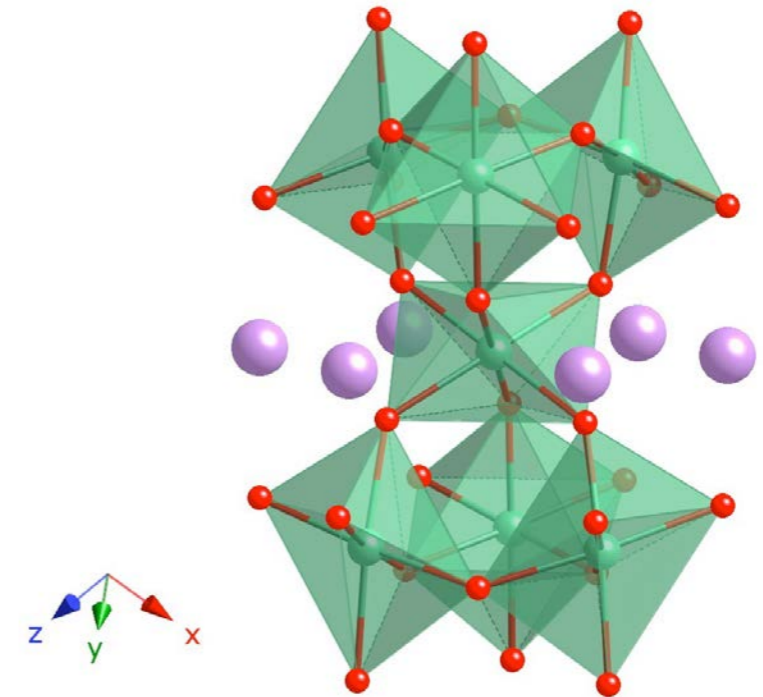
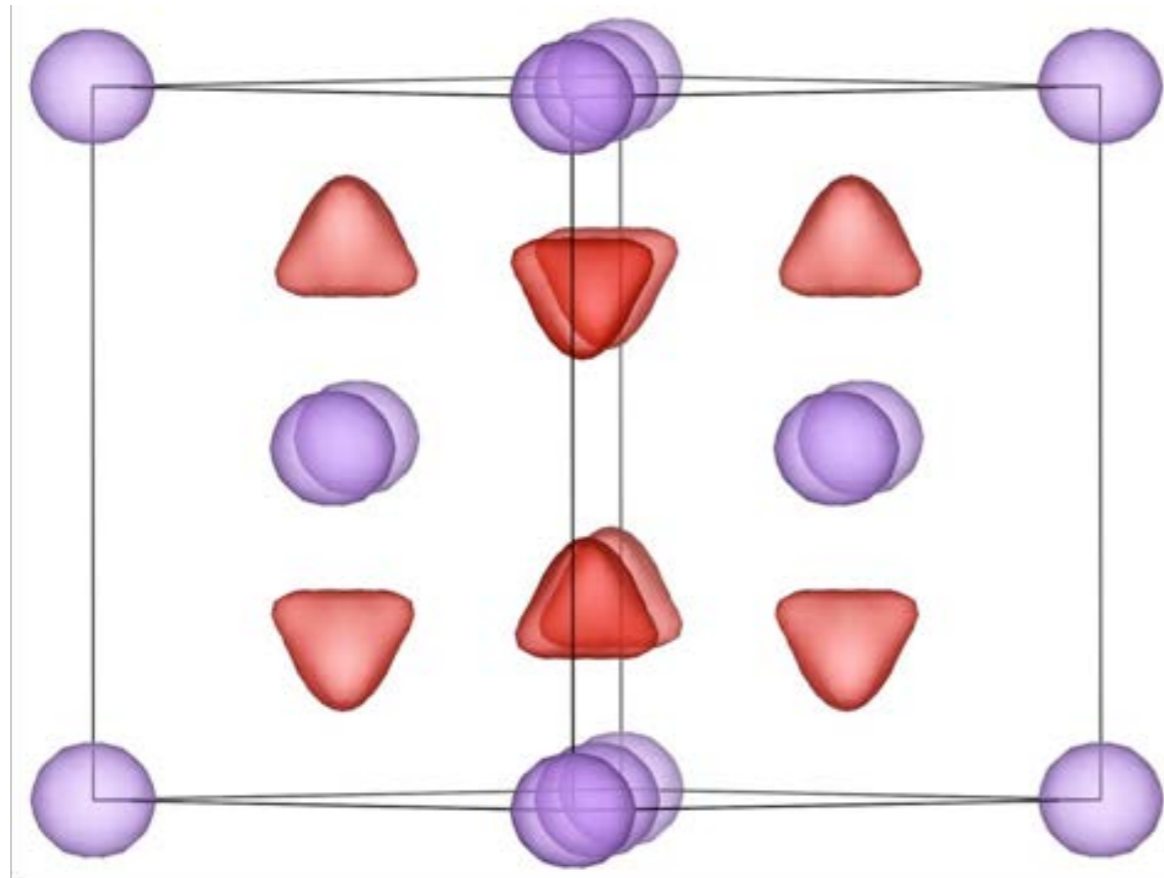


