



UNIVERSITY OF SOUTH ALABAMA

# GY 302: Crystallography & Mineralogy

---

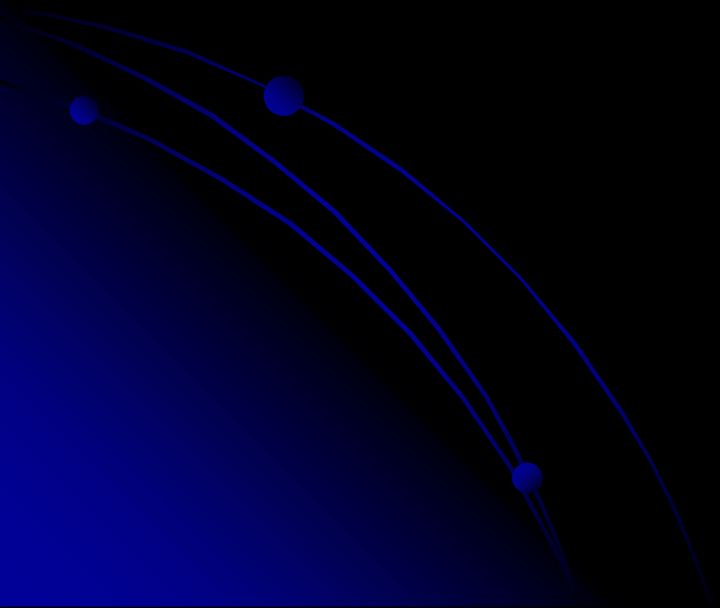
## Lecture 6: Polymorphism and Crystal Habit

Instructor: Dr. Douglas Haywick

# Online Lecture Review

---

1. Polymorphs and Polymorphism
2. Pseudomorphs and other definitions
3. Crystal Habit

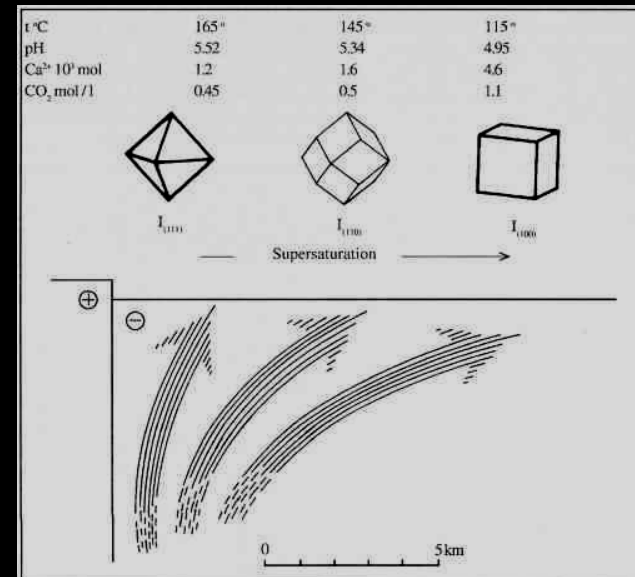


# Polymorphs & Polymorphism

**Polymorphism:** literally translates as “many forms”

In crystallography/mineralogy, it generally means that one mineral can exist in more than one crystal form under certain conditions (normally pressure/temperature).

Crystal forms of fluorite versus temperature (this is not polymorphism. It is only a change of **Crystal Habit**)

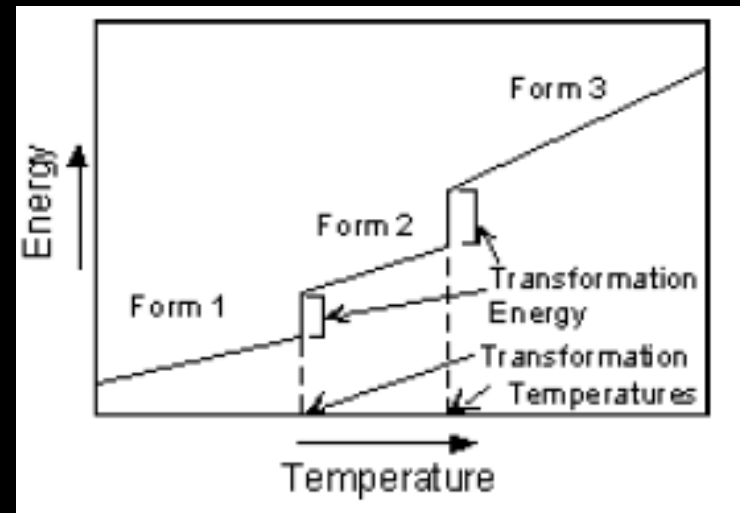


# Polymorphs & Polymorphism

Polymorphic transformations can occur in one of 3 ways:

1) **Reconstructive**: requires extensive breaking and recombining of chemical bonds in the crystal lattice. This transformation generally occurs only when the pressure/temperature threshold is reached and may be very slow\*.

\* really slow transformations may produce long-lived metastable Polymorphs (e.g., diamond)

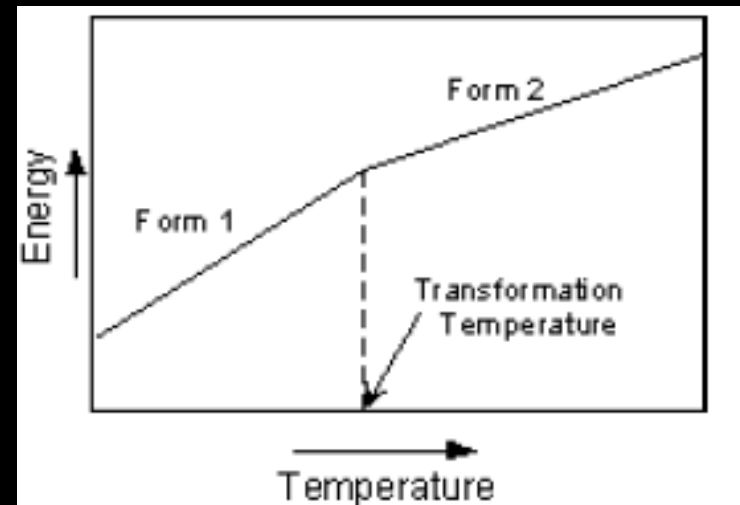


# Polymorphs & Polymorphism

Polymorphic transformations can occur in one of 3 ways:

