

# *Viscoelasticity of Supramolecular Center-functionalized Polymer*



*Effect of the **strength of Hydrogen Bonding** Stickers*

***Xavier Callies***

*Costantino Creton, Guylaine Ducouret*

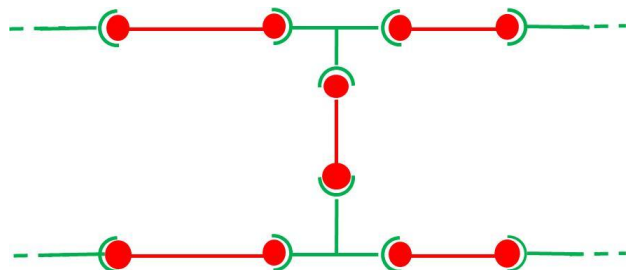
*AERC 2015 - Nantes*



# Supramolecular Polymers ?

## Supramolecular Chemistry

Self-assembly of small molecules by **non covalent** bonds (H-bonds, ionic...) in solution

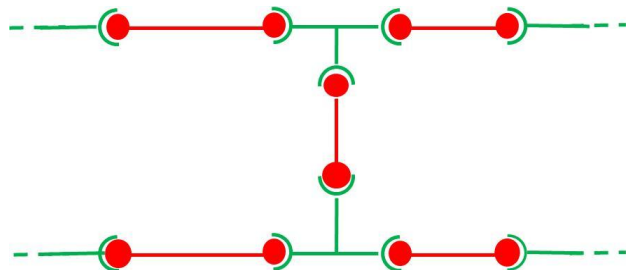


Lehn, J. *Angew. Chem.-Int. Ed. Engl.* **1990**, 29, 1304–1319.

# Supramolecular Polymers ?

## Supramolecular Chemistry

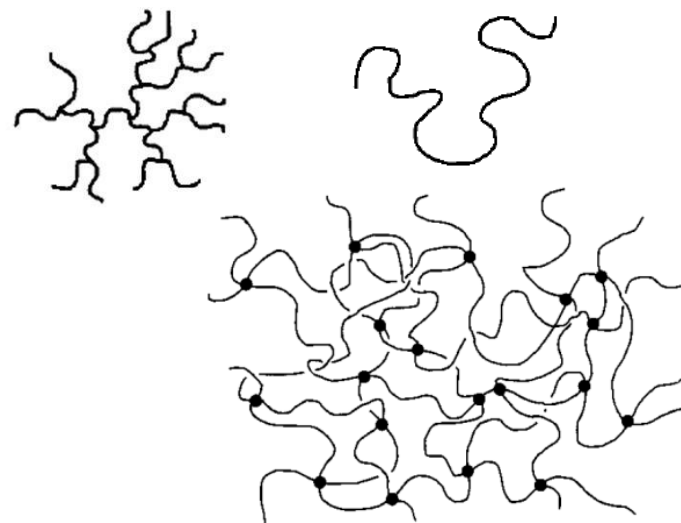
Self-assembly of small molecules by **non covalent** bonds (H-bonds, ionic...) in solution



Lehn, J. *Angew. Chem.-Int. Ed. Engl.* **1990**, 29, 1304–1319.

## Polymer Physics

Association by **covalent** bonds of monomers

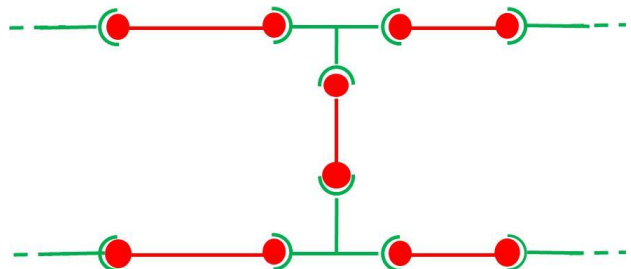


Rubinstein, M.; Colby, R. H. *Polymer Physics*; Oxford University Press, 2003.

# Supramolecular Polymers ?

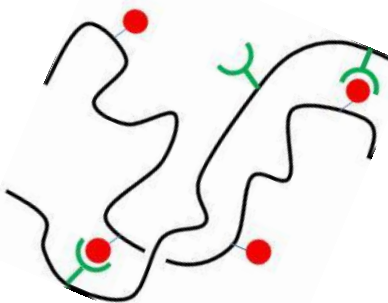
## Supramolecular Chemistry

Self-assembly of small molecules by **non covalent** bonds (H-bonds, ionic...) in solution



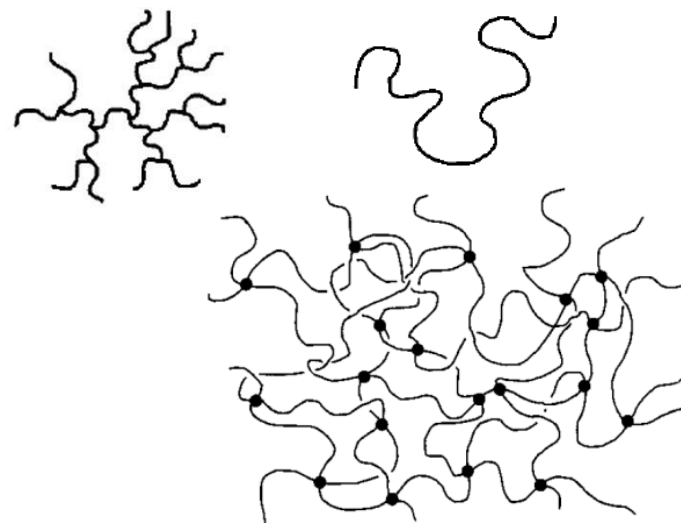
Lehn, J. *Angew. Chem.-Int. Ed. Engl.* **1990**, 29, 1304–1319.

## Supramolecular Polymers



## Polymer Physics

Association by **covalent** bonds of monomers

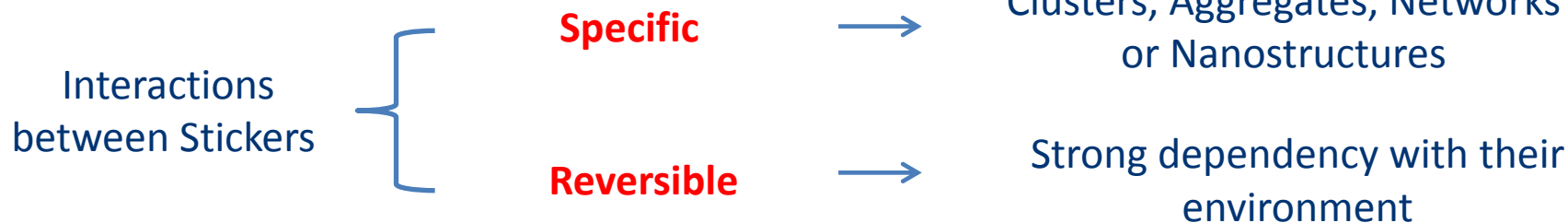


Rubinstein, M.; Colby, R. H. *Polymer Physics*; Oxford University Press, 2003.

## Interest of Supramolecular Polymers ?

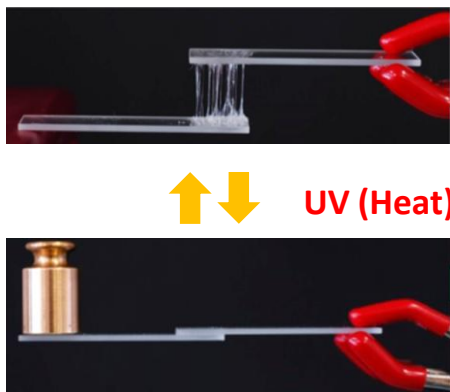


# Interest of Supramolecular Polymers ?



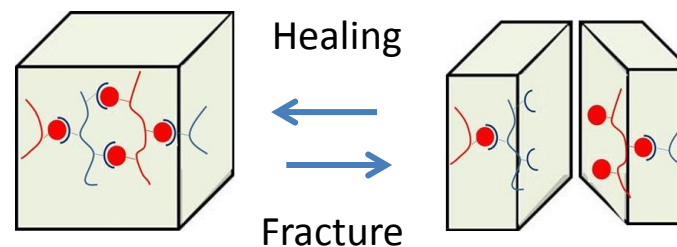
→ *A wide diversity of applications*

## Stimuli-responsive Materials



Heinzmann, C. et al,  
*ACS Appl. Mater. Inter.*  
**2014**, 6,4713–4719.

## Self-healing by simple contact at RT

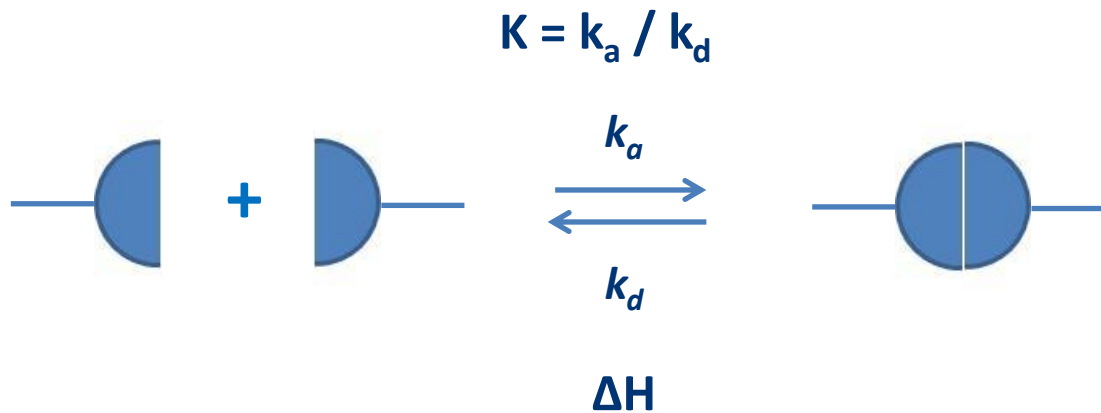


Cordier, P. et al, *Nature* **2008**, 451, 977–980.

## Challenge for all applications

How to link the **rheological** properties and the **chemical structure** of the polymer chain ?