Tunable Energy/Wavelength

Greater accessible $\sin\theta/\lambda$ range

More information about atomic occupancies

Anharmonic atomic displacement parameters



Tuneable Energy/Wavelength

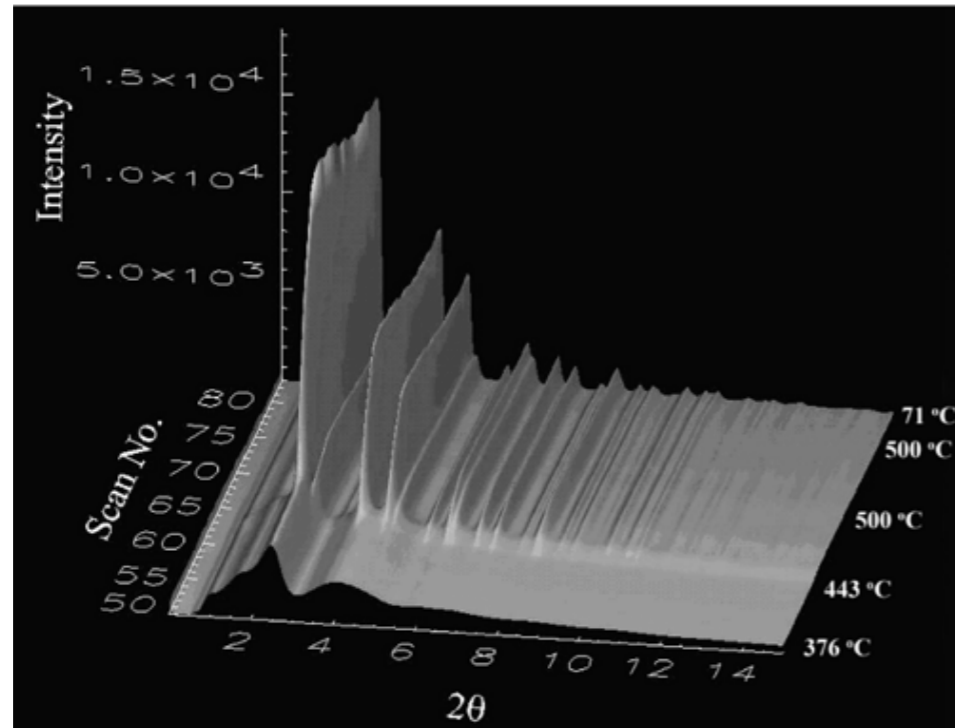
Greater accessible $\sin\theta/\lambda$ range