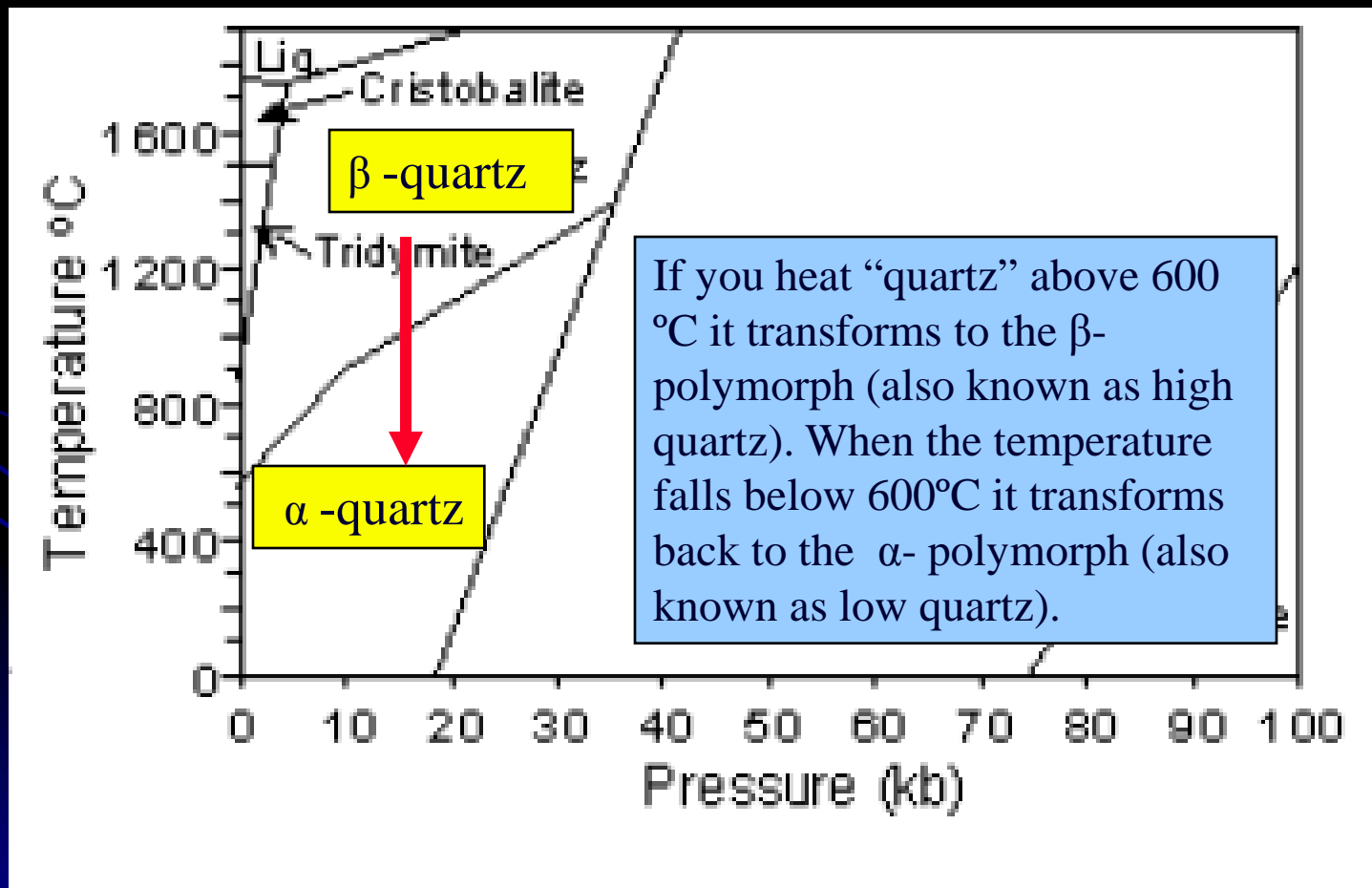
2) **Displacive**: requires relatively minor changes in the crystal lattice (e.g., modification of  $\alpha$ ,  $\beta$  or  $\gamma$  angles). There is generally no change in energy at the transformation threshold so polymorphic transformations are instantaneous and reversible.

No metastable polymorphics exist (e.g.,  $\alpha$ -quartz and  $\beta$ -quartz)



# Polymorphs & Polymorphism

Quartz has 6 polymorphs related to pressure and temperature



# Polymorphs & Polymorphism

---

Polymorphic transformations can occur in one of 3 ways:

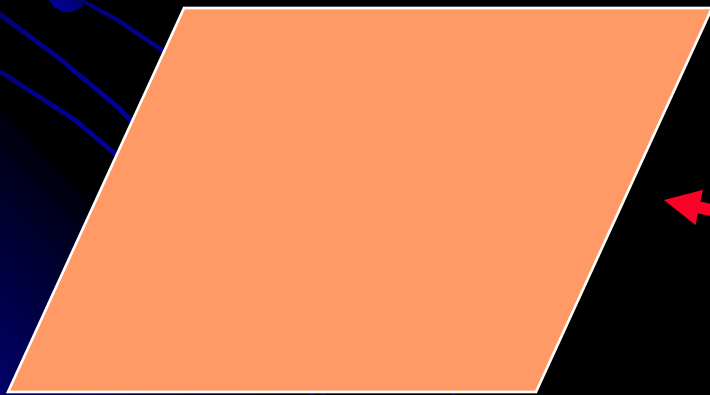
3) **Order-Disorder transformations**: all crystals have a certain amount of disorder to them (e.g., “saddle” dolomite) and in some cases, polymorphic transitions can occur once an appropriate level of order-disorder is attained.



# Polymorphs & Polymorphism

---

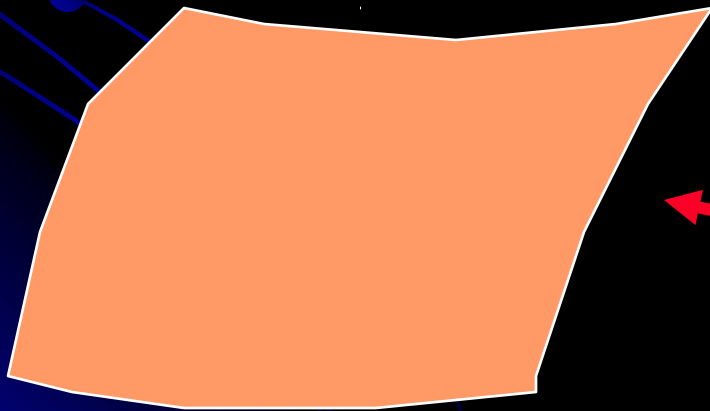
Well ordered dolomite  
 $\text{CaMg}(\text{CO}_3)_2$