→ What is the effect of the **strength** of stickers on the rheological behavior ?

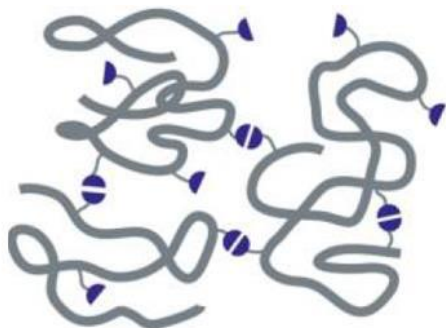


Strength of Stickers

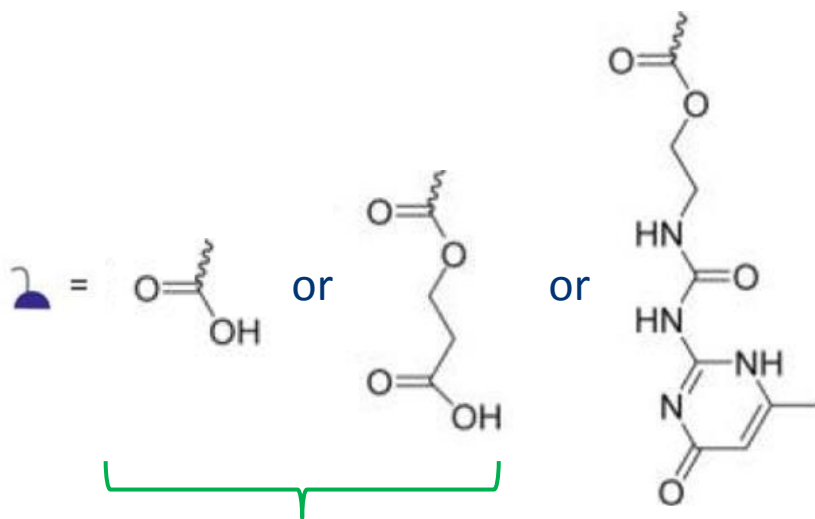


Chemistry of stickers

Polarity of Polymer Matrix

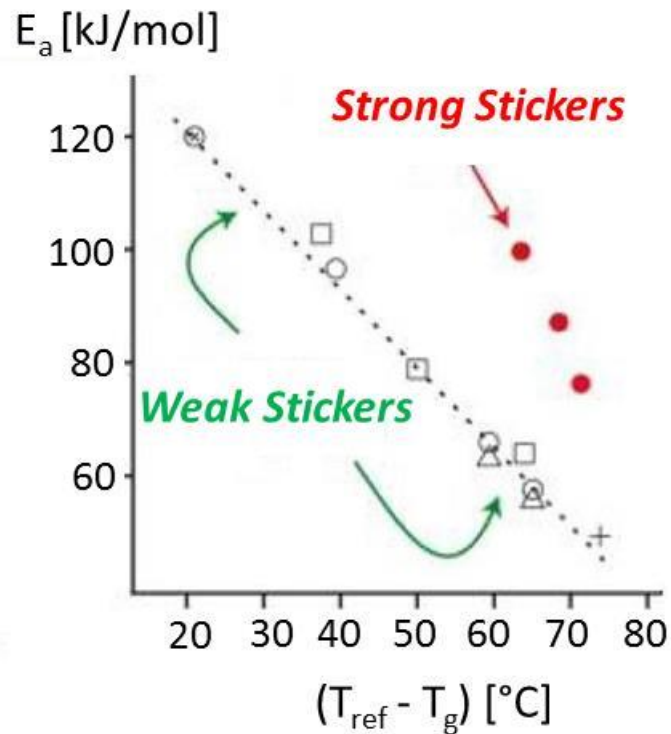


Poly(butylacrylate)  
Copolymers

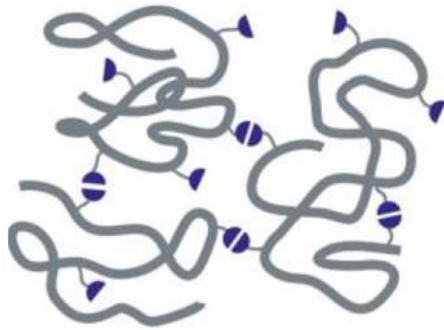


**Weak**  
Stickers

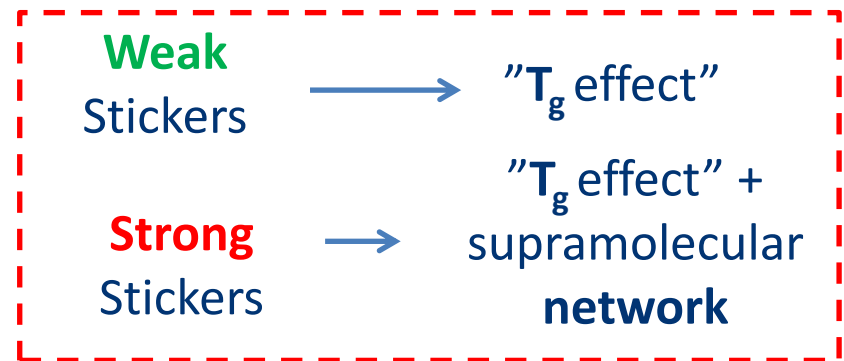
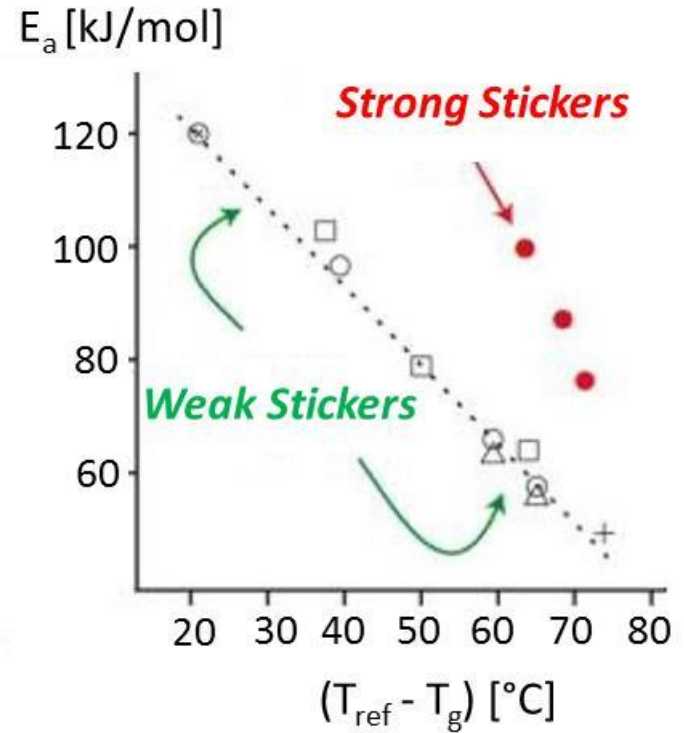
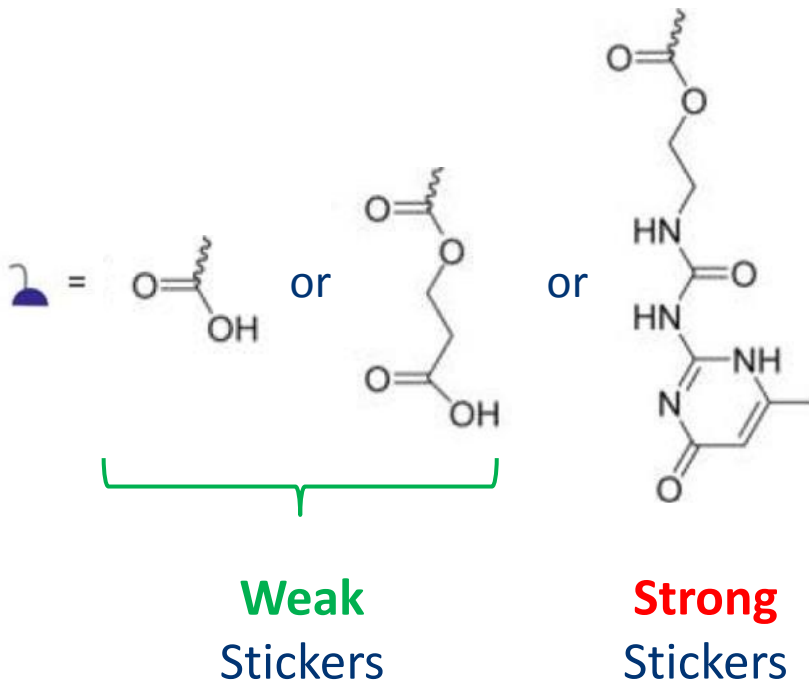
**Strong**  
Stickers



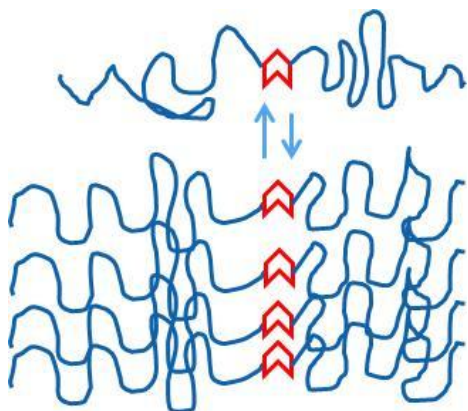




Poly(butylacrylate)  
Copolymers



## Center-functionalized Polymers ?



**Self-assembly** of stickers  
into **filaments** is favored

Bouteiller, L. *et al.*. *Adv. Funct. Mater.* 20, 1803–1811 (2010).

Frauenrath, H. *et al.* *Chem. Sci.* 3, 1512-1521 (2012)

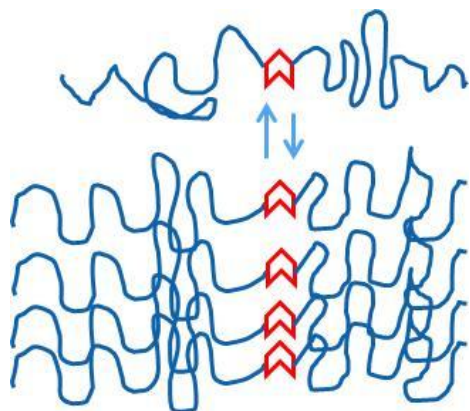


**Predictable shape**  
of the aggregates



Link the rheology and  
the supramolecular Chemistry

## Center-functionalized Polymers ?



**Self-assembly** of stickers  
into **filaments** is favored

Bouteiller, L. *et al.*. *Adv. Funct. Mater.* 20, 1803–1811 (2010).