Site occupancies in disordered structures

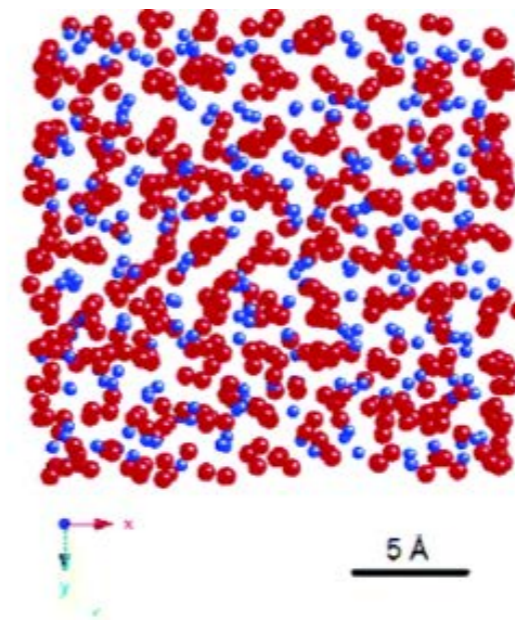
Electron density, maximum entropy (MEM)

Anharmonic atomic displacement parameters

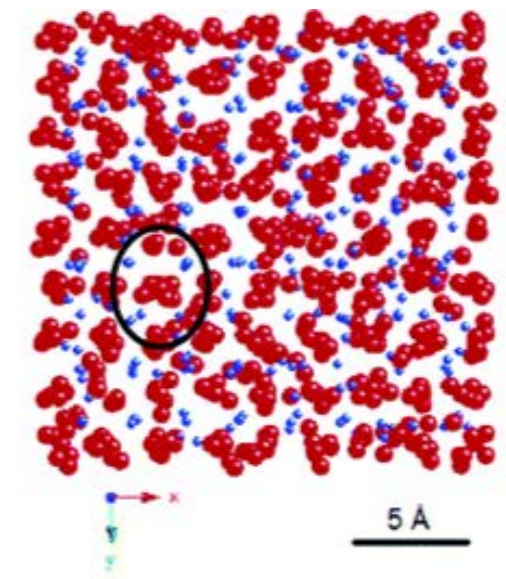
Pair distribution function (PDF) analysis



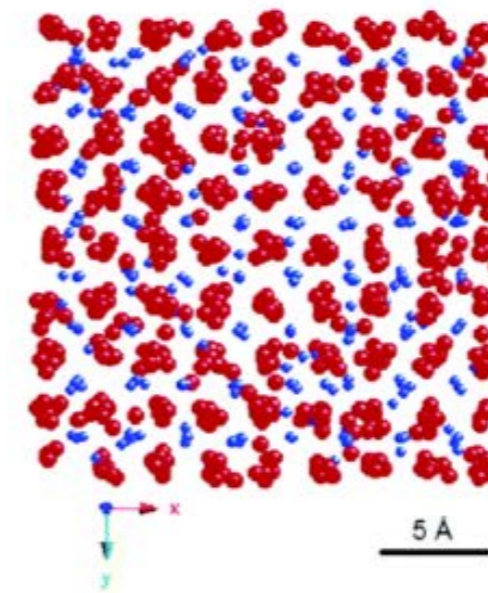
1-minute S-XRD scans at $\lambda = 0.124 \text{ \AA}$ of cubic ZrO₂ being reduced to an amorphous phase



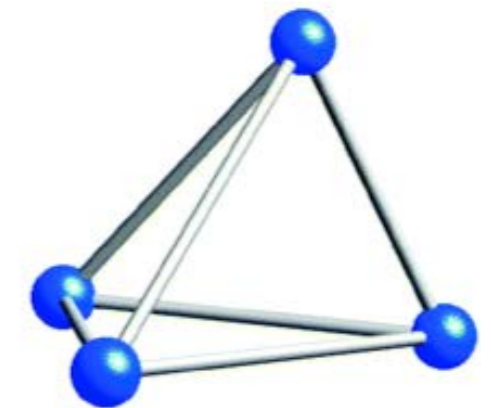
(a) Amorphous (Scan No.57)



(b) Transition (Scan No. 60)