# Polymorphs & Polymorphism

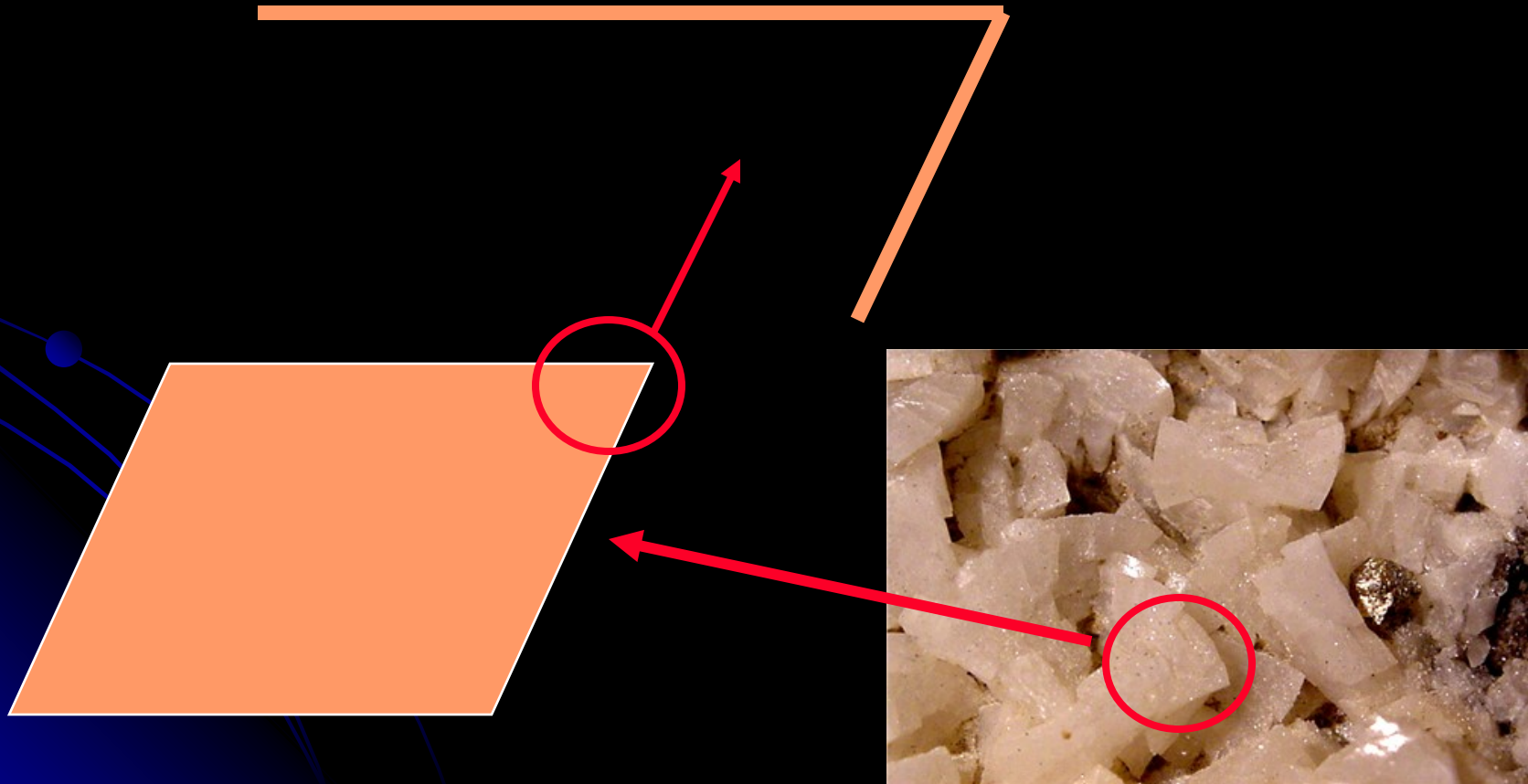
---

Less well ordered dolomite  
 $\text{CaMg}(\text{CO}_3)_2$

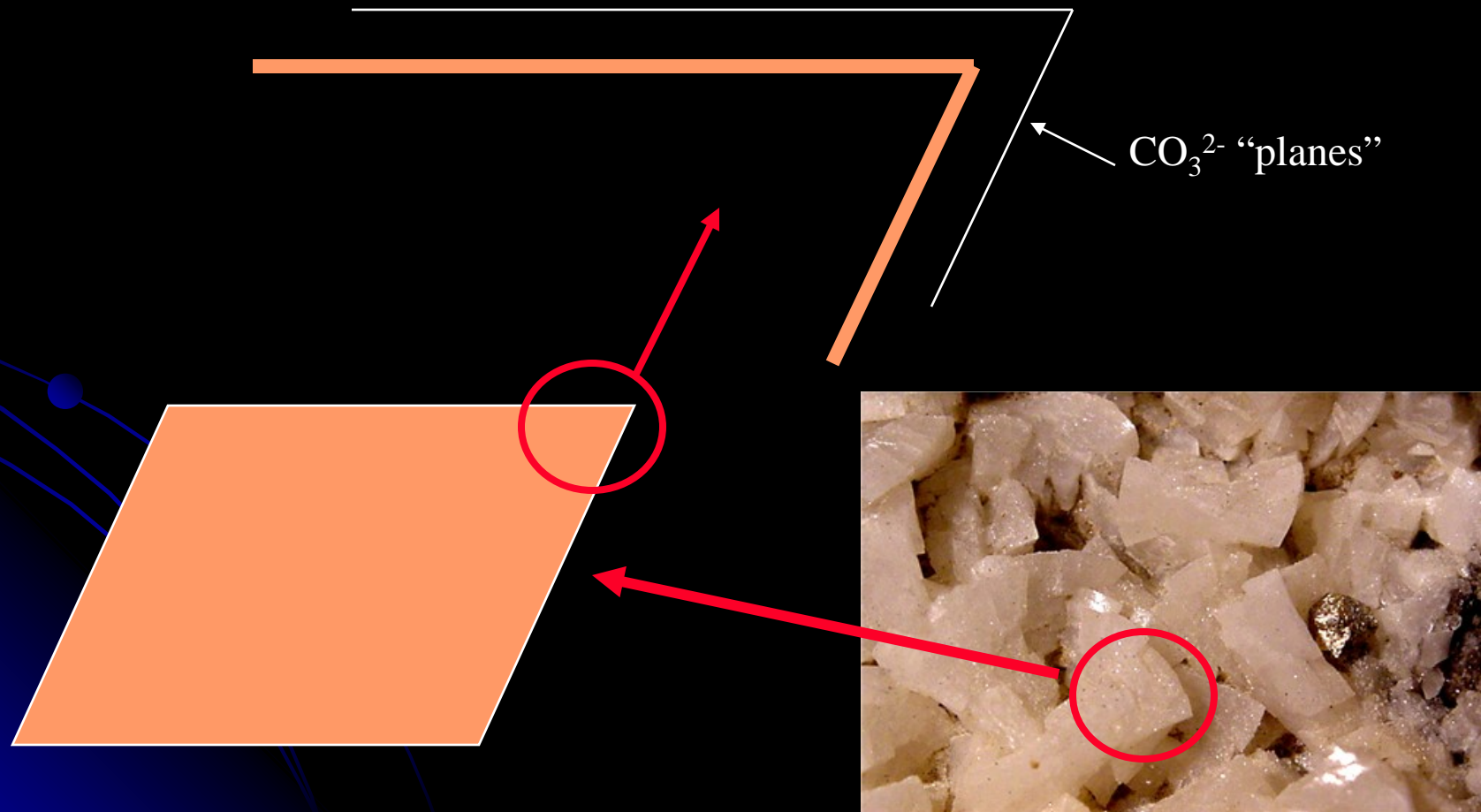


# Polymorphs & Polymorphism

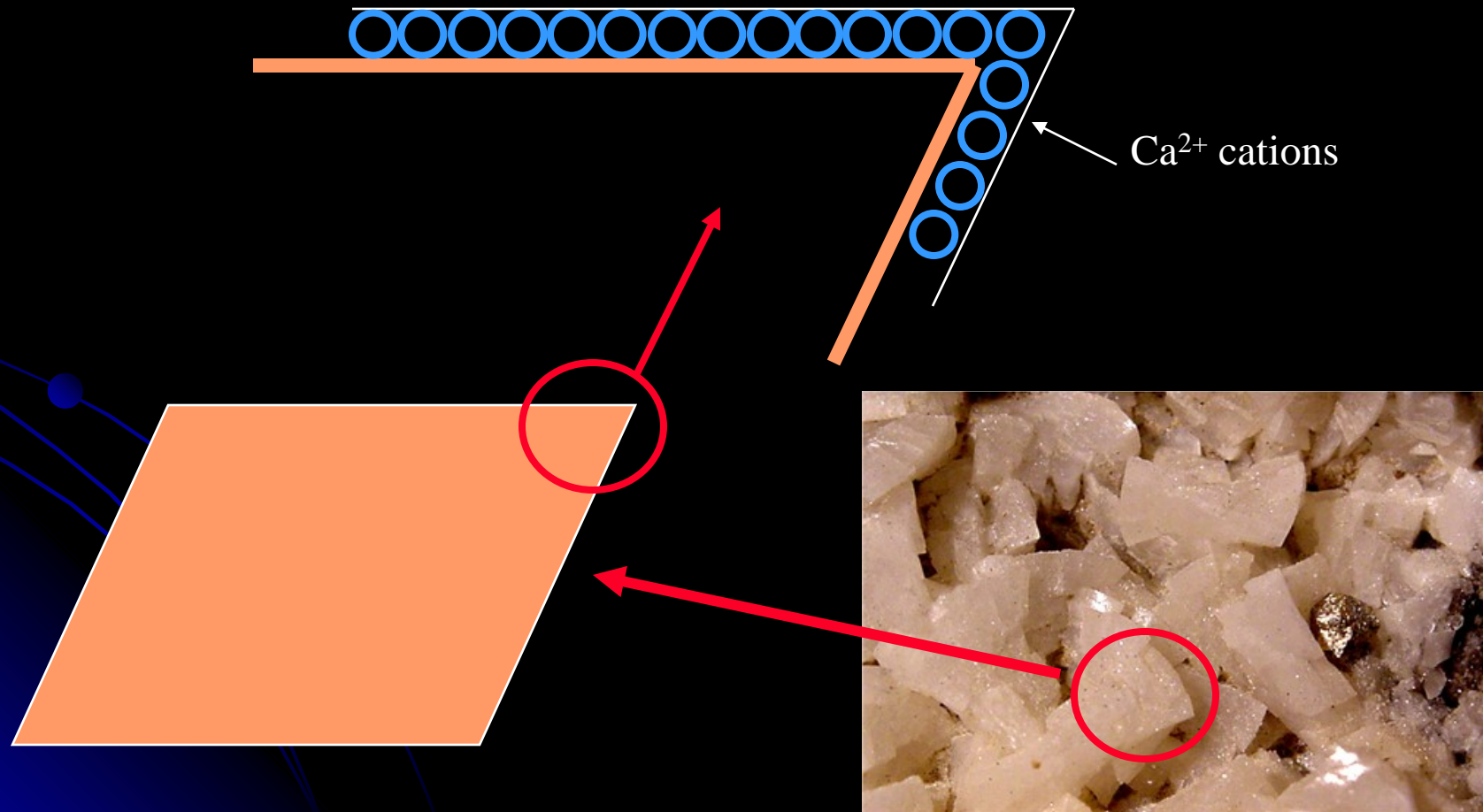