Frauenrath, H. *et al.* *Chem. Sci.* 3, 1512-1521 (2012)



**Predictable shape**  
of the aggregates



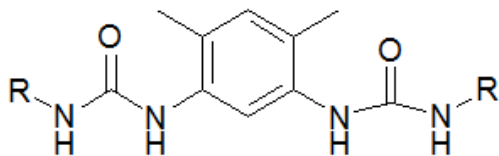
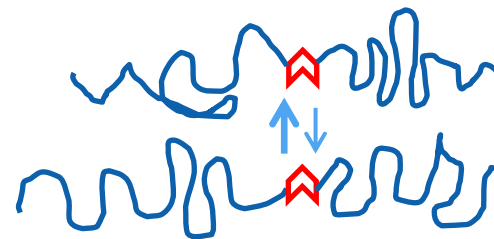
Link the rheology and  
the supramolecular Chemistry

### Strategy of our Project

- Synthesis of **monodisperse and linear** center-functionalized polymers
- Change the molecular parameters in a **highly controlled** way
- Systematic characterization of the **nanostructure** and linear **rheology**.

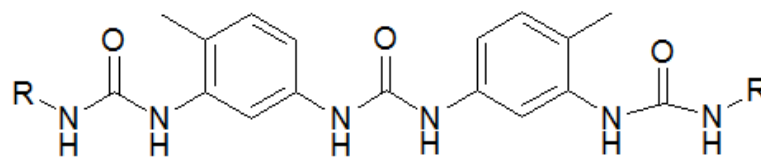
## Key-molecular parameters

1. Level of interactions : Two **hydrogen bonding** stickers.



Bis-urea Xylene

”**Weak** Sticker”

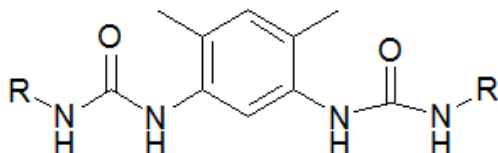
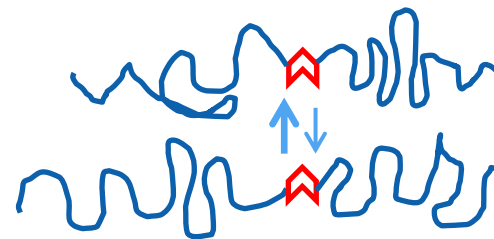


Tri-Urea Toluene

”**Strong** Sticker”

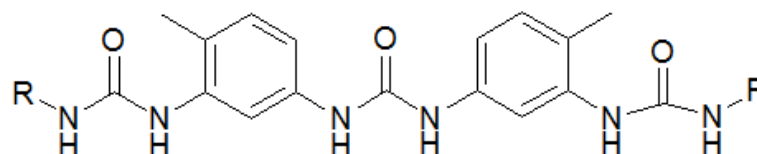
## Key-molecular parameters

1. Level of interactions : Two **hydrogen bonding** stickers.



Bis-urea Xylene

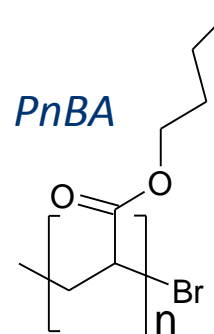
”Weak Sticker”



Tri-Urea Toluene

”Strong Sticker”

2. the interacting moieties density : the **size** of the linear non polar chains

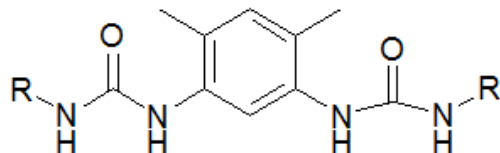
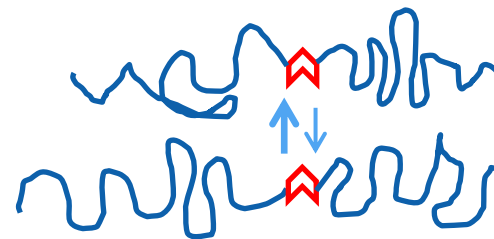


$$I_p \leq 1,4$$

$$5 \text{ kg/mol} \leq M_w \leq 100 \text{ kg/mol} \longleftrightarrow 4\% \geq \text{Sticker (w\%)} \geq 0,2\%$$

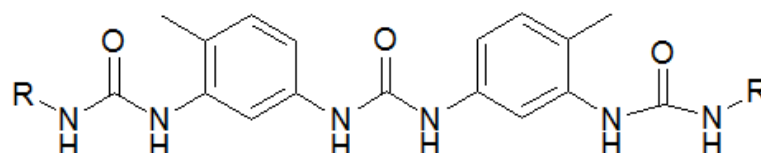
## Key-molecular parameters

1. Level of interactions : Two **hydrogen bonding** stickers.



Bis-urea Xylene

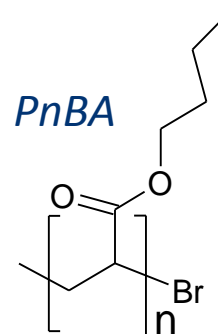
"Weak Sticker" **2<sup>nd</sup> Part**



Tri-Urea Toluene

"Strong Sticker" **1<sup>st</sup> Part**

2. the interacting moieties density : the **size** of the linear non polar chains



$$I_p \leq 1,4$$

$$5 \text{ kg/mol} \leq M_w \leq 100 \text{ kg/mol} \longleftrightarrow 4\% \geq \text{Sticker (w\%)} \geq 0,2\%$$

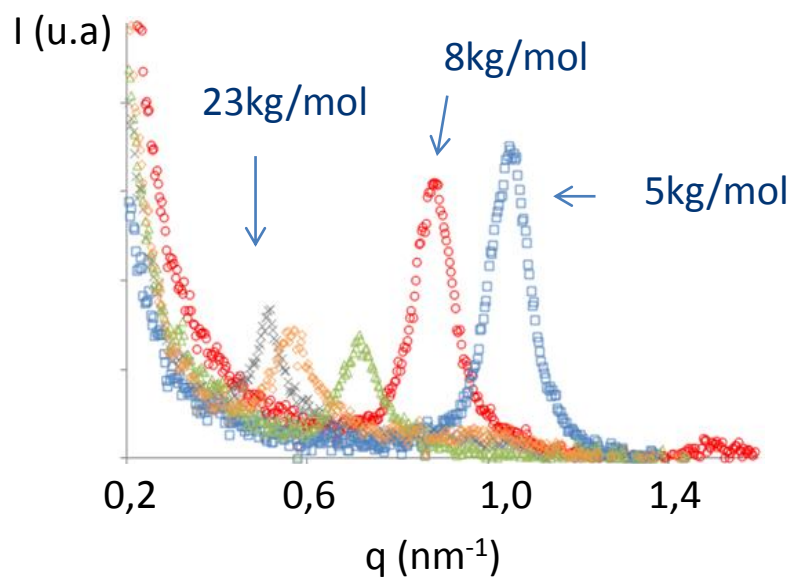
## $T_g$ and Nanostructure for **Strong** Stickers

- No variation of  $T_g$  with  $M_n$  (between 5 and 100kg/mol):  $T_g = -49 \pm 1^\circ\text{C}$  (DSC)

## $T_g$ and Nanostructure for **Strong** Stickers

- No variation of  $T_g$  with  $M_n$  (between 5 and 100kg/mol):  $T_g = -49 \pm 1^\circ\text{C}$  (DSC)

### SAXS Investigation



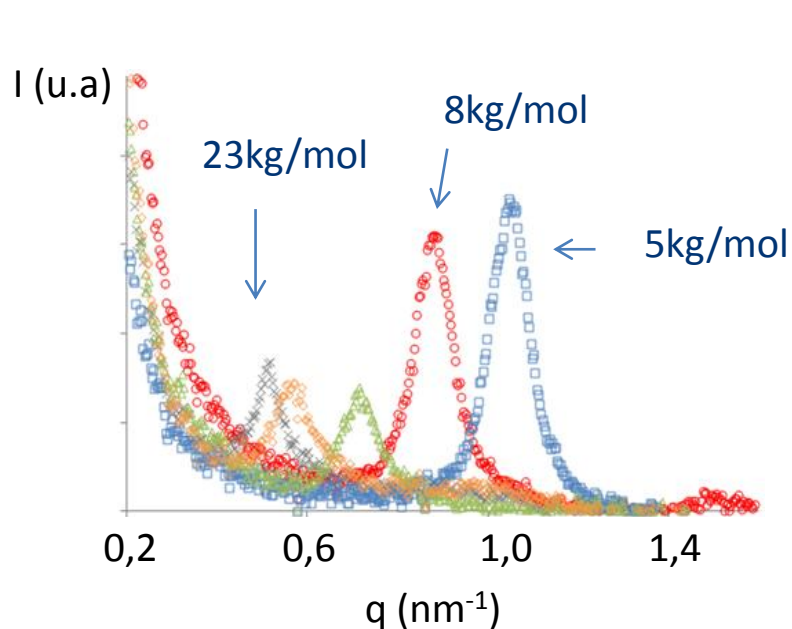
No peak for  $M_n \geq 40 \text{ kg/mol}$



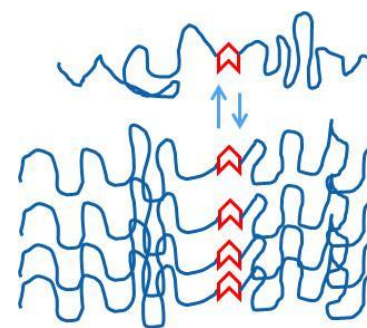
## $T_g$ and Nanostructure for **Strong** Stickers