(c) Cubic 500°C



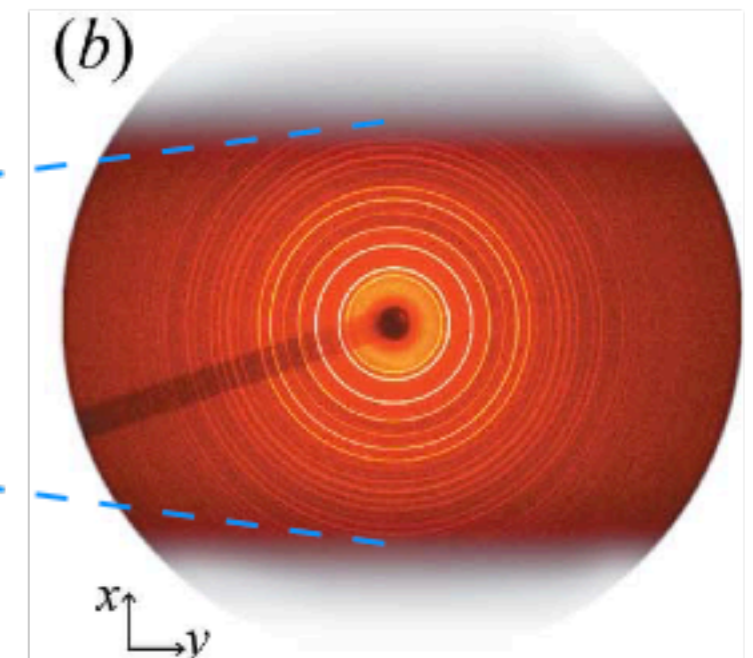
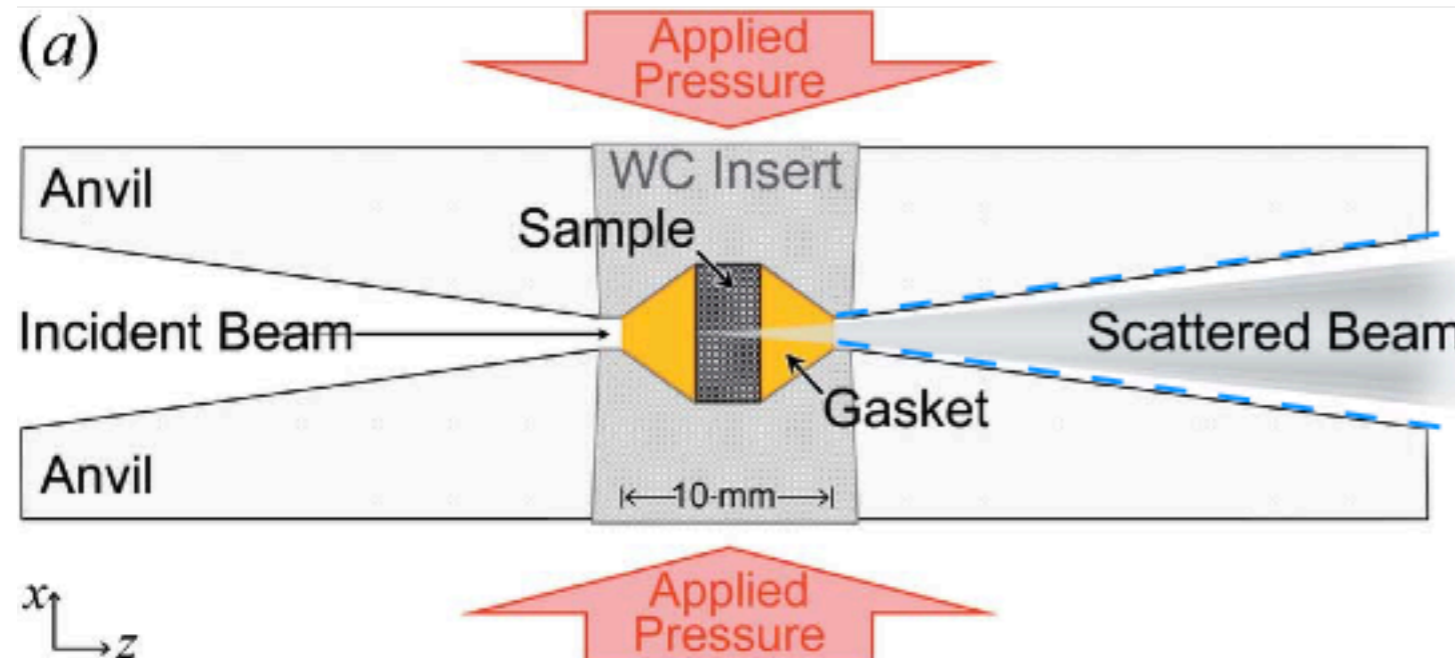