---



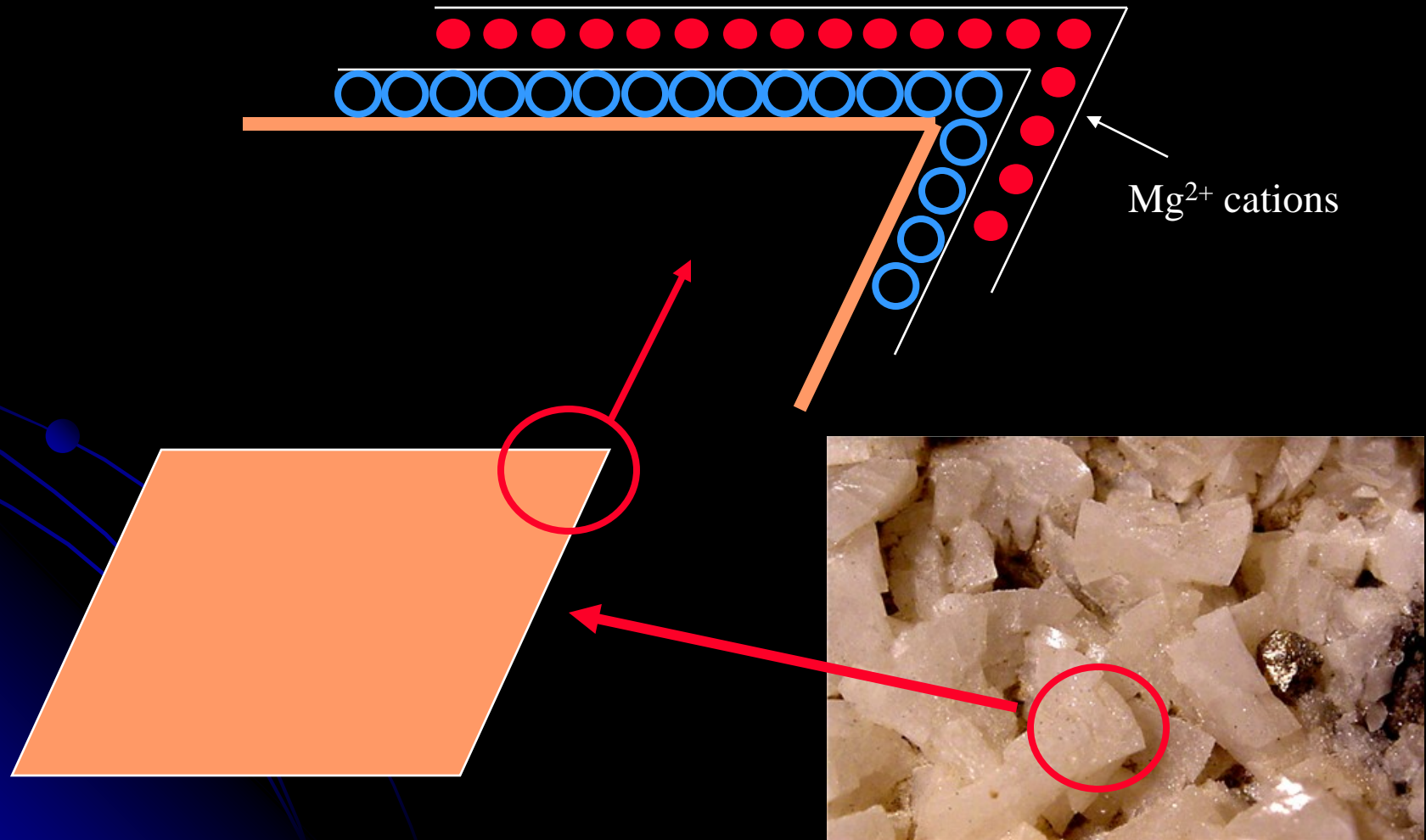
# Polymorphs & Polymorphism



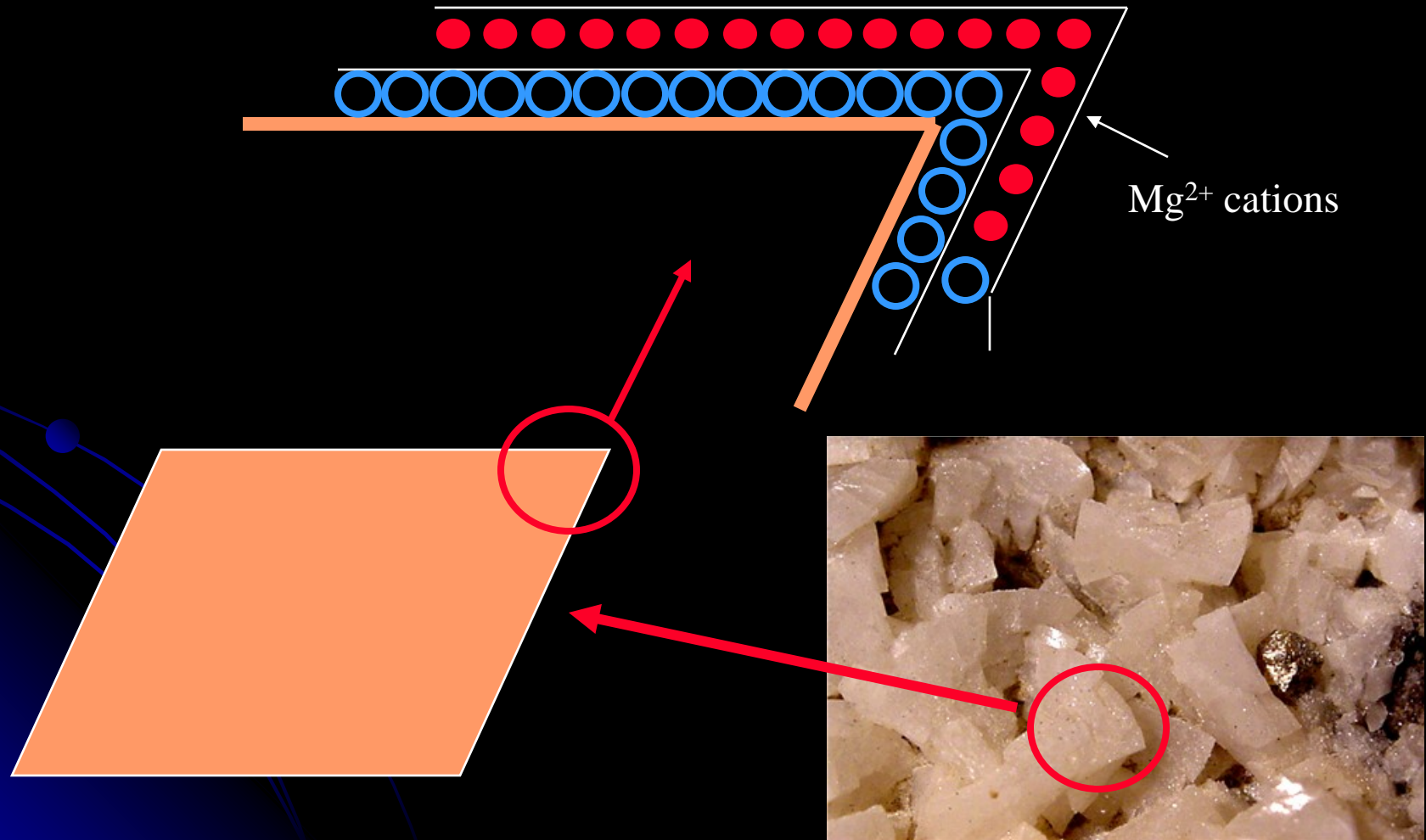
# Polymorphs & Polymorphism



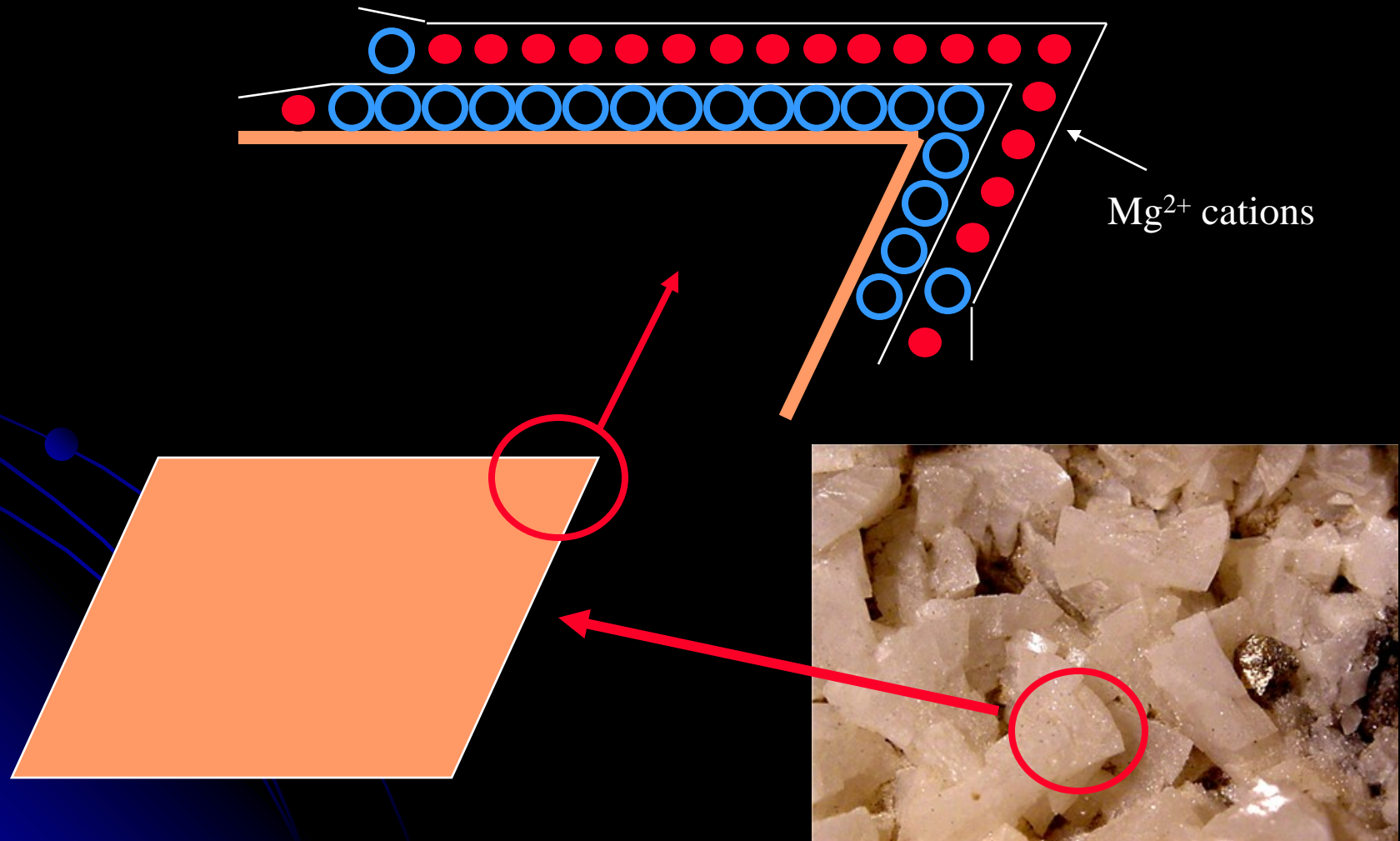