- No variation of  $T_g$  with  $M_n$  (between 5 and 100kg/mol):  $T_g = -49 \pm 1^\circ\text{C}$  (DSC)

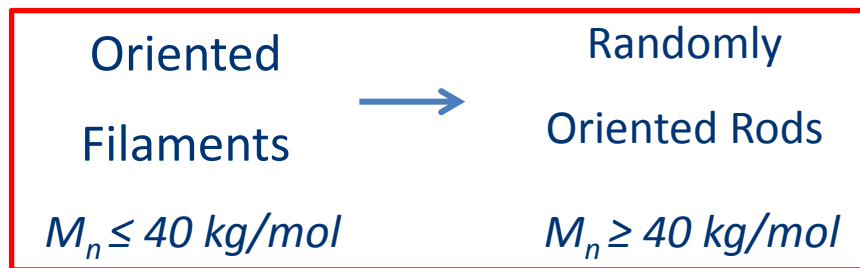
### SAXS Investigation



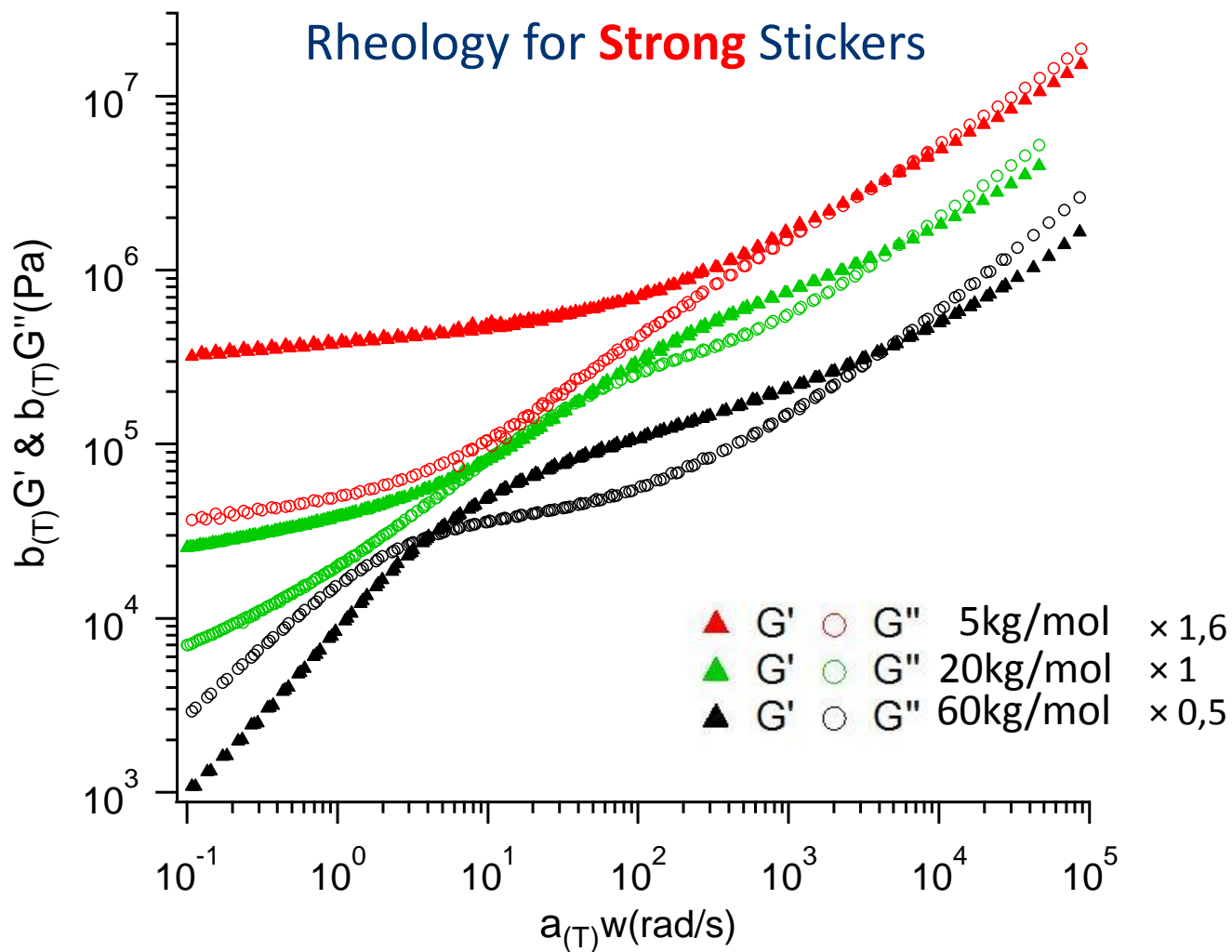
No peak for  $M_n \geq 40$  kg/mol

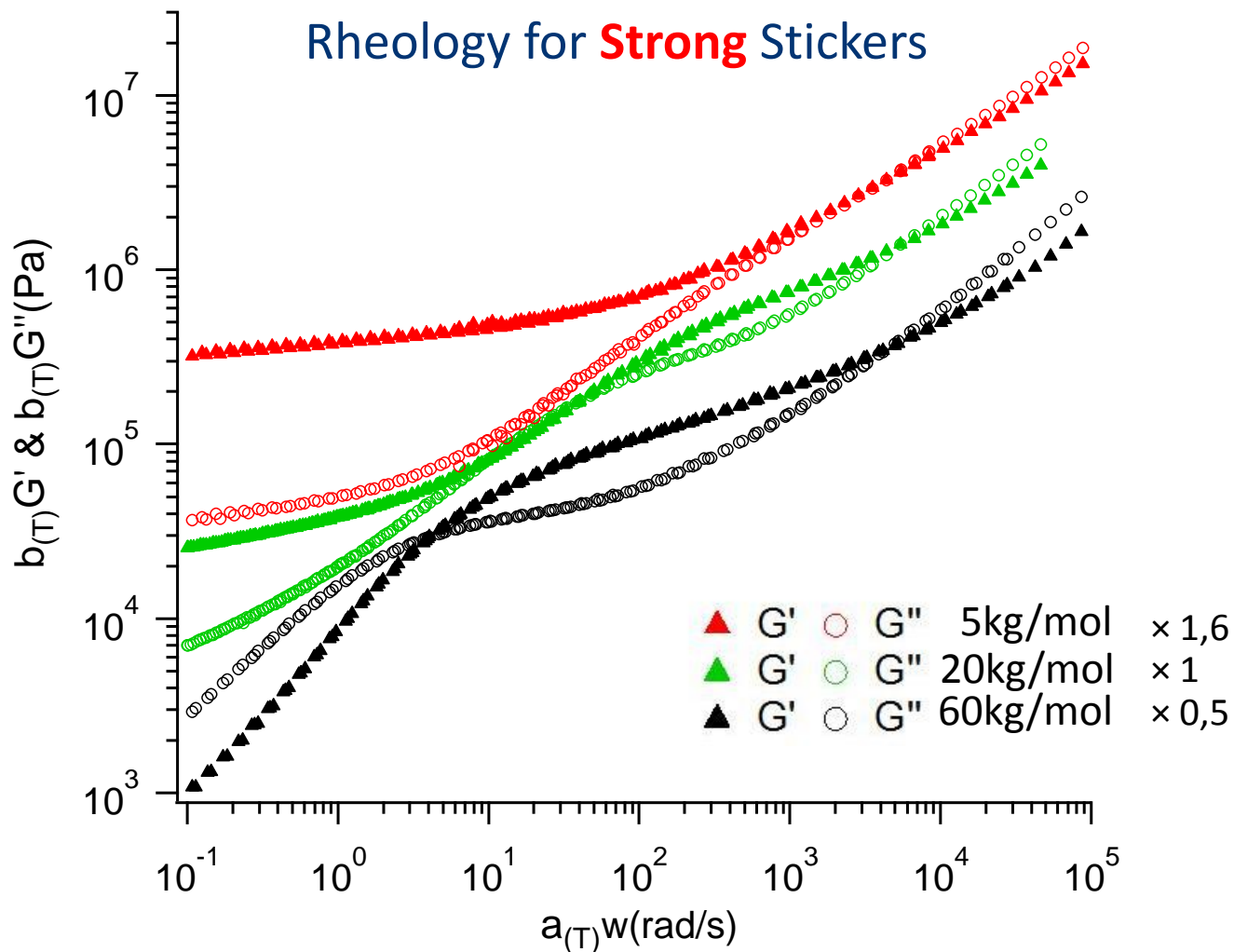


Self-assembly of  
stickers  
into filaments



# Rheology for **Strong** Stickers





$M_n \leq 40\text{kg/mol}$