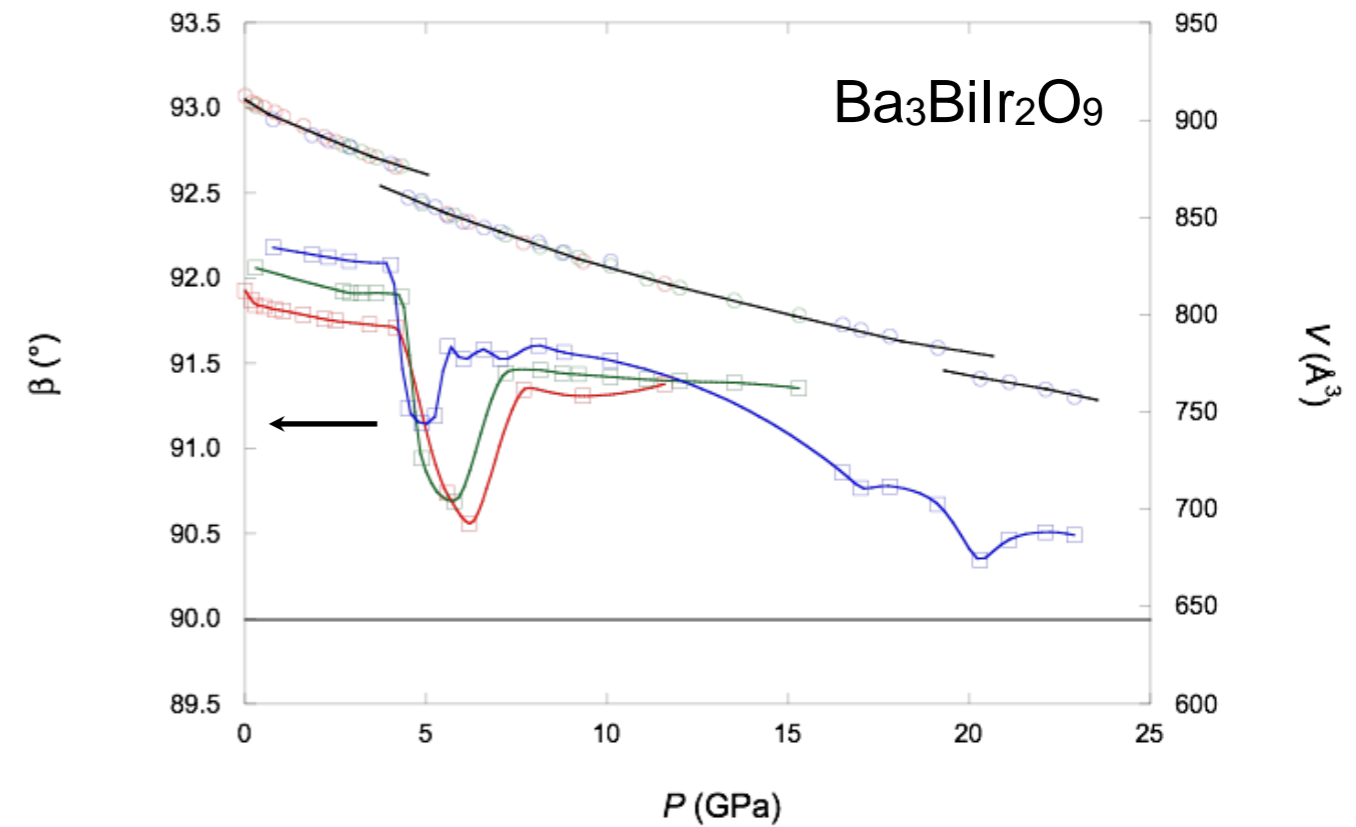
(d) Zr tetrahedron unit