# Polymorphs & Polymorphism



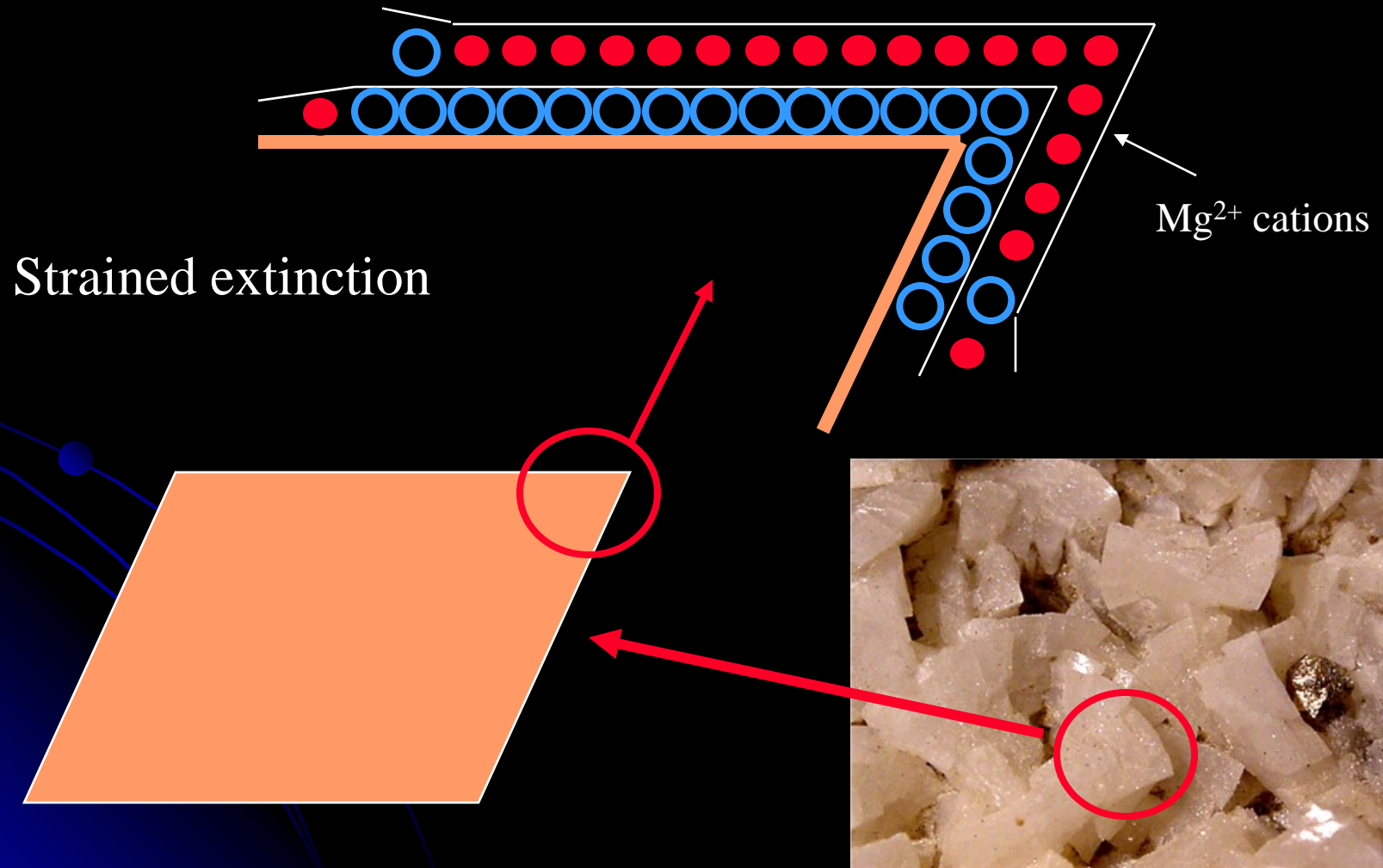
# Polymorphs & Polymorphism



# Polymorphs & Polymorphism



# Polymorphs & Polymorphism



# Polymorphs & Polymorphism

---

Polymorphic transformations can occur in one of 3 ways:

3) **Order-Disorder transformations**: all crystals have a certain amount of disorder to them (e.g., “saddle” dolomite) and in some cases, polymorphic transitions can occur once an appropriate level of order-disorder is attained. But...