$M_n \geq 40\text{kg/mol}$

**Oriented  
Filaments**

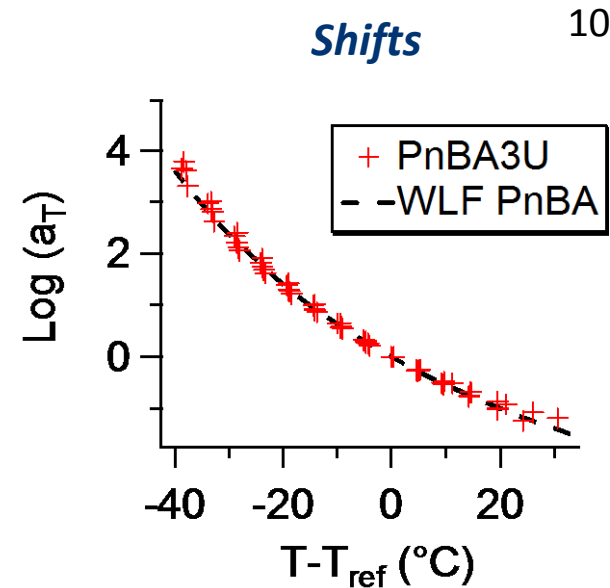
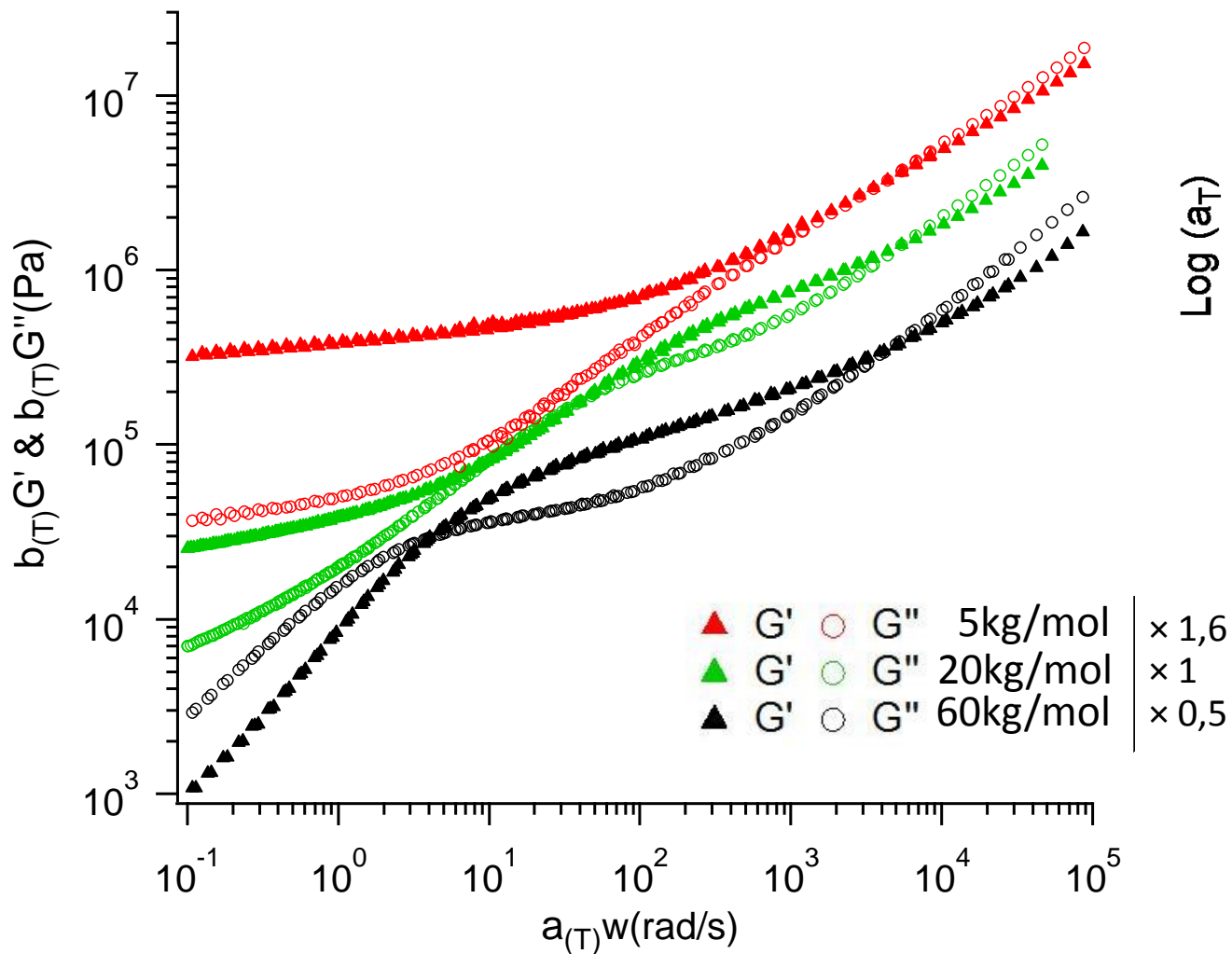


Dissipative  
**Gel**

**Randomly  
Oriented Rods**

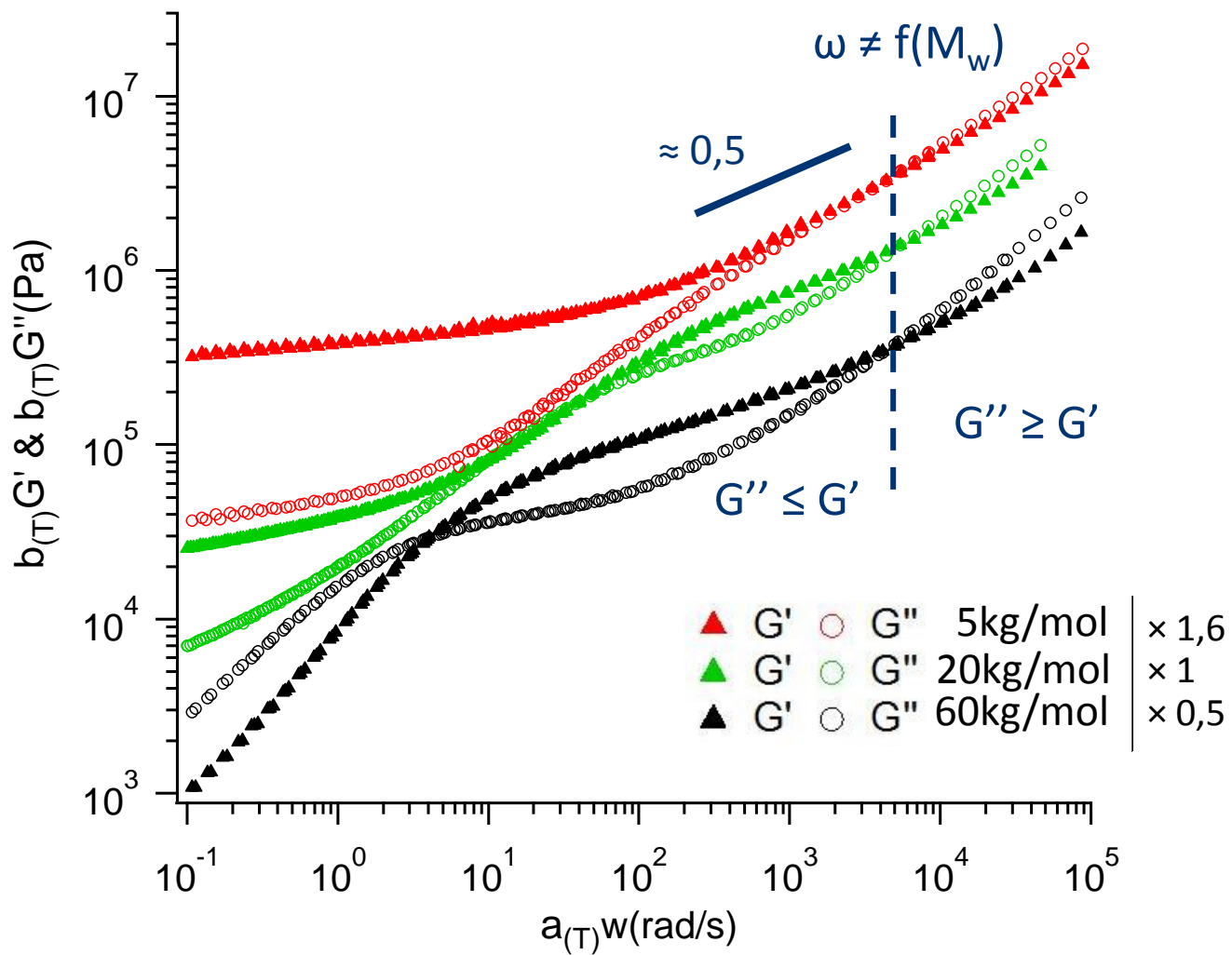


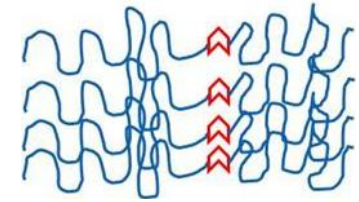
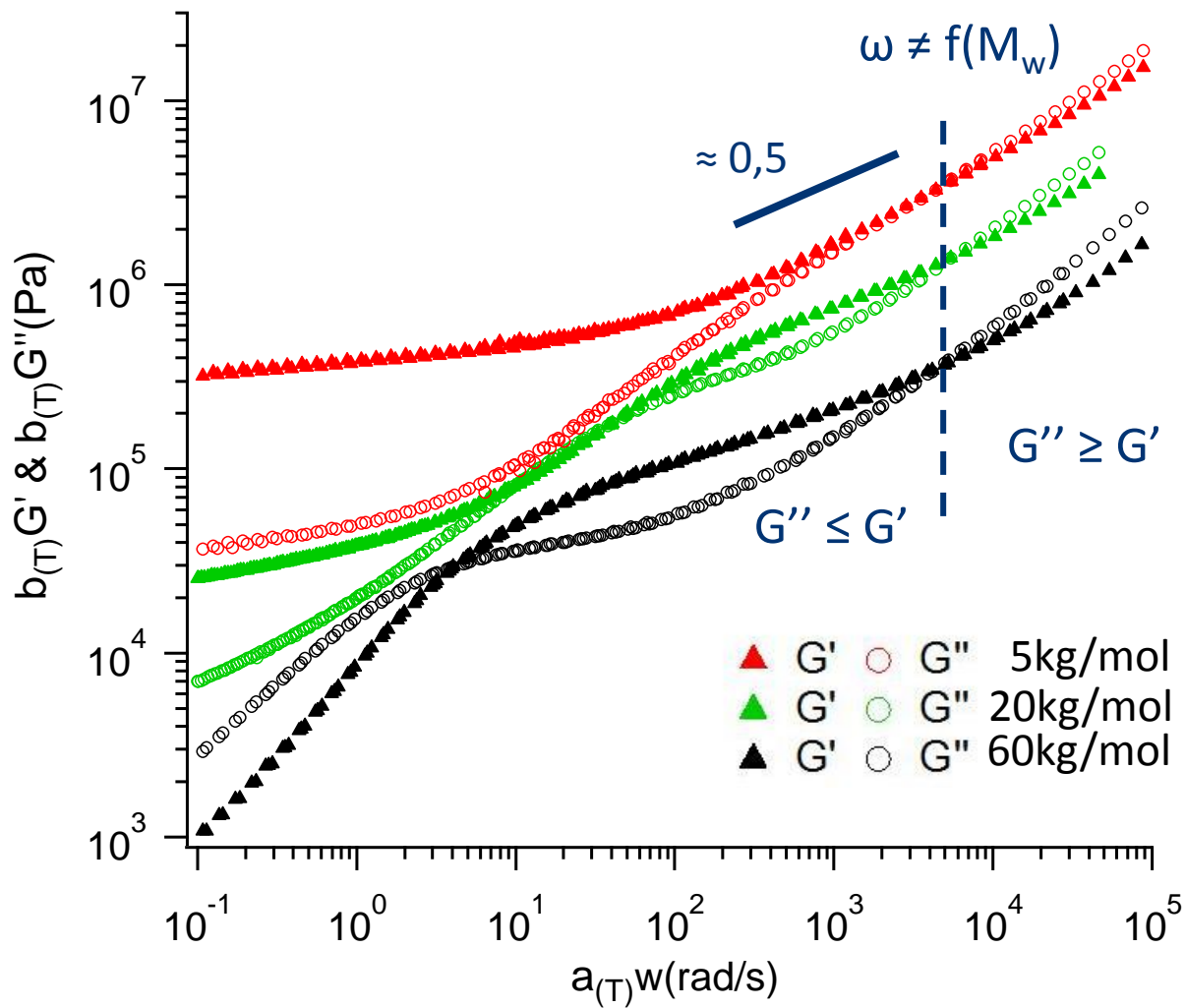
Viscoelastic  
**Fluid**