Tunable Energy/Wavelength

Difficult sample environments

Pattern more compressed

Better penetration of gaskets *etc*

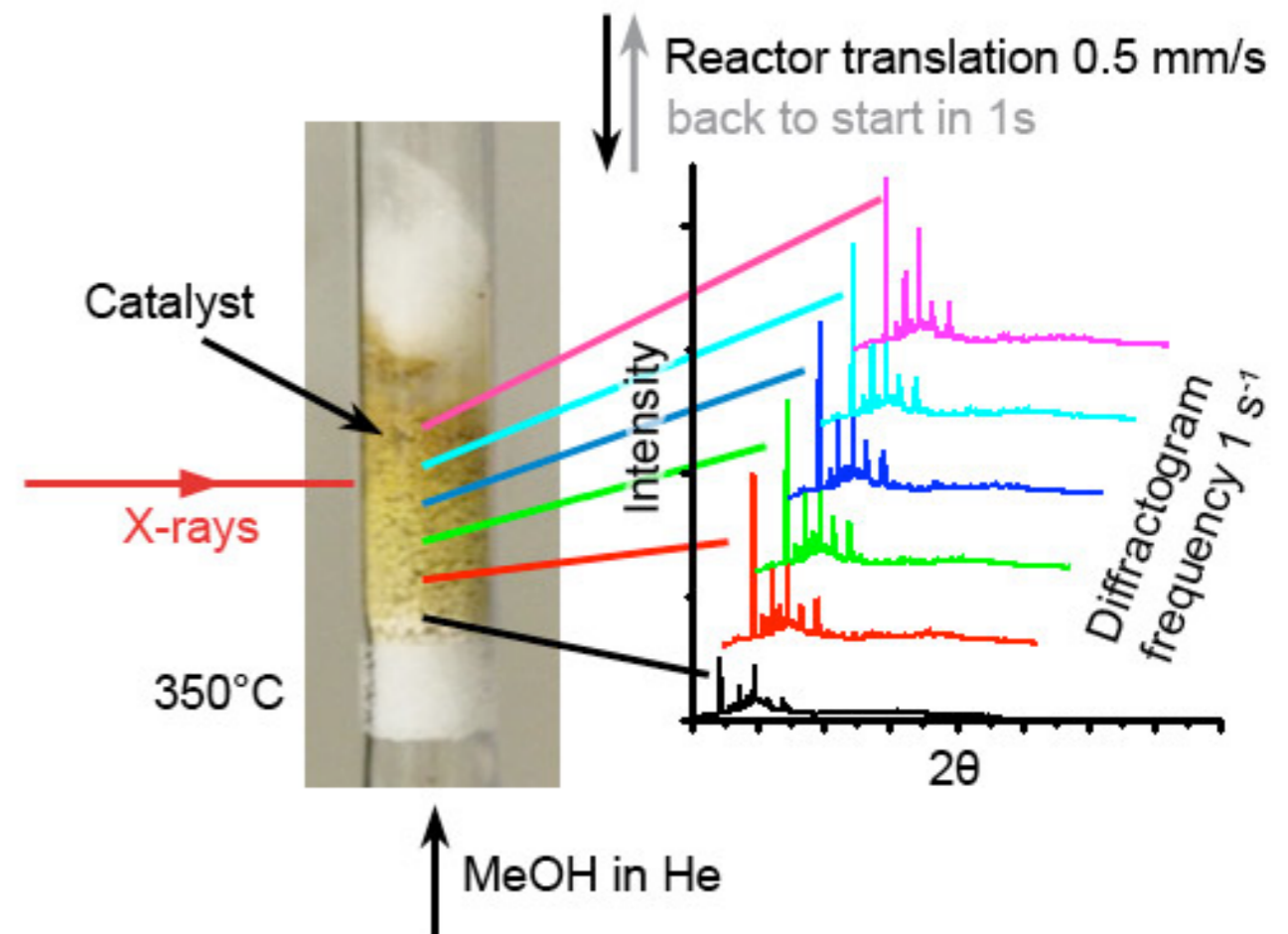
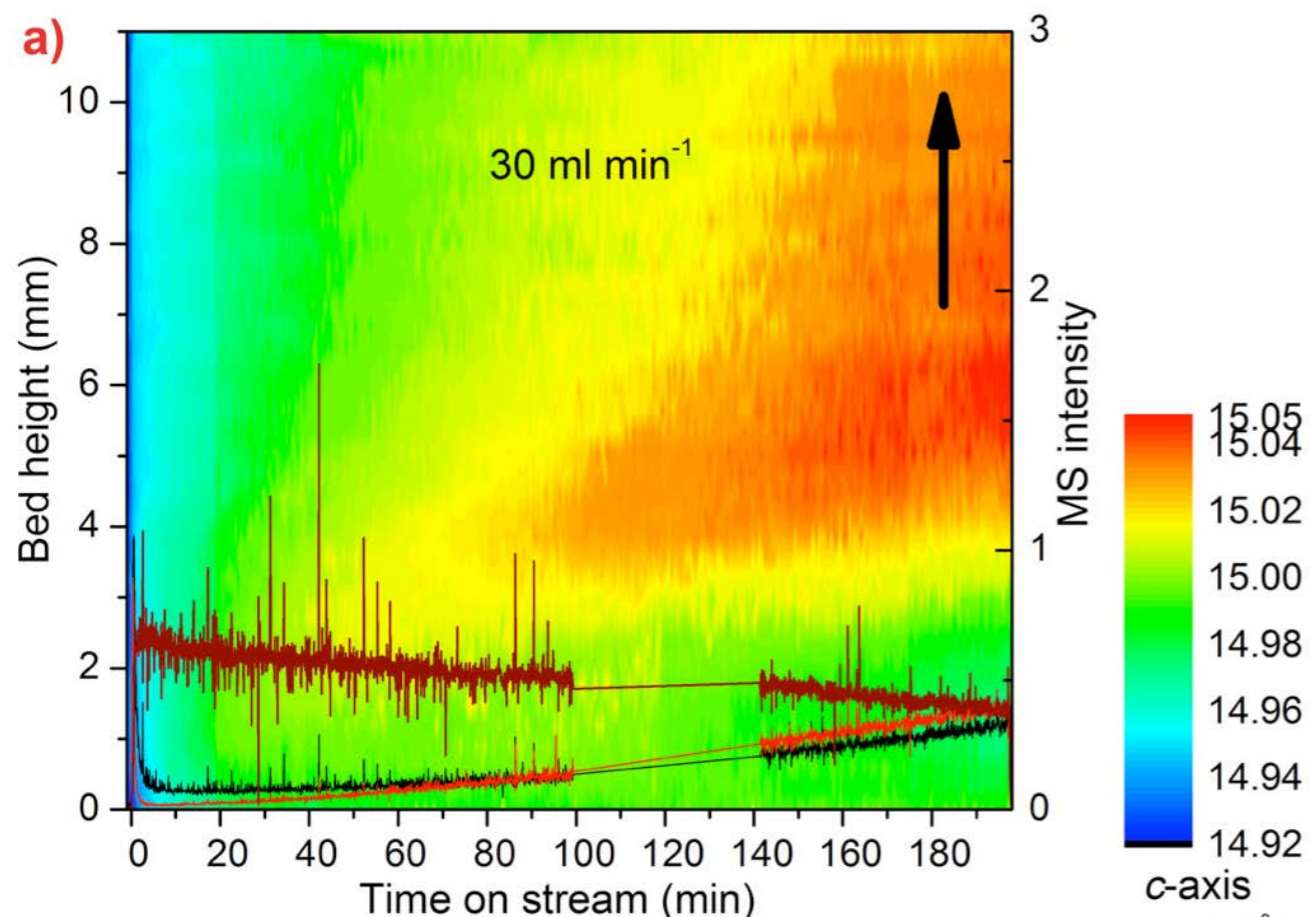


Tuneable Energy/Wavelength

In situ reactions

High spatial resolution, high data collection speed and high sample penetration allow the study of industrial reactions under “real-world” conditions in real time.

Lattice expansion of a zeolite catalyst in a methanol-to-olefin reactor, due to coking, mapped in time and space



Summary: synchrotron killer apps

Small samples

Heavy elements

Big, complex and/or low-symmetry structures

High-resolution ADPs

Modulated structures

Short-range order (PDF)

Medium-range order (diffuse scattering)

Torturing insects



産美収集箱
2017/11/16



西側入口