- ... there is no threshold temperature or pressure for the transformation.

e.g., potassium feldspars



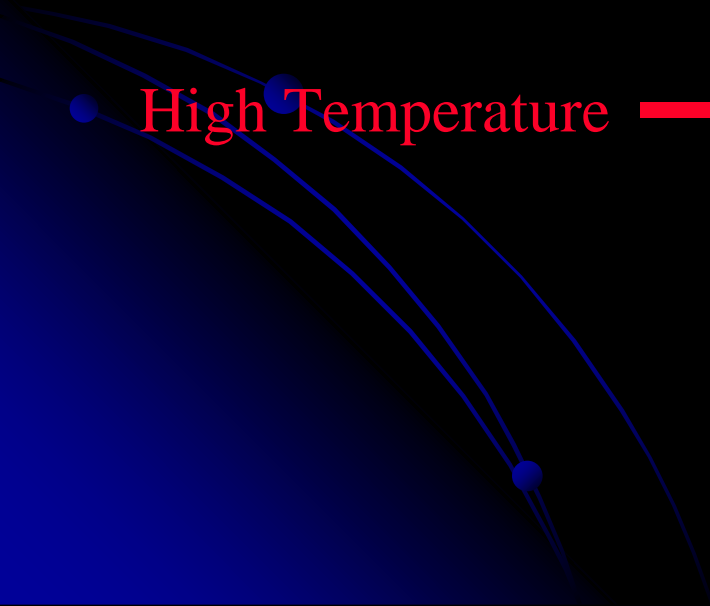
# Polymorphs & Polymorphism

---



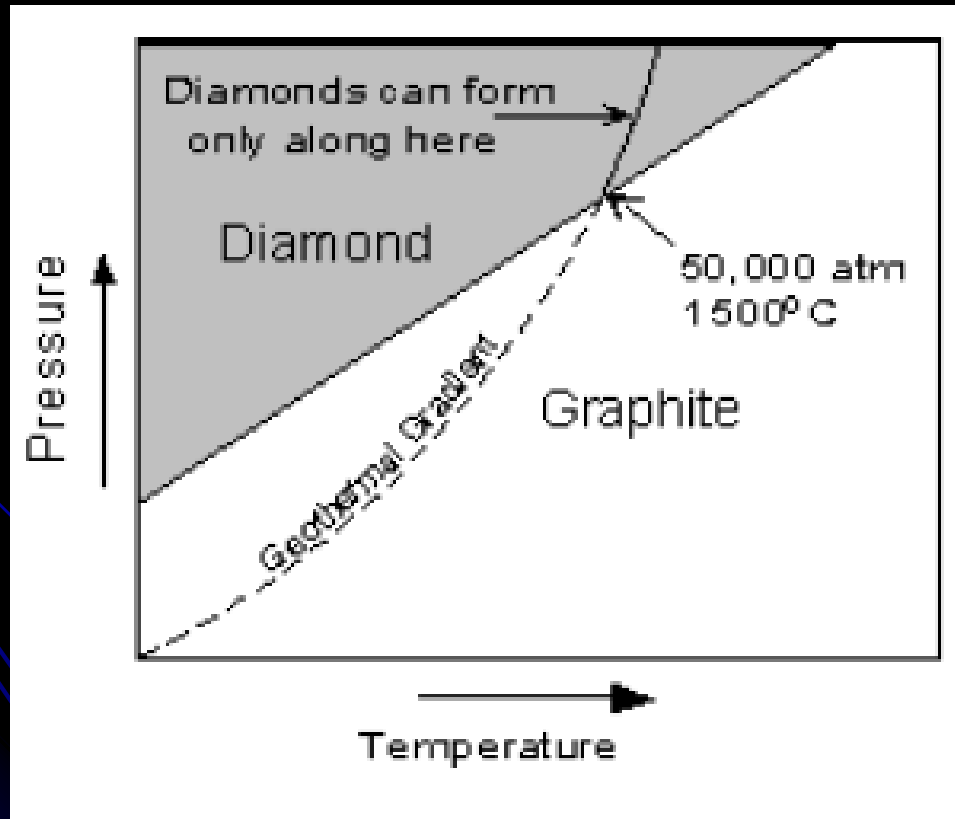
Sanidine  $\longrightarrow$  Orthoclase  $\longrightarrow$  Microcline  
(triclinic) (monoclinic) (monoclinic)

High Temperature  $\longrightarrow$  Low Temperature



# Polymorphs & Polymorphism

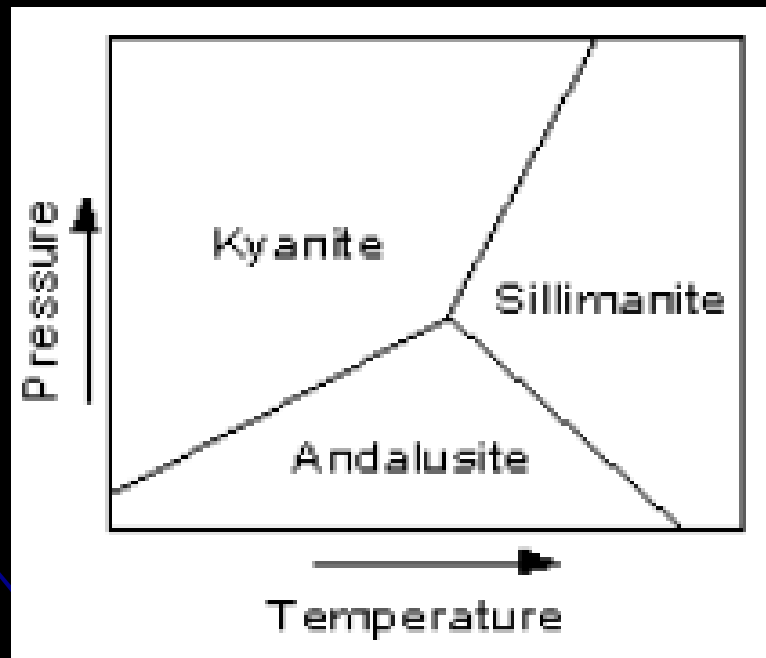
Examples of Polymorphs: 1) C- Diamond-Graphite  
(Reconstructive)