« Frozen » Aggregates

**Dissipative** Relaxation  
of the **side chains**

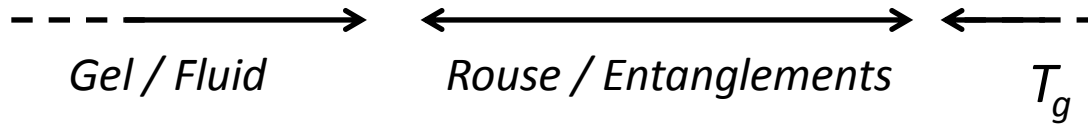




Comb-shaped Aggregates  
 $\approx$   
 Comb-shaped Polymers

**Self-assembly of Stickers**

**Polymer Matrix**



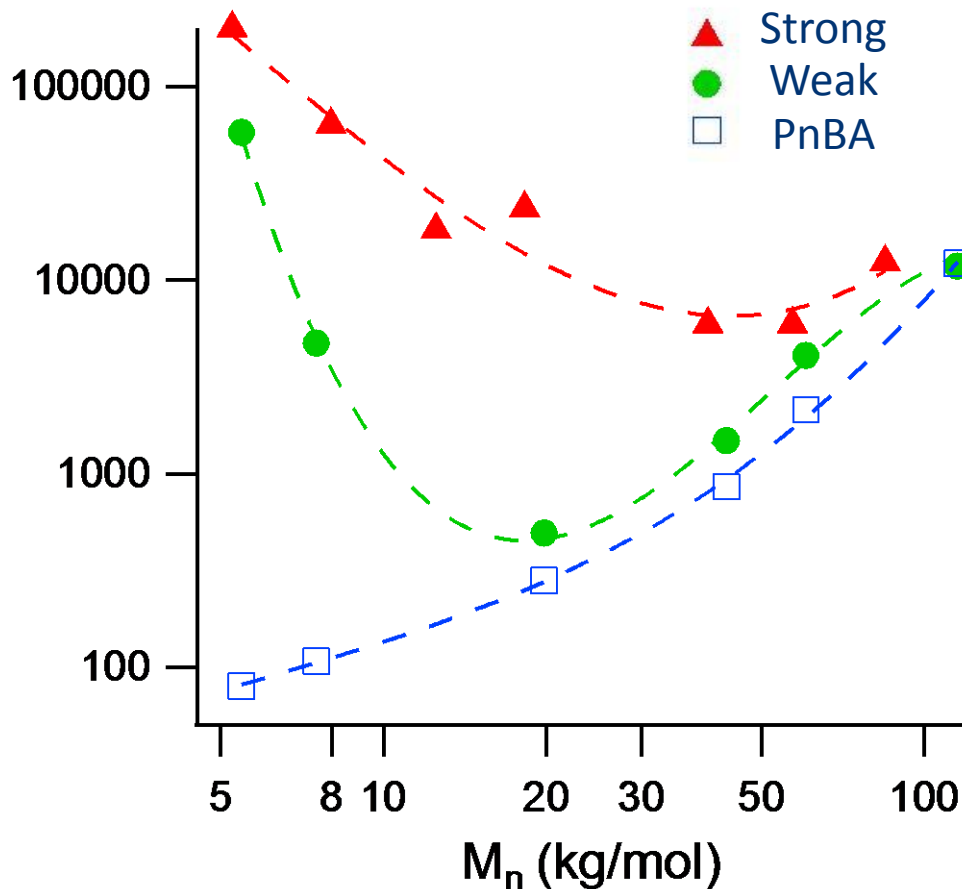
## **Strong** Stickers vs **Weak** Stickers ?

Molecular weight dependency ?

# Strong Stickers vs Weak Stickers ?

Molecular weight dependency ?

$\eta^*$  (1rad/s, T=25°C)



Critical Molecular weight ( $M_c$ ) ?

Stickers' Regime

$$M_n \leq M_c$$

Entanglements' Regime

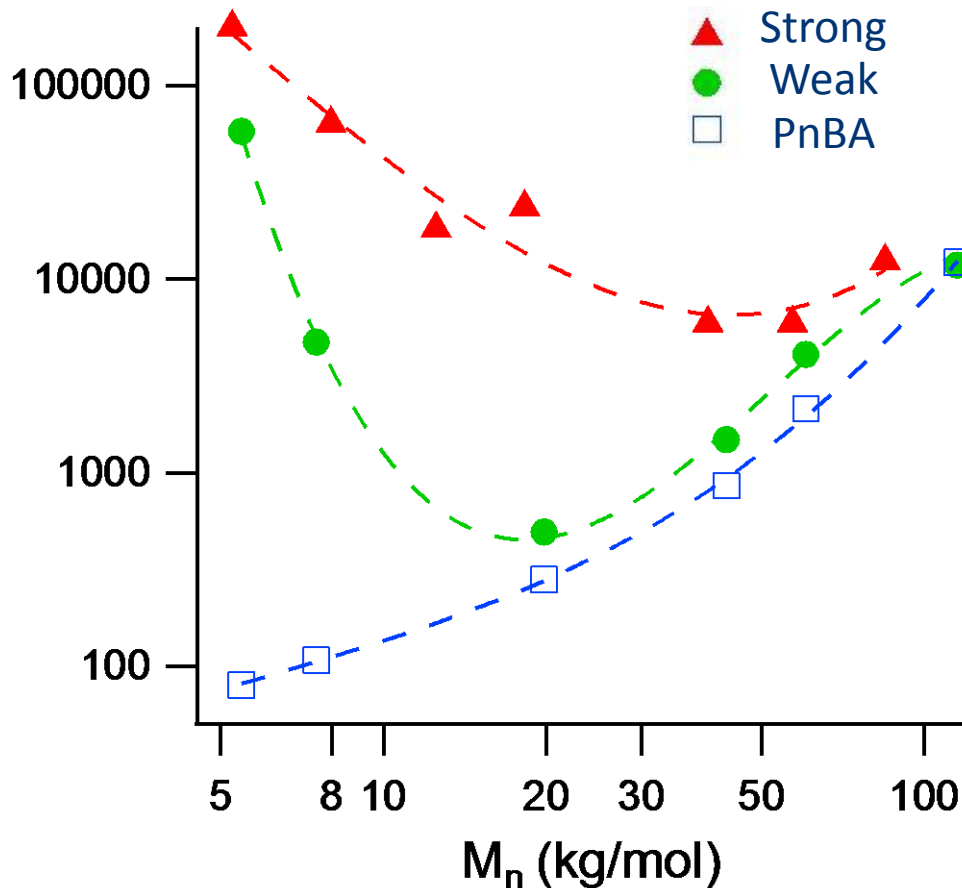
$$M_n \geq M_c$$



# Strong Stickers vs Weak Stickers ?

Molecular weight dependency ?

$\eta^*$  (1rad/s, T=25°C)



Critical Molecular weight ( $M_c$ ) ?