# Polymorphs & Polymorphism

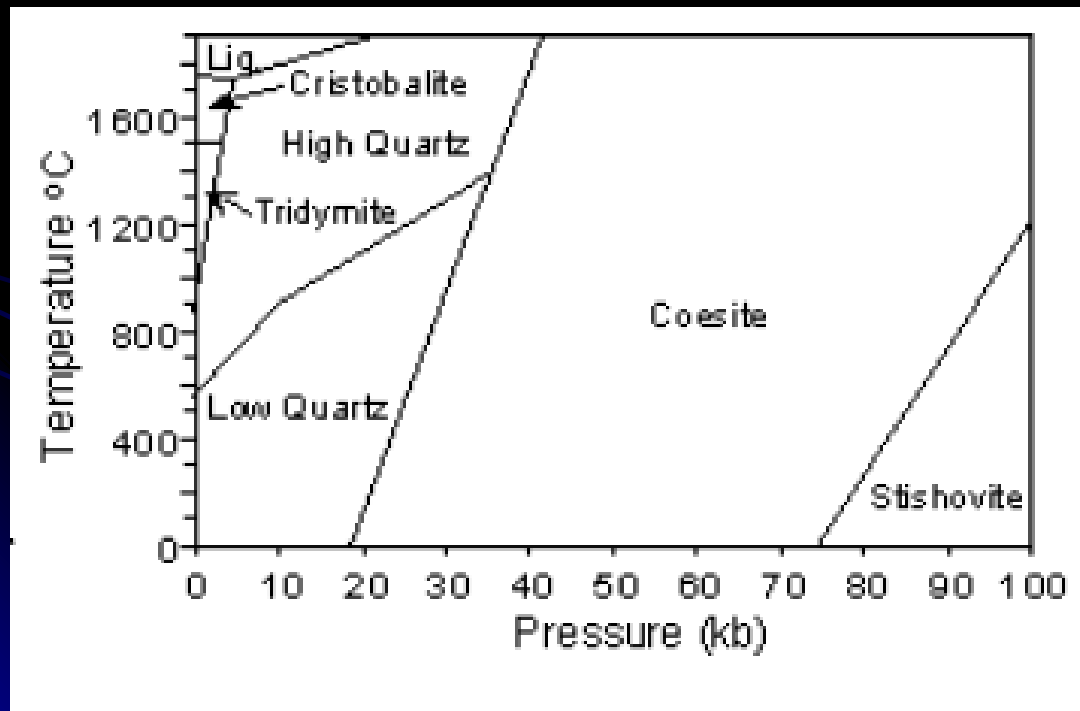
Examples of Polymorphs: 2)  $\text{Al}_2\text{SiO}_5$

Kyanite-Sillimanite-Andalusite (Reconstructive)



# Polymorphs & Polymorphism

Examples of Polymorphs: 3)  $\text{SiO}_2$   
(all reconstructive except  $\alpha$  to  $\beta$  quartz)



**Cristobalite:** isometric

**Tridymite:** hexagonal

**$\alpha$ ,  $\beta$ :** hexagonal

**Coesite:** monoclinic

**Stishovite:** tetragonal

# Other Crystal “Structures”

---

**Metamict Minerals:** crystals with partially “nuked” lattices (e.g., zircon;  $\text{Zr(U,Th)SiO}_4$ ).

**Mineraloids:** Amorphous minerals that do not have crystalline form (e.g., limonite:  $\text{FeOOH} \cdot n\text{H}_2\text{O}$ )

- **Pseudomorph:** A mineral which has the crystal form of another mineral (nasty, sneaky minerals in disguise)- several types

# Other Crystal “Structures”

---

Pseudomorphs: Form due to one of three mechanisms:



Dolomite crystals mimicking aragonite <http://www.mindat.org/gallery-4223.html>

# Other Crystal “Structures”

---

Pseudomorphs: Form due to one of three mechanisms:

- 1) **Substitution**: molecule by molecule replacement of one mineral by another (e.g., quartz replacement of aragonite)
- 2) **Encrustation**: a coating of a new mineral top of an existing mineral (then the first mineral is dissolved; e.g., hematite on pyrite)
- 3) **Alteration**: Partial removal of one mineral and infilling by a second can also result in pseudomorphing (e.g., anhydrite in gypsum)

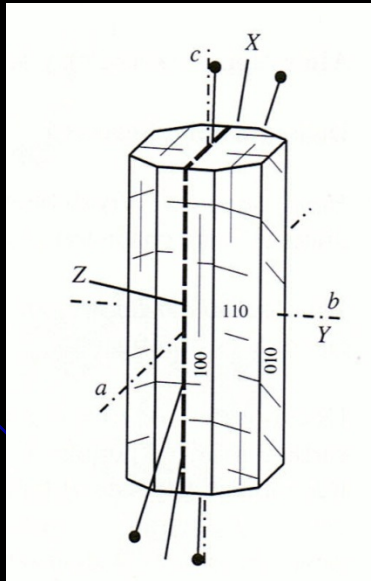
Terminology: anhydrite pseudomorphed after gypsum

# Mineral Habits

Habit: The external appearance of a mineral (not necessarily the same as crystal form)

e.g., malachite  
 $\text{Cu}_2\text{CO}_3(\text{OH})_2$

(monoclinic:  $2/m$ )

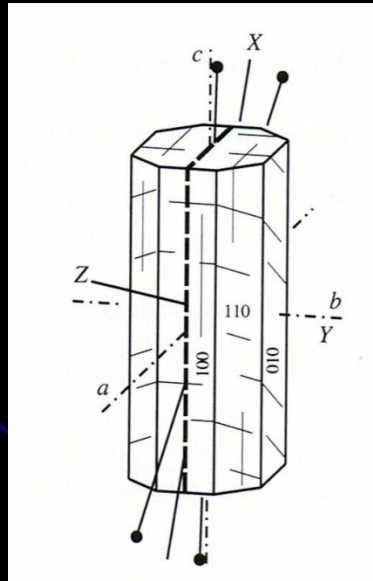


# Mineral Habits

Habit: The external appearance of a mineral (not necessarily the same as crystal form)

e.g., malachite  
 $\text{Cu}_2\text{CO}_3(\text{OH})_2$

(monoclinic:  $2/m$ )



# Today's Homework

---

1. Assignments 2 and 3 now next week
2. SI Session request by email (first come; first serve)
3. Study and organize: crystallography exam (take home) issued next Tuesday

## Next Time

---

1. Optical Microscopy

# GY 302: Crystallography and Mineralogy

---

## Lecture 6: Polymorphism and Crystal Habit

Instructor: Dr. Doug Haywick

[dhaywick@southalabama.edu](mailto:dhaywick@southalabama.edu)

This is a free open access lecture, but not for commercial purposes.

For personal use only.