Stickers' Regime

$$M_n \leq M_c$$

Entanglements' Regime

$$M_n \geq M_c$$

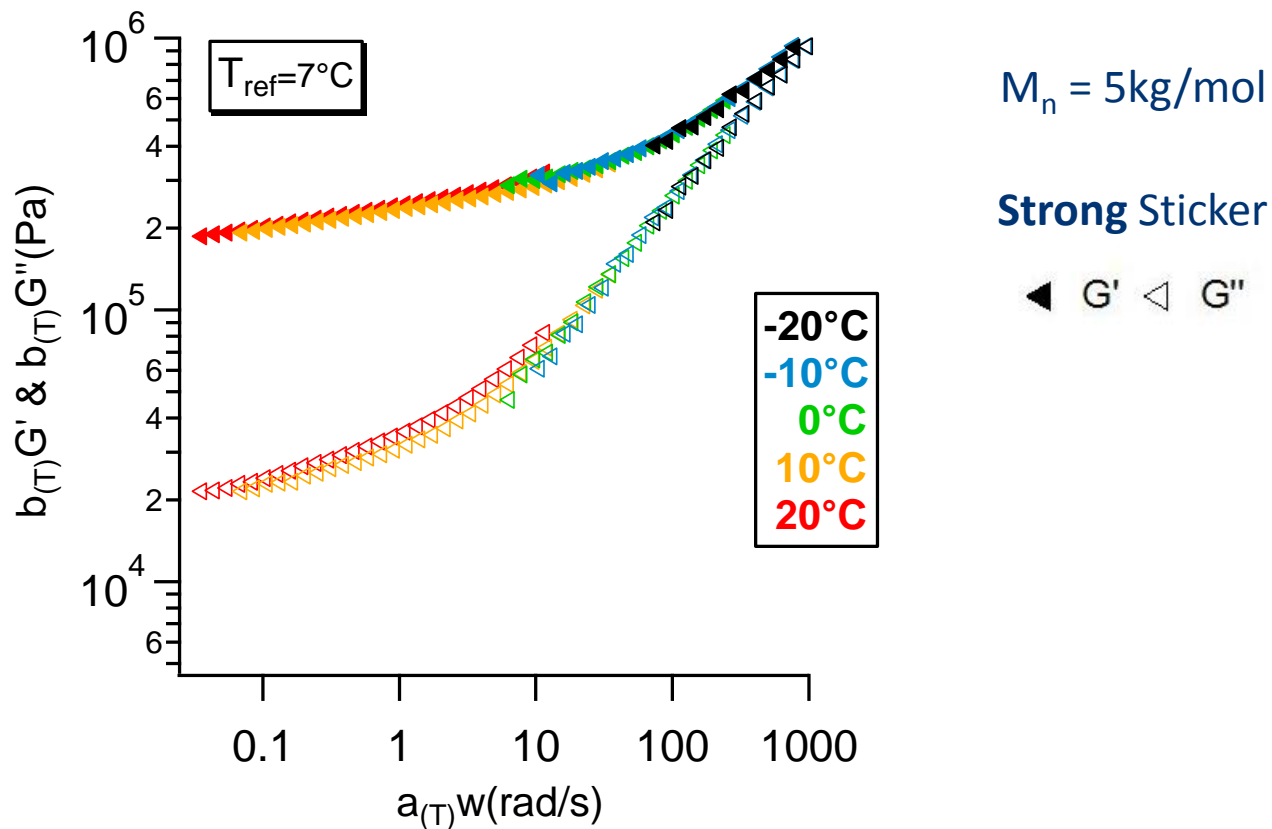
$M_c \approx 40 \text{ kg/mol}$

$M_c \approx 20 \text{ kg/mol}$

Strength  $\nearrow$ ,  $M_c \nearrow$

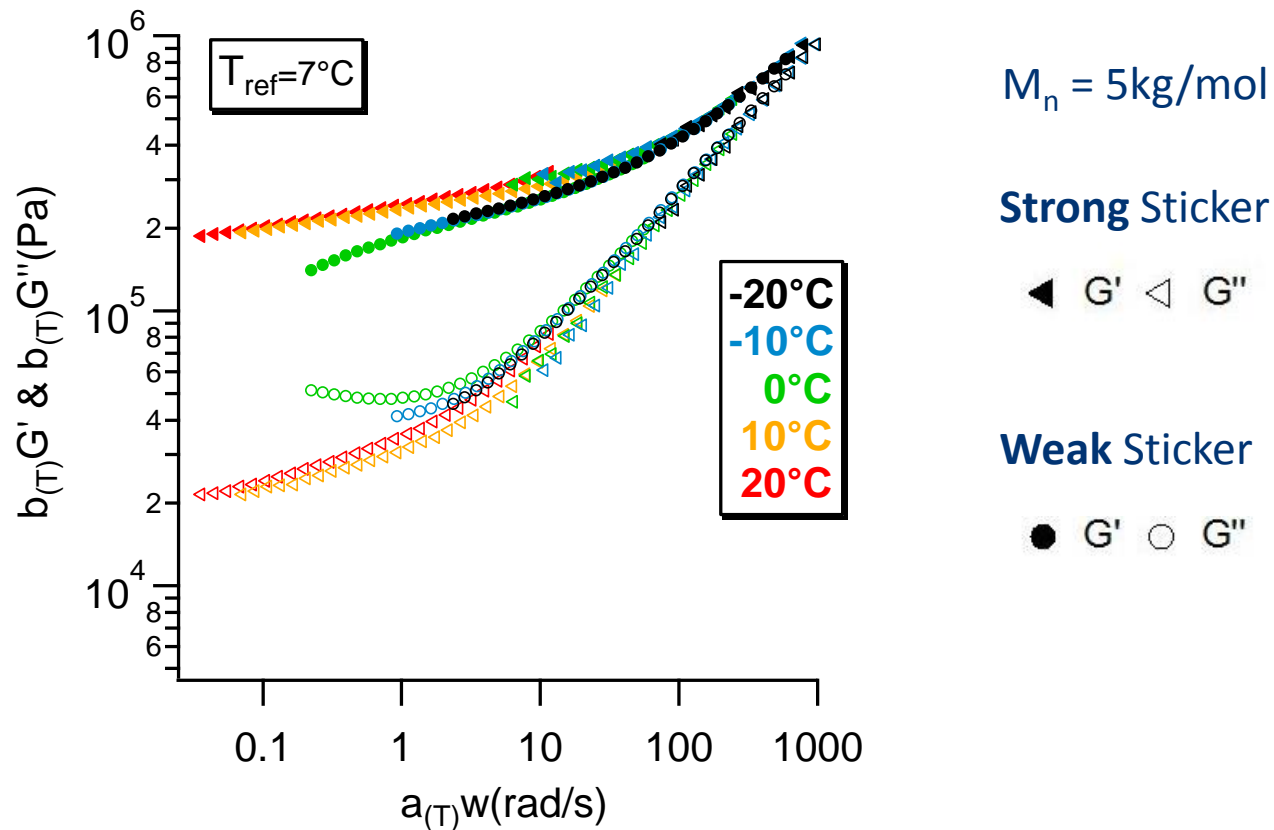
## Strong Stickers vs Weak Stickers ?

Stability of the nanostructure (below  $M_c$ ) ?



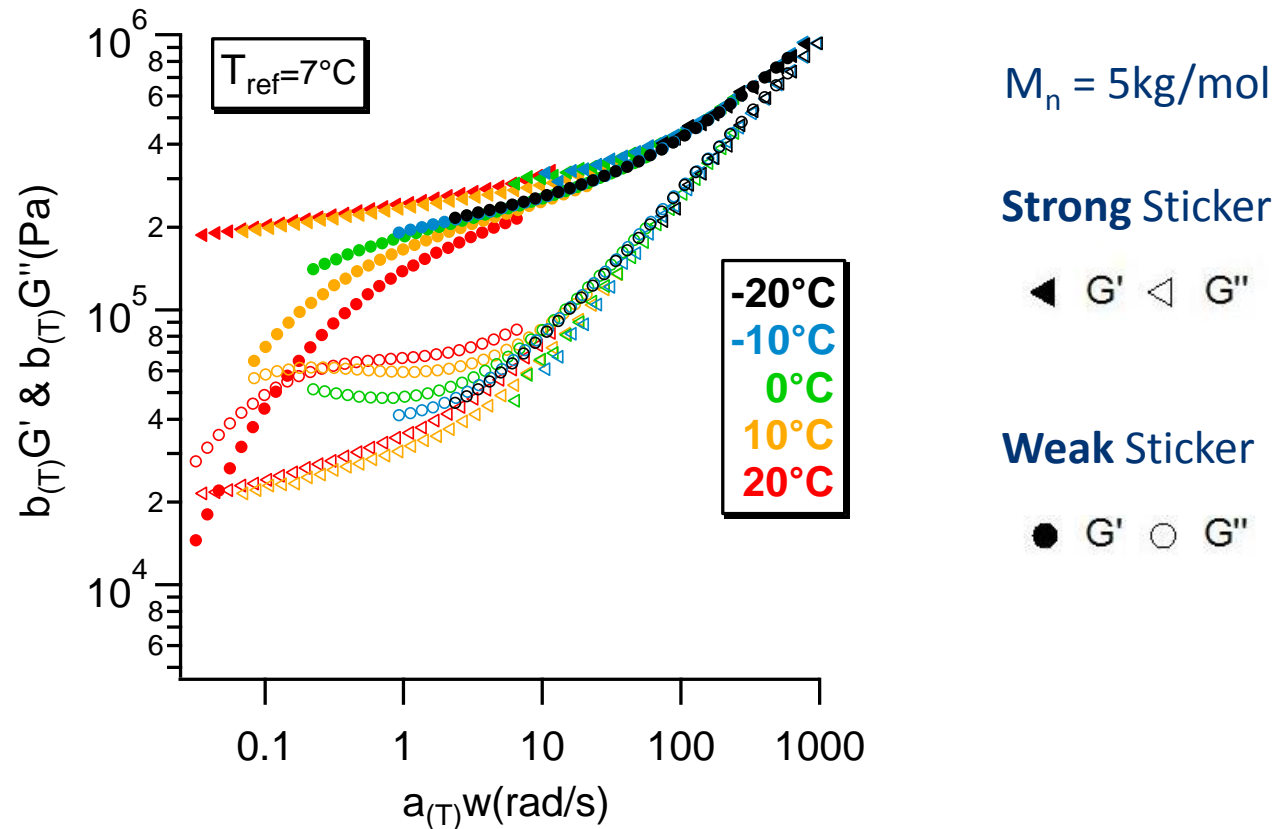
## Strong Stickers vs Weak Stickers ?

Stability of the nanostructure (below  $M_c$ ) ?

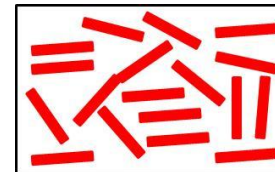
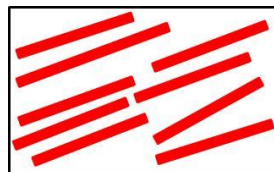


# Strong Stickers vs Weak Stickers ?

Stability of the nanostructure (below  $M_c$ ) ?



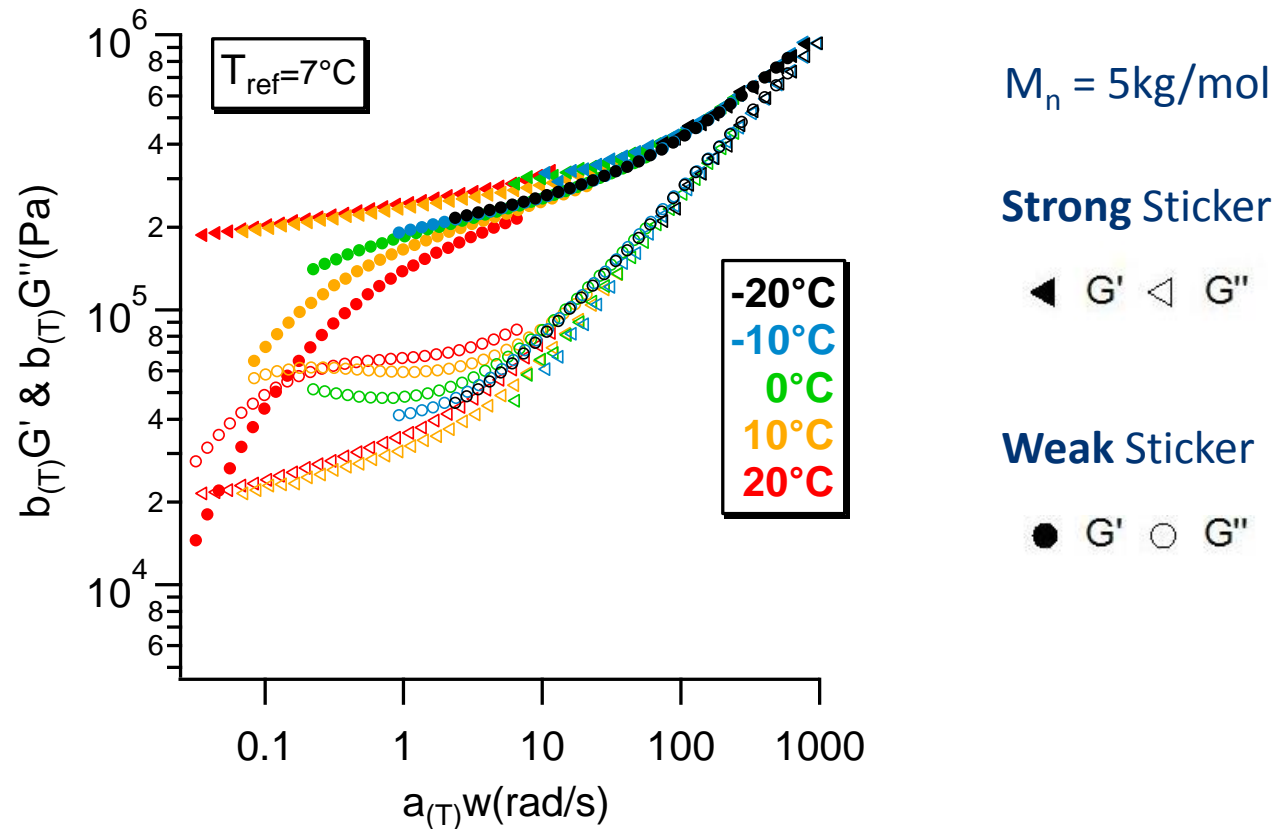
$T \leq T_{ODT}$



$T \geq T_{ODT}$

## Strong Stickers vs Weak Stickers ?

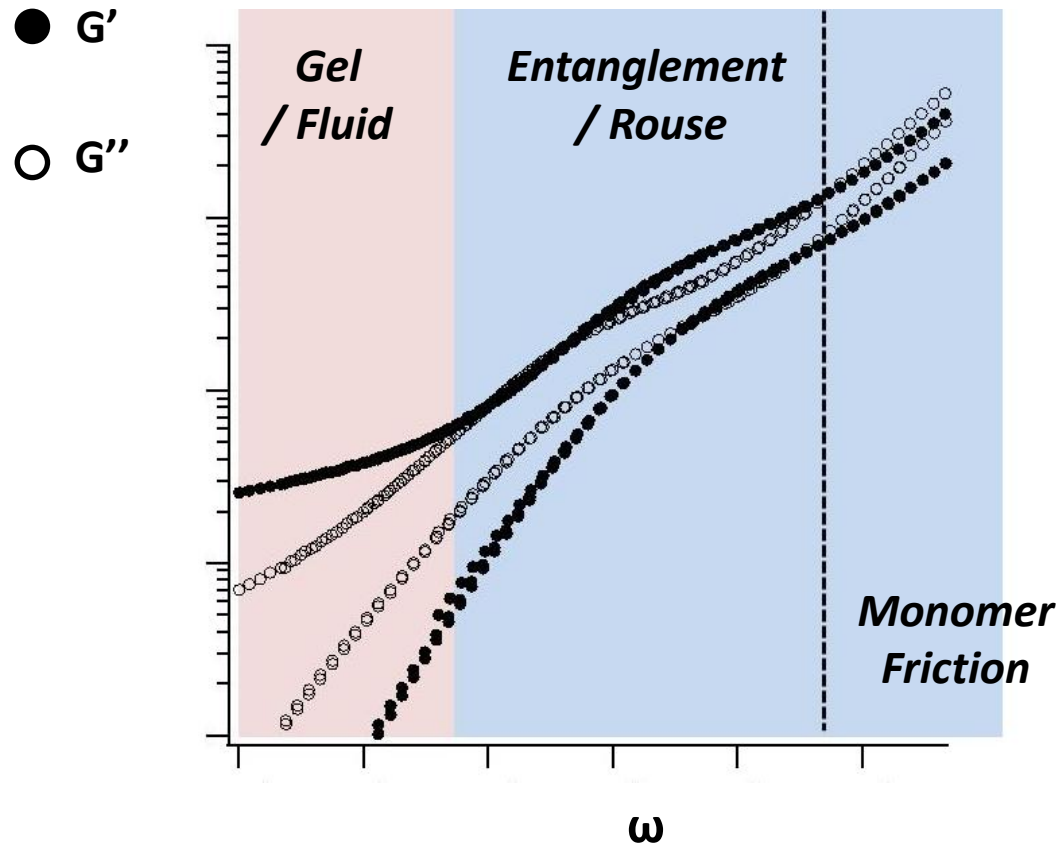
Stability of the nanostructure (below  $M_c$ ) ?



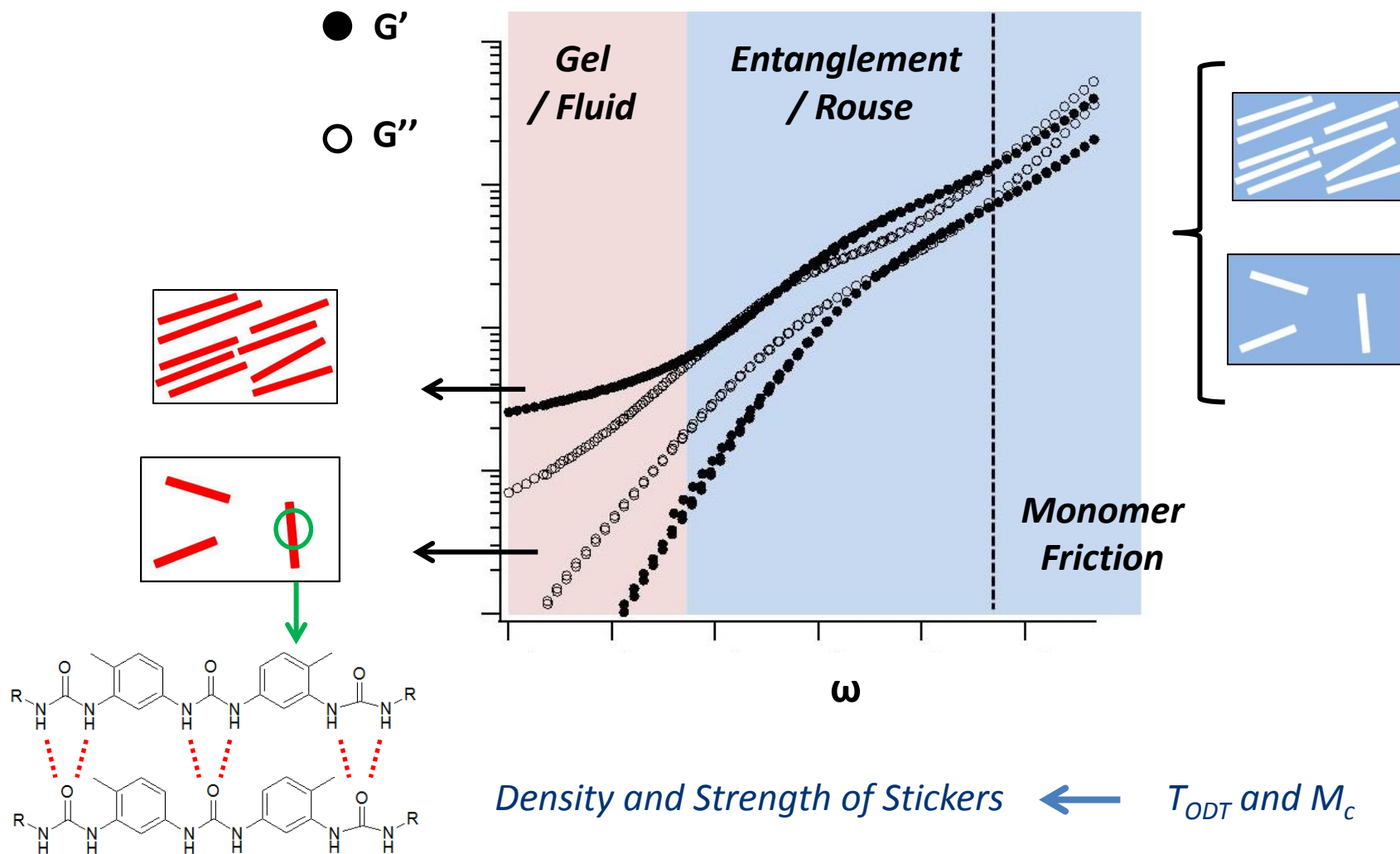
$T \leq T_{ODT} \rightarrow$  Frozen Structure over a long distance range  $\rightarrow$  Gel plateau

$T \geq T_{ODT} \rightarrow$  Scission / Association of Stickers  $\rightarrow$  Viscoelastic Fluids

# Conclusion on linear rheology of center-functionalized Polymers



# Conclusion on linear rheology of center-functionalized Polymers



Fine Control of the viscoelasticity via the chemical structure

*Thank you for all People  
in Project ANR SUPRADHESION*



**Guylaine Ducouret**  
Costantino Creton



**Cécile Fonteneau**  
Sandrine Pensec, Laurent Bouteiller

**Cyril Véchambre**  
Jean-Marc Chenal, Laurent Chazeau





*Thank you for your attention !